

Electronic Supplementary Information

Metal Salt Assisted Electrospray Ionization Mass Spectrometry for Soft Ionization of GAP Polymers in Negative Ion Mode

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Table S1. Comparison of the observed peaks in the systems with NaNO₃ and related parameters in negative ion mode with variation in applied voltage

condition	NaNO ₃ or not	series	composition	observed peaks										M _n	M _w	PDI				
			basic unit	M ₁	486,	585,	684,	783,	882,	981,	1080,	1179,	1278,	1377,	1476,	1575,	1674			
-2.5 kV	N-with	A	[M ₁ +NO ₃] ⁻	m/z	647,	746,	845,	944,	1043,	1142,	1241,	1340,	1439,	1538,	1637		949.1	989.2	1.042	
-3.0 kV	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548,	647,	746,	845,	944,	1043,	1142,	1241,	1340,	1439,	1538,	1637,	1736	942.7	997.7	1.058
-3.5 kV	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548,	647,	746,	845,	944,	1043,	1142,	1241,	1340,	1439,	1538,	1637,	1736	979.5	1031.8	1.053
-4.0 kV	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548,	647,	746,	845,	944,	1043,	1142,	1241,	1340,	1439,	1538,	1637,	1736	990.6	1040.6	1.050
			basic unit	M ₂	414,	513,	612,	711,	810,	909,	1008,	1107,	1206,	1305,	1404					
-2.5 kV	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575,	674,	773,	872,	971,	1070,	1169,	1268,	1367				810.2	833.2	1.028	
-3.0 kV	N-with	B	[M ₂ +NO ₃] ⁻	m/z	476,	575,	674,	773,	872,	971,	1070,	1169,	1268,	1367,	1466		802.6	851.7	1.061	
-3.5 kV	N-with	B	[M ₂ +NO ₃] ⁻	m/z	476,	575,	674,	773,	872,	971,	1070,	1169,	1268,	1367,	1466,	1565	812.0	860.4	1.060	
-4.0 kV	N-with	B	[M ₂ +NO ₃] ⁻	m/z	476,	575,	674,	773,	872,	971,	1070,	1169,	1268,	1367,	1466,	1565	797.9	841.0	1.054	

Note: "N-with" means "negative ion mode and with NaNO₃". A and B mean series A and B in Figure 3 in the main text. Due the relative low peak intensity for other series, the M_n, M_w and PDI values were not calculated.

Table S2. Comparison of the observed peaks in the systems with NaNO₃ and related parameters in negative ion mode with variation in concentration of NaNO₃

condition	NaNO ₃ or not	series	composition	observed peaks										M _n	M _w	PDI				
			basic unit	M ₁	486,	585,	684,	783,	882,	981,	1080,	1179,	1278,	1377,	1476,	1575,	1674			
0 mmol L ⁻¹	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548,	647,	746,	845,	944,	1043,	1142,	1241					693.1	713.1	1.029	
0.1 mmol L ⁻¹	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548,	647,	746,	845,	944,	1043,	1142,	1241,	1340,	1439,	1538,	1637	804.8	854.0	1.061	
0.5 mmol L ⁻¹	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548,	647,	746,	845,	944,	1043,	1142,	1241,	1340,	1439,	1538,	1637,	1736	1010.0	1053.6	1.043
1.0 mmol L ⁻¹	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548,	647,	746,	845,	944,	1043,	1142,	1241,	1340,	1439,	1538,	1637,	1736	1042.6	1093.1	1.048
1.5 mmol L ⁻¹	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548,	647,	746,	845,	944,	1043,	1142,	1241,	1340,	1439,	1538,	1637,		1027.2	1070.5	1.042
2.0 mmol L ⁻¹	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548,	647,	746,	845,	944,	1043,	1142,	1241,	1340,	1439,	1538,	1637,		994.8	1040.9	1.046
3.0 mmol L ⁻¹	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548,	647,	746,	845,	944,	1043,	1142,	1241,	1340,	1439,	1538,	1637,		1038.8	1093.2	1.052
5.0 mmol L ⁻¹	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548,	647,	746,	845,	944,	1043,	1142,	1241					857.9	889.9	1.037	
			basic unit	M ₂	414,	513,	612,	711,	810,	909,	1008,	1107,	1206,	1305,	1404					
0 mmol L ⁻¹	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575,	674,	773,	872									673.9	687.3	1.020	
0.1 mmol L ⁻¹	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575,	674,	773,	872,	971,	1070,	1169,	1268					732.4	756.0	1.032	
0.5 mmol L ⁻¹	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575,	674,	773,	872,	971,	1070,	1169,	1268,	1367,	1466			836.9	874.4	1.045	
1.0 mmol L ⁻¹	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575,	674,	773,	872,	971,	1070,	1169,	1268,	1367,	1466			823.3	870.8	1.057	
1.5 mmol L ⁻¹	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575,	674,	773,	872,	971,	1070,	1169,	1268,	1367,	1466			803.4	848.0	1.056	
2.0 mmol L ⁻¹	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575,	674,	773,	872,	971,	1070,	1169,	1268,	1367,	1466			789.1	834.3	1.057	
3.0 mmol L ⁻¹	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575,	674,	773,	872,	971,	1070,	1169,	1268,	1367,	1466			798.4	859.9	1.077	
5.0 mmol L ⁻¹	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575,	674,	773,	872,	971,	1070							718.9	741.9	1.032	

Note: "N-with" means "negative ion mode and with NaNO₃". A and B mean series A and B in Figure 4 in the main text. Due the relative low peak intensity for other series, the M_n, M_w and PDI values were not calculated.

