

**Aggregation-Induced Enhanced Photoluminescence in Magnetic Graphene Oxide
Quantum dots as Fluorescence Probe for Arsenic (III) Sensing**

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Results

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -11.1	Peak 1: -11.1	100.0	6.88
Zeta Deviation (mV): 6.88	Peak 2: 0.00	0.0	0.00
Conductivity (mS/cm): 0.0210	Peak 3: 0.00	0.0	0.00
Result quality : Good			

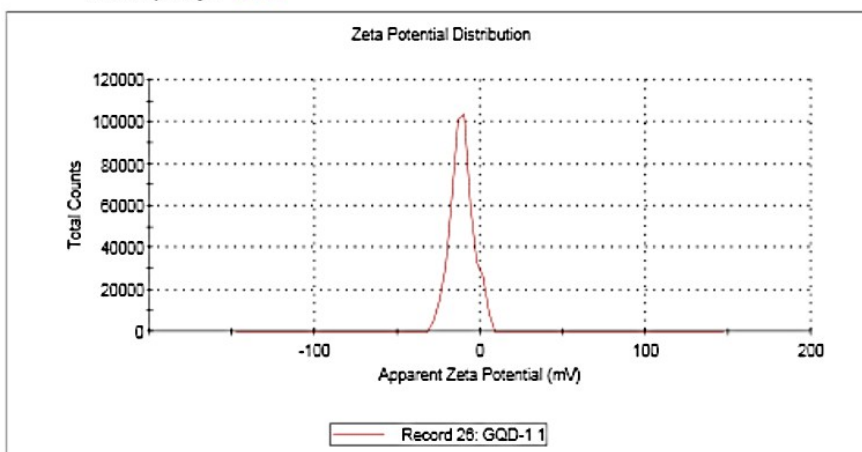


Figure S1 Zeta potential graph of GQDs

Results

	Mean (mV)	Area (%)	St Dev (mV)
Zeta Potential (mV): -21.6	Peak 1: -21.6	100.0	7.09
Zeta Deviation (mV): 7.09	Peak 2: 0.00	0.0	0.00
Conductivity (mS/cm): 0.00890	Peak 3: 0.00	0.0	0.00
Result quality : Good			

