# Electronic Supplementary Information

### Sulfur-bridged tetraphenylethylene AIEgens for deep-blue organic lightemitting diodes

Yuanjing Cai,<sup>‡a</sup> Changsheng Shi,<sup>‡c</sup> Han Zhang,<sup>‡a</sup> Bin Chen,<sup>a</sup> Kerim Samedov,<sup>d</sup> Ming Chen,<sup>b</sup> Zhiming Wang, <sup>a</sup> Zujin Zhao, <sup>a</sup> Xinggui Gu,<sup>\*e</sup> Dongge Ma,<sup>\*ac</sup> Anjun Qin <sup>a</sup> and Ben Zhong Tang <sup>\*ab</sup>

<sup>a</sup> Center for Aggregation-Induced Emission, NSFC Center for Luminescence from Molecular Aggregates, SCUT-HKUST Joint Research Institute, State Key Laboratory of Luminescent Materials and Devices, South China University of Technology, Guangzhou 510640, China. \*E-mail: <u>tangbenz@ust.hk</u>; <u>msdgma@scut.edu.cn</u>.

<sup>b</sup> Department of Chemistry, Hong Kong Branch of Chinese National Engineering Research Center for Tissue Restoration and Reconstruction, The Hong Kong University of Science & Technology, Clear Water Bay, Kowloon, Hong Kong SAR, China.

<sup>c</sup> State Key Laboratory of Polymer Physics and Chemistry, Changchun Institute of Applied Chemistry, University of Chinese Academy of Sciences, Changchun 130022, China.

<sup>d</sup> Department of Chemistry, University of British Columbia, 2036 Main Mall, Vancouver, British Columbia, Canada V6T 1Z1.

<sup>e</sup> Beijing Advanced Innovation Center for Soft Matter Science and Engineering, Beijing University of Chemical Technology, Beijing 100029, China. \*E-mail: <u>guxinggui@mail.buct.edu.cn</u>.

<sup>‡</sup> These authors contributed equally to this work.

# Table of Contents

- 1. HRMS and NMR spectra
- 2. Single Crystal X-ray Crystallography
- 3. AIE properties
- 4. OLEDs performances
- 5. Theoretical calculation

### 6. References

### 1. HRMS and NMR spectra



Figure S1. HRMS of compound 1.





--8.17 --8.15

Figure S2. <sup>1</sup>H NMR spectrum of compound 1 in CDCl<sub>3</sub>.



Figure S3. <sup>13</sup>C NMR spectrum of compound 1 in CDCl<sub>3</sub>.

S4



Figure S4. HRMS spectrum of compound 2.







Figure S6. <sup>13</sup>C NMR spectrum of compound 2 in CDCl<sub>3</sub>.



Figure S7. HRMS spectrum of compound 3.





Figure S8. <sup>1</sup>H NMR spectrum of compound 3 in CDCl<sub>3</sub>.

147.75 141.75 137.15 135.57 135.67 135.67 135.67 125.58 126.58 126.58 126.58 126.58 126.58 126.58 122.83	77.48 76.84
┶─ <del>╘───────────────────────────────────</del>	$\sim$



Figure S9. <sup>13</sup>C NMR spectrum of compound 3 in CDCl<sub>3</sub>.



Figure S10. HRMS spectrum of compound 4.





Figure S11. <sup>1</sup>H NMR spectrum of compound 4 in CDCl<sub>3</sub>.



Figure S12. <sup>13</sup>C NMR spectrum of compound 4 in CDCl<sub>3</sub>.

--8.06

### 2. Single Crystal X-ray Crystallography

Suitable single crystals of sulfur-bridged TPE derivatives were selected under oil under ambient conditions. Single crystal X-ray diffraction intensity data were collected in a stream of cold nitrogen at 100K on a SuperNova, dual Cu/Mo Atlas diffractometer or at 90K on a Bruker APEX DUO diffractometer. Using Olex2, <sup>[1]</sup> the structure was solved with the ShelXS <sup>[2]</sup> structure solution program using Direct Methods and refined with the ShelXL<sup>[3]</sup> refinement package using Least Squares minimization. The details concerning x-ray crystallographic structure solutions and refinement for TPE derivatives are tabulated in **Table S1**.

