## An aluminum(III) picket fence phthalocyanine-based heterogeneous catalyst for ring-expansion carbonylation of epoxides

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## **Supporting Information**



Fig. S1 <sup>1</sup>H NMR spectra of the ligand and AlPc'Cl monomer.



Fig. S2 Elemental analysis of the ligand (left) and AlPc'Cl monomer (right).



Fig. S3 TEM images of the network 2.



Operator ID: Company name: Method filename: Method name: Analysed: Printed:	LEE HK SOGANG LINC C:\Eager for FLAy NCHS 2018-05-08 12:09 2018-05-09 09:14	SH\유기반응센터\ 9 4	N C H S syster	n 170	316.mth
Sampler method:					
Sample ID:	JJ3 (# 52)				
Analysis type:	UnkNown				
Chromatogram filename:	Q051.dat				
Calibration method:	K Factors				
Sample weight:	1.343				
Protein factor:	6.25				
Element Name		Ret.Time	Area	BC	Area ratio
Nitrogen	5 3248	48	161400	RS	27 257450
Carbon	67.7286	74	4399339	RS	1.000000
Hydrogen Totals	5.3120 78.3654	215	1033964 5594703	RS	4.254828

.202058E+07 .480874E+07 .144163E+08 48 74 215 161400 RS 4399339 RS 1033964 RS 5594703 5.3248 67.7286 5.3120 78.3654

K factor

Fig. S4 Elemental analysis of the network 2.



**Fig. S5**. <sup>1</sup>H NMR (a) and <sup>13</sup>C NMR (b) spectra of monomer **1**. The solid-state <sup>13</sup>C NMR of network **2** (c).



Fig. S6 XRD pattern of the network 2.



the atomic ratio of Al/N:  $\frac{0.61}{5.79} = 0.1053$ 

Theoretically, the ratio of Al/N: 1/8 = 0.125

Therefore, the proportion of AI metalated Pc' ring in the networs:

$$\frac{0.1053}{0.125} \times 100\% = 84.2\%$$

Fig. S7 SEM and EDX of the network 2.



Fig. S8 SEM images of the network 2 (a, b) and catalyst 3 (c, d).



Fig. S9 Al  $(2p_{3/2})$  electron binding energy of the network 2 and catalyst 3.



Element	Wt%	Atom %
С	79.81	84.86
Ν	6.55	5.97
0	9.96	7.95
Al	1.18	0.56
Cl	0.55	0.20
К	0.40	0.13
Со	1.56	0.34
Total:	100.0	100.0

Fig. S10 SEM and EDX of catalyst 3.



Fig. S11 BJH pore-size distribution for the network 2 and catalyst 3.



**Fig. S12-1** <sup>1</sup>H NMR spectrum of the product from PO carbonylation in DME solvent using AlPc'-based heterogeneous catalyst **3**, 40 bar CO, 23 °C, 1 h (Table 1, entry 1).



**Fig. S12-2** <sup>1</sup>H NMR spectrum of the product from PO carbonylation using an *in situ* generated catalyst from AlPcCl and Co<sub>2</sub>(CO)<sub>8</sub>). (4 mol% AlPcCl, 6 mol% Co<sub>2</sub>(CO)<sub>8</sub>, 0.5 M PO in THF, 40 bar CO, 23 °C, 1 h).



**Fig. S13** <sup>1</sup>H NMR spectrum of the product from PO carbonylation in DME solvent using catalyst **3**, 10 bar CO, 23 °C, 1 h (Table 1, entry 2).



**Fig. S14** <sup>1</sup>H NMR spectrum of the product from allyl glycidyl ether carbonylation (Table 1, entry 3).



**Fig. S15** <sup>1</sup>H NMR spectrum of the product from benzyl glycidyl ether carbonylation (Table 1, entry 4).



**Fig. S16** <sup>1</sup>H NMR spectrum of the product from epichlorohydrin carbonylation (Table 1, entry 5).



**Fig. S17** <sup>1</sup>H NMR spectrum of the product from oxetane carbonylation in DME (Table 1, entry 6).



**Fig. S18** <sup>1</sup>H NMR spectrum of the product from oxetane carbonylation in THF (Table 1-6).



Fig. S19 <sup>1</sup>H NMR spectrum of the product from oxetane carbonylation using the network 2 or  $KCo(CO)_4$  alone.



Fig. S20 <sup>1</sup>H NMR spectrum of the product from oxetane carbonylation using an equimolar mixture of the network 2 and  $KCo(CO)_4$ .



Fig. S21 <sup>1</sup>H NMR spectrum of the product from filteration test.



Fig. S22 SEM images of catalyst 3 before (a, b) and after (c, d) 3 catalytic cycles.



**Fig. S23** The BET surface areas (a) of catalyst before and after three catalytic cycles, with corresponding BJH pore-size distributions (b).