

Supporting Information for *Journal of Materials Chemistry A*

**MXene-derived TiO₂/reduced graphene oxide composite with
an enhanced capacitive capacity for Li and K ion batteries**

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Table S1. Determination of the size in TiO₂/RGO.

Sample	Index	2θ (degree)	β (degree)	Diameter(nm)
TiO ₂ /RGO	(101)	25.303	0.49	16.44

Calculation method: the Scherrer equation [1]:

$$\text{Diameter} = 0.89 * \lambda / (\beta * \cos\theta) \quad [1]$$

The λ is the wavelength of the X-ray (0.15418 nm) and β is the full width at half maximum of the diffraction peak.

Table S2. The cycling performance at low and high current density contrast with literature.

Refer	Current density (mA/g)	Cycles and retention	Current density (mA/g)	Cycles and retention
The work	200	200, 78%	1000	1000, 85%
1	140	150,83%	280	500,75%
2	100	300,68%	100	1000,65%
3	200	200, 90%	1000	1000, 70%
4	200	200, 100%	1000	600, 100%
5	100	100, 94%	500	500, 80%
6	200	100, 62.5%	--	--
7	55.8	100, 90%	558	500, 94.6%
8	200	500, 60%	--	--
9	500	200, 78%	--	--
10	50	100, 96%	--	--
11	400	400,72.8%	--	--

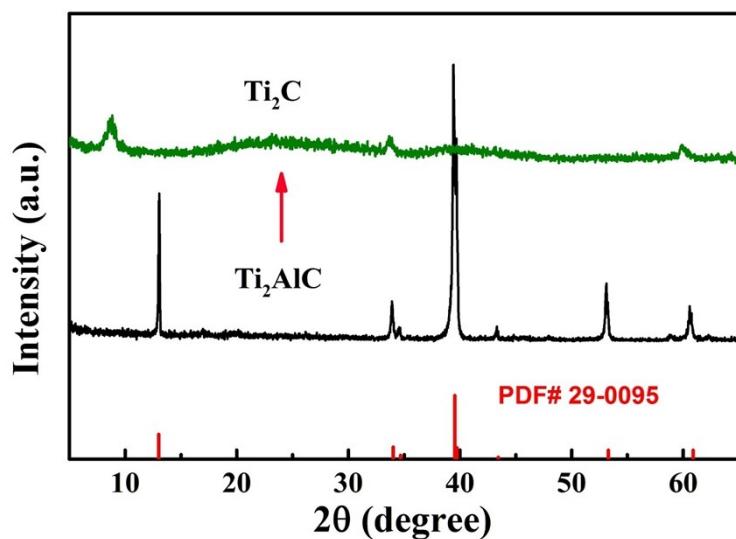


Figure S1. XRD of the Ti_2AlC and the Ti_2C .

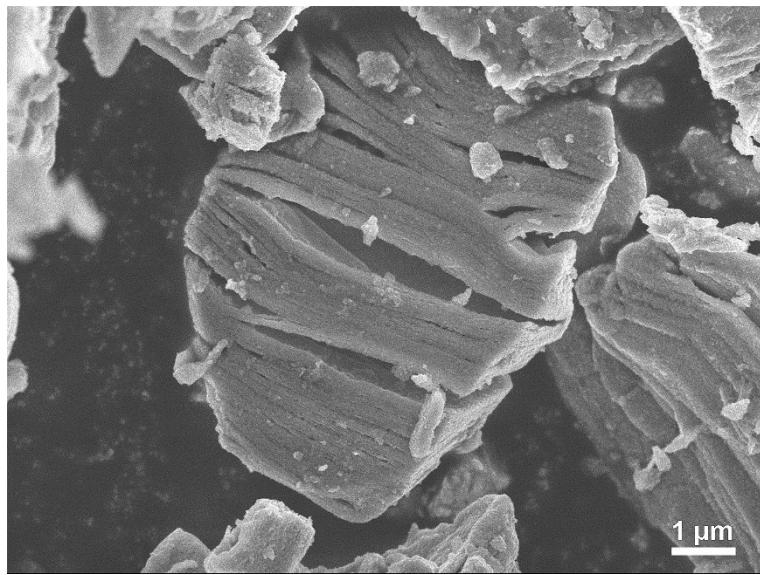


Figure S2. SEM of the multilayer Ti_2C .

