

*Supplementary Information*

**Star-shaped Magnesium Tetraethynylporphyrin Bearing Four Peripheral Electron-accepting Diketopyrrolopyrrole Functionalities for Organic Solar Cells**

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## 1. Experimental Section

### 1.1. General

Materials were purchased from Tokyo Kasei (TCI) Co., Sigma-Aldrich Inc., and other commercial suppliers and used after appropriate purification. Anhydrous solvents (stabilizer-free) were purchased from WAKO Pure Chemical. Compound **7** was synthesized as according to previous literature.<sup>1</sup> Compounds **11**, **DPP**, **DPP-Br** were prepared according to previous papers.<sup>2-4</sup> Compounds **DPP-CHO**, **Br-DPP-CHO** were synthesized according to previously reported procedures.<sup>5-6</sup> All reactions dealing with air- or moisture-sensitive compounds were carried out in a dry reaction vessel under nitrogen or argon. All reactions were monitored by thin layer chromatography (TLC, eluent, CHCl<sub>3</sub>/CH<sub>2</sub>Cl<sub>2</sub>). The NMR spectra were measured on a Bruker US400 for <sup>1</sup>H NMR and <sup>13</sup>C NMR, respectively, reported in parts per million from tetramethylsilane. High-resolution mass spectra were acquired by MALDI using a time-of-flight mass analyzer on Bruker Ultra exTOF/TOF spectrometer. Elemental analyses were carried out using a Vario EL cube&III elemental analyzer. UV-vis absorption was recorded on Shimadzu UV-3600. Cyclic voltammetry (CV) and differential pulse voltammetry (DPV) were performed using a HOKUTO DENKO HZ-5000 voltammetric analyzer. All CV measurements were carried out in a one-compartment cell under argon gas, equipped with a glassy-carbon working electrode, a platinum wire counter electrode, and an Ag/Ag<sup>+</sup> reference electrode. The solvent with supporting electrolyte was a 0.1 mol L<sup>-1</sup> acetonitrile solution of tetrabutylammonium hexafluorophosphate (TBAPF<sub>6</sub>). Ionization potential was measured by a RIKEN KEIKI AC-3 photoemission yield spectroscopy in air. Current–voltage (*J–V*) characteristics were measured using a source meter (Keithley 2400) under sun AM 1.5G simulated sunlight irradiation (100 mW/cm<sup>2</sup>) from a solar simulator (EMS-35AAA, Ushio Spax Inc.), which was calibrated using a silicon diode (BS-520BK, Bunkoukeiki).

**Mobility Measurements:** Hole-only and electron-only devices were fabricated by using hole-only devices with a configuration of ITO/PEDOT:PSS/**3a**:PC<sub>61</sub>BM/MoO<sub>3</sub>/Ag and electron-only devices with a configuration of ITO/ZnO/**3a**:PC<sub>61</sub>BM/PFN/Al, The mobility was extracted by fitting the current density–voltage curves using space charge limited current (SCLC). The equation is as follows (Figure S11):

$$J = 9\varepsilon_0\varepsilon_r\mu V^2/8d^3$$

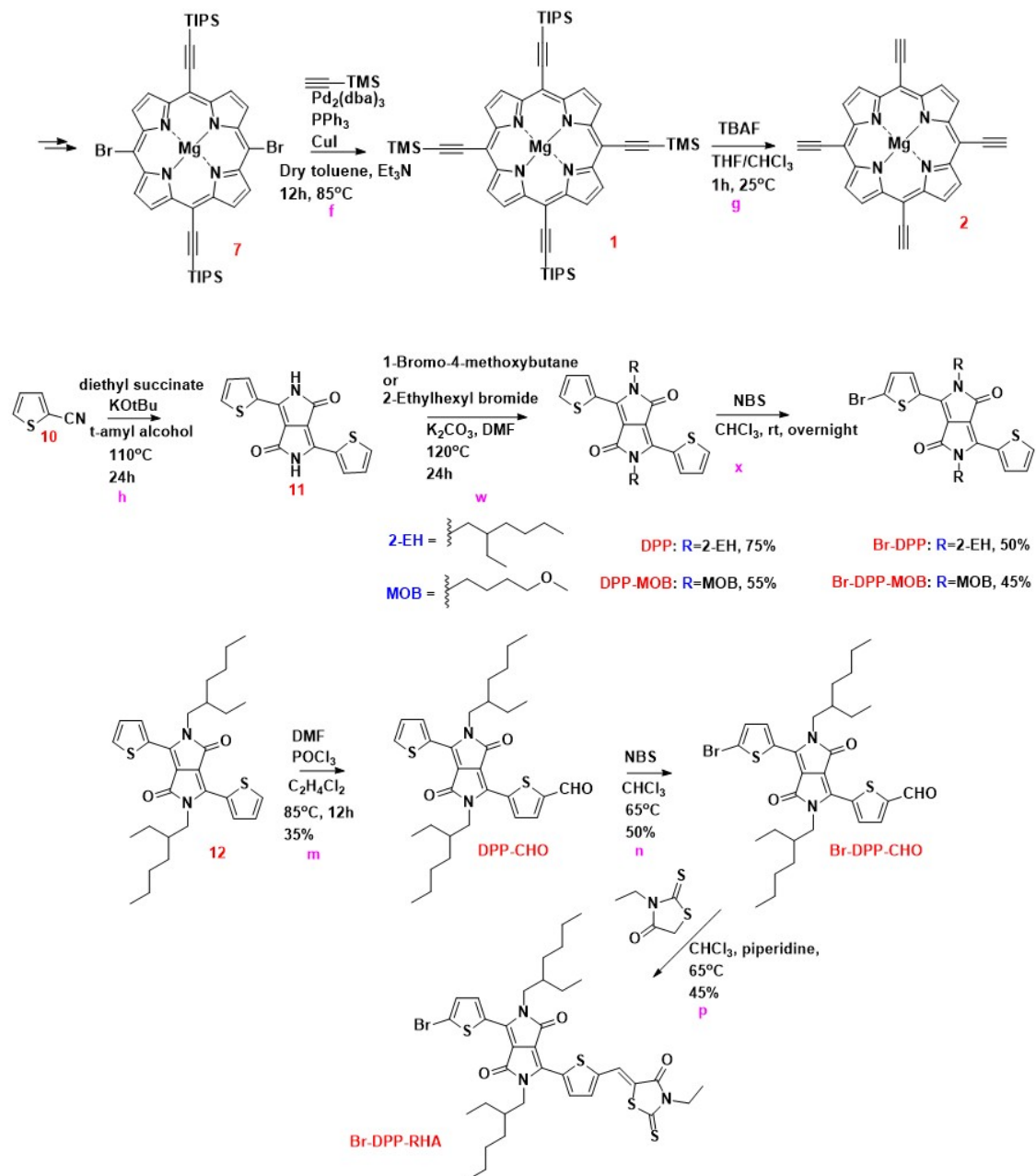
By the plots of  $J^{0.5}$  vs  $V$ , hole and electron mobilities can be calculated.

Where  $\mu$  is the mobility,  $J$  is the current,  $\varepsilon_0$  is the permittivity of free space,  $\varepsilon_r$  is the relative permittivity of the material,  $V$  is the effective voltage, and  $d$  is the thickness of the active layer.

**Theoretically  $J_{SC}$  calculation:** According to the following equation, the theoretically estimated  $J_{SC}^{cal}$  values were also obtained by integrating the EQE spectra with the AM1.5G solar spectra (Table S4):

$$J_{SC}^{cal} = q \int EQE(\lambda) AM1.5G(\lambda) d\lambda$$

## 1.2. Synthesis of the Starting Materials



**Scheme S1.** Synthetic routes of the starting materials.

### 1.3. Synthetic procedures

**Magnesium(II) 5,15-bis(triisopropylsilylethynyl)-10,20-bis(trimethylsilylethynyl) porphyrinate (1).** Porphyrin **7** (500 mg, 0.59 mmol) was dissolved in dry THF (40 mL) and dry triethylamine (20 mL) was added. The mixture was purged with Argon for 30 min. Then Pd<sub>2</sub>(dba)<sub>3</sub> (54.0 mg, 0.059 mmol), PPh<sub>3</sub> (15.0 mg, 0.059 mmol), CuI (5.5 mg, 0.029 mmol), and trimethylsilylacetylene (578 mg, 5.88 mmol) were added. The mixture was stirred at 80 °C for 12 h under argon, the reaction was quenched with saturated brine. After the mixture was extracted with chloroform (50 mL × 2), the combined organic layers were dried with anhydrous MgSO<sub>4</sub> and concentrated. Finally, the residue was purified with silica gel column by using ethyl acetate/petroleum ether (1/10) to afford compound **1** as a dark green powder (411 mg, 85% yield). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 9.62 (d, *J* = 4.4 Hz, 4H), 9.59 (d, *J* = 4.4 Hz, 4H), 1.52–1.41 (m, 42H), 0.62 (s, 18H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 151.86, 151.60, 131.57, 131.42, 109.08, 107.26, 103.21, 102.79, 101.15, 98.03, 60.22, 18.93, 11.47, 0.66. MALDI-TOF-HRMS (+) (*m/z*): calcd for C<sub>52</sub>H<sub>68</sub>MgN<sub>4</sub>Si<sub>4</sub> (M<sup>+</sup>): 884.4371, found 884.4355. Anal. Calc. for C<sub>52</sub>H<sub>68</sub>MgN<sub>4</sub>Si<sub>4</sub>: C, 70.51; H, 7.74; Mg, 2.74; N, 6.33; Si, 12.68. Found: C, 70.29; H, 7.63; N, 6.27.

**Magnesium(II) 5,10,15,20-tetraethynylporphyrin (2).** To a solution of porphyrin **9** (200 mg, 0.23 mmol) in a mixture of CH<sub>2</sub>Cl<sub>2</sub> (30 mL) and THF (15 mL) was added TBAF (1.5 mL, 1M in THF). After the mixture was stirred at room temperature for 1 h under nitrogen, water (20 mL) was added. Then the mixture was extracted with chloroform (50 mL × 2), the combined organic layers were dried with anhydrous MgSO<sub>4</sub> and concentrated, the residue was purified with silica gel column by using ethyl acetate/DCM/petroleum ether (1/5/10) to afford compound **2** as dark green solid, and washed with petroleum ether and methanol (62.6 mg, 65% yield). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 9.57 (d, *J* = 1.3 Hz, 8H), 4.67 (d, *J* = 0.9 Hz, 4H). MALDI-TOF-HRMS (+) (*m/z*): calcd for C<sub>28</sub>H<sub>12</sub>MgN<sub>4</sub> (M<sup>+</sup>): 428.0912, found 428.0901. Anal. Calc. for C<sub>28</sub>H<sub>12</sub>MgN<sub>4</sub>: C, 78.44; H, 2.82; Mg, 5.67; N, 13.07. Found: C, 77.74; H, 2.91; N, 12.68.

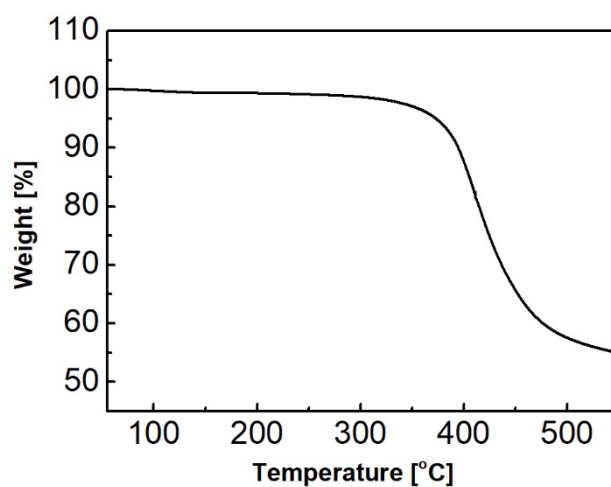
**3-(5-Bromothiophen-2-yl)-6-(5-((3-ethyl-4-oxo-2-thioxothiazolidin-5-ylidene)methyl)thiophen-2-yl)-2,5-bis(2-ethylhexyl)-2,5-dihydropyrrolo[3,4-c]pyrrole-1,4-dione (Br-DPP-RHA).** 3-Ethylrhodanine (123 mg, 0.76 mmol) was added to a solution of compound **Br-DPP-CHO** (300 mg, 0.48 mmol) in dry chloroform (30 mL) in a two-necked 100-mL round-bottomed flask under an Ar atmosphere. Piperidine (1.5 mL) was then added and the reaction mixture was stirred at 50 °C overnight. The reaction was quenched with water and extracted with chloroform (30 mL × 2). The organic phase was dried with Na<sub>2</sub>SO<sub>4</sub> and the solvent removed under reduce pressure. The crude mixture was purified by column chromatography (silica gel; eluent: ethyl acetate/CH<sub>2</sub>Cl<sub>2</sub> = 1:20) and washed with methanol to give a copper-color solid (276 mg, 75% yield). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 9.08 (d, *J* = 4.3 Hz, 1H), 8.78 (d, *J* = 4.2 Hz, 1H), 7.88 (s, 1H), 7.49 (d, *J* = 4.3 Hz, 1H), 7.24 (d, *J* = 4.2 Hz, 1H), 4.20 (q, *J* = 7.1 Hz, 2H), 4.11–3.89 (m, 4H), 1.93–1.79 (m, 2H), 1.46–1.20 (m, 19H),

0.97–0.82 (m, 12H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  190.83, 166.99, 161.27, 161.08, 141.88, 140.28, 138.37, 136.40, 136.37, 135.57, 134.33, 131.67, 130.98, 123.53, 123.49, 120.02, 109.84, 108.31, 46.08, 40.09, 39.61, 39.08, 30.25, 30.10, 28.47, 28.26, 28.23, 23.51, 23.47, 23.42, 23.40, 23.18, 23.04, 14.13, 14.04, 12.30, 10.47, 10.42. MALDI-TOF-HRMS (+) ( $m/z$ ): calcd for  $\text{C}_{28}\text{H}_{12}\text{MgN}_4$  ( $\text{M}^+$ ): 773.1449, found 773.1438. Anal. Calc. for  $\text{C}_{36}\text{H}_{44}\text{BrN}_3\text{O}_3\text{S}_4$ : C, 55.80; H, 5.72; Br, 10.31; N, 5.42; O, 6.19; S, 16.55. Found: C, 55.81; H, 5.80; N, 5.33; O, 7.00; S, 16.57.

