

## Supporting Information

### **Silicate conductive filaments assisted broadband light emission of HfO<sub>2</sub> high-k solid state incandescent devices**

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### 1. JV curve for ITO electrode device and silver paste electrode

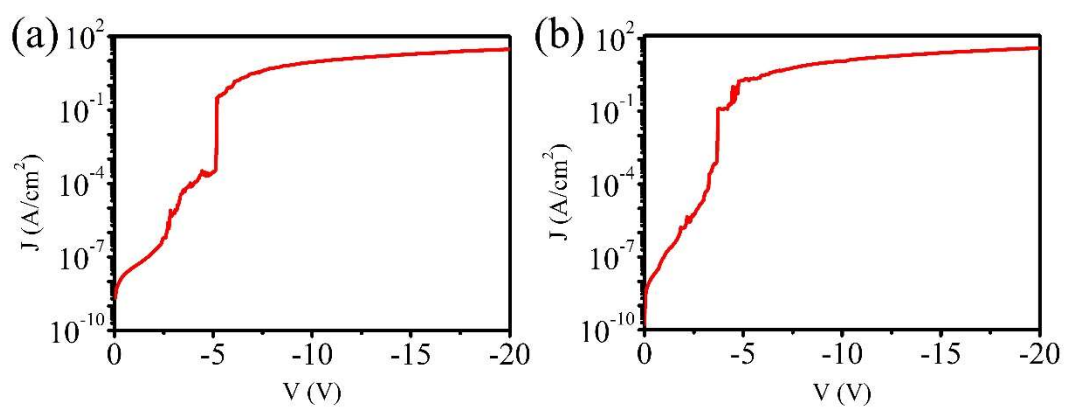


Figure S1 JV curve of (a) ITO electrode device, (b) silver paste electrode device

## 2. TEM and EDX line scan

Fig. S1 (a) shows the cross-sectional HRTEM views (bright-field) of Ti-embedded HfO<sub>2</sub> after RTA. The thickness of the HfO<sub>2</sub>-Ti-HfO<sub>2</sub> stack layer can be extracted as 9 nm. Fig. 1 (b) shows the line scan from point A to point D of the energy dispersive X-ray spectrometry (EDXS) for the HfO<sub>2</sub>-Ti-HfO<sub>2</sub> stack layer. It can be seen from Fig. S1 (a) that the layer with darker color (point B region) is related to the Ti-HfO<sub>2</sub> layer with a high Hf (with a higher atomic number) concentration (red curve in (b)), while the layer with brighter color correlates to the silicate layer (point C region) with a high Si (with a smaller atomic number) concentration (green curve in (b)). It can be noticed here that the Ti (black curve) has diffused into these two areas because of RTA procession thus the Ti-embedded layer cannot be observed in Fig. S1 (a).

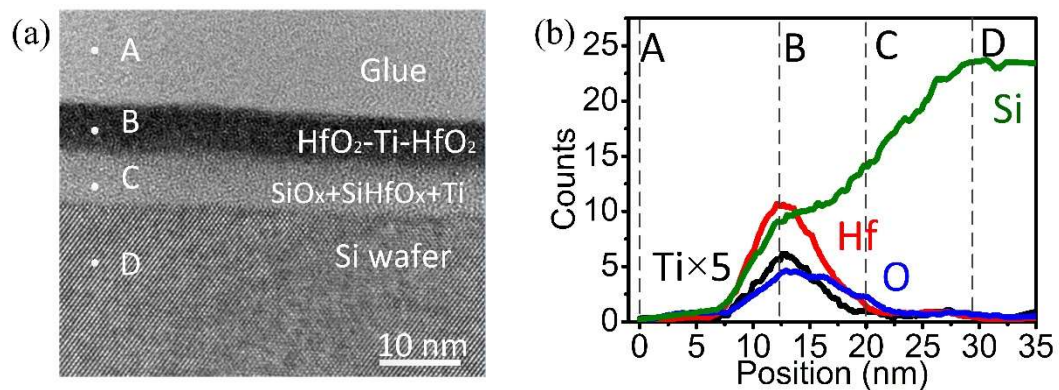


Figure S2 TEM image (a) and EDS line scan of Ti-embedded HfO<sub>2</sub> thin film after RTA

