Electronic Supplementary Information

Efficient flash memory devices Based on non-conjugated ferrocenecontaining copolymers

Jing Xiang, Xiangling Li, Yun Ma, Qiang Zhao,* Cheuk-Lam Ho,* Wai-Yeung Wong*

Synthesis of VM1

4-Di-*p*-tolylamino-benzaldehyde (860 mg, 2.85 mmol) in dry THF was added into a mixture solution of potassium *t*-butoxide (480 mg, 4.28 mmol) and methyltriphenyl phosphonium iodide (1.73 g, 4.28 mmol). They were then stirred at r.t. for 24 h. The solution was poured into ice water and extracted with ethyl acetate. The organic layer was concentrated under reduced pressure and the product was then purified by column chromatography (silica). 80% yield. ¹H NMR (acetone-*d*₆, 400 MHz): δ 7.33~7.30 (m, 2H, Ar), 7.12~7.09 (m, 4H, Ar), 6.96~6.89 (m, 6H, Ar), 6.67 (dd, 1H, *J*₁ = 17.6 Hz, *J*₂ = 10.8 Hz, vinyl CH), 5.65 (dd, 1H, *J*₁ = 17.6 Hz, *J*₂ = 0.8 Hz, *trans*-vinyl C=CH₂), 5.10 (dd, 1H, *J*₁ = 10.8 Hz, *J*₂ = 0.8 Hz, *cis*-vinyl C=CH₂). ¹³C NMR (acetone-*d*₆, 100 MHz): δ 148.86, 146.07, 137.32, 133.48, 131.94, 130.86, 127.91, 125.52 (Ar), 122.98 (vinyl CH), 112.02 (vinyl CH₂), 21.01 (CH₃).

Synthesis of VM2

80% yield. ¹H NMR (CDCl₃, 400 MHz): δ 8.04 (dd, 3H, $J_1 = 12.3$, $J_2 = 8.1$ Hz, Ar), 7.85 (d, 1H, J = 8.0 Hz), 7.47 (dd, 3H, J = 11.2, 7.8 Hz, Ar), 7.35 (t, 1H, J = 7.6 Hz, Ar), 6.73 (dd, 1H, $J_1 = 17.6$, $J_2 = 10.9$ Hz, vinyl CH), 5.83 (d, 1H, J = 17.6 Hz, transvinyl C=CH₂), 5.34 (d, 1H, J = 10.9 Hz, cis-vinyl C=CH₂). ¹³C NMR (CDCl₃, 100 MHz): δ 167.64, 154.18, 140.13, 136.06, 135.03, 132.88, 127.75, 126.78, 126.34, 125.18, 123.19 (Ar), 121.60 (vinyl CH), 115.62 (vinyl CH₂).

Synthesis of VM3

83% yield. ¹H NMR (tetrahydrofuran- d_8 , 400 MHz): δ 7.21~7.18 (m, 2H, Ar), 7.15~7.13 (m, 1H, Ar), 7.08~7.06 (m, 1H, Ar), 6.90~6.86 (m, 2H, Ar), 6.83~6.81 (m, 1H, Ar), 6.59 (dd, 1H, J_1 = 17.6 Hz, J_2 = 10.8 Hz, vinyl CH), 5.62 (dd, 1H, J_1 = 17.6 Hz, J_2 = 0.8 Hz, trans-vinyl C=CH₂), 5.07 (dd, 1H, J_1 = 10.8 Hz, J_2 = 0.8 Hz, cis-vinyl C=CH₂), 3.34 (s, 3H, CH₃). ¹³C NMR (tetrahydrofuran- d_8 , 100 MHz): δ 143.71, 143.54, 133.82, 130.33, 125.32, 124.76, 123.61, 122.21, 121.60, 121.10, 120.26, 112.12 (Ar), 111.92 (vinyl CH), 109.19 (vinyl CH₂), 32.72 (CH₃).