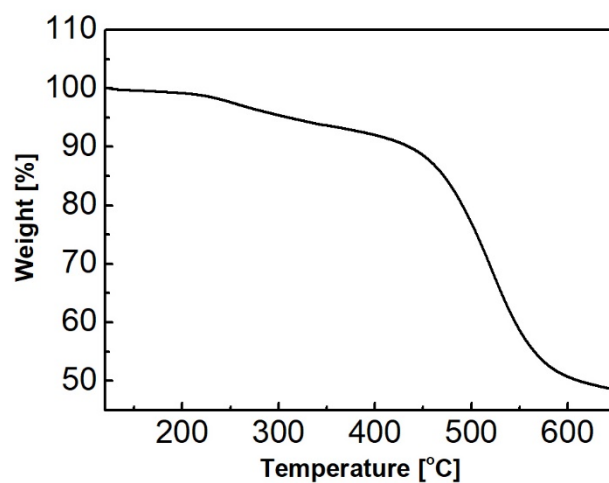
**2,5-Bis(4-methoxybutyl)-3,6-di(thiophen-2-yl)-2,5-dihydropyrrolo[3,4-c]pyrrole-1,4-dione (DPP-MOB).** Potassium carbonate (4.84 g, 35.0 mmol) was added to a solution of **11** (3.01 g, 10.0 mmol) in dry DMF (60 mL), and the mixture was heated at 120 °C for 1 h. Then 1-Bromo-4-methoxybutane (5.35 g, 32.0 mmol) was added and the mixture was stirred for 24 h at the same temperature. The suspension was then filtered and the solvent removed under reduce pressure. The crude mixture was washed with methanol to give a copper-color solid, the solid was dried under vacuum and used for next step without further purification (3.31g, 70% yield). MALDI-TOF-HRMS (+) ( $m/z$ ): calcd for  $\text{C}_{24}\text{H}_{28}\text{N}_2\text{O}_4\text{S}_2$  ( $\text{M}^+$ ): 472.1490, found 472.1475. Anal. Calc. for  $\text{C}_{24}\text{H}_{28}\text{N}_2\text{O}_4\text{S}_2$ : C, 60.99; H, 5.97; N, 5.93; O, 13.54; S, 13.57. Found: C, 70.68; H, 4.96; N, 5.58; O, 12.28; S, 14.23.

**3-(5-Bromothiophen-2-yl)-2,5-bis(4-methoxybutyl)-6-(thiophen-2-yl)-2,5-dihydropyrrolo[3,4-c]pyrrole-1,4-dione (Br-DPP-MOB).** DPP-MOB (0.76 g, 1.60 mmol) and N-bromosuccinimide (0.28 g, 1.60 mmol) were dissolved in  $\text{CHCl}_3$  (50 mL), then the solution was protected from light and stirred at room temperature for 24 h. Water (25 mL) was added and the mixture was extracted with chloroform (50 mL  $\times$  2). The organic layer was separated and dried over magnesium sulfate. After the solvent was removed under reduced pressure, the crude product was purified by column chromatography on silica gel with dichloromethane:ethyl acetate from 50:1 to 20:1 (v/v) to afford a red powder (0.49 g, 55% yield).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  8.90 (dd,  $J = 3.9$  Hz, 1.1 Hz, 1H), 8.64 (d,  $J = 4.2$  Hz, 1H), 7.65 (dd,  $J = 5.0$ , 1.1 Hz, 1H), 7.29–7.27 (m, 1H), 7.22 (d,  $J = 4.2$  Hz, 1H), 4.13–4.01 (m, 4H), 3.41 (m, 4H), 3.33 (s, 3H), 3.32 (s, 3H), 1.88–1.60 (m, 8H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  161.24, 161.08, 140.36, 138.46, 135.49, 135.06, 131.56, 131.12, 131.06, 129.61, 128.70, 118.90, 107.82, 107.53, 72.14, 72.07, 58.62, 58.59, 41.94, 41.91, 26.95, 26.93, 26.91, 26.84. MALDI-TOF-HRMS (+) ( $m/z$ ): calcd for  $\text{C}_{24}\text{H}_{27}\text{BrN}_2\text{O}_4\text{S}_2$  ( $\text{M}^+$ ): 550.0596, found 550.0582. Anal. Calc. for  $\text{C}_{24}\text{H}_{27}\text{BrN}_2\text{O}_4\text{S}_2$ : C, 52.27; H, 4.93; Br, 14.49; N, 5.08; O, 11.60; S, 11.63. Found: C, 52.65; H, 4.96; N, 4.77; O, 12.01; S, 11.99.

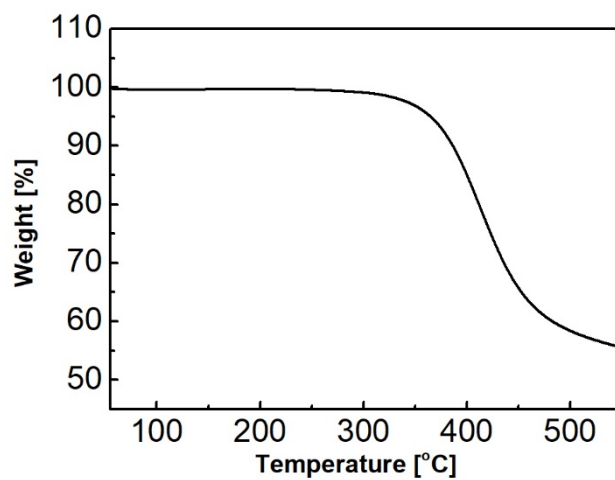
## 2. Thermal Properties of the Porphyrin Derivatives



**Figure S1.** TGA data for **3b** under a N<sub>2</sub> gas flow with temperature ramp rate of 10 °C/min until 550 °C. Temperature with 5% weight loss is 211.71 °C.

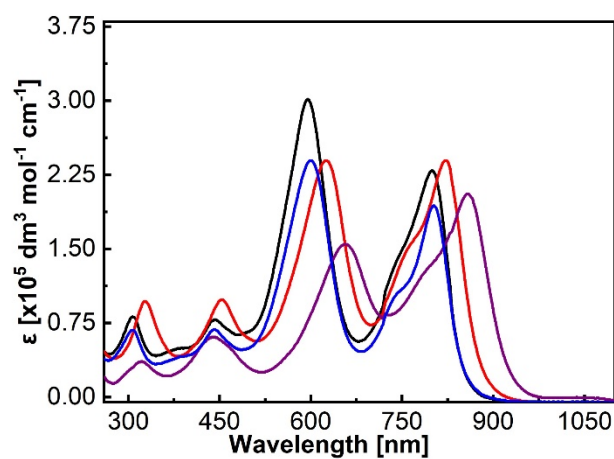


**Figure S2.** TGA data for **3c** under a N<sub>2</sub> gas flow with temperature ramp rate of 10 °C/min until 650 °C. Temperature with 5% weight loss is 210.26 °C.



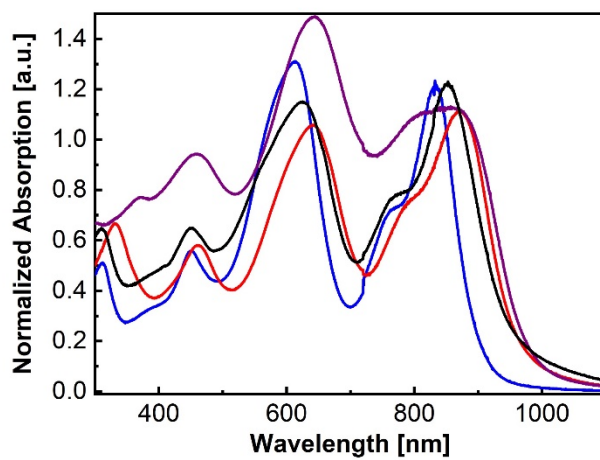
**Figure S3.** TGA data for **3d** under a N<sub>2</sub> gas flow with temperature ramp rate of 10 °C/min until 550 °C. Temperature with 5% weight loss is 242.44 °C.

### 3. Absorption Spectra and Electrochemical Data

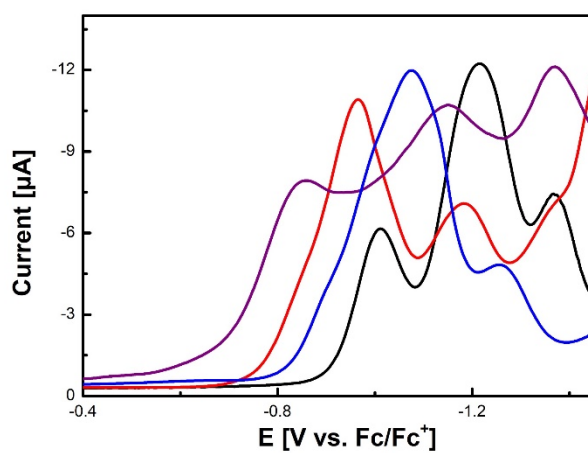


**Figure S4.** UV-vis absorption spectra of **3a** (black), **3b** (red), **3c** (purple), and **3d** (blue) in DCM.

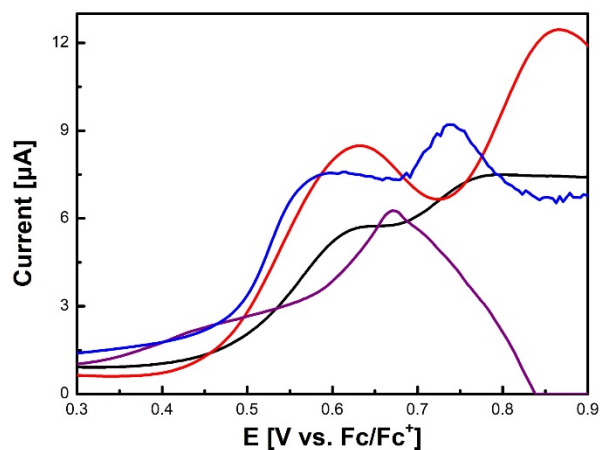




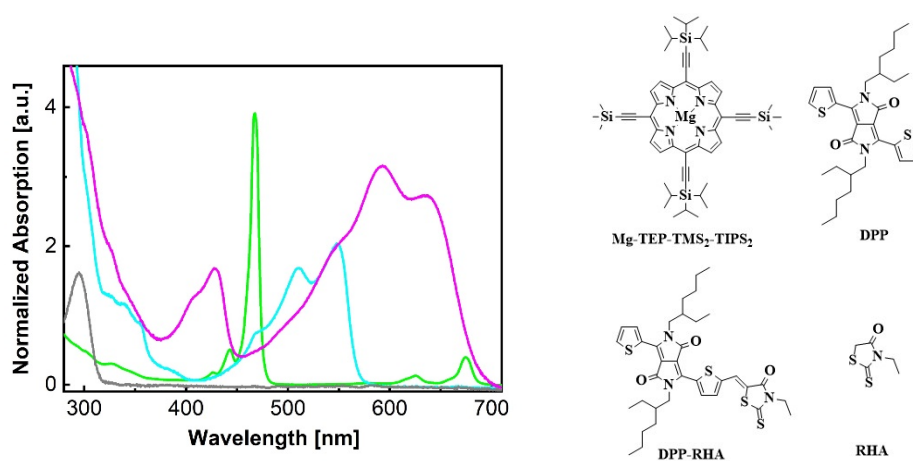
**Figure S5.** UV-vis absorption spectra of **3a** (black), **3b** (red), **3c** (purple), and **3d** (blue) as solid thin films, respectively.



