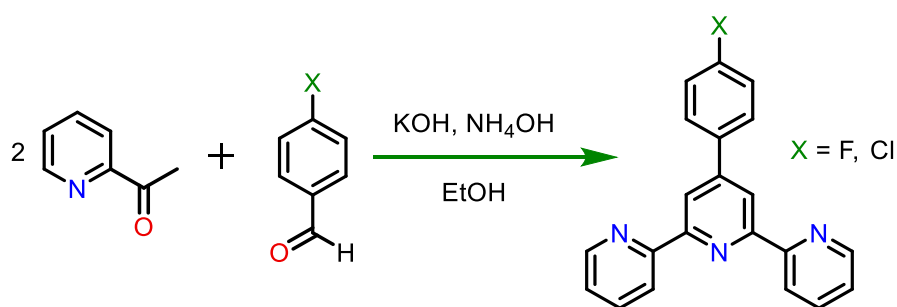


Electronic Supplementary Information

White-Light-Emitting Lanthanide and Lanthanide-Iridium Doped Supramolecular Gels: Modular Luminescence and Stimuli-Responsive Behaviour

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Scheme S1. Facile one-pot synthesis of 4'-p-halophenyl-2,2':6',2''-terpyridine

Characterization of L-F and L-Cl

L-F

Yield: (3.588 g, 45.67%). ¹H NMR (500 MHz, CDCl₃) δ 8.71 (d, *J* = 4.6 Hz, 2H), 8.67 (s, 2H), 8.65 (d, *J* = 8.0 Hz, 2H), 7.90–7.83 (m, 4H), 7.33 (dd, *J* = 7.0, 5.0 Hz, 2H), 7.22–7.13 (m, 2H). HRMS (ESI): calcd for C₂₁H₁₅N₃F [M+H]⁺ 328.1250 found 328.1289 and calcd for C₂₁H₁₄N₃FNa [M+Na]⁺ 350.1069 found 350.1091.

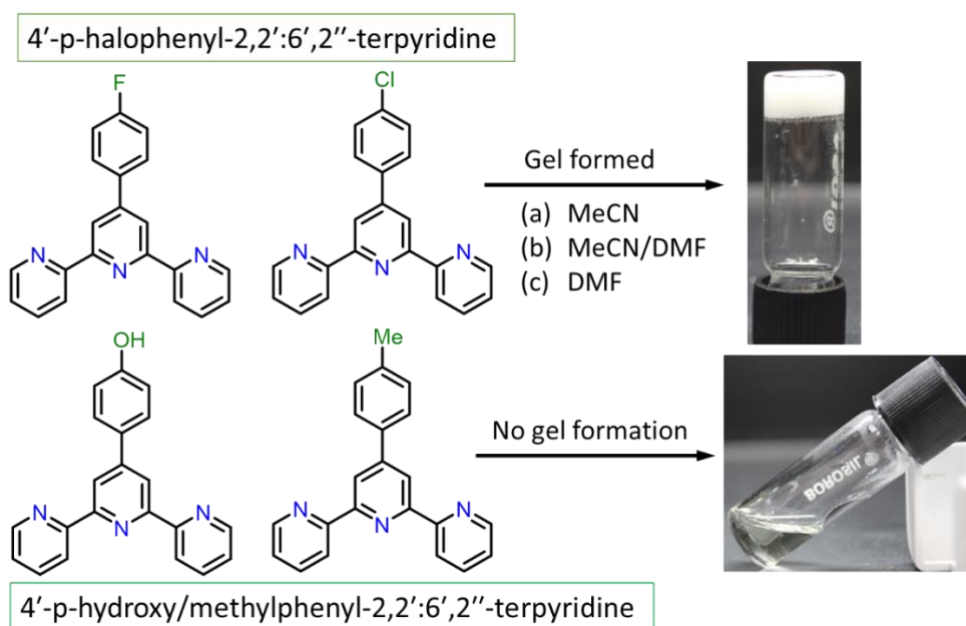
L-Cl

Yield: (3.490 g, 42.30%). ¹H NMR (500 MHz, CDCl₃) δ 8.71 (d, *J* = 4.4 Hz, 2H), 8.68 (s, 2H), 8.64 (d, *J* = 7.9 Hz, 2H), 7.88–7.79 (m, 4H), 7.46 (d, *J* = 8.6 Hz, 2H), 7.36–7.30 (m, 2H). HRMS (ESI): calcd for C₂₁H₁₅N₃Cl [M+H]⁺ 344.0955 found 344.0967.

Table S1. Crystal Parameters and Refinement Data for **L-F**.

