

Kinetics of Dye Regeneration in Liquid Electrolyte Unveils Efficiency of 10.5% in Dye-Sensitized Solar Cells

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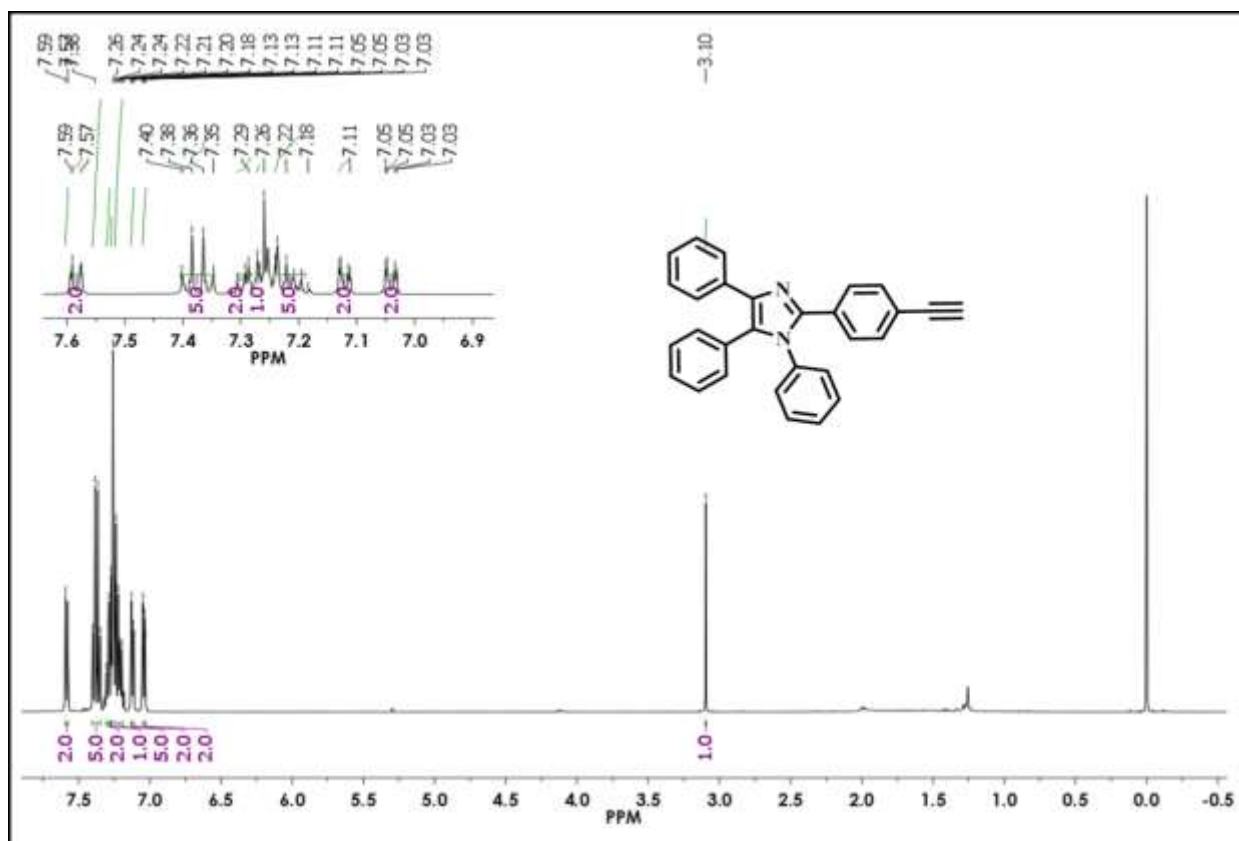


Figure S1. ¹H NMR spectrum (500 MHz, CDCl₃) of 1a.

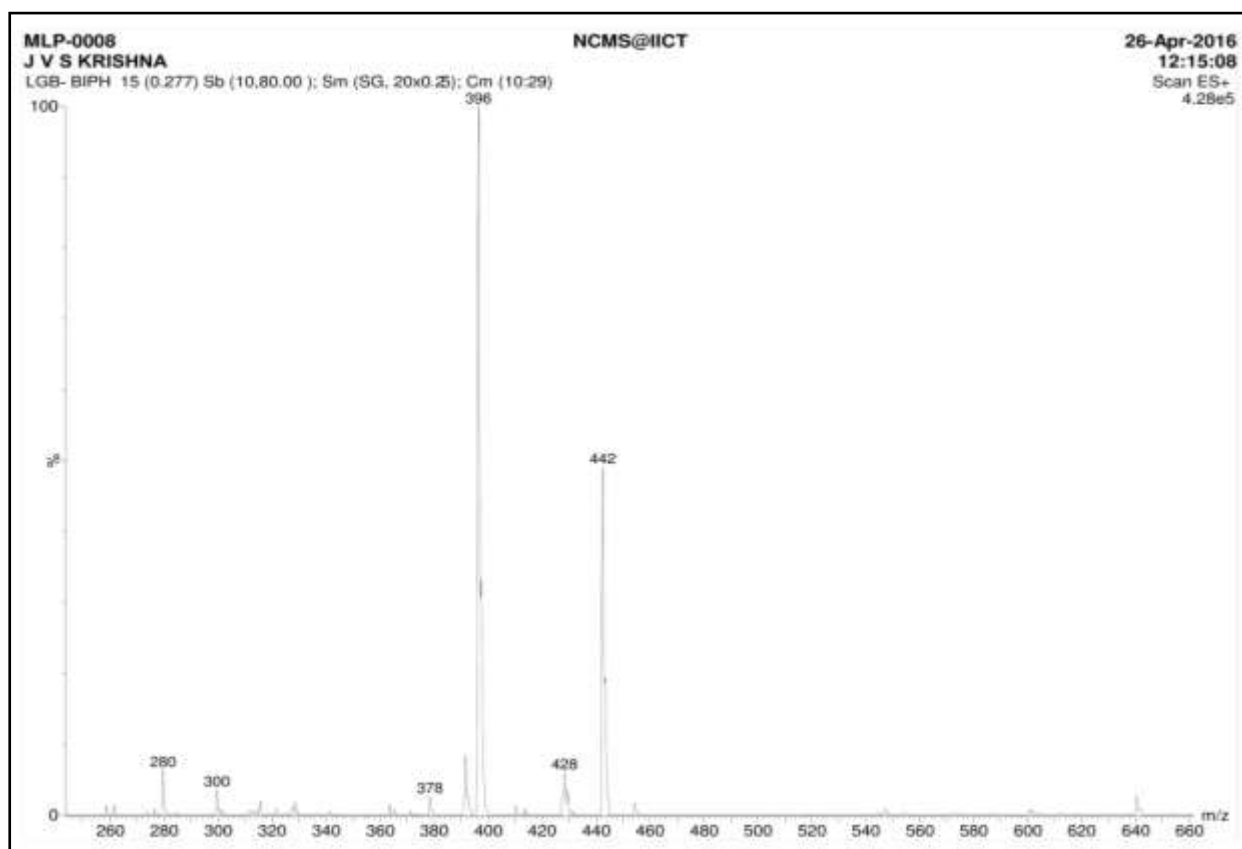


Figure S2. ESI-MS of 1.