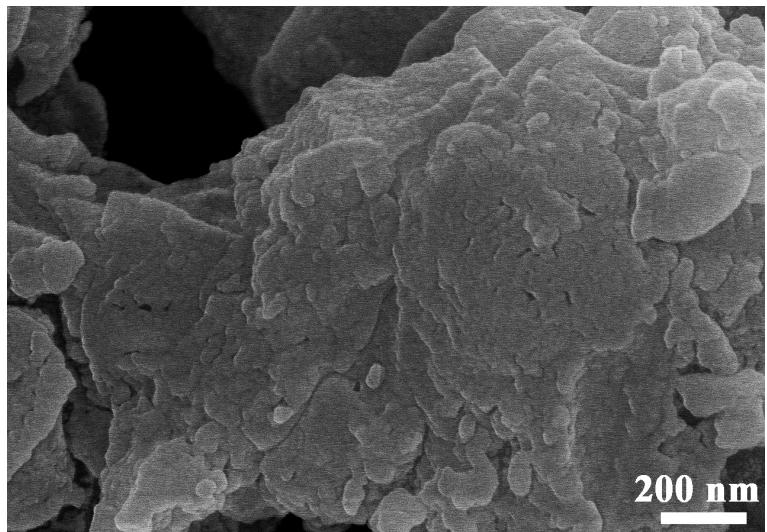


Figure S3. SEM of the amorphous Ti-based intermediates.

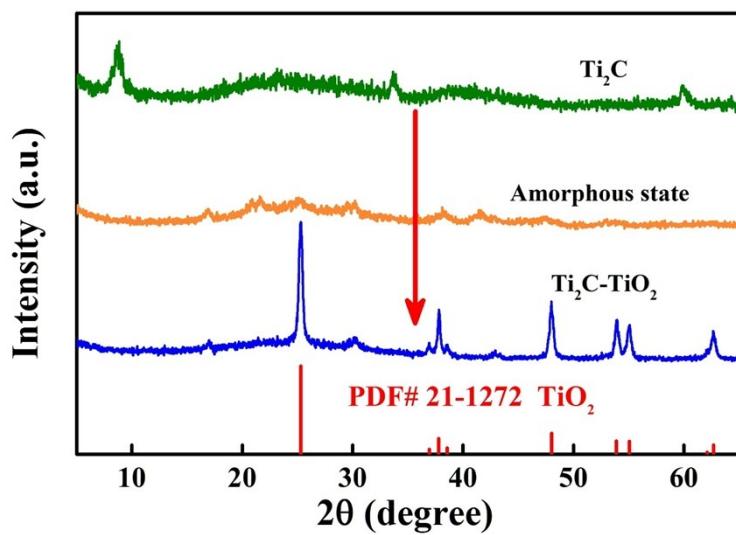


Figure S4. XRD of the change process from the Ti₂C to amorphous state to the Ti₂C - derived TiO₂.