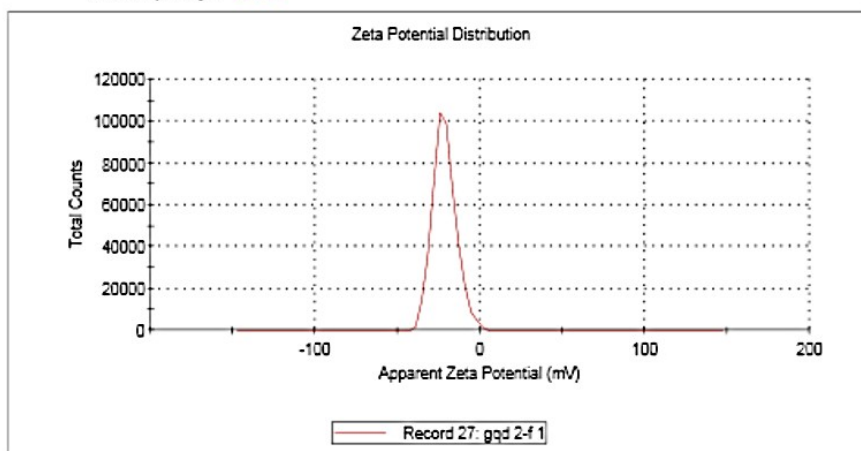


Figure S2 Zeta potential graph of Fe-GQDs

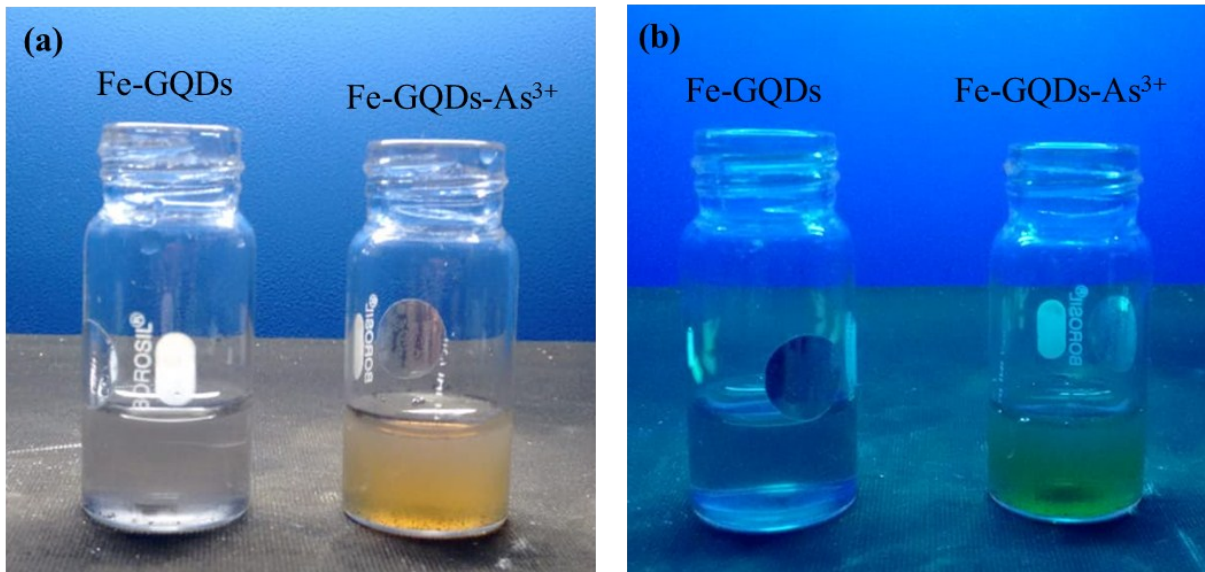


Figure S3. Digital photographs of Fe-GQDs and Fe-GQDs-As³⁺ (a) in white light (b) UV light

