

## Electronic Supplementary Information

### Facile Synthesis of Metal-Organic Framework-Derived SiW<sub>12</sub>@Co<sub>3</sub>O<sub>4</sub> and its Peroxidase-Like Activity in Colorimetric Assay

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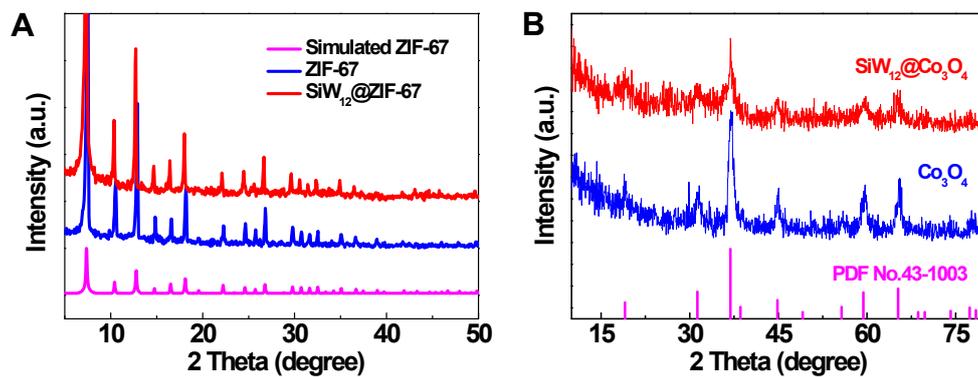
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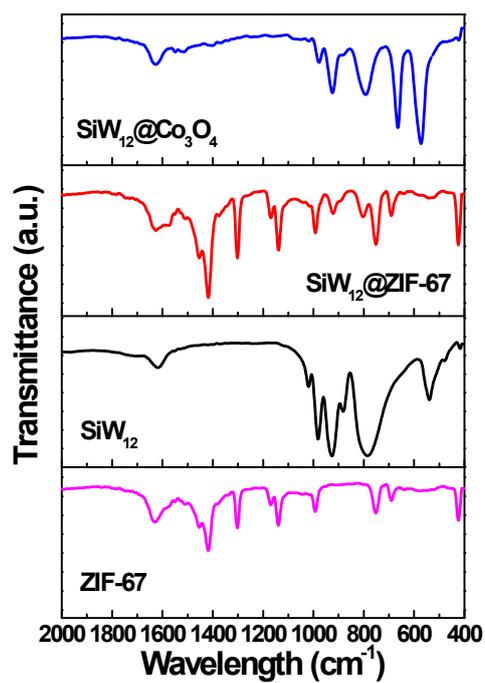
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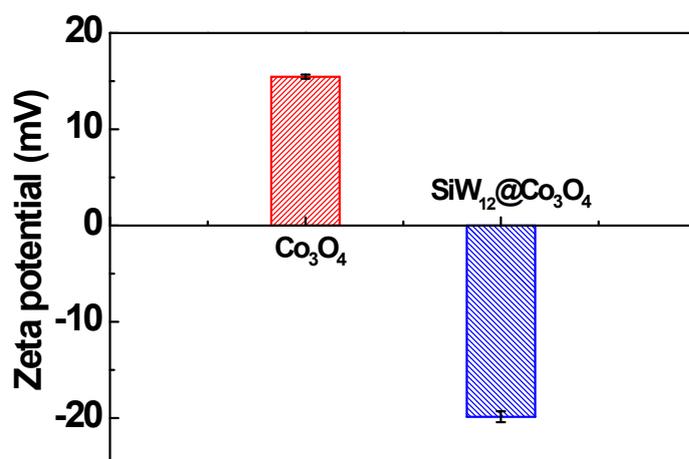
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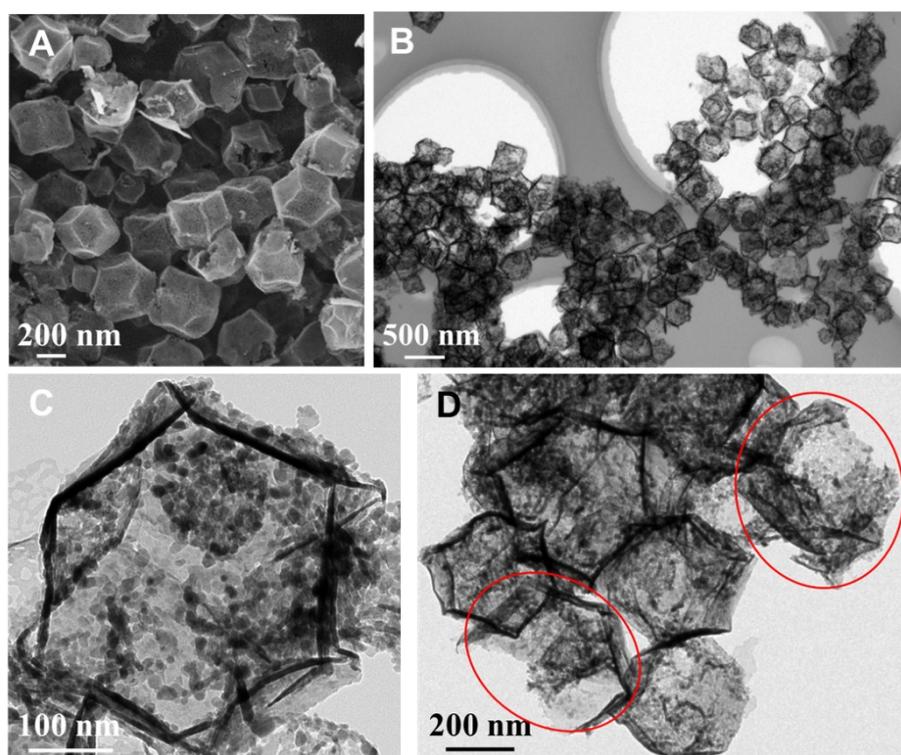
**Fig. S1** XRD curves of (A) ZIF-67 and SiW<sub>12</sub>@ZIF-67, (B) Co<sub>3</sub>O<sub>4</sub> and SiW<sub>12</sub>@Co<sub>3</sub>O<sub>4</sub>.



