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Supporting information

3Dgraphene/hydroxypropyl-β-cyclodextrin nanocomposite as electrochemical chiral sensor for the recognition of tryptophan enantiomers

Wenting Liang^a*, Yanqin Rong^a, Lifang Fan^a, Wenjuan Dong^a, Qingchen Dong^c, Cheng Yang^b*, Zhihui Zhong^b, Chuan Dong^a*, Shaomin Shuang^a, Wai-Yeung Wong^d*

^a Institute of Environmental Sciences, Department of Chemistry, Shanxi University, Taiyuan 030006, China. E-mail: liangwt@sxu.edu.cn; dc@sxu.edu.cn.

^b State Key Laboratory of Biotherapy, West China Medical School, College of Chemistry, Sichuan University, 29 Wangjiang Road, Chengdu 610064, China. E-mail: yangchengyc@scu.edu.cn

^c Ministry of Education Key Laboratory of Interface Science and Engineering in Advanced Materials, Research Center of Advanced Materials Science and Technology, Taiyuan University of Technology, Taiyuan, 030024, China.

^d Department of Applied Biology and Chemical Technology, The Hong Kong Polytechnic University, Hung Hom, Hong Kong, China. E-mail: wai-yeung.wong@polyu.edu.hk



Fig. S1. SEM image of 3D-G



Fig. S2. Nyquist plots of bare GCE, 3D-Gand 3D-G/HP- β -CD modified GCE in 0.1 M KCl containing 10 mM $[Fe(CN)_6]^{3-/4-}(pH 7.0) \text{ at } 25 \text{ °C}.$



Fig. S3.Cyclic voltammograms of bare GCE, 3D-G and 3D-G/HP- β -CD modified GCE in 0.1 M KCl containing 10 mM [Fe(CN)₆]^{3-/4-} (PBS of pH 7.0) at 25 °C.



Fig. S4. TGA of 3D-G (a) and 3D-G/HP- β -CD containing different amounts of HP- β -CD (b-d).



Fig. S5. The HP- β -CD amounts of 3D-G/HP- β -CD (32.1%, 46.7% and 77.5%) influence on the enantiorecognition efficiency ($I_{\rm L}/I_{\rm D}$) of 3D-G/HP- β -CD/GCE toward Trp isomers. Errors bars represent the standard deviation for three independent measurements.



Fig. S6. Influence of pH on the enantiorecognition efficiency (I_L/I_D) of 3D-G/HP- β -CD/GCE toward Trp isomers at 25 °C. Errors bars represent the standard deviation for three independent measurements.



Fig. S7. Influence of the accumulation time on the enantiorecognition efficiency (I_L/I_D) of 3D-G/HP- β -CD/GCE toward Trp isomers at 25 °C. Errors bars represent the standard deviation for three independent measurements.



Fig. S8. Influence of the 3D-G/HP- β -CD amount on the enantiorecognition efficiency (I_L/I_D) of 3D-G/HP- β -CD/GCE toward Trp isomers at 25 °C. Errors bars represent the standard deviation for three independent measurements.

Modified electrode	Peak current ratio	Linear range	Detection limit	Ref.
	(L-Trp to D-Trp)	(mol L ⁻¹)	(mol L ⁻¹)	
PTCA-CS/GCE	2.60	—	—	40
P-L-Glu/β-CD/GCE	2.30	_	—	9
GQDs/β-CD/GCE	2.10	_	—	39
GQDs-CS/GCE	2.06	_	—	10
Pd-Cu@Cu ₂ O/N-RGO/GCE	—	$1.0 imes 10^{-8} - 4.0 imes 10^{-5}$	1.9×10 ⁻⁹ (L/D)	41
β-CD-MNPs/GCE	—	$8.0 imes 10^{-7} - 3.0 imes 10^{-4}$	$5.0 \times 10^{-7} (L/D)$	42
ds-DNA/Thi-GR/GCE	—	$5.0 \times 10^{7} - 2.5 \times 10^{3}$	$1.7 \times 10^{-7} (L/D)$	43
NH ₂ -GQDs/β-CD/GCE	—	$1.0 \times 10^{\text{-6}} - 3.0 \times 10^{\text{-5}}$	6.5 × 10 ⁻⁷ (L)	44
			1.2×10^{-7} (D)	
NH2-B-CD/Au@Pt/PEI/	—	$1.0 \times 10^{-5} - 5.0 \times 10^{-3}$	4.3 × 10 ⁻⁶ (L)	45
MWCNTs/GCE			$5.6 \times 10^{-6} (D)$	
PLC/MWCNTs/GCE	—	$1.0 imes 10^{-4} - 1.0 imes 10^{-3}$	$3.3 \times 10^{-5} (L/D)$	46
mNBFs/mNC-GPEs	1.40	$1.0 imes 10^{-7} - 1.0 imes 10^{-5}$	—	47
β-CD-PtNPs/GNs/GCE	1.30	$5.0 \times 10^{\text{-5}} - 5.0 \times 10^{\text{-3}}$	1.7×10^{-5} (L)	36
			2.1 × 10 ⁻⁵ (D)	
3D-G/HP-β-CD/GCE	2.13	$5.0 imes 10^{-7} - 1.75 imes 10^{-4}$	9.6 × 10 ⁻⁹ (L)	This
			3.8 × 10 ⁻⁸ (D)	work

Table S1. Comparison with other electrochemical methods for the recognition of tryptophan enantiomers

Chiral amino acids	K (M ⁻¹) of L-isomer with HP- β -CD	K (M ⁻¹) of D-isomer with HP- β -CD
Phenylalanine (Phe)	23.75	19.73
Tyrosine (Tyr)	32.02	25.31
Tryptophan (Trp)	235.3	176.2

Table S2. Inclusion constant (*K*) for the 1:1 inclusion complexation of L-, D-phenylalanine, L, D-tryptophan and L-, D-tyrosine with HP- β -CD.