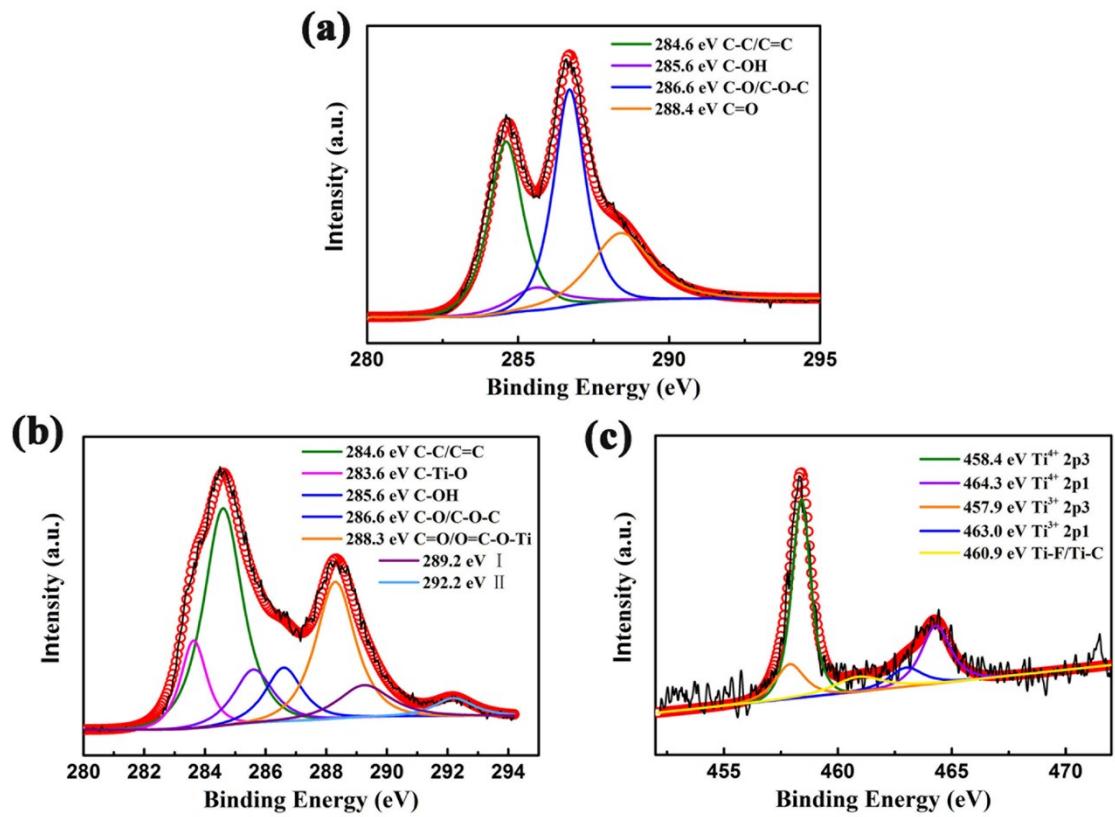


Figure S5. XPS images of GO and TiO_2 /RGO composite. (a) C1s image of GO. (b) C1s image of TiO_2 /RGO. (c) Ti2p of TiO_2 /RGO.

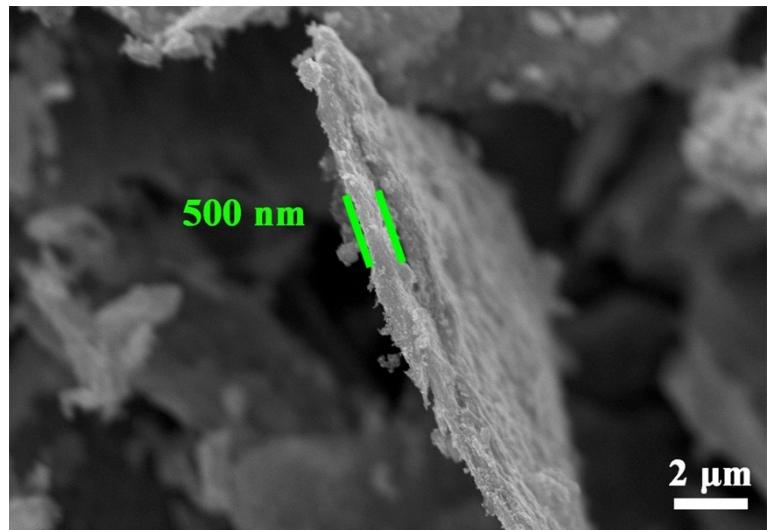


Figure S6. Cross-section SEM of the sandwich sheet-like TiO_2/RGO . The TiO_2/RGO sheets show a tight and unstacked structure, which indicates a good interaction between TiO_2 and RGO. The thickness of TiO_2/RGO sheets is about 500 nm.

