

Electronic Supporting Information

A 3D flower-like VO₂/MXene hybrid architecture with superior anode performance for sodium ion batteries

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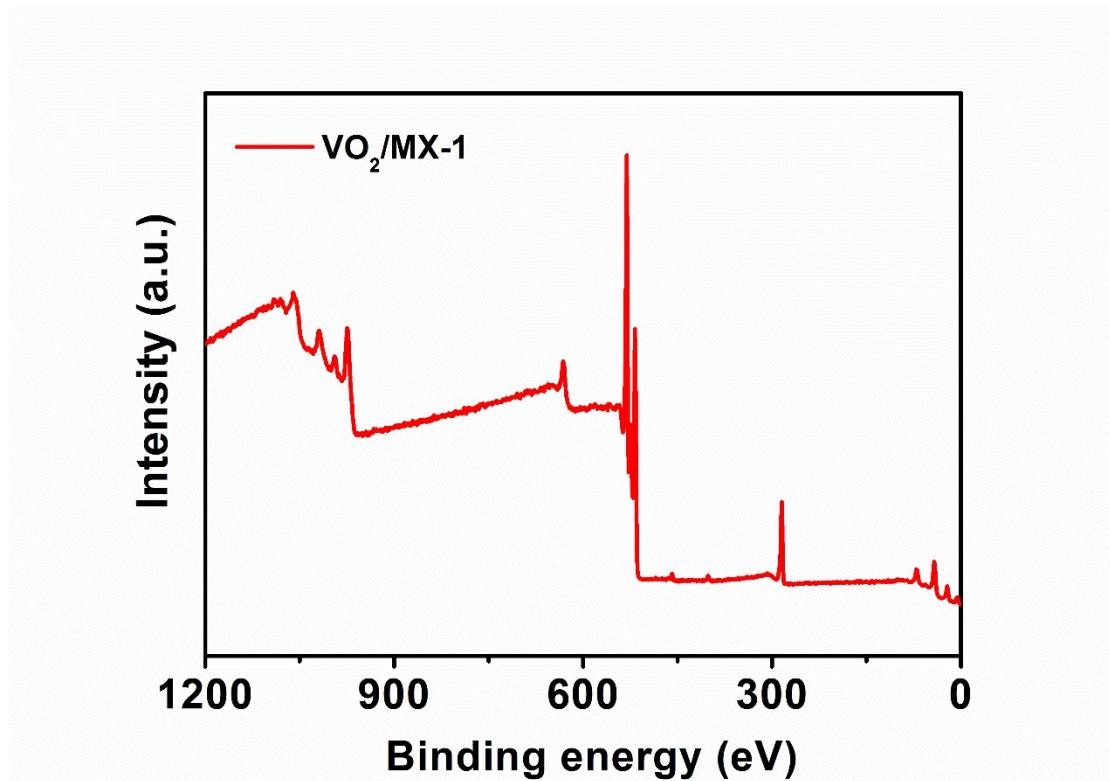


Fig. S1 The survey XPS spectrum of $\text{VO}_2/\text{MX}-1$ hybrid.