Table S3. Comparison of the observed peaks in the systems with NaNO₃ and related parameters in negative ion mode with variation in the types of metal salts

metal salt	NaNO ₃ or not	series	composition	observed peaks											M_n	M_w	PDI			
		basic unit	M ₁	486, 585, 684, 783, 882, 981, 1080, 1179, 1278, 1377, 1476, 1575, 1674																
AgNO ₃	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548	647	746	845	944	1043	1142	1241	1340	1439	1538	1637	1736	1014.1	1054.3	1.040
Mg(NO ₃) ₂	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548	647	746	845	944	1043	1142	1241	1340	1439	1538	1637	1736	946.8	997.4	1.053
Co(NO ₃) ₂	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548	647	746	845	944	1043	1142	1241	1340	1439	1538	1637		1010.5	1054.3	1.043
Ni(NO ₃) ₂	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548	647	746	845	944	1043	1142	1241	1340	1439	1538	1637		894.8	949.1	1.061
Cu(NO ₃) ₂	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548	647	746	845	944	1043	1142	1241	1340	1439	1538	1637	1736	1020.2	1070.8	1.050
Zn(NO ₃) ₂	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548	647	746	845	944	1043	1142	1241	1340	1439	1538	1637	1736	980.2	1032.4	1.053
Ba(NO ₃) ₂	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548	647	746	845	944	1043	1142	1241	1340	1439	1538	1637		923.0	968.3	1.049
Pb(NO ₃) ₂	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548	647	746	845	944	1043	1142	1241	1340	1439	1538	1637		952.9	997.2	1.046
Cr(NO ₃) ₃	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548	647	746	845	944	1043	1142	1241	1340	1439	1538	1637	1736	1000.2	1048.0	1.048
Al(NO ₃) ₃	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548	647	746	845	944	1043	1142	1241	1340	1439	1538			905.5	952.8	1.052
Fe(NO ₃) ₃	N-with	A	[M ₁ +NO ₃] ⁻	m/z	548	647	746	845	944	1043	1142	1241	1340	1439	1538	1637		905.7	949.3	1.048
		basic unit	M ₂	414, 513, 612, 711, 810, 909, 1008, 1107, 1206, 1305, 1404																
AgNO ₃	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575	674	773	872	971	1070	1169	1268	1367	1466	1565			841.7	875.3	1.040
Mg(NO ₃) ₂	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575	674	773	872	971	1070	1169	1268	1367	1466				830.8	867.1	1.044
Co(NO ₃) ₂	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575	674	773	872	971	1070	1169	1268	1367	1466				824.6	868.5	1.053
Ni(NO ₃) ₂	N-with	B	[M ₂ +NO ₃] ⁻	m/z	476	575	674	773	872	971	1070	1169	1268	1367				763.9	801.9	1.050
Cu(NO ₃) ₂	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575	674	773	872	971	1070	1169	1268	1367	1466				847.1	891.6	1.053
Zn(NO ₃) ₂	N-with	B	[M ₂ +NO ₃] ⁻	m/z	476	575	674	773	872	971	1070	1169	1268	1367	1466			808.2	854.1	1.057
Ba(NO ₃) ₂	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575	674	773	872	971	1070	1169	1268	1367					825.4	858.6	1.040
Pb(NO ₃) ₂	N-with	B	[M ₂ +NO ₃] ⁻	m/z	476	575	674	773	872	971	1070	1169	1268	1367	1466	1565		795.1	833.8	1.049
Cr(NO ₃) ₃	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575	674	773	872	971	1070	1169	1268	1367	1466	1565			854.4	893.1	1.045
Al(NO ₃) ₃	N-with	B	[M ₂ +NO ₃] ⁻	m/z	476	575	674	773	872	971	1070	1169	1268	1367	1466			753.9	787.5	1.044
Fe(NO ₃) ₃	N-with	B	[M ₂ +NO ₃] ⁻	m/z	575	674	773	872	971	1070	1169	1268	1367					779.6	807.1	1.035

Note: "N-with" means "negative ion mode and with NaNO₃". A and B mean series A and B in Figure S1 as below. Due the relative low peak intensity for other series, the M_n , M_w and PDI values were not calculated.