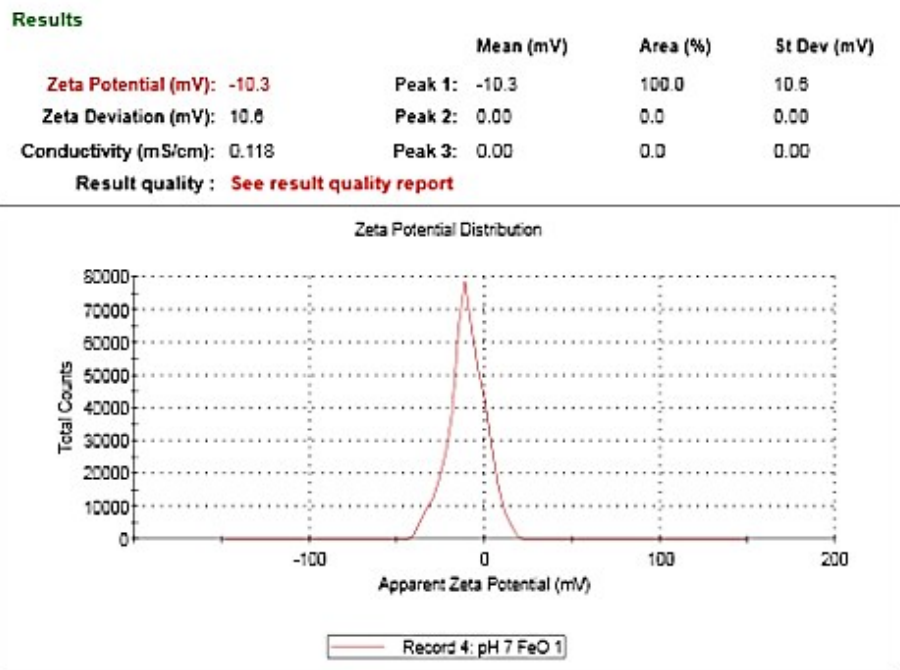


Figure S4 Zeta potential graph of Fe-GQDs at pH 7 after arsenic ion addition

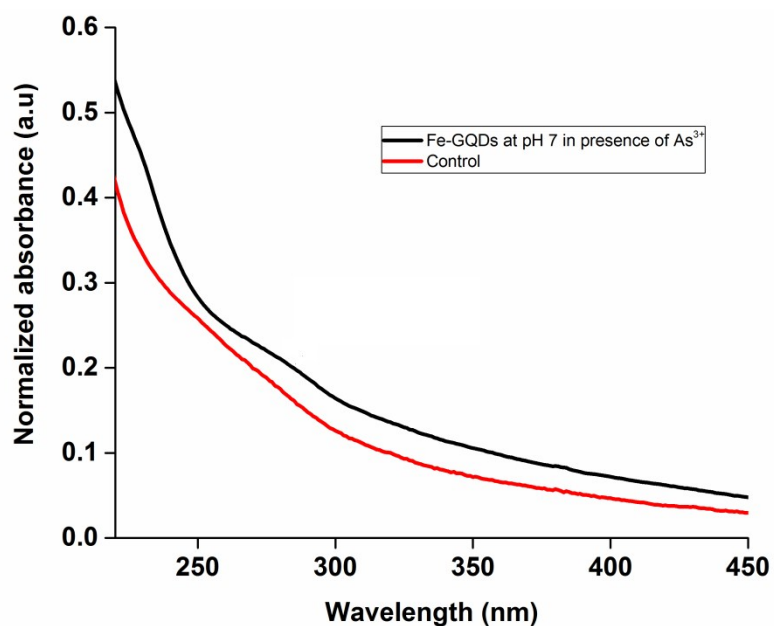


Figure S5 UV-visible absorption spectra of Fe-GQDs before and after arsenic addition

Results

	Size (d.nm):	% Intensity:	St Dev (d.n...)
Z-Average (d.nm): 1784	Peak 1: 636.8	100.0	62.80
Pdi: 0.935	Peak 2: 0.000	0.0	0.000
Intercept: 0.932	Peak 3: 0.000	0.0	0.000
Result quality :	Refer to quality report		

