Hierarchical electrodes of $NiCo_2S_4$ nanosheets anchored sulfur-doped Co_3O_4 nanoneedles with advanced performance for batterysupercapacitor hybrid devices

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Fig. S1 SEM image of NF/NiCo₂S₄.



Fig. S2 Nitrogen adsorption/desorption of the NF/S-Co₃O₄ and NF/S-Co₃O₄@NiCo₂S₄.



Fig. S3 XPS wide-scan spectra of S-Co₃O₄@NiCo₂S₄.



Fig. S4 High-resolution XPS spectra for Co of S-Co₃O₄.



Figure S5. The equivalent electrical circuit in NF/S-Co₃O₄@NiCo₂S₄, NF/S-Co₃O₄, NF/Co₃O₄, and NF/NiCo₂S₄.



Fig. S6 Electrochemical performance of $S-Co_3O_4$ @Ni Co_2S_4 with different molar weight (6, 10, 12 mmol) of NH₄F. (a) EIS curves; (b) CV curves; (c) GCD curves; and (d) Specific capacity at different current densities from 8 to 30 mA/cm².



Fig. S7 Capacitive contribution of the NF/S-Co₃O₄@NiCo₂S₄.



Fig. S8 SEM of NF/S-Co₃O₄@NiCo₂S₄ after 5,000 cycles.



Fig. S9 (a) CV curves of AC at different scan rates. (b) GCD curves of AC at different current densities. (c) Specific capacity at different current densities from 0.5 to 30 A/g.



Fig. S10. CV curves of BSH device with different window voltages at 5 mV/s.

Table S1. Comparison of electrochemical performance between the NF/S- Co_3O_4 @NiCo_2S4 and the reported sulfides-based electrode materials.

Electrode material	Specific capacitance (capacity)	Cycling performance	Ref
hollow ellipsoid Ni–Mn sulfides	1636.8 F/g at 2 A/g	95.1% (4000 cycles)	[1]
NiCo ₂ S ₄ /NCF	1231 F/g at 2 A/g	90.4% (2000 cycles)	[2]
NiCo ₂ S ₄ @Ni-Mn LDH arrays/GS	1.48 F/cm ² at 5 mA/cm ²	88.3% (1000 cycles)	[3]
porous NiCo2S4 aerogel	1268 F/g at 1 A/g		[4]
P-Doped NiCo ₂ S ₄ nanotubes	8.03 F/cm ² at 2 mA/cm ²	87.5% (5000 cycles)	[5]
NiCo ₂ O ₄ @NiCo ₂ S ₄ Nanocomposite	1872 F/g at 1 A/g	65% (4000 cycles)	[6]
Onion-like NiCo ₂ S ₄ particles	1016 F/g at 2 A/g	87% (10000 cycles)	[7]
FeCo ₂ S ₄ -NiCo ₂ S ₄ composite	1519 F/g at 5 mA/cm ²		[8]
NF/S-Co ₃ O ₄ @NiCo ₂ S ₄	1886 F/g at 2.5 A/g or (6 F/cm ²	97.3% (5000 cycles)	This
	at 8 mA/cm ²)		work

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