Electronic supplementary information

for

Asymmetric *tris*-Heteroleptic Iridium(III) Complexes Containing Three Different 2-Phenylpyridine-Type Ligands: A New Strategy for Improving Electroluminescent Ability of Phosphorescent Emitters[†]

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Experimental



Scheme S1Synthetic scheme for the organic ligands.

2-bromo-9,9-diethylfluorene.

To a stirred solution of 2-bromofluorene (5.00 g, 20.40 mmol) in 50 ml DMSO, NaOH (2.44 g, 61.00 mmol) was added. After 5 min of stirring, bromoethane (14.00 g, 44.00 mmol) was added slowly. The reaction mixture were stirred continuously for 24 h at 90 °C. Then the reaction mixture was poured into cold water (100 mL), and extracted with dichloromethane (3×30 mL). The organic phase was washed with water, and dried over MgSO₄ before the solvent was evaporated. Then the mixture was purified by column chromatograph with hexane. The product was obtained as a white solid (5.40 g, Yield: 88%). ¹H NMR (400MHz, CDCl₃): δ (ppm) 7.69–7.65 (m, 1H), 7.57–7.55 (m, 1H), 7.47–7.44 (m, 2H), 7.34–7.29 (m, 3H), 2.06–1.96 (m, 4H), 0.32 (t, J = 7.4 Hz, 6H). ¹³C NMR (100MHz, CDCl₃): δ (ppm) 152.13, 149.45, 140.52, 140.41, 129.91, 127.45, 126.98, 126.19, 122.89, 120.96, 119.69, 99.96, 56.40, 32.67, 8.41.

2-Bromo-9,9-diethyl-7-iodofluorene.

A mixture of 2-bromo-9,9-diethylfluorene (4.5 g, 15.00 mmol), iodine (1.52 g, 6.00 mmol), iodic acid(0.6 g, 3.41mmol), concentrated sulfuric acid (0.8 mL), acetic acid (40 mL), water (4 mL) and carbon tetrachloride (5 mL) was maintained at 80-85 °C for 2.5 h with mechanical stirring. The product slurry was cooled to room temperature and filtered, and the solids was washed with acetic acid-water (1:1) and water. The product was obtained as light yellow solid after dried (5.71 g, Yield: 92%). ¹H NMR (400MHz, CDCl₃): δ (ppm) 7.68–7.64 (m, 1H), 7.57–7.54 (m, 1H), 7.47–7.40 (m, 3H), 7.35–7.30 (m, 1H), 2.04–1.96 (m, 4H), 0.31 (t, J = 7.4 Hz, 6H). ¹³C NMR (100MHz, CDCl₃): δ (ppm) 151.84, 151.49, 140.04, 139.50, 136.10, 132.12, 130.09, 126.16, 121.38, 121.10, 92.89.

L1-Br.

Under N₂ atmosphere, the mixture of 2-bromo-9,9-diethyl-7-iodofluorene (4.50 g, 10.50 mmol), diphenylamine (1.96 g, 11.60 mmol), phenanthroline (0.34 mg, 1.86 mmol), CuI (0.22 mg, 1.18 mmol) and p-xylene (30 mL) was heated to 100 °C for 15 min. Then KOH (4.48 g, 0.08 mol) were added. The reaction mixture was stirred vigorously and refluxed for 2 days. After the solution was cooled to room temperature, glacial acetic acid (10 g) was added. The mixture was dissolved in toluene (100 mL) and washed with water (100 mL). Then the organic layer was washed with conc. Na₂CO₃ (100 mL) and water (100 mL) and dried over MgSO₄. The solvent was removed and the crude product was purified by column chromatography with hexane. The product was obtained as white solid (3.12 g, Yield: 64%). ¹H NMR (400MHz, CDCl₃): δ (ppm) 7.52 (d, J = 8.4 Hz, 1H), 7.46 (d, J = 8.0 Hz, 1H), 7.43–7.40 (m, 3H), 7.25–7.23 (m, 4H), 7.11 (d, J = 7.6 Hz, 4H), 7.07–7.01 (m, 3H), 1.94–1.84 (m, 4H), 0.35 (t, J = 7.4Hz, 6H). ¹³C NMR (100MHz, CDCl₃): δ (ppm) 152.01. 151.10, 147.97, 147.50, 140.43, 135.63, 130.22, 129.50, 129.32, 129.10, 126.13, 124.23, 124.05, 123.79, 123.73, 123.63, 122.90, 122.72, 122.45, 120.51, 120.43, 120.20, 119.11, 56.41, 32.63, 8.60.