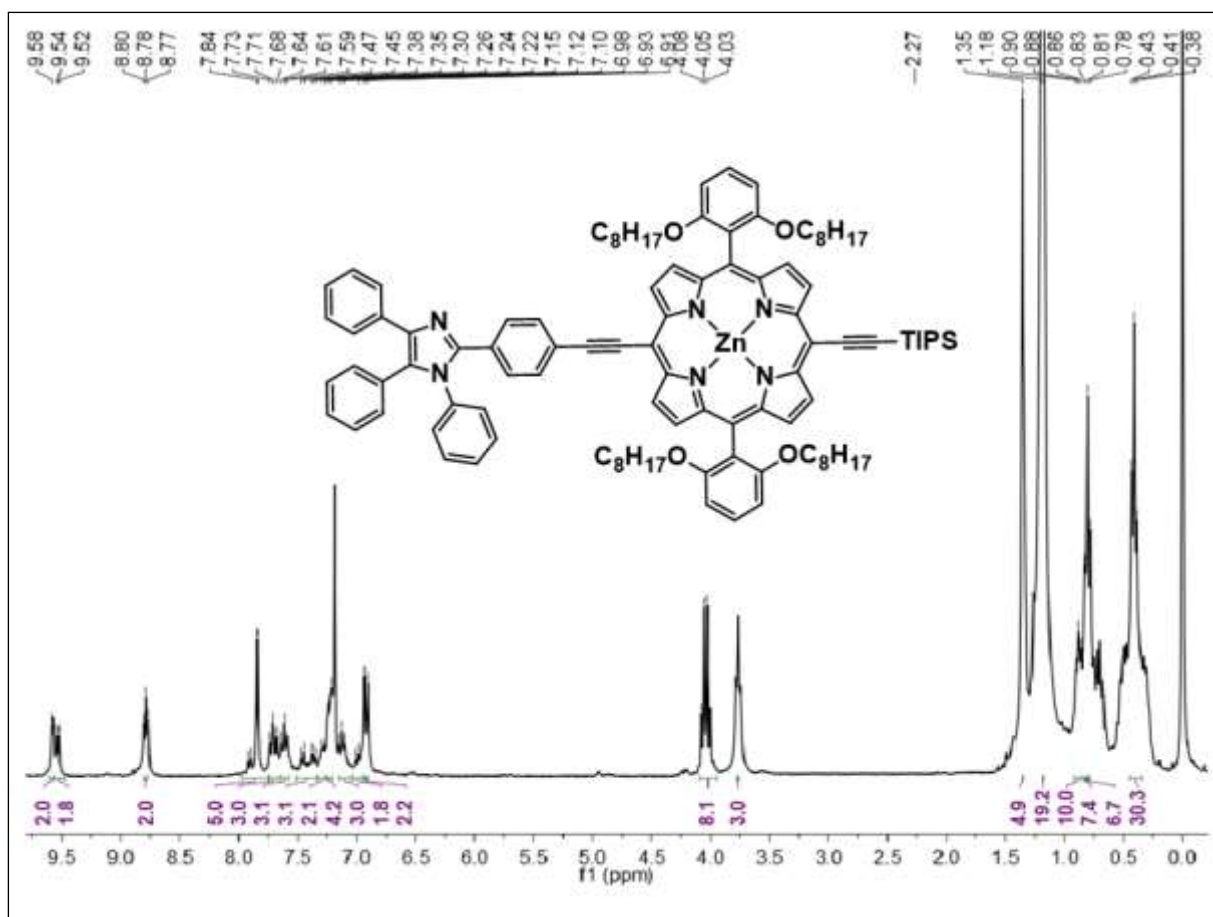


Figure S3. ^1H NMR spectrum (500 MHz, CDCl_3) of 3.

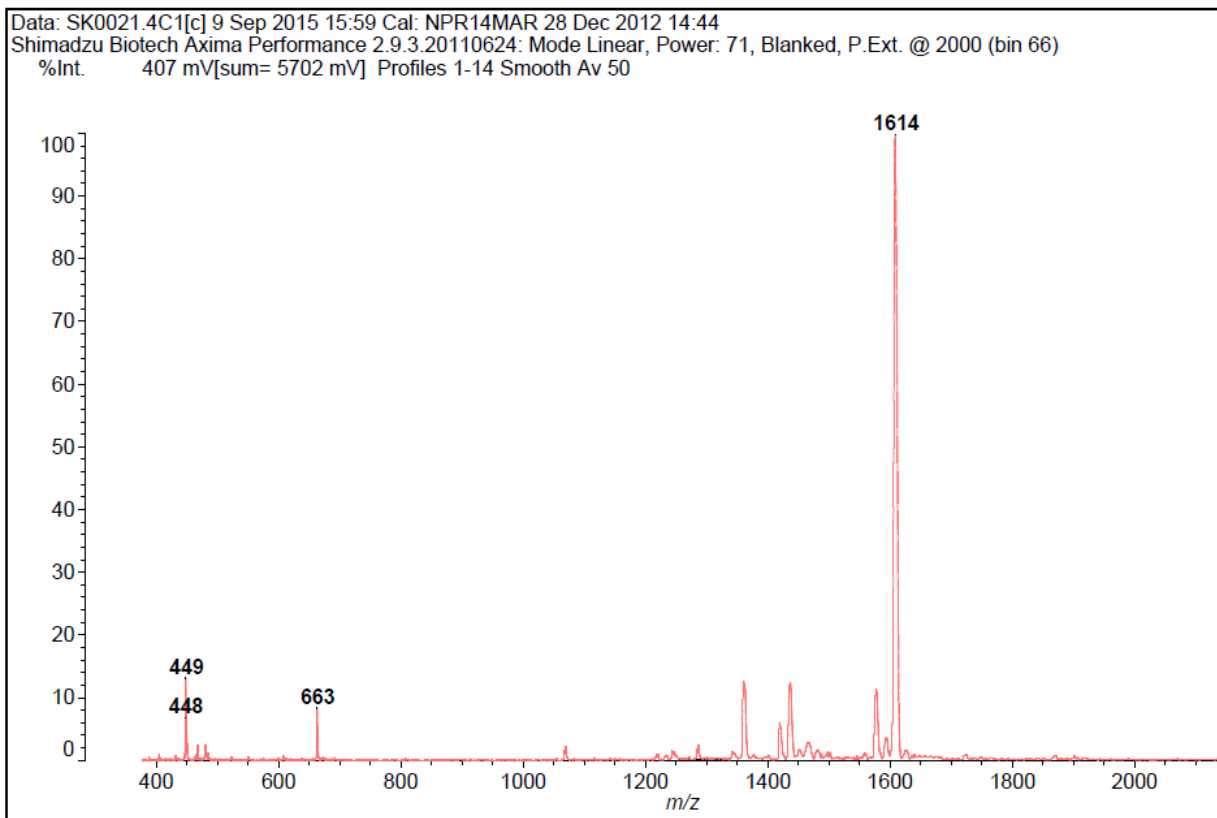


Figure S4. MALDI-TOF of 3.

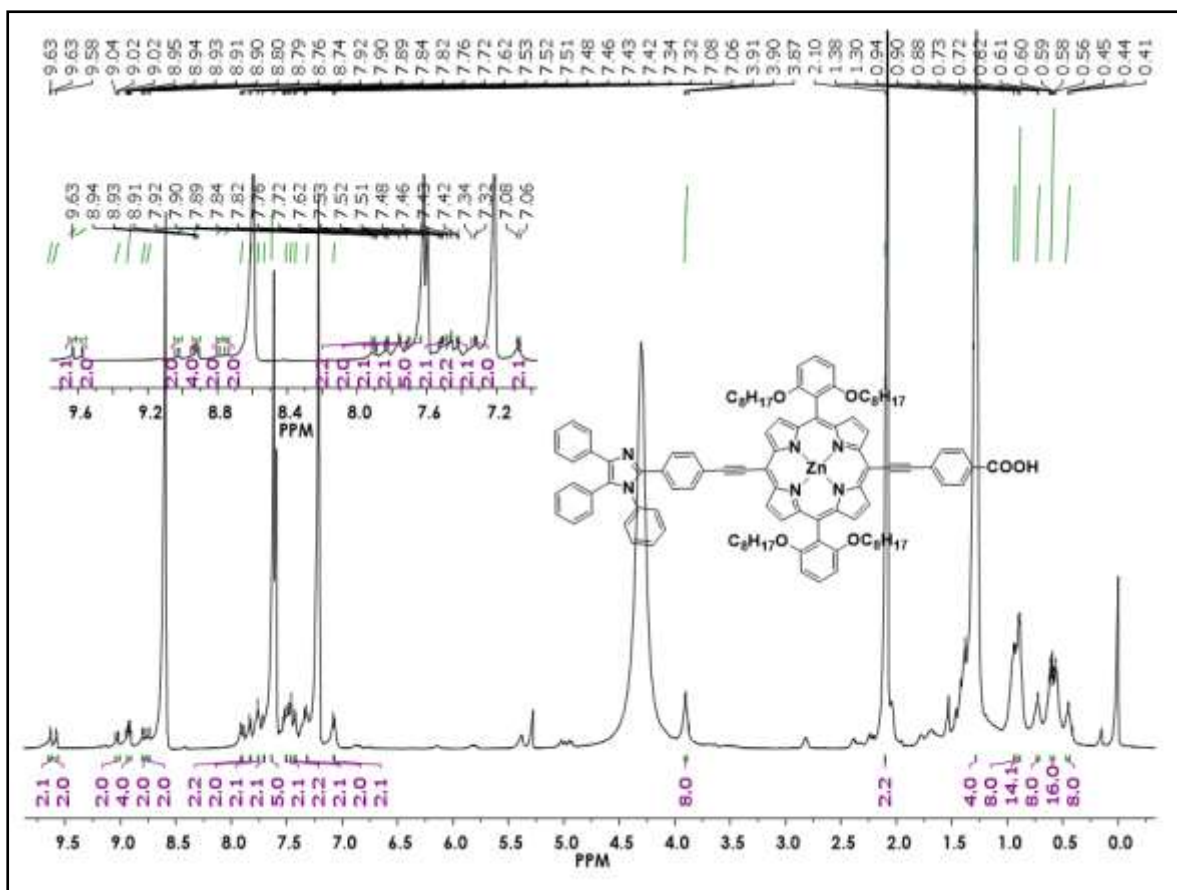


Figure S5. ^1H NMR spectrum (500 MHz, CDCl_3) of LG15.