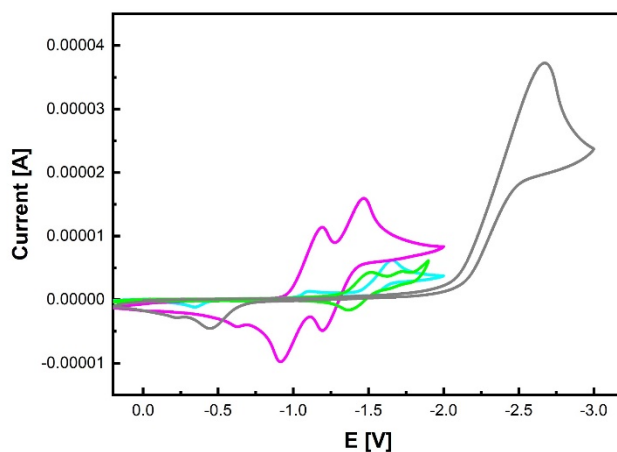
**Figure S6.** DPV of the reduction range for **3a** (black), **3b** (red), **3c** (purple), and **3d** (blue) in THF.



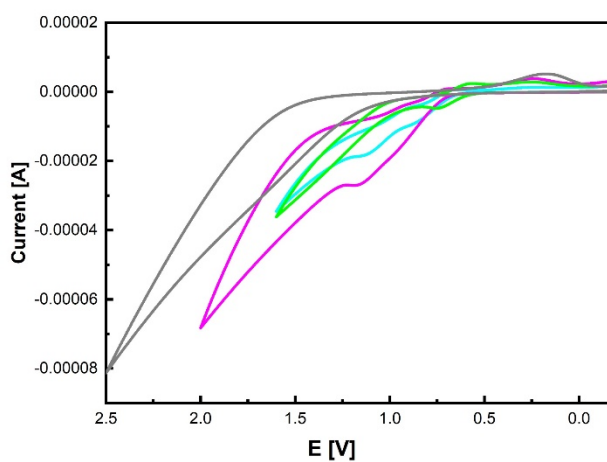
**Figure S7.** Differential pulse voltammetry (DPV) of the oxidation range for **3a** (black), **3b** (red), **3c** (purple), and **3d** (blue) in THF.



**Figure S8.** UV-vis absorption spectra of **Mg-TEP-TMS<sub>2</sub>-TIPS<sub>2</sub>** (green), **DPP** (cyan), **DPP-RHA** (magenta), and **RHA** (gray) in THF.

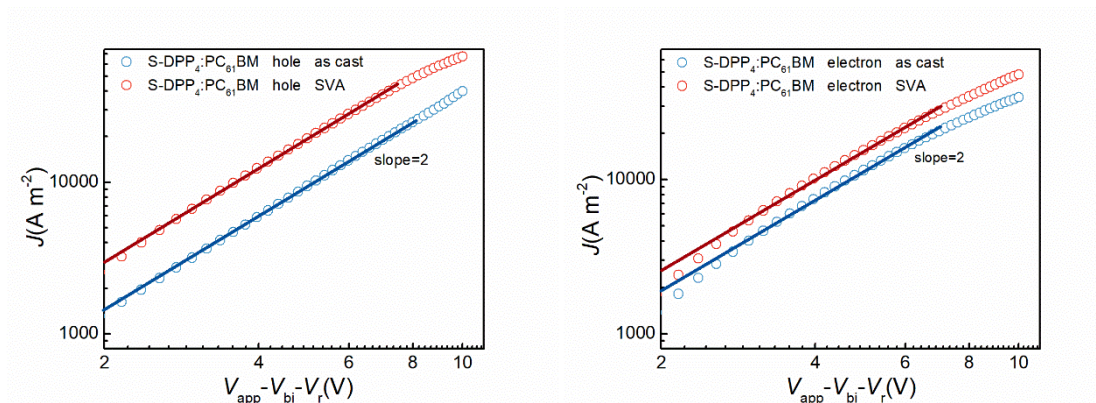


**Figure S9.** Cyclic voltammety of the reduction range of **Mg-TEP-TMS<sub>2</sub>-TIPS<sub>2</sub>** (green), **DPP** (cyan), **DPP-RHA** (magenta), and **RHA** (gray) in THF containing TBAPF<sub>6</sub> (0.1 M) as a supporting electrolyte.



**Figure S10.** Cyclic voltammety of the oxidation range of **Mg-TEP-TMS<sub>2</sub>-TIPS<sub>2</sub>** (green), **DPP** (cyan), **DPP-RHA** (magenta), and **RHA** (gray) in THF containing TBAPF<sub>6</sub> (0.1 M) as a supporting electrolyte.

#### 4. Photovoltaic Data



**Figure S11.** The hole mobility and electron mobility of **3a**:PC<sub>61</sub>BM before and after SVA at the best optimized device condition measured by SCLC method.

**Table S1.** Photovoltaic performance of **3a** in conventional BHJ devices with different D/A ratio.

D:A	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
1:1	0.74	14.07	63.21	6.56
1:1.5	0.72	16.42	62.24	7.40
1:2	0.72	15.26	60.41	6.70

**Table S2.** Thickness influence on photovoltaic performance of **3a** in BHJ devices after SVA for 20s.

Thickness [nm]	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
90	0.73	15.21	63.23	7.04
110	0.72	16.42	62.24	7.40
130	0.72	16.54	56.56	6.80

**Table S3.** Photovoltaic performance of the **3a** in the BHJ devices with different SVA time.

SVA	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
as-cast	0.75	16.35	47.45	5.80
THF 10s	0.74	16.78	51.14	6.37
THF 20s	0.72	16.42	62.24	7.40
THF 30s	0.73	15.08	60.84	6.69
THF 40s	0.72	13.75	59.96	6.00

**Table S4.** Photovoltaic performance of the devices under 100 mW/cm<sup>2</sup> simulated solar irradiation.

<b>3a:PC<sub>61</sub>BM</b>	<b><math>J_{SC}</math> [mA/cm<sup>2</sup>]</b>	<b><math>J_{SC}^{cal}</math> [mA/cm<sup>2</sup>]</b>	<b>error</b>	<b><math>J_{SAT}</math> [mA cm<sup>-2</sup>]</b>	<b>P(E,T)<sup>a</sup></b>	<b>P(E,T)<sup>b</sup></b>
as cast	16.35	15.74	-3.7%	19.25	84.9%	57.6%
SVA	16.42	15.88	-3.3%	17.25	95.2%	77.9%

<sup>a</sup> Exciton dissociation probability under short-circuit condition. <sup>b</sup> Exciton dissociation probability under maximum power output condition.

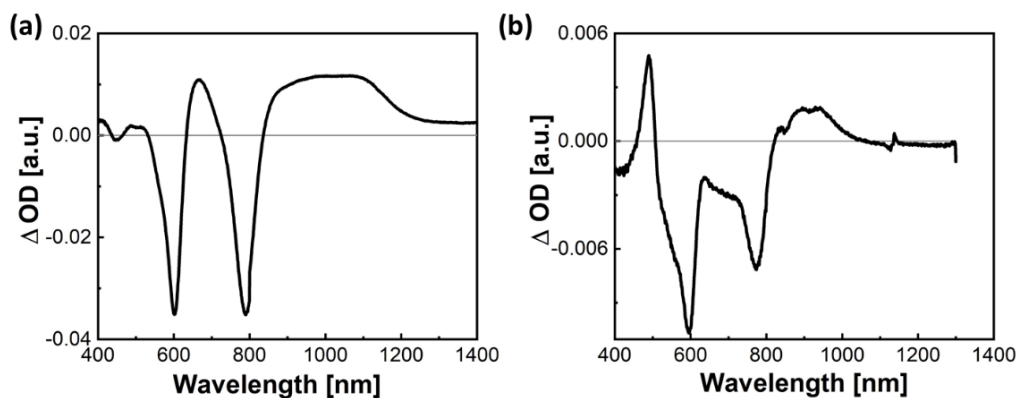
**Table S5.** Photovoltaic performance of the devices under 100 mW/cm<sup>2</sup> simulated solar irradiation.

<b>donor</b>	<b>acceptor</b>	<b>Conc.</b> [mg/mL]	<b>additive</b>	<b>TA</b> [°C]	<b>SVA</b> [s]	<b><math>V_{OC}</math></b> [V]	<b><math>J_{SC}</math></b> [mA/cm <sup>2</sup> ]	<b>FF</b> [%]	<b>PCE</b> [%]
<b>3d</b>	PC <sub>61</sub> BM	30	1%Py	–	–	0.67	9.40	43.0	2.71
<b>3b</b>	PC <sub>61</sub> BM	30	–	–	–	0.63	0.92	37.0	0.22
<b>3c</b>	PC <sub>61</sub> BM	30	–	–	–	0.59	1.58	28.9	0.27

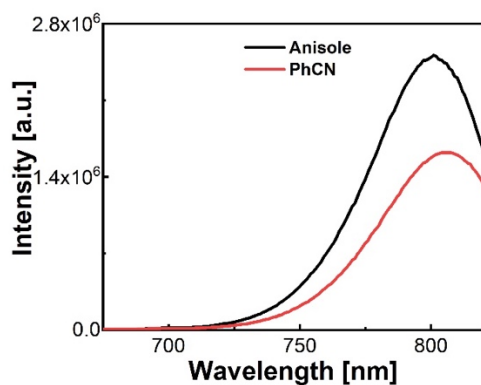
**Table S6.** Photovoltaic performance of **3a** in conventional BHJ devices under 100 mW/cm<sup>2</sup> simulated solar irradiation.

<b>donor</b>	<b>acceptor</b>	<b>Conc.</b> [mg/mL]	<b>D:A</b>	<b>TA</b> [°C]	<b>SVA</b> [s]	<b><math>V_{OC}</math></b> [V]	<b><math>J_{SC}</math></b> [mA/cm <sup>2</sup> ]	<b>FF</b> [%]	<b>PCE</b> [%]
<b>3a</b>	PC <sub>71</sub> BM	30	1:1	–	–	0.73	6.23	49.34	2.23
<b>3a</b>	PC <sub>71</sub> BM	30	1:1	–	THF 20s	0.73	8.49	59.61	3.70
<b>3a</b>	PC <sub>71</sub> BM	30	1:1.5	–	–	0.72	4.59	52.45	1.73
<b>3a</b>	PC <sub>71</sub> BM	30	1:1.5	–	THF 20s	0.73	3.81	56.20	1.57
<b>3a</b>	PC <sub>71</sub> BM	30	1:2	–	–	0.71	4.29	54.20	1.66
<b>3a</b>	PC <sub>71</sub> BM	30	1:2	–	THF 20s	0.71	3.60	55.81	1.43

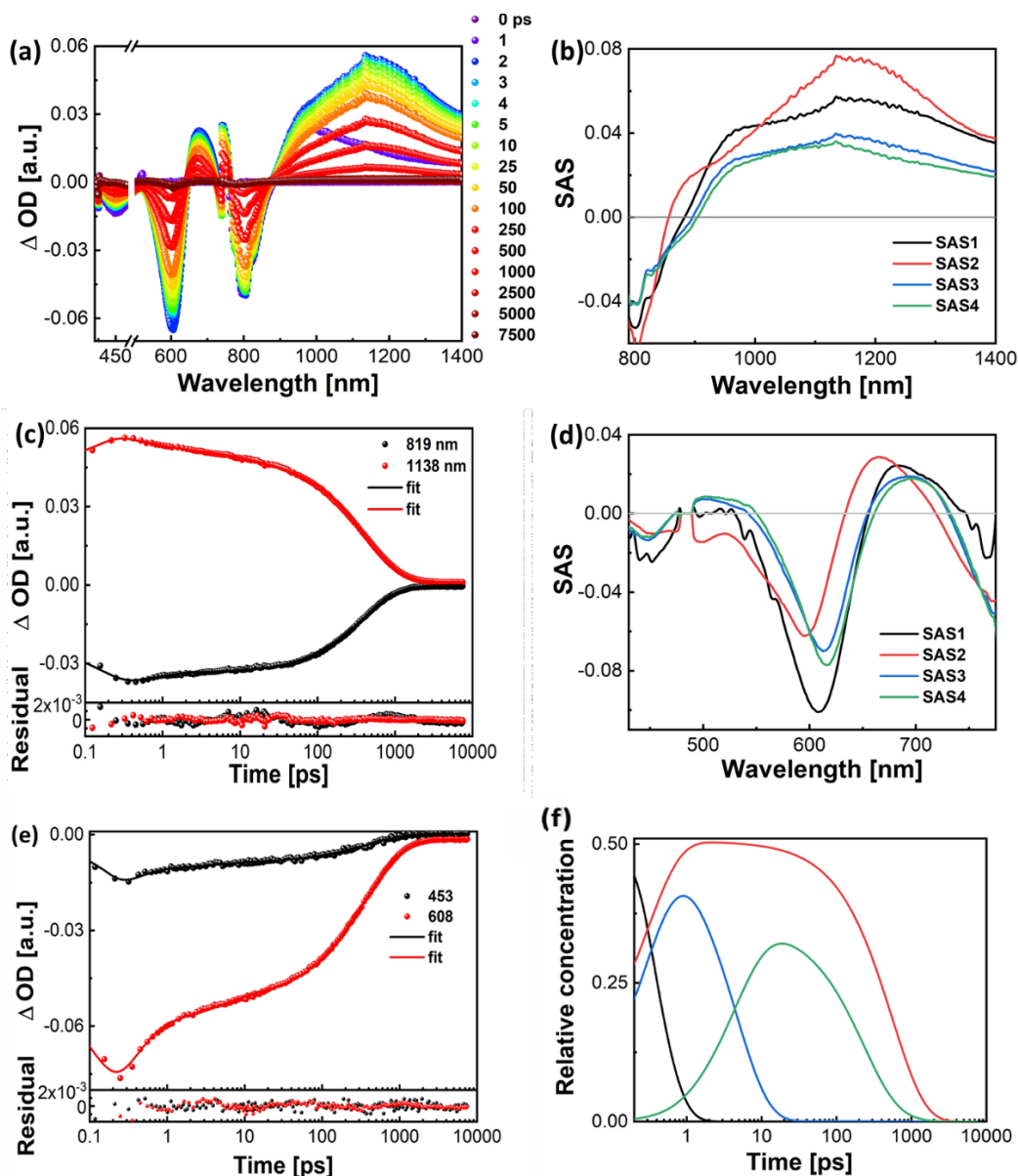
## 5. Spectroelectrochemistry, Fluorescence, and Time-resolved Photophysics Data



