

Supporting Information

Polyelectrolyte-Based Physical Adhesive Hydrogels with Excellent Mechanical Strength for Biomedical Applications

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1.

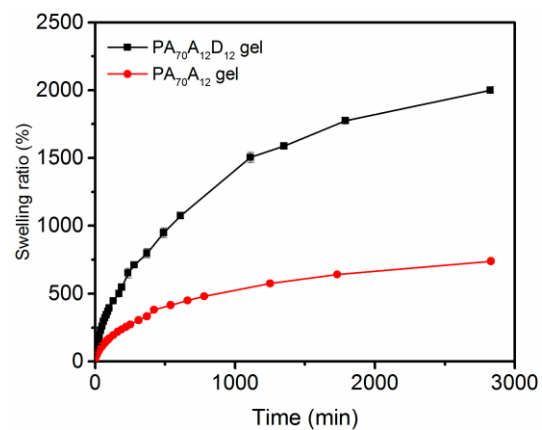


Figure S1. The swelling ratio of hydrogels at room temperature. Each data point in the average of 3 measurements, with the error bars showing the standard deviation.

2.

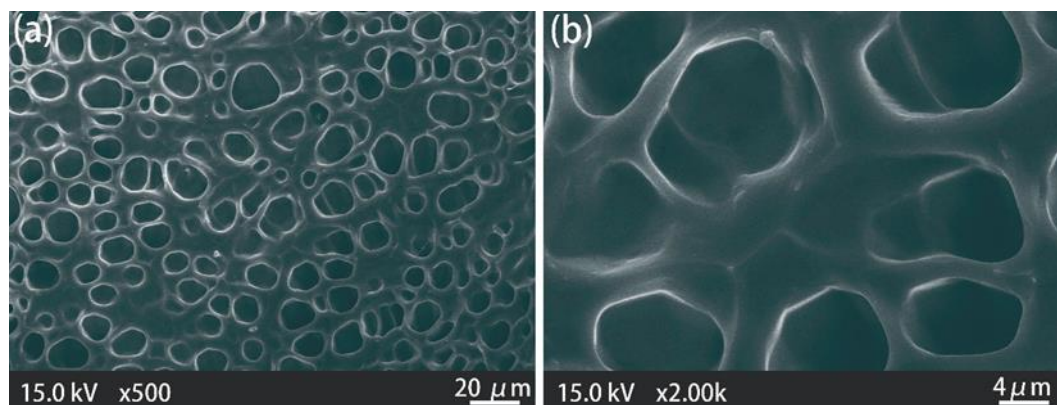


Figure S2. SEM images of the PA₇₀A₁₂D₁₂ hydrogel.

3.

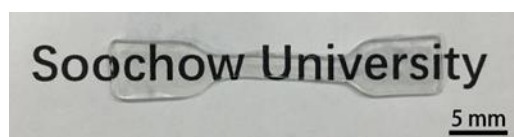


Figure S3. Photograph of a dumbbell shaped the PA₇₀A₁₂D₁₂ hydrogel (GB/T 1040-1992).

4.



Figure S4. Photographs of the PA₇₀A_yD_z hydrogel with various content of ionic pairs.

5.

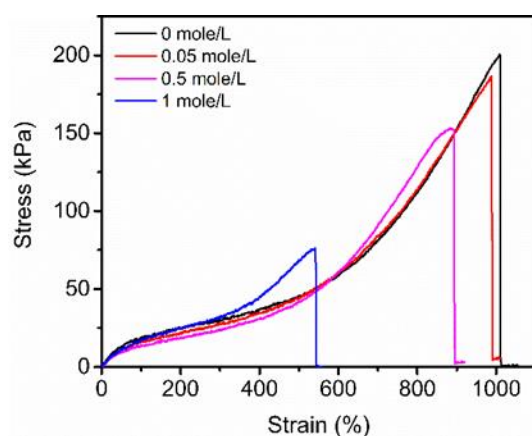


Figure S5. The tensile stress-strain curves of PA₇₀A₁₂D₁₂ hydrogels which were prepared through the precursor solution with various ionic strength.

6.

