

Supporting Information

Highly Integrated All-manganese Battery with Supported Oxide Nanoparticles on Cathode and Anode by Super-aligned Carbon Nanotubes

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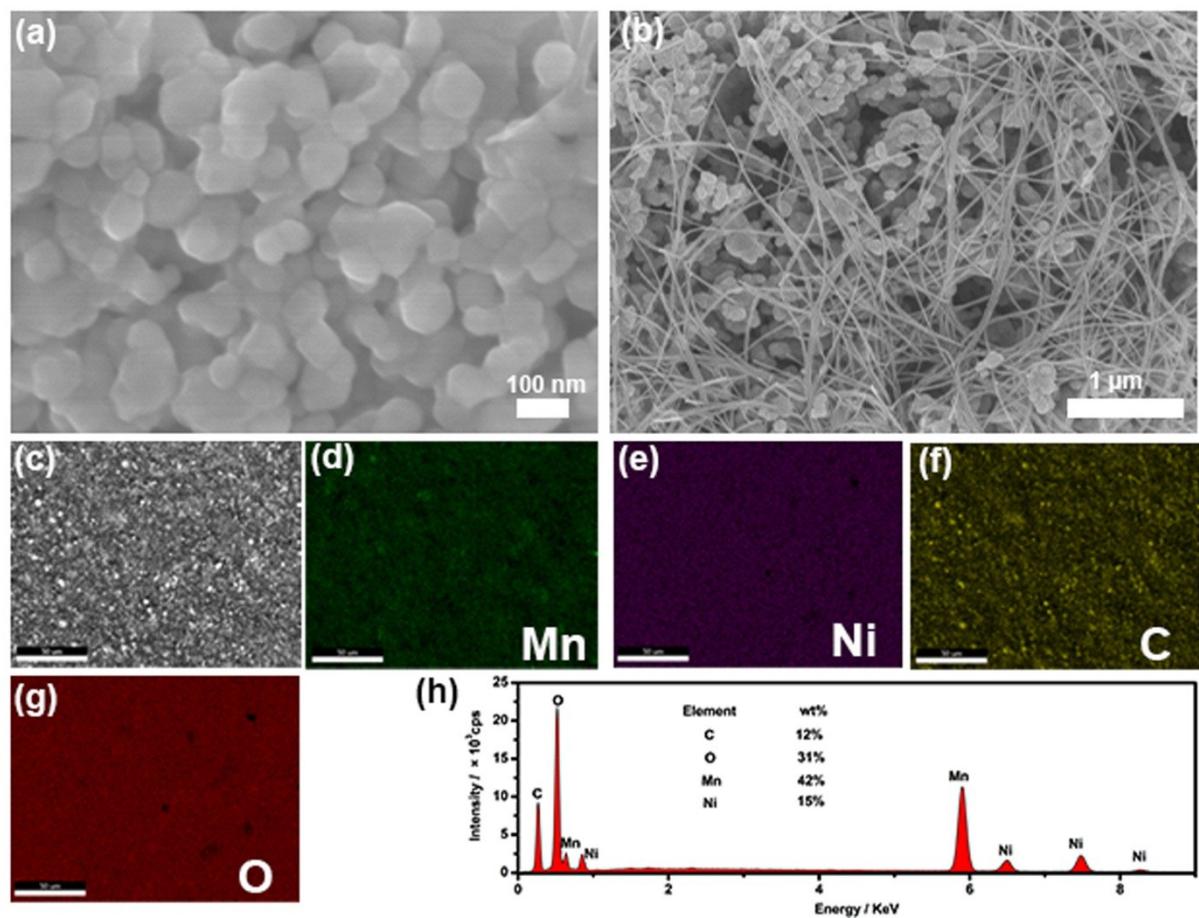


Fig. S1. (a) SEM image of LMN nanoparticles; (b) The top-view SEM image of LMN/SACNT; (c) SEM image of LMN/SACNT and corresponding element mappings of (d) Mn, (e)Ni, (f) C, and (g) O; (h) EDS profile and element contents.

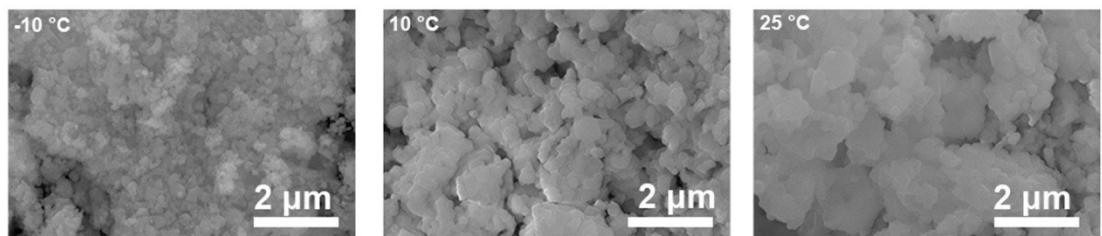


Fig. S2. SEM images of LMN from the precursor synthesized at -10°C, 10 °C and 25 °C.

