

Supporting Information

TiO₂/(BiO)₂CO₃ Nanocomposites for Ultraviolet Filtration with Reduced Photocatalytic Activity

Kathrin Bogusz,^{†,‡} Moeava Tehei,^{‡,§,||} Michael Lerch,^{‡,||} Shi Xue Dou,[†] Hua Kun Liu,[†] and
Konstantin Konstantinov^{*,†,‡}

[†] Institute for Superconducting and Electronic Materials, Australian Institute for Innovative
Materials, University of Wollongong, NSW 2522, Australia.

[‡] Illawarra Health and Medical Research Institute, University of Wollongong, NSW 2522,
Australia.

[§] School of Chemistry, Faculty of Science, Medicine and Health, University of Wollongong,
NSW 2522, Australia.

^{||} Centre for Medical and Radiation Physics, Faculty of Engineering and Information Science,
University of Wollongong, NSW 2522, Australia.

*Email - konstan@uow.edu.au

Supporting Figures

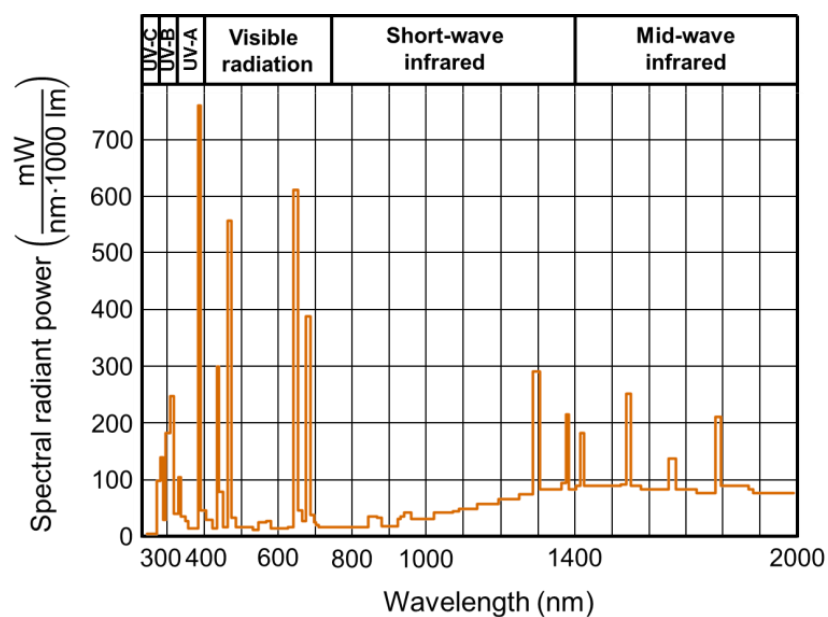


Figure S1. Adapted light emission profile of the OSRAM Ultra-Vitalux® 300 W Sunlamp.¹

Reference

- (1) Deka, M.; Humar, M.; Rep, G.; Kričej, B.; Šentjurc, M.; Petrič, M. Effects of UV Light Irradiation on Colour Stability of Thermally Modified, Copper Ethanolamine Treated and Non-Modified Wood: EPR and DRIFT Spectroscopic Studies. *Wood Sci. Technol.* **2008**, *42*, 5 – 20.

