

Supporting Information

Selective integration of hierarchical nanostructured energy materials: An effective approach to boost energy storage performance of flexible hybrid supercapacitor

Chandu V.V. Muralee Gopi,^a Prem Jyoti Singh Rana,^{a,b} R. Padma,^c Rajangam Vinodh,^a and Hee-Je Kim^{a*}

^a School of Electrical and Computer Engineering, Pusan National University, Gumjeong-Ku, Jangjeong-Dong, Busan 46241, South Korea.

^b School of Materials Science and Engineering, Nanyang Technological University, Nanyang Avenue 639798, Singapore

^c School of Mechanical Engineering, Yonsei University, Seoul 120-749, South Korea.

***Corresponding Author**

E-mail: heeje@pusan.ac.kr; naga5673@gmail.com

Tel: +82 51 510 2364. Fax: +82 51 513 0212.

Experimental Section

Preparation of NiO@NiO-NiCo₂O₄ NSAs on Cu-Ni fabric:

Typically, 0.1 M NiSO₄·6H₂O and 0.2 M C₆H₁₂N₄ were dissolved in 80 mL DI water by constant stirring. The above solution with NiO-NiCo₂O₄ loaded fabrics were transferred to autoclave and maintained at 100 °C for 2 h. After, the NiO@NiO-NiCo₂O₄ NSAs material loaded electrodes were washed and dried at 80 °C for 12 h.

Preparation of Co₃O₄@NiO-NiCo₂O₄ NSAs on Cu-Ni fabric:

0.1 M CoCl₂·6H₂O and 0.2 M CH₄N₂O were dissolved in a 80 mL of DI water by constant stirring. The above solution with NiO-NiCo₂O₄ loaded fabrics were transferred to autoclave and maintained at 100 °C for 2 h. After cooling down to room temperature, the Co₃O₄@NiO-NiCo₂O₄ NSAs material loaded electrodes were washed and dried at 80 °C for 12 h.

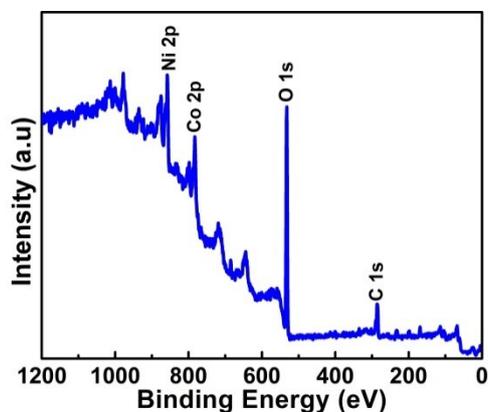


Figure S1. XPS survey spectrum of Ni(OH)₂@NiO-NiCo₂O₄

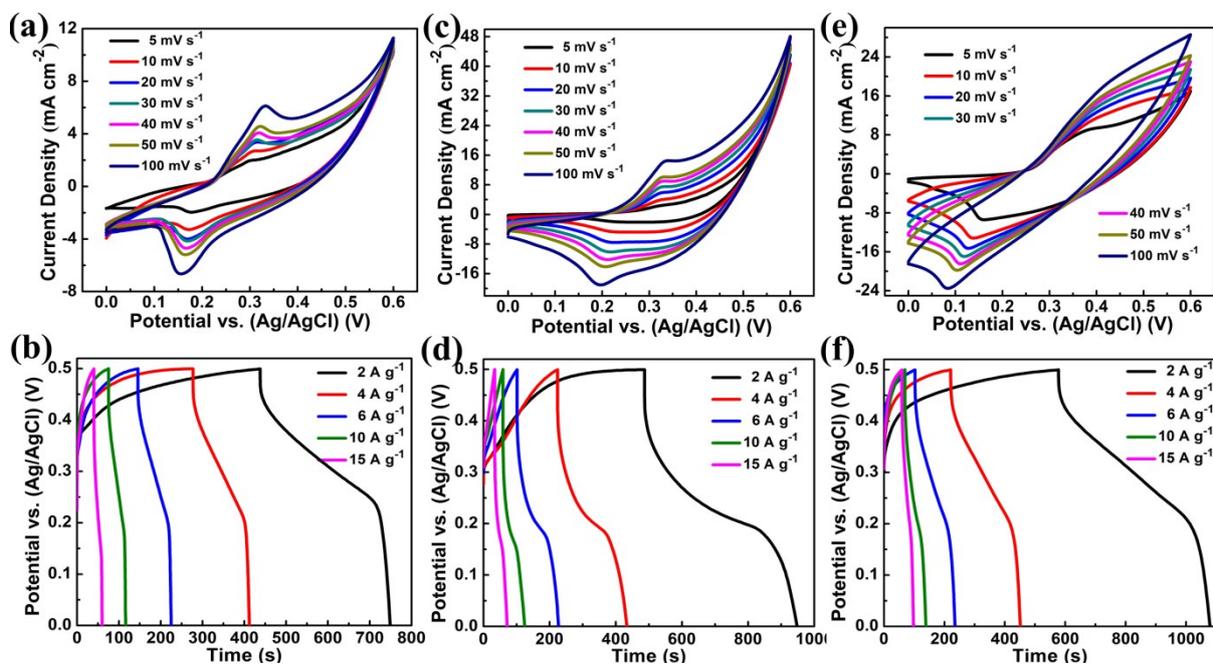


Figure S2. (a,c,e) CV and (b,d,f) GCD curves of (a,b) NiO-NiCo₂O₄ NSAs (c,d) NiO@NiO-NiCo₂O₄ NSAs and (e,f) CO₃O₄@NiO-NiCo₂O₄ NSAs fabric electrodes in 3 M KOH electrolyte solution

