

Supporting Information

Polarity Engineering of Porous Aromatic Frameworks for Specific Water Contaminants Capture

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Materials Characterizations

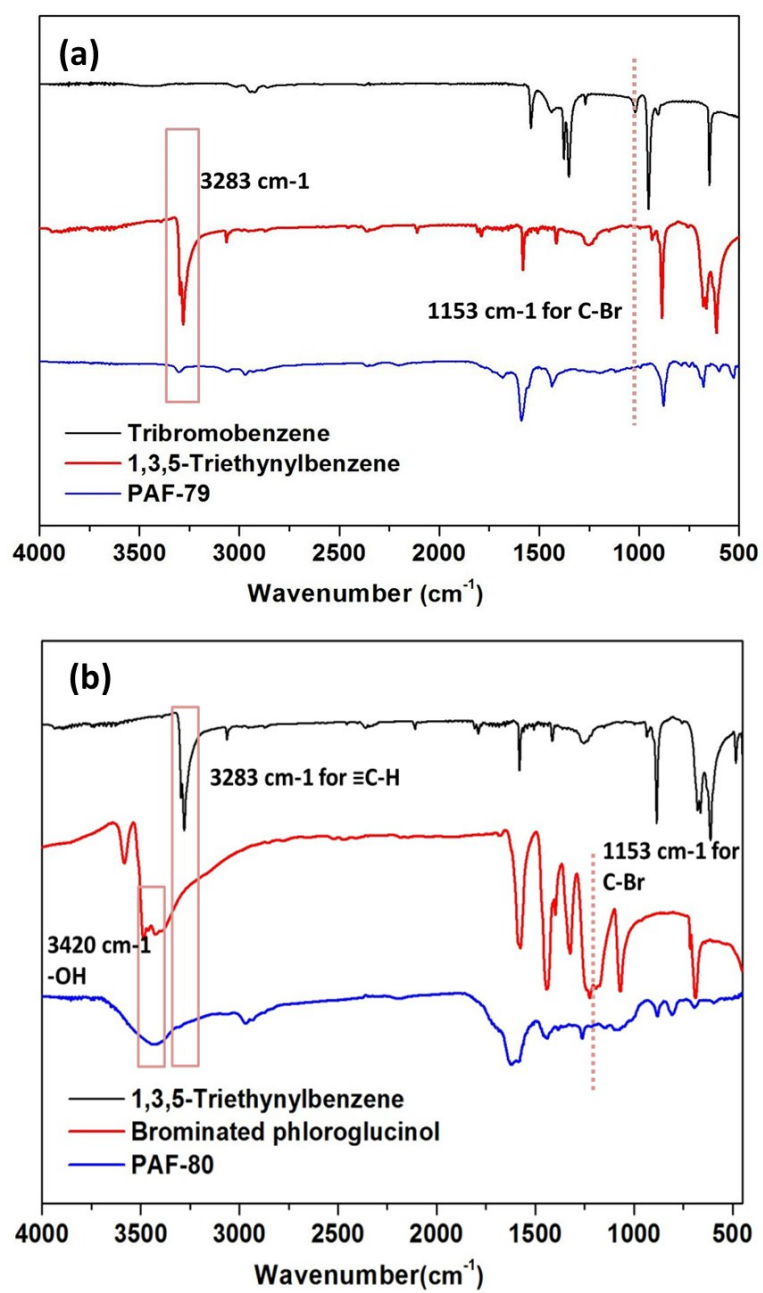


Fig. S1 FT-IR spectra of (a) PAF-79 and (b) PAF-80 and their respective starting materials ^{1,2}