	1	3	4
CCDC No.	1826997	1826670	1826998
empirical formula	$C_{50}H_{32}N_2S$	$C_{50}H_{36}N_2S$	C <sub>50</sub> H <sub>36</sub> N <sub>2</sub> O <sub>2</sub> S • CHCl <sub>3</sub>
formula weight	692.83	696.91	848.23
crystal dimensions, mm	$0.4 \times 0.04 \times 0.03$	0.3  imes 0.1  imes 0.1	0.4  imes 0.05  imes 0.03
crystal system	monoclinic	monoclinic	monoclinic
space group	P21	$P2_1/n$	P2 <sub>1</sub> /c
a/Å	10.2848(6)	13.4729(6)	15.9689(4)
b/Å	9.0120(5)	11.6041(5)	28.4291(4)
c/Å	19.3052(11)	24.7449(10)	9.49878(19)
α/°	90	90	90
β/°	91.856(5)	97.034(2)	106.722(2)
$\gamma/^{\circ}$	90	90	90
Volume/Å <sup>3</sup>	1788.41(17)	3839.5(3)	4129.93(15)
Ζ	2	4	4
$\rho_{calc}, mg/mm^3$	1.287	1.206	1.364
F(000)	724.0	1464.0	1760.0
λÅ	Cu Kα, λ = 1.54184	Mo Kα, $\lambda = 0.71073$	Cu Ka , 1.54184
µ/mm <sup>-1</sup>	1.100	0.122	2.833
temperature/K	100.00(10)	90.00(2)	99.98(10)
$2\theta$ range for data collection	8.602° to 134.988°	3.3° to 61.2°	6.562° to 135°
reflections collected	5324	87810	12935
independent reflections. (Rint)	4010 (0.0412)	11714 (0.0545)	7349 (0.0291)
data/restraints/parameters	4010/1/478	11714/0/478	7349/0/532
goodness-of-fit on F <sup>2</sup>	1.001	1.027	1.022
$R_1, wR_2 [I \ge 2\sigma (I)]$	0.0543, 0.1306	0.0430, 0.1038	0.0478, 0.1297
R <sub>1</sub> ,wR <sub>2</sub> [all data]	0.0683, 0.1409	0.0599, 0.1141	0.0606, 0.1399
largest diff. peak/hole / e Å-3	0.27/-0.37	0.36/-0.36	0.38/-0.58

Table S1. The crystallographic details of sulfur-bridged TPE derivatives.



Figure S13. Thermal ellipsoids drawing of sulfur-bridged TPE derivative 1.



Figure S14. Thermal ellipsoids drawing of sulfur-bridged TPE derivative 3.



Figure S15. Thermal ellipsoids drawing of sulfur-bridged TPE derivative 4.



**3. AIE properties** 

**Figure S16**. (a) **1**, (b) **2**, (c) **3**, (d) **4** AIE curves of sulfur-bridged tetraphenylethylene derivatives **1-4 in** THF/H<sub>2</sub>O mixtures ( $10^{-5}$  M) with different water fractions ( $f_w$ ).

### 4. OLEDs performances



**Figure S17**. Power efficiency *versus* luminance curves of the non-doped OLEDs based on sulfurbridged tetraphenylethylene derivatives **1-4** as emitters, device configuration: ITO/MoO<sub>3</sub> (10nm)/NPB (40nm)/TCTA (10nm)/EML (10nm)/BmPyPB (40nm)/LiF (1nm)/Al.



**Figure S18**. Power efficiency *versus* luminance curves of the doped OLEDs based on sulfur-bridged tetraphenylethylene derivative **3** as dopant, device configuration: ITO/MoO<sub>3</sub> (10nm)/NPB (40nm)/TCTA (10nm)/EML (10nm)/BmPyPB (40nm)/LiF (1nm)/Al.