Table S4. Comparison of the observed peaks in the systems with different salts and related parameters in negative ion modes

salt	with or not	series	composition	observed peaks							M_n	M_w	PDI
	basic unit		M_1	486, 585, 684, 783, 882, 981, 1080, 1179, 1278, 1377									
NaF	N-with		$[M_1+F]^-$	m/z	n/a								
NaCl	N-with	C	$[M_1+Cl]^-$	m/z	521, 620, 719, 818, 917, 1016, 1115, 1214, 1313, 1412, 1511, 1610, 1709					986.8	1029.2	1.043	
NaBr	N-with	E	$[M_1+Br]^-$	m/z	567, 666, 765, 864, 963, 1062, 1161, 1260, 1359, 1458, 1557, 1656, 1755					1082.1	1126.4	1.041	
NaI	N-with	H	$[M_1+I]^-$	m/z	712, 811, 910, 1009, 1108, 1207, 1306, 1405, 1504, 1603, 1702					1065.6	1103.5	1.035	
NaNO ₃	N-with	A	$[M_1+NO_3]^-$	m/z	548, 647, 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538, 1637, 1736					968.2	1033.8	1.068	
Na ₂ SO ₄	N-with	L	$[M_1+HSO_4]^-$	m/z	682, 781, 880, 979, 1078, 1177, 1276, 1375, 1474, 1573					1028.8	1063.4	1.034	
Na ₃ PO ₄	N-with	L	$[M_1+H_2PO_4]^-$	m/z	583, 682, 781, 880, 979, 1078, 1177, 1276					821.6	856.5	1.043	
	basic unit		M_2	414, 513, 612, 711, 810, 909, 1008, 1107, 1206									
NaF	N-with		$[M_2+F]^-$	m/z	n/a								
NaCl	N-with	A	$[M_2+Cl]^-$	m/z	548, 647, 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538					816.6	858.0	1.051	
NaBr	N-with	D	$[M_2+Br]^-$	m/z	594, 693, 792, 891, 990, 1089, 1188, 1287, 1386, 1485, 1584					900.8	950.5	1.055	
NaI	N-with	G	$[M_2+I]^-$	m/z	640, 739, 838, 937, 1036, 1135, 1234, 1333, 1432, 1531					904.0	935.9	1.035	
NaNO ₃	N-with	O	$[M_2+NO_3]^-$	m/z	476, 575, 674, 773, 872, 971, 1070, 1169, 1268, 1367, 1466					805.9	858.9	1.066	
Na ₂ SO ₄	N-with	M	$[M_2+HSO_4]^-$	m/z	511, 610, 709, 808, 907, 1006, 1105					810.0	841.6	1.039	
Na ₃ PO ₄	N-with	M	$[M_2+H_2PO_4]^-$	m/z	511, 610, 709, 808, 907, 1006					666.6	691.5	1.037	
	basic unit		M_3	442, 541, 640, 739									
NaF	N-with		$[M_3+F]^-$	m/z	n/a								
NaCl	N-with	B	$[M_3+Cl]^-$	m/z	477, 576, 675, 774, 873, 972					736.6	759.2	1.031	
NaBr	N-with	C	$[M_3+Br]^-$	m/z	622, 721, 820, 919, 1018, 1117, 1216					868.4	905.9	1.043	
NaI	N-with	I	$[M_3+I]^-$	m/z	688, 767, 866, 965,					801.7	814.8	1.016	
NaNO ₃	N-with	K	$[M_3+NO_3]^-$	m/z	504, 603, 702, 801					641.4	650.0	1.013	
Na ₂ SO ₄	N-with		$[M_3+HSO_4]^-$	m/z	n/a								
Na ₃ PO ₄	N-with	N	$[M_3+NaH_2PO_4]^-$	m/z	759, 858, 957, 1056, 1155, 1254, 1353, 1452, 1551					1052.2	1083.5	1.030	
	basic unit		M_4	756, 855, 954, 1053, 1152, 1251									
NaF	N-with		$[M_4+F]^-$	m/z	n/a								
NaCl	N-with	D	$[M_4+Cl]^-$	m/z	890, 989, 1088, 1187, 1286, 1385, 1484, 1583, 1682					1206.8	1249.5	1.035	
NaBr	N-with	F	$[M_4+Br]^-$	m/z	936, 1035, 1134, 1233, 1332, 1431, 1530, 1629, 1728					1292.6	1326.9	1.027	
NaI	N-with	J	$[M_4+I]^-$	m/z	1081, 1180, 1279, 1378, 1477, 1576, 1675, 1774					1352.0	1379.3	1.020	
NaNO ₃	N-with	C	$[M_4+NO_3]^-$	m/z	521, 620, 719, 818, 917, 1016, 1115, 1214, 1313, 1412, 1511, 1610, 1709					1052.3	1123.9	1.068	
Na ₂ SO ₄	N-with		$[M_4+HSO_4]^-$	m/z	n/a								
Na ₃ PO ₄	N-with		$[M_4+H_2PO_4]^-$	m/z	n/a								

Note: "N-with" means negative ion mode and with related salt, and "n/a" means no related peaks in mass spectrum.