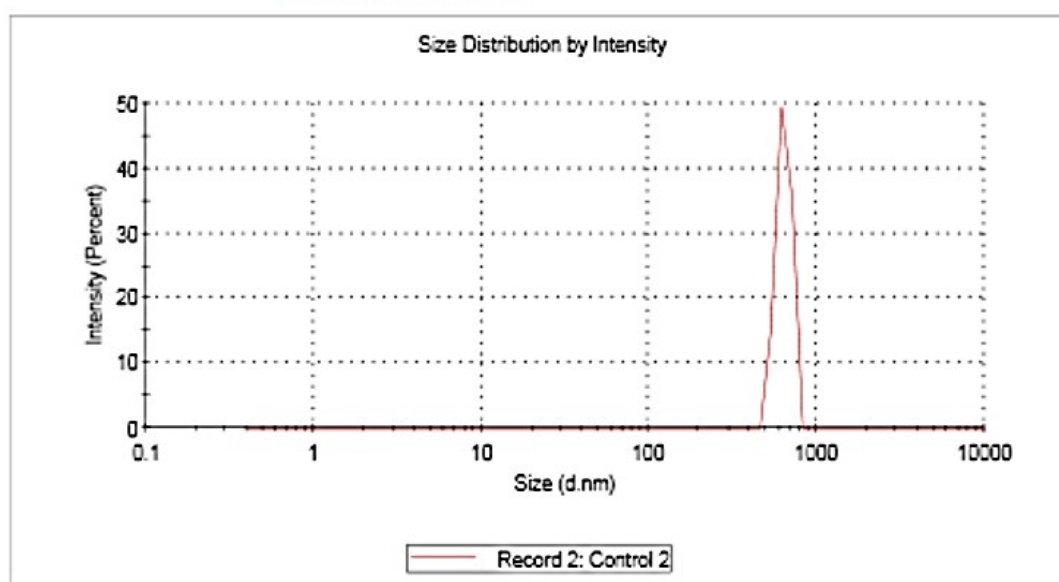


Figure S6 DLS graph of Fe-GQDs without As^{3+} ions

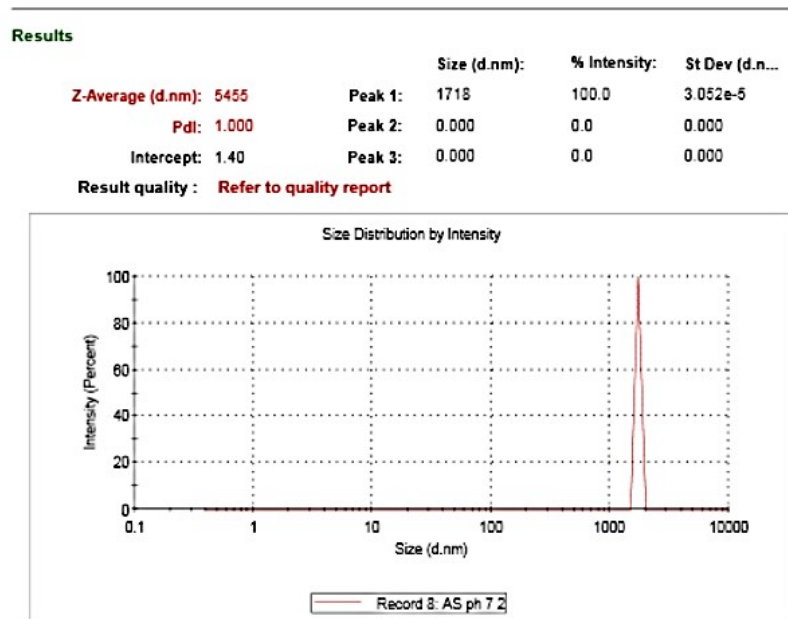


Figure S7 DLS graph of Fe-GQDs with As^{3+} ions at pH 7

Table S2 Photophysical parameters of Fe-GQDs before and after arsenic addition

Sample	τ_f (avg. ns)	χ^2
Fe-GQDs	4.0	1.35
Fe-GQDs- As^{3+}	6.02	1.27

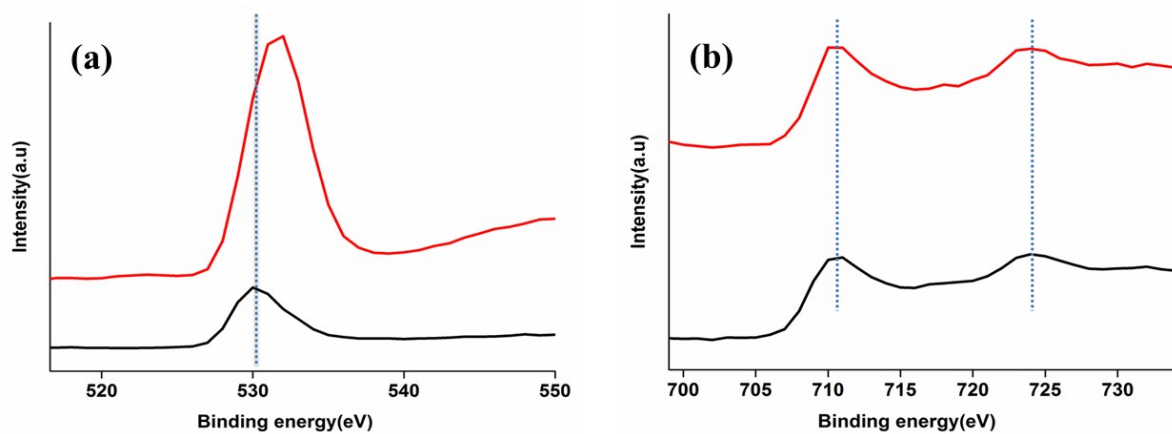


Figure S8 Enlarged view of O1s and Fe2p of Fe-GQDs after arsenic interaction

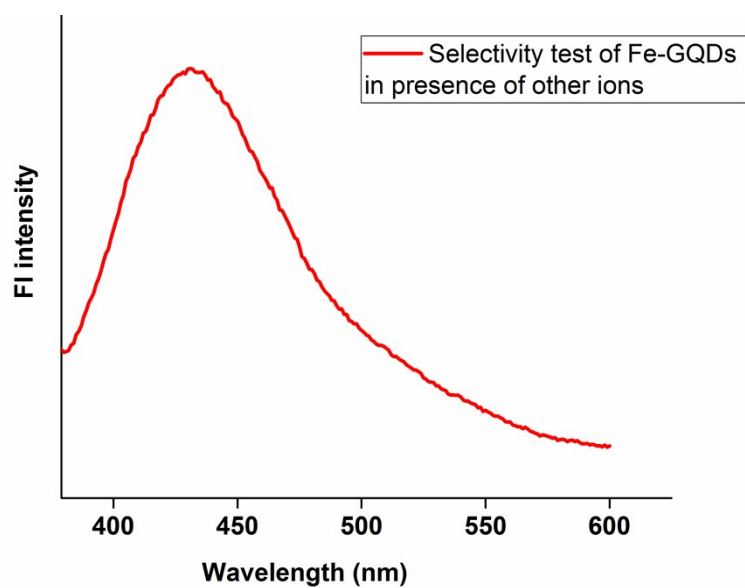


Figure S9 Interfering studies of Fe-GQDS in a mixture of other ions along with As^{3+} ions