### 5. Theoretical calculation

All density functional theory (DFT) calculations were performed using Gaussian 09 (Revision D.01)<sup>[4]</sup> employing Becke, Lee, Yang and Parr (B3LYP) functional<sup>[5]</sup> in combination with the 6-311G(d) basis set as implemented in Gaussian 09 (Revision D.01). All geometry optimizations were performed without imposing any symmetry constraints using tightened cutoffs (keyword *Tight*) on forces and step size to ensure adequate convergence and reliability of frequencies computed in a subsequent job step. Ultrafine (pruned (99,590)) integration grid (keyword *Ultrafine*) as implemented in Gaussian 09 (Revision D.01) was used for all calculations. Frequency calculations were used to confirm the identity of all obtained minima (no negative eigenvalues of the Hessian matrix).

**Table S2.** The optimized geometric structure of **1** in gas phase at the ground state ( $S_0$ ) performed at B3LYP/6-311G(d) using the Gaussian 09 package.

Atom	X	У	Z
Ν	4.937600	-1.500100	-0.043100
С	3.704900	-0.803900	-0.105500
С	3.119600	-0.298400	1.061500
С	1.911300	0.389500	0.993400
С	1.278000	0.613200	-0.236500
С	-0.014900	1.367100	-0.280200
С	-0.058900	2.724600	-0.237200
С	-1.338200	3.484600	-0.280500
С	-1.582500	4.487100	0.675900
S	-0.349100	4.857300	1.911300
С	1.121900	4.581200	0.934600
С	1.149100	3.576800	-0.054700
С	2.314300	3.442300	-0.825000
Η	2.337200	2.706900	-1.620000
С	3.438800	4.224200	-0.574500
Н	4.332600	4.090400	-1.177300
С	3.413700	5.177500	0.445700
Η	4.287800	5.790600	0.646500
С	2.251100	5.367300	1.189200
H	2.208900	6.135800	1.955800
С	-2.800200	5.177200	0.692900
H	-2.971800	5.948000	1.438900
С	-3.776400	4.885400	-0.257100
H	-4.720000	5.423700	-0.241400
С	-3.528200	3.926300	-1.243900
H	-4.276700	3.714200	-2.002100
С	-2.316000	3.242900	-1.25//00
H	-2.123600	2.48/200	-2.012400
C	-1.243/00	0.520400	-0.234100
C	-1.449000	-0.534100	-1.136800
H	-0.700300	-0./51300	-1.891/00
U	-2.619300	-1.288500	-1.10/900
п	-2.790600	-2.068100	-1.843300
C NI	-3.396100	-1.029600	-0.139900
	-4.804000	-1.700100	-0.120400
C	-4.920800	-3.102100	-0.109500
U U	-2 870700	-3 861500	-0 044000
п	-2.070700	-5 471300	-0 040000
ц	-4.JIOIUU _3 550100	-6 2/8300	-0.040900
11	J.JJJ100	0.240500	0.000100

С	-5.677200	-5.831900	-0.053500
Н	-5.953200	-6.882400	-0.035500
С	-6.669100	-4.855400	-0.082900
Н	-7.720100	-5.132400	-0.081900
С	-6.296400	-3.507200	-0.108200
С	-7.041300	-2.266300	-0.123500
С	-8.406600	-1.961200	-0.145500
Н	-9.145300	-2.758300	-0.138700
С	-8.801600	-0.626700	-0.182700
Н	-9.858500	-0.376200	-0.198100
С	-7.843800	0.402600	-0.206900
Н	-8.172700	1.437600	-0.245900
С	-6.478000	0.126700	-0.184700
Н	-5.742900	0.924400	-0.214800
С	-6.089800	-1.214900	-0.131200
С	-3.375400	-0.022700	0.808500
Н	-4.115300	0.150400	1.583300
С	-2.218500	0.744100	0.753500
Н	-2.061600	1.531000	1.483900
С	1.873400	0.102300	-1.398100
С	3.069800	-0.608900	-1.337300
Н	3.520400	-1.012200	-2.238500
Н	1.403600	0.276200	-2.362600
Н	1.463800	0.788100	1.899100
Н	3.624200	-0.436600	2.012500
С	6.086800	-1.190400	-0.778500
С	6.305600	-0.151700	-1.687000
Н	5.522200	0.561000	-1.922500
С	7.567600	-0.058200	-2.270700
Н	7.763200	0.739900	-2.981900
С	8.590400	-0.970400	-1.956300
Н	9.562300	-0.870800	-2.431300
С	8.369600	-1.993900	-1.038600
Н	9.162700	-2.692700	-0.785900
С	7.111600	-2.109500	-0.437400
С	6.555300	-3.012400	0.547400
С	7.064300	-4.116100	1.240500
Н	8.090200	-4.437000	1.080200
С	6.237900	-4.796600	2.130600
Н	6.621200	-5.653900	2.676300
С	4.906900	-4.387500	2.327100
Н	4.275400	-4.937200	3.019800
С	4.375500	-3.291700	1.649700
Н	3.345300	-2.985100	1.796600
С	5.214800	-2.605900	0.768200