Table S5. Comparison of the observed peaks and related parameters of different GAP samples without and with addition of NaNO₃ in negative ion modes with paper spray

sample	with or not	series	composition	observed peaks	<i>M_n</i>	<i>M_w</i>	<i>PDI</i>
			M ₁	387, 486, 585, 684, 783, 882, 981, 1080, 1179, 1278, 1377, 1476, 1575, 1674, 1773, 1872			
GAP #2	N-without	A	[M ₁ +NO ₃] ⁻	m/z 449, 548, 647, 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538, 1637, 1736, 1835, 1934	789.2	881.7	1.117
	N-with	A	[M ₁ +NO ₃] ⁻	m/z 548, 647, 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538, 1637, 1736, 1835, 1934	934.2	1033.0	1.106
	basic unit		M ₂	513, 612, 711, 810, 909, 1008, 1107, 1206, 1305, 1404, 1503, 1602, 1701, 1800			
	N-without	B	[M ₂ +NO ₃] ⁻	m/z 575, 674, 773, 872, 971, 1070, 1169, 1268, 1367, 1466, 1565, 1664, 1763, 1862	924.9	1028.6	1.112
	N-with	B	[M ₂ +NO ₃] ⁻	m/z 575, 674, 773, 872, 971, 1070, 1169, 1268, 1367, 1466, 1565, 1664, 1763, 1862	957.2	1068.7	1.116
	basic unit		M ₃	541, 640, 739, 838, 937, 1036, 1135, 1234, 1333, 1432, 1531, 1630, 1729			
N-without	C	[M ₃ +NO ₃] ⁻	m/z 603 702, 801, 900, 999, 1098, 1197, 1296	785.9	831.3	1.058	
N-with	C	[M ₃ +NO ₃] ⁻	m/z 603 702, 801, 900, 999, 1098, 1197, 1296, 1395, 1494, 1593, 1692, 1791	1036.4	1116.2	1.077	
			M ₁	387, 486, 585, 684, 783, 882, 981, 1080, 1179, 1278, 1377, 1476, 1575, 1674, 1773, 1872			
GAP #3	N-without	A	[M ₁ +NO ₃] ⁻	m/z 548, 647, 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538, 1637, 1736	795.0	869.6	1.094
	N-with	A	[M ₁ +NO ₃] ⁻	m/z 548, 647, 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538, 1637, 1736, 1835, 1934	1127.1	1198.2	1.063
	basic unit		M ₂	513, 612, 711, 810, 909, 1008, 1107, 1206, 1305, 1404, 1503, 1602, 1701, 1800			
	N-without	B	[M ₂ +NO ₃] ⁻	m/z 575, 674, 773, 872, 971, 1070, 1169, 1268	758.0	790.6	1.043
	N-with	B	[M ₂ +NO ₃] ⁻	m/z 575, 674, 773, 872, 971, 1070, 1169, 1268, 1367, 1466, 1565, 1664, 1763, 1862	1093.0	1163.9	1.065
	basic unit		M ₃	541, 640, 739, 838, 937, 1036, 1135, 1234, 1333, 1432, 1531, 1630, 1729			
	N-without	C	[M ₃ +NO ₃] ⁻	m/z 603 702	648.4	652.2	1.006
	N-with	C	[M ₃ +NO ₃] ⁻	m/z 603 702, 801, 900, 999, 1098, 1197, 1296, 1395	929.0	973.2	1.048
	basic unit		M ₄	1152, 1251, 1350, 1449, 1548, 1647, 1746, 1845			
	N-without	D	[M ₄ +NO ₃] ⁻	m/z 1214, 1313, 1412, 1511, 1610, 1709, 1808	1426.5	1447.2	1.015
N-with	D	[M ₄ +NO ₃] ⁻	m/z 1214, 1313, 1412, 1511, 1610, 1709, 1808, 1907	1498.3	1523.9	1.017	
			M ₁	387, 486, 585, 684, 783, 882, 981, 1080, 1179, 1278, 1377, 1476, 1575, 1674, 1773, 1872			
GAP #4	N-without	A	[M ₁ +NO ₃] ⁻	m/z 548, 647, 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538, 1637, 1736, 1835	897.7	981.8	1.094
	N-with	A	[M ₁ +NO ₃] ⁻	m/z 548, 647, 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538, 1637, 1736, 1835, 1934	1134.3	1195.0	1.054
	basic unit		M ₂	513, 612, 711, 810, 909, 1008, 1107, 1206, 1305, 1404, 1503, 1602, 1701, 1800			
	N-without	B	[M ₂ +NO ₃] ⁻	m/z 575, 674, 773, 872, 971, 1070, 1169, 1268, 1367, 1466, 1565, 1664	962.4	1046.1	1.087
	N-with	B	[M ₂ +NO ₃] ⁻	m/z 575, 674, 773, 872, 971, 1070, 1169, 1268, 1367, 1466, 1565, 1664, 1763, 1862, 1961	1154.1	1241.3	1.076
	basic unit		M ₃	541, 640, 739, 838, 937, 1036, 1135, 1234, 1333, 1432, 1531, 1630, 1729			
	N-without	C	[M ₃ +NO ₃] ⁻	m/z 603 702, 801	691.3	699.5	1.012
	N-with	C	[M ₃ +NO ₃] ⁻	m/z 603 702, 801, 900, 999, 1098, 1197, 1296, 1395, 1494, 1593, 1692, 1791	1105.8	1164.0	1.053

