

Supporting information

Stabilized the Layered Manganese Oxide by Cationic ion Substitution Doping

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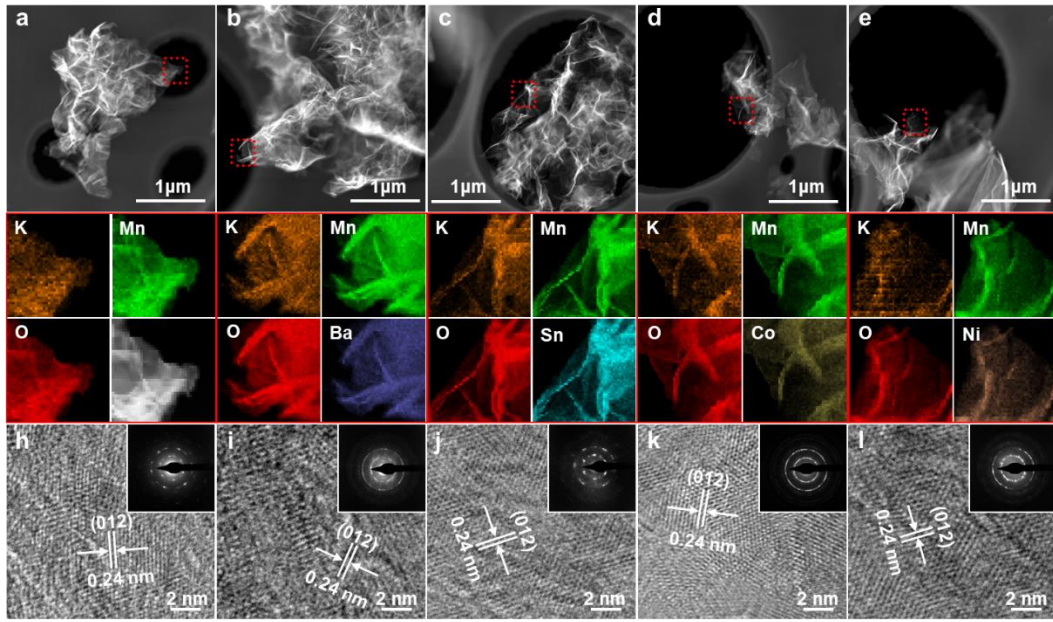


Fig. S1. Morphology and structure of 2D KMnO and 2D M-KMnO for a reaction time of 10 minutes. The HAADF-TEM images and the corresponding mapping images of a) 2D KMnO, b) 2D Ba-KMnO, c) 2D Sn-KMnO, d) 2D Co-KMnO and e) Ni-KMnO. The HR-TEM images of h) 2D KMnO, i) 2D Ba-KMnO, j) 2D Sn-KMnO, k) 2D Co-KMnO and l) 2D Ni-KMnO; insert shows the SAED patterns.

