

Electronic Supplementary Information

# All-inorganic $\text{CsPbBr}_3$ perovskite quantum dots as photoluminescent probe for ultrasensitive $\text{Cu}^{2+}$ detection

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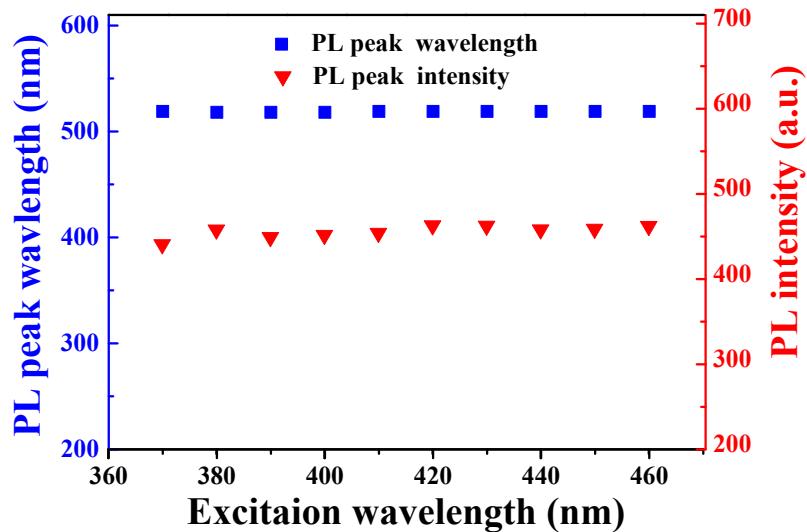
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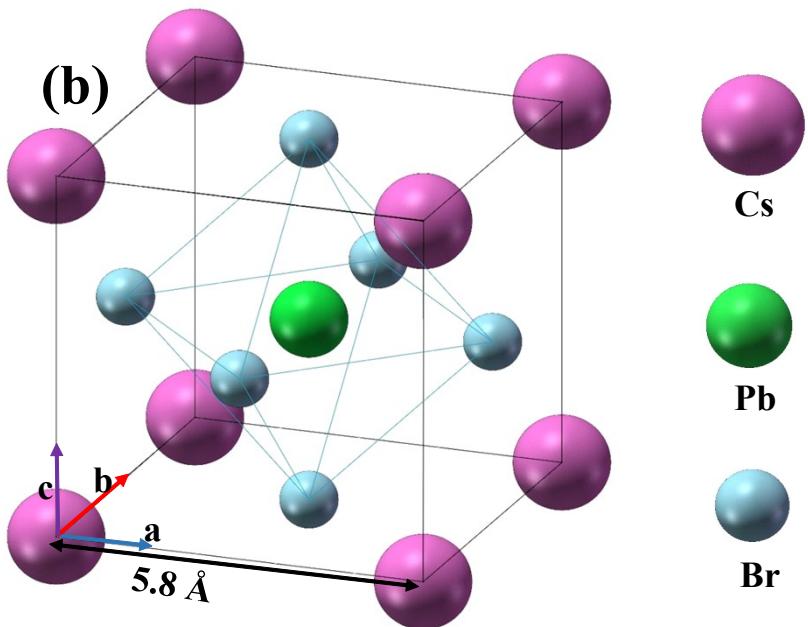
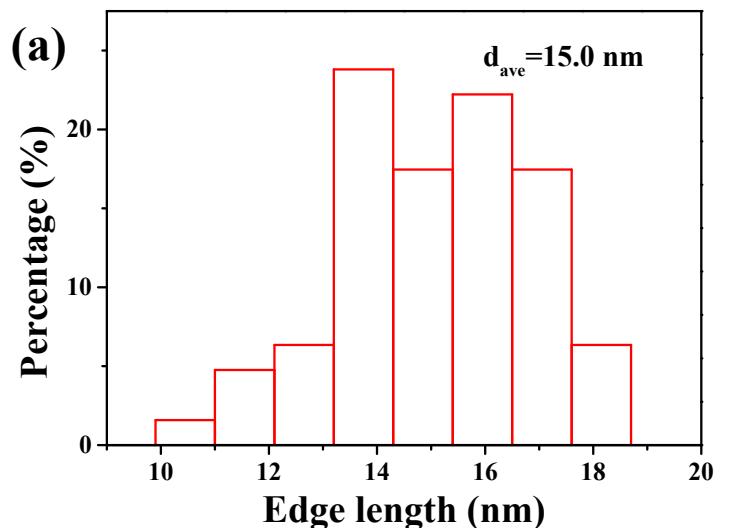
## Supporting Figures and Tables