Compound reference	L-F
Chemical formula	C ₂₁ H ₁₄ N ₃ F
Formula Mass	327.35
Crystal system	Monoclinic
Space group	<i>P</i> 2 ₁ / <i>c</i>
Crystal color	Colourless
Crystal size/mm ³	0.49×0.08×0.06
<i>a</i> /Å	3.8357(9)
<i>b</i> /Å	21.751(5)
<i>c</i> /Å	18.396(5)
α /°	90.0
β /°	92.713(4)
γ /°	90.0
<i>V</i> /Å ³	1533.0(6)
<i>Z</i>	4
<i>D_c</i> /g cm ⁻³	1.418
μ (mm ⁻¹)	0.094
<i>F</i> (000)	680
<i>T</i> /°K	150(2)
Total reflns	7902
<i>R</i> (int)	0.0433
Unique reflns	3001
Observed reflns (<i>I</i> > 2σ(<i>I</i>))	2519
Parameters	282
<i>R</i> ₁ (<i>I</i> > 2σ(<i>I</i>)) <i>wR</i> ₂ (all reflns)	0.0775,0.1546
GOF (<i>F</i> ²)	1.200
CCDC number	1837781

$$R_1 = \Sigma(|F_o| - |F_c|) / \Sigma|F_o|, \quad wR_2 = [\Sigma w(|F_o| - |F_c|)^2 / \Sigma w(F_o)^2]^{1/2}, \quad w = 0.75 / (\sigma^2(F_o) + 0.0010F_o^2)$$



Scheme S2. Role of halogen atom in terpyridine moiety in gel formation

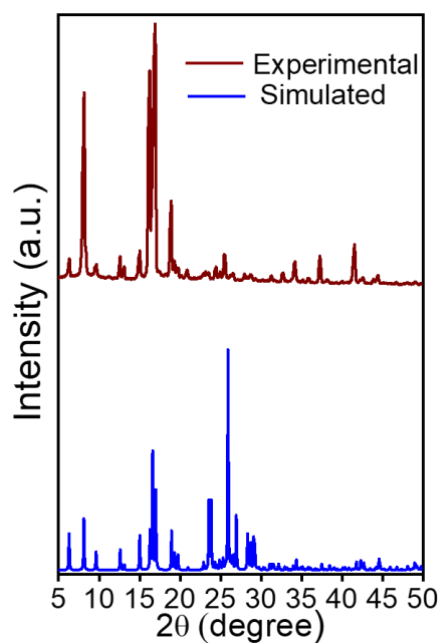


Fig. S1. Comparison of experimental (blue line) and simulated (brown line) PXRD patterns for L-F

