Thick Junction Photodiodes Based on Crushed Perovskite Crystal/Polymer Composite Films

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- 1. Experimental details
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1. Experimental details

Materials: Lead iodide (PbI₂, 99% trace metals basis) was purchased from Sigma-Aldrich. Methylammonium iodide (CH₃NH₃I, 99.99%) was purchased from Greatcell Solar. Nickel acetate (C₄H₁₄NiO₈, AR, 99%) was purchased from Heowns. Polymethylmethacrylate (PMMA) were purchased from Sinopharm Chemical Reagent Co., Ltd. Fullerene-C70 (C70, >99%) and bathocuproine (BCP, >99%) were purchased from Xi'an Polymer Light Technology Corp. Gamma-Butyrolactone (GBL, >99.0%) was purchased from Tokyo Chemical Industry (TCI). Ethylene glycol monomethyl ether (C₃H₈O₂, AR, \geq 99.0%) was purchased from Greagent. Chlorobenzene (C₆H₅Cl, AR, 99%) was purchased from Aladdin. Acetone (C₃H₆O, AR) and ethanol (C₂H₆O, AR, anhydrous, \geq 99.7%) were purchased from Sinopharm Chemical Reagent Co., Ltd. 2-aminoethanol (C₂H₇NO, AR, 99%) was purchased from Aladdin. All commercial products were used as received.

Preparation of perovskite single crystal powders: The CH₃NH₃PbI₃ perovskite single crystals were prepared via inverse temperature crystallization^[S1]. Typically, 461 mg PbI₂ and 159 mg CH₃NH₃I were added into 0.8 ml GBL solvent with a molar ratio of 1:1 to form 1.25 M perovskite precursor solution. The mixed solution was stirred and heated on a hot plate at 60°C for 2 h, and filtered (0.22 µm pore size) after all solids were dissolved. The obtained solution was transferred to clean containers, which were kept on a stable hot-plate and gradually heated to 120°C and kept for another 6 h. Crystals were formed in the bottom of the containers. Finally, the crystals were collected and dried at 60°C in vacuum oven for 12 h. Then, the single crystals were ground and sieved carefully in glovebox (O₂ < 1 ppm, H₂O < 1 ppm) to prevent materials degradation during the process, as the perovksite are sensitive to moisture. CH₃NH₃PbI₃ single crystal powders were collected with a vial and stored in a glovebox.

Device fabrication: All the photodetector devices were fabricated based on commercial patterned Indium Tin Oxide (ITO) substrates (Meijingyuan Glass). These electrodes were cleaned with detergent solution at 80°C, followed with sonication in sequence in de-ionized water, acetone and ethanol for 10 min each. These substrates

were dried on a hot plate at 120°C for 10 min before depositing thin films according to the device structure. The cleaned substrates were dried with nitrogen before being sustained ozone treatment for 15 min. The NiO_x precursor solution was prepared by adding 35.36 mg nickel acetate and 12.22 µL 2-aminoethanol in 1 mL ethylene glycol monomethyl ether and stirred for 1h under 70°C, according to the method reported by W. Chen et al.^[S2] The obtained NiO_x precursor was spin-coated on the substrates at 3000 rpm for 30 s in air to form hole transport layer (HTL). The NiO_x-coated substrates were pre-annealed at 150°C for 5 min on a hot plate, followed with another annealing treatment in a muffle furnace at 300°C for 60 min. After cooling, the substrates were transferred to a nitrogen-filled glovebox for device fabrication ($O_2 < 1$ ppm, H₂O < 1 ppm). 25 mg PMMA were dissolved in 0.5 mL chlorobenzene to form a viscous solution (50 mg/ml). Then, another 500 mg CH₃NH₃PbI₃ single crystal powders were added in the prepared 0.5 mL PMMA solution, followed with ultrasonication and stirring until powders were uniformly dispersed. Then, the CH₃NH₃PbI₃/PMMA suspension was spin-coated on the NiO_x layer at 500 rpm for 100 s to form thick perovskite/polymer composite films, which were annealed at 100°C for 30 min in glovebox afterwards. In order to deposit electron transfer layer, 60 nm C70 were evaporated under vacuum (~ 4 $\times 10^{-4}$ Pa, at a rate of ~0.1 Å/ s). Another 3 nm BCP were evaporated at the same conditions. Finally, 60 nm Cu electrodes were evaporated under vacuum (~ 4×10^{-4} Pa, at a rate between 0.1 and 0.5 Å/ s).

Characterization: Scanning electron microscopy (SEM) images of the films were obtained using a SIGMA 500 field-emission scanning electron microscope operated at 5 keV. The surface morphology of the perovskite films were imaged using a micro microscope. The crystal structure was characterized with X-ray diffraction (XRD) using an D8 Advance X-ray diffractometer (CuKa=1.5418 Å) with a scanning range from 10° to 80°. Film thicknesses were determined using a film thickness gauge (ThetaMetrisis, FR-Basic-UV/NIR-HR). The roughness and thickness of the crushed CH₃NH₃PbI₃ crystal/PMMA composite films were recorded with an Alpha-step D-600 profile meter (KLA-Tencor).