**Table S3**. The optimized geometric structure of **2** in gas phase at the ground state ( $S_0$ ) performed at B3LYP/6-311G(d) using the Gaussian 09 package.

Atom	X	У	Z
S	-0.231500	4.388400	1.505500
С	-1.554300	4.154500	0.322700
С	-1.371700	3.141100	-0.632800
С	-0.073200	2.404300	-0.667300
С	-0.002800	1.052000	-0.649500
С	1.302800	0.339200	-0.507600
С	2.018000	0.480600	0.691900
С	3.239100	-0.161800	0.870600

С	3.759100	-0.976000	-0.143500
N	5 004100	-1 628200	0 032200
C	6 004200	-1 556900	_0 942700
C	0.094200	1.00000	0.042700
C	6.232800	-0.83/500	-2.032500
Η	5.423100	-0.225100	-2.415000
С	7.450000	-0.927100	-2.705400
Н	7.582200	-0.378600	-3.634100
C	8 507700	-1 707600	-2 206600
	0.007700	1 750100	2.200000
Н	9.442900	-1./58100	-2.757000
С	8.367800	-2.409700	-1.012400
Н	9.188300	-3.004300	-0.619400
С	7.156200	-2.335900	-0.317100
C	6 685600	-2 902400	0 928800
C	7 261700	2 752100	1 070200
0	7.201700	-3.752100	1.0/9300
Н	8.2/9/00	-4.109300	1./48100
С	6.511500	-4.135100	2.987700
Н	6.947300	-4.792700	3.734300
С	5.189900	-3,682500	3.148300
U U	4 617700	-3 000000	1 015700
	4.017700	3.999900	2.010/00
C	4.592900	-2.835600	2.216400
Н	3.569500	-2.497600	2.339800
С	5.356800	-2.443900	1.114200
С	3.043000	-1.138100	-1.335600
C	1 827800	-0 480600	-1 514400
U U	1 201000	-0 601700	-2 451200
п	1.291900	-0.001700	-2.451200
Н	3.442400	-1./81500	-2.113000
Н	3.803200	-0.026600	1.787800
Н	1.617300	1.121000	1.473100
С	-1.233700	0.215200	-0.529000
C	-1 605900	-0 724600	-1 498900
	0.001200	0.066200	2 202500
Н	-0.991200	-0.866300	-2.382500
С	-2.790800	-1.446100	-1.373100
Н	-3.102000	-2.136700	-2.150300
С	-3.606200	-1.263000	-0.249800
N	-4.829300	-1,966200	-0.132000
C	-5 000300	-3 350300	-0 244600
a	1.040000	1 240000	0.244000
C	-4.042900	-4.349900	-0.433900
H	-2.989400	-4.107600	-0.526500
С	-4.488200	-5.669300	-0.488700
Н	-3.763700	-6.465700	-0.636000
С	-5.850500	-5.989700	-0.353000
ч	-6 164300	-7 028500	-0 402000
C C	6 707200	1 000100	0.1502000
C	-6.797200	-4.989100	-0.150200
H	-7.849200	-5.236300	-0.034200
С	-6.376500	-3.656200	-0.091500
С	-7.065900	-2.400600	0.117200
С	-8.408600	-2.058200	0.309800
U U	-0 171700	-2 931600	0 337600
п	-9.171700	-2.031000	0.337000
C	-8./50800	-0./16800	0.458300
Н	-9.789500	-0.437800	0.611100
С	-7.763900	0.283300	0.406900
Н	-8.052700	1.325200	0.516000
C	-6 419300	-0 030300	0 217200
с ц	-5 660000	0.030300	0.21/200
п	-3.002800	U./401UU	0.10/400
С	-6.082000	-1.380100	0.085500
С	-3.214700	-0.364800	0.750900
Н	-3.833700	-0.249100	1.634700
С	-2.045500	0.373500	0.606000
н	-1 748800	1 083200	1 373500
C	1 1 2 5 6 0 0	1.000200	1.373300
C	I.I33600	3.2/5200	-0.3/6600