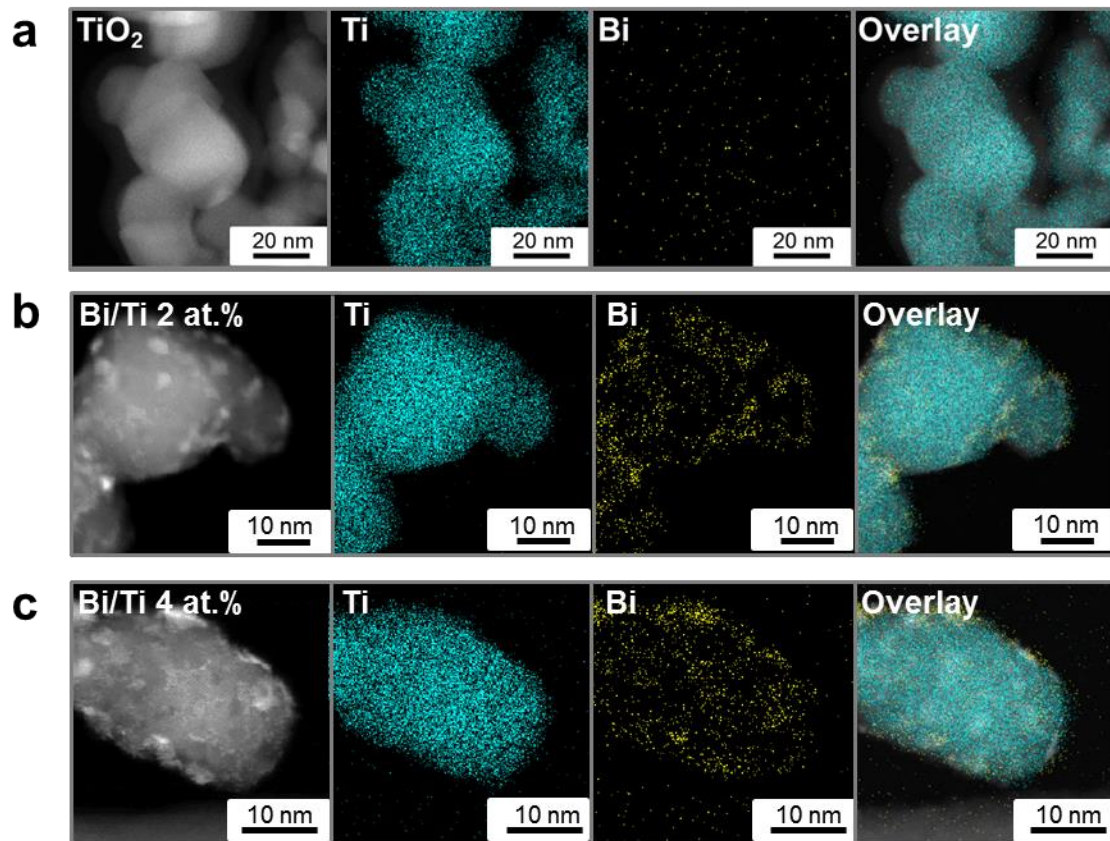


Figure S2. TEM images with high resolution EDS mapping of a) TiO₂ NPs, b) TiO₂/(BiO)₂CO₃ sample Bi/Ti 2 at.%, and c) TiO₂/(BiO)₂CO₃ sample Bi/Ti 4 at.%.