### 3. AFM and CAFM

The topography and current maps of HfO<sub>2</sub> thin film after stressed under -35 V are shown in Fig. S2 (a) and (b) respectively. After a careful searching on the large smooth surface of the HfO<sub>2</sub> surface, the pitted regions can be found. In Fig. S2(a), two black, rectangle regions with a size of ~900 nm×240 nm can be distinguished, corresponding to the deepest pits, the depth is ~7 nm extracted from the quantitative analysis. Fig. S2 (b) shows the current map of the same region shown in Fig. S2 (b). It is found that pits regions are conductive with a maximum current of approximately 12 nA and the conductive regions is exactly same as the pits shown in figure 5 (a). The other areas are insulating. After the dielectric hard breakdown, the current would only pass along these conductive paths which generate heat and emits light, quite similar as the current flow through the filaments.

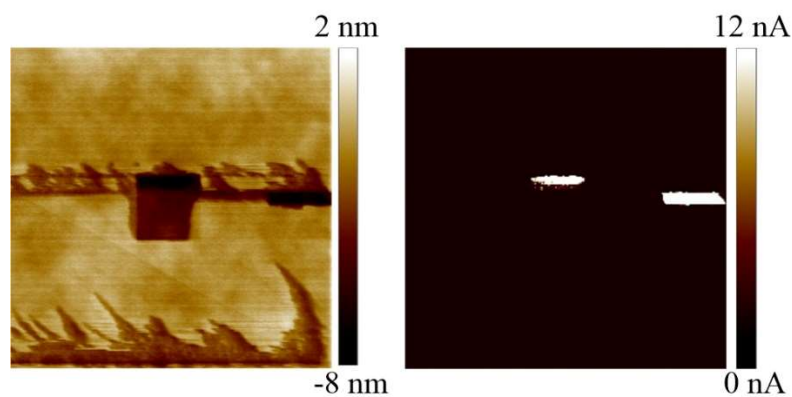


Figure S3 (a) Topography map and (b) current map of the surface of the HfO<sub>2</sub> thin film after hard breakdown;

#### 4. First and second electrical measurement of devices

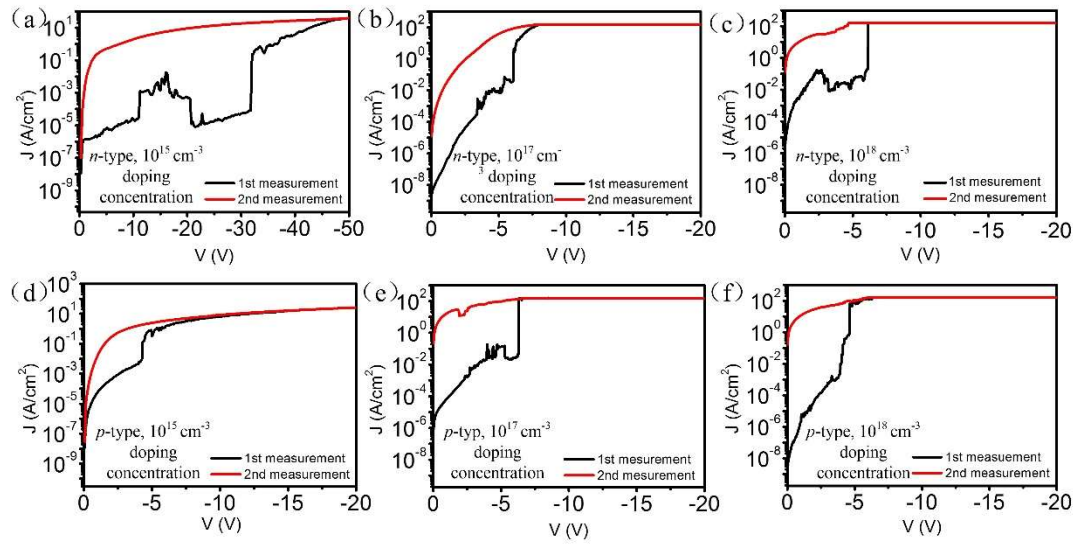


Figure S4 J-V curve for the devices fabricated on different substrates:  $n$ -type Si with (a)  $10^{15}$  cm<sup>-3</sup>; (b)  $10^{17}$  cm<sup>-3</sup>; (c)  $10^{18}$  cm<sup>-3</sup> doping concentration;  $p$ -type Si with (d)  $10^{15}$  cm<sup>-3</sup>; (e)  $10^{17}$  cm<sup>-3</sup>; (f)  $10^{18}$  cm<sup>-3</sup> doping concentration;

