

Supporting Information

Cu/Cu₂O nanocomposite as a p-type transparent-conductive-oxide for efficient bifacial-illuminated perovskite solar cells

You-Jyun Chen^a, Ming-Hsien Li^a, Jung-Chun-Andrew Huang^{a,b,*}, and Peter Chen^{a,c,*}

^aDepartment of Photonics, National Cheng Kung University, Tainan 701, Taiwan

^bDepartment of Physics, National Cheng Kung University, Tainan 701, Taiwan

^cHierarchical Green-Energy Materials (Hi-GEM) Research Center, National Cheng Kung University, Tainan, 701, Taiwan

Corresponding authors.

E-mail addresses: jcahuang@mail.ncku.edu.tw (J.-C. Huang),
petercyc@mail.ncku.edu.tw (P. Chen).

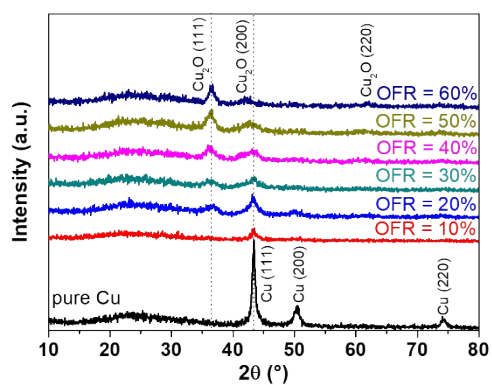
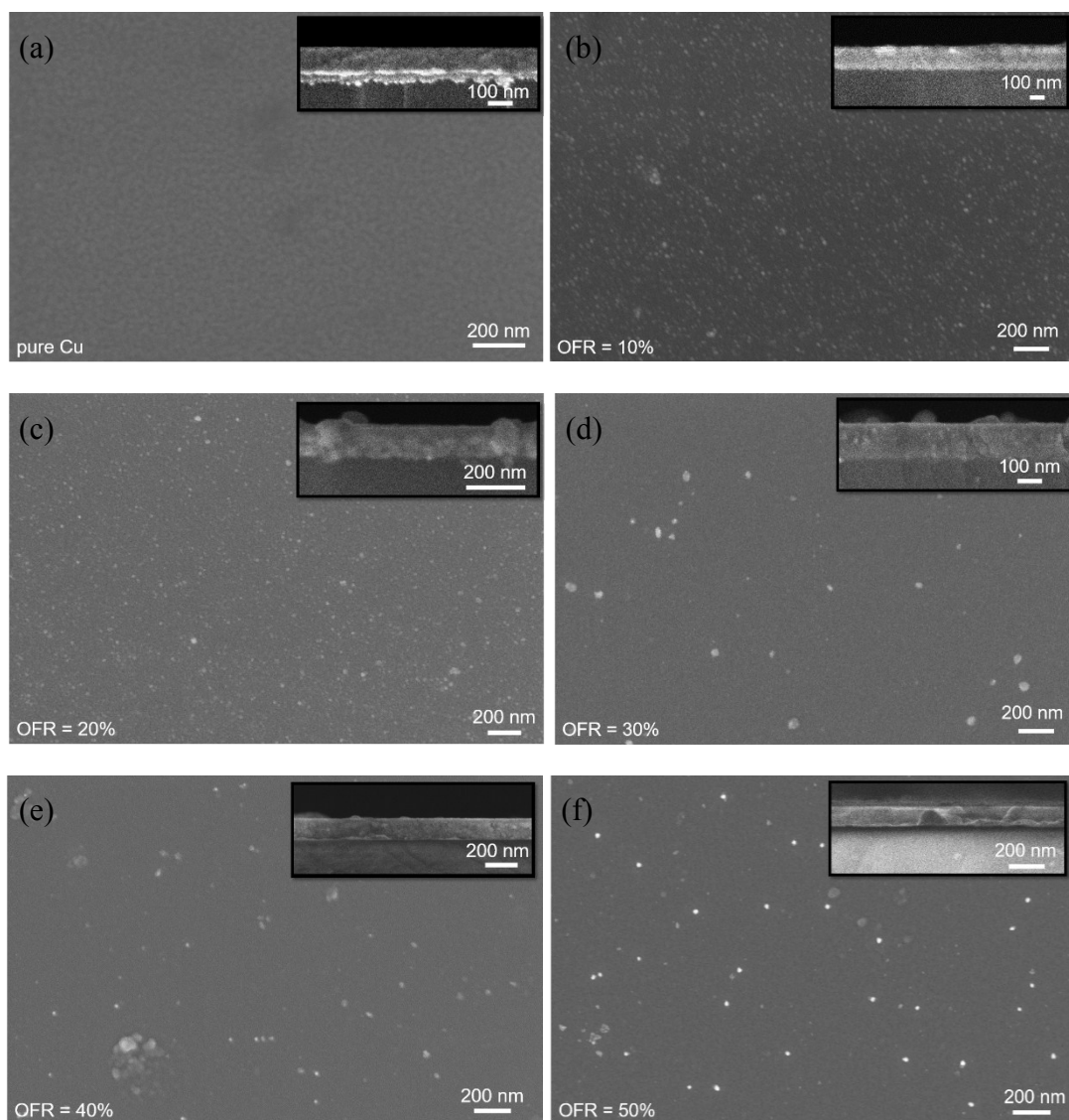