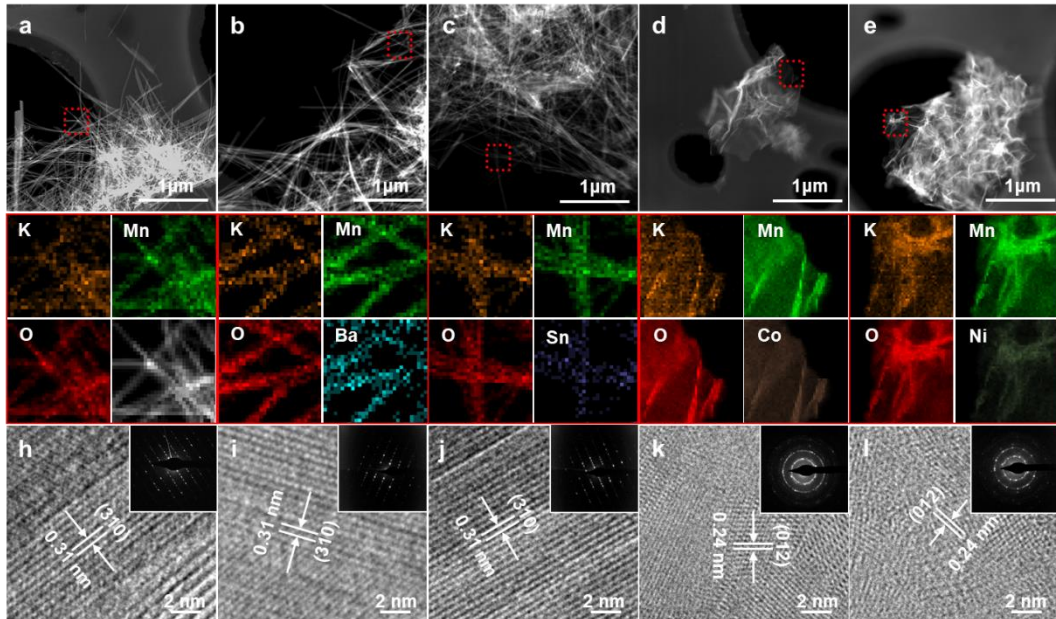


Fig. S2. Morphology and structure of KMnO-30 and M-KMnO-30 for a reaction time of 30 minutes. The HAADF-TEM images and the corresponding mapping images of a) KMnO-30, b) Ba-KMnO-30, c) Sn-KMnO-30, d) Co-KMnO-30 and e) Ni-KMnO-30. The HR-TEM images of h) KMnO-30, i) Ba-KMnO-30, j) Sn-KMnO-30, k) Co-KMnO-30 and l) Ni-KMnO-30; insert shows the SAED patterns.

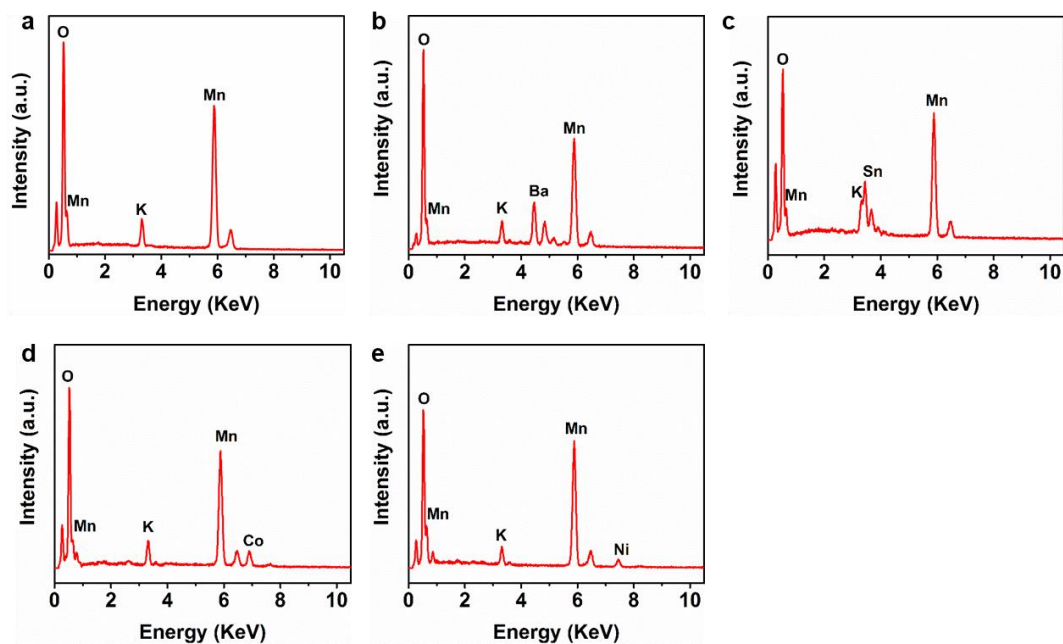


Fig. S3. EDS patterns for a) 2D KMnO, b) 2D Ba-KMnO, c) 2D Sn-KMnO, d) 2D Co-KMnO and e) 2D Ni-KMnO.