**Fig. S2** IR spectra of ZIF-67, SiW<sub>12</sub>, SiW<sub>12</sub>@ZIF-67 and SiW<sub>12</sub>@Co<sub>3</sub>O<sub>4</sub>.



**Fig. S3** Zeta potential of  $\text{Co}_3\text{O}_4$  and  $\text{SiW}_{12}@Co_3O_4$ .



**Fig. S4** (A) SEM images of  $\text{Co}_3\text{O}_4$ . (B), (C) and (D) TEM images of  $\text{Co}_3\text{O}_4$ .

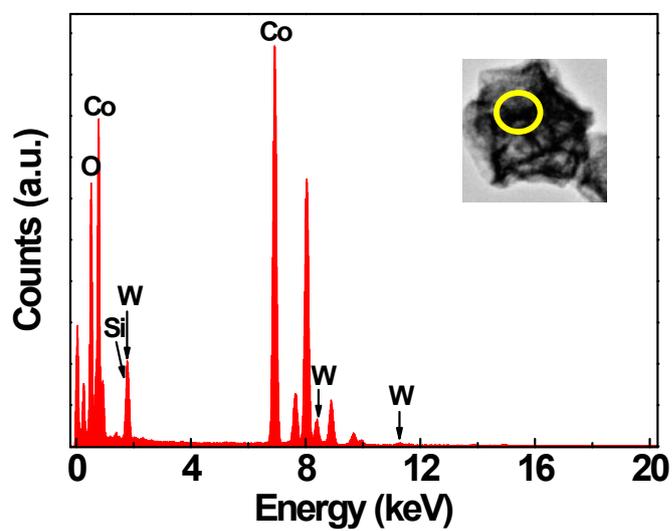


Fig. S5 EDS spectrum of SiW<sub>12</sub>@Co<sub>3</sub>O<sub>4</sub>.