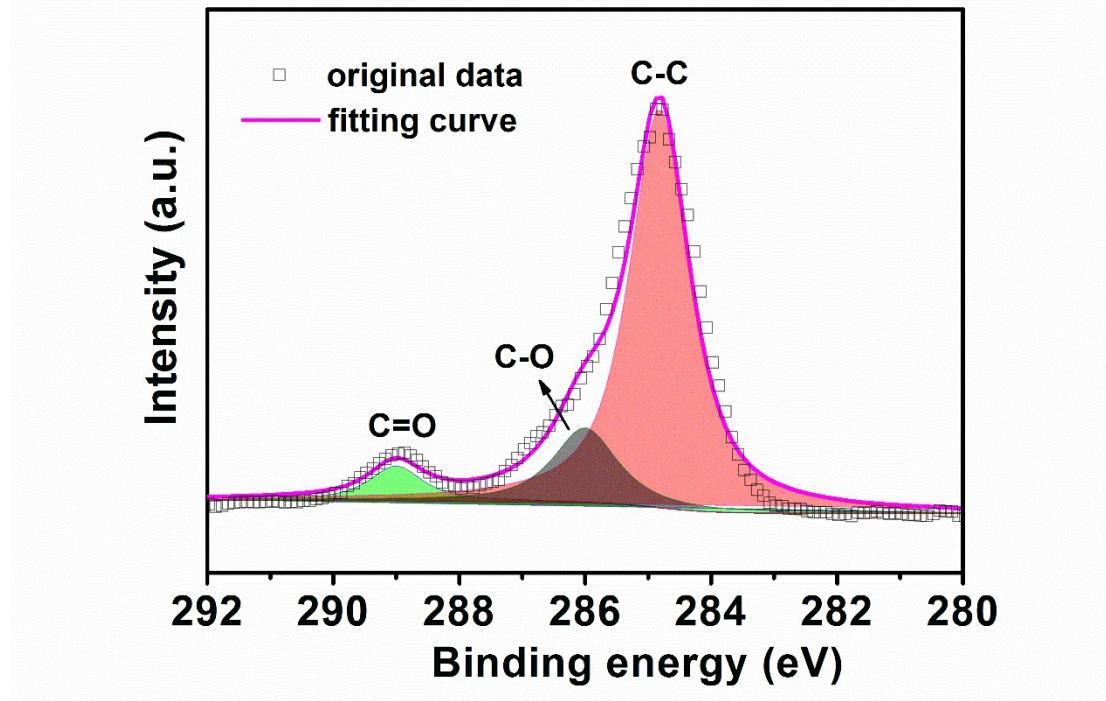


Fig. S2 XPS of C 1s spectra of $\text{VO}_2/\text{MX}-1$ hybrid.

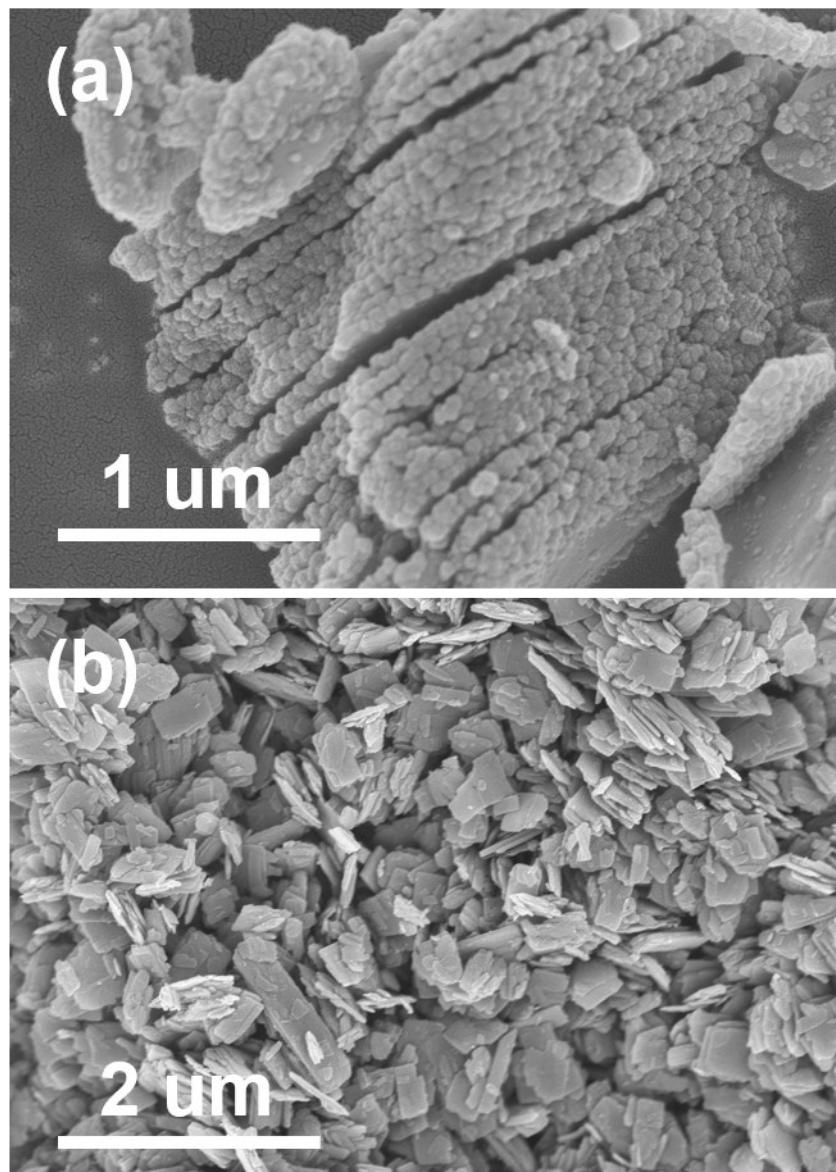


Fig. S3 SEM images of the (a) MX and (b) VO₂.