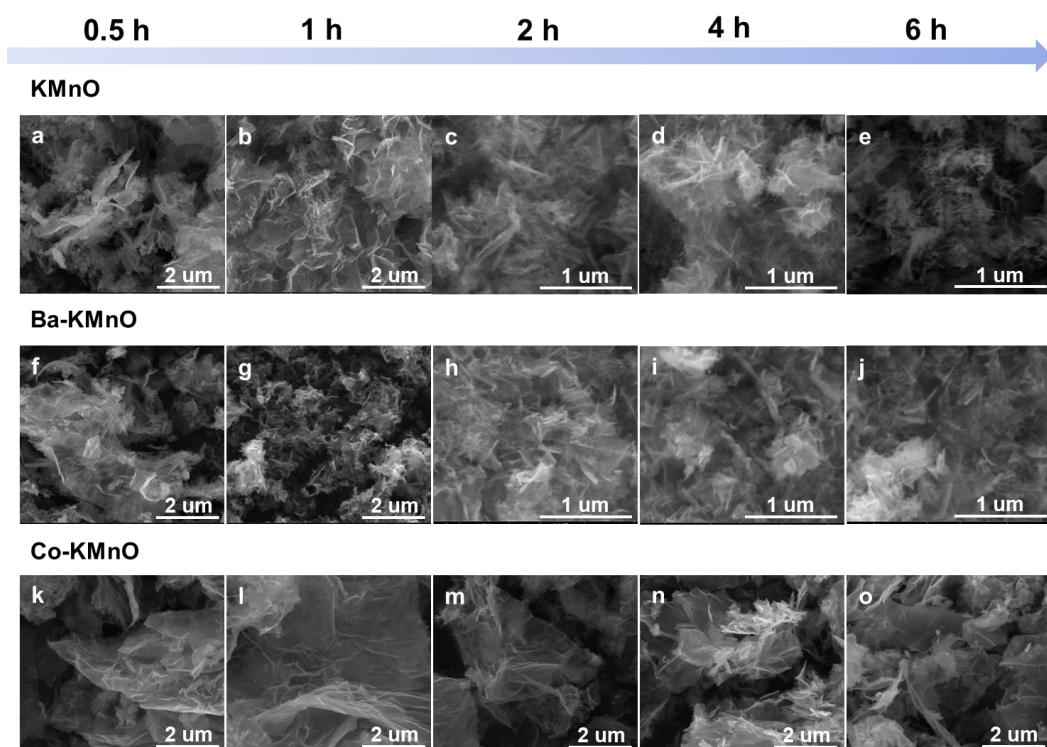


Fig. S4. Morphology conversion of the 2D KMnO, 2D Ba-KMnO and 2D Co-KMnO samples under heat treatment at 400 °C for varying times. a-e) SEM images of 2D KMnO under heat treatment from 0.5 to 6 h. f-j) SEM images of Ba-KMnO under heat treatment from 0.5 to 6 h. k-o) SEM images of 2D Co-KMnO under heat treatment from 0.5 to 6 h.

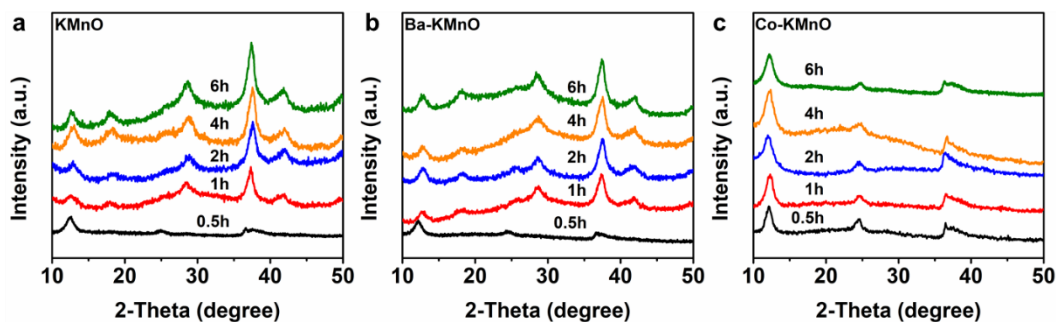


Fig. S5. Structure conversion of 2D KMnO, 2D Ba-KMnO and 2D Co-KMnO samples under heat treatment at 400 °C for varying times. a) XRD patterns for 2D KMnO under heat treatment from 0.5 to 6 h. b) XRD patterns for 2D Ba-KMnO under heat treatment from 0.5 to 6 h. c) XRD patterns for 2D Co-KMnO under heat treatment from 0.5 to 6 h.