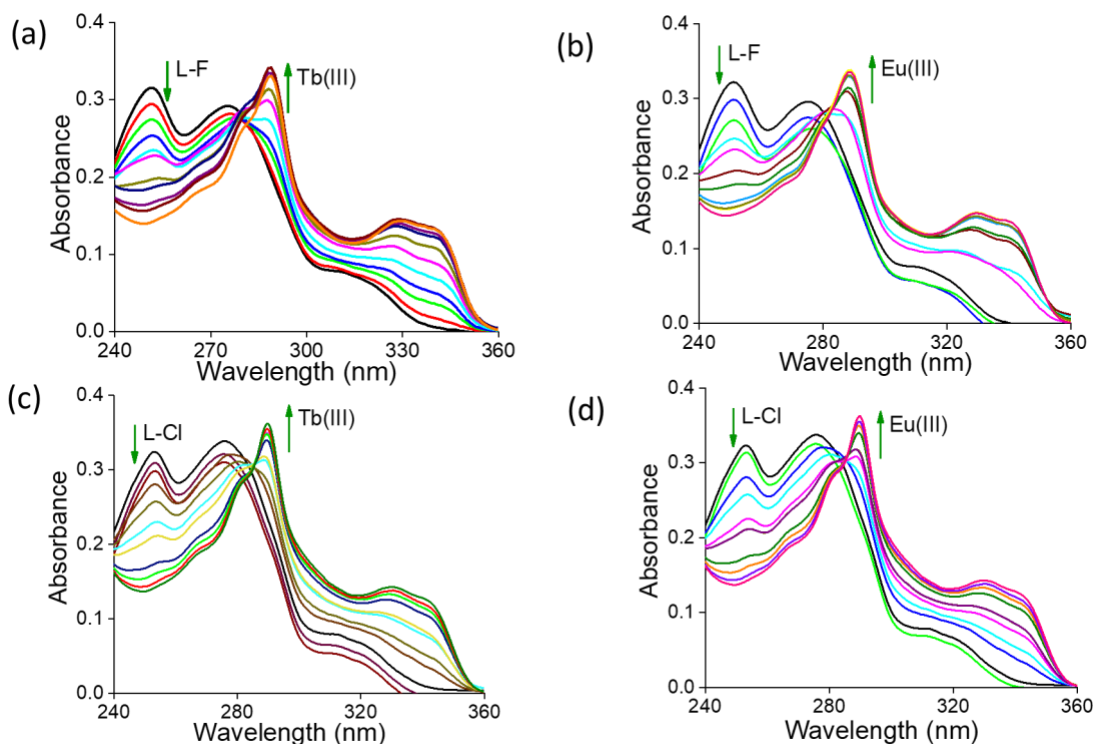


Fig. S2. Changes in the absorption spectra of (a and b) L-F (1×10^{-5} M) after addition of Tb(III) and Eu(III) (0 $\rightarrow$ 1 equiv) and (c and d) L-Cl (1×10^{-5} M) after addition of Tb(III) and Eu(III) (0 $\rightarrow$ 1 equiv) in MeCN.

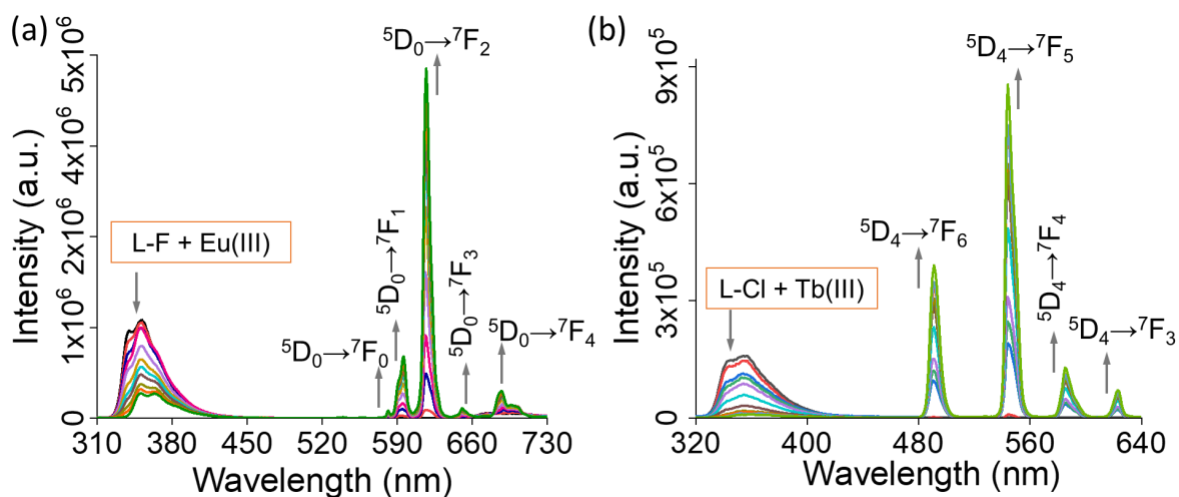
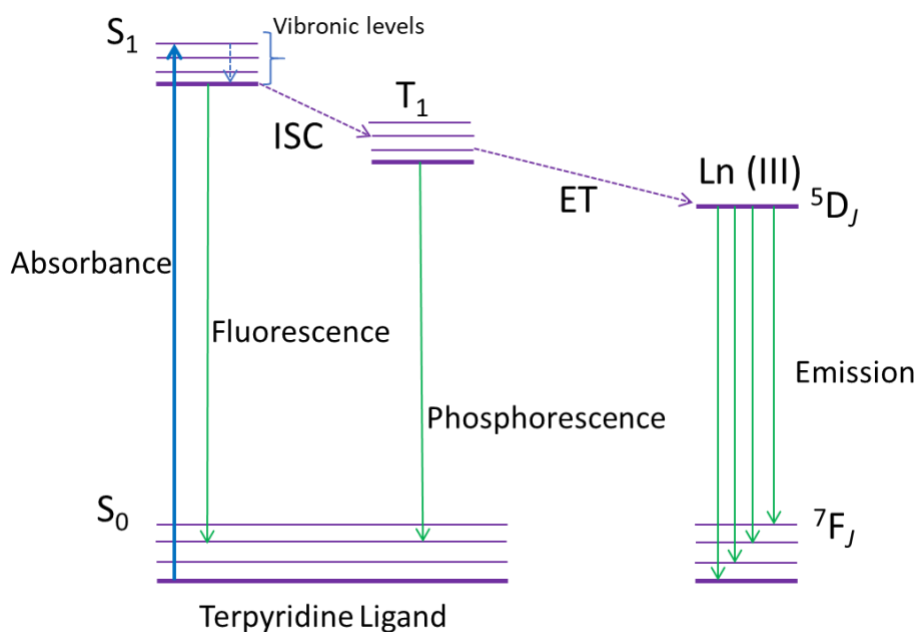


Fig. S3. Spectral changes in (a) the Eu(III) and (b) Tb(III) centred luminescence spectra upon titrating L-F or L-Cl (1.0×10^{-5} M) with Tb(III)/Eu(III) (0 $\rightarrow$ 1 mole equiv) in MeCN.