Figure S1. X-ray diffraction spectra of the Cu_2O films deposited at different OFRs from 10 to 60%. The pure Cu (OFR = 0%) film is also shown for comparison.[1]



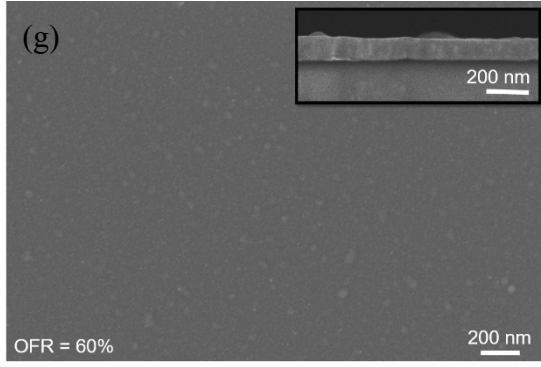


Figure S2. SEM images of the top-view of Cu/Cu₂O composite films with different oxygen flow ratios (a) 0%, (b) 10%, (c) 20%, (d) 30%, (e) 40%, (f) 50%, and (g) 60%. Insets show the corresponding cross-sectional SEM images.[1]

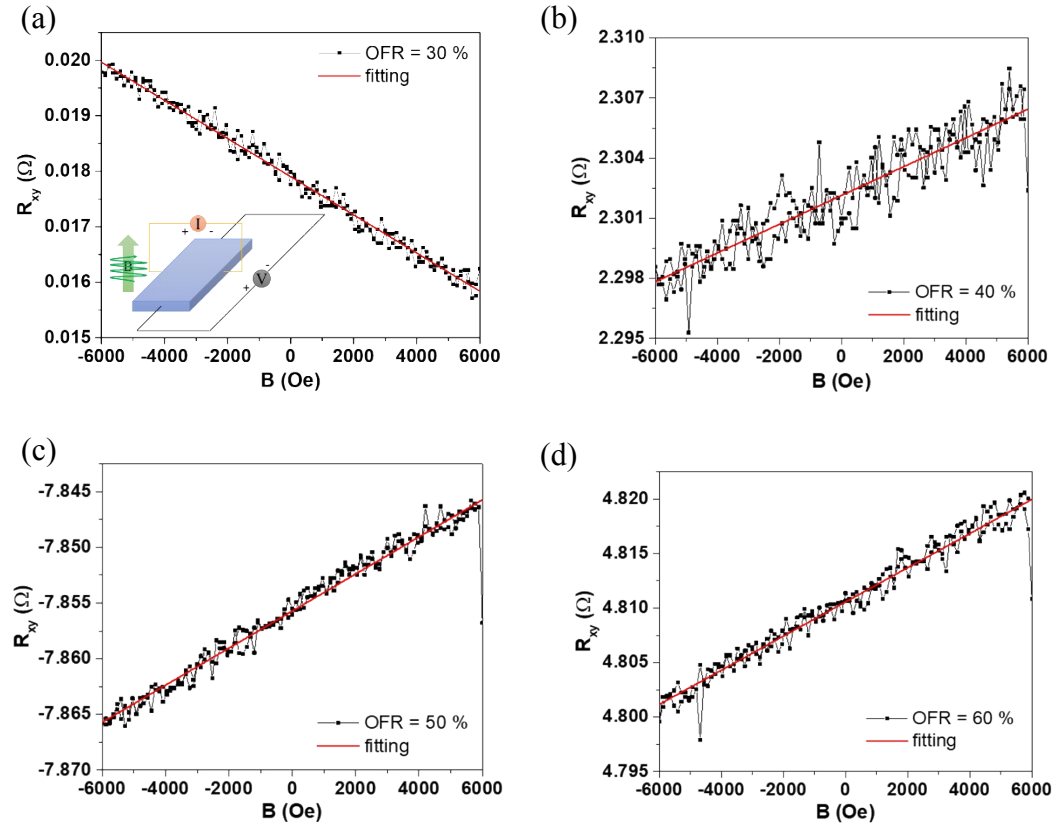


Figure S3. Hall effect measurements of the Cu/Cu₂O films with different oxygen flow ratios (a) 30%, (b) 40%, (c) 50%, and (d) 60%. Inset of (a) shows the input orientation of voltage, current, and magnetic field.[1]

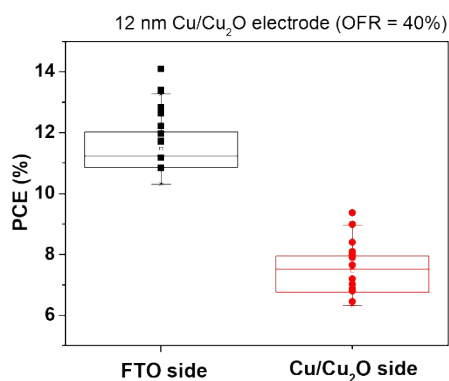


Figure S4. Statistical efficiency diagram of perovskite solar cells with Cu/Cu₂O electrodes (OFR = 40%) illuminated from FTO side and Cu/Cu₂O side, respectively.