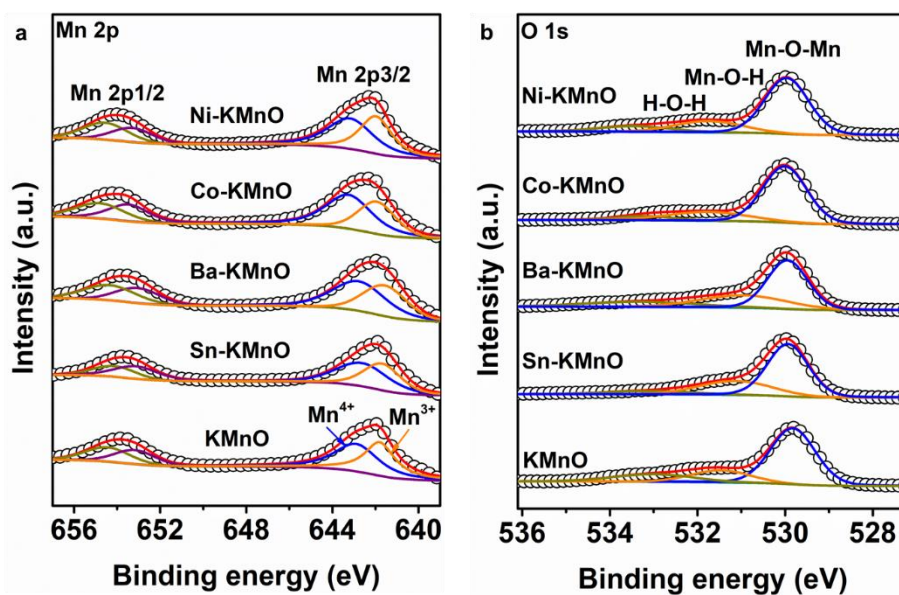


Fig. S6. XPS patterns for 2D KMnO and 2D M-KMnO. a) Mn 2p. b) O 1s.

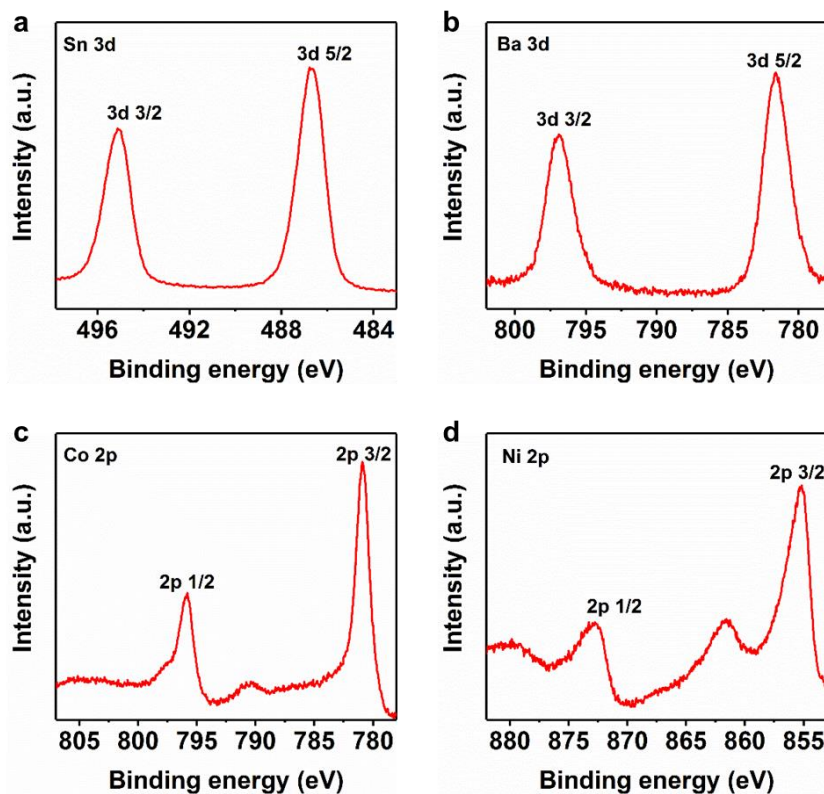


Fig. S7. XPS patterns for a) Sn 3d, b) Ba 3d, c) Co 2p and d) Ni 2p.