Scheme S3. Characteristic energy level diagram describing a general mechanism for emissive chromophore-appended lanthanide complexes sensitized through a ligand-centred triplet excited state (ISC- intersystem crossing, ET- Energy Transfer).

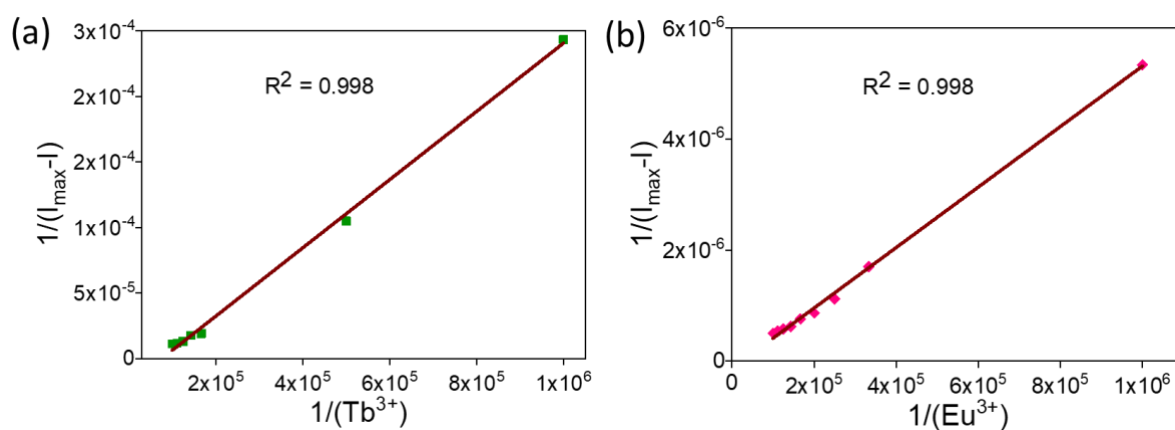


Fig. S4. Benesi-Hildebrand (B-H) plots obtained from the emission titration of (a) L-F ($1 \times 10^{-5} \text{ M}$) with Tb(III) and (b) L-Cl ($1 \times 10^{-5} \text{ M}$) with Eu(III) supported 1:1 binding stoichiometry ($R^2 = 0.998$).