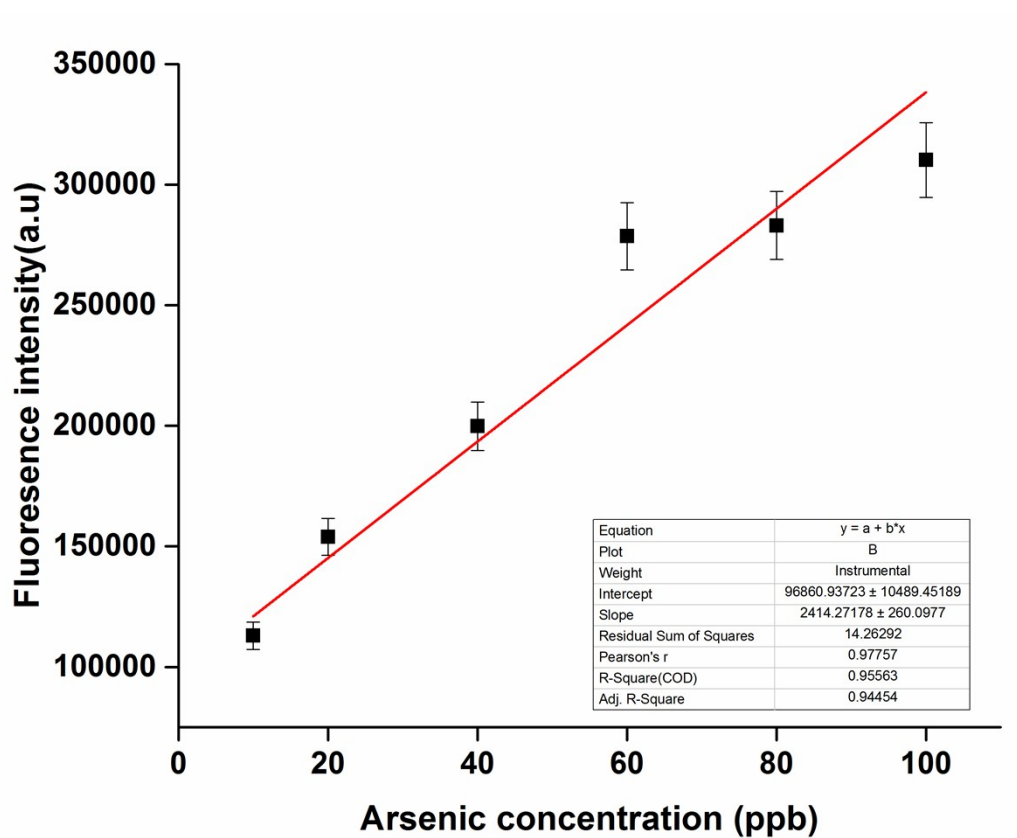


Figure S10 Linear graph showing fluorescence efficiency as a function of different concentrations of As^{3+} in Tap water