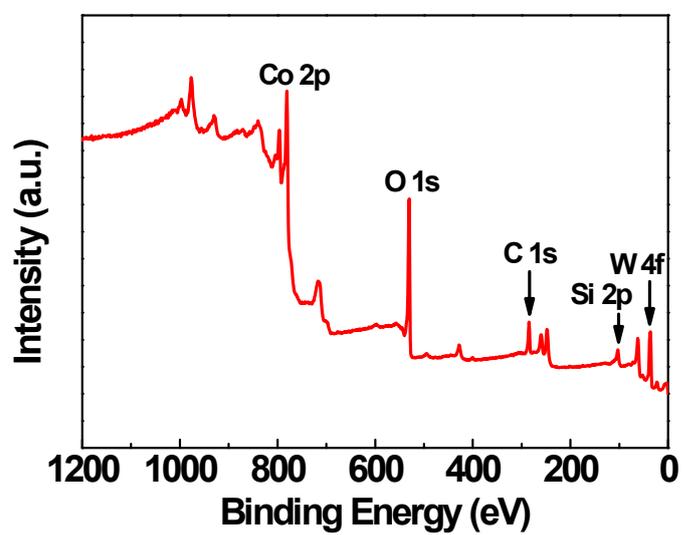


Fig. S6 XPS full survey spectrum of SiW<sub>12</sub>@Co<sub>3</sub>O<sub>4</sub>.

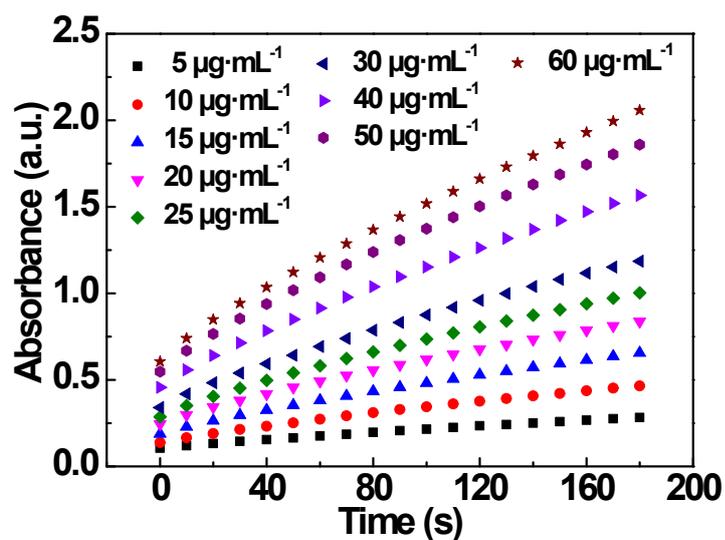


Fig. S7 Absorbance at 652 nm changing with time with different SiW<sub>12</sub>@Co<sub>3</sub>O<sub>4</sub> dosages.