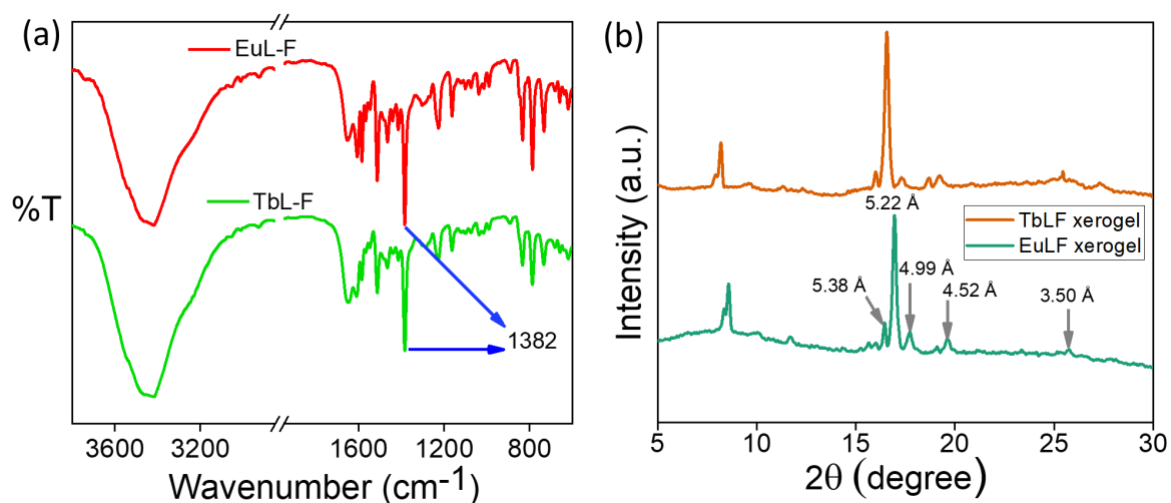
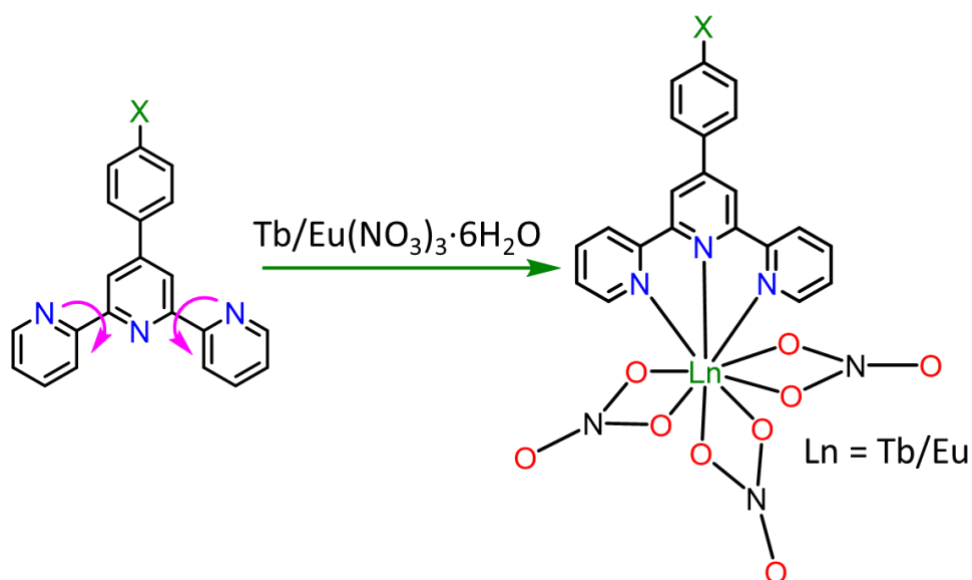


Fig. S5. (a) Comparison of (a) FT-IR spectra of Eu•L-F and Tb•L-F xerogels and (b) PXRD patterns of Eu•L-F and Tb•L-F xerogels.



Scheme S4. Probable coordination environment around Eu(III) and Tb(III) in Eu•L-X and Tb•L-X (X = F and Cl)