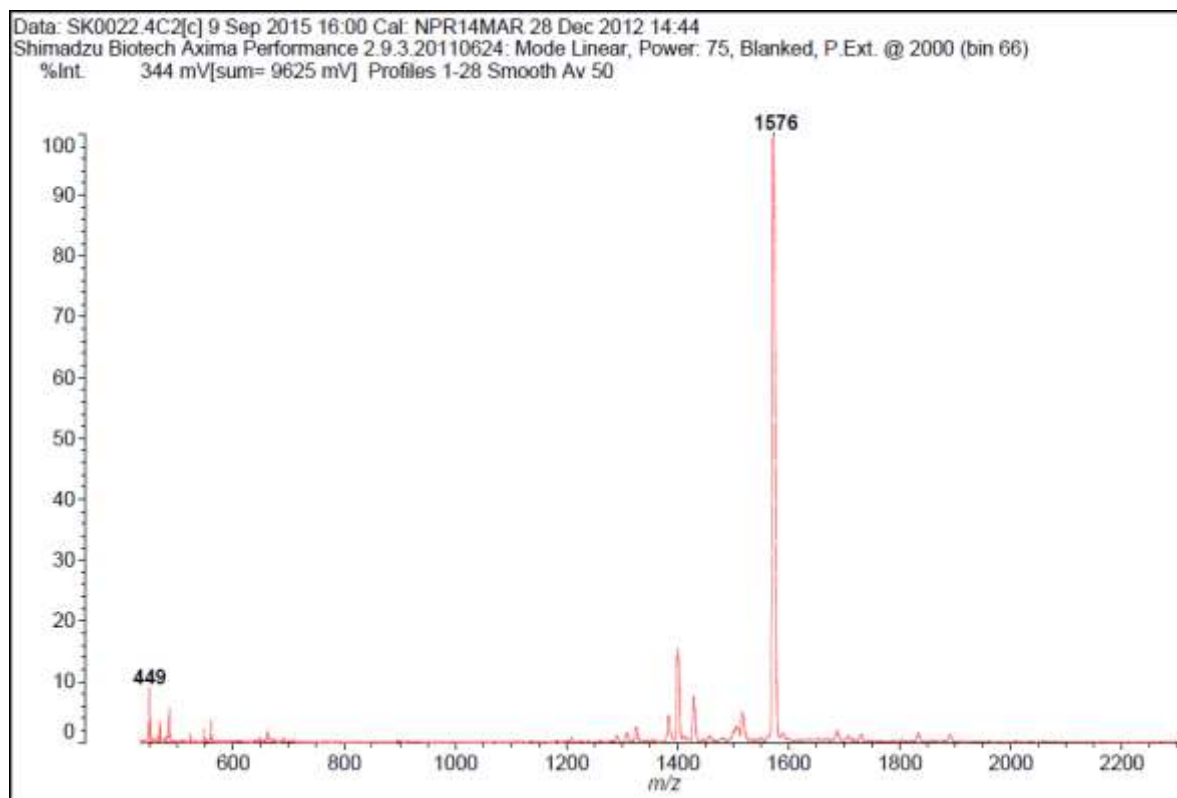
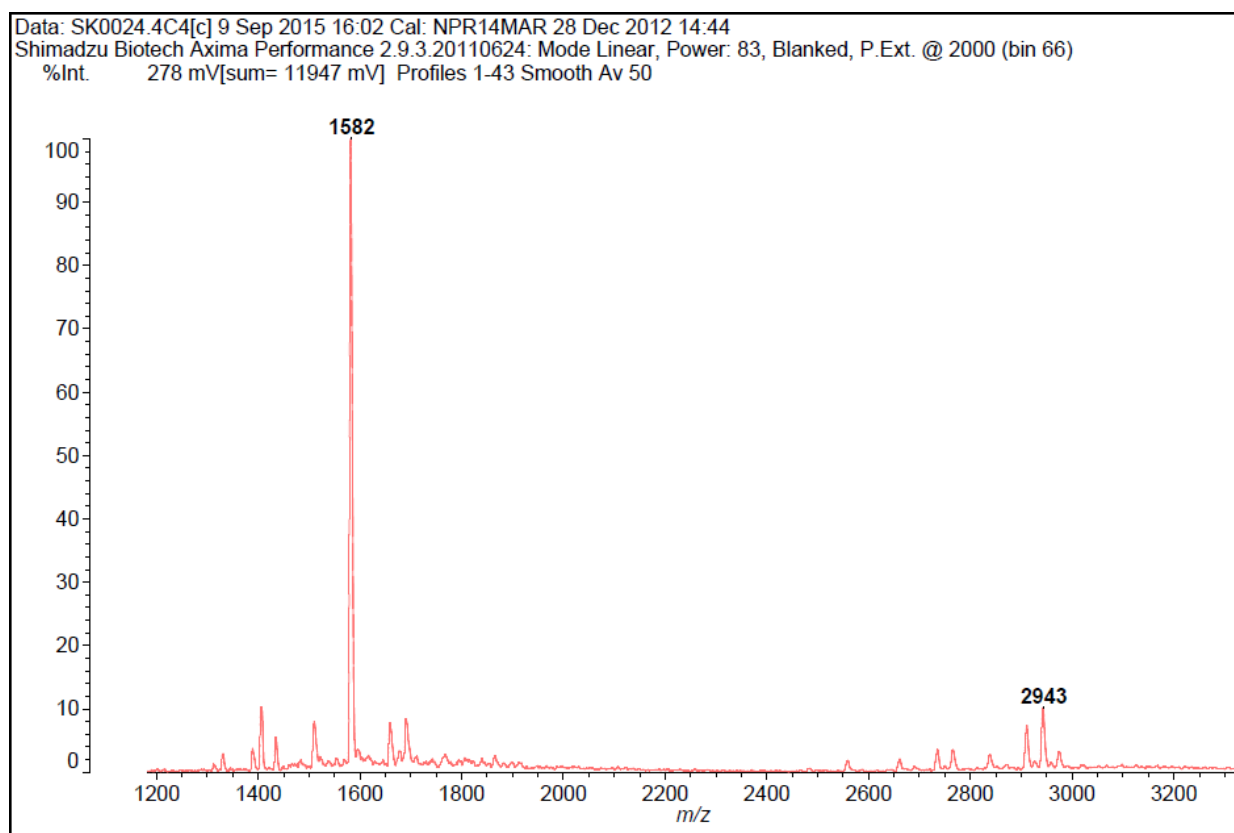
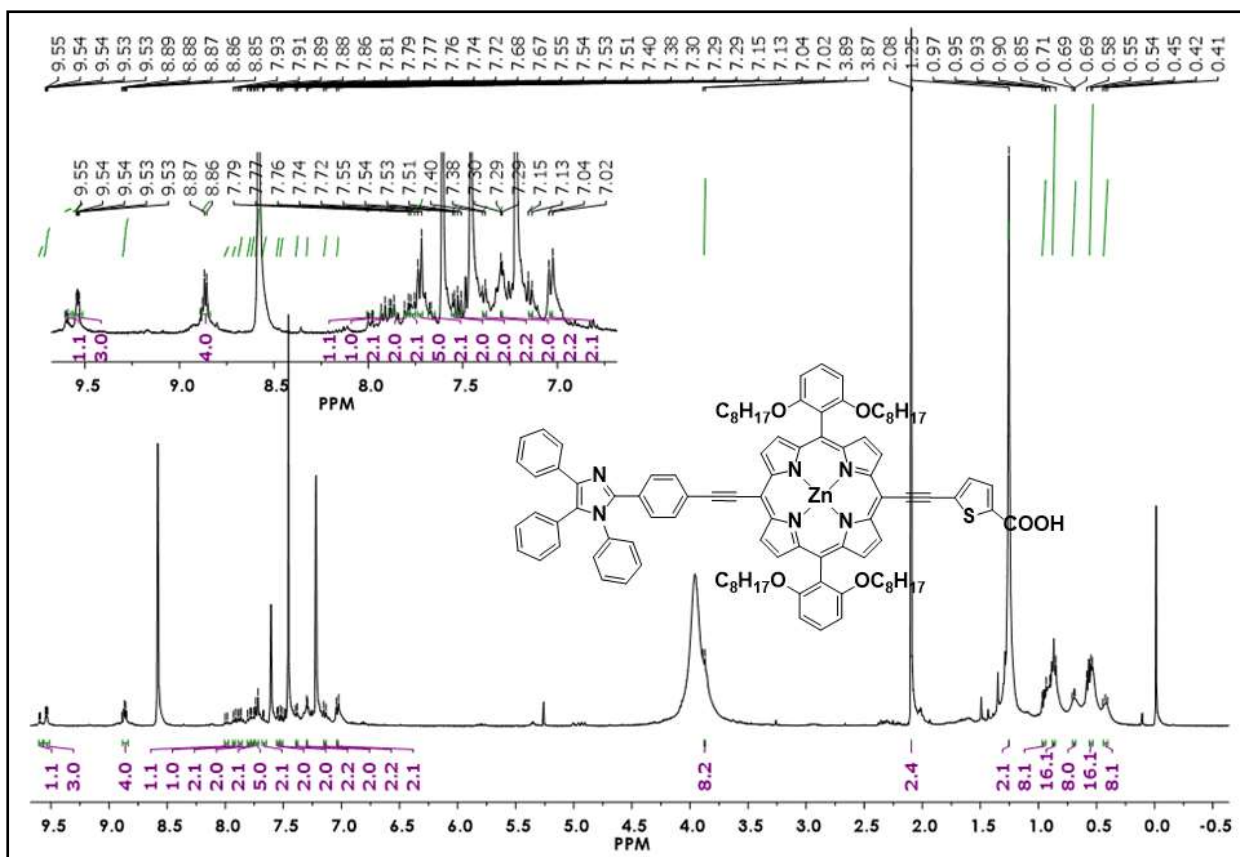