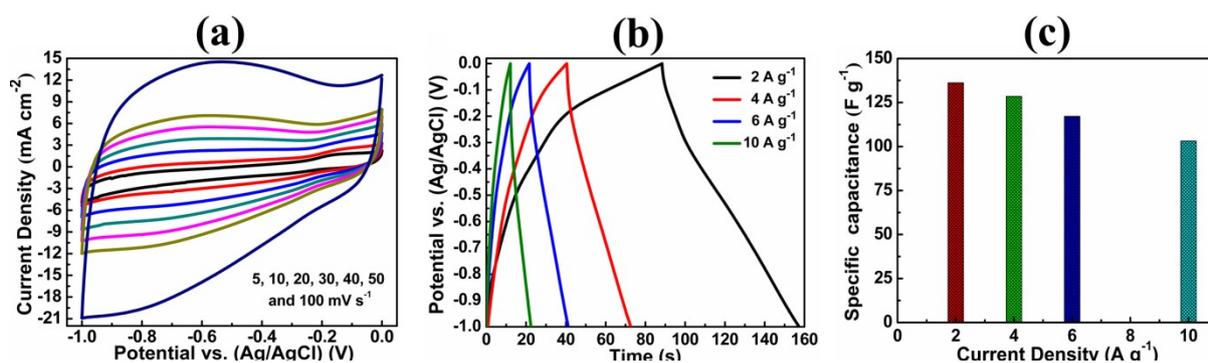


Figure S3. (a) CV and (b) GCD curves investigated at various scan rates and current densities, respectively. (c) Specific capacitance of the graphene-ink (G-ink)@Cu-Ni fabric electrode at different current densities.

Graphene-ink (G-ink) was received from MExplorer Co., Ltd. and it shows nanosheets morphology.^[S1] The mass of the G-ink on Cu-Ni fabric was obtained to be $\sim 3.2 \text{ mg cm}^{-2}$. The electrochemical properties of negative electrode were characterized in a three-electrode system by taking G-ink@Cu-Ni fabric, Ag/AgCl and Pt-wire as working, reference and counter electrodes, respectively, in 3 M KOH aqueous electrolyte. Fig. S3(a) shows the cyclic voltammetry curves (CVs) of G-ink@Cu-Ni fabric electrode measured at different scan rates ranging from 5-100 mV s^{-1} in the potential window of -1.0 to 0.0 V. All the CV plots exhibited symmetrical rectangular shapes, representing the typical EDLC behavior of negative electrode. Moreover, it is clear that all the CV curves maintained their shapes even at higher sweep rates, specifying the rapid charge transfer process between the electrolyte and G-ink electrode. The galvanic charge-discharge (GCD) of G-ink was measured at different current densities from 2 to 10 A g^{-1} in the voltage window of -1.0 to 0.0 V, as shown in Fig. S3(b). The linear and symmetric shapes of GCD curves further revealed the capacitive behavior of G-ink material, which is well consistent with the CV results. The specific capacitance values of G-ink are calculated using Eq. (2) based on GCD curves and the obtained values were 136.24, 128.52, 117.18 and 103.2 F g^{-1} at 2, 4, 6, and 10 A g^{-1} , respectively. The specific capacitance of G-ink was plotted against current density values, as displayed in Fig. S3(c).

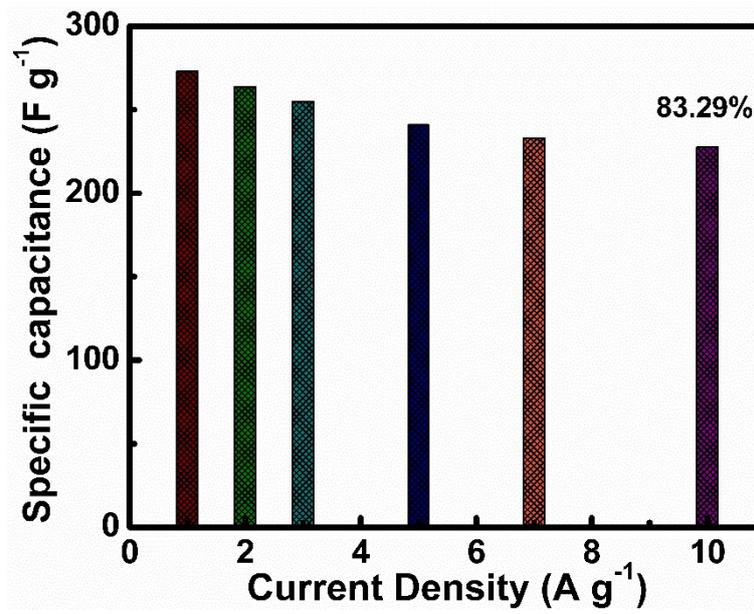


Figure S4. Specific capacitance values of FHSC

References:

[S1] http://www.mexplorer.co.kr/bbs/board.php?bo_table=grp01_eng&wr_id=34