Device performance measurements: J-V characteristics (both in dark and under illumination) of the devices were tested at room temperature by using a Keithley 2400 source meter. A Xenon lamp (Solar-500) was used as light source to provide various light intensities for measuring light *J-V* curves. Frequency response were obtained using different color LEDs (Thorlabs) modulated with an arbitrary wave function generator (Agilent, 33612A), and the photocurrent responses of these devices were recorded with a digital storage oscilloscope (LeCroy Waverunner 8254). Responsivity was measured with a lock-in amplifier, and the light intensity was calibrated with a standard certified photodiode (Thorlabs).

2. Supporting figures



Fig. S1 SEM images of solution-processed (**a**) CH₃NH₃PbI₃ thin films and (**b**) crushed CH₃NH₃PbI₃ crystal/PMMA composite films.



Fig. S2 Recorded roughness and thickness of the crushed CH₃NH₃PbI₃ crystal/PMMA composite films via profile meter.



Fig. S3 Comparison of the X-ray diffraction patterns of (**a**) CH₃NH₃PbI₃ thin films and (**b**) crushed CH₃NH₃PbI₃ crystal/PMMA composite films.



Fig. S4 Typical optical photos of (**a**) the crushed single crystal/PMMA composite thick films, (**b**) the spin-coated $CH_3NH_3PbI_3$ thin films and (**c**)-(**f**) $CH_3NH_3PbI_3$ /PMMA thin films prepared by directly mixing of PMMA and perovskite precursors with PMMA concentration of (**c**) 0.5 mg/mL, (**d**) 5 mg/mL, (**e**) 25 mg/mL and (**f**) 50 mg/mL.



Fig. S5 Comparison of reflectance and transmittance spectra of (**a**), (**b**) CH₃NH₃PbI₃ thin films and (**c**), (**d**) crushed CH₃NH₃PbI₃ crystal/PMMA composite films.



Fig. S6 Optical photos of (**a**) as-prepared CH₃NH₃PbI₃ thin films, (**b**) CH₃NH₃PbI₃ thin films exposed in air for 120 min, (**c**) as-prepared crushed CH₃NH₃PbI₃ crystal/PMMA composite films and (**d**) crushed CH₃NH₃PbI₃ crystal/PMMA composite films exposed in air for 120 min.



Fig. S7 Schematic illustration of (**a**) the device architecture of thick junction perovskite photodetectors and (**b**) proposed energy level diagram.



Fig. S8 (a) External quantum efficiency and (b) responsivity of the crushed CH₃NH₃PbI₃ crystal/PMMA composite film based TJPPDs.



Fig. S9 Typical RC decay of the optimized thick junction photodiodes based on crushed CH₃NH₃PbI₃ crystal/PMMA composite films, measured with R_{load} =100 Ω .



Fig. S10 Recorded photovoltage decay of the crushed CH₃NH₃PbI₃ crystal/PMMA composite film based TJPPDs.



Fig. S11 Optical photos of (**a**) as-prepared CH₃NH₃PbI₃ thin film based devices, (**b**) CH₃NH₃PbI₃ thin film based devices stored in air after 128 h (20 ℃/35% RH), (**c**) as-prepared crushed CH₃NH₃PbI₃ crystal/PMMA composite film based TJPPDs and (**d**) crushed CH₃NH₃PbI₃ crystal/PMMA composite film based TJPPDs stored in air after 128 h (20 ℃/35% RH).



Fig. S12 Long-term stability of the responsivity and external quantum efficiency of (**a**) CH₃NH₃PbI₃ thin film based photodiodes and (**b**) crushed CH₃NH₃PbI₃ crystal/PMMA composite film based TJPPDs measured without bias.

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 Table S1 Comparison of the key parameters for perovskite-based photodetectors.

References:

- [S1] M. I. Saidaminov, A. L. Abdelhady, B. Murali, E. Alarousu, V. M. Burlakov, W. Peng, I. Dursun, L. Wang, Y. He, G. Maculan, A. Goriely, T. Wu, O. F. Mohammed and O. M. Bakr, *Nat. Comm.*, 2015, 6, 7586.
- [S2] W. Chen, F. Z. Liu, X. Y. Feng, A. B. Djurišić, W. K. Chan and Z. B. He, Adv. Energy Mater., 2017, 7, 1700722.
- [S3] C. Liu, K. Wang, C. Yi, X. Shi, P. Du, A. W. Smith, A. Karim and X. Gong, J. Mater. Chem. C., 2015, 3, 6600-6606.
- [S4] R. Dong, Y. Fang, J. Chae, J. Dai, Z. Xiao, Q. Dong, Y. Yuan, A. Centrone, X.
 C. Zeng and J. Huang, *Adv. Mater.*, 2015, 27, 1912-1918.
- [S5] L. Dou, Y. M. Yang, J. You, Z. Hong, W. H. Chang, G. Li and Y. Yang, Nat. Comm., 2014, 5, 5404.
- [S6] J. Ding, H. Fang, Z. Lian, J. Li, Q. Lv, L. Wang, J. L. Sun and Q. Yan, Cryst. Eng. Comm., 2016, 18, 4405-4411.
- [S7] H. Fang, Q. Li, J. Ding, N. Li, H. Tian, L. Zhang, T. Ren, J. Dai, L. Wang and Q. Yan, J. Mater. Chem. C., 2016, 4, 630-636.