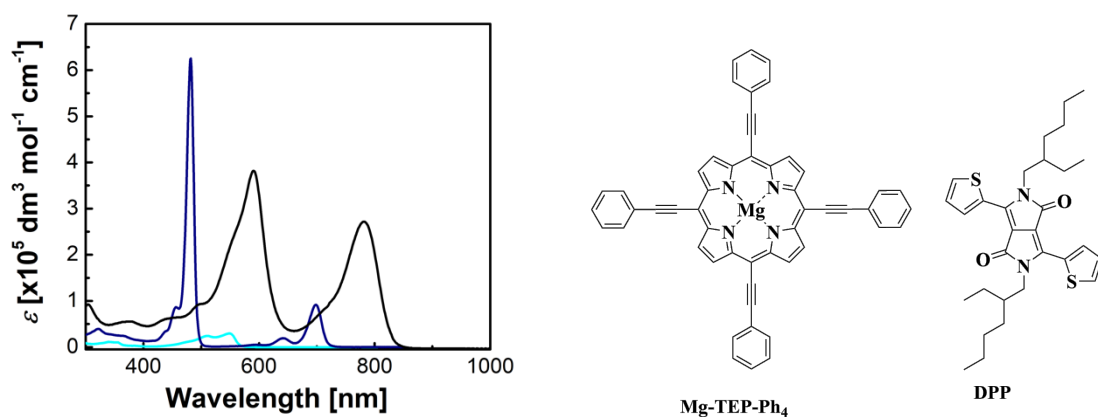
**Figure S12.** The spectroelectrochemical spectra of **3a** in *o*-dichlorobenzene upon application of (a) +0.8 V and (b) -1.2V versus Ag/Ag<sup>+</sup>, with tetrabutylammonium perchloride used as the electrolyte.



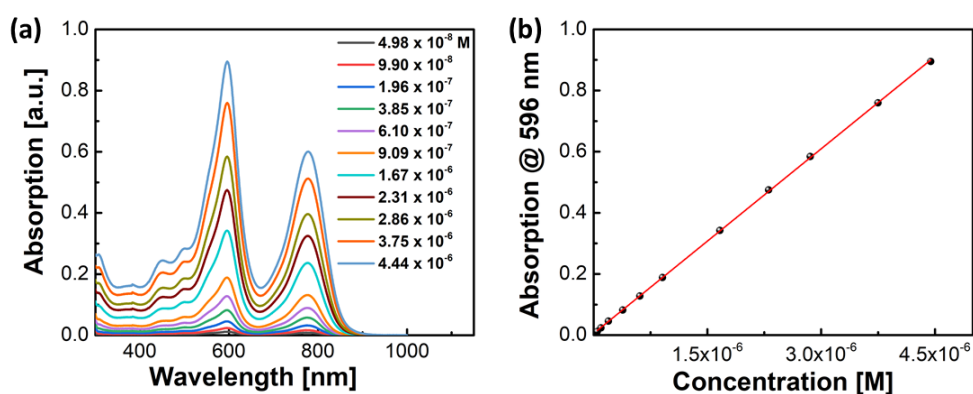
**Figure S13.** Room-temperature fluorescence spectra of **3a** in anisole and PhCN upon 480 nm excitation with matching optical density of 0.05.



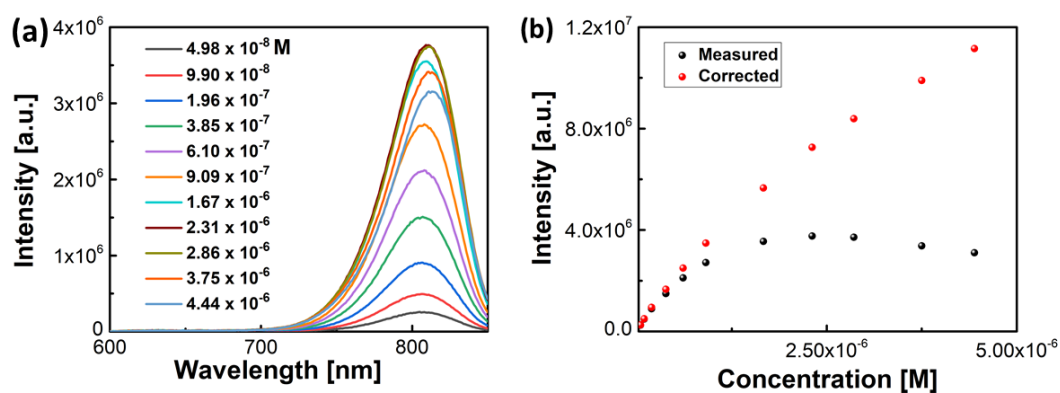
**Figure S14.** (a) Differential absorption spectra (visible and near-infrared) obtained upon femtosecond flash photolysis (480 nm) of **3a** in argon-saturated PhCN with several time delays between 0 and 7500 ps at room temperature. (b) Deconvoluted transient absorption spectra of  $\text{MgP}^{\delta+}\text{-(DPP}_4\text{)}^{\delta-}$  ( $\text{MgP}^{\delta+}\text{-(DPP}_4\text{)}^{\delta-}$ ) (black),  $\{\text{MgP}^{\delta+}\text{-(DPP}_4\text{)}^{\delta-}\}_{\text{relaxed}}$  (red),  $\text{MgP}^+\text{-(DPP)}^{\cdot-}\text{(DPP}_3\text{)}$  (blue), and  $\text{MgP}^+\text{-(DPP}_4\text{)}^{\cdot-}$  (green) as obtained by global target analysis (NIR). (c) Time absorption profiles of **3a** and the corresponding fits (NIR). (d) Deconvoluted transient absorption spectra of  $\text{MgP}^{\delta+}\text{-(DPP}_4\text{)}^{\delta-}$  (black),  $\{\text{MgP}^{\delta+}\text{-(DPP}_4\text{)}^{\delta-}\}_{\text{relaxed}}$  (red),  $\text{MgP}^+\text{-(DPP)}^{\cdot-}\text{(DPP}_3\text{)}$  (blue), and  $\text{MgP}^+\text{-(DPP}_4\text{)}^{\cdot-}$  (green) as obtained by global target analysis (Vis). (e) Time absorption profiles of **3a** and the corresponding fits (Vis). (f) Evolution of the population of the involved states.



**Figure S15.** UV-Vis absorption spectra of **3a** (black), **Mg-TEP-Ph<sub>4</sub>** (navy), and **DPP** (cyan) in THF.



**Figure S16.** a) Concentration-dependent UV-Vis absorption spectra of **3a** in PhCN. b) Correlation of the absorption intensity as a function of concentration detected at 596 nm in PhCN.

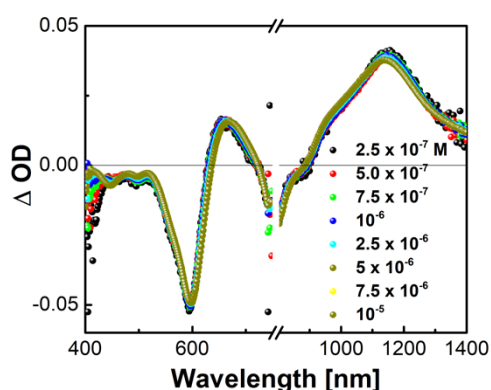




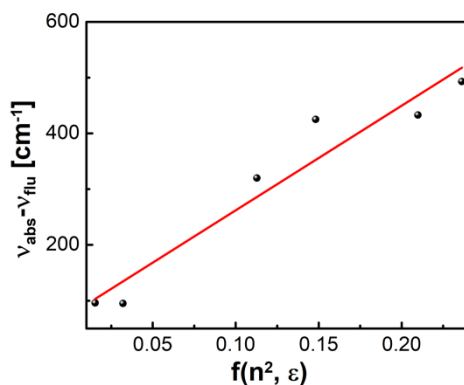
**Figure S17.** a) Concentration-dependent fluorescence spectra of **3a** in PhCN upon 575 nm excitation. b) Correlation of the fluorescence intensity detected at 809 nm as a function of concentration in PhCN. The spectra was corrected for the primary inner-filter effect using equation:

$$F_{corr.} = F_{obs.} \times 10^{\frac{A_{exc.} + A_{em.}}{2}},$$

where  $F_{corr.}$  is the corrected fluorescence value,  $F_{obs.}$  is the measured fluorescence value,  $A_{exc.}$  the absorption value at the excitation wavelength,  $A_{em.}$  is the absorption value at the emission wavelength. Owing to the overlap of the absorption and fluorescence spectra, the non-linearity of the corrected values stems from the secondary inner filter-effect.<sup>7-8</sup>



**Figure S18.** Differential absorption spectra with the time delay of 5 ps with increasing concentration of **3a**. The spectra were obtained upon femtosecond pump-probe experiment (775 nm) in argon-saturated benzonitrile.

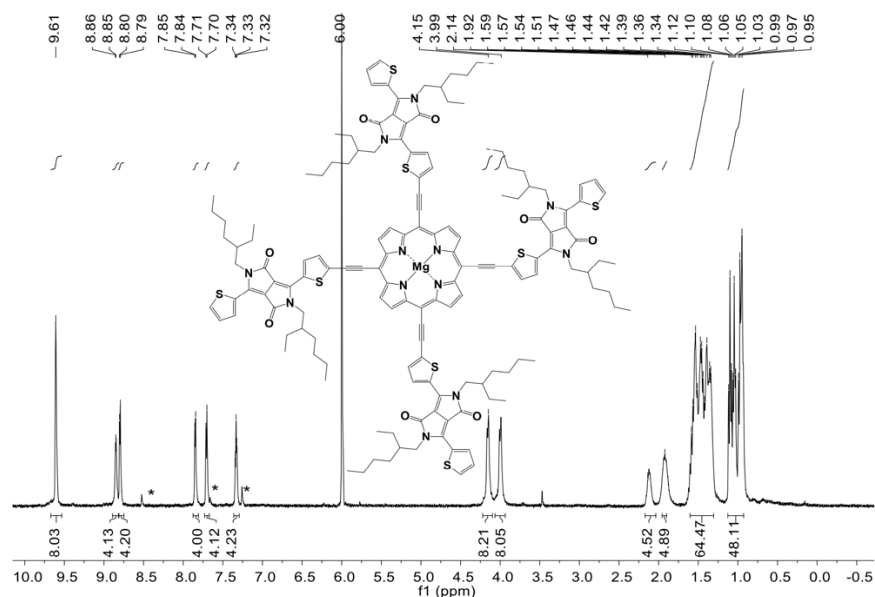


**Figure S19.** Lippert-Mataga plot showing Stokes shift as a function of solvent orientation polarizability  $f(n^2, \epsilon)$ :

$$f(n^2, \epsilon) = \frac{\epsilon - 1}{2\epsilon + 1} - \frac{n^2 - 1}{2n^2 + 1}$$

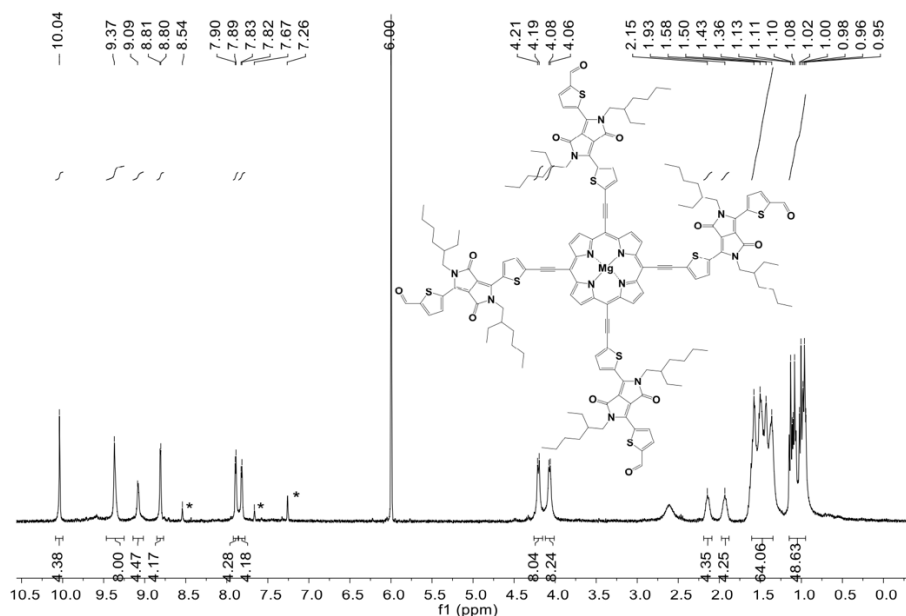
Where  $\epsilon$  stands for the dielectric constant and  $n$  is the refractive index. The Stokes shifts were measured in toluene, chlorobenzene, anisole, chloroform, tetrahydrofuran, and benzonitrile.