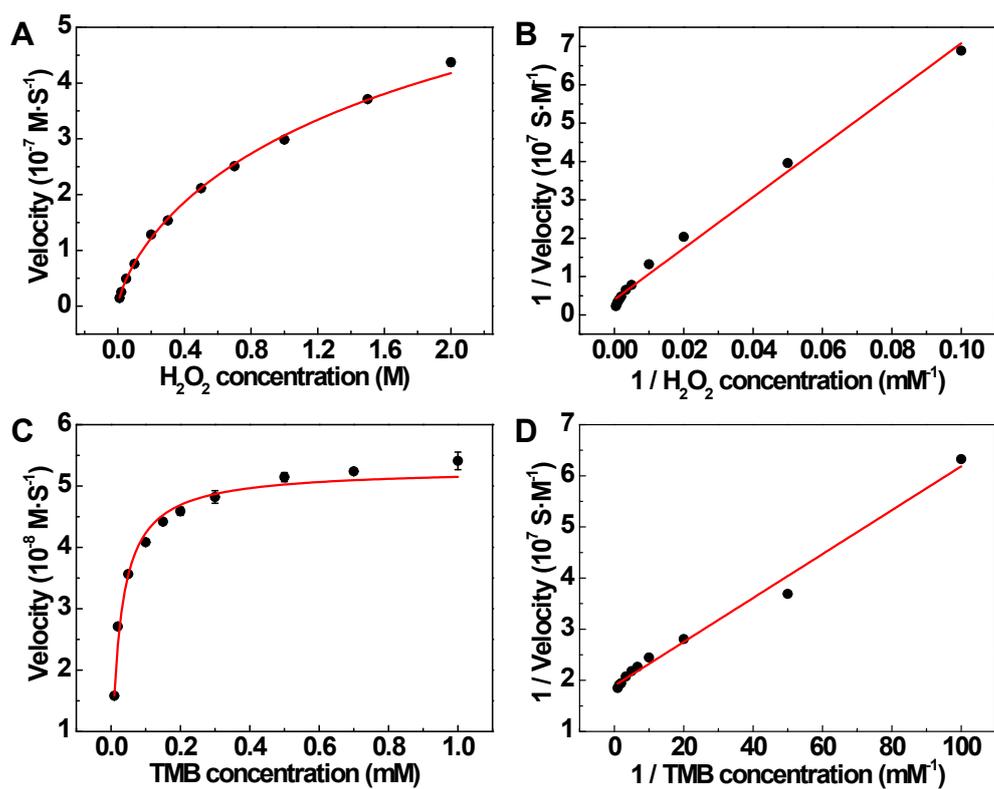
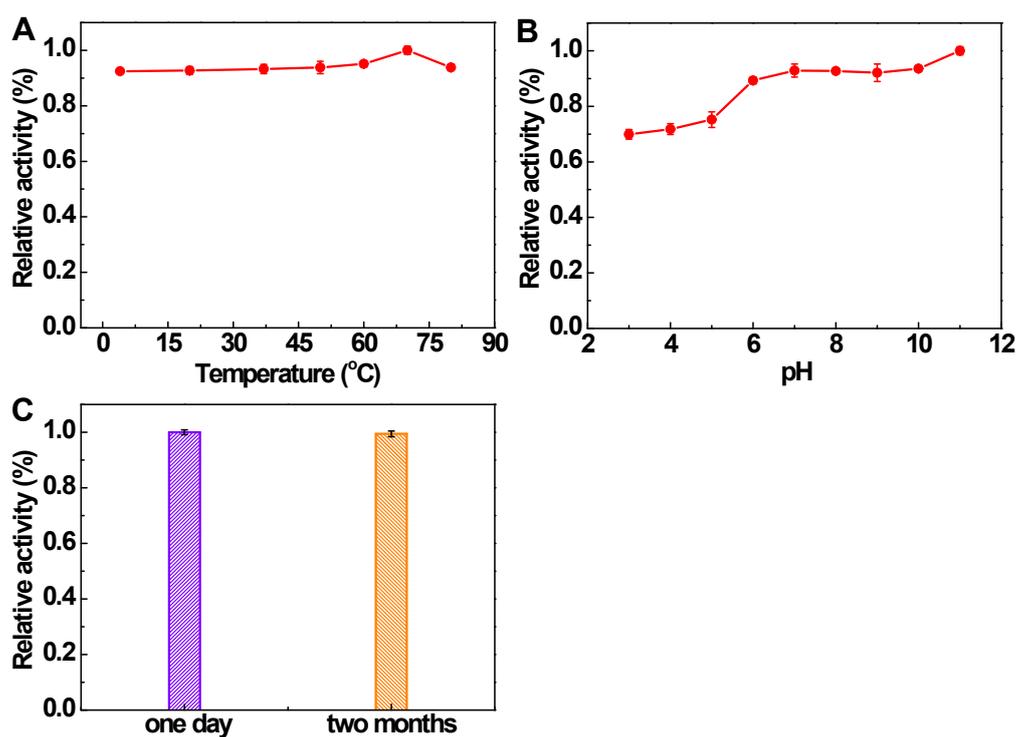
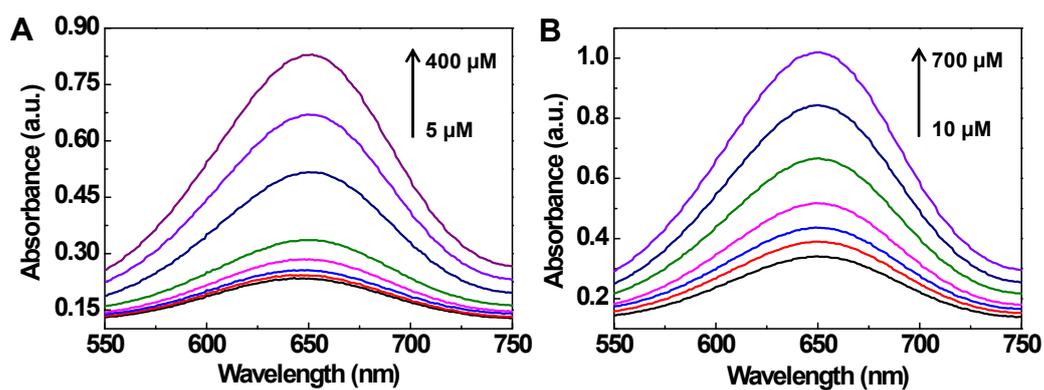