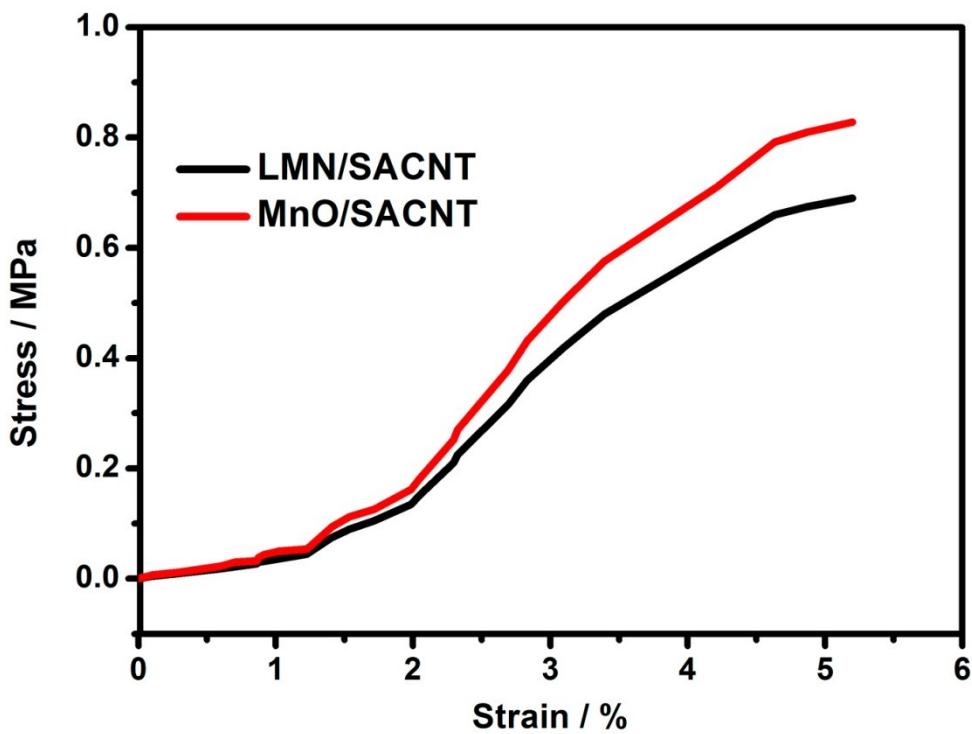


Fig. S3. Stress-strain curves of LMN/SACNT and MnO/SACNT electrodes.

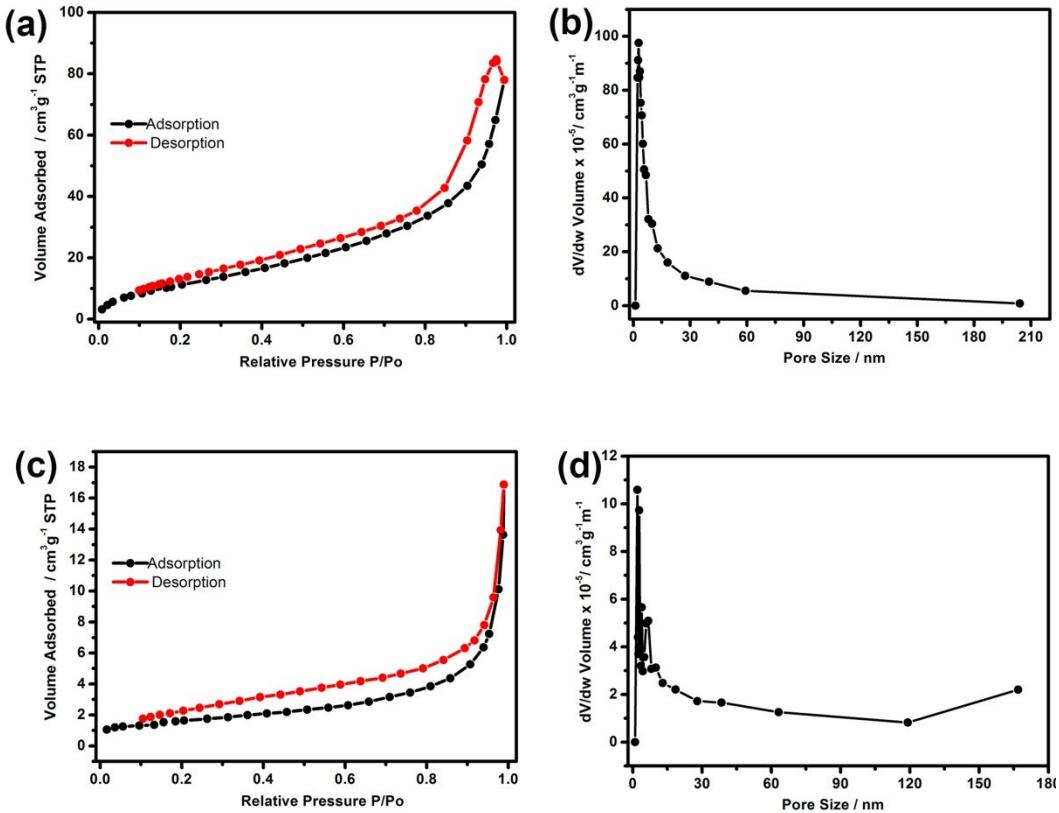


Fig. S4. (a, c) Nitrogen adsorption and desorption isotherms and (b, d) pore distribution curves of LMN/SACNT (a, b) and LMN (c, d).

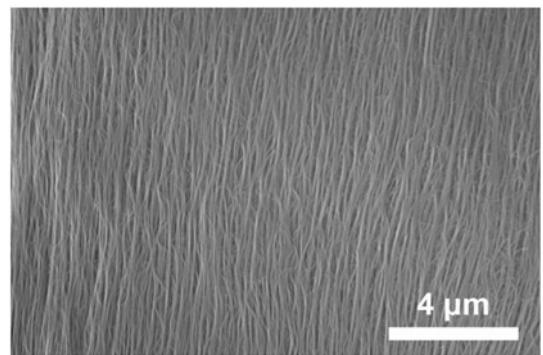


Fig. S5. SEM image of SACNT, indicating the orderly arranging of CNTs in long range.