**Table S1** Absorption peak, PL peak, Stokes shift, FWHM, and PL QY of  $\text{CsPbBr}_3$  perovskites.

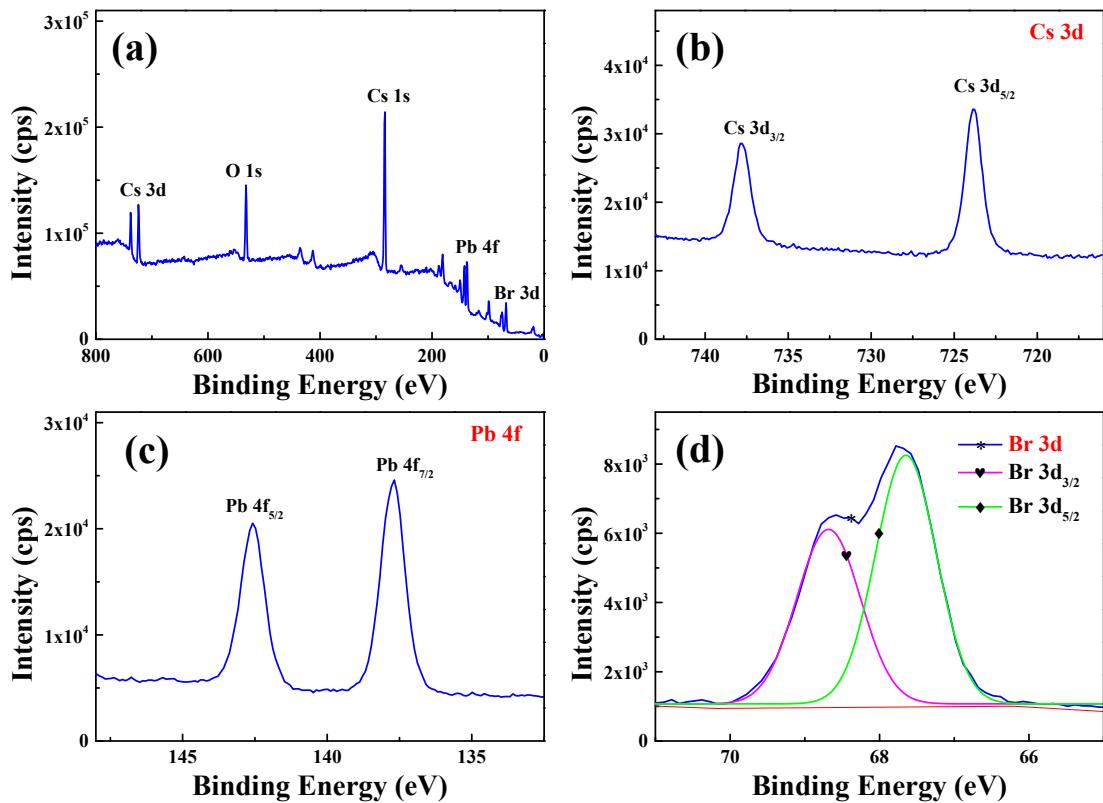
| QDs               | Abs peak (nm)/E1 (eV) | PL peak (nm)/E2 (eV) | Stokes shift (nm) | Stokes shift (meV) | FWHM (nm) | QY (%) |
|-------------------|-----------------------|----------------------|-------------------|--------------------|-----------|--------|
| $\text{CsPbBr}_3$ | 508.11/2.44           | 518.05/2.39          | 9.94              | 50                 | 16        | 90.12  |