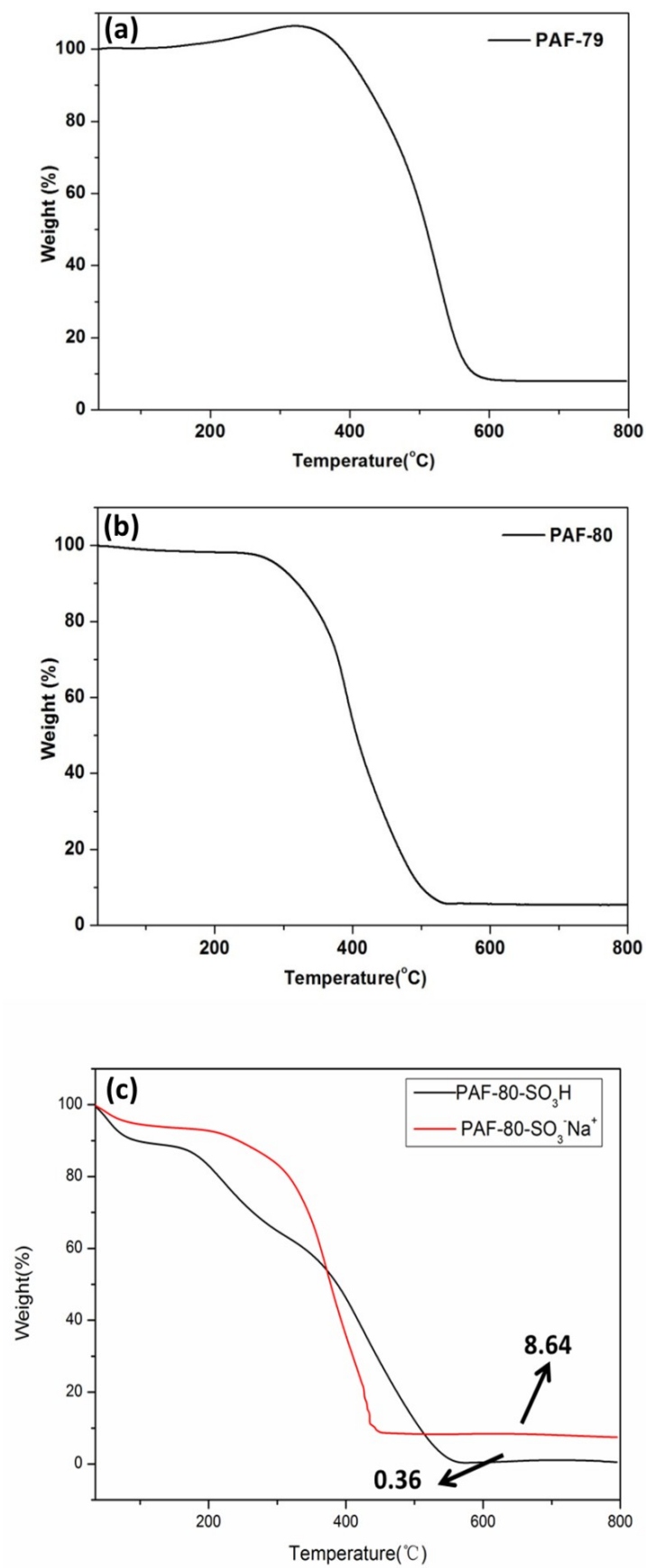


Fig. S2 TGA curves of (a) PAF-79, (b) PAF-80 and (c) PAF-80-SO₃Na and PAF-80-SO₃H