Figure S6. MALDI-TOF of LG15.



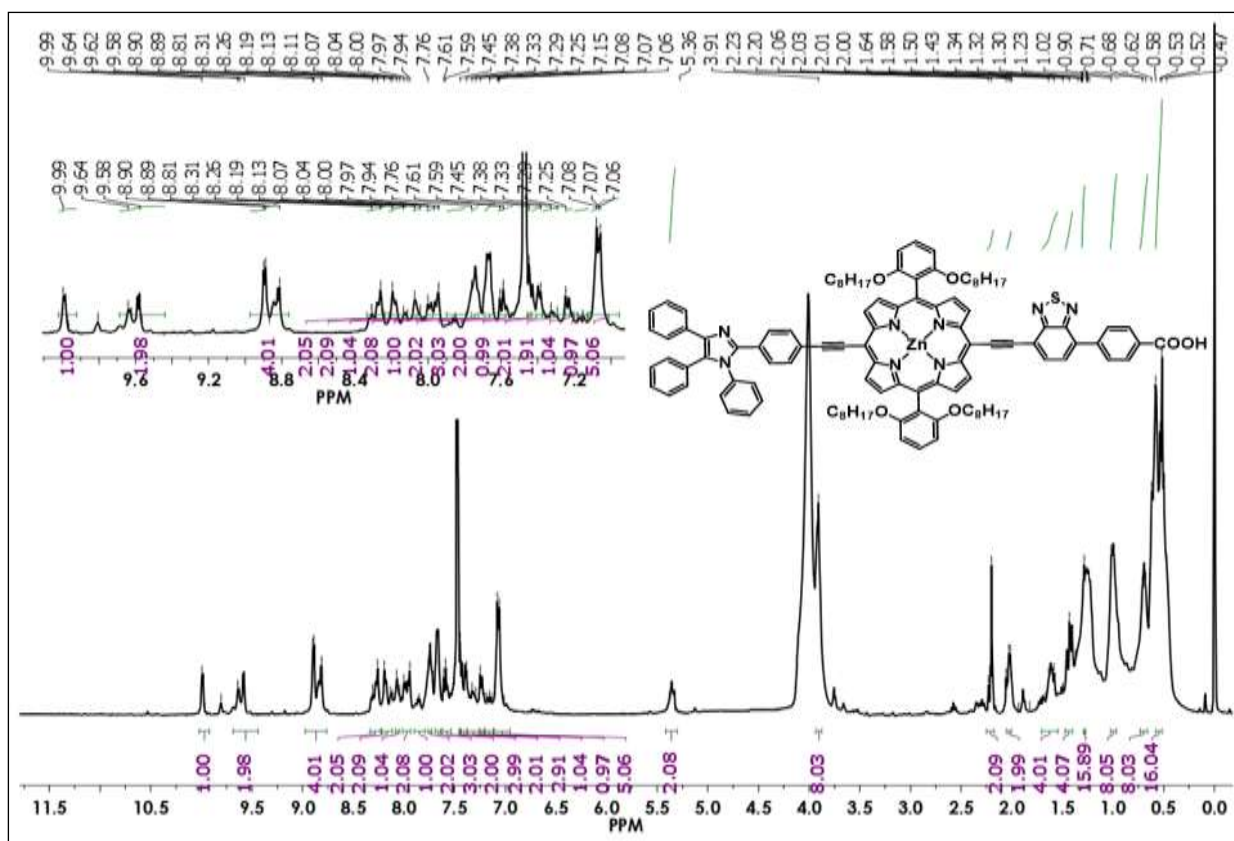


Figure S9. ^1H NMR spectrum (500 MHz, CDCl_3) of LG17.

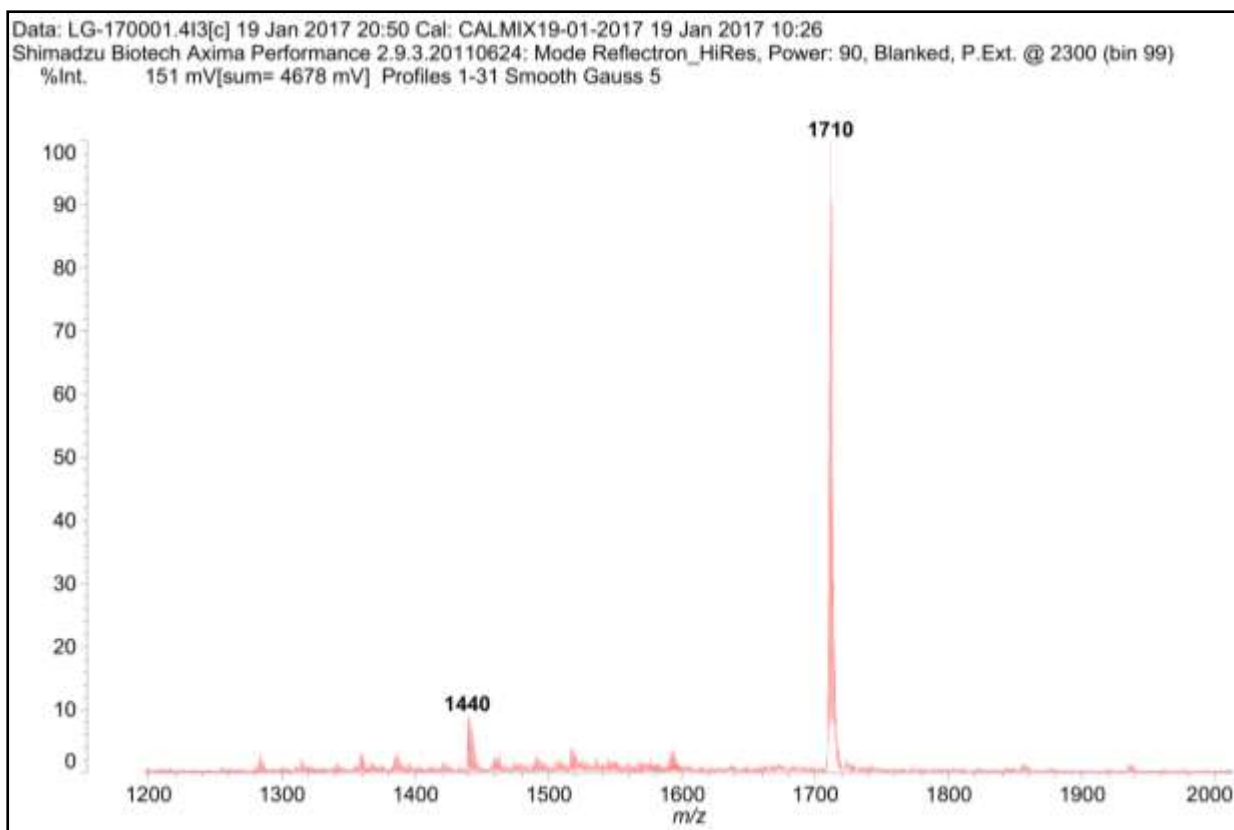


Figure S10. MALDI-TOF of LG17.