## 7. $^1\text{H}$ and $^{13}\text{C}$ NMR Spectra



**Figure S20.**  $^1\text{H}$  NMR spectrum for **3a**.

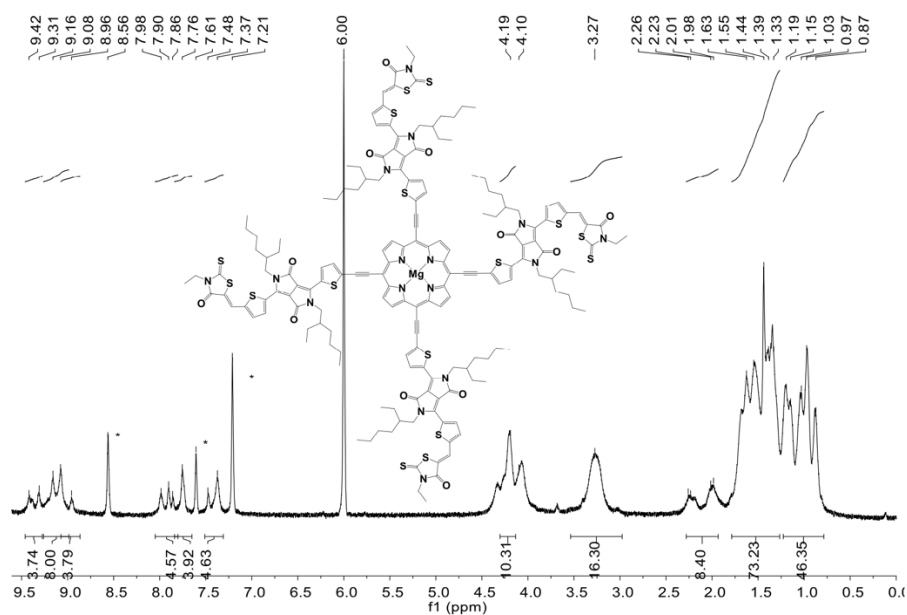
$^1\text{H}$  NMR (400 MHz, tetrachloroethane- $d_2$  with 1% pyridine- $d_5^*$ , 100 °C):  $\delta$  9.61 (s, 8H, porphyrin), 8.85 (s, 4H, Th), 8.80 (d,  $J = 3.6$  Hz, 4H, Th), 7.85 (d,  $J = 4.1$  Hz, 4H, Th), 7.71 (d,  $J = 4.7$  Hz, 4H, Th), 7.33 (t,  $J = 4.0$  Hz, 4H, Th), 4.15 (d,  $J = 6.8$  Hz, 8H,  $\text{NCH}_2$ ), 4.00 (d,  $J = 6.9$  Hz, 8H,  $\text{NCH}_2$ ), 2.13 (m, 4H, CH), 1.92 (m, 4H, CH), 1.61–1.28 (m, 64H,  $\text{CH}_2$ ), 1.15–0.88 (m, 48H,  $\text{CH}_3$ ).



**Figure S21.**  $^1\text{H}$  NMR spectrum for **3b**.

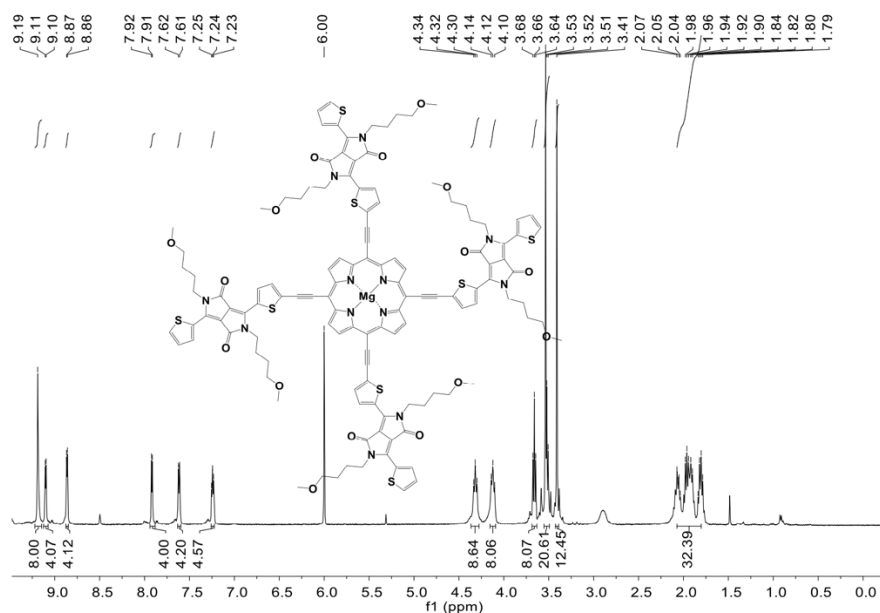
$^1\text{H}$  NMR (400 MHz, tetrachloroethane- $d_2$  with 1% pyridine- $d_5^*$ , 100 °C):  $\delta$  10.04 (s, 4H, CHO), 9.37 (s, 8H, porphyrin), 9.08 (m, 4H, Th), 8.81 (d,  $J = 3.9$  Hz, 4H, Th), 7.89 (d,  $J = 3.4$  Hz, 4H, Th), 7.82 (m, 4H, Th), 4.20 (d,  $J = 7.6$  Hz, 8H,  $\text{NCH}_2$ ), 4.07 (d,  $J = 6.3$  Hz, 8H,  $\text{NCH}_2$ ), 2.15 (m, 4H, CH), 1.93

(m, 4H, CH), 1.75–1.47 (m, 64H, CH<sub>2</sub>), 1.18–0.92 (m, 48H, CH<sub>3</sub>).



**Figure S22.** <sup>1</sup>H NMR spectrum for **3c**.

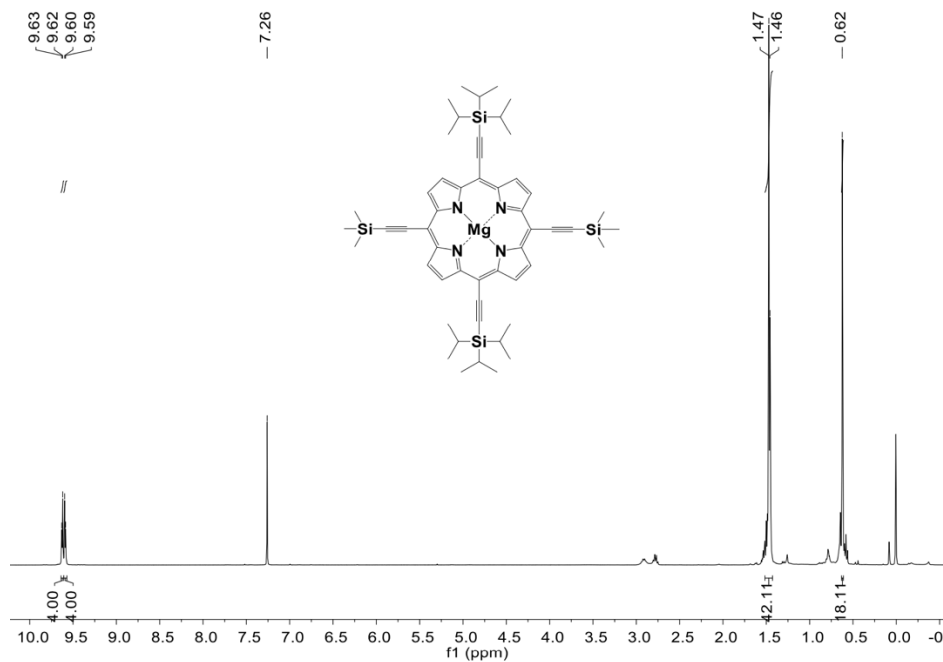
<sup>1</sup>H NMR (400 MHz, tetrachloroethane-*d*<sub>2</sub> with 5% pyridine-*d*<sub>5</sub><sup>\*</sup>, 100°C): δ 9.36 (m, 4H, Th), 9.12 (m, 8H, porphyrin), 8.96 (s, 4H, Th), 8.01–7.82 (m, 4H, Th), 7.75 (s, 4H, Th), 7.42 (m, 4H, Th), 4.13 (m, 8H, CSNCH<sub>2</sub>), 3.26 (m, 16H, NCH<sub>2</sub>), 2.12 (m, 8H, CH), 1.72–1.29 (m, 76H, CH<sub>2</sub> and CH<sub>3</sub>), 1.24–0.82 (m, 48H, CH<sub>2</sub>).



**Figure S23.** <sup>1</sup>H NMR spectrum for **3d**.

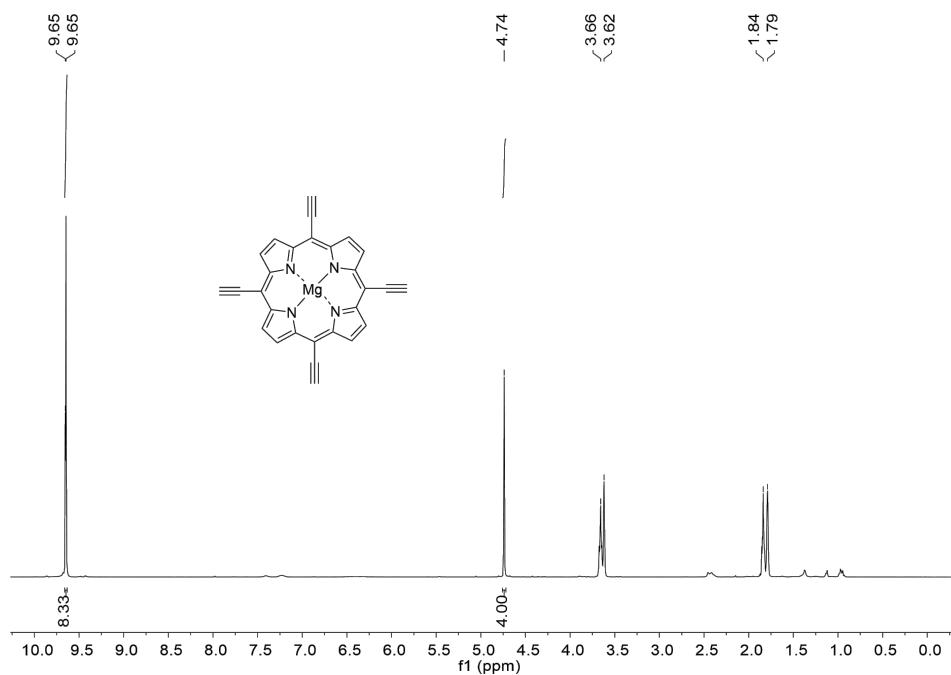
<sup>1</sup>H NMR (400 MHz, tetrachloroethane-*d*<sub>2</sub> with 1% pyridine-*d*<sub>5</sub>, 100 °C): δ 9.19 (s, 8H, porphyrin), 9.10 (d, *J* = 3.7 Hz, 4H, Th), 8.86 (d, *J* = 3.7 Hz, 4H, Th), 7.92 (d, *J* = 4.0 Hz, 4H, Th), 7.62 (d, *J* = 5.6 Hz, 4H, Th), 7.27–7.21 (br, 4H, Th), 4.37–4.26 (br, 8H, NCH<sub>2</sub>), 4.20–4.06 (br, 8H, NCH<sub>2</sub>), 3.66

(t,  $J = 6.3$  Hz, 8H, OCH<sub>2</sub>), 3.55–3.50 (m, 20H, OCH<sub>2</sub> and OCH<sub>3</sub>), 3.41 (m, 12H, OCH<sub>3</sub>), 2.11–1.77 (m, 32H, CH<sub>2</sub>).



