# **Electronic Supplementary Information**

# Ligand structure and charge state-dependent separation of monolayer protected Au<sub>25</sub> clusters using non-aqueous reversed-phase HPLC

Korath Shivan Sugi,<sup>a</sup> Shridevi Bhat,<sup>a</sup> Abhijit Nag,<sup>a</sup> Ganesan Paramasivam,<sup>a</sup> Ananthu Mahendranath,<sup>a,b</sup> and Thalappil Pradeep<sup>\*a</sup>

<sup>a</sup> DST Unit of Nanoscience (DST UNS) and Thematic Unit of Excellence (TUE), Department of Chemistry, Indian Institute of Technology Madras, Chennai 600 036, India <sup>b</sup> Department of Metallurgical and Materials Engineering, Indian Institute of Technology Madras, Chennai 600036, India.

\*E-mail: <u>pradeep@iitm.ac.in</u>

# **Table of contents**

SL. No.	Description	Page No.	
	Calculation of selectivity ( $\alpha$ )	3	
	Computational methods	3	
Fig. S1	Schematic representation of functionalized silica beads of reversed-phase columns	5	
Fig. S2	Negative ion MALDI MS of cluster I, II, and III.	5	
Fig. S3	Linear gradient program used for the separation	6	
Fig. S4	Flow rate vs. eluent volume required in linear gradient [10] program	6	
Fig. S5	Online UV-vis spectra of cluster I in C <sub>18</sub> column with different flow rates	7	
Fig. S6	Online UV-vis spectra of cluster II in C <sub>18</sub> column with different flow rates	8	
Fig. S7	Online UV-vis spectra of cluster III in $C_{18}$ column with different flow rates	9	
Fig. S8	Variation of peak width of half maxima at different	10	

	flow rates of cluster I, II, and III in $C_{18}$ column	
Fig. S9	Chromatograms of cluster a) I, b) II, and c) III at different flow rates in $C_8$ column	10
Fig. S10	Online UV-vis spectra of cluster I, II, and III in $C_8$ column with different flow rates	11
Fig. S11	Variation of peak width of half maxima at different flow rates of cluster I, II, and III in C <sub>8</sub> column	12
Fig. S12	Online UV-vis spectra of cluster I and II in a phenylhexyl column with different flow rates	12
Fig. S13	Variation of peak width of half maxima at different flow rates of cluster I, II, and III in phenylhexyl column	13
Table S1	Fitting parameters for the exponential fits	14
Fig. S14	Online UV-vis spectra of cluster IIIa, III, and mixture of both eluted in phenylhexyl column with 1.0 mL/min flow rate	15
Fig. S15	The energy minimum structures of octadecyl silyl functionality docked with cluster I, II, and III	15
Fig. S16	The energy minimum structures of octyl silyl functionality docked with cluster I, II, and III	16
Fig. S17	Concentration-dependent chromatograms of cluster III in the phenylhexyl column at 1.0 mL/min flow rate	16
Fig. S18	Peak area vs. concentration of cluster III in the phenylhexyl column at 1mL/min flow rate	17

#### **Calculation of selectivity (α)**

The selectivity ( $\alpha$ ) of the method is estimated using equation (1).

$$\alpha = \frac{K_2}{K_1} = \frac{t_{R2} - t_0}{t_{R1} - t_0} \tag{1}$$

Here, K = retention factor of the separation

 $t_{\rm R}$  = retention time of the peak

 $t_0 = \text{dead time}$ 

#### **Computational methods**

#### **DFT calculations**

The structures of Au<sub>25</sub> cluster protected with different thiols such as dodecanethiol (DDT) and octanethiol (OT) were built from the crystal structure of  $Au_{25}(PET)_{18}$  cluster by replacing the phenylethanethiol (PET) ligands. All the geometries were optimized using density functional theory (DFT) with projector augmented waves (PAW) as implemented in the GPAW software package.<sup>1</sup> The atomic PAW setup was used as Au(5d<sup>10</sup>6s<sup>1</sup>), S(3s<sup>2</sup>3p<sup>4</sup>), C(2s<sup>2</sup>2p<sup>2</sup>) and H(1s<sup>1</sup>), with scalar-relativistic effects included for Au. Further, these atomic orbitals were described by the DZP (double zeta plus polarization) basis set with the Perdew-Burke-Ernzerhof (PBE) exchange-correlation functional<sup>2</sup> in LCAO mode<sup>3</sup> to effectively include the interactions in the considered clusters. The geometries were relaxed with a grid spacing of 0.2 Å, and the convergence was set to minimize the residual forces without any symmetry constraints by 0.05 eV/Å.