Synthesis of FcCP1

VM1 (278 mg, 0.929 mmol), vinylferrocene (197 mg, 0.929 mmol) and AIBN (30 mg,

0.18 mmol) were delivered into a Fisher-Porter aerosol compatibility tube equipped with a valve. 8 mL dry toluene was then added into the tube to dissolve the mixture followed by degassing using three alternate freeze-thaw cycles. After degassing, the tube was placed in a constant-temperature bath controlled to 70 °C for 24 h. The reacted solution was added dropwise to hexane with rapid stirring to precipitate the copolymer. It was filtered and precipitated in hexane again followed by drying under vacuum to obtain the pure product in 50% yield. ¹H NMR (CDCl₃, 400 MHz): δ 6.84 (br, 36H, triphenylamine H), 3.85 (br, 9H, Fc), 2.19 (br, 19H, CH₃). IR (KBr pellet, cm⁻¹): 3089 (C-H stretching, Fc), 1605 (C=C stretching, Ar), 1502 (C=C stretching, Ar), 1448 (C=C stretching, Ar), 1412 (C=C stretching, Fc), 1270 (C-N stretching), 1106 (C-C stretching, Fc), 1000 (C-C stretching, Fc), 813 (C-H bending, Fc), 504 (C-Fe stretching, Fc).

Synthesis of FcCP2

The synthetic procedures of **FcCP1** were applied using **VM2** instead of **VM1** to afford **FcCP2** in 60% yield. ¹H NMR (CD₂Cl₂, 400 MHz): δ 7.90~6.76 (m, 23H, Ar), 4.00 (br, 9H, Fc). IR (KBr pellet, cm⁻¹): 3060 (C-H stretching, Fc), 1607 (C=C stretching, Ar), 1562 (C=C stretching, Ar), 1483 (C=C stretching, Ar), 1453 (C=C stretching, Ar), 1413 (C=C stretching, Fc), 1107 (C-C stretching, Fc), 483 (C-Fe stretching, Fc).

Synthesis of FcCP3

The synthetic procedures of FcCP1 were applied using VM3 instead of VM1 to get FcCP3 in 58% yield. ¹H NMR (CD₂Cl₂, 400 MHz): δ 7.23~6.56 (m, 24H, Ar), 4.04~3.71 (m, 9H, Fc), 3.25 (br, 10H, CH₃). IR (KBr pellet, cm⁻¹): 3058 (C-H stretching, Fc), 1879 (C-H oop bending, para-substituted-Ar), 1763 (C-H oop bending, para-substituted-Ar), 1763 (C-H oop bending, para-substituted-Ar), 1602 (C=C stretching, Ar), 1576 (C=C stretching, Ar), 1497 (C=C stretching, Ar), 1469 (C=C stretching, Ar), 1256 (C-N stretching), 1109 (C-C stretching, Fc), 1000 (C-C stretching, Fc), 813 (C-H bending, Fc), 485 (C-Fe stretching, Fc).

 Table S1
 Compositions of FcCP1-FcCP3 calculated from ¹H NMR spectra

Polymer	ferrocene unit: aromatic moiety (mole ratio)
FcCP1	0.24 : 0.76
FcCP2	0.26: 0.74
FcCP3	0.23 : 0.77



Figure S1. IR spectra of FcCP1-FcCP3 using KBr pellet method



Figure S2. TGA curves of FcCP1-FcCP3



Figure S3. Cyclic voltammograms of FcCP1-FcCP3 in thin layer electrodes on glassy carbon electrode under 0.1 M *n*-Bu₄NPF₆ acetonitrile solution at a scan rate of 50 mV s⁻¹



Figure S4. ¹H NMR spectrum of FcCP1 in CDCl₃



Figure S5. ¹H NMR spectrum of FcCP2 in CDCl₃



Figure S6. ¹H NMR spectrum of FcCP3 in CDCl₃