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Supporting infromation

Title: Ultrasensitive electrochemical detection of ochratoxin A based on signal amplification by one-pot synthesized flower-like PEDOT-AuNFs supported on graphene oxide sponge

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Fig.S1

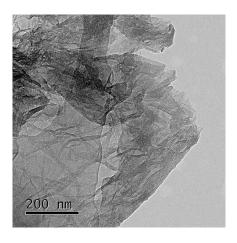


Fig. S1 HRTEM image of GOS

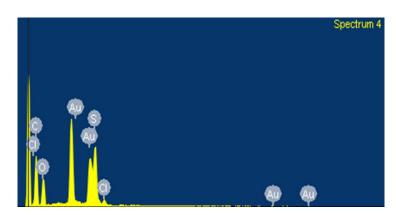


Fig. S2 EDX of PEDOT-AuNFs

Fig.S2



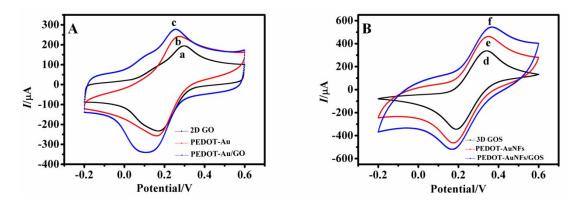


Fig. S3 (A) CVs of electrodes modified by (a) 2D GO, (b) PEDOT-Au, (c) PEDOT-Au/GO; (B) CVs of (d) 3D GOS, (e) PEDOT-AuNFs, (f) PEDOT-AuNFs/GOS, respectively. All of CV curves were recorded from -0.2 V to 0.6 V in the solution of 5 mM $Fe(CN)_6^{3-/4-}$ as a redox probe and 0.2 M KCl at a scan rate of 100mV/s.

Fig.S4

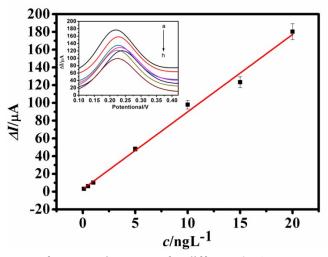


Fig. S4 Calibration curve of DPV peak currents for different OTA concentrations from 0.01 ng/L to 20 ng/L. The inset shows DPV responses of the electrochemical aptasensor to different concentrations of OTA (from a to h: 0, 0.1, 0.5, 1, 5, 10, 15, 20 ng/L). $\Delta I(\mu A)=8.7c(ng/L)+2.8$ ($\Delta I=I_{(BSA/aptamer/PEDOT-AuNFs/GOS)}-I_{(OTA/BSA/aptamer/PEDOT-AuNFs/GOS)}$), R²=0.9921. DPV curves were recorded from -0.2 V to 0.6 V in the solution of 5 mM Fe(CN)₆^{3-/4-} as a redox probe and 0.2 M KCl at a scan rate of 100mV/s.

Table S1

Table S1 The electroactive surface area (A) of different modified electrodes	
Electrode	A (cm ²)
GCE	0.094
GO/GCE	0.171
PEDOT-Au/GCE	0.213
PEDOT-Au/GO/GCE	0.256
GOS/GCE	0.299
PEDOT-AuNFs/GCE	0.367
PEDOT-AuNFs /GOS/GCE	0.444

Table S1 The electroactive surface area (A) of different modified electrodes

Note: The data are calculated based on the Randles–Sevcike equation $Ip = 2.65 \times 10^5 n^{3/2} AD^{1/2} v^{1/2}C$, where Ip is the peak current, n is the transferring electron number, A is the electroactive area (cm²), D is the diffusion coefficient, v is the scanning rate, and C is the concentration of the substrate. The diffusion coefficient of K₃[Fe(CN)₆] is 7.6 × $10^6 cm^2/s$.