

Imidazolium iodide salt as a bifunctional co-adsorbent for quasi-solid-state dye-sensitized solar cells: improvements of electron lifetime and charge collection efficiency

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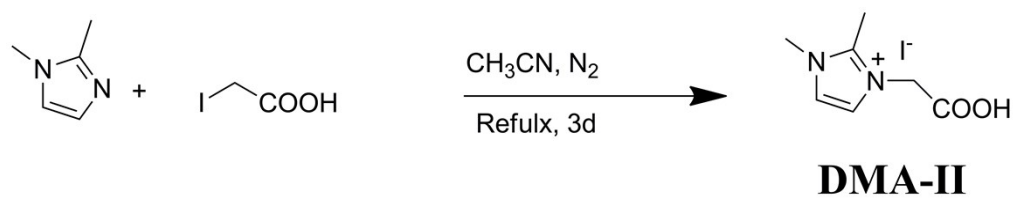


Figure S1 The synthetic route of **DMA-II**.

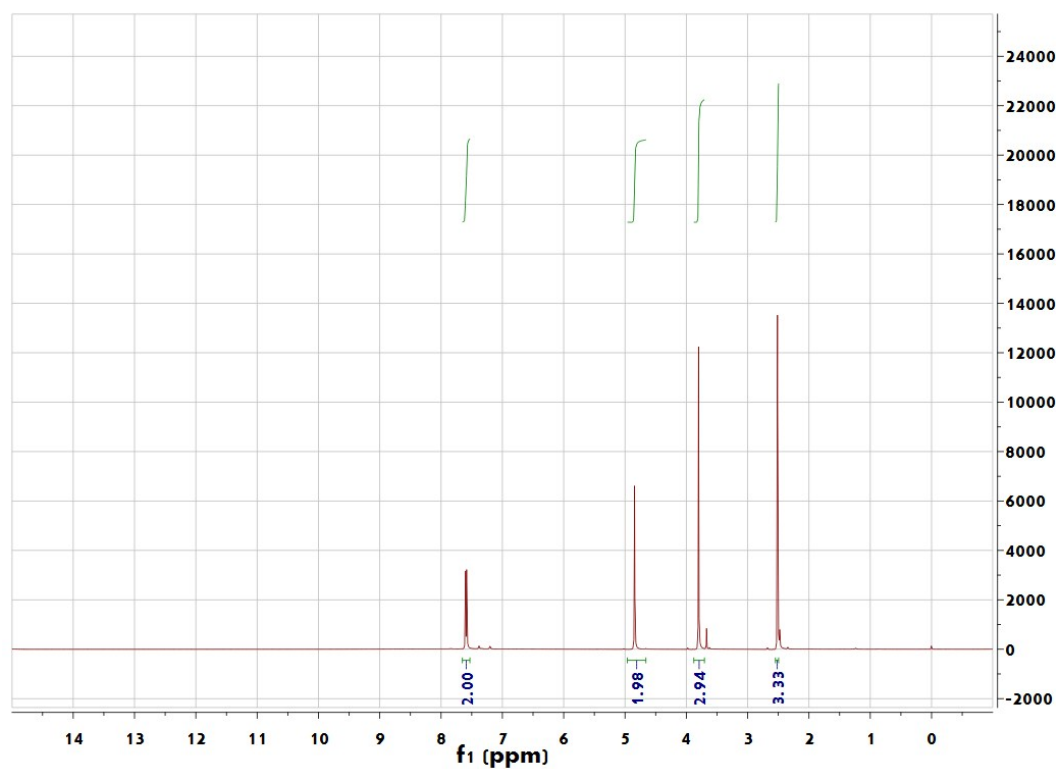


Figure S2. ¹H NMR of DMA-II.