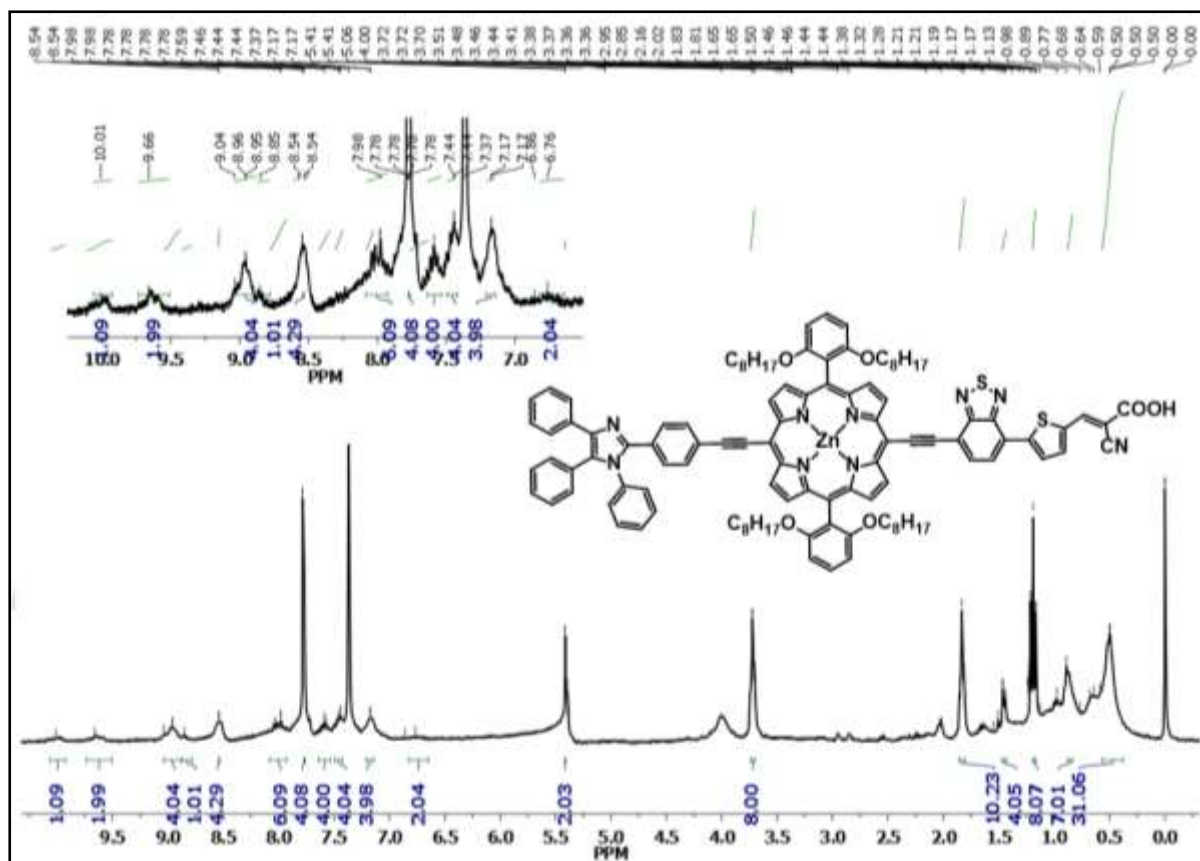


Figure S11. ^1H NMR spectrum (500 MHz, CDCl_3) of LG18.

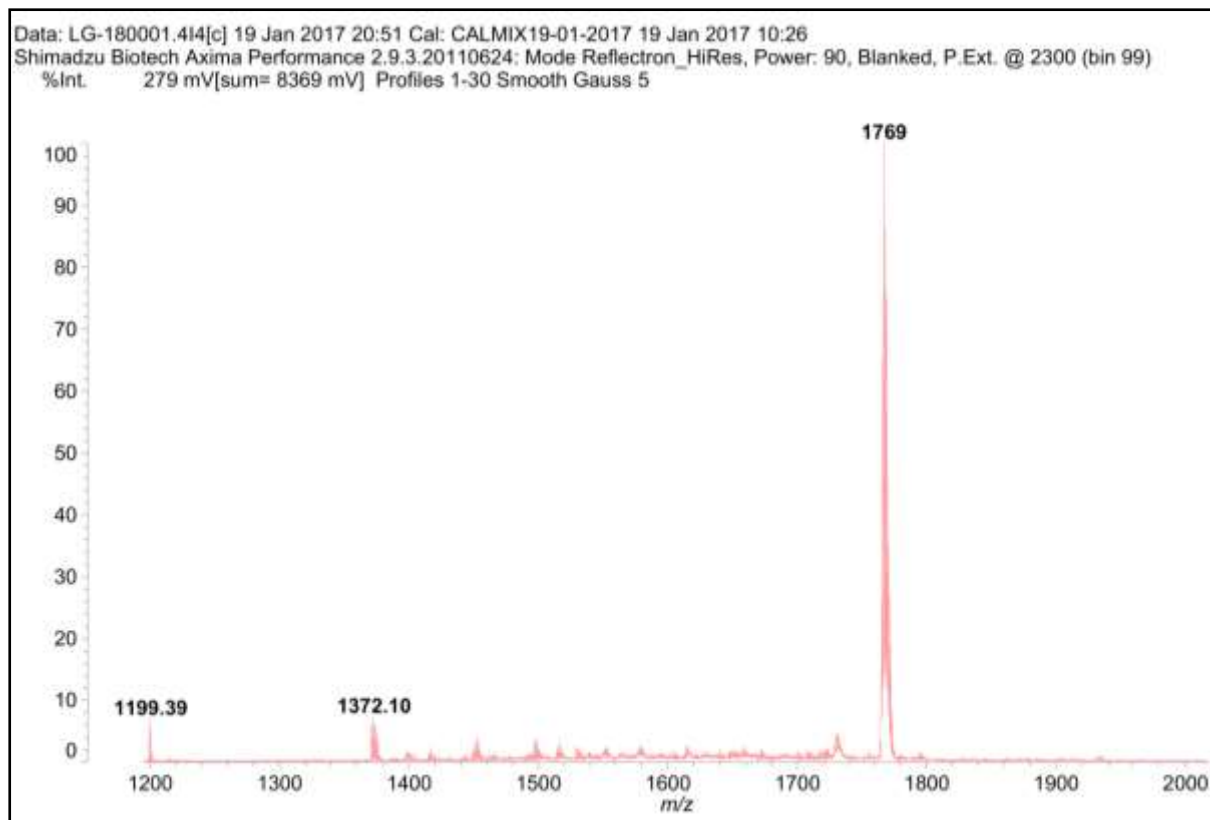


Figure S12. MALDI-TOF of LG18.

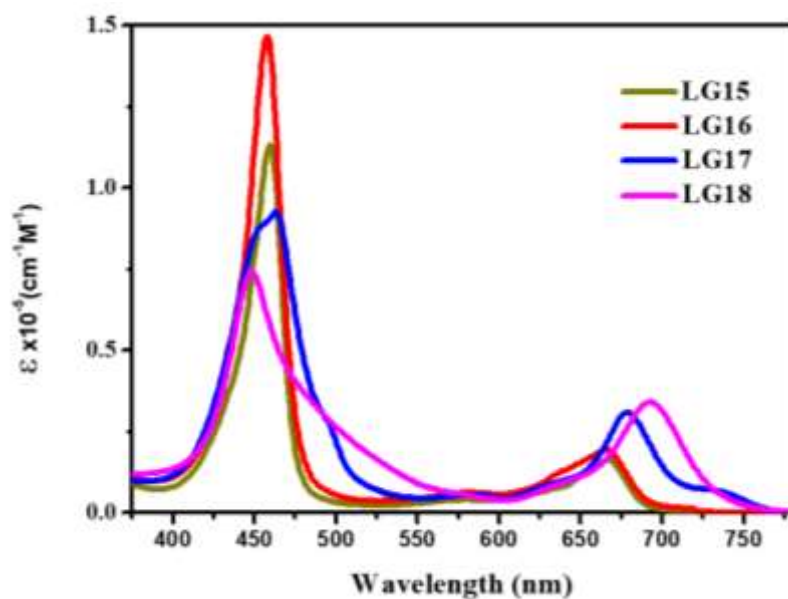


Figure S13. UV-visible absorption spectra of LG15-LG18 in THF solvent.