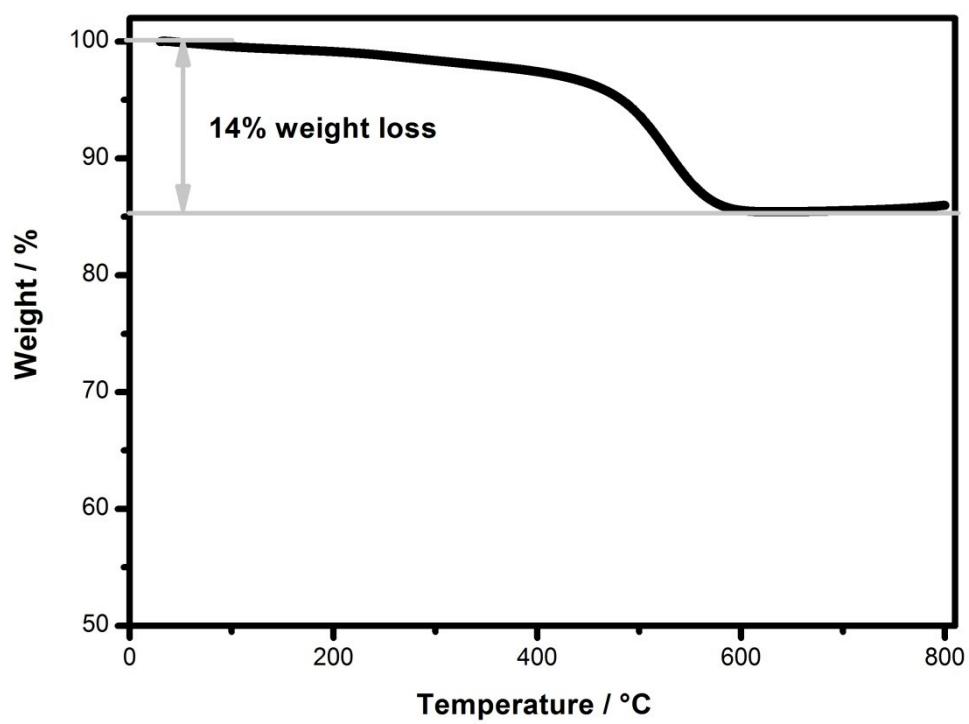


Fig. S6. TGA curve of LMN/SACNT.

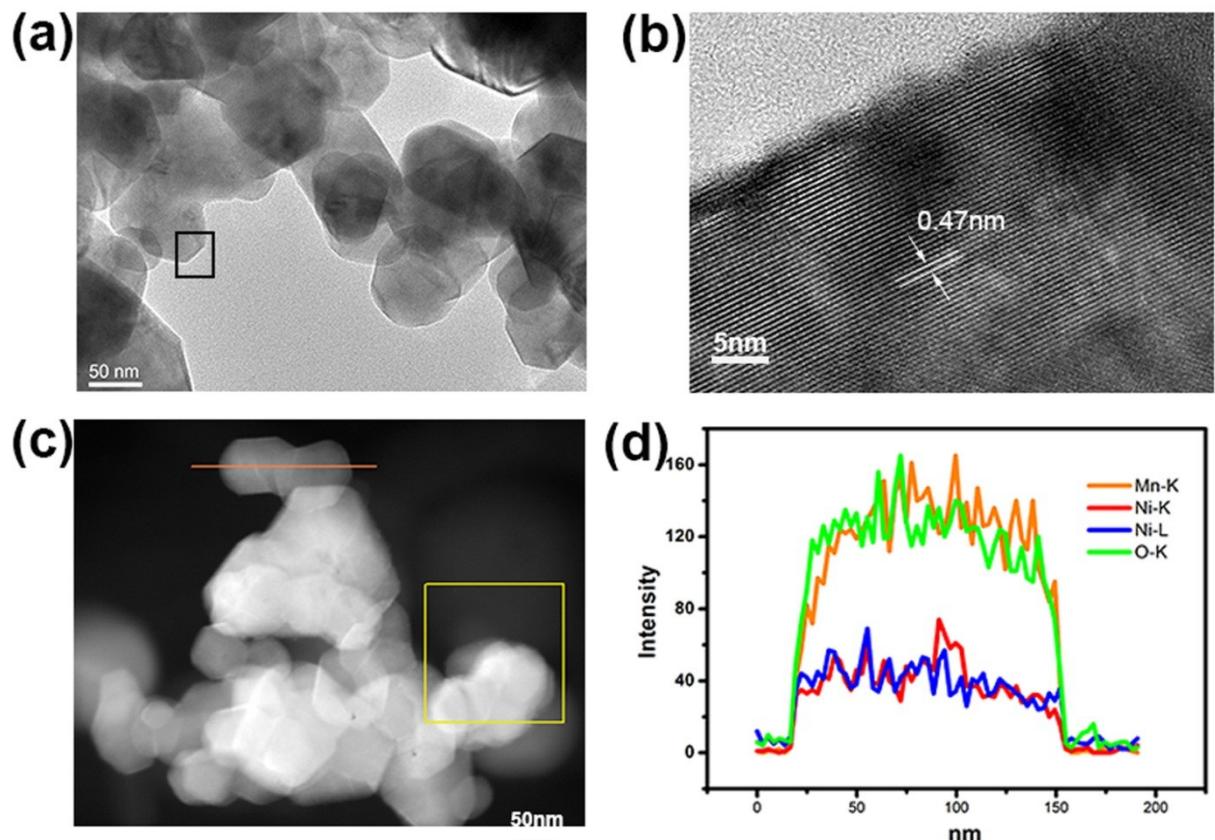


Fig. S7. (a) TEM, (b) HRTEM and (c) HAADF STEM images of LMN nanoparticles; (d) Line scan EDS profile in (c).