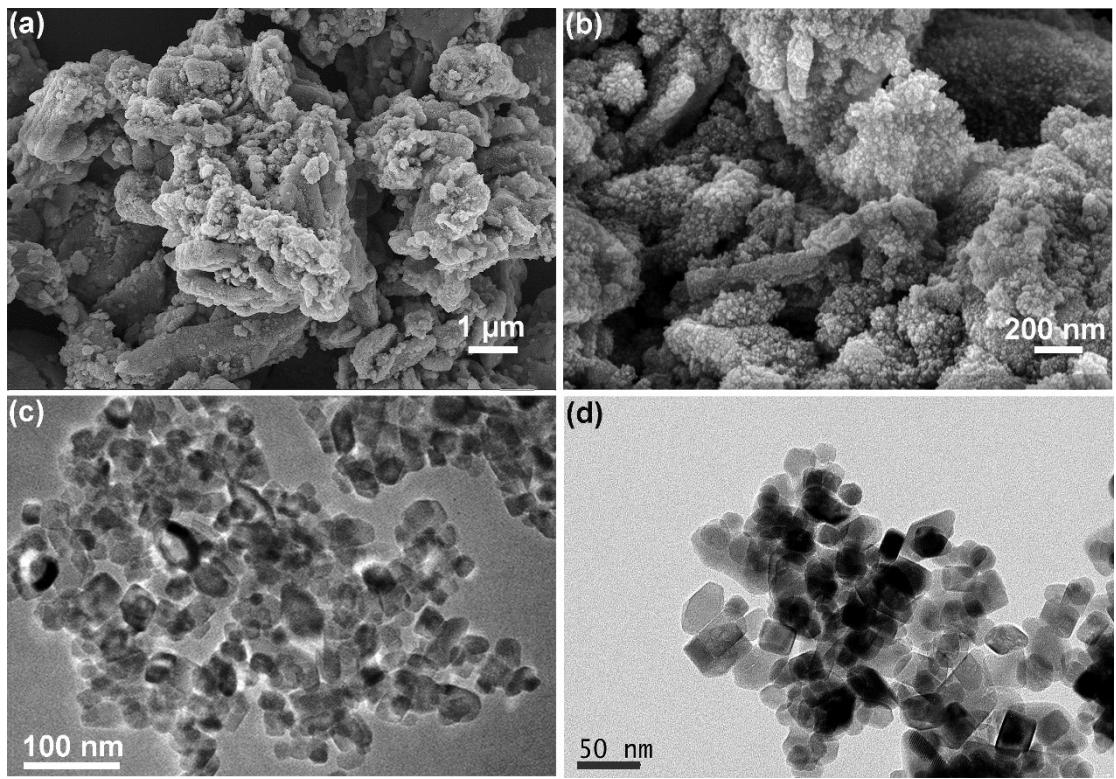


Figure S7. The microtopography of $\text{Ti}_2\text{C}-\text{TiO}_2$. (a), (b) SEM images. (c), (d) TEM images.

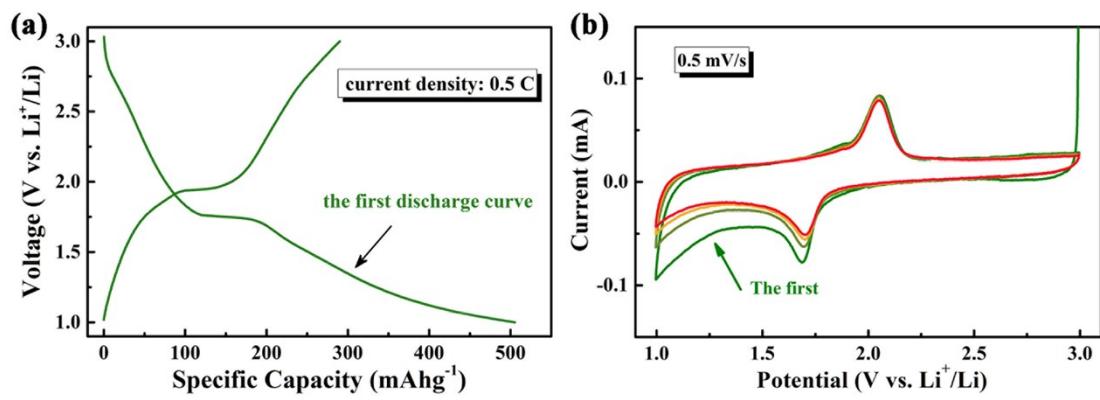


Figure S8. (a) the charge-discharge curves in the first cycle; (b) the initial CV curves.

of TiO₂/RGO.

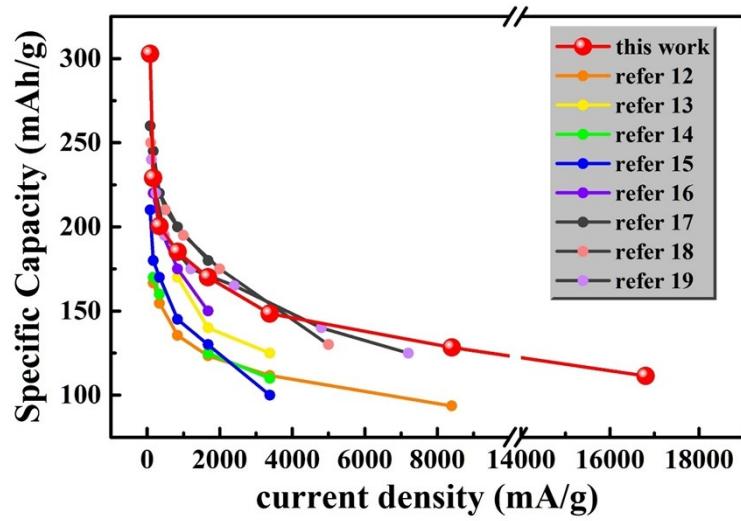


Figure S9. The rate performance comparison of LIBs between TiO₂/RGO and other

TiO₂ anode.