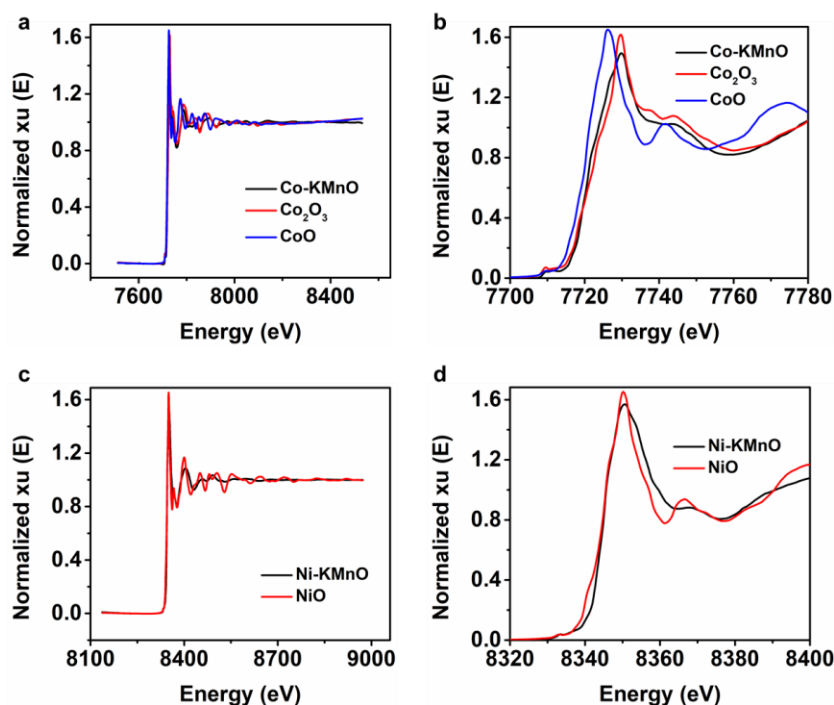


Fig. S8. a) XANES patterns for 2D Co-KMnO, Co_2O_3 and CoO. b) XANES patterns for 2D Co-KMnO, Co_2O_3 and CoO measured in the range of 7704 to 7944 eV. c) XANES patterns for 2D Ni-KMnO and NiO. d) XANES patterns for 2D Ni-KMnO and NiO in the range of 8337 to 8494 eV.

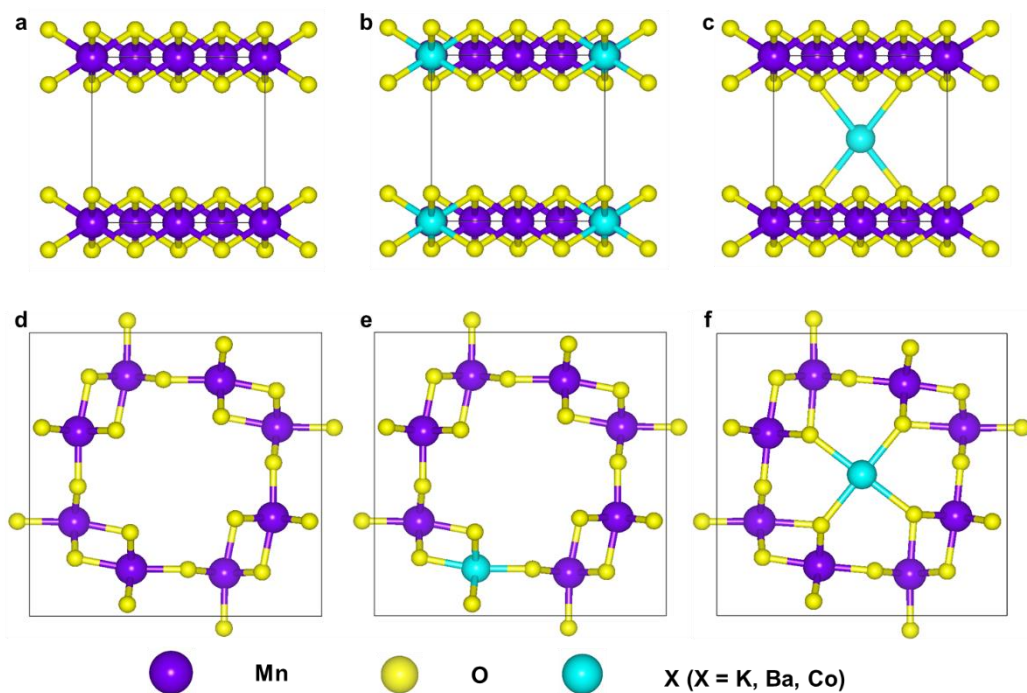


Fig. S9. Calculated Structures. a) pure δ - Mn_8O_{16} ; b) δ - $\text{AMn}_7\text{O}_{16}$ ($A = \text{K}, \text{Ba}, \text{Co}$); c) δ - $\text{AMn}_8\text{O}_{16}$ ($A = \text{K}, \text{Ba}, \text{Co}$); d) pure α - Mn_8O_{16} ; e) α - $\text{AMn}_7\text{O}_{16}$ ($A = \text{K}, \text{Ba}, \text{Co}$) and f) α - Mn_8O_{16} ($A = \text{K}, \text{Ba}, \text{Co}$).