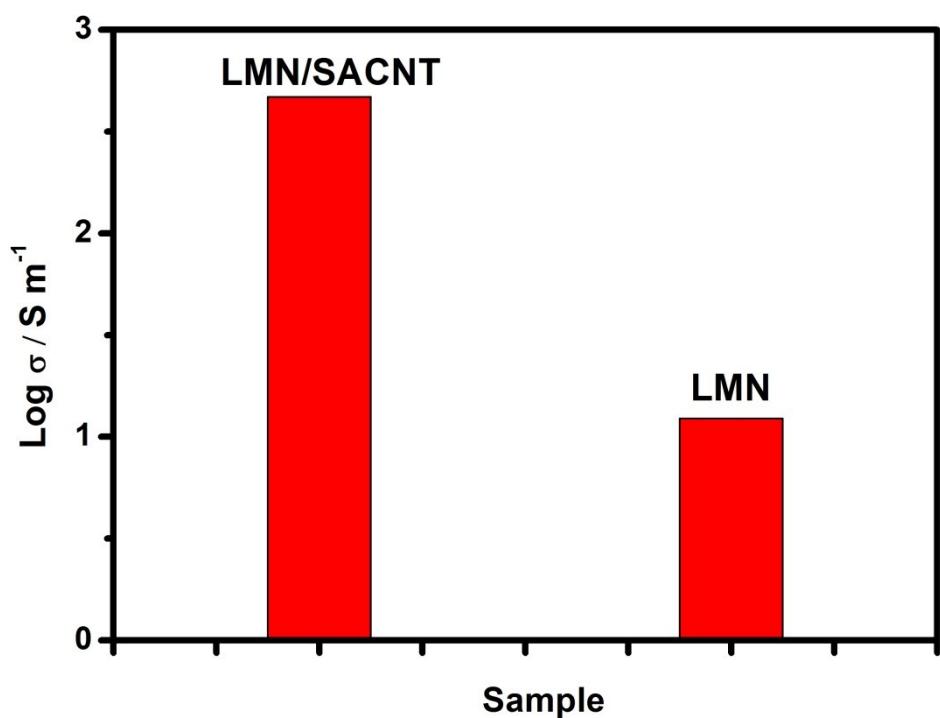


Fig. S8. Electrical conductivity of LMN/SACNT and LMN.

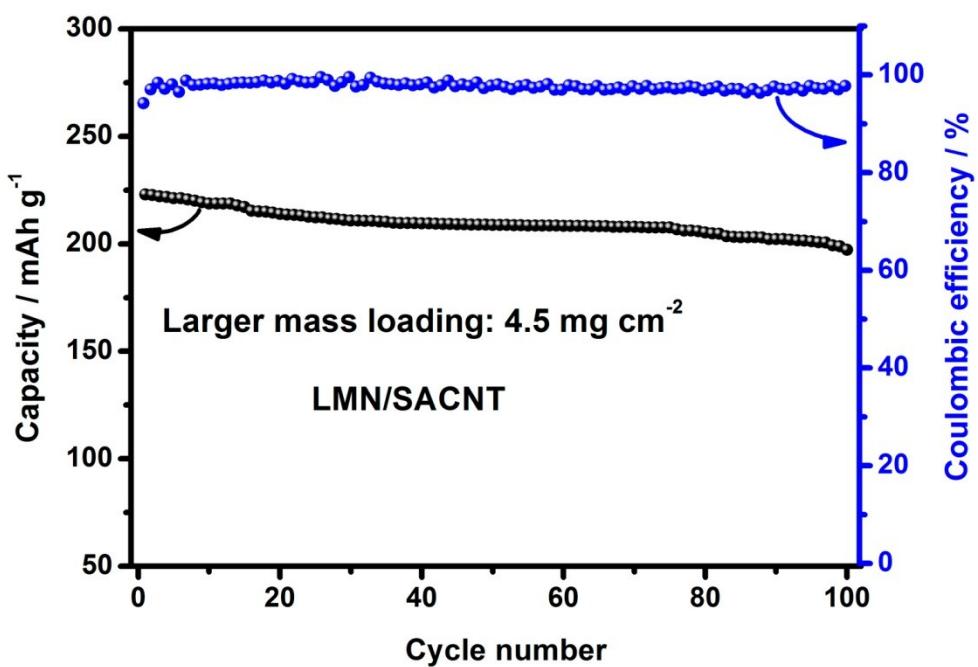


Fig. S9. Cycling performance and corresponding Coulombic efficiency of LMN/SACNT electrode with a mass loading of 4.5 mg cm^{-2} at 0.3C . Before cycling, an initial activation was performed at 0.1 C for 5 cycles.