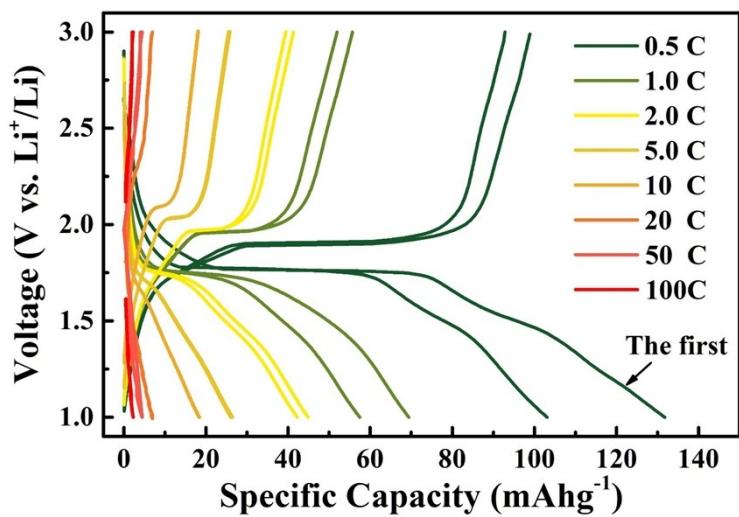


Figure S10. The charge-discharge curves of $\text{Ti}_2\text{C}-\text{TiO}_2$ at different current density

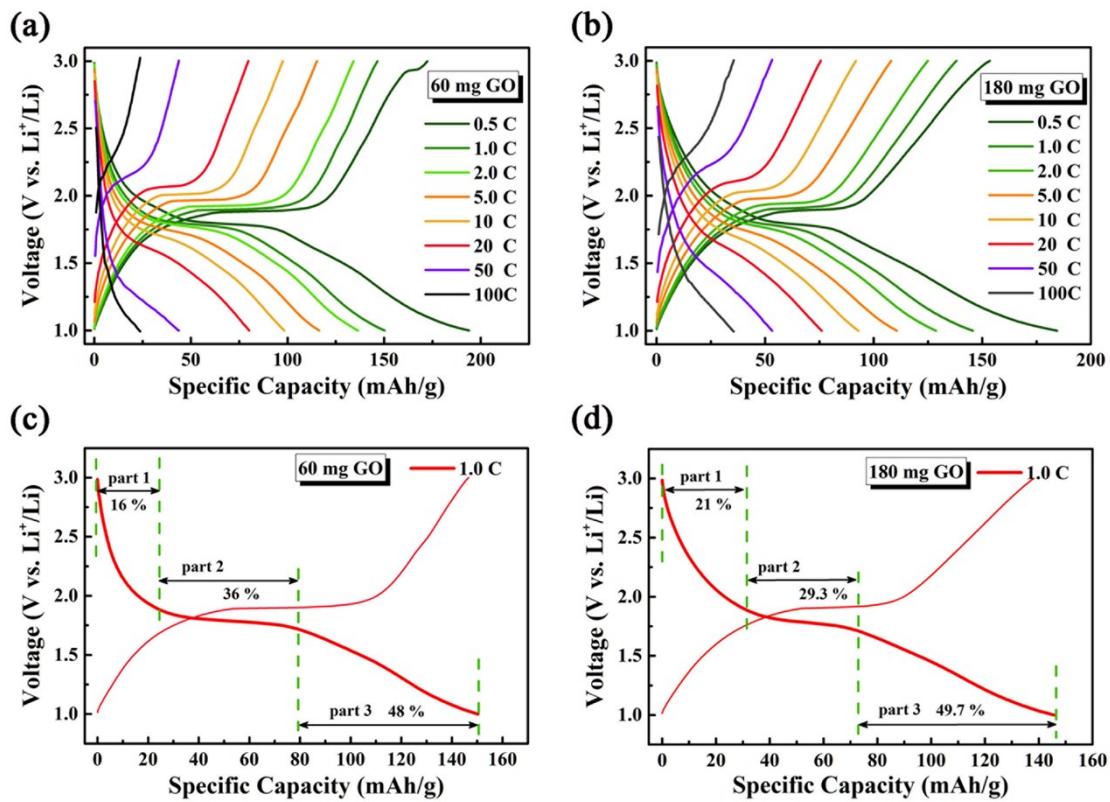


Figure S11. The charge-discharge curves of (a) $\text{TiO}_2/\text{RGO-60}$ and (b) $\text{TiO}_2/\text{RGO-180}$ at different current density; charge-discharge curves of (c) $\text{TiO}_2/\text{RGO-60}$ and (d) $\text{TiO}_2/\text{RGO-180}$ at 1.0 C.