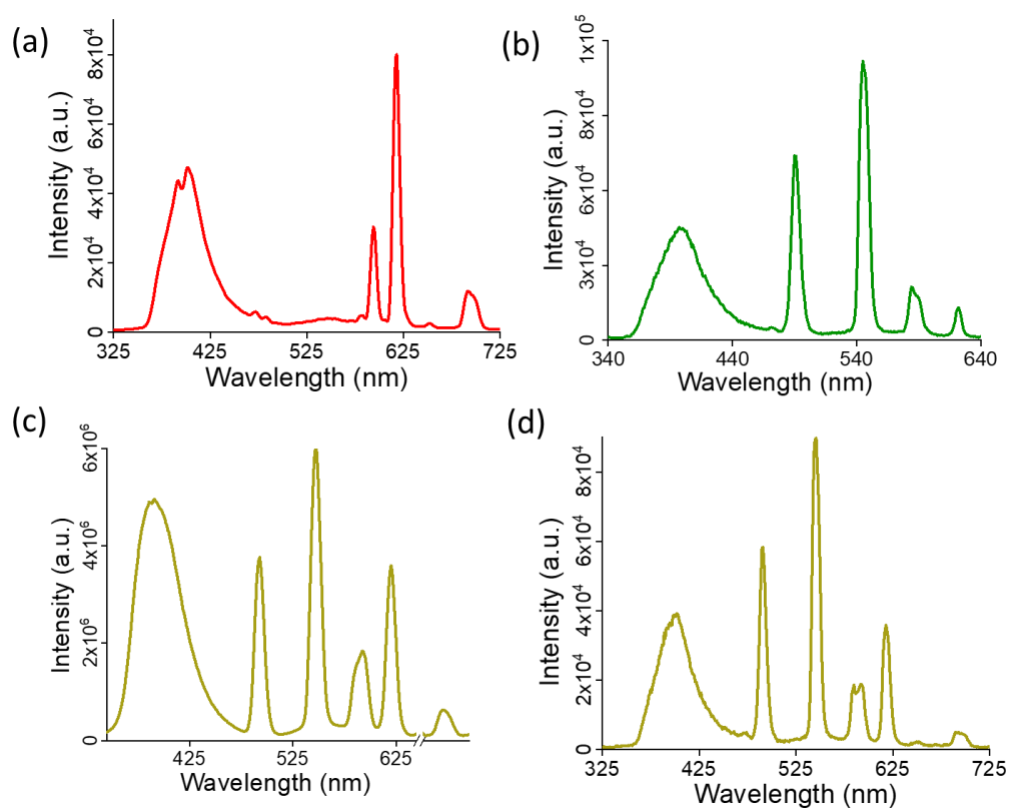


Fig. S6. Luminescence spectra of (a) Eu•L-Cl, (b) Tb•L-Cl, (c) Eu•L-Cl-Tb•L-Cl, and (d) Eu•L-Cl-2(Tb•L-Cl).

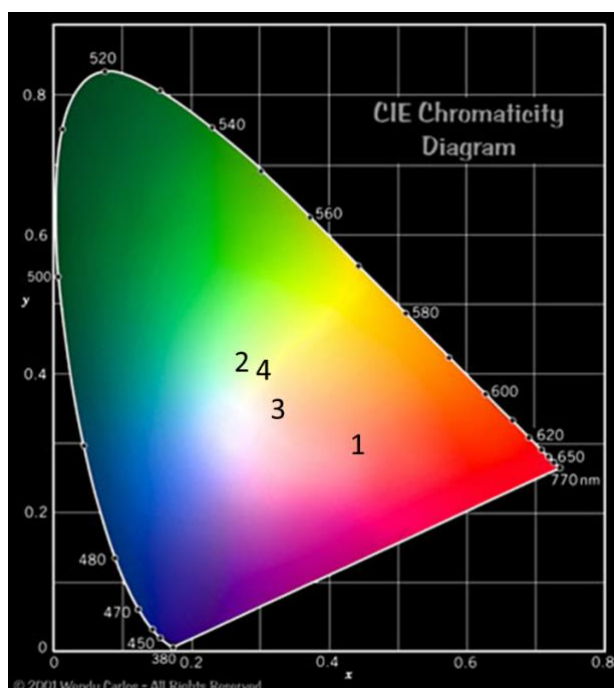
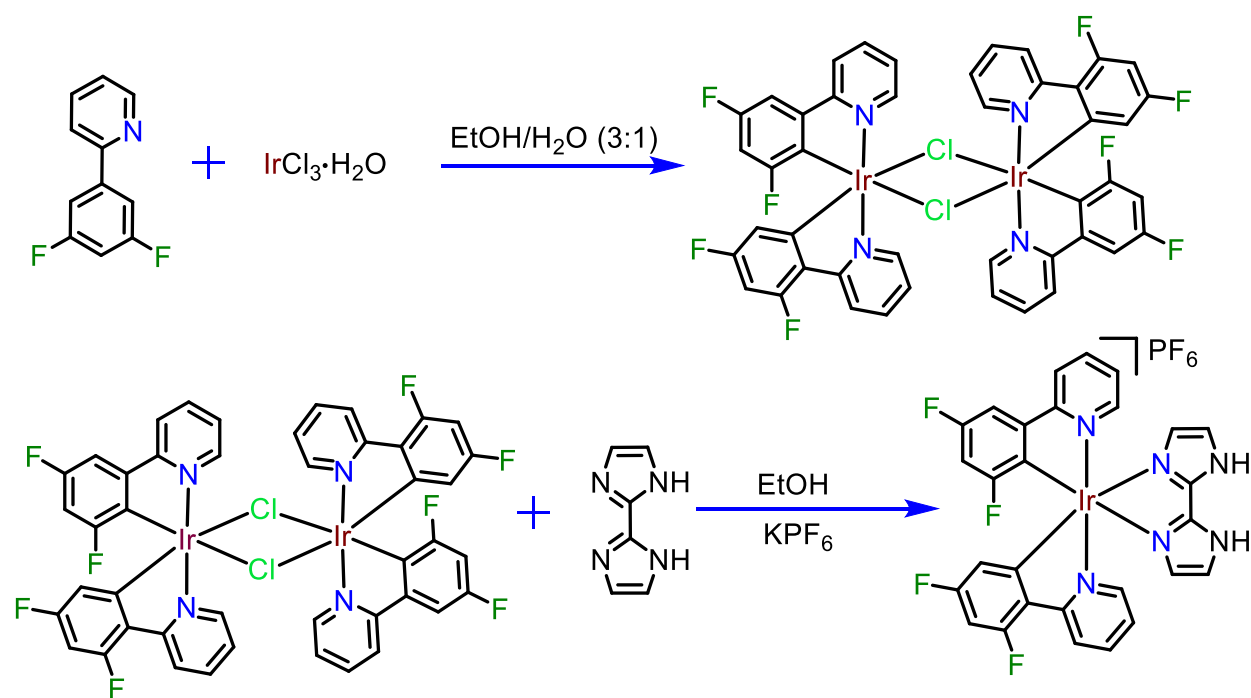