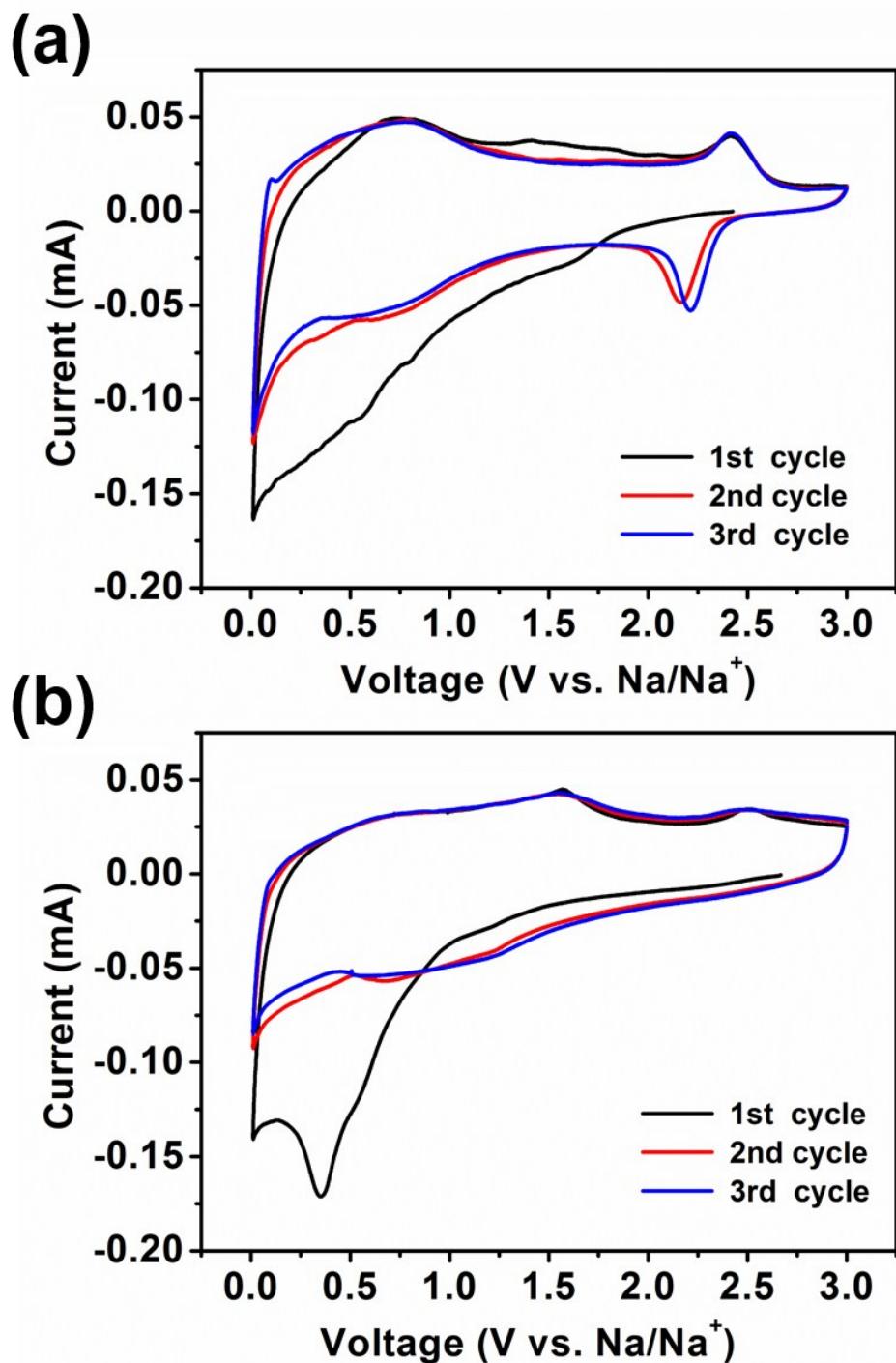


Fig. S4 CV curves of (a) MX and (b) VO₂ electrodes at a scanning rate of 0.1 mV s⁻¹.