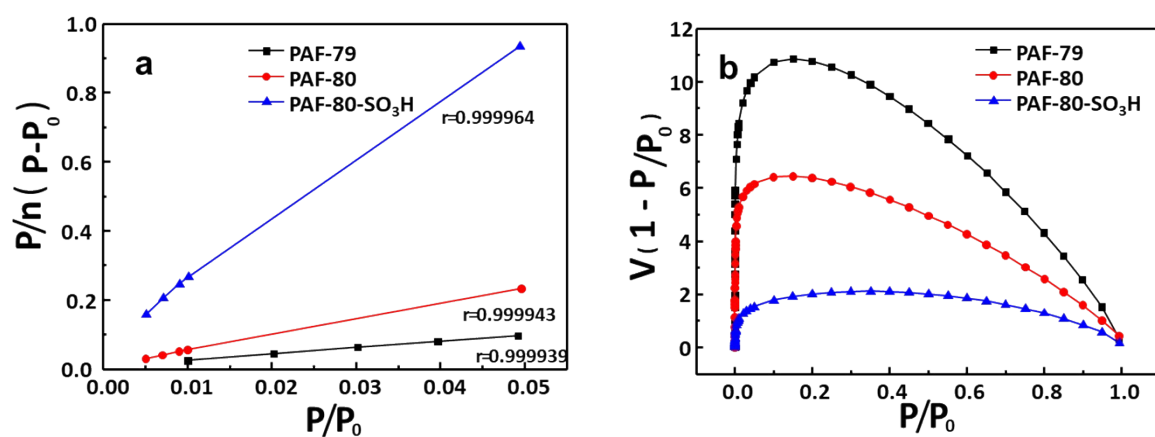


Fig. S3 Represents (a) BET plots and (b) Rouquerol plots of PAF materials and leads to selection of an appropriate pressure range (0.01-0.05, 0.01-0.08 and 0.01-0.1 P/P_0 for PAF-79, PAF-80 and PAF-80-SO₃H, respectively).