Fig. S7. CIE chromaticity diagram of Eu•L-Cl (1), Tb•L-Cl (2), Eu•L-Cl-Tb•L-Cl (3), and Eu•L-Cl-2(Tb•L-Cl) (4) gels



Scheme S5. Synthetic route for the formation of $[\text{Ir}^{\text{III}}(\text{F}_2\text{ppy})_2(\text{biimid})]\text{PF}_6$ complex

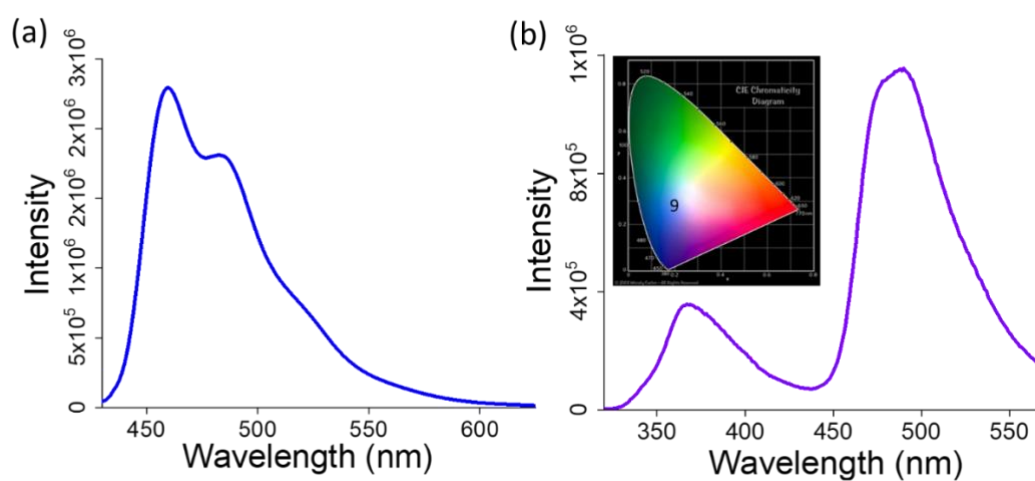


Fig. S8. Luminescence spectra of (a) $[\text{Ir}^{\text{III}}(\text{F}_2\text{ppy})_2(\text{biimid})]\text{PF}_6$ complex in MeCN ($1.0 \times 10^{-5} \text{ M}$) ($\lambda_{\text{ex}} = 385 \text{ nm}$) and (b) Ir-L-F gel ($\lambda_{\text{ex}} = 285 \text{ nm}$). Inset: CIE chromaticity diagram of Ir-L-F gel