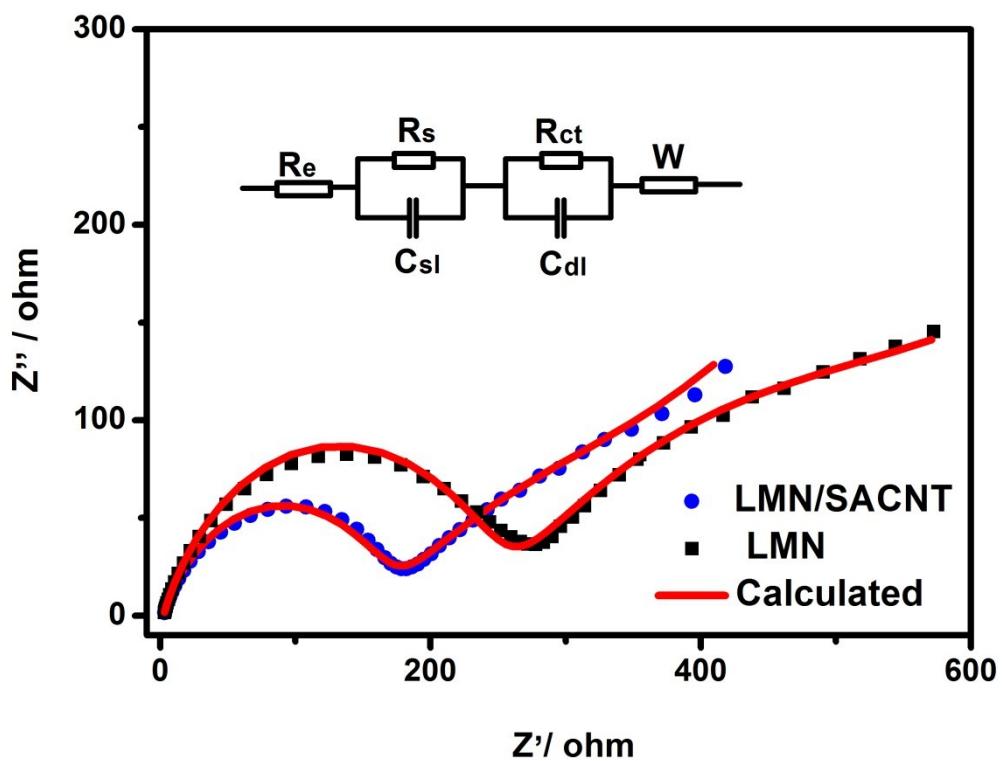


Fig. S10. Electrochemical impedance spectra of tLMN/SACNT and LMN electrodes after 200 cycles at 0.3C.

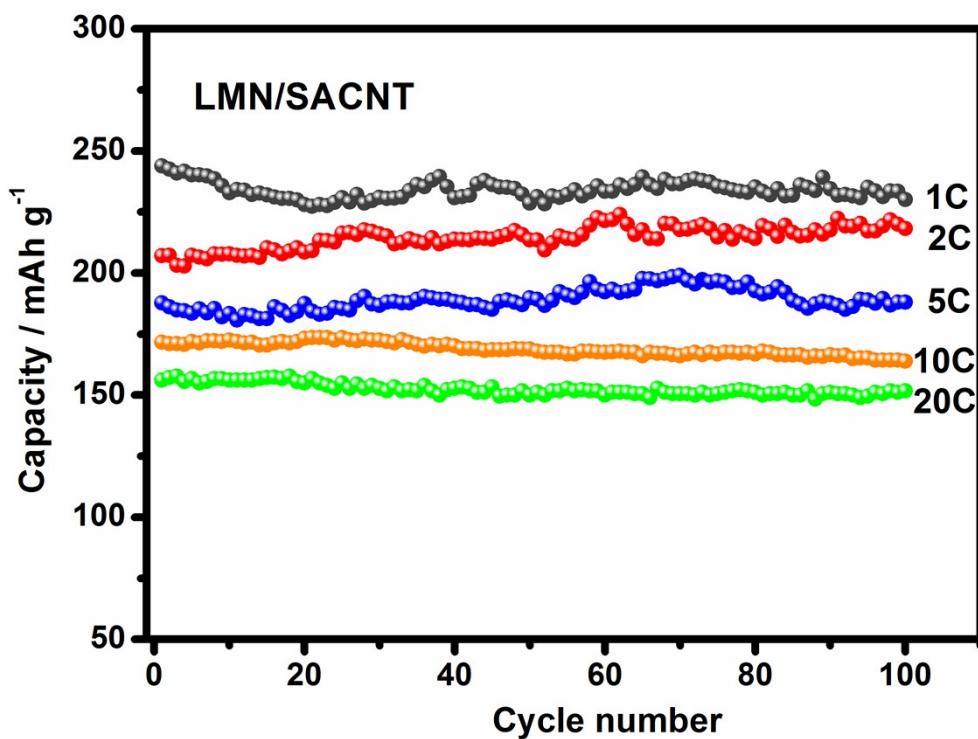


Fig. S11. Cycling performance of LMN/SACNT electrode at various rates (after activation at 0.1 C for 5 cycles) between 2.0-4.8 V.

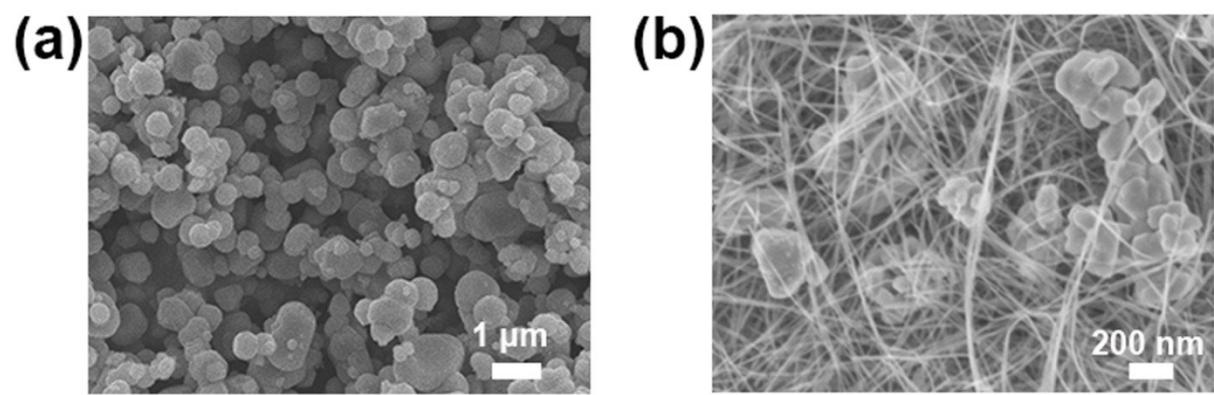


Fig. S12. SEM images of (a) MnO and (b) MnO/SACNT.