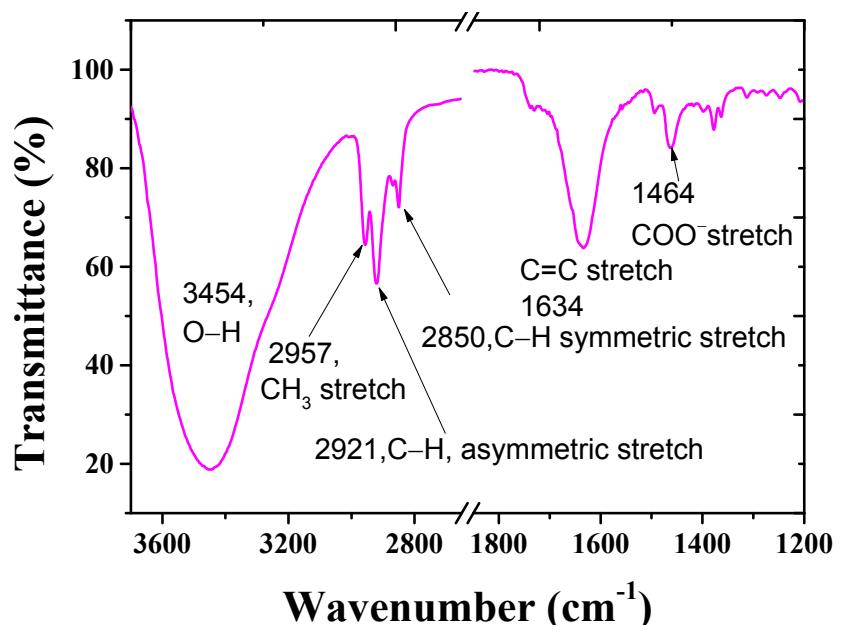
**Figure S1** PL peak wavelength and intensity dependence of  $\text{CsPbBr}_3$  on the excitation wavelength.



**Figure S2** (a) Size distribution and (b) the typical cubic crystal structure diagram of  $\text{CsPbBr}_3$ .



**Figure S3** XPS spectra of CsPbBr<sub>3</sub> PQDs. (a) Survey XPS and high-resolution XPS of (b) Cs 3d, (c) Pb 4f, and (d) Br 3d.



**Figure S4** FTIR spectrum of  $\text{CsPbBr}_3$  QDs.

**Table S2** Fitting parameters of fluorescence decay curves.

| Parameters<br>QDs \                | $\tau_1$<br>(ns) | $\tau_2$<br>(ns) | $\tau_3$<br>(ns) | $A_1$<br>(%) | $A_2$<br>(%) | $A_3$<br>(%) | $\tau_{ave}$<br>(ns) | $\chi^2$ |
|------------------------------------|------------------|------------------|------------------|--------------|--------------|--------------|----------------------|----------|
| CPB QDs in 0 nM Cu <sup>2+</sup>   | 4.54             | 20.24            | 99.77            | 19.48        | 46.65        | 33.87        | 80.84                | 1.233    |
| CPB QDs in 50 nM Cu <sup>2+</sup>  | 3.95             | 17.36            | 88.28            | 19.52        | 46.63        | 33.85        | 71.83                | 1.161    |
| CPB QDs in 100 nM Cu <sup>2+</sup> | 3.48             | 15.64            | 78.39            | 23.53        | 48.15        | 28.31        | 60.92                | 1.193    |

**Table S3** Average lifetime of the CPB NCs in the presence of metal ions.

| Metal ions       | no    | In <sup>3+</sup> | Ag <sup>+</sup> | Cd <sup>2+</sup> | Fe <sup>3+</sup> | Hg <sup>2+</sup> | Mg <sup>2+</sup> | Mn <sup>2+</sup> | Na <sup>+</sup> | Pb <sup>2+</sup> | Zn <sup>2+</sup> | Ni <sup>2+</sup> | Cu <sup>2+</sup> |
|------------------|-------|------------------|-----------------|------------------|------------------|------------------|------------------|------------------|-----------------|------------------|------------------|------------------|------------------|
| $\tau_{av}$ (ns) | 80.84 | 78.01            | 80.82           | 77.44            | 80.03            | 80.44            | 78.33            | 75.67            | 78.01           | 78.90            | 80.51            | 80.03            | 60.92            |