#### **Molecular docking calculations**

Molecular docking studies were performed using AutoDock4.2 and its associated tools.<sup>4</sup> DFToptimized geometries of clusters I, II, and III were used for the study. The different columns were computationally modeled by constructing a SiO<sub>2</sub> surface of ~ 5 nm by cutting out from its crystal structure,<sup>5</sup> and the column functionalities such as octadecyl silyl, octyl silyl, and phenylhexyl silyl were attached to the silicon atoms on the SiO<sub>2</sub> surface. All the structures were built using Avogadro software package.<sup>6</sup> We used the cluster as "ligand" and stationary phase as "receptor". Receptor grids were generated using  $126 \times 126 \times 126$  grid points in XYZ, with a grid spacing of 0.375 Å, and map types were created using AutoGrid-4.2. The grid parameter file (.gpf) was saved using MGL Tools-1.4.6.50. The docking parameter files (.dpf) were generated using MGLTools-1.4.6.50. The results of AutoDock generated an output file (.dlg), and the generated conformers were scored and ranked as per the interaction energy. Ten lowest-energy conformers were obtained. We used the Lamarckian genetic algorithm for the output file using MGLTools-1.4.6.



**Fig. S1** Schematic representation of three different silica beads of reversed-phase HPLC columns. Color labels: blue, C; pink, H; red, Si; yellow, O.





**Fig. S3** Linear gradient program used for the separation of clusters. The label (e.g., [10]) indicates the time (in minutes) taken to replace the mobile phase fully with THF.



**Fig. S4** Eluent volume required to replace 100% MeOH to 100% THF in a linear gradient [10] program with varying flow rate.





Fig. S5 Online UV-vis spectra of cluster I in  $C_{18}$  column at different flow rates, extracted at 400 nm wavelength obtained by PDA detector.



Fig. S6 Online UV-vis spectra of cluster II in  $C_{18}$  column at different flow rates, extracted at 400 nm wavelength obtained by PDA detector.



**Fig. S7** Online UV-vis spectra of cluster III in  $C_{18}$  column at different flow rates, extracted at 400 nm wavelength obtained by PDA detector.



Fig. S8 Variation of peak width of half maxima with different flow rates of cluster I, II, and III in  $C_{18}$  column.



Fig. S9 Chromatograms of cluster a) I, b) II, and c) III at different flow rates in C<sub>8</sub> column.



flow rate obtained by PDA detector.



Fig. S11 Variation of peak width half maxima with different flow rates of cluster I, II, and III in  $C_8$  column.





**Fig.S12** a) and b) are the online UV-vis spectra of cluster I and II in a phenylhexyl column with different flow rates obtained by PDA detector.



**Supplementary information 13** 

**Fig. S13** Variation of peak width of half maxima with different flow rates of cluster I, II, and III in the phenylhexyl column.

## **Table S1** Fitting parameters for the exponential fits shown in Figure 7.

Column	Cluster	A (in min)	b	С	R <sup>2</sup>
C <sub>18</sub>	Au <sub>25</sub> (DDT) <sub>18</sub>	28.136	0.687	18.74	0.992
C <sub>18</sub>	Au <sub>25</sub> (OT) <sub>18</sub>	26.055	1.407	11.013	0.993
C <sub>18</sub>	Au <sub>25</sub> (PET) <sub>18</sub>	37.033	0.951	12.380	0.991
C <sub>8</sub>	Au <sub>25</sub> (DDT) <sub>18</sub>	28.680	0.515	19.530	0.997
C <sub>6</sub>	Au <sub>25</sub> (OT) <sub>18</sub>	27.498	0.523	18.990	0.998
C <sub>6</sub>	Au <sub>25</sub> (PET) <sub>18</sub>	26.914	0.559	15.55	0.998
Phenylhexyl	Au <sub>25</sub> (DDT) <sub>18</sub>	20.980	0.520	17.328	0.998
Phenylhexyl	Au <sub>25</sub> (OT) <sub>18</sub>	22.960	0.5118	16.815	0.998
Phenylhexyl	Au <sub>25</sub> (PET) <sub>18</sub>	22.151	0.545	16.597	0.997

