

Supplementary information

Proposal of a Stable B₃S Nanosheet as an Efficient Hydrogen Evolution Catalyst

Hong Wu,^a Xingxing Li,^{a,b} Ruiqi Zhang^a and Jinlong Yang^{*a,b}

^a Hefei National Laboratory of Physical Sciences at the Microscale, University of Science and Technology of China, Hefei, Anhui 230026, China.

^b Synergetic Innovation Center of Quantum Information & Quantum Physics, University of Science and Technology of China, Hefei, Anhui 230026, China.

*Corresponding author: jlyang@ustc.edu.cn

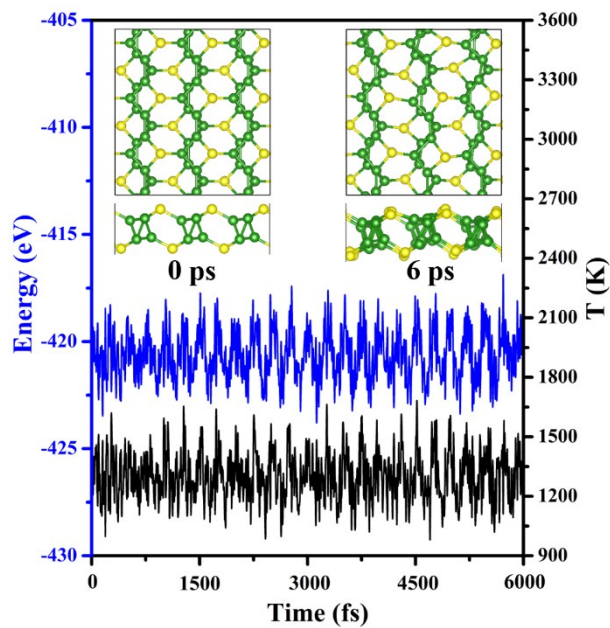


Figure S1. The fluctuation of total energy (blue) and temperature T (black) during the AIMD simulation at 1300 K. The insets are the top and side views of B_3S structure at the time of 0 and 6 ps, respectively.

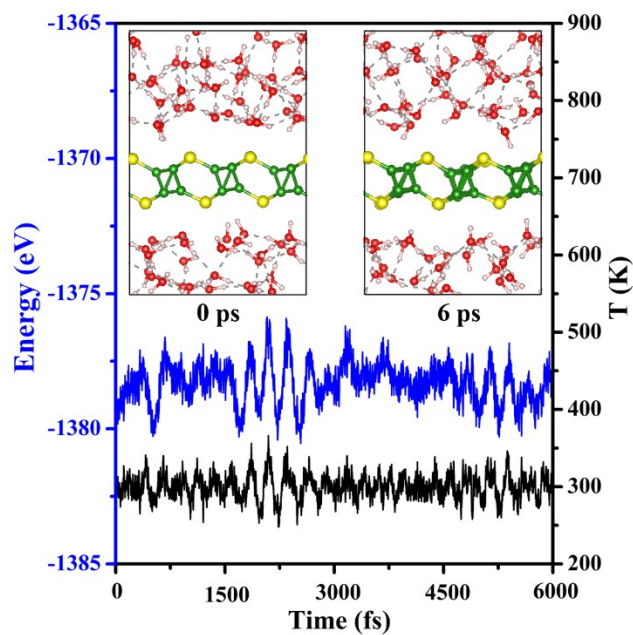


Figure S2. The fluctuation of total energy (blue) and temperature T (black) during 300 K AIMD in water environment. The insets are the side views of B₃S structure with 64 H₂O molecules at the time of 0 and 6 ps, respectively. Green, yellow, red and pale pink balls represent B, S, O and H atoms, respectively.

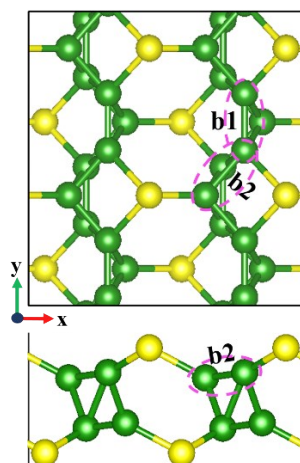


Figure S3. Two different B-B bridge sites (labelled as b1 and b2) on the B_3S surface that are possible for hydrogen adsorption. We find only the B-B bridge sites parallel to y-axis (b1) can stabilize atomic hydrogen, while a hydrogen atom initially placed on b2 site migrates to b1 site after structure optimization due to the steric repulsion of the surrounding S atoms.

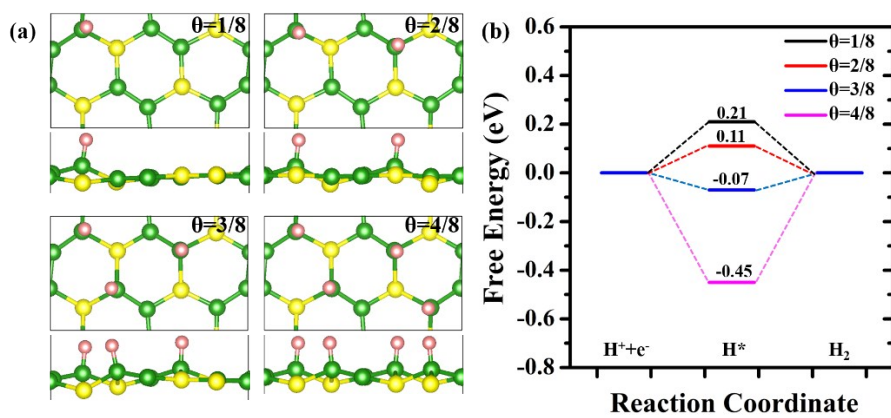


Figure S4. (a) The most stable configurations and (b) corresponding free energies of hydrogen adsorption on B₂S monolayer under different hydrogen coverages. Green, yellow and pink balls represent B, S and H atoms, respectively.