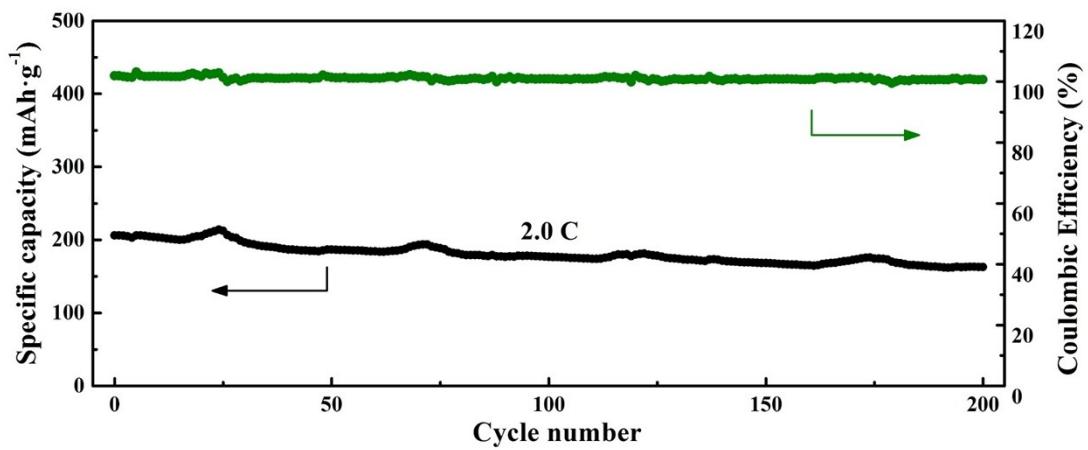


Figure S12. The 200 cycles performance of TiO_2/RGO at 2.0 C.

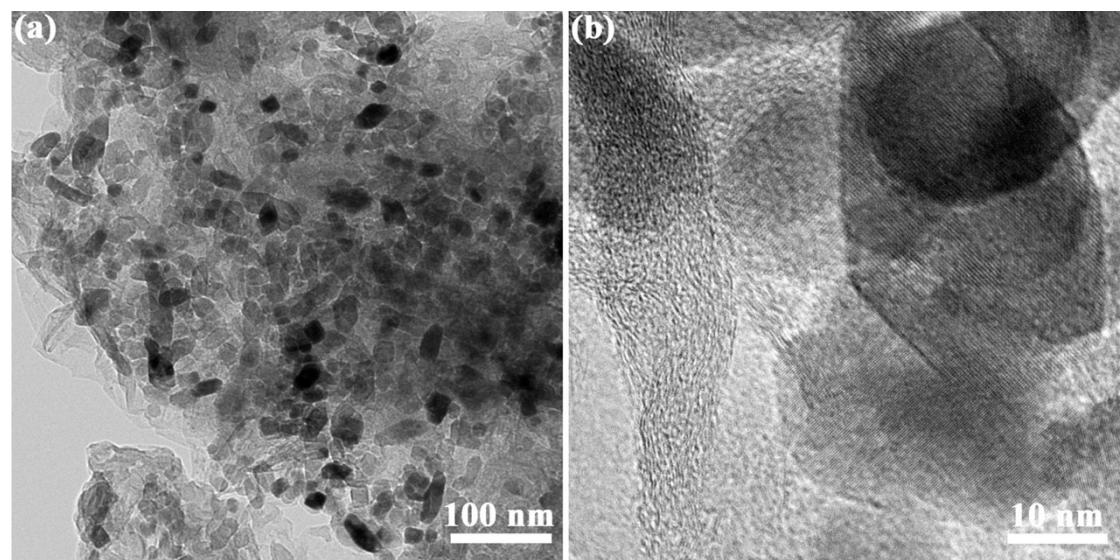


Figure S13. TEM image of TiO_2/RGO after cycling.