Note: "N-without" means negative ion mode and without NaNO₃, and "N-with" means negative ion mode and with NaNO₃.

Table S6. Comparison of the observed peaks and related parameters of different GAP samples without and with addition of NaNO₃ in negative ion modes using nanoESI

sample	with or not	series	composition	observed peaks	<i>M_n</i>	<i>M_w</i>	<i>PDI</i>
GAP #1			M ₁	387, 486, 585, 684, 783, 882, 981, 1080, 1179, 1278, 1377			
		A	[M ₁ +NO ₃] ⁻	m/z 449, 548, 647, 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538	839.6	891.4	1.062
		E	[M ₁ +X*] ⁻	m/z 484, 583, 682, 781, 880, 979, 1078, 1177	831.6	864.4	1.039
		A	[M ₁ +NO ₃] ⁻	m/z 647, 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538, 1637	1079.5	1111.5	1.030
			M ₂	315, 414, 513, 612, 711, 810, 909, 1008, 1107, 1206, 1305, 1404			
		B	[M ₂ +NO ₃] ⁻	m/z 674, 773, 872, 971, 1070	883.6	911.2	1.031
		D	[M ₂ +X*] ⁻	m/z 412, 511, 610, 709, 808	615.7	638.7	1.038
		B	[M ₂ +NO ₃] ⁻	m/z 674, 773, 872, 971, 1070, 1169, 1268, 1367, 1466	938.5	969.0	1.033
			M ₃	442, 541, 640, 739			
			[M ₃ +NO ₃] ⁻	m/z n/a			
			[M ₃ +NO ₃] ⁻	m/z n/a			
	GAP #2			M ₄	558, 657, 756, 855, 954, 1053, 1152, 1251, 1350, 1449		
		C	[M ₄ +NO ₃] ⁻	m/z 620, 719, 818, 917, 1016, 1115, 1214, 1313, 1412, 1511	941.1	976.8	1.038
		C	[M ₄ +NO ₃] ⁻	m/z 917, 1016, 1115, 1214, 1313, 1412, 1511, 1610	1225.2	1260.5	1.029
			M ₁	387, 486, 585, 684, 783, 882, 981, 1080, 1179, 1278, 1377, 1476, 1575, 1674, 1773, 1872			
		A	[M ₁ +NO ₃] ⁻	m/z 449, 548, 647, 746, 845, 944, 1043, 1142, 1241	906.9	964.1	1.063
			[M ₁ +NO ₃] ⁻	m/z n/a			
			M ₂	315, 414, 513, 612, 711, 810, 909, 1008, 1107, 1206, 1305, 1404, 1503, 1602, 1701, 1800			
		D	[M ₂ +X*] ⁻	m/z 412, 511, 610, 709, 808, 907, 1006, 1105, 1204, 1303, 1402	817.6	902.6	1.104
		B	[M ₂ +NO ₃] ⁻	m/z 674, 773, 872, 971, 1070, 1169, 1268, 1367, 1466, 1565, 1664, 1763, 1862	1227.6	1298.3	1.058
			M ₃	640, 739, 838, 937, 1036, 1135, 1234, 1333, 1432, 1531, 1630, 1729, 1828			
			[M ₃ +NO ₃] ⁻	m/z n/a			
		F	[M ₃ +NO ₃] ⁻	m/z 702, 801, 900, 999, 1098, 1197, 1296, 1395, 1494, 1593, 1692, 1791, 1890	1333.0	1380.1	1.035
GAP #3			M ₄	950, 1049, 1148, 1247, 1346, 1445, 1544			