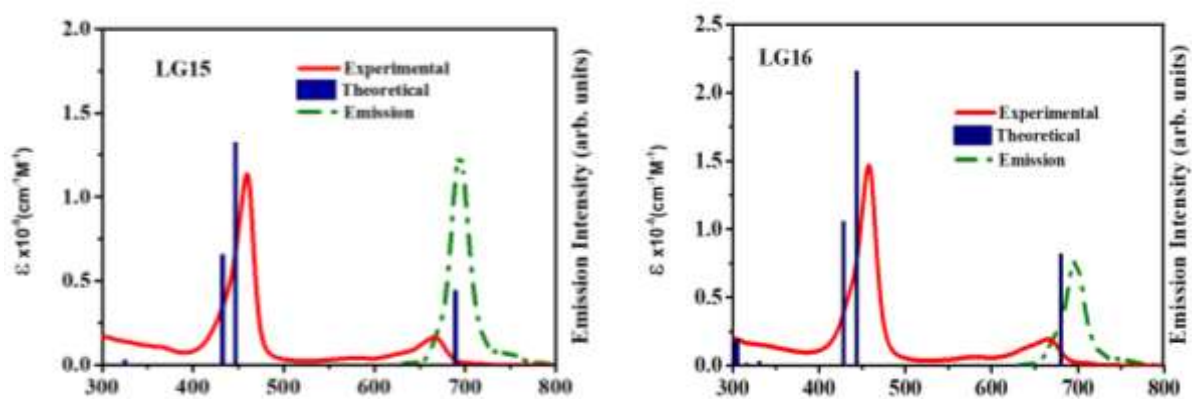


Figure S14. Theoretical absorption spectra of LG15, LG16 Dyes by using B3LYP method PCM model in tetrahydrofuran solvent with M06-2X function.

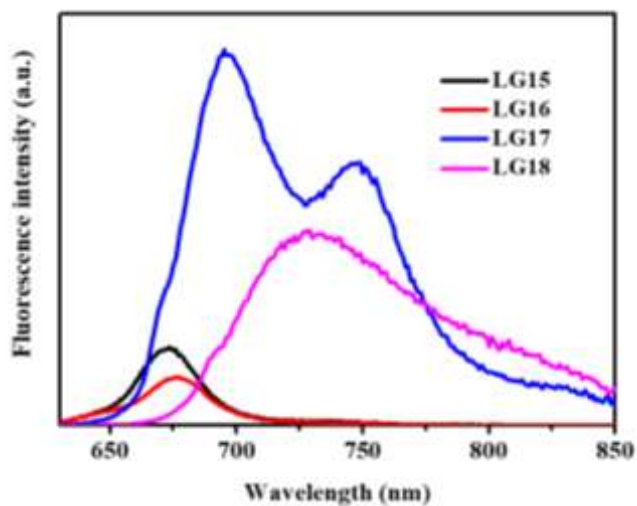


Figure S15. Fluorescence spectra of LG15, LG16, LG17 and LG18 in THF solution.

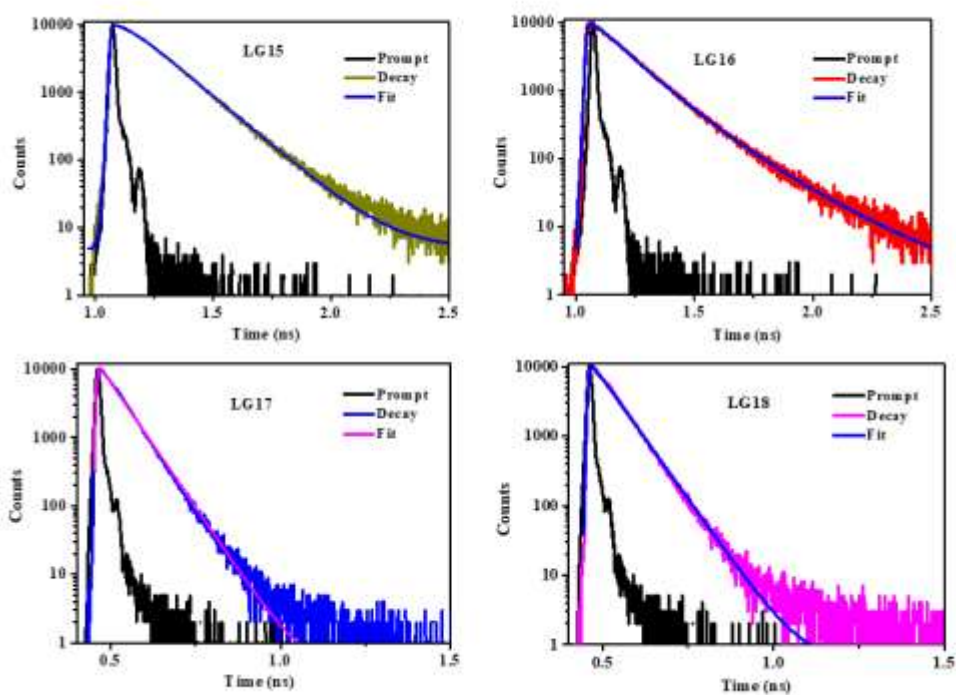


Figure S16. Singlet excited-state lifetimes of *LG15*, *LG16*, *LG17* and *LG18* in THF solution.

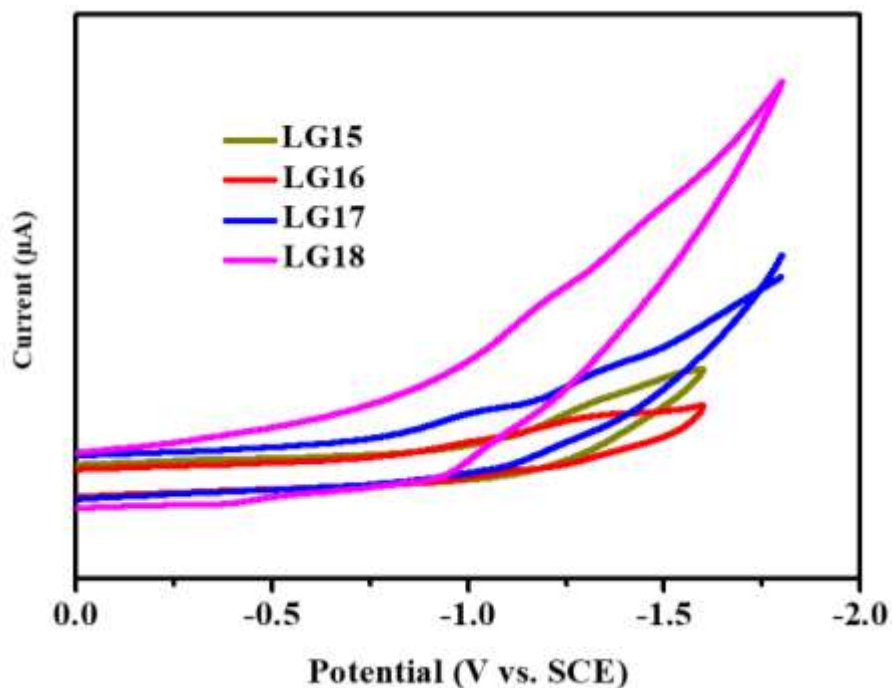


Figure S17. Cyclic Voltammogram of *LG15*, *LG16*, *LG17* and *LG18* in THF.