Under N₂ atmosphere, **L1-Br** (2.8 g, 5.98 mmol) and 2-tributylstannanyl-pyridine (2.42 g, 6.58 mmol) and Pd(PPh₃)₄ (0.35 g, 0.30 mmol) were added to 30 mL dry toluene. The reaction mixture was stirred at 110 °C for 24 h. After it was cooled to room temperature, the reaction mixture was poured into separation funnel. Dichloromethane (50 mL) was added and washed with water (3×100 mL). The organic phase was dried over MgSO₄. Then solvent was removed and the crude product was purified by column chromatography with dichloromethane-hexane (1:1) as eluent. The product was obtained as yellow solid (1.67 g, Yield: 60%). ¹H NMR (400MHz, CDCl₃): δ (ppm) 8.70 (d, J = 4.6 Hz, 1H), 7.98–7.58 (m, 6H), 7.27–6.97 (m, 13H), 2.07–1.91 (m, 4H), 0.38 (t, J = 7.3 Hz, 6H). ¹³C NMR (100MHz, CDCl₃): δ (ppm) 157.64, 151.71, 150.31, 149.44, 147.81, 147.30, 142.18, 137.31, 136.49, 135.98, 129.05, 125.81, 123.76, 123.44, 122.42, 121.60, 121.01, 120.54, 120.38, 119.15, 119.13, 56.27, 32.72, 8.71.

L2-Br.

The synthetic method was similar to **L1-Br**. **L2-Br** was obtained as yellow solid (0.17 g, Yield: 60%) using 2-tributylstannanyl-pyridine (2.42 g, 6.58 mmol) as raw material. ¹H NMR (400MHz, CDCl₃): δ (ppm) 7.44 (t, J = 8.0 Hz, 2H), 7.39 (d, J = 9.6 Hz, 2H), 7.05 (d, J = 8.8 Hz, 4H), 6.93 (s, 1H), 6.88 (m, 1H), 6.82 (d, J = 8.8 Hz, 4H), 3.87 (s, 6H), 1.94–1.79 (m, 4H), 0.34 (t, J = 7.4 Hz, 6H). ¹³C NMR (100MHz, CDCl₃): δ (ppm) 155.45, 151.75, 150.68, 148.49, 141.20, 140.48, 133.43, 129.75, 125.93, 125.88, 120.50, 120.03, 119.94, 119.45, 115.99, 114.50, 56.30, 32.63, 8.64.

L2

The synthetic method was similar to L1. L2 was obtained as orange solid (0.67 g, Yield: 56%) using bis(4-methoxyphenyl)amine (1.20 g, 2.27mmol) as raw material. ¹H NMR (400MHz, CDCl₃): δ

(ppm) 8.71 (d, J = 4.4 Hz, 1H), 7.95–7.92 (m, 2H), 7.80–7.72 (m, 2H), 7.66 (d, J = 8.0 Hz, 1H), 7.53 (d, J = 8.0 Hz, 1H), 7.22–7.19 (m, 1H), 7.07 (d, J = 8.8 Hz, 4H), 6.99 (s, 1H), 6.92–6.90 (m, 1H), 6.85–6.81 (m, 4H), 3.81(s, 6H), 2.07–1.87 (m, 4H), 0.37 (t, J = 7.4 Hz, 6H). ¹³C NMR (100MHz, CDCl₃): δ (ppm) 157.76, 155.32, 151.63, 150.10, 149.43, 148.38, 142.52, 141.28, 136.86, 136.49, 134.18, 125.88, 125.77, 121.54, 120.95, 120.56, 120.33, 120.29, 118.80, 116.19, 114.53, 56.72, 55.47, 32.74, 8.68.