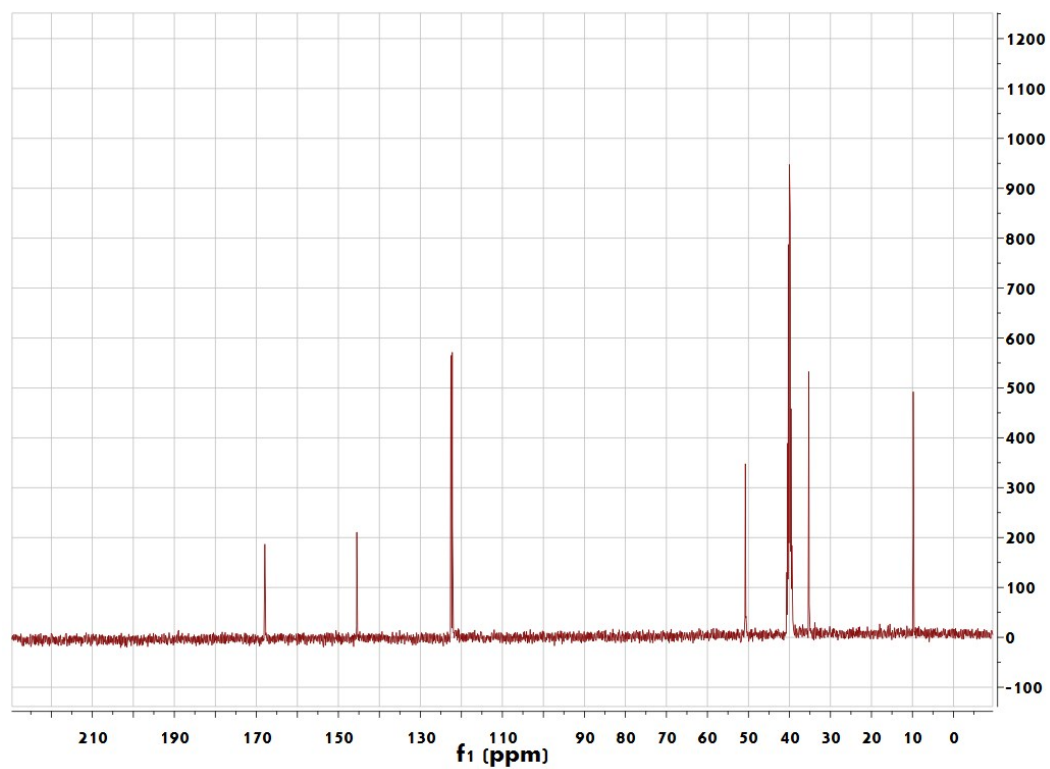


Figure S3. ^{13}C NMR of DMA-II.

