

## A dual-signal output ratiometric electrochemiluminescent sensor for NADH detection

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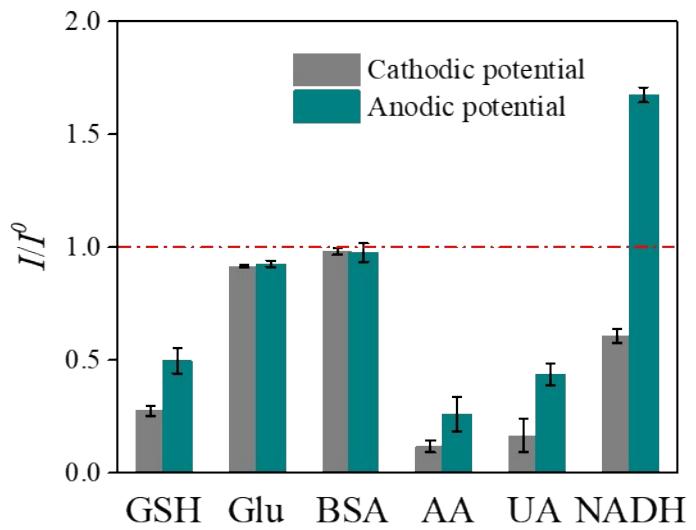
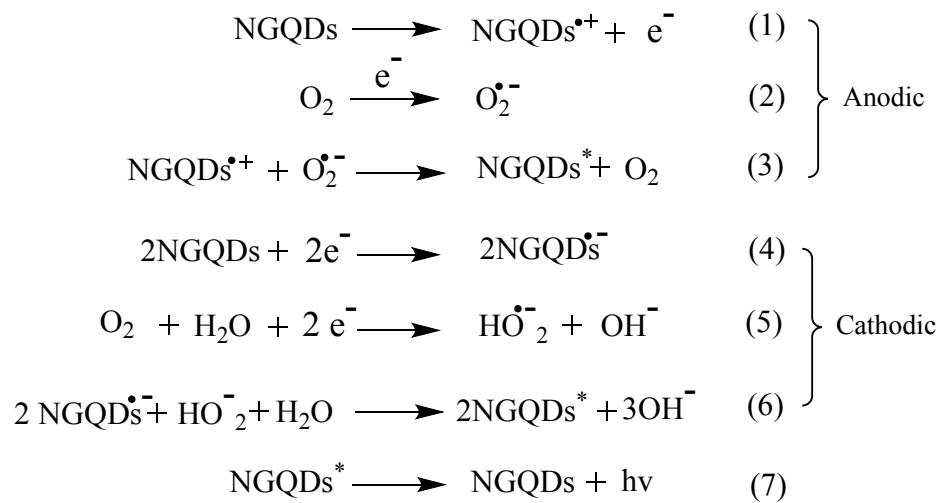


Fig. S1. Ratio values of different potential intensities of the dual-potential ECL system toward 100  $\mu\text{M}$  NADH against 200  $\mu\text{M}$  GSH, Glu, AA, uric acid (UA) and 10 mg  $\text{mL}^{-1}$  BSA interfering species (note:  $I$  is the anodic or cathodic ECL peak intensity, the  $I^0$  is the initial anodic and cathodic ECL intensity of the dual-potential ECL system).



Scheme S1. The entire dual-potential ECL mechanism equations of NGQDs.

**Table S1.** Comparison of various methods and its composite used towards the detection of NADH.

method	Material or electrode	Linear range	Limit of detection	Reference
Colorimetric	MnO <sub>2</sub> nanosheets	5.0–100 μM	-	[1]
Electrochemical	PTH/SPCE	5.0–100 μM	3.0 μM	[2]
Electrochemical	MWCNT/AuNP/PNR	12.4–150 μM	3.72 μM	[3]
Electrochemical	PAA–MWNT/GCE	4.0–400 μM	1.0 μM	[4]
Electrochemical	PEDOT/GCE	5.0–45 μM	3.8 μM	[5]
Electrochemical	GC/MWCNT/Flu/GCE	15–84 μM	5.0 μM	[6]
Electrochemical	Poly Xa/FAD/MWCNT/GCE	5.0–170 μM	1.0 μM	[7]
Electrochemical	PGE/SWCNT/PCV/GCE	1.3–280 μM	1.3 μM	[8]
Electrochemical	AuNPs/PB/EGP	0.5–1000 μM	0.21 μM	[9]
Ratiometric ECL	NGQDs	10–400 μM	2.5 μM	This work

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