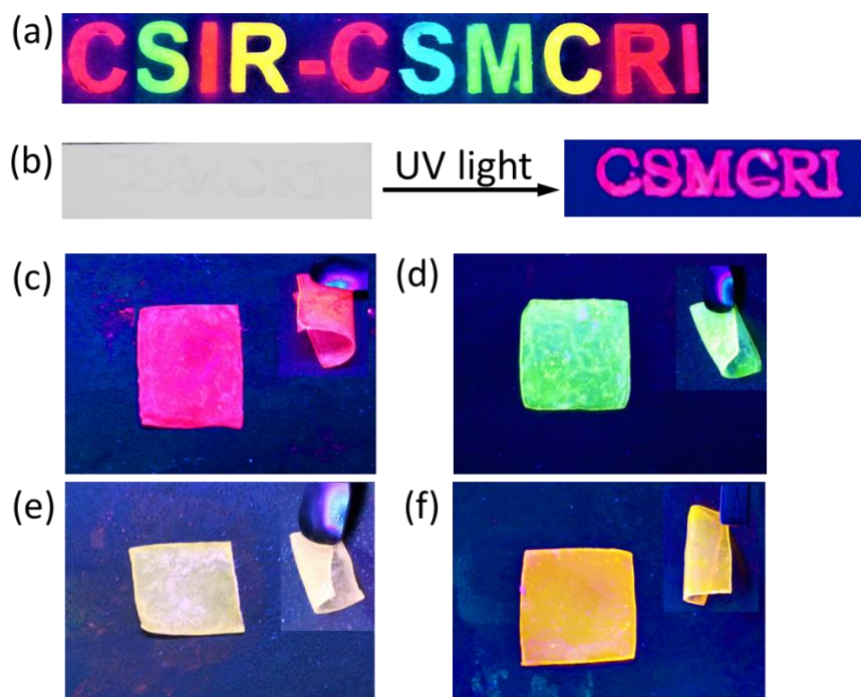


Fig. S9. (a) Photographs of write-up on a nonfluorescent glass plate using the metalloids under UV light irradiation ($\lambda_{\text{ex}} = 365 \text{ nm}$), (b) Photographs of write-up on silica gel plate in daylight and after UV light irradiation, (c-f) The digital images of the (c) Eu•L-F, (d) Tb•L-F, (e) Eu•L-F-Tb•L-F, and (f) Eu•L-F-2(Tb•L-F) coated-agarose gel films. Inset: Corresponding luminescent films after bending

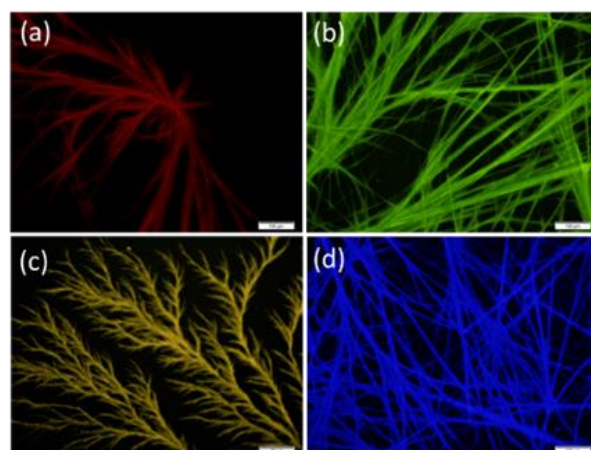


Fig. S10. Fluorescent microscope images of (a) Eu•L-F, (b) Tb•L-F, (c) Eu•L-F-2(Tb•L-F), and (d) Ir-L-F under fluorescence light

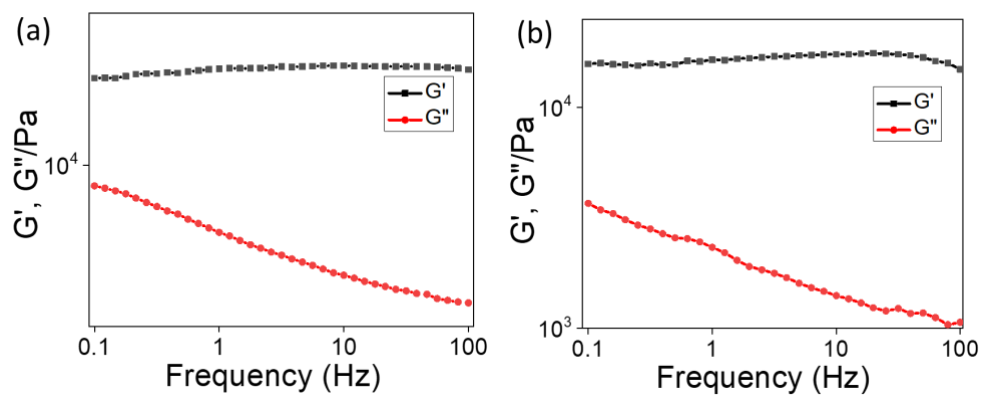


Fig. S11. Frequency sweeps at 0.1% strain amplitude of the storage modulus G' (■) and loss modulus G'' (●) for (a) Eu•L-F and (b) Eu•L-F-Ir gels

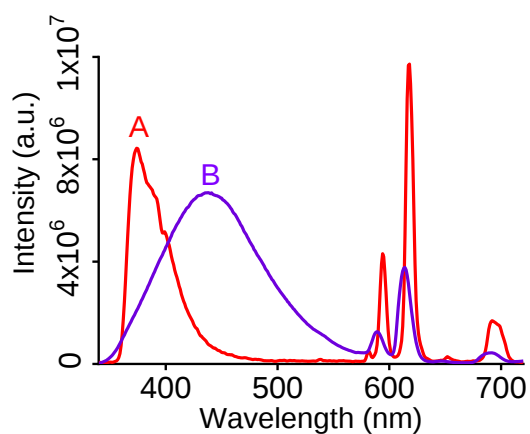


Fig. S12. Assessment of luminescence spectra of (A) Eu•L-F and (B) TFA exposed Eu•L-F sol generated after gel–sol transition