С	1.161900	4.299800	0.385000
С	2.228500	5.182400	0.519300
С	3.324600	5.040000	-0.332200
С	3.325000	4.032000	-1.300000
С	2.240600	3.162800	-1.428200
Н	2.246800	2.389400	-2.188600
Н	4.174200	3.926000	-1.969200
Н	4.170400	5.715400	-0.245100
Н	2.186200	5.960900	1.274400
С	-2.418900	2.907100	-1.532000
С	-3.599800	3.646400	-1.452100
С	-3.754900	4.645800	-0.487000
С	-2.718800	4.910600	0.409100
Н	-2.797000	5.687400	1.163100
Н	-4.674700	5.221000	-0.437800
Н	-4.401700	3.446400	-2.157200
Н	-2.305800	2.137400	-2.288000
0	-0.316900	5.734800	2.092800
0	-0.190500	3.200000	2.384700

**Table S4**. The optimized geometric structure of **3** in gas phase at the ground state ( $S_0$ ) performed at B3LYP/6-311G(d) using the Gaussian 09 package.

Atom	X	У	Z
С	1.315100	-4.423300	-0.031900
S	0.037700	-4.891400	1.124300
С	-1.391900	-4.320200	0.216600
С	-1.314400	-3.164800	-0.587200
С	-0.034700	-2.399500	-0.617600
С	1.174800	-3.248400	-0.792900
С	2.184700	-2.927000	-1.713100
С	3.328700	-3.711100	-1.825800
С	3.475000	-4.850700	-1.028900
С	2.463600	-5.215800	-0.143300
Н	2.553800	-6.119800	0.452300
Н	4.365300	-5.467900	-1.113000
Н	4.104200	-3.436500	-2.535400
Н	2.072100	-2.033000	-2.318000
С	0.030800	-1.054300	-0.424000
С	1.317800	-0.327900	-0.232900
С	2.269200	-0.798700	0.689500
H	2.048800	-1.689900	1.268200
С	3.478500	-0.145400	0.879200
H	4.195100	-0.525400	1.599900
С	3.793300	1.001900	0.132200
Ν	5.051300	1.635500	0.283900
С	6.227400	0.850700	0.416900
С	6.383700	-0.333500	-0.322200
H	5.595000	-0.649200	-0.997600
С	7.535800	-1.103300	-0.178000
H	7.638400	-2.018600	-0.755200
С	8.558000	-0.700400	0.684500
H	9.45/800	-1.300000	0./88600
C	8.408300	0.481800	1.413000
H	9.190900	0.805300	2.094400
C	7.251300	1.248500	1.291500
H	/.132800 F 140400	2.159900	1.868600
C	5.140400	3.053000	0.25/300