			[M ₄ +NO ₃] ⁻	m/z n/a			
		H	[M ₄ +NO ₃] ⁻	m/z 1012, 1111, 1210, 1309, 1408, 1507, 1606	1312.0	1329.6	1.013
			M ₅	999, 1098, 1197, 1296, 1395, 1494, 1593, 1692			
			[M ₅ +NO ₃] ⁻	m/z n/a			
		G	[M ₅ +NO ₃] ⁻	m/z 1061, 1160, 1259, 1358, 1457, 1556, 1655, 1754	1324.3	1348.1	1.018
			M ₁	387, 486, 585, 684, 783, 882, 981, 1080, 1179, 1278, 1377, 1476, 1575, 1674, 1773, 1872			
		A	[M ₁ +NO ₃] ⁻	m/z 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538	1079.5	1118.6	1.036
		A	[M ₁ +NO ₃] ⁻	m/z 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538, 1637, 1736, 1835, 1934	1329.4	1382.6	1.040
			M ₂	315, 414, 513, 612, 711, 810, 909, 1008, 1107, 1206, 1305, 1404, 1503, 1602			
		D	[M ₂ +X*] ⁻	m/z 412, 511, 610, 709, 808, 907, 1006, 1105, 1204, 1303	870.7	917.1	1.053
		B	[M ₂ +NO ₃] ⁻	m/z 872, 971, 1070, 1169	1027.6	1038.2	1.010
	B	[M ₂ +NO ₃] ⁻	m/z 674, 773, 872, 971, 1070, 1169, 1268, 1367, 1466, 1565, 1664, 1763, 1862	1109.0	1155.5	1.042	
GAP #4			M ₃	640, 739, 838, 937, 1036, 1135, 1234, 1333, 1432, 1531, 1630			
			[M ₃ +NO ₃] ⁻	m/z n/a			
		F	[M ₃ +NO ₃] ⁻	m/z 702, 801, 900, 999, 1098, 1197, 1296, 1395	1022.4	1051.0	1.028
			M ₁	387, 486, 585, 684, 783, 882, 981, 1080, 1179, 1278, 1377, 1476, 1575, 1674, 1773, 1872			
		A	[M ₁ +NO ₃] ⁻	m/z 449, 548, 647, 746, 845, 944, 1043, 1142, 1241, 1340, 1439, 1538, 1637	1119.6	1186.9	1.060
			[M ₁ +NO ₃] ⁻	m/z n/a			
			M ₂	315, 414, 513, 612, 711, 810, 909, 1008, 1107, 1206, 1305, 1404, 1503, 1602			
		D	[M ₂ +X*] ⁻	m/z 412, 511, 610, 709, 808, 907, 1006, 1105, 1204, 1303, 1402	910.1	971.5	1.067
		B	[M ₂ +NO ₃] ⁻	m/z 872, 971, 1070, 1169, 1268, 1367, 1466	1150.5	1180.6	1.026
		B	[M ₂ +NO ₃] ⁻	m/z 674, 773, 872, 971, 1070, 1169, 1268, 1367, 1466, 1565, 1664, 1763, 1862	1255.1	1309.0	1.043
			M ₃	640, 739, 838, 937, 1036, 1135, 1234, 1333, 1432, 1531, 1630, 1729, 1828			
			[M ₃ +NO ₃] ⁻	m/z n/a			
	F	[M ₃ +NO ₃] ⁻	m/z 702, 801, 900, 999, 1098, 1197, 1296, 1395, 1494, 1593, 1692, 1791, 1890	1269.5	1313.0	1.034	
		M ₄	788, 887, 986, 1085, 1184, 1283, 1382, 1481, 1580, 1679, 1778				
		[M ₄ +NO ₃] ⁻	m/z n/a				
	H	[M ₄ +NO ₃] ⁻	m/z 850, 949, 1048, 1147, 1246, 1345, 1444, 1543, 1642, 1741, 1840	1345.7	1378.8	1.025	