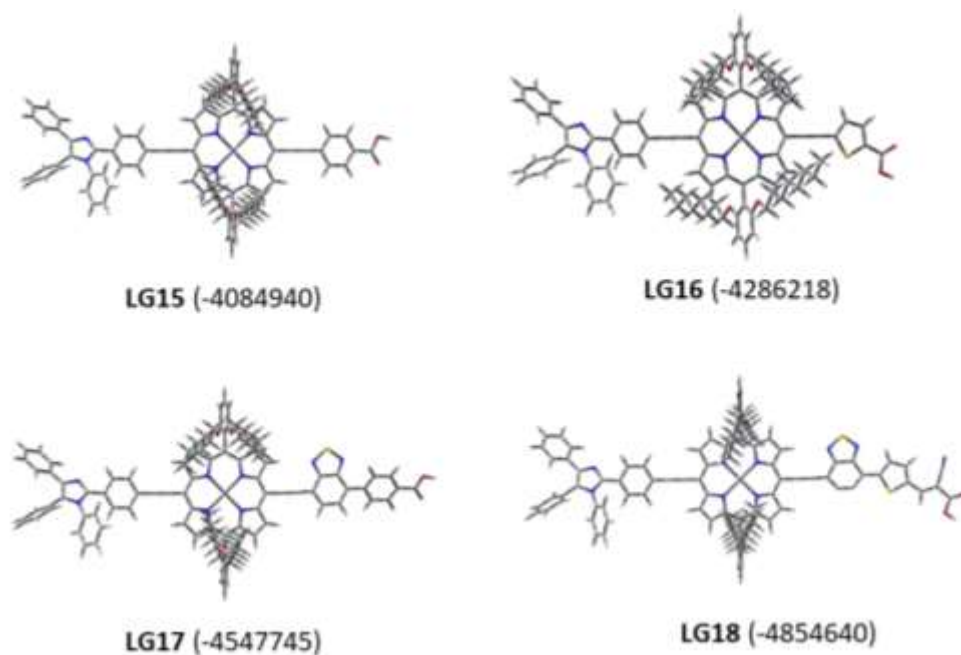


Figure S18. Optimized structure of *LG15*, *LG16*, *LG17* and *LG18* dyes.

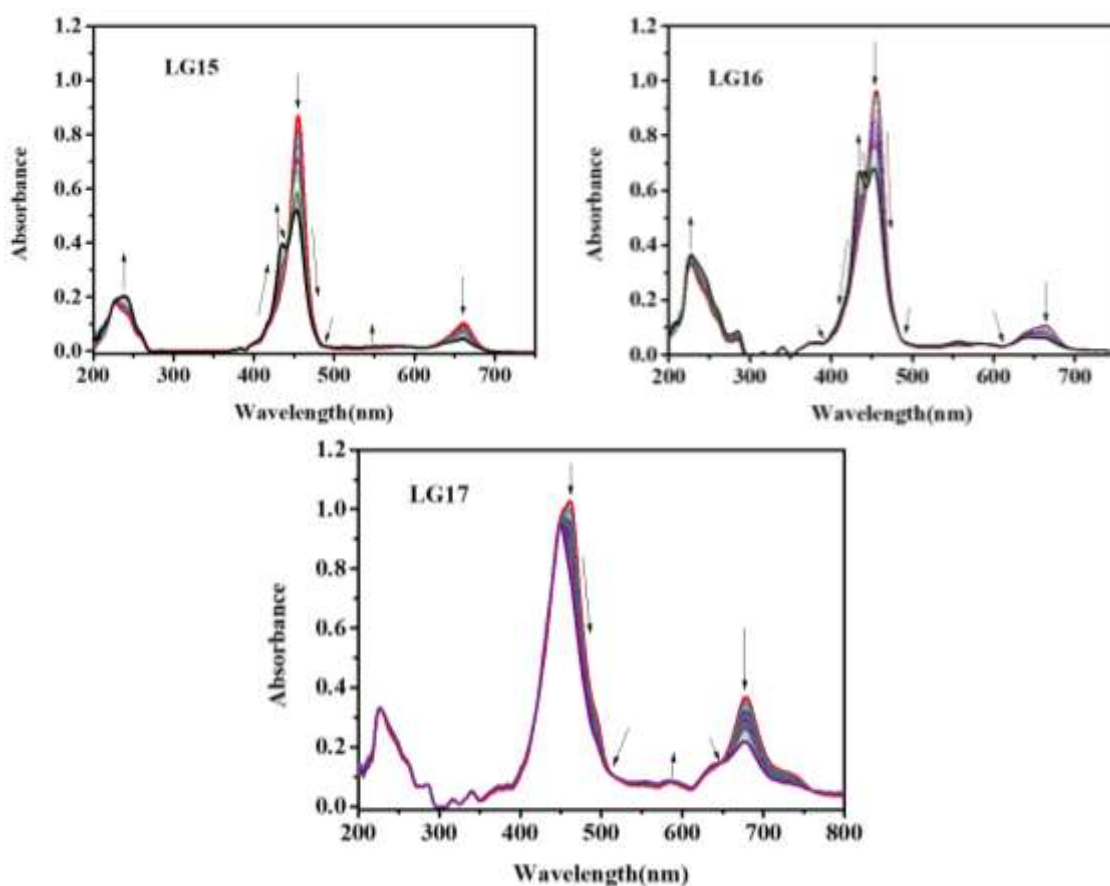


Figure S19. Oxidative OTTLE studies of *LG15*, *LG16*, *LG17* series sensitizers in 0.3M TBAP/THF with an applied potential of +0.90V (vs. SCE/KCl).

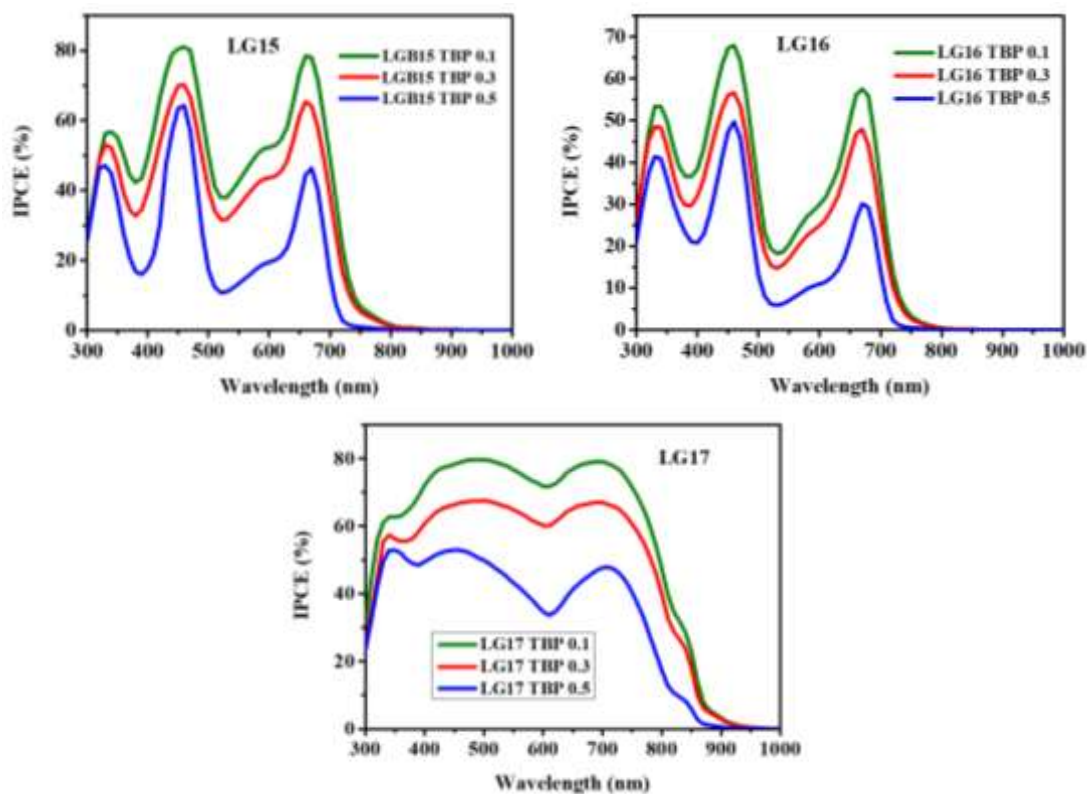


Figure S20. Photocurrent action spectra porphyrin sensitizers using different concentrations of 4-tert butylpyridine.