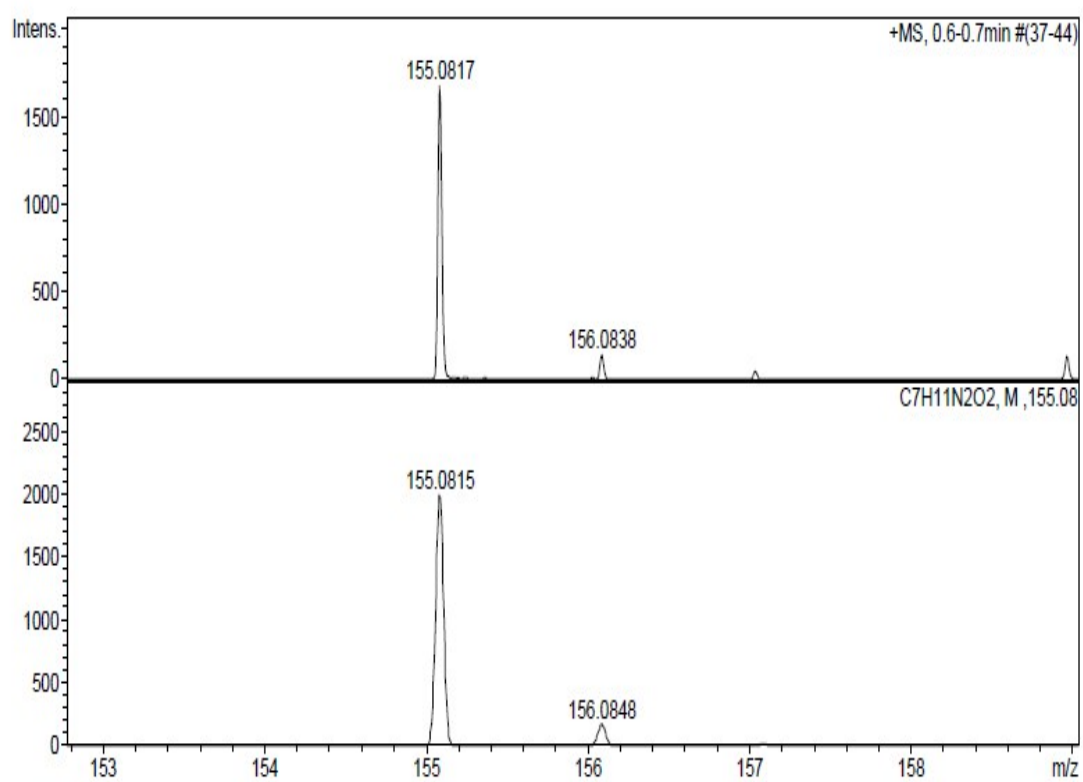


Figure S4. HRMS of DMA-II.

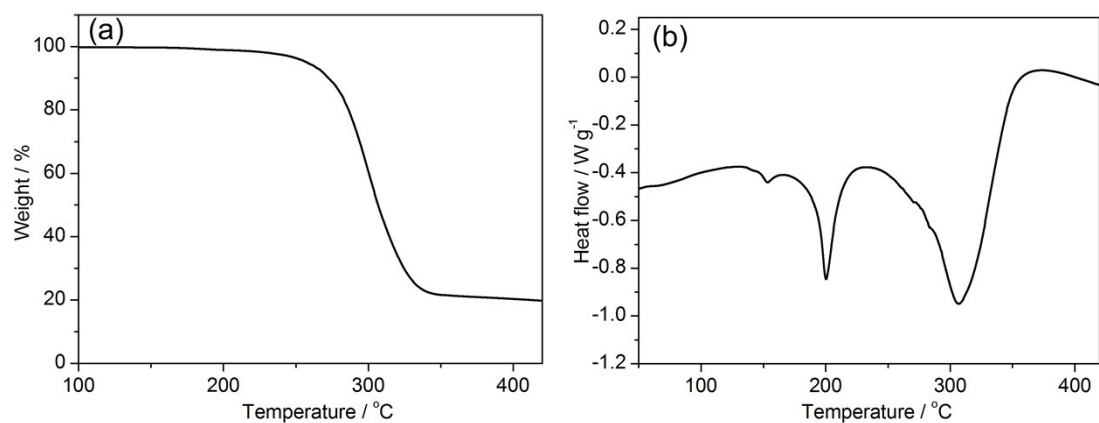


Fig. S5 (a) TG and (b) DSC curves of DMA-II.