Note: "N-without" means negative ion mode and without NaNO₃, and "N-with" means negative ion mode and with NaNO₃; X* is either HSO₄⁻ or H₂PO₄⁻; "n/a" means no signal.

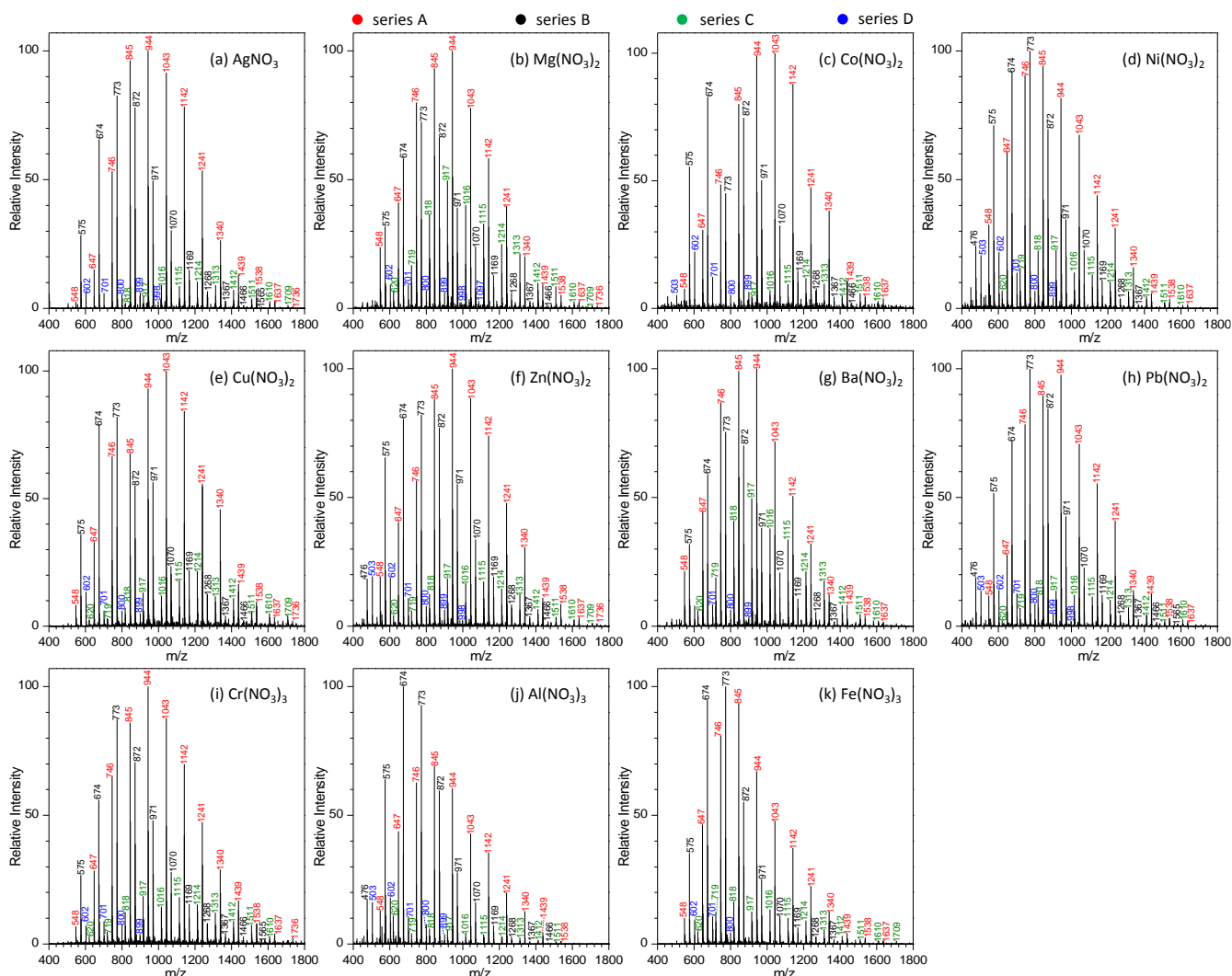


Figure S1. Effect of the type of metal salt on the analysis of 100 $\mu\text{g mL}^{-1}$ GAP #1: (a) AgNO_3 , (b) $\text{Mg}(\text{NO}_3)_2$, (c) $\text{Co}(\text{NO}_3)_2$, (d) $\text{Ni}(\text{NO}_3)_2$, (e) $\text{Cu}(\text{NO}_3)_2$, (f) $\text{Zn}(\text{NO}_3)_2$, (g) $\text{Ba}(\text{NO}_3)_2$, (h) $\text{Pb}(\text{NO}_3)_2$, (i) $\text{Cr}(\text{NO}_3)_3$, (j) $\text{Al}(\text{NO}_3)_3$, and (k) $\text{Fe}(\text{NO}_3)_3$ (sample volume: 25 μL for each spray; applied voltage: -3.5 kV; concentration of each metal salt in GAP #1 solution: 1.0 mmol L^{-1}).

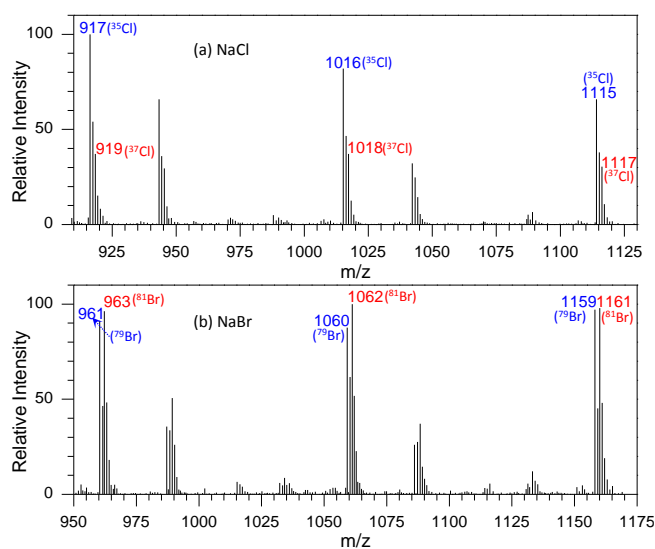


Figure S2. Mass spectra of GAP #1 solution in the presence of (a) NaCl and (b) NaBr (sample volume: 25 μL for each spray; applied voltage: -3.5 kV; concentration of each metal salt in GAP #1 solution: 1.0 mmol L^{-1}).