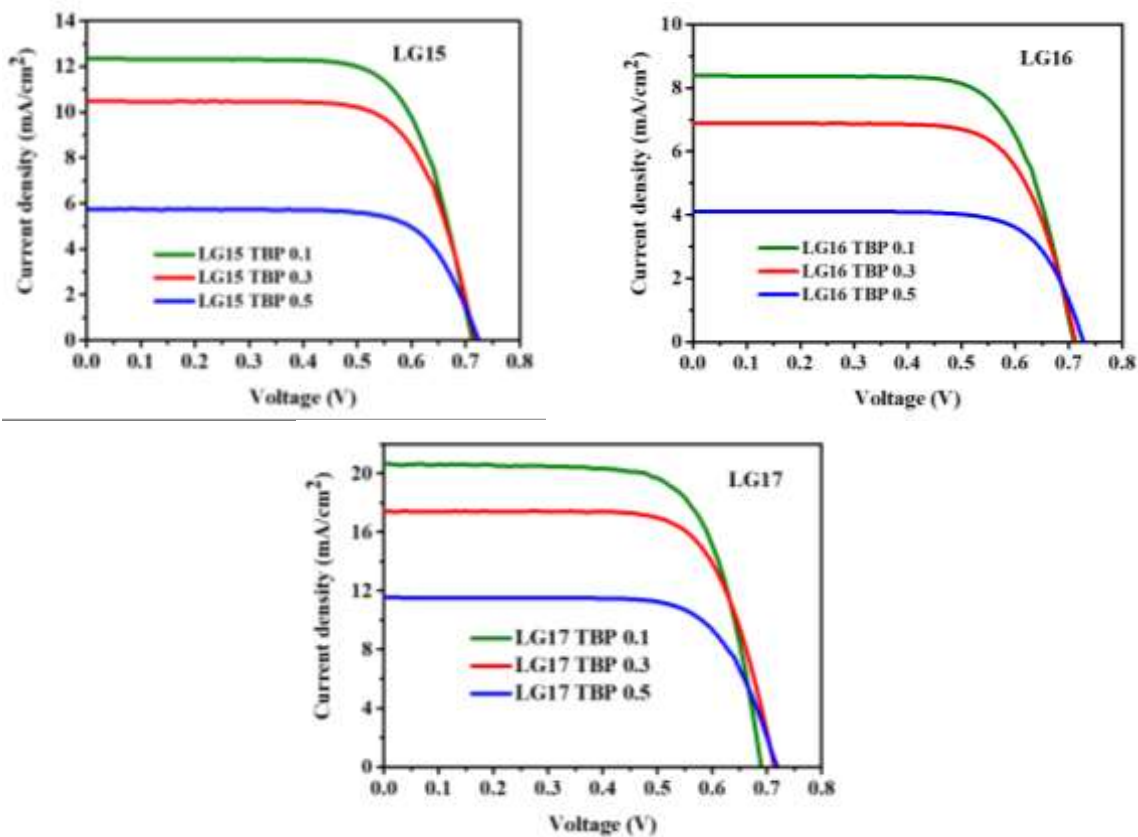


Figure S21. Current-voltage characteristics of porphyrin sensitizers using different concentrations of 4-tert butylpyridine.

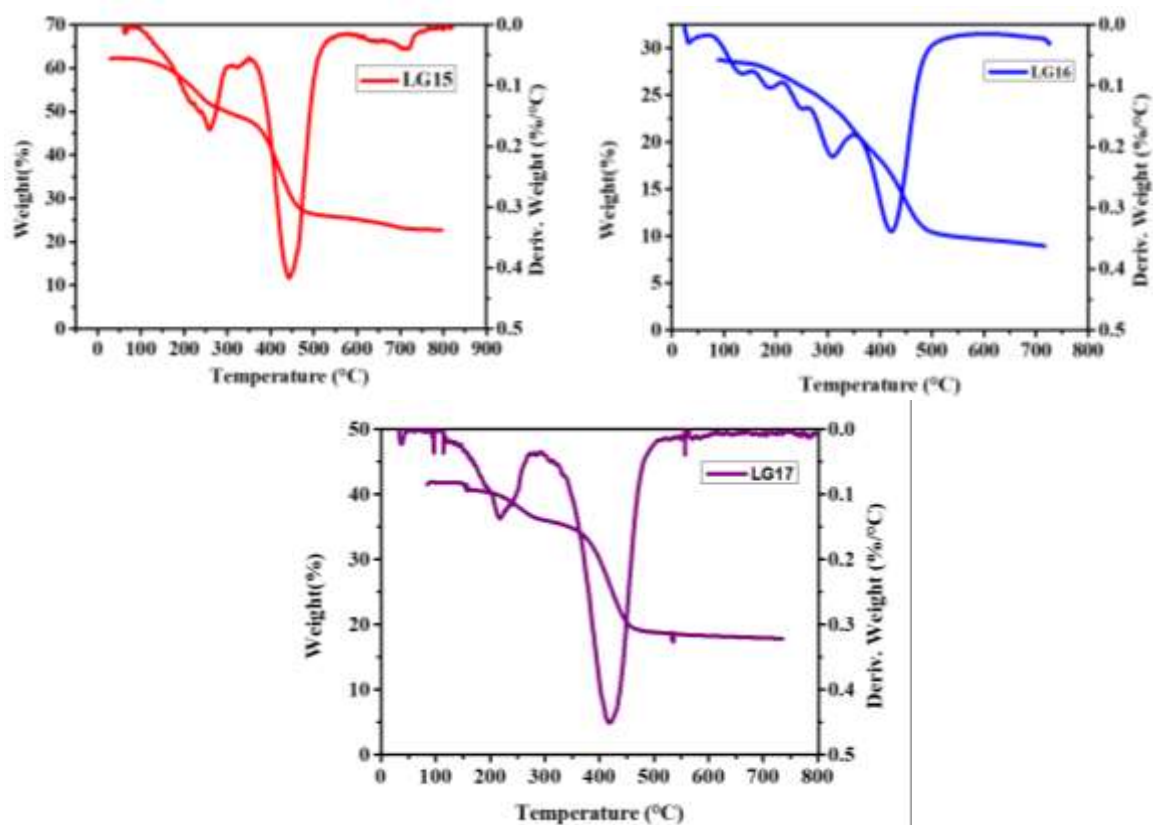


Figure S22. TG/DTG curves of **LG15**, **LG16** and **LG17** porphyrins with heating rate 10 °C.min⁻¹ under nitrogen.

Table 1: Optimized energies, HOMO-LUMO energies and ground state dipole moment by DFT studies by using B3LYP/6-31G (d,p) in vacuum.

Dye	^a E, K.cal./mol	^b HOMO (H),	^b LUMO (L)	^b H-Lgap	^c μ
LG15	-4084940)	-4.689	-2.431	2.25	4.785
LG16	-4286218	-4.698	-2.498	2.20	6.193
LG17	-4547745	-4.678	-2.701	1.97	6.284
LG18	-4854640	-4.849	-3.222	1.62	15.281

^aTotal minimum energy of **LG15-LG18**, ^bvalues in eV, ^cvalues in debye units.

Table 2: Singlet excited state properties of dyes by B3LYP method and M06-2X function in tetrahydrofuran solvent in PCM model.

Dye	^a λ_{max}	^b f	^c E (eV)	% of Molecular Orbital Contribution
LG 15	441	2.652	3.014	H-1→L+1 (74%), HOMO→LUMO (17%) H-2→LUMO (3%), HOMO→L+3 (2%)
	555	0.014	2.231	H-1→LUMO (59%), HOMO→L+1 (39%)
	595	0.886	2.081	H-1→L+1 (16%), HOMO→LUMO (81%)
LG 16	440	2.592	2.991	H-1→L+1 (75%), HOMO→LUMO (15%) H-2→LUMO (2%)
	557	0.016	2.223	H-1→LUMO (59%), HOMO→L+1 (38%)
	603	0.987	2.052	H-1→L+1 (15%), HOMO→LUMO (82%)
LG 17	450	1.049	2.754	H-1→L+2 (30%), HOMO→L+1 (44%) H-3→LUMO (7%), H- 3→L+1 (3%), H-2→L+1 (2%), HOMO→LUMO (9%)
	559	0.017	2.215	H-1→LUMO (54%), HOMO→L+2 (37%) H-1→L+1 (7%)
	616	1.273	2.011	H-1→L+2 (12%), HOMO→LUMO (81%), HOMO→L+1 (3%)
LG 18	441	0.226	2.807	H-1→LUMO (44%), H-1→L+1 (14%), HOMO→L+2 (37%), H-1→L+4 (3%)
	578	0.041	2.143	H-1→LUMO (54%), H-1→L+1 (19%), HOMO→L+2 (25%)
	692	2.384	1.789	HOMO→LUMO (83%) H-3→L+1 (2%), H-2→LUMO (3%), H-1→L+2 (4%), HOMO→L+1 (6%)

^aTheoretical absorbance in nm, ^bOscillator strength, and ^cExcited state energy in eV.