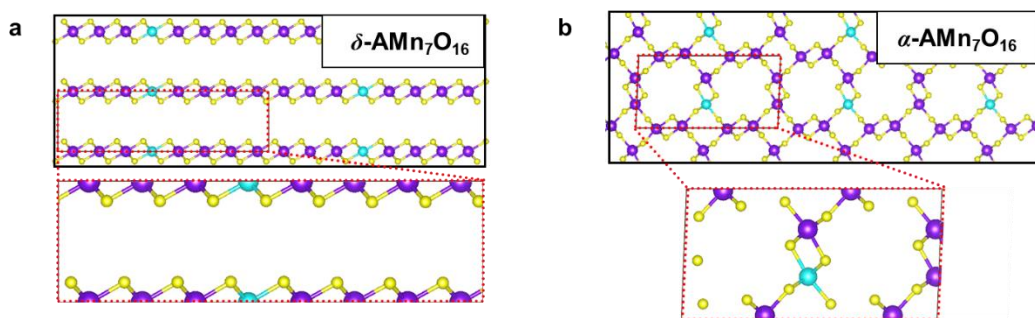


Fig. S10. Snapshots of the key states in the δ - to α - $\text{AMn}_7\text{O}_{16}$ ($A = \text{Mn}, \text{Co}$) transitions. a) δ - $\text{AMn}_7\text{O}_{16}$; b) α - $\text{AMn}_7\text{O}_{16}$. $A = \text{Mn}$ denotes pure Mn_8O_{16} .

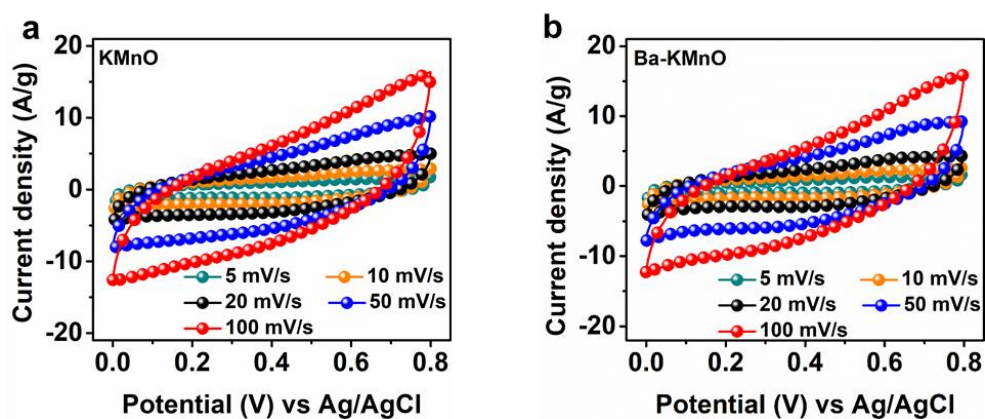


Fig. S11. CV curves for a) 2D KMnO and b) 2D Ba-KMnO at different scan rates ranging from 5 to 100 mV/s.