## 5. O 1s and Ti 2p XPS measurement of the Ti-HfO<sub>2</sub> thin film surface

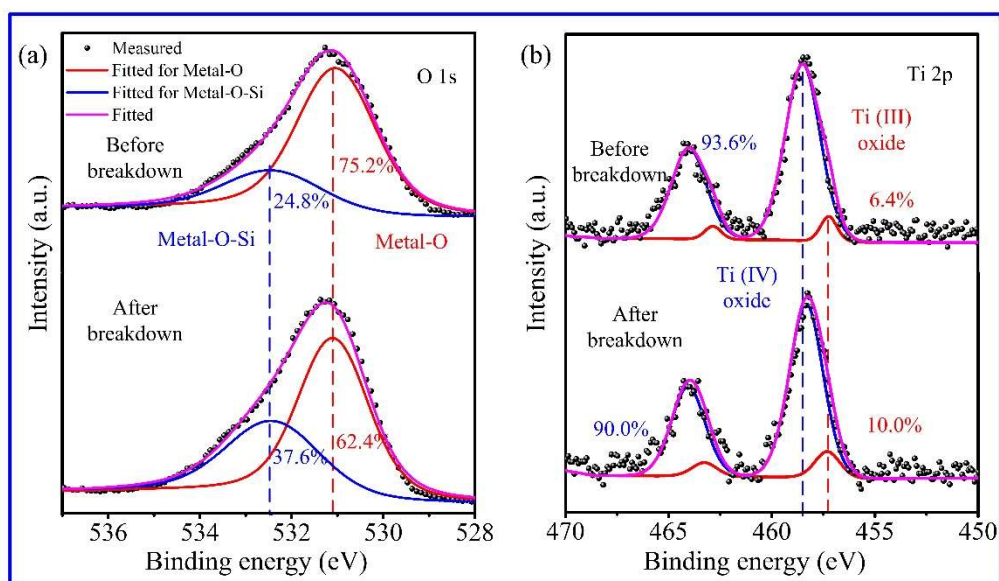


Figure S5 Deconvolutions of the spectra taken from HfO<sub>2</sub> thin film before and after breakdown, (a) O 1s (b) Ti 2p

In Fig. S5 (a), O 1s XPS spectra were fitted into two peaks. It should be noted here that the Metal-O-Si bonds observed from the sample before breakdown belong to the silicate located at interface between Si substrate and HfO<sub>2</sub>, which was revealed by TEM in Fig. S2. One is located at 531.2 eV attributing to Metal-O bonds<sup>1, 2</sup> and the other located at 532.4 eV corresponding to Metal-O-Si bonds<sup>1, 2</sup>. As shown in Fig. S5 (b), the Ti 2p spectra consists Ti (VI)<sup>3</sup> located at 458.6 eV attributing to TiO<sub>2</sub> and Ti (III)<sup>3</sup> located at 457.1 eV corresponding to Ti<sub>2</sub>O<sub>3</sub>. The area ratio of Metal-O-Si increases from 24.8% to 37.6% and there is a slight augment of Ti (III) (from 6.4% to 10%). These results indicate the fact that the diffusion of Si ions into Ti-HfO<sub>2</sub> layer, which react with Ti-HfO<sub>2</sub> and cause the increase of Metal-O-Si. Numbers of O is got by Si ions thus lots of Ti (VI) being transformed into Ti (III) (reduction reaction).

## Reference

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2. R. Tan, Y. Azuma and I. Kojima, *Appl. Surf. Sci.*, 2005, **241**, 135-140.
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