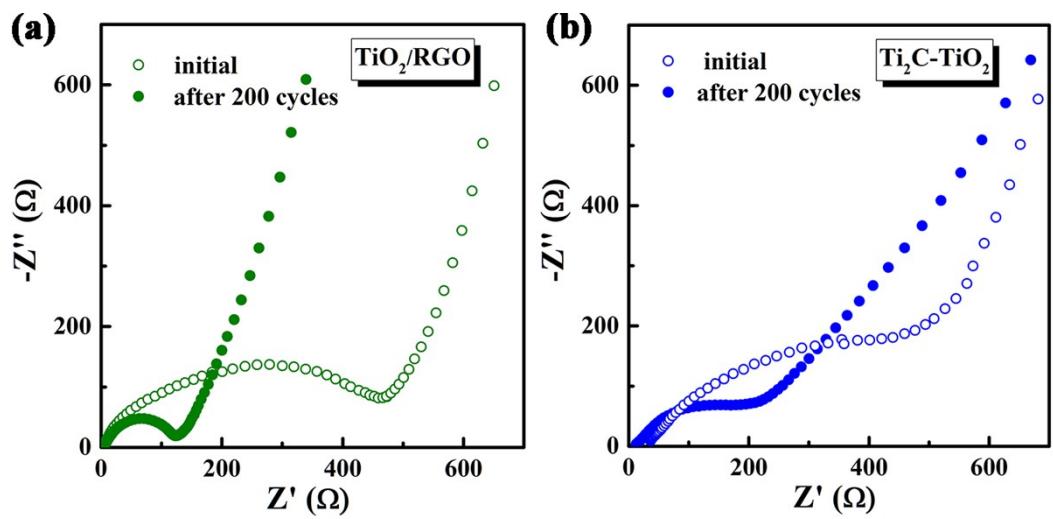


Figure S14. EIS curves at initial and after 200 cycles. (a) TiO_2/RGO (b) $\text{Ti}_2\text{C}-\text{TiO}_2$.

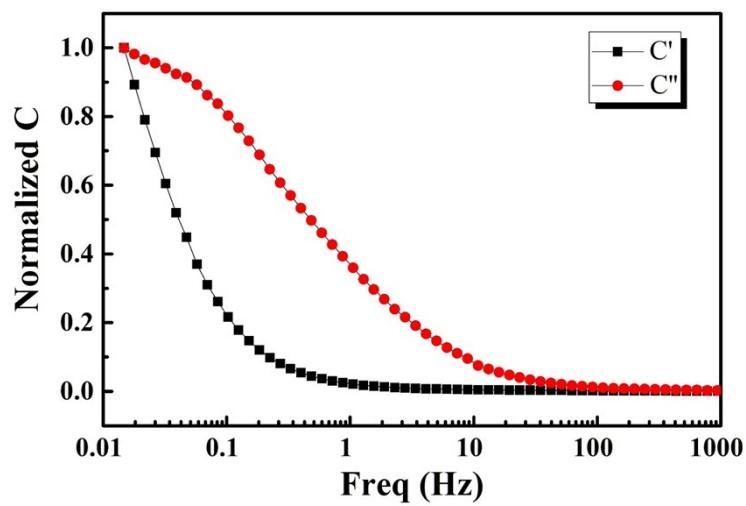


Figure S15. Normalized real and imaginary capacitances of the $\text{Ti}_2\text{C}-\text{TiO}_2$.

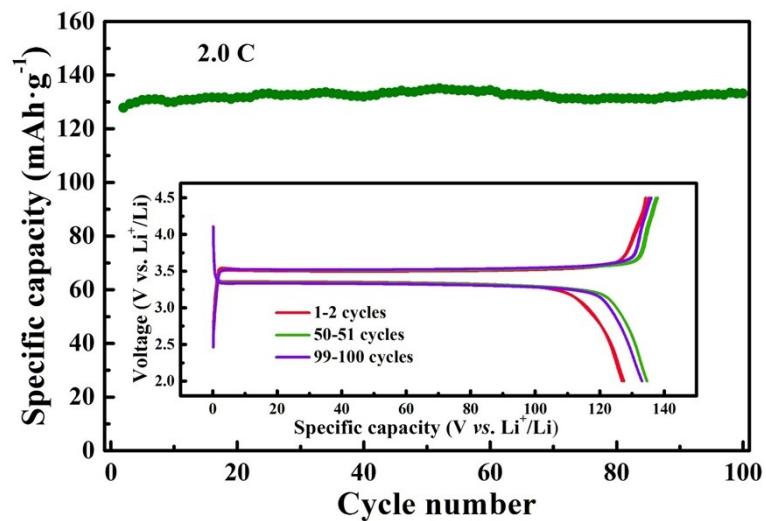


Figure S16. The cycle performance of the LiFePO₄. The charge-discharge curves in the illustration.