L3-Br

4-Bromothiophenol (1.00g, 5.29mmol) and iodobenzene (1.19g, 5.82mmol) were mixed in dioxane under a N₂ atmosphere. Then, KOH (0.59g, 10.60mmol), CuI (0.10g, 0.53mmol) and glycine (0.08g, 1.06mmol) were added into the reaction mixture. The mixture was heated to 110 °C for 16 h. After cooling to room temperature, water was added and the mixture was extracted three times with CH₂Cl₂. The organic layer was collected and dried over MgSO₄. After the solvent was removed, the residue was dissolved in CHCl₃ and m-chloroperoxybenzoic acid (mCPBA, 1.47g, 8.52mmol) was added slowly at 0 °C. After the addition, the reaction mixture was warmed to room temperature. slowly and stirred for 2 days to ensure completion of the reaction. Chloroform (50 mL) was added and the mixture was washed with 2 M sodium thiosulfate solution (3*100 mL) and water (3*100 mL). The organic phase was separated and dried over MgSO₄. The solvent was removed under reduced pressure and the residue was purified by column chromatography using CH₂Cl₂ as eluent. The product was obtained as white crystals (0.94 g, 80%). ¹H NMR (400MHz, CDCl₃): δ (ppm) 7.93–7.91 (m, 2H), 7.80–7.78 (m, 2H), 7.63–7.48 (m, 5H). ¹³C NMR (100MHz, CDCl₃): δ (ppm) 140.98, 140.53, 133.40, 132.50, 129.34, 129.09, 128.36, 127.54.

L3

Under N₂ atmosphere, **L3-Br** (0.59 g, 1.98mmol) and 2-(tributylstannyl) pyridine (0.80g, 2.17mmol) were mixed in dry toluene (50 mL), and Pd(PPh₃)₄ (0.114g, 0.10mmol) was added. The resulting mixture was stirred at 110 °C for 24 h. After cooling to room temperature, the reaction mixture was poured into a separating funnel and CH₂Cl₂ (50mL) was added followed by washing with water (3*100 mL). The organic phase was dried over MgSO₄. The solvent was then removed and the residue was purified by column chromatography eluting with CH₂Cl₂/ ethyl acetate (15:1, v/v). The product was obtained as a white solid (0.33 g, 65%). ¹H NMR (400MHz, CDCl₃): δ (ppm) 8.71 (d, J = 4.8 Hz, 1H), 8.13 (d, J = 8.4 Hz, 2H), 8.04 (d, J = 8.8 Hz, 2H), 7.98–7.96 (m, 2H), 7.81–7.74 (m, 2H), 7.59–7.49 (m, 3H), 7.32–7.29 (m, 1H). ¹³C NMR (100MHz, CDCl₃): δ (ppm) 155.28, 149.99, 143.97, 141.55, 141.49, 137.02, 133.21, 129.28, 128.14, 127.71, 127.61, 123.26, 121.05.



(a)

S7



(b)



S8



(**d**)



(e)





(g)



(h)

Fig. S1 The ¹H,¹³C NMR spectra of Ir3-1, Ir3-2, Ir2-1 and Ir2-2.



Fig. S2 TGA curves for the phosphorescent Ir(III) complexes.



Fig. S3 PL spectra for the phosphorescent Ir(III) complexes measured in PMMA film (6% wt doping

level) at 298 K.



Fig. S4 PL spectra for the phosphorescent Ir(III) complexes measured in CH₂Cl₂ solution at 77 K.





Fig. S5 EL spectra for the OLEDs except the optimized ones at ca. 10 V.





Fig. S6 The Current density-Voltage-Luminance(*J*-*V*-*L*) curves for the devices except the optimized ones.









Fig. S7 Relationship between the electroluminescence efficiency and luminance for the devices except the optimized ones.

complexes	coordinate covalent bonds	bond length(Å)
I2 1	Ir–C (ppy)	2.02865
113-1	Ir-N (ppy)	2.18316
L-2 0	Ir–C (ppy)	2.02817
113-2	Ir–N (ppy)	2.18394
I2 1	Ir-O1 (acac)	2.19689
112-1	Ir–O2 (acac)	2.20403
I-2 2	Ir-O1 (acac)	2.19751
1f2-2	Ir–O2 (acac)	2.20506

Table 1 The coordinate covalent bonds length based on the TD-DFT calculations