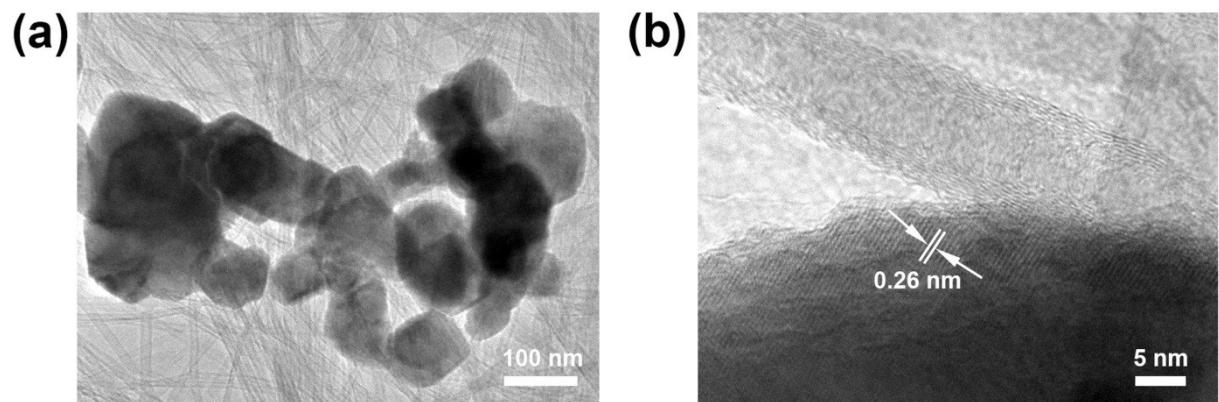


Fig. S13. (a) TEM and (b) HRTEM images of MnO/SACNT.

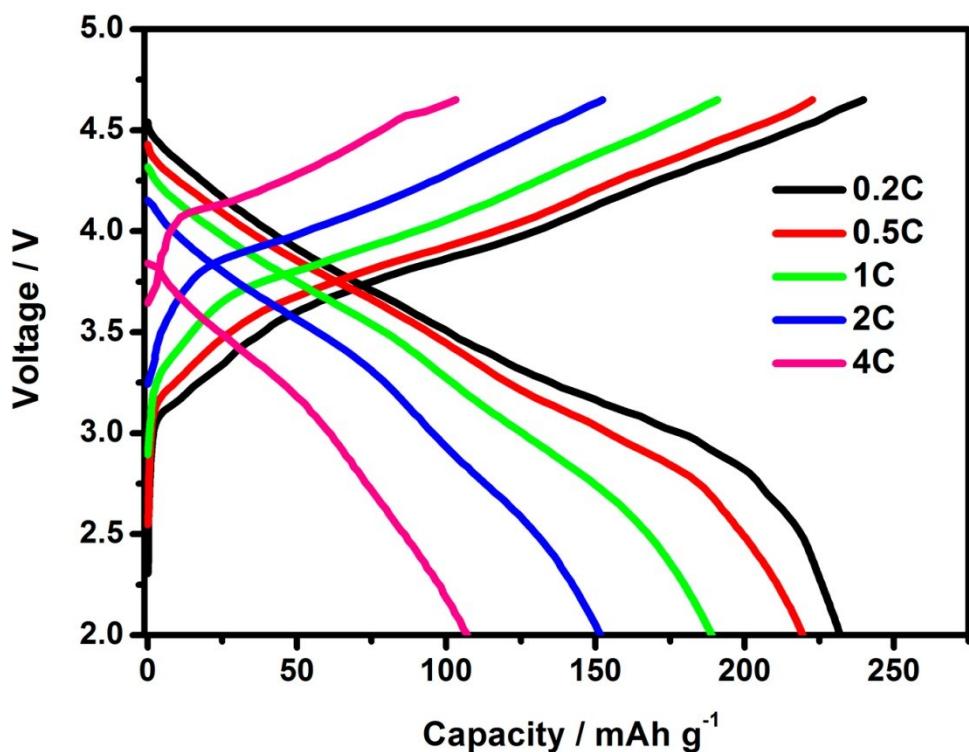


Fig. S14. Typical charge/discharge profiles of LMN/SACNT||MnO/SACNT full cell when performed with rate capability test in Fig. 5c.