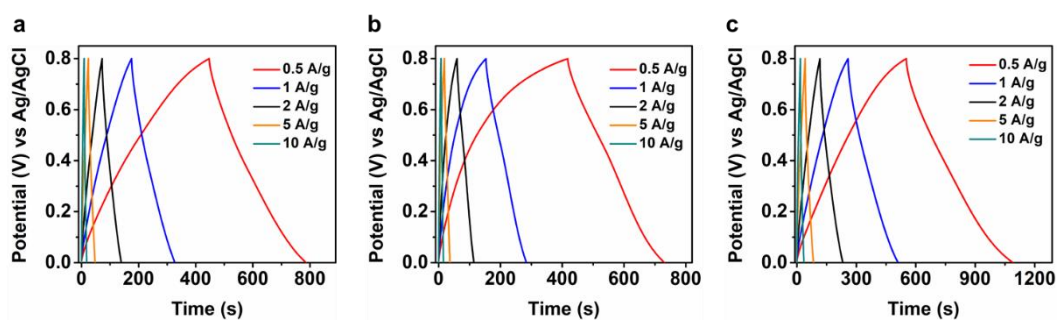


Fig. S12. Charge and discharge curves for a) 2D KMnO, b) 2D Ba-KMnO and c) 2D Co-KMnO with a current density ranging from 0.5 to 10 A/g.

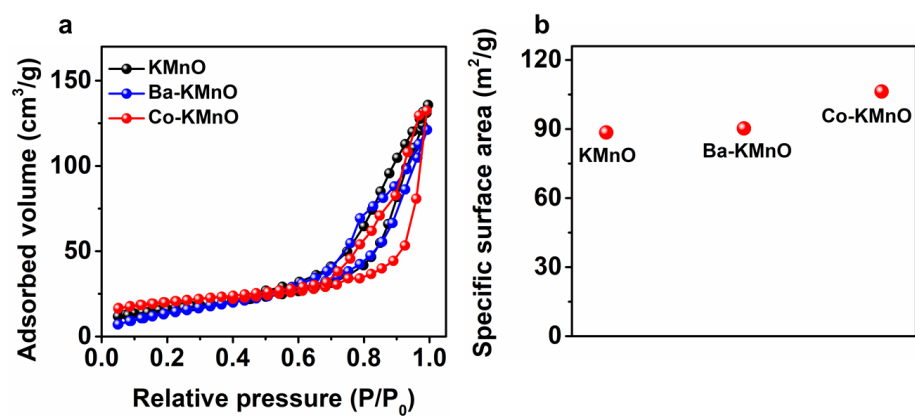


Fig. S13. a) N_2 adsorption-desorption isotherms of samples 2D KMnO, 2D Ba-KMnO and 2D Co-KMnO. b) Specific surface area of 2D KMnO, 2D Ba-KMnO and 2D Co-KMnO.

Table S1. Ionic radius of different ion species.

Ion species	Mn ⁴⁺	Ba ²⁺	Sn ²⁺	Co ³⁺	Ni ²⁺	K ⁺
Ionic radius	54 pm	135 pm	112 pm	54.5 pm	69 pm	138 pm

Table S2. Elemental assay before and after H⁺ exchange from ICP-OES measurements.

Sample	2D KMnO		2D Ba-KMnO		2D Sn-KMnO		2D Co-KMnO		2D Ni-KMnO	
Initial	K	4.12%	K	0.87%	K	0.75%	K	1.86%	K	1.69%
			Ba	1.97%	Sn	1.76%	Co	3.62%	Ni	3.53%
After H ⁺ exchange	K	0.03%	K	0.01%	K	0.02%	K	0.05%	K	0.03%
			Ba	0.02%	Sn	0.04%	Co	2.14%	Ni	1.81%

Table S3. EXAFS fitting parameters

Sample	Shell	N	R (Å)	ΔE ₀ (eV)	σ ² (10 ⁻³ Å ²)	R-factor
KMnO	Mn-O	4.3±0.5	1.90±0.01	-6.3±1.8	4.2±1.3	0.007
	Mn-Mn	4.2±0.9	2.86±0.02	-10.7±2.3	8.3±2.2	
Ba-KMnO	Mn-O	4.2±0.5	1.90±0.01	-6.9±1.8	4.3±1.3	0.007
	Mn-X	4.3±0.9	2.86±0.02	-10.8±2.2	8.4±2.1	
Co-KMnO	Mn-O	4.2±0.6	1.90±0.01	-8.8±2.1	3.5±1.4	0.008
	Mn-X	4.2±0.8	2.84±0.01	-13.6±2.0	5.4±1.6	
	Co-O	4.3±0.4	1.92±0.01	-0.1±1.5	3.7±1.2	0.005
	Co-X	4.4±0.6	2.85±0.01	-0.6±1.5	4.3±1.1	

X presents Mn, Ba or Co; N is coordination numbers; R is the internal atomic distance; σ² is Debye-Waller factor; ΔE₀ is the edge-energy shift.