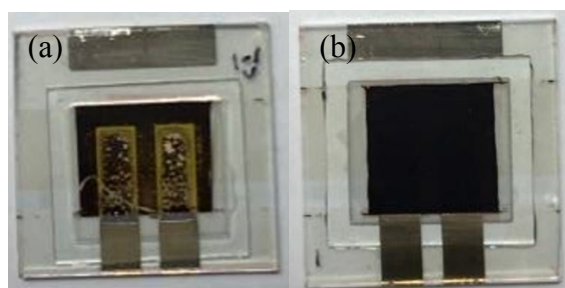


Figure S5. The photographs of the PSC using an (a) Ag as electrode and a (b) Cu/Cu₂O as electrode in dark surroundings for 55 days.

Table S1. The color change of growth films by bombarding copper target with 30 min with different argon and oxygen flow ratios.

color							
$Ar(sccm)$	5	4.5	4	3.5	3	3	3
$O_2(sccm)$	0	0.5	1	1.5	2	3	4.5
$\frac{O_2}{Ar + O_2}$	0%	10%	20%	30%	40%	50%	60%

Table S2. Electrical properties of Cu/Cu₂O composite films under different oxygen flow ratios (OFRs).[1]

	Type	Thickness (nm)	Resistivity (Ω /square)	Resistance (Ω -cm)	Mobility (cm ² /V-s)	Concentration (cm ⁻³)
OFR = 30%	n	20	78.82	1.58×10^{-4}	2.21	1.79×10^{22}
OFR = 40%	p	20	72.5	1.45×10^{-4}	1.87	2.3×10^{22}
OFR = 50%	p	20	210.1	4.2×10^{-4}	1.23	8.54×10^{21}
OFR = 60%	p	20	83360	0.1667	60.5	6.19×10^{17}

Table S3. Comparison of performance of bifacial PSCs featuring various types of transparent electrodes.

Transparen t electrode	Devcie architecture	Illumination side	PCE (%)	Reference
Carbon	Glass/FTO/compact TiO ₂ /mesoporous	FTO	6.87	[2]
Nanotube	TiO ₂ / CH ₃ NH ₃ PbI ₃ / Carbon Nanotube			
	Glass/FTO/compact TiO ₂ /mesoporous	FTO	9.9	
	TiO ₂ /CH ₃ NH ₃ PbI ₃ /Spiro- OMeTAD/Carbon Nanotube			
Au	Glass/FTO/compact TiO ₂ / CH ₃ NH ₃ PbI ₃ - _x Cl _x /Spiro-OMeTAD/Au	FTO Au	7.5 3	[3]

Au	Glass/FTO/TiO ₂ / HC(NH ₂) ₂ PbI ₃ /Spiro-OMeTAD/Au	FTO	5.2	[4]
Au/LiF	Glass/ITO/PEDOT:PSS/ CH ₃ NH ₃ PbI ₃ /PC ₆₀ BM/Au/LiF	ITO Au/LiF	7.73 3.39	[5]
MoO ₃ /Au/	Glass/FTO/TiO ₂ /CH ₃ NH ₃ PbI ₃ /Spiro-	FTO	13.0	[6]
MoO ₃	OMeTAD/ MoO ₃ /Au/MoO ₃	MoO ₃ /Au/MoO ₃	9.7	
Graphene	Glass/FTO/compact TiO ₂ /CH ₃ NH ₃ PbI ₃ . xCl _x /Spiro-OMeTAD/PEDOT:PSS+D-sorbitol/Graphene	FTO Graphene	12.02 11.65	[7]
PEDOT	Glass/FTO/compact TiO ₂ /mesoporous TiO ₂ /CH ₃ NH ₃ PbI ₃ /PEDOT	FTO PEDOT	12.33 11.78	[8]
PEIE/Ag/	Glass/ITO/PEDOT:PSS/CH ₃ NH ₃ PbI ₃ /PCB	ITO	13.55	[9]
MoO _x	M/PEIE/Ag/MoO _x	PEIE/Ag/ MoO _x	8.41	
BCP/Ag/	Glass/ITO/PEDOT:PSS/CH ₃ NH ₃ PbI ₃ /PCB	ITO	13.49	[10]
MoO ₃	M/C ₆₀ /BCP/Ag/ MoO ₃	BCP/Ag/MoO ₃	9.61	
Cu/Cu ₂ O	Glass/FTO/compact TiO ₂ /mesoporous TiO ₂ /CH ₃ NH ₃ PbI ₃ /Spiro-OMeTAD/(Cu/Cu ₂ O)	FTO Cu/Cu ₂ O	14.1 9.37	This work

Reference

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