С	6.168000	3.690800	-0.455000
Н	6.896300	3.089400	-0.989600
С	6.252500	5.081600	-0.469100
н	7 054900	5 560300	-1 024500
C	5 307000	5 957000	0.206400
	5.307900	5.057900	0.200400
Н	5.3/2900	6.942000	0.186900
С	4.278800	5.224600	0.907500
Н	3.540500	5.815100	1.443700
С	4.197500	3.834200	0.943800
Н	3.404200	3.342800	1.498100
С	2.837700	1,500800	-0.766200
н	3 060100	2 394000	-1 340900
C	1 615100	0 855900	-0.925600
	1.013100	1 267200	1 625100
H	0.894500	1.267200	-1.625100
C	-1.21/800	-0.248200	-0.252200
С	-1.748100	0.535900	-1.285600
Н	-1.231700	0.587600	-2.240500
С	-2.974400	1.180400	-1.142600
Н	-3.394400	1.748400	-1.966600
С	-3.694700	1.068400	0.055000
N	-4 984600	1 647200	0 186000
C	-5 227700	2 947100	-0.327500
	-3.227700	2.947100	-0.327300
	-4.2/5300	3.963000	-0.145300
Н	-3.363000	3.742800	0.400100
С	-4.499000	5.236900	-0.662600
Н	-3.750700	6.010800	-0.512600
С	-5.679100	5.527200	-1.351600
Н	-5.854100	6.523800	-1.746400
С	-6.631300	4,520400	-1.526600
н	-7 551100	4 728200	-2 067200
C	-6 408300	3 237600	-1 030100
	7 145100	3.237000	1 170700
н	-7.145100	2.455200	-1.1/9/00
C	-6.042600	0.863500	0./1/600
С	-6.109300	-0.516300	0.459700
Н	-5.345100	-0.987000	-0.150500
С	-7.140200	-1.281500	1.000600
Н	-7.170800	-2.348500	0.794900
С	-8.131100	-0.689200	1.786500
Н	-8.936600	-1.288700	2 200900
C	-8 071100	0 683900	2 036400
U U	-9 929100	1 150400	2.653400
п	-0.029100	1.159400	2.033400
C	-7.033100	1.455300	1.518200
Н	-6.982600	2.519200	1./26500
С	-3.139400	0.337200	1.115600
Н	-3.687500	0.257400	2.048900
С	-1.923000	-0.316100	0.957300
Н	-1.520300	-0.912300	1.771000
С	-2.453600	-2.798400	-1.321500
н	-2 401500	-1 939300	-1 979100
C	-3 6/0/00	-3 501200	_1 100000
	-3.049400	-3.JUIJUU	-I.JJJJUU
п	-4.518600	-3.18//00	-1.//0/00
C	-3.725500	-4.608500	-0.350000
Н	-4.655400	-5.161300	-0.248800
С	-2.592500	-5.028800	0.342500
ц	-2.629500	-5.915300	0.969400

Table S5. . The optimized geometric structure of 4 in gas phase at the ground state (S<sub>0</sub>) performed at B3LYP/6-311G(d) using the Gaussian 09 package.