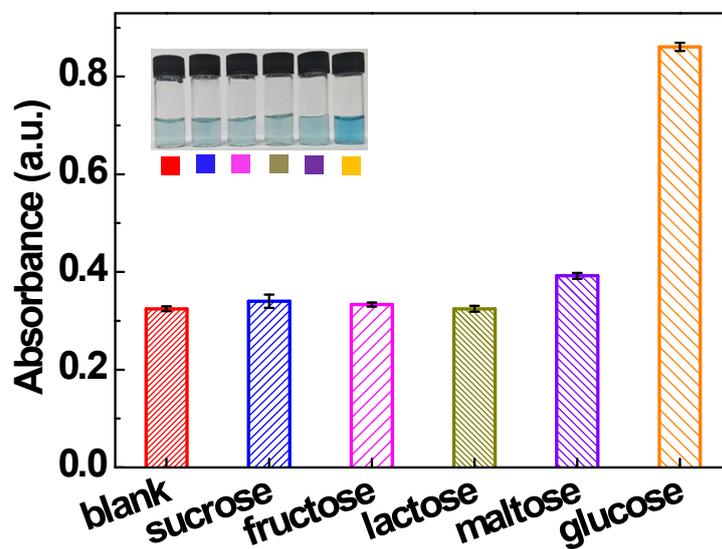
Fig. S8 Steady-state kinetic of SiW<sub>12</sub>@Co<sub>3</sub>O<sub>4</sub>.



**Fig. S9** Relative peroxidase-mimicking activity of  $\text{SiW}_{12}\text{@Co}_3\text{O}_4$  after incubation for 2 h at various (A) temperatures and (B) pH. (C) Relative activity of  $\text{SiW}_{12}\text{@Co}_3\text{O}_4$  after being stored in water for different times.



**Fig. S10** The absorption spectra of oxidation product of TMB in various (A)  $\text{H}_2\text{O}_2$  concentrations (5, 10, 20, 50, 100, 200, 300, and 400  $\mu\text{M}$ ) and (B) glucose concentrations (10, 50, 100, 200, 300, 500, and 700  $\mu\text{M}$ ).



**Fig. S11** Selectivity of colorimetric methods for glucose. Inset image: photographs of corresponding solutions.

**Table S1** Kinetic parameters of  $\text{SiW}_{12}@ \text{Co}_3\text{O}_4$  and HRP.

Catalysts	$K_m$ [mM]		$V_{\max}$ [ $10^{-8}\text{M s}^{-1}$ ]		Reference
	$\text{H}_2\text{O}_2$	TMB	$\text{H}_2\text{O}_2$	TMB	
HRP	3.7	0.434	8.71	10.00	<sup>1</sup>
$\text{SiW}_{12}@ \text{Co}_3\text{O}_4$	167.8	0.023	25.1	5.3	This work

**Table S2** Performance comparison between the proposed sensing method and other H<sub>2</sub>O<sub>2</sub> and glucose colorimetric sensors.

Catalysts	H <sub>2</sub> O <sub>2</sub> detection (μM)		Glucose detection (μM)		Reference
	Linear range	Detection limit	Linear range	Detection limit	
Co <sub>3</sub> O <sub>4</sub> -MMT NPS	10-100	8.7	—	—	2
Cu NCs	10-1000	10	100-2000	100	3
Au@Pt NRs	45-1000	45	45-400	45	4
N-GODs	20-1170	5.3	25-375	16	5
H <sub>2</sub> TCPP-NiO	20-100	8.0	50-500	20	6
CeO <sub>2</sub> /NT-TiO <sub>2</sub> @0.1	5-100	3.2	10-500	6.1	7
NiFe-LDHNS	10-500	4.4	50-2000	23	8
GO-FeTPyP	20-500	72	—	—	9
CuO-Au	—	—	0-30	6.75	10
SiW <sub>12</sub> @Co <sub>3</sub> O <sub>4</sub>	5-400	1.0	10-700	3.3	This work

**Table S3** Determination of glucose in orange juice and human urine.

Samples	Spiked (mM)	Found (mM)	Recovery (%)	RSD (%; n=3)
Orange juice	0	0.209	—	2.8
	0.1	0.312	103	0.8
	0.2	0.412	102	1.6
	0.3	0.515	102	1.7
Urine	0	0	—	—
	0.1	0.104	104	2.4
	0.2	0.203	102	1.6
	0.3	0.321	107	2.4

## References

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