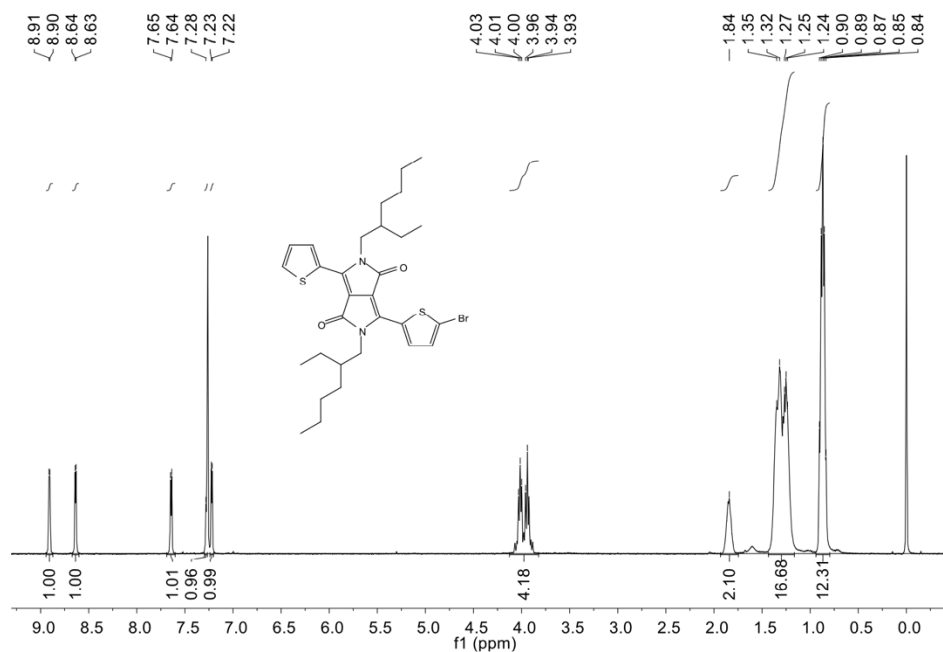
**Figure S24.** <sup>1</sup>H NMR spectrum for 1.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  9.62 (d,  $J = 4.4$  Hz, 4H, porphyrin), 9.59 (d,  $J = 4.4$  Hz, 4H, porphyrin), 1.52–1.41 (m, 42H, TIPS), 0.62 (s, 18H, TMS).



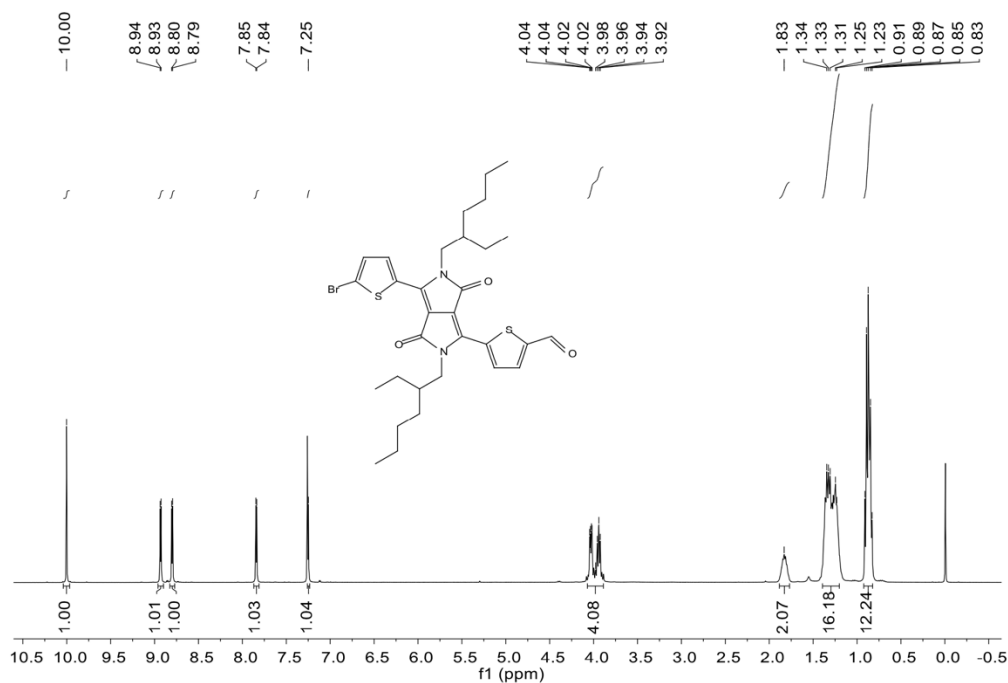
**Figure S25.** <sup>1</sup>H NMR spectrum for 2.

<sup>1</sup>H NMR (400 MHz, THF-*d*<sub>8</sub>):  $\delta$  9.65 (d,  $J = 1.3$  Hz, 8H, porphyrin), 4.74 (s, 4H, Ethynyl).



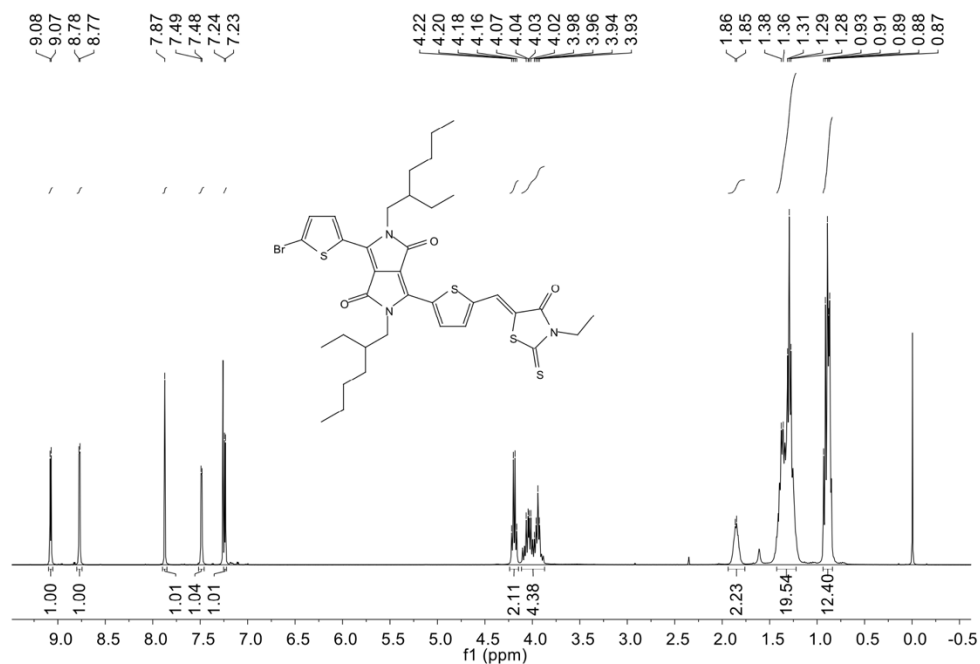
**Figure S26.**  $^1\text{H}$  NMR spectrum for **Br-DPP**.

$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  8.91 (d,  $J = 3.7$  Hz, 1H, Th), 8.63 (d,  $J = 3.7$  Hz, 1H, Th), 7.64 (d,  $J = 4.9$  Hz, 1H, Th), 7.27 (d,  $J = 3.6$  Hz, 1H, Th), 7.22 (d,  $J = 3.6$  Hz, 1H, Th), 4.05–3.90 (m, 4H,  $\text{NCH}_2$ ), 1.84 (s, 2H, CH), 1.39–1.20 (m, 16H,  $\text{CH}_2$ ), 0.93–0.81 (m, 12H,  $\text{CH}_3$ ).



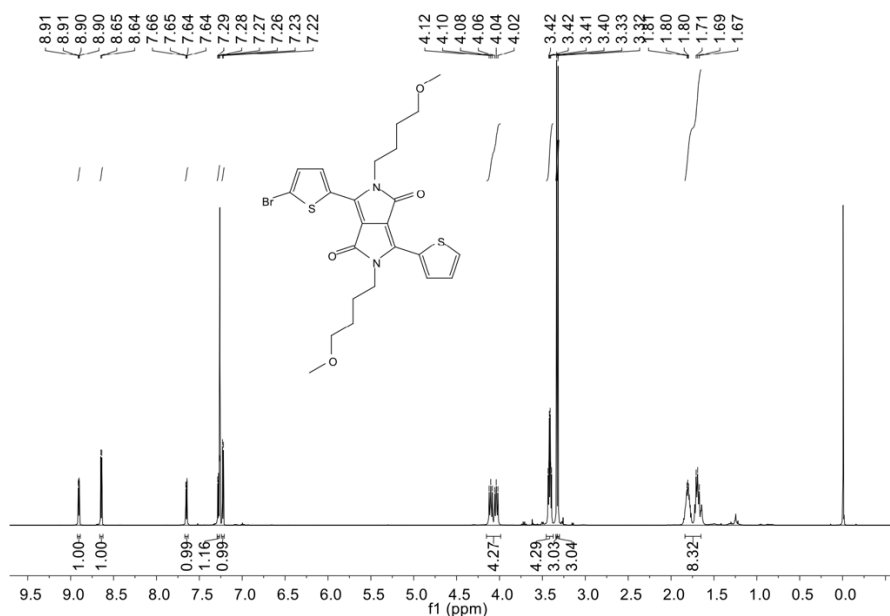
**Figure S27.**  $^1\text{H}$  NMR spectrum for **Br-DPP-CHO**.

$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  10.00 (s, 1H, CHO), 8.93 (d,  $J = 4.1$  Hz, 1H, Th), 8.80 (d,  $J = 4.2$  Hz, 1H, Th), 7.84 (d,  $J = 4.2$  Hz, 1H, Th), 7.25 (d,  $J = 4.7$  Hz, 1H, Th), 4.09–3.88 (m, 4H,  $\text{NCH}_2$ ), 1.82 (m, 2H, CH), 1.42–1.17 (m, 16H,  $\text{CH}_2$ ), 0.87 (m, 12H,  $\text{CH}_3$ ).



**Figure S28.**  $^1\text{H}$  NMR spectrum for Br-DPP-RHA.

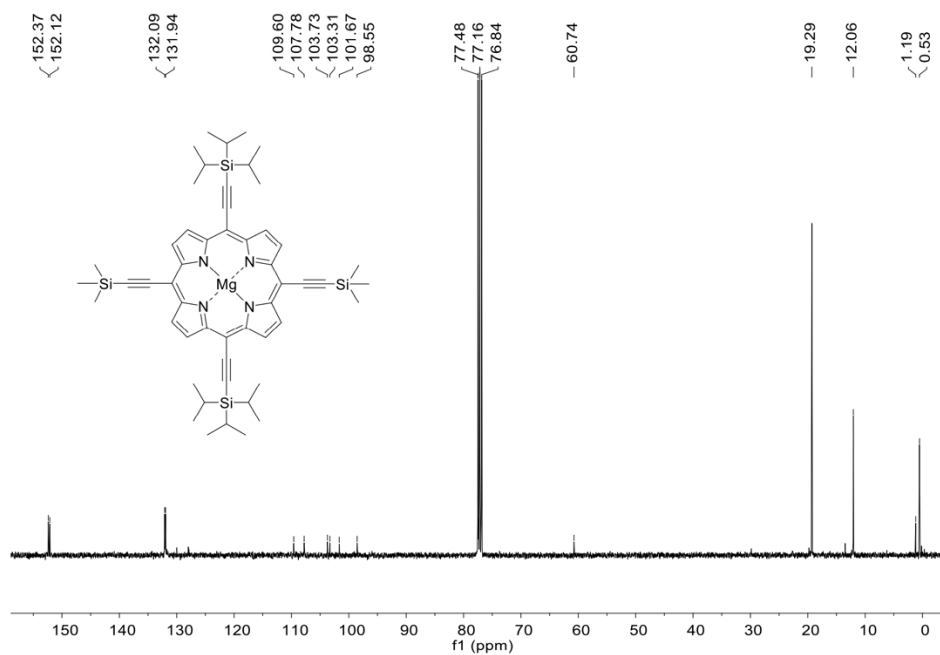
$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  9.08 (d,  $J = 4.3$  Hz, 1H, Th), 8.78 (d,  $J = 4.2$  Hz, 1H, Th), 7.88 (s, 1H, Vinyl), 7.49 (d,  $J = 4.3$  Hz, 1H, Th), 7.24 (d,  $J = 4.2$  Hz, 1H, Th), 4.20 (q,  $J = 7.1$  Hz, 2H,  $\text{CSNCH}_2$ ), 4.11–3.89 (m, 4H,  $\text{NCH}_2$ ), 1.93–1.79 (m, 2H, CH), 1.46–1.20 (m, 19H,  $\text{CH}_2$  and  $\text{CH}_3$ ), 0.97–0.82 (m, 12H,  $\text{CH}_2$ ).