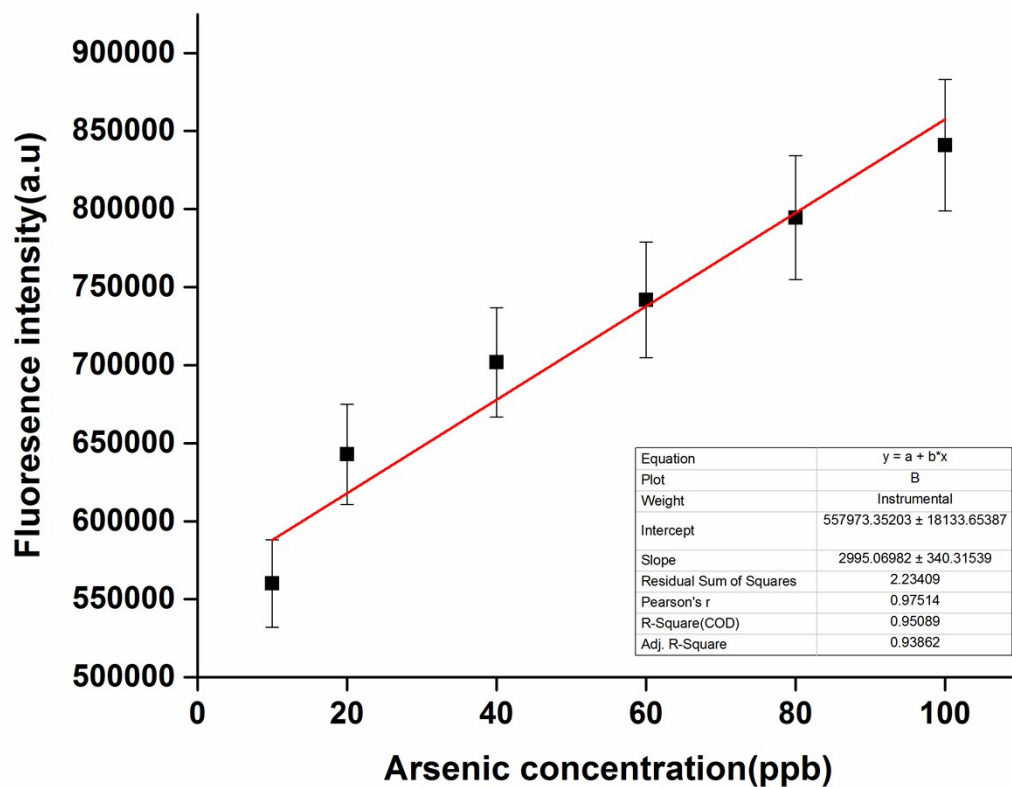


Figure S11. Linear graph showing fluorescence efficiency as a function of different concentrations of As^{3+} in recycled water