The fitting function has the form  $t_R = A^*(\exp(-x/b) + c$  where 'x' is flow rate in mL/min



**Fig.S14** a), b), and c) are the online UV-vis spectra of cluster IIIa, III, and mixture of both eluted in phenylhexyl column with 1.0 mL/min flow rate obtained by PDA detector.



**Fig. S15** Energy minimum conformers of octadecyl functionality docked with clusters; a) I, b) II, and c) III. Color labels: golden yellow, Si; red, O; grey, C; white, H; teal, Au; yellow, S; green, DDT, OT, and PET ligands.



**Fig. S16** Energy minimum conformers of octyl functionality docked with clusters; a) I, b) II, and c) III. Color labels: golden yellow, Si; red, O; grey, C; white, H; teal, Au; yellow, S; green, DDT, OT, and PET ligands.



**Fig. S17** Concentration-dependent chromatograms of cluster III in the phenylhexyl column at 1.0 mL/min flow rate.



Fig. S18 Peak area vs. concentration of cluster III in the phenylhexyl column at 1.0 mL/min

flow rate.

## References

1. Enkovaara, J.; Rostgaard, C.; Mortensen, J. J.; Chen, J.; Dułak, M.; Ferrighi, L.; Gavnholt, J.; Glinsvad, C.; Haikola, V.; Hansen, H. A.; Kristoffersen, H. H.; Kuisma, M.; Larsen, A. H.; Lehtovaara, L.; Ljungberg, M.; Lopez-Acevedo, O.; Moses, P. G.; Ojanen, J.; Olsen, T.; Petzold, V.; Romero, N. A.; Stausholm-Møller, J.; Strange, M.; Tritsaris, G. A.; Vanin, M.; Walter, M.; Hammer, B.; Häkkinen, H.; Madsen, G. K. H.; Nieminen, R. M.; Nørskov, J. K.; Puska, M.; Rantala, T. T.; Schiøtz, J.; Thygesen, K. S.; Jacobsen, K. W., Electronic structure calculations with GPAW: a real-space implementation of the projector augmented-wave method. *Journal of Physics: Condensed Matter* **2010**, *22* (25), 253202.

2. Perdew, J. P.; Burke, K.; Ernzerhof, M., Generalized Gradient Approximation Made Simple [Phys. Rev. Lett. 77, 3865 (1996)]. *Physical Review Letters* **1997**, *78* (7), 1396-1396.

3. Larsen, A. H.; Vanin, M.; Mortensen, J. J.; Thygesen, K. S.; Jacobsen, K. W., Localized atomic basis set in the projector augmented wave method. *Physical Review B* **2009**, *80* (19), 195112.

4. Morris, G. M.; Huey, R.; Lindstrom, W.; Sanner, M. F.; Belew, R. K.; Goodsell, D. S.; Olson, A. J., AutoDock4 and AutoDockTools4: Automated docking with selective receptor flexibility. *Journal of Computational Chemistry* **2009**, *30* (16), 2785-2791.

5. Prewitt, C. T.; Weidner, D. J.; Levien, L., Structure and elastic properties of quartz at pressure. *American Mineralogist* **1980**, *65* (9-10), 920-930.

6. Hanwell, M. D.; Curtis, D. E.; Lonie, D. C.; Vandermeersch, T.; Zurek, E.; Hutchison, G. R., Avogadro: an advanced semantic chemical editor, visualization, and analysis platform. *Journal of Cheminformatics* **2012**, *4* (1), 17.