**Figure S29.**  $^1\text{H}$  NMR spectrum for Br-DPP-MOB.

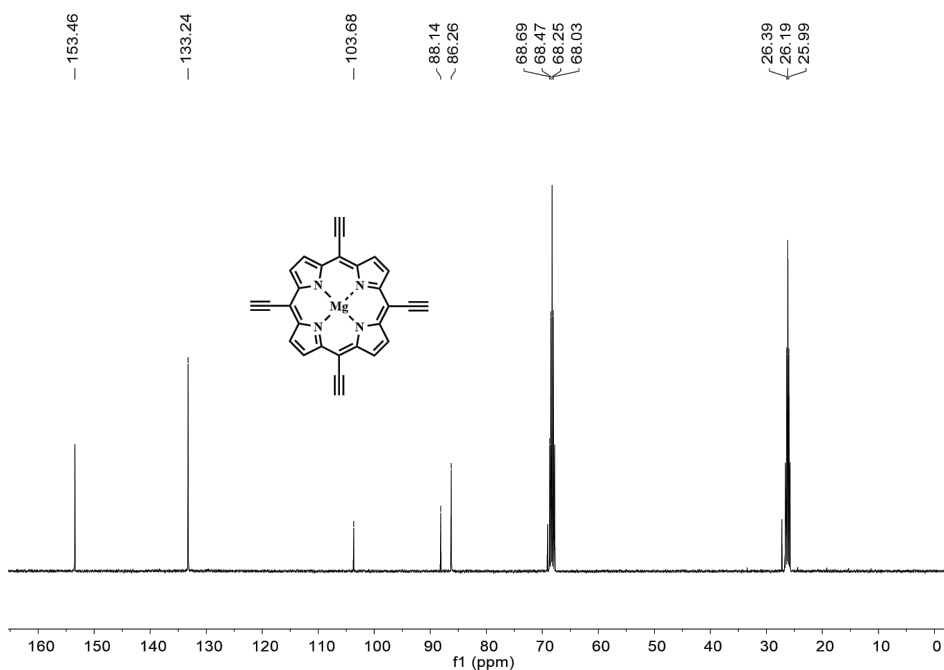
$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  8.90 (dd,  $J = 3.9$  Hz, 1.1 Hz, 1H, Th), 8.64 (d,  $J = 4.2$  Hz, 1H, Th), 7.65 (dd,  $J = 5.0$  Hz, 1.1 Hz, 1H, Th), 7.29–7.27 (m, 1H, Th), 7.22 (d,  $J = 4.2$  Hz, 1H, Th), 4.13–4.01 (m, 4H,  $\text{NCH}_2$ ), 3.41 (m, 4H,  $\text{OCH}_2$ ), 3.33 (s, 3H,  $\text{OCH}_3$ ), 3.32 (s, 3H,  $\text{OCH}_3$ ), 1.86–1.60 (m, 8H,

CH<sub>2</sub>).



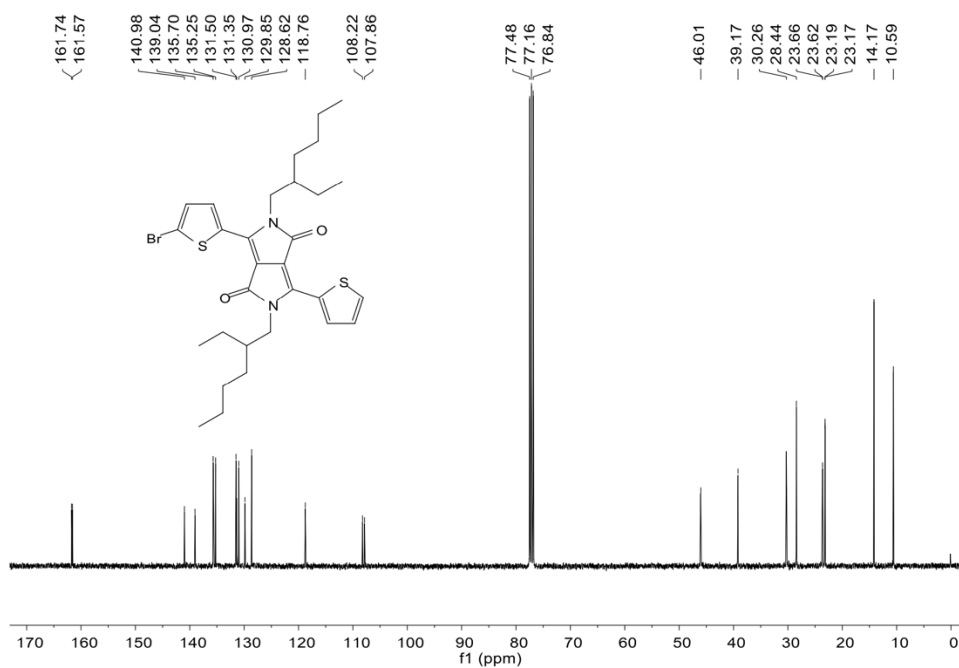
**Figure S30.** <sup>13</sup>C NMR spectrum for **1**.

<sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  152.37, 152.12, 132.09, 131.94, 109.60, 107.78, 103.73, 103.31, 101.67, 98.55, 60.74, 19.29, 12.06, 1.19, 0.53.



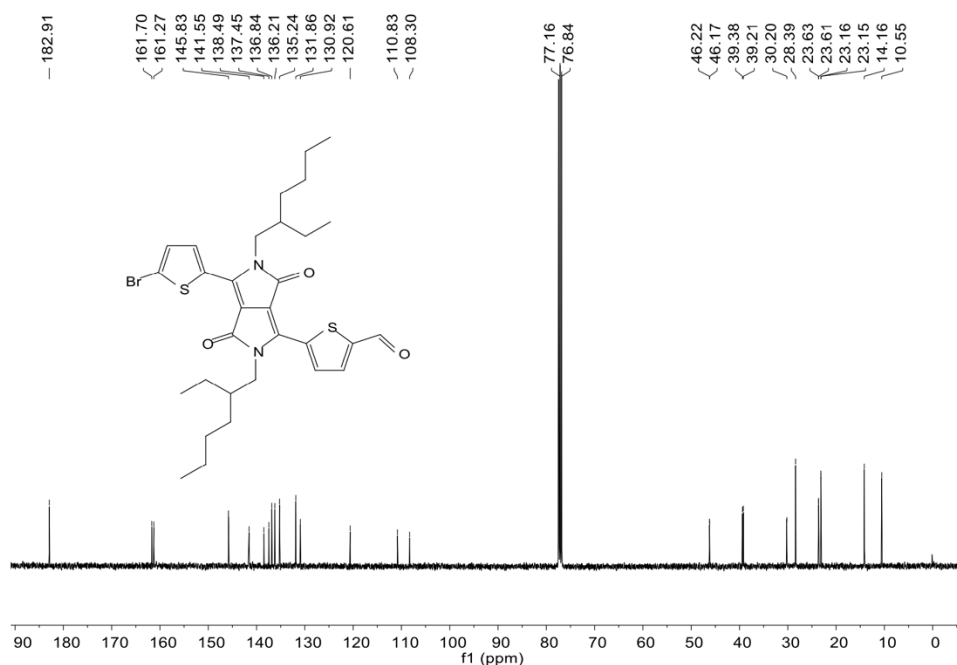
**Figure S31.** <sup>13</sup>C NMR spectrum for **1**.

<sup>13</sup>C NMR (100 MHz, THF-*d*<sub>8</sub>):  $\delta$  153.46, 133.24, 103.68, 88.14, 86.26.



**Figure S32.**  $^{13}\text{C}$  NMR spectrum for **Br-DPP**.

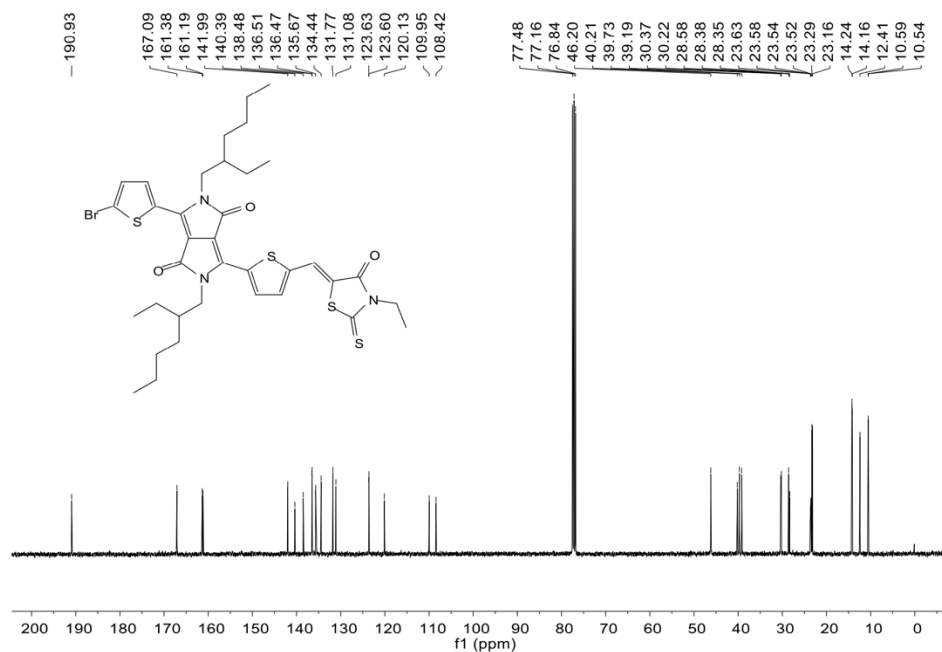
$^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  161.74, 161.57, 140.98, 139.04, 135.70, 135.25, 131.50, 131.35, 130.97, 129.85, 128.61, 118.76, 108.22, 107.86, 46.01, 39.17, 30.26, 28.44, 23.66, 23.62, 23.19, 23.17, 14.17, 10.59.



**Figure S33.**  $^{13}\text{C}$  NMR spectrum for **Br-DPP-CHO**.

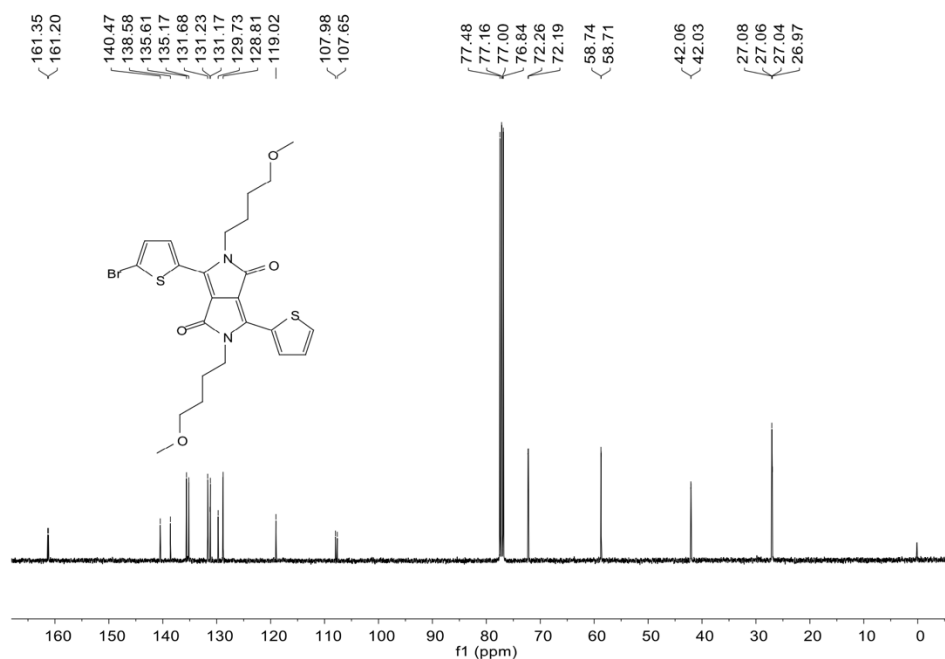
$^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  182.91, 161.70, 161.27, 145.83, 141.55, 138.49, 137.45, 136.84, 136.21, 135.24, 131.86, 130.92, 120.61, 110.83, 108.30, 46.22, 46.17, 39.38, 39.21, 30.20, 28.39, 23.63, 23.61, 23.16, 23.15, 14.16, 10.55.