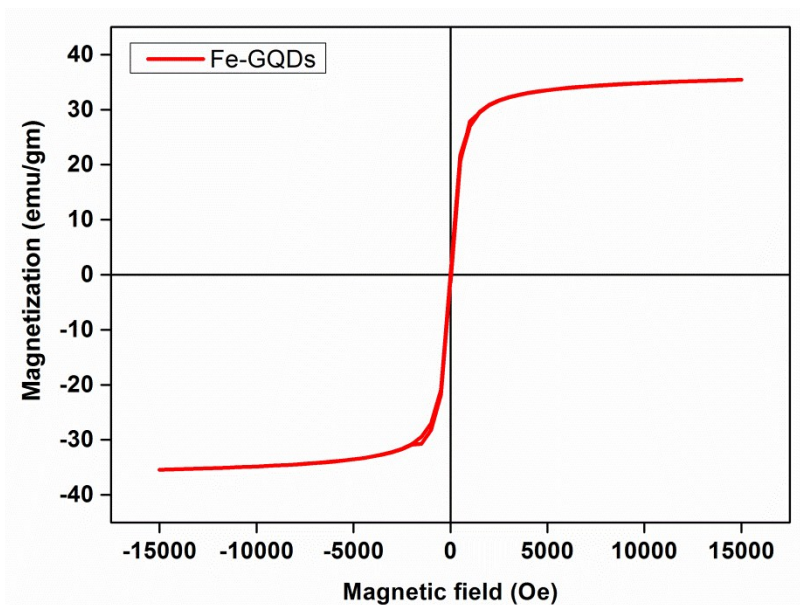


Figure S12 Magnetization curve of Fe-GQDs

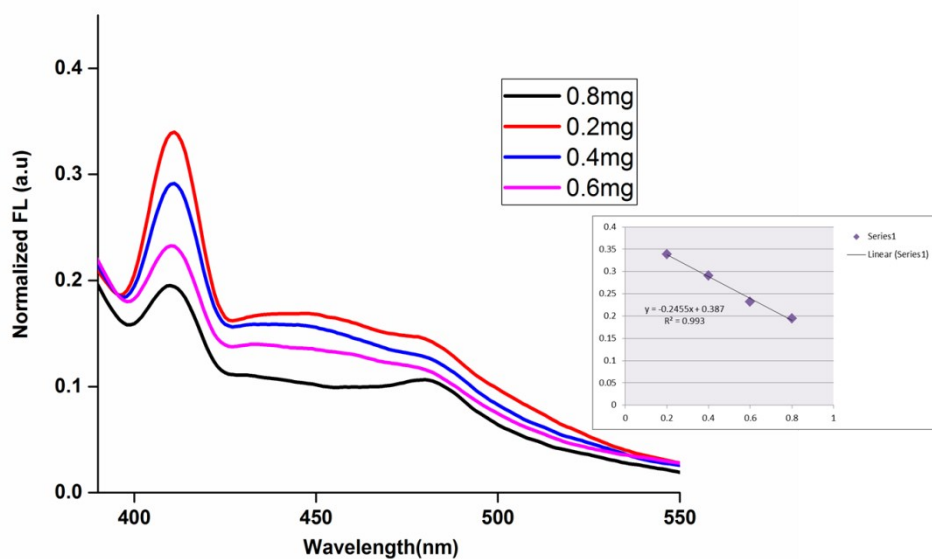


Figure S13 Concentration dependent FL emission spectra

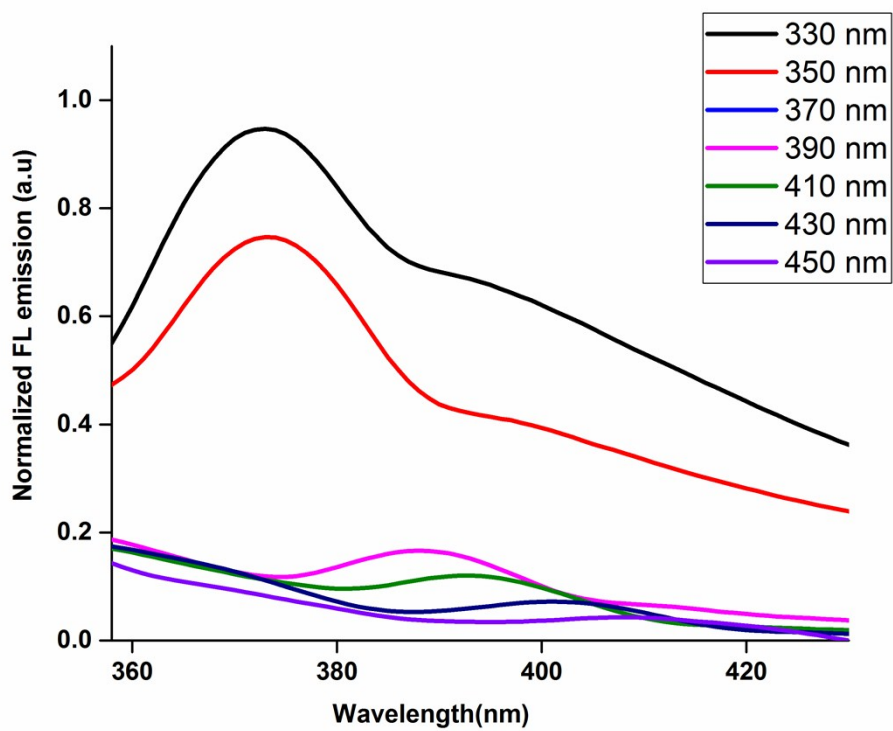


Figure S14 FL emission at different wavelength

Antibacterial study of Fe-GQDs

Reactive Oxygen Species (ROS) Quantification in E.coli cell

When bacterial cells are subjected to various environmental or chemical environments, it induces stresses on them. These stresses are in form of intracellular reactive oxygen species. A strong co-relation exists between these stresses and the bacterial reduction.¹⁻³ For this, the Fe-GQDs were incubated with 100 μ L of E.coli culture for 2 h and DCFH-DA (dichlorodihydrofluorescein diacetate) dye was used to evaluate the generation of ROS. As evident from Figure S15a and S15b, no ROS was observed in control, however, significant green fluorescence is observed in Fe-GQDs treated bacterial cells. These ROS species interacts and disrupt the cell integrity of the bacterial cell and finally led to the cell death.