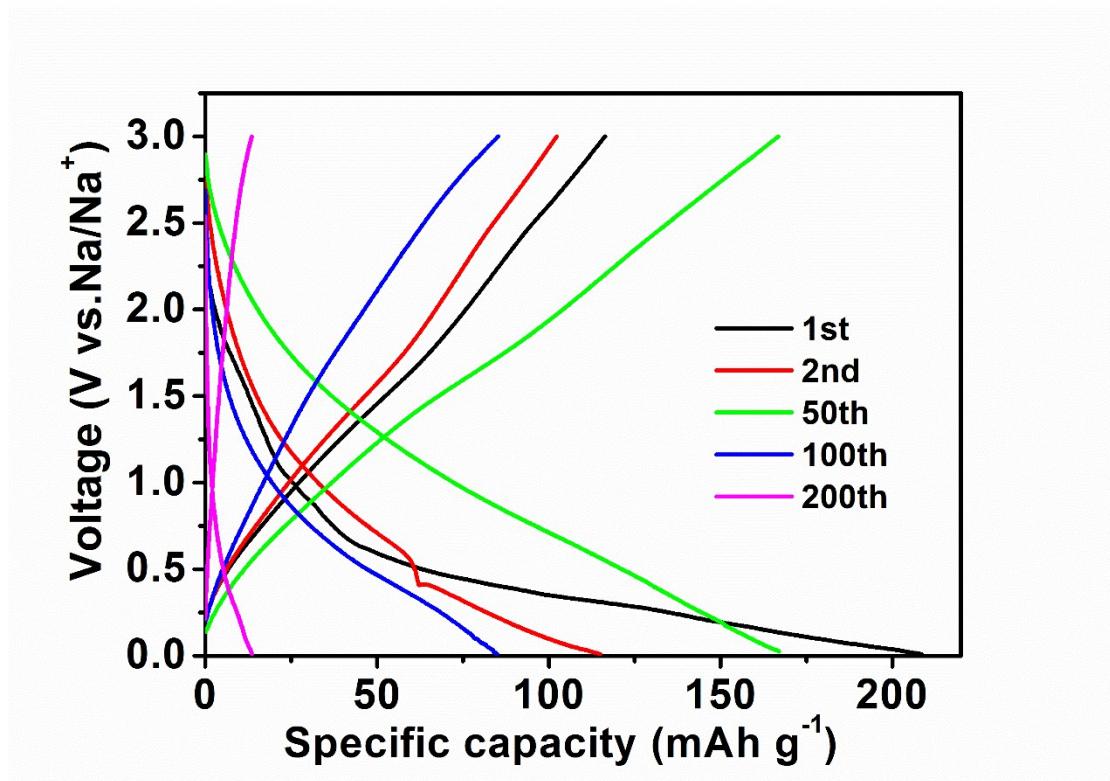


Fig. S5 Charge/discharge curves of the VO_2 electrode cycled for the 1st, 2nd, 50th, 100th, and 200th cycle at a current of 0.1 A g^{-1}