Atom	X	У	Z
S	0.066100	-4.477300	0.924600
0	0.066500	-5.910000	1.263300
0	0.139900	-3.466300	2.000300
С	1.368400	-4.131600	-0.252200
С	2.475600	-4.969700	-0.344000
Н	2.515700	-5.869600	0.261500
С	3.503100	-4.625100	-1.221700
Н	4.378300	-5.262000	-1.309400
С	3.395900	-3.463800	-1.993300
Н	4.190000	-3.200500	-2.686800
С	2.272600	-2.642600	-1.897100
Н	2.195100	-1.746100	-2.503000
С	1.234500	-2.957000	-1.010300
С	-0.009400	-2.141500	-0.884600
С	-1.274000	-2.936700	-0.891700
С	-2.393900	-2.617600	-1.670800
Н	-2.367300	-1.736200	-2.301300
С	-3.536800	-3.417100	-1.633900
Н	-4.393300	-3.148300	-2.246400
С	-3.584900	-4.563000	-0.834800
Н	-4.476400	-5.182900	-0.817900
С	-2.475500	-4.916100	-0.067000
Н	-2.467100	-5.807700	0.551900
С	-1.348900	-4.100000	-0.106100
С	0.012200	-0.802900	-0.664300
С	1.288200	-0.071900	-0.428100
С	2.140600	-0.500100	0.604400
Н	1.838500	-1.338800	1.225500
С	3.347600	0.141200	0.847400
Н	3.991000	-0.198800	1.652200
С	3.755300	1.221000	0.046400
Ν	5.013700	1.835300	0.255400
С	6.149800	1.038000	0.559100
С	6.341500	-0.202000	-0.071800
Н	5.610600	-0.552100	-0.793800
С	7.453200	-0.982100	0.238600
Н	7.584200	-1.941300	-0.255600
С	8.400700	-0.534800	1.162300
Н	9.269400	-1.143200	1.396800
С	8.216500	0.702800	1.783100
Н	8.940400	1.061100	2.510300
С	7.097500	1.481300	1.494800
Н	6.950300	2.436600	1.988300
С	5.150000	3.243000	0.119600
С	6.254900	3.791400	-0.549900
Н	7.006300	3.128300	-0.966700
С	6.385900	5.173400	-0.671100
Н	7.248000	5.582700	-1.191200
С	5.412900	6.027800	-0.147200
Н	5.515000	7.104400	-0.249900
С	4.307800	5.483000	0.511200
Н	3.546700	6.135500	0.930800
С	4.178300	4.103200	0.654700
Н	3.325800	3.681100	1.177400
С	2.894500	1.671800	-0.967500
Н	3.193200	2.510600	-1.587500
С	1.671100	1.045300	-1.183200
Н	1.026700	1.409700	-1.977400

С	-1.256000	-0.048100	-0.436300
С	-1.766500	0.872600	-1.359700
Н	-1.229300	1.069000	-2.283500
С	-2.996700	1.489900	-1.148400
Н	-3.400300	2.171500	-1.890100
С	-3.735600	1.218700	0.013300
N	-5.011900	1.802500	0.213600
С	-5.235000	3.157100	-0.148700
С	-4.271600	4.134300	0.148300
Н	-3.362100	3.843100	0.664200
С	-4.481700	5.461500	-0.219500
Н	-3.725500	6.205400	0.017600
С	-5.658800	5.841000	-0.869100
Н	-5.823100	6.878200	-1.146800
С	-6.622500	4.871700	-1.156700
Н	-7.540500	5.150400	-1.667600
С	-6.412900	3.538200	-0.810700
Н	-7.158300	2.785300	-1.045600
С	-6.080200	1.003100	0.700600
С	-6.211900	-0.336100	0.297600
Н	-5.487800	-0.761800	-0.389800
С	-7.254800	-1.117400	0.790500
Н	-7.338000	-2.153400	0.472400
С	-8.193700	-0.577500	1.672400
Н	-9.008700	-1.188400	2.049500
С	-8.069600	0.756800	2.066500
Н	-8.786700	1.189700	2.759000
С	-7.018500	1.541000	1.595900
Н	-6.917400	2.573400	1.914900
С	-3.201700	0.336100	0.966000
Н	-3.760900	0.130100	1.872800
С	-1.985900	-0.295300	0.736600
Н	-1.594200	-1.003500	1.461900

#### 6. References

- [1] Dolomanov, O. V., Bourhis, L. J., Gildea, R. J., Howard, J. A. K. & Puschmann, H. OLEX2: a complete structure solution, refinement and analysis program. *J. Appl. Cryst.* **42**, 339-341 (2009).
- [2] Sheldrick, G. M. A short history of SHELX. Acta Cryst. A. 64, 112-122 (2008).
- [3] Sheldrick, G. M. Crystal structure refinement with SHELXL. Acta Cryst. C. 71, 3-8 (2015).
- [4] M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski and D. J. Fox, EM64L-Gaussian 09, Revision D.01. Gaussian, Inc., Wallingford CT, 2013.

[5]. (a) A. D. Becke, J. *Chem. Phys.* **98**, 5648-5652 (1993); (b) C. Lee, W. Yang, R.G. Parr, *Phys. Rev. B* **37**, 785-789 (1988).