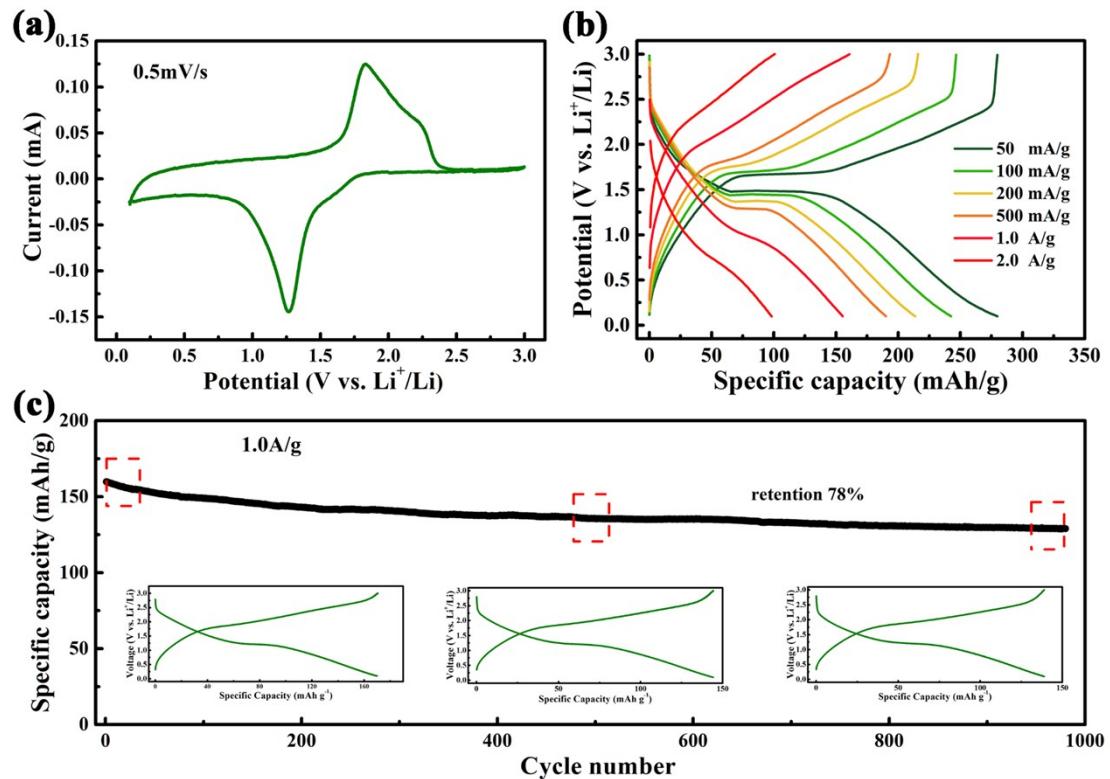


Figure S17. The performance of full cell with LiFePO_4 as cathode and TiO_2/RGO as anode. (a) the CV curve at 0.5 mV/s. (b) The charge-discharge curves at different current density. (c) The 1000 cycles performance at 1.0 A/g, the capacity retention was 78%.

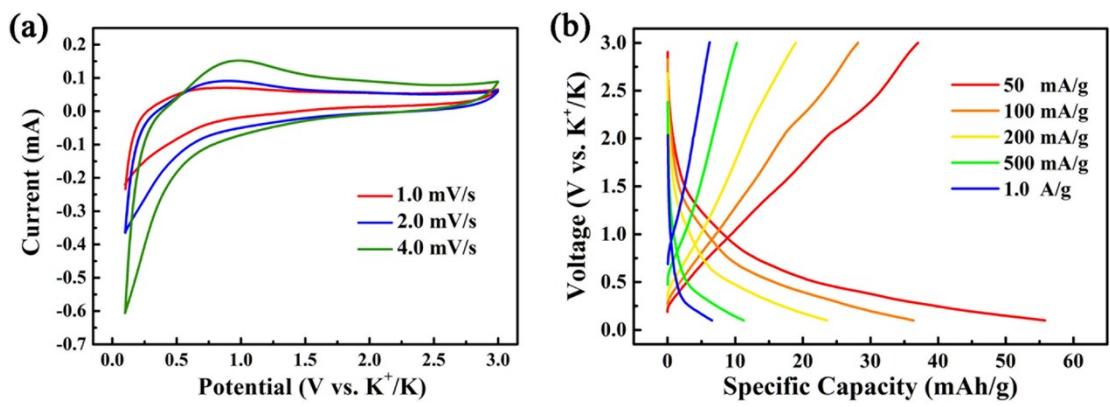


Figure S18. The performance of Ti₂C-TiO₂. (a) The CV curves at different scan rates.

(b) The charge-discharge curves at different current density.

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