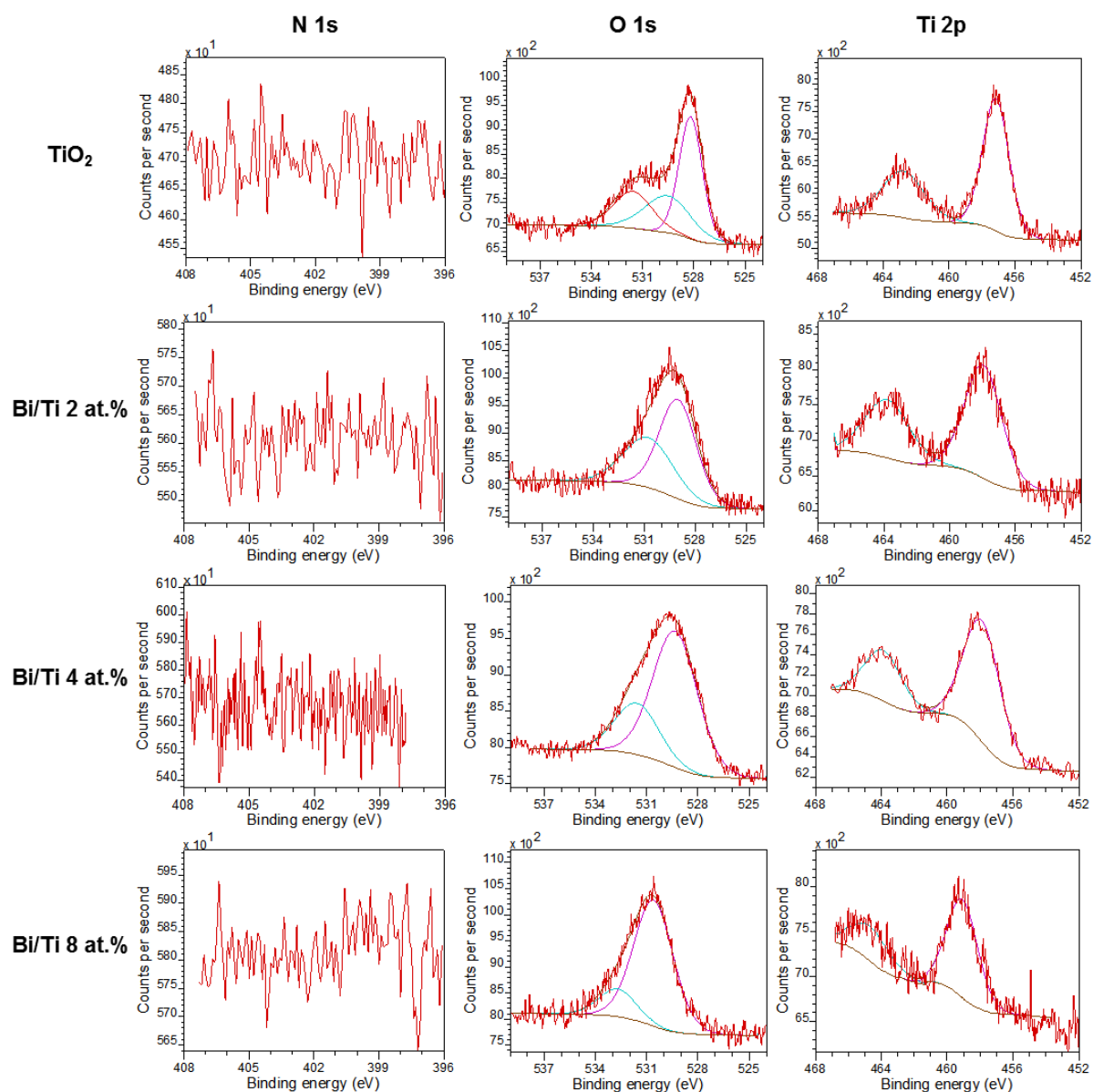


Figure S3. High resolution XPS spectra of the N 1s (left column), O 1s (centre column), and Ti 2p (right column) regions for the commercial TiO_2 NPs, and $\text{TiO}_2/(\text{BiO})_2\text{CO}_3$ nanocomposites with different Bi/Ti atomic ratios.

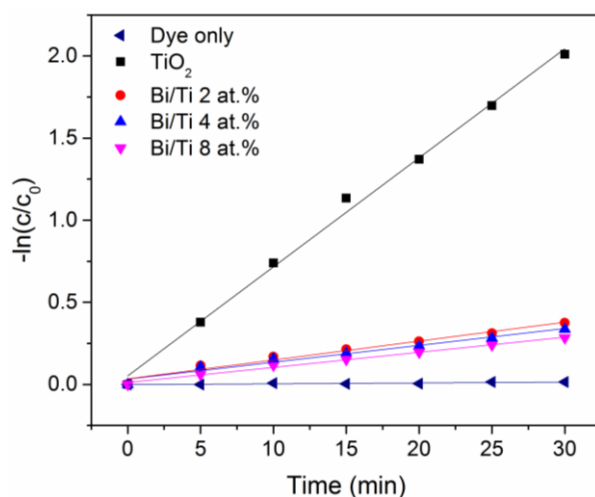


Figure S4. Apparent rate constant curves for the UV-visible dye degradation of crystal violet, as shown in Figure 5a. The concentration of all nanomaterials is 5 mg/L.

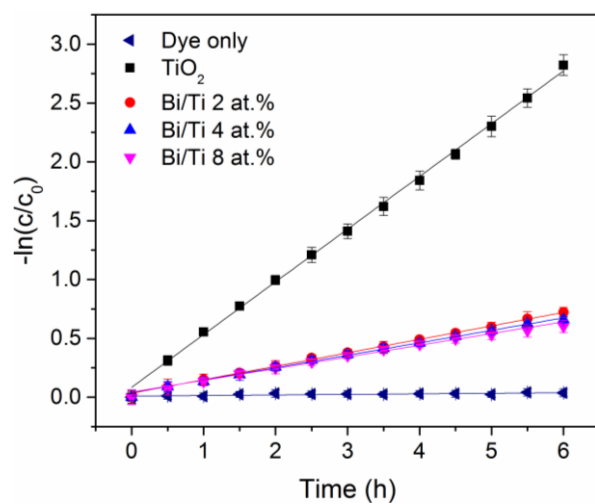


Figure S5. Apparent rate constant curves for the dye degradation of crystal violet under exposure to AM 1.5 G one sun (100 mW/cm^2), as shown in Figure 5b. The concentration of all nanomaterials is 5 mg/L.

Supporting Table

Table S1. High Resolution XPS Data^a

Sample	Bi 4f _{5/2} (eV)	Bi 4f _{7/2} (eV)	Ti 2p _{1/2} (eV)	Ti 2p _{3/2} (eV)	C 1s c-o (eV)
TiO ₂	-	-	462.83 ± 0.24	457.23 ± 0.32	286.92 ± 0.28
Bi/Ti 2 at. %	164.68 ± 0.45	159.31 ± 0.28	464.41 ± 0.41	458.54 ± 0.44	287.57 ± 0.24
Bi/Ti 4 at. %	164.85 ± 0.40	159.34 ± 0.21	464.56 ± 0.39	458.56 ± 0.36	287.56 ± 0.24
Bi/Ti 8 at. %	165.85 ± 0.39	160.51 ± 0.43	465.26 ± 0.48	459.42 ± 0.50	287.25 ± 0.23

^aData is shown for peaks of Bi 4f, Ti 2p, and C 1s for the commercial TiO₂ NPs, and the TiO₂/(BiO)₂CO₃ nanocomposites with different Bi/Ti atomic ratios. The C 1s peak of adventitious carbon for all samples is found at 284.83 eV.