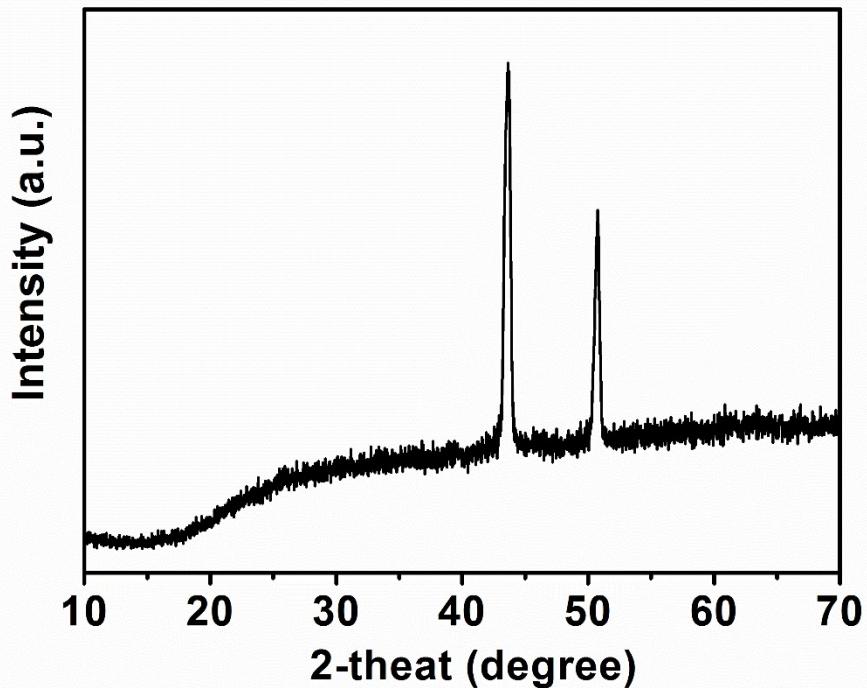


Fig. S6 Ex-situ XRD pattern of the $\text{VO}_2/\text{MX}-1$ electrode after the 10th cycle discharged to 0.01V.

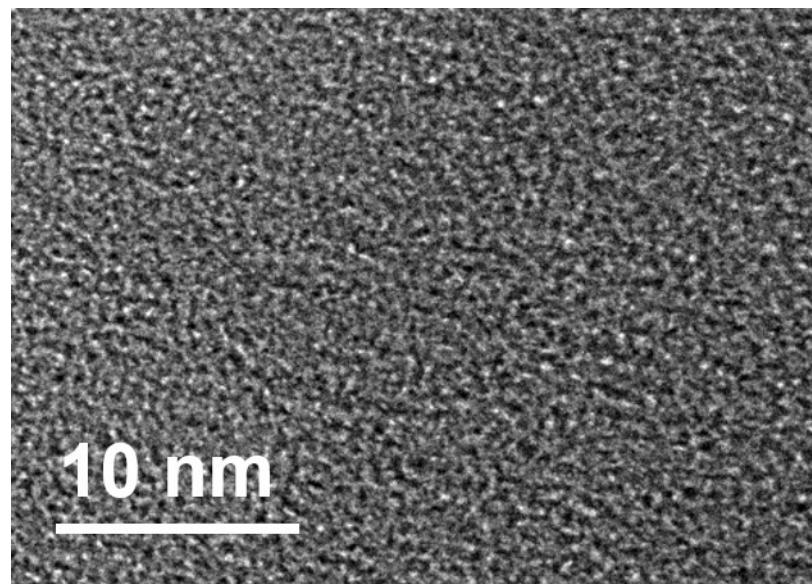


Fig. S7 HRTEM image of the VO₂/MX-1 electrode after the 10th cycle discharged to 0.01V.

Table S1 The electrochemical performance comparison of MXene based and VO₂ materials for SIBs

Electrode materials	Current density (mA g ⁻¹)	Final capacity (mA h g ⁻¹)	Capacity retention (%)	Ref.
Ti ₃ C ₂ MXene-Derived Sodium Nanoribbons (NaTi _{1.5} O _{8.3})	200	171 mA h g ⁻¹ after 50 cycles	~81 (C _{50th} /C _{2nd})	1
alkalized Ti ₃ C ₂ MXene nanoribbons (a-Ti ₃ C ₂ MNRs)	50	113 mA h g ⁻¹ after 200 cycles	~67 (C _{200th} /C _{2nd})	2
MoS ₂ /Ti ₃ C ₂ T _x composite	100	250.9 mA h g ⁻¹ after 100 cycles.	~88 (C _{100th} /C _{2nd})	3
dimethyl sulfoxide intercalate into Ti ₃ C ₂ T _x (d-D-Ti ₃ C ₂ T _x)	100	103 mA h g ⁻¹ after 500 cycles	~86 (C _{500th} /C _{2nd})	4
Sb ₂ O ₃ /MXene	100	472 mA h g ⁻¹ after 100 cycles	~106 (C _{100th} /C _{2nd})	5
VO ₂ /rGO nanorods	60	173 mA h g ⁻¹ after 100 cycles	~70 (C _{100th} /C _{2nd})	6
VO ₂ /crumpled rGO	100	260 mA h g ⁻¹ after 500 cycles	~70 (C _{500th} /C _{2nd})	7
3D flower-like VO ₂ /MX-1	100	280.9 mA h g ⁻¹ after 200 cycles	~141 (C _{200th} /C _{2nd})	This work

References

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