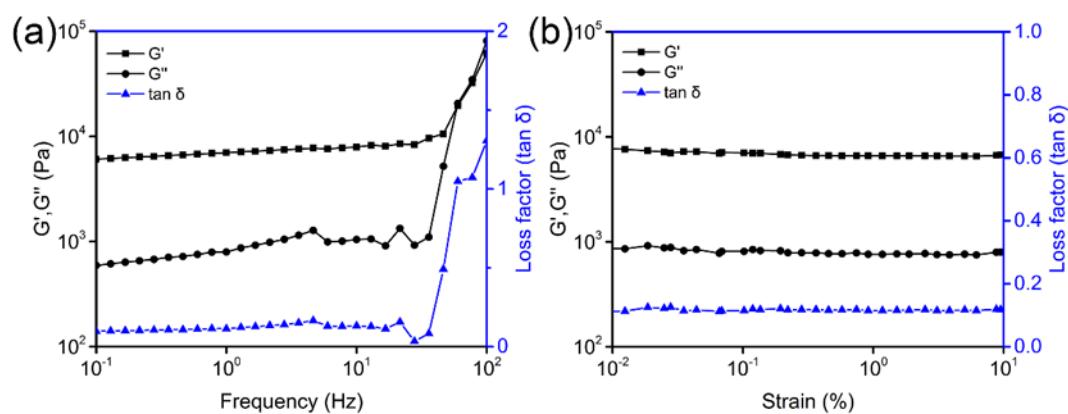


Figure S6. Dynamic mechanical properties of the PA₇₀A₁₂D₁₂ hydrogel. (a) Frequency-sweep and (b) strain-sweep measurements.

7.

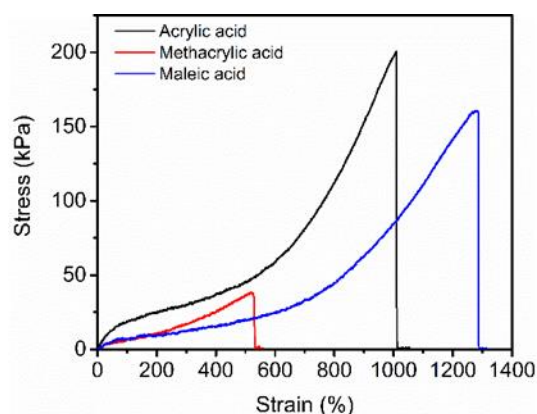


Figure S7. Stress-strain curves of hydrogels with different anionic reactive monomers, such as poly(AAm-*co*-methacrylic acid)/PDADMAC and poly(AAm-*co*-maleic acid)/PDADMAC hydrogels. Herein, both hydrogels were synthesized by using the same method as PA_xA_yD_z hydrogels. Briefly, 12 mmole of methacrylic acid (or 6 mmole of maleic acid) was added to replace AAc (12 mmole).

8.

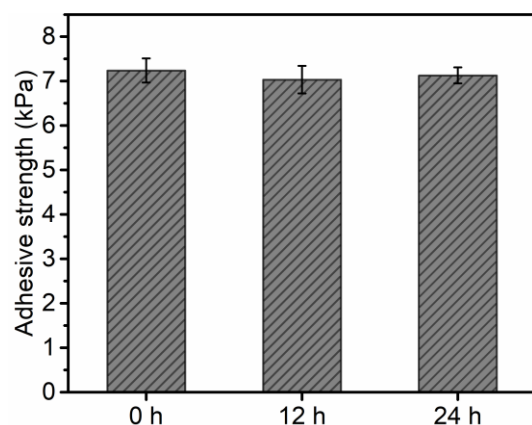


Figure S8. The adhesive strength of the hydrogel on PE substrates after different storing time. Each data point is the average of 3 measurements, with the error bars showing the standard deviation.

9.