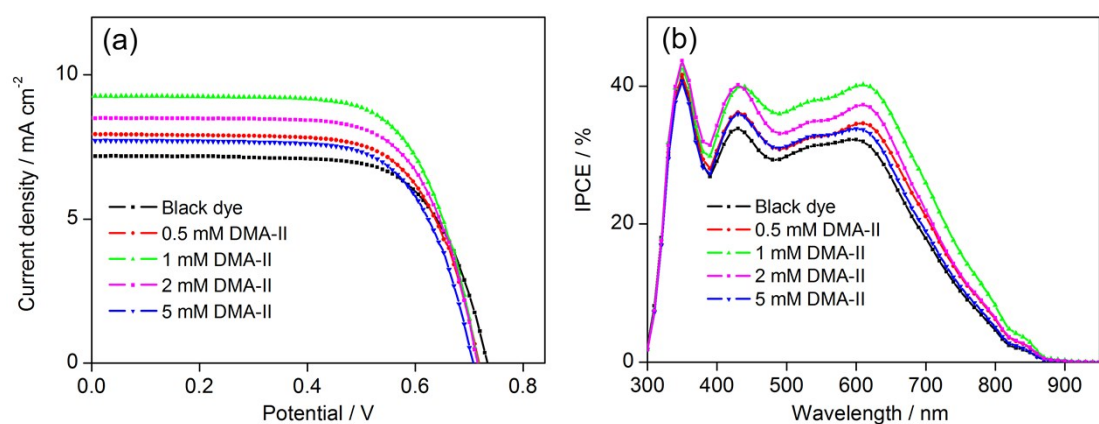


Fig. S6 (a) Current-potential curves and (b) IPCE action spectra of the DSSCs with a film thickness of 5 μm .

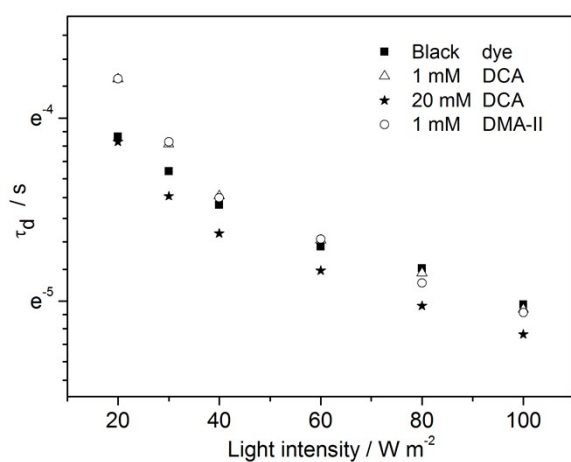


Fig. S7 Electron transport time as a function of light intensity.

Table S1. Photovoltaic performance of DMA-II under different concentrations with a film thickness of 5 μm .

Coadsorption	V_{oc} (mV)	J_{sc} (mA cm^{-2})	FF	PCE (%)
No	735	7.18	0.694	3.66
0.5 mM DMA-II	719	7.93	0.686	3.92
1 mM DMA-II	716	9.26	0.689	4.57
2 mM DMA-II	715	8.49	0.694	4.22
5 mM DMA-II	707	7.72	0.688	3.76

Table S2. Contributions to V_{oc} gain vs. the experimental change of V_{oc}

Coadsorption	ΔE_{CB} (mV) ^a	Q ($\mu\text{C cm}^{-2}$) ^b	$m_{\text{c}}\ln(Q_2/Q_1)$ (mV) ^c	$\Delta E_{\text{CB}}+m_{\text{c}}\ln(Q_2/Q_1)$ (mV)	ΔV_{oc} (mV) ^d
No	0	125.08	0	0	0
1 mM DCA	-65	184.24	48	-17	-21
20 mM DCA	-86	170.84	39	-47	-39
1 mM DMA-II	-102	235.32	79	-23	-18

^a $\Delta E_{\text{CB}} = E_{\text{CB}}(\text{adsorbent-dye}) - E_{\text{CB}}(\text{black dye})$ is obtained from Fig. 4(a). The negative value indicates a positive shift of CB. ^b The charge density at open circuit is obtained at 100 W m^{-2} LED light (532 nm). ^c m_{c} is 125 mV obtained from Fig. 4(a). ^d $\Delta V_{\text{oc}} = V_{\text{oc}}(\text{adsorbent-dye}) - V_{\text{oc}}(\text{black dye})$. The V_{oc} for the DSSCs is measured under AM 1.5G illumination sun-light (100 W m^{-2}).