**Figure S34.**  $^{13}\text{C}$  NMR spectrum for **Br-DPP-RHA**.

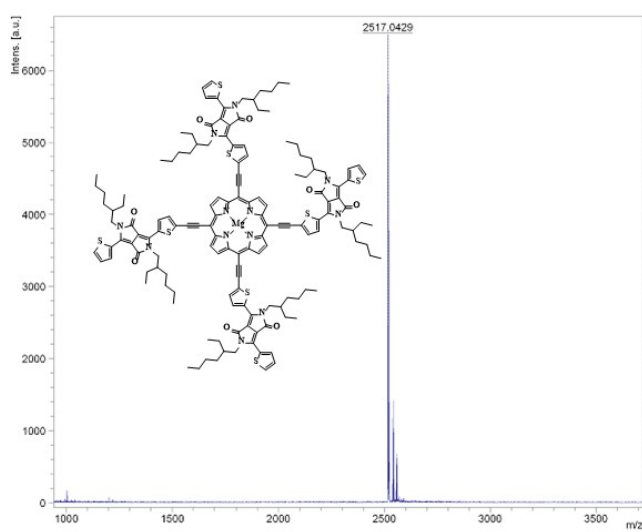
$^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  190.93, 167.09, 161.38, 161.19, 141.99, 140.39, 138.48, 136.51, 136.47, 135.67, 134.44, 131.77, 131.08, 123.63, 123.60, 120.13, 109.95, 108.42, 46.20, 40.21, 39.73, 39.19, 30.37, 30.22, 28.58, 28.38, 28.35, 23.63, 23.58, 23.54, 23.52, 23.29, 23.16, 14.24, 14.16, 12.41, 10.59, 10.54.



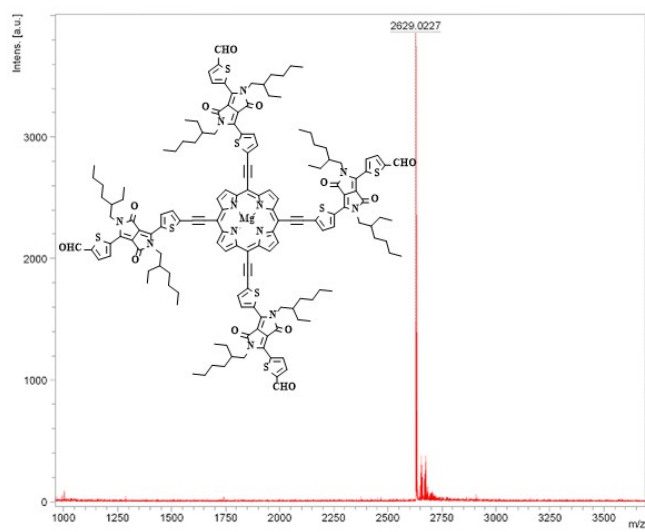
**Figure S35.**  $^{13}\text{C}$  NMR spectrum for **Br-DPP-MOB**.

$^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  161.35, 161.20, 140.47, 138.58, 135.61, 135.17, 131.68, 131.23, 131.17, 129.73, 128.81, 119.02, 107.98, 107.65, 72.26, 72.19, 58.74, 58.71, 42.06, 42.03, 27.08, 27.06, 27.04, 26.97.

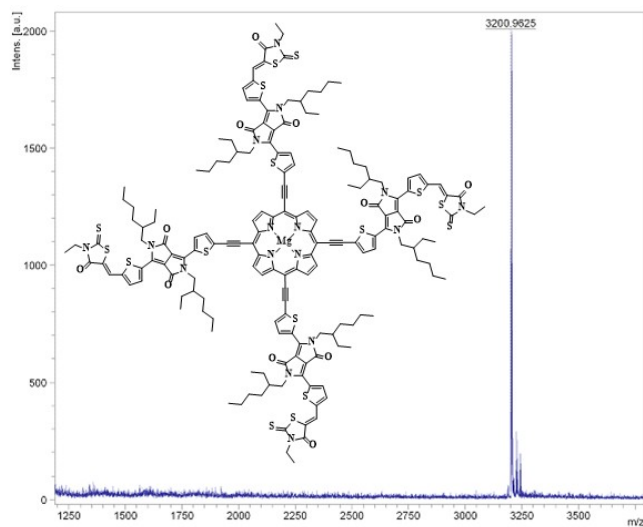
## 8. MALDI-TOF HRMS Data



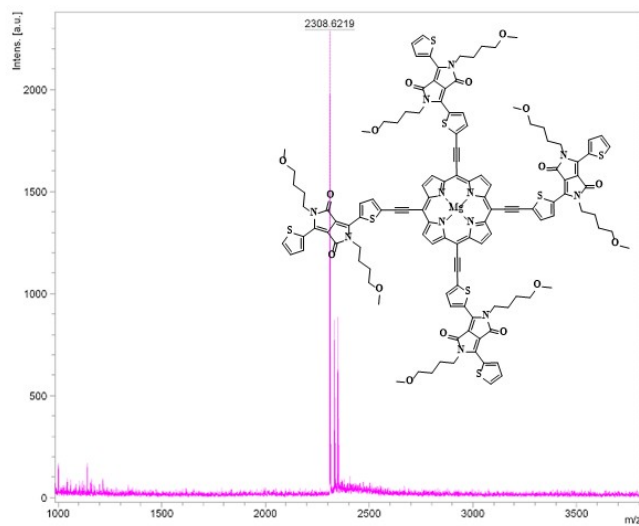
**Figure S36.** HRMS spectrum for **3a**.



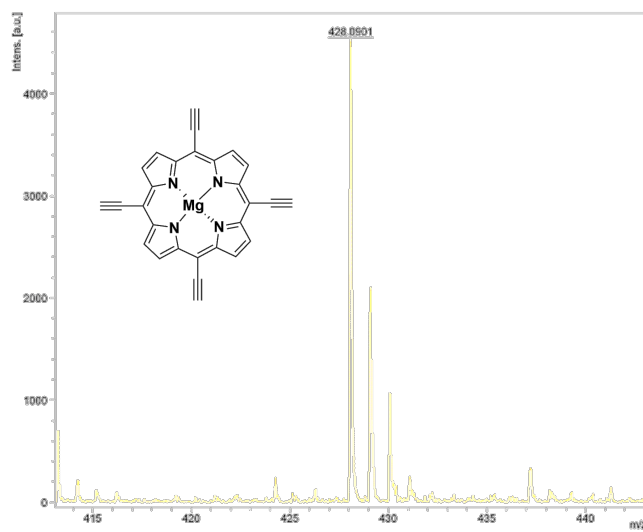
**Figure S37.** HRMS spectrum for **3b**.



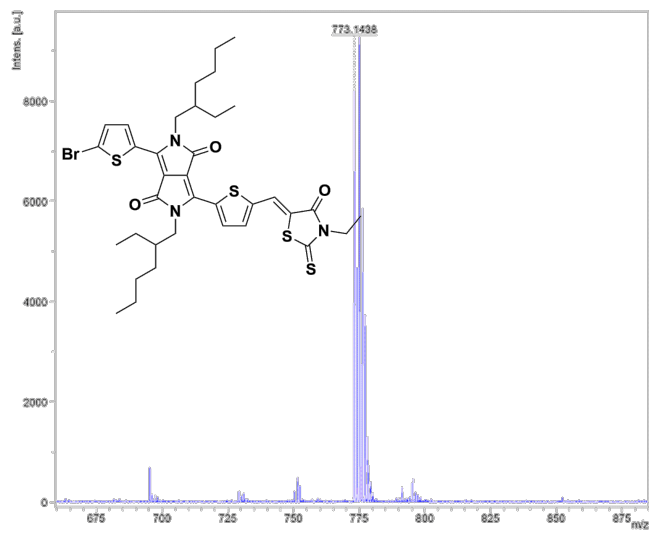
**Figure S38.** HRMS spectrum for **3c**.



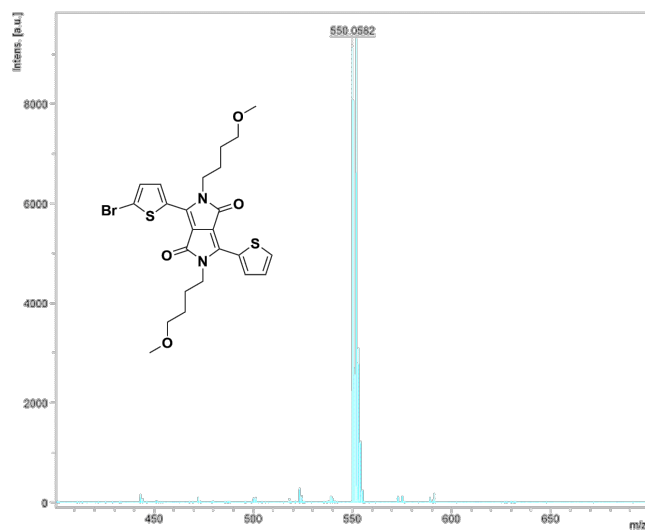
**Figure S39.** HRMS spectrum for **3d**.



**Figure S40.** HRMS spectrum for **2**.

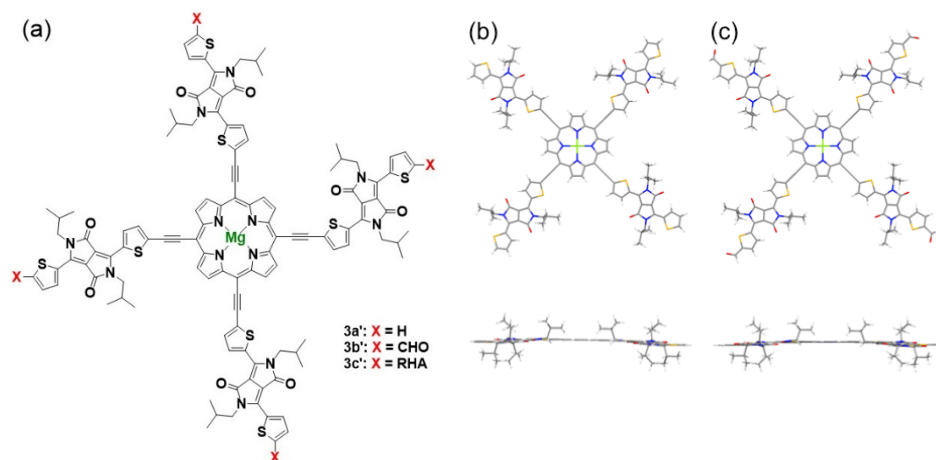


**Figure S41.** HRMS spectrum for **Br-DPP-RHA**.



**Figure S42.** HRMS spectrum for **Br-DPP-MOB**.

## 9. Computational Calculation



**Figure S43.** Calculated structures for **3a'–c'** (a) and optimized structures for **3a'** (b) and **3b'** (c) as top and side view (for **3c'**, see Figure 8). Calculations were carried out by using Gaussian09<sup>9</sup> package at the B3LYP/6-31G(d) level.

**Table S7.** Calculated energies and the numbers of imaginary frequency for the optimized structures.

	Number of imaginary frequencies	free energy <sup>[a]</sup> / Hartree	HOMO/ Hartree	LUMO/ Hartree
<b>3a'</b>	0	-9123.294193	-0.17242	-0.11533
<b>3b'</b>	0	-9576.573404	-0.18481	-0.12892
<b>3c'</b>	0	-13755.062886	-0.18335	-0.13028

<sup>[a]</sup> in standard conditions.

**Table S8.** Cartesian coordinates for the optimized structures.

**3a'**

Symbol	X	Y	Z
C	0.023381	-0.001906	-0.022580
C	0.023091	-0.000785	1.342415
C	1.400569	-0.002224	1.760065
N	2.221664	-0.004499	0.660357
C	1.400923	-0.003860	-0.439661
C	4.974621	-0.004940	-3.607263
C	3.609624	-0.005215	-3.607552
C	3.191979	-0.005676	-2.230068
N	4.291684	-0.006046	-1.408971
C	5.391697	-0.005126	-2.229715
C	1.839287	-0.004119	-1.792213
C	8.559294	-0.001185	1.343985
C	8.559591	-0.002235	-0.021011
C	7.182108	-0.004093	-0.438662
N	6.361009	-0.004538	0.661045
C	7.181750	-0.002290	1.761059
C	6.744242	-0.003752	-1.791361
C	3.608055	0.001904	4.928664
C	4.973050	0.001527	4.928962
C	5.390705	-0.001266	3.551482
N	4.290998	-0.002963	2.730386
C	3.190980	-0.000594	3.551120

C	1.838440	0.000161	3.112763
C	6.743405	-0.000429	3.113602
C	7.745909	0.000893	-2.792093
C	0.836759	0.006574	4.113471
C	8.603554	0.007453	-3.662485
C	-0.021076	0.014689	4.983661
C	9.594842	0.019829	-4.647371
C	-1.013084	0.028040	5.967811
C	0.838556	0.000197	-2.793881
Mg	4.291339	-0.004633	0.660708
C	7.744158	0.005499	4.115241
C	8.614631	0.013114	4.972794
C	-0.031707	0.006386	-3.651661
C	9.599429	0.026059	5.964163
C	-7.184700	0.494757	-13.96530
C	-5.823628	0.453334	-14.11992
C	-5.127166	0.430314	-12.88789
C	-5.959152	0.459137	-11.77509
S	-7.642862	0.513135	-12.30024
C	-5.486591	0.410086	-10.41126
C	-4.153257	0.257655	-10.01143
C	-4.083972	0.213432	-8.594920
C	-5.426795	0.340639	-8.076360
N	-6.248447	0.470770	-9.243617
C	-2.808597	0.136258	-10.53127
N	-1.987213	0.010871	-9.361614
C	-2.751452	0.070075	-8.194934
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### 3b'

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