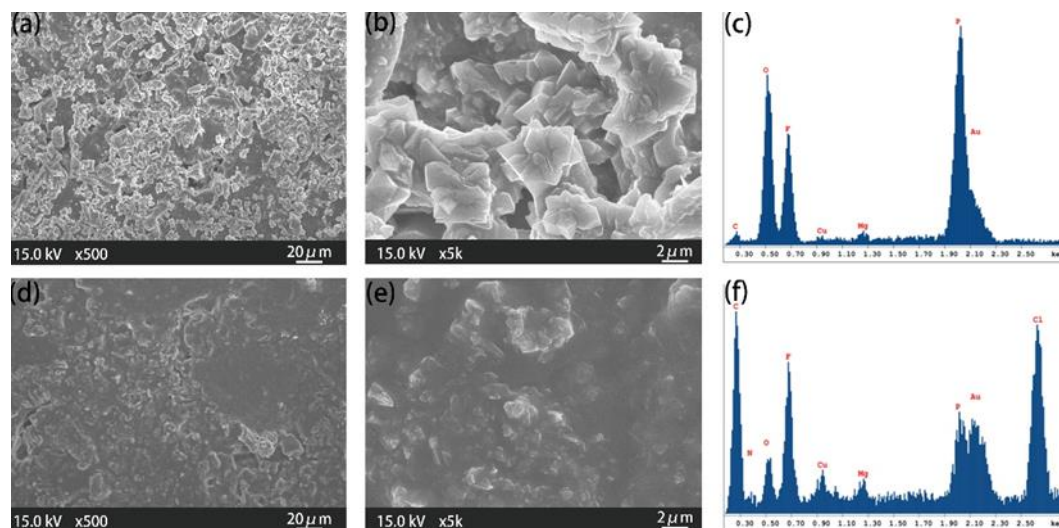


Figure S9. SEM EDX images and spectra of (a-c) pristine TLDs and (d-f) TLD-hydrogel hybrids.

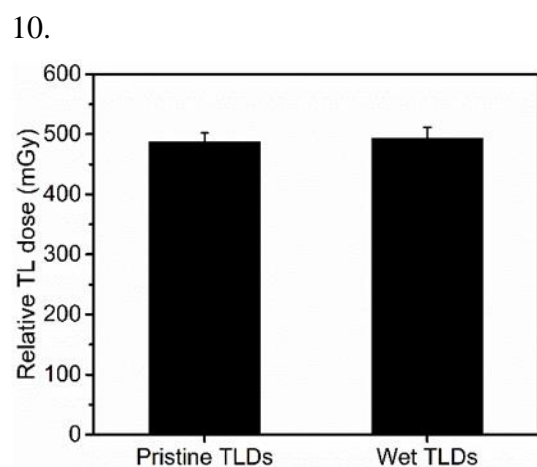


Figure S10. The effect of water on the relative TL doses for samples. Wet TLDs were prepared by immersing TLDs in DI water prior to the exposure to X-ray radiation. Each data point is the average of 3 measurements, with the error bars showing the standard deviation.

11.

When the incident angle is $\sim 87^\circ$, the penetration of X-rays ($d_{X\text{-ray}}$) is ~ 2.5 mm via a simple geometric calculation. According to Figure S10, $D_{r(c3/c1)}$ should have detected ~ 328.691 mGy

(91.7% of $D_{r(c2)}$) if all TLDs have been located at the same SSD, which is higher than our detection. As a result, the lower the magnitudes of $D_{r(c1)}$ and $D_{r(c3)}$ is caused by the reduced SSD and increased incident angle in this case.

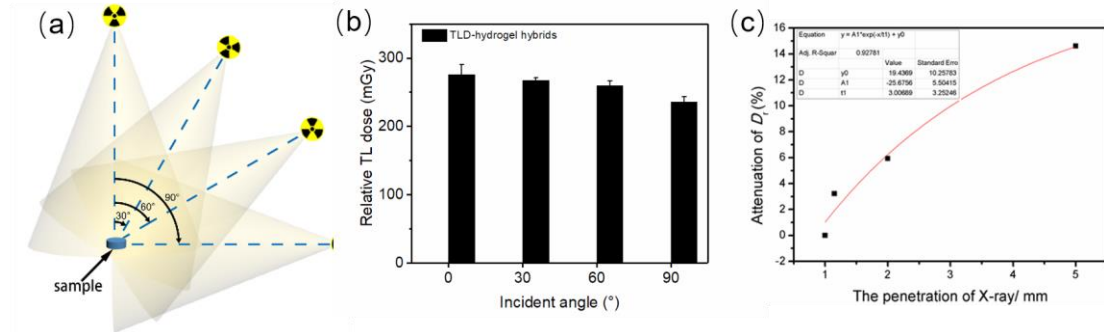


Figure S11. The relative TL dose of samples vary as a function of (a, b) incident angles at 100 mGy min^{-1} under X-ray. Accumulated dose was 1000 mGy . Each data point in the average of > 3 measurements, with the error bars showing the standard deviation. (c) The attenuation of D_f as a function of the penetration depth of X-rays. The penetration of X-rays was depended on the incident angles of X-rays and the geometry size of the TLDs.