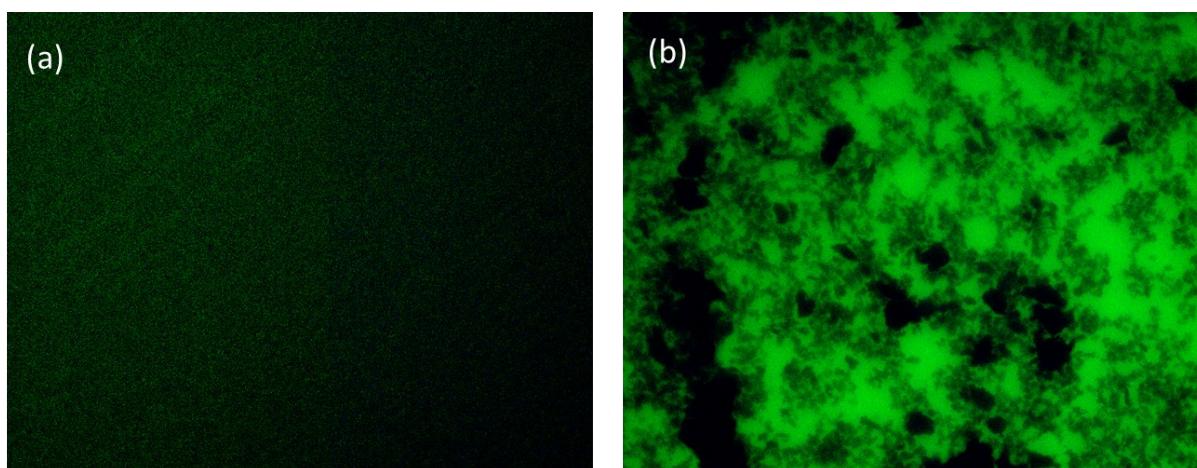


Figure S15 ROS generation (a) in the absence and (b) in the presence of Fe-GQDs

Further, we assessed the reactive oxygen species generation (in cps) as a function of different time periods (5 min, 10 min and 15 min) in presence of UV light. It was observed that the level of ROS generation (Figure S16) was more for Fe-GQDs as compared to the control sample upon UV irradiation. Additionally, a significant amount of ROS was generated after 10 mins which attained saturation after 15 mins. An intracellular ROS is a preliminary marker for bacterial stress and a potential indicator of bacterial cell death.⁴ From the obtained results

we hypothesize that the prepared Fe-GQDs nanoparticles are antibacterial in nature. The facile and quick response in terms of intense oxygen species generation from Fe-GQDs can be ascribed from the synergistic effect of Fe₃O₄ nanoparticles and GQDs. Herein, we assume Fe-GQD based nanoparticles in *E.coli* suspension mediated a series of different redox reactions via Fenton's chemistry which generates hydroxyl radicals. This effect was aggravated by GQDs. Absorption of light in semiconductor GQDs led to the generation of electron-hole pair, which later reacted with the surrounding molecular oxygen or water to form various oxygen moieties. Further, some of the trapped electrons with the adsorbed oxygen molecules would have formed O^{2•-} and holes would have reacted with the H₂O molecules forming •OH. ⁵⁻⁸These radicals concern are toxic to bacterial cells hence these particles can effectively remediate bacterial contaminated water besides removing As³⁺ ions.

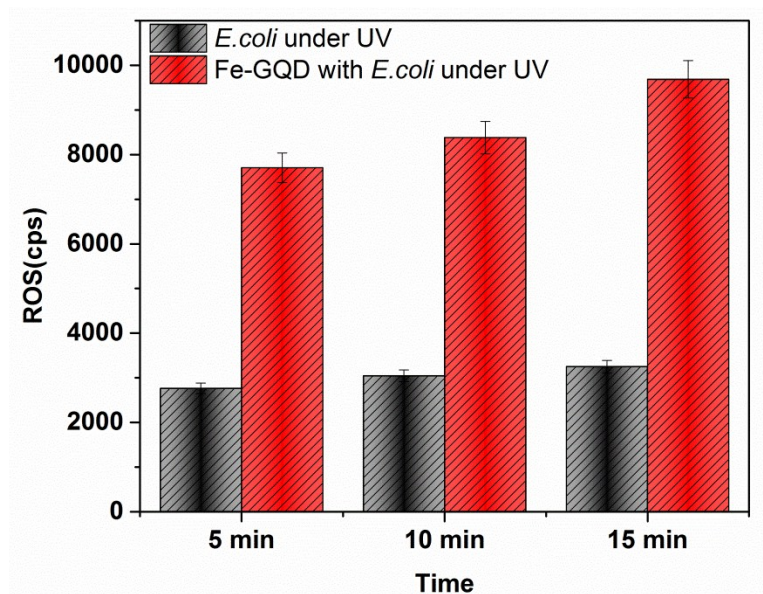


Figure S16 Photo induced ROS production for Fe-GQDs and control

References:

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- (3) Samantaray, P. K.; Madras, G.; Bose, S. Water Remediation Aided by a Graphene-Oxide-Anchored Metal Organic Framework through Pore-and Charge-Based Sieving of Ions. *ACS Sustainable Chemistry & Engineering* **2018**, *7* (1), 1580-1590.
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- (6) Chong, Y.; Ge, C.; Fang, G.; Tian, X.; Ma, X.; Wen, T.; Wamer, W. G.; Chen, C.; Chai, Z.; Yin, J.-J. Crossover between anti-and pro-oxidant activities of graphene quantum dots in the absence or presence of light. *ACS nano* **2016**, *10* (9), 8690-8699.
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- (8) Bose, S.; Samantaray, P. Cationic biocide anchored graphene oxide-based membranes for water purification. *Proceedings of the Indian National Science Academy* **2018**, *84* (3), 669-679.