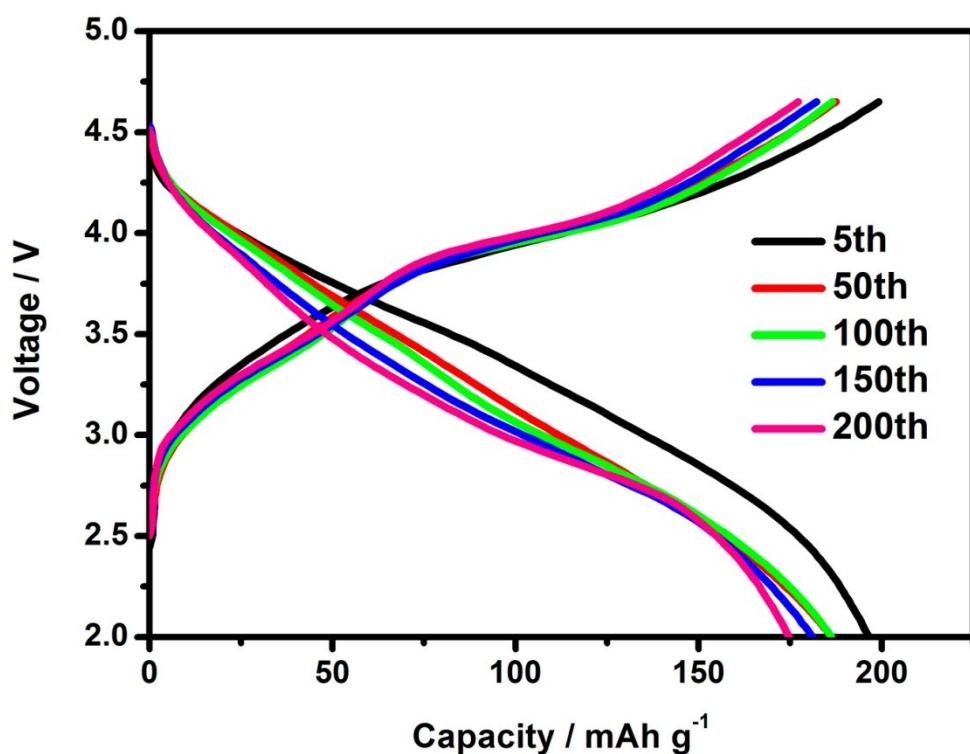


Fig. S15. Typical charge/discharge profiles of LMN/SACNT||MnO/SACNT full cell when performed with cyclic stability test in Fig. 5d

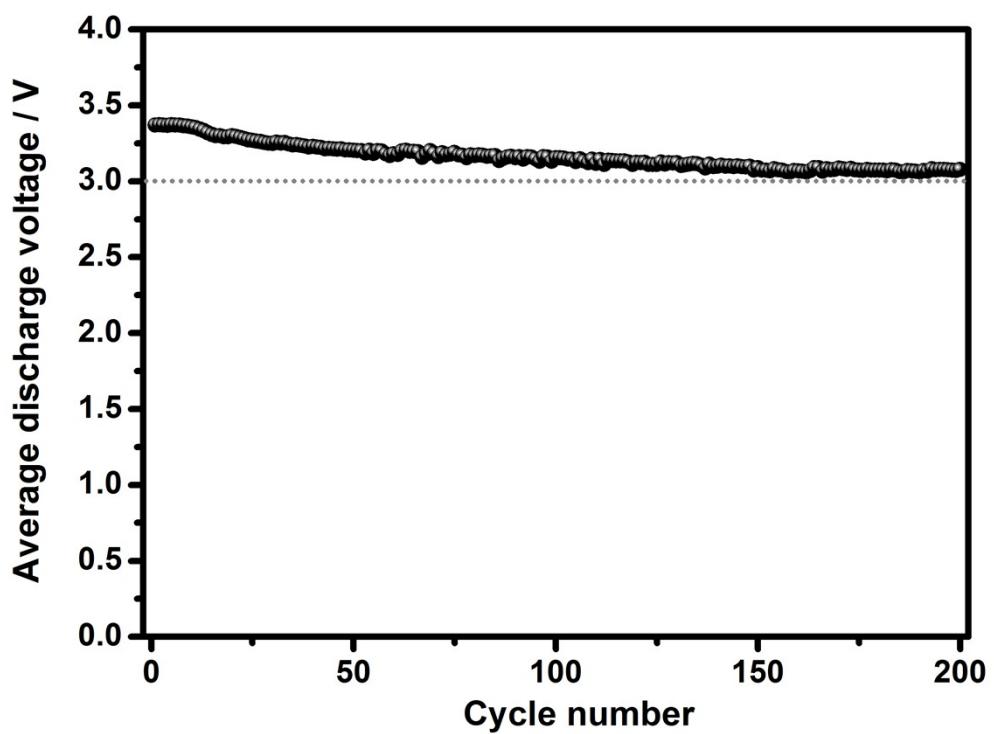


Fig. S16. Evolution of average discharge voltage of LMN/SACNT||MnO/SACNT full cell with cycling at 1 C

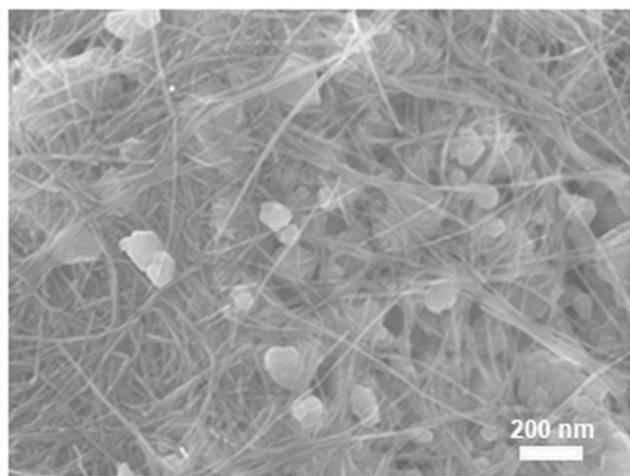


Fig. S17. SEM image of LMN/SACNT electrodde from the cycled LMN/SACNT||MnO/SACNT full cell at 1 C for 200 cycles.