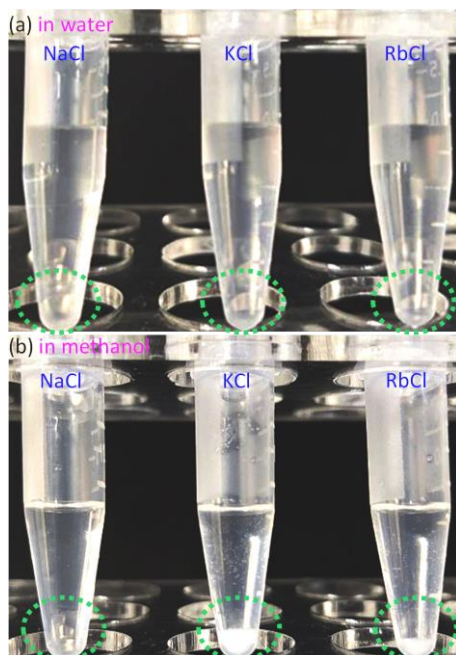


Figure S3. Comparison of the solubility of NaCl, KCl, and RbCl **(a)** in water and **(b)** in methanol (*Note:* For dissolving different salts into water, exactly measured 2.5 mmol of corresponding salts into 1.0 mL deionized water followed by vibration for 5.0 min; for dissolving different salts into methanol, transferred 100 μL of 2.5 mmol mL^{-1} the related aqueous solutions into 900 μL methanol followed by vibration for 5.0 min).

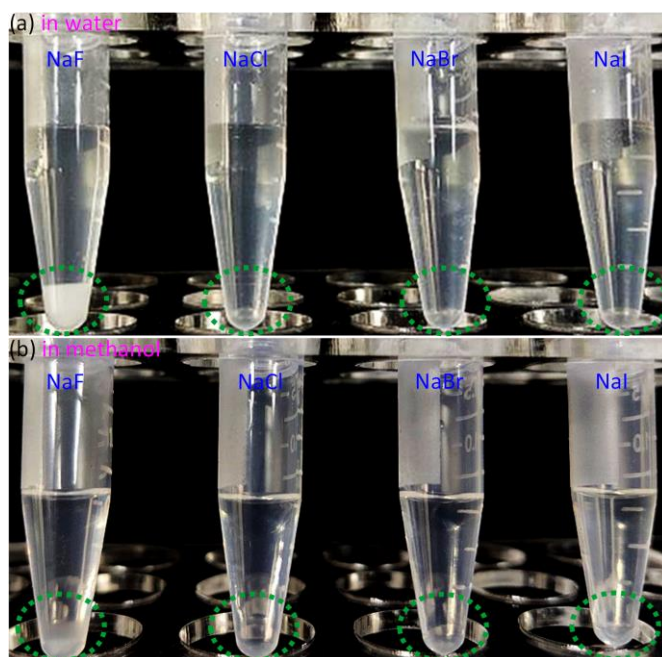


Figure S4. Comparison of the solubility of NaF, NaCl, NaBr, and NaI **(a)** in water and **(b)** in methanol (*Note:* For dissolving different salts into water, exactly measured 2.5 mmol of corresponding salts into 1.0 mL deionized water followed by vibration for 5.0 min; for dissolving different salts into methanol, transferred 100 μL of 2.5 mmol mL^{-1} the related aqueous solutions into 900 μL methanol followed by vibration for 5.0 min).

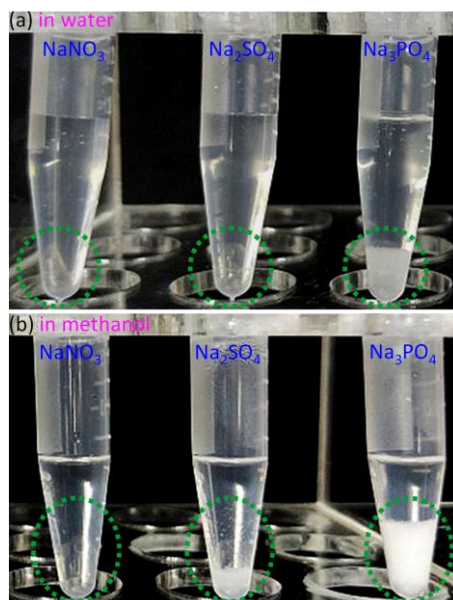


Figure S5. Comparison of the solubility of NaNO_3 , Na_2SO_4 , and Na_3PO_4 (a) in water and (b) in methanol (Note: For dissolving different salts into water, exactly measured 2.5 mmol of corresponding salts into 1.0 mL deionized water followed by vibration for 5.0 min; for dissolving different salts into methanol, transferred 100 μL of 2.5 mmol mL^{-1} the related aqueous solutions into 900 μL methanol followed by vibration for 5.0 min).