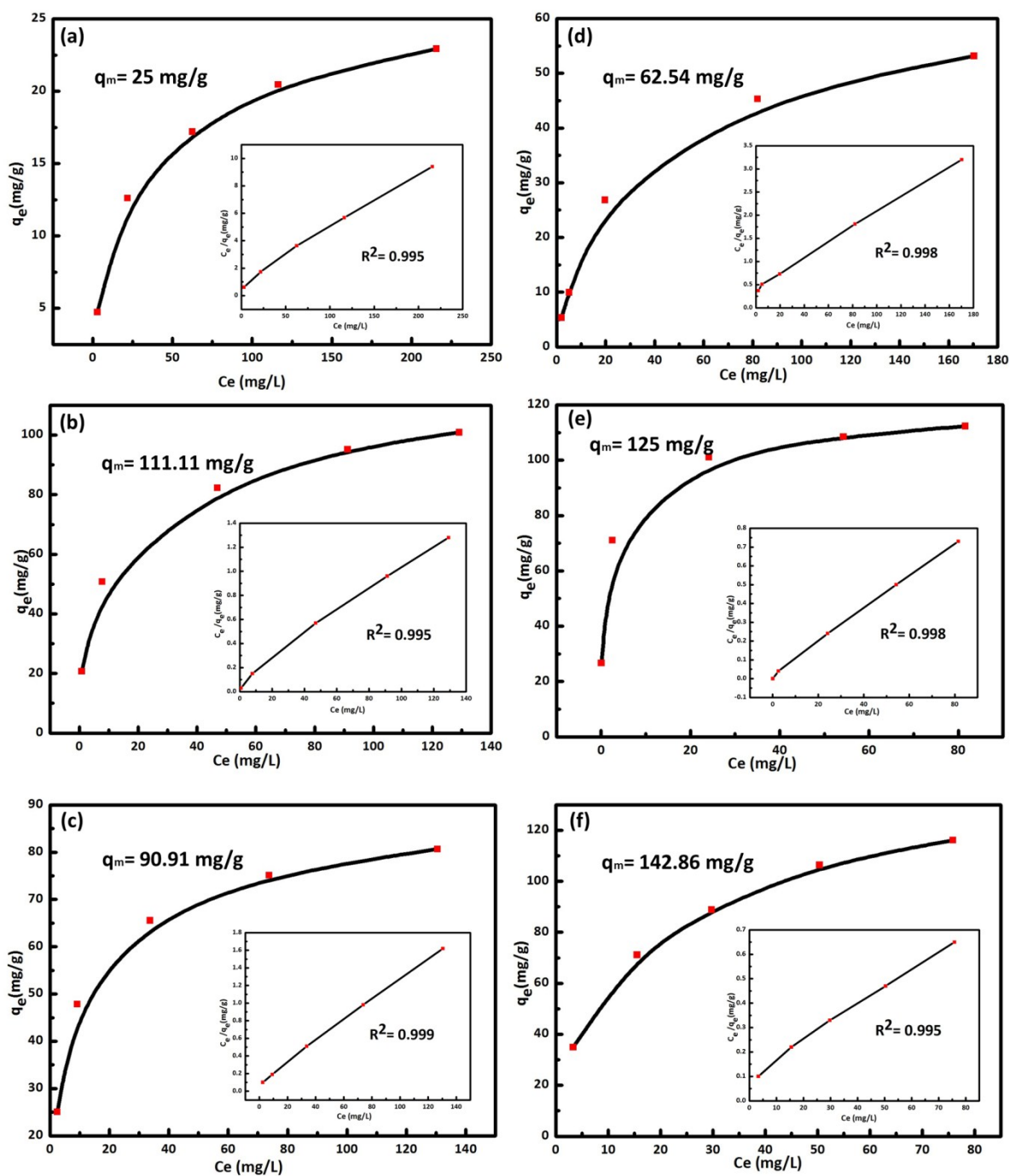


Fig. S4 Antibiotic adsorption isotherms of (a) PAF-79, (b) PAF-80, (c) PAF-80-SO₃H for tetracycline (TC) and (d) PAF-79, (e) PAF-80, (f) PAF-80-SO₃H for doxycycline hydrochloride (DOX).

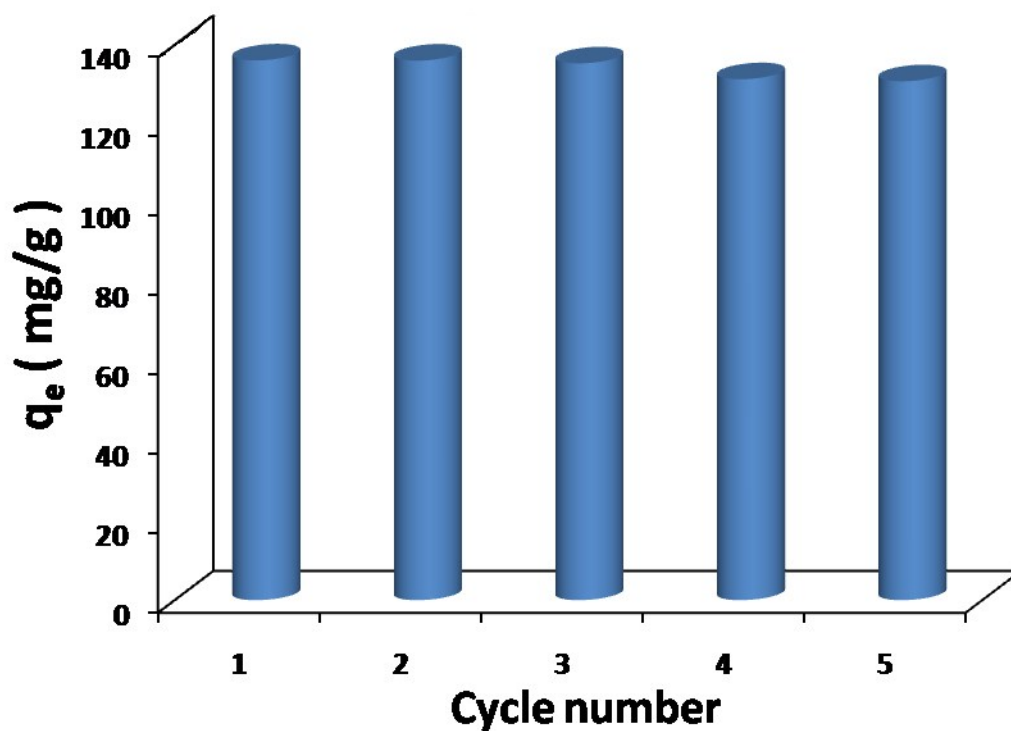


Fig. S5 Recyclability of PAF-80-SO₃H for doxycycline hydrochloride (DOX) adsorption by shaking DOX/PAF-80-SO₃H for 12h in aqueous solution to displace adsorbed species.

References

- (1) D. Tan, W. Fan, W. Xiaong, H. Sun, Y. Cheng, X. Liu, C. Meng, A. Li and W. Q. Deng, *Macromol. Chem. Phys.* 2012, 213, 1435–1440.
- (2) Z. Yan, H. Ren, H. Ma, R. Yuan, Y. Yuan, X. Zou, F. Sun and G. Zhu, *Microporous and Mesoporous Materials*, 2013, 173, 92–98.