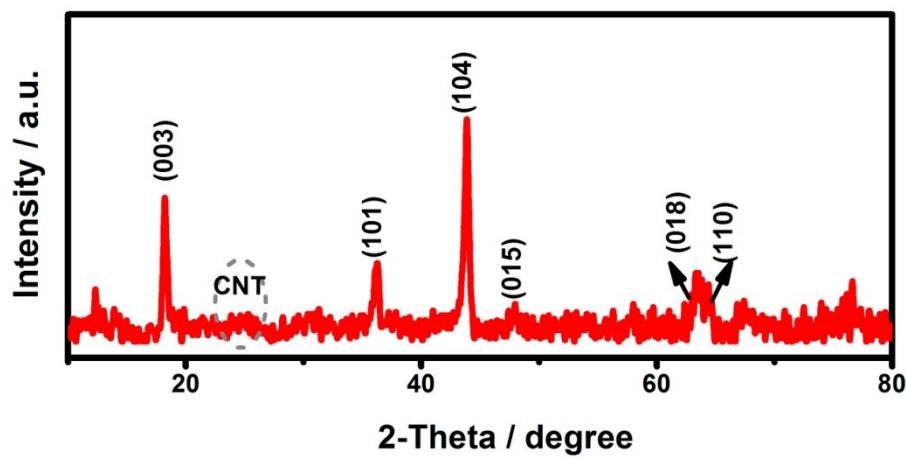


Fig. S18. XRD pattern of LMN/SACNT electrode from the cycled LMN/SACNT||MnO/SACNT full cell at 1 C for 200 cycles.

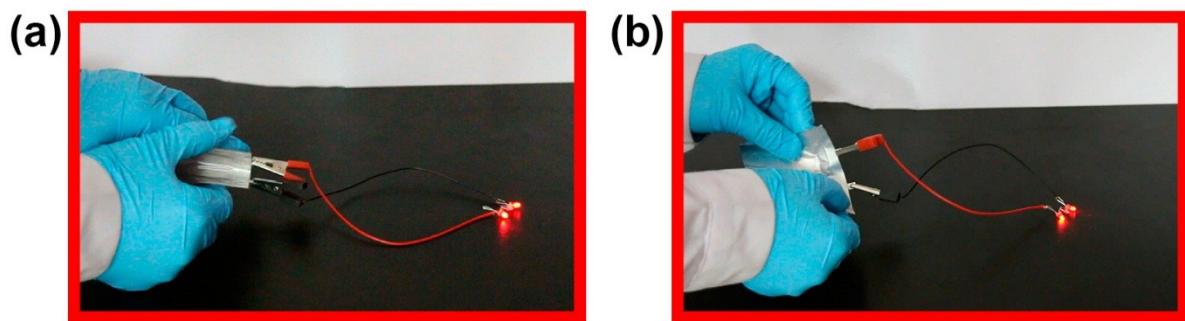


Fig. S19. LMN/SACNT||MnO/SACNT pouch cell that power two LEDs under bending conditions.

Table S1. Fitting parameters for EIS of two samples.

| Samples | R_e / Ω | R_s / Ω | R_{ct} / Ω | R_{total} / Ω |
|-----------|----------------|----------------|-------------------|----------------------|
| LMN/SACNT | 2.11 | 165 | 200 | 367.11 |
| LMN | 2.21 | 248 | 355 | 605.21 |

Table S2. Performance comparisons of our full cell with those reported in literature.

| Sample | Current density / mA g ⁻¹ | Voltage window / V | Capacity / mAh g ⁻¹ | Energy density / Wh kg ⁻¹ | Ref. |
|---|--------------------------------------|--------------------|--|--|-----------|
| rGO/Mn ₃ O ₄ LiMn ₂ O ₄ | 100 | 2.0-4.1 | 100 | | [1] |
| MnO _x /CNT LiMn ₂ O ₄ /CNT | 150 | 2.4-3.7 | ~97 | | [2] |
| Mn ₂ O ₃ LiMn ₂ O ₄ | 100 | 1.5-4.2 | ~99 | | [3] |
| MnO@C LiMn ₂ O ₄ | 25.16 | 2.4-4.2 | 397 (Based on the weight of cathode active material) | | [4] |
| MnO _x /C LiMn ₂ O ₄ | 29.6 | 2.0-4.1 | ~100 | 350 (Based on the weight of cathode active material) | [5] |
| MnO LiNi _{0.5} Mn _{1.5} O _{4-δ} | 14.8 | 2.3-4.6 | | 285.5 (Based on the weight of cathode active material) | [6] |
| LMN/SACNT MnO/SACNT | 200 | 2.0-4.65 | 253 | 411 (Based on the total weight of cathode and anode electrodes, separator and electrolyte) | This work |

Reference:

- [1] J. G. Wang, D. Jin, R. Zhou, X. Li, X. R. Liu, C. Shen, K. Xie, B. Li, F. Kang, B. Wei, *ACS Nano* 2016, **10**, 6227.
- [2] T. Gu, Z. Cao, B. Wei, *Adv. Energy Mater.* 2017, **7**, 1700369.
- [3] Y. Wang, Y. Wang, D. Jia, Z. Peng, Y. Xia, G. Zheng, *Nano Lett.* 2014, **14**, 1080.
- [4] J.-G. Wang, D. Jin, H. Liu, C. Zhang, R. Zhou, C. Shen, K. Xie, B. Wei, *Nano Energy* 2016, **22**, 524.
- [5] C. Chae, H. Park, D. Kim, J. Kim, E.-S. Oh, J. K. Lee, *J. Power Sources* 2013, **244**, 214.
- [6] G. L. Xu, Y. F. Xu, J. C. Fang, F. Fu, H. Sun, L. Huang, S. Yang, S. G. Sun, *ACS Appl. Mater. Interfaces* 2013, **5**, 6316.