12.

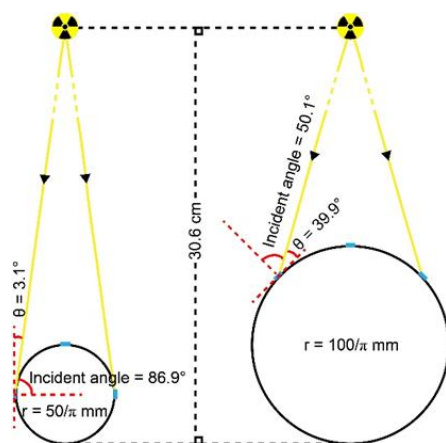


Figure S12. The geometry of TLD (shown in blue)-hydrogel hybrids deposited on curved surfaces with different radii.

Table S1. Mechanical properties and structural parameters of PA_xA₁₂D₁₂ hydrogels.

hydrogels	E (kPa)	σ_f (kPa)	ϵ_f (%)	U (kJ m ⁻³)	τ (kPa)	N^* (mol m ⁻³)
PA ₄₀ A ₁₂ D ₁₂	3.8±0.6	47.6±8.9	654±23	100.0±11.2	7.8±0.6	1.8
PA ₅₀ A ₁₂ D ₁₂	9.6±1.9	113.3±16.4	817±16	285.9±28.8	12.7±1.3	2.9
PA ₆₀ A ₁₂ D ₁₂	19.9±2.1	174.2±12.8	900±27	489.5±24.3	17.1±1.0	3.9
PA ₇₀ A ₁₂ D ₁₂	26.9±2.7	200.2±14.4	1013±54	679.4±31.2	25±3.7	5.7
PA ₈₀ A ₁₂ D ₁₂	36.5±3.1	214.8±21.4	934±35	705.2±33.0	28.8±2.1	6.6
PA ₉₀ A ₁₂ D ₁₂	44.1±4.1	232.4±19.4	845±11	671.1±25.1	33.8±1.9	7.8

Table S2. Mechanical properties and structural parameters of PA₇₀A_yD_z hydrogels with varying AAc concentration (y=z).

PDADMAC	E (kPa)	σ_f (kPa)	ϵ_f (%)	U (kJ m ⁻³)	τ (kPa)	N^* (mol·m ⁻³)
2	11.1±2.6	29.6±4.3	2120±62	347.5±24.2	6.2±1.0	1.4
7	21.4±1.9	157.9±11.2	1380±45	732.2±28.1	13.2±1.7	3
12	26.9±2	200.2±14.4	1013±54	679.4±31.2	18.6±2.1	4.3
17	39±2.2	150±17.1	704±48	395.7±19.2	21.6±1.5	5.3
22	43.3±3.1	128.0±9.5	623±29	340.9±24.1	25.0±2.7	5.7

Table S3. Mechanical properties and structural parameters of PA₇₀A_yD_z hydrogels with varying mole ratio of anions to cations (y:z).

y:z	E (kPa)	σ_f (kPa)	ϵ_f (%)	U (kJ m ⁻³)	τ (kPa)	N^* (mol m ⁻³)
1:9	54.4±7.9	435±34	14.9±2.8	114.0±37.1	22.4±1.2	5.1
3:7	135.1±12.4	713±42	20.1±2.4	325.7±28.9	22.6±0.8	5.2
5:5	200.2±14.4	1013±54	26.9±2	679.4±31.2	25±1.6	5.7
7:3	182.6±16.8	1097±27	23.7±1.9	643.1±42.1	17.8±2.0	4.1
9:1	60.6±9.7	1839±82	22.1±0.7	433.0±33.5	10.3±1.5	2.4