## **Supporting Information**

Colorimetric detection of Hg (II) based on the gold amalgamtriggered reductase mimetic activity in aqueous solutions by employing AuNP@Fe-TCPP-MOF nanoparticles

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Fig. S1 TEM image of AuNP@MOF after being treated with Hg2+ ions



**Fig. S2** The UV spectrum of the MB system before (black line) and after incubation with AuNPs@MOF (red line) and the simple mixture of Au NPs and MOF (blue line)



**Fig. S3** Optimization of sensing conditions. (A) The effect of amount of AuNP@MOF on the response signal;  $A_0$  represents the absorbance of the original MB in the sensing system at 665 nm, A represents the absorbance of MB in sensing system by adding various dosage of AuNP@MOF wtih Hg<sup>2+</sup> at 665 nm. (B) The effect of MB concentration on the response signal.  $A_0$  represents the absorbance of the original MB in the sensing system at 665 nm; A represents the absorbance of MB in sensing system at 665 nm; A represents the absorbance of MB in sensing system at 665 nm; A represents the absorbance of ne original MB in the sensing system at 665 nm; A represents the absorbance of MB in sensing system by adding certain Hg<sup>2+</sup> at 665 nm. (C) The effect of NaBH<sub>4</sub> concentration on the response signal.  $A_0$  represents the absorbance of the original NaBH<sub>4</sub> in the sensing system at 665 nm; A represents the absorbance of NaBH<sub>4</sub> in sensing system by adding certain Hg<sup>2+</sup> at 665 nm; A represents the absorbance of NaBH<sub>4</sub> in sensing system by adding certain Hg<sup>2+</sup> at 665 nm; A represents the absorbance of NaBH<sub>4</sub> in sensing system by adding certain Hg<sup>2+</sup> at 665 nm; A represents the absorbance of NaBH<sub>4</sub> in sensing system by adding certain Hg<sup>2+</sup> at 665 nm; A represents the absorbance of NaBH<sub>4</sub> in sensing system by adding certain Hg<sup>2+</sup> at 665 nm; A represents the absorbance of NaBH<sub>4</sub> in sensing system by adding certain Hg<sup>2+</sup> at 665 nm; A represents the absorbance of NaBH<sub>4</sub> in sensing system by adding certain Hg<sup>2+</sup> at 665 nm

Sample	Hg <sup>2+</sup> spiked (pM)	Hg <sup>2+</sup> detected (pM) Mean ± SD	Recovery (%)	
Tap water 1	200	$194.68\pm0.11$	$97.34\pm0.06$	
Tap water 2	250	$253.40\pm0.05$	$101.36\pm0.02$	
Tap water 3	300	$298.68\pm0.13$	$99.56\pm0.04$	
Yihe river water 1	200	$202.92\pm0.03$	$101.46\pm0.02$	
Yihe river water 2	250	$249.65\pm0.05$	$99.86\pm0.02$	
Yihe river water 3	300	$303.90 \pm 0.04$	$101.13 \pm 0.01$	

Table S1 Recovery of tap water, Yihe river water samples containing different concentrations of Hg<sup>2+</sup> measured by UV–Vis spectrometry

Table S2 The comparison of different methods for  $Hg^{2+}$  detection

Material	Method	Linear range	Detection limit	Time	Ref
Au/Hexanedithiol/ Rhodamine B	Fluorescence	0-19.9 µM	2.49 nM	2 mins	1
Colorimetric Sensor based on G- Quadruplex	Colorimetric method	250-1250 nM	50 nM	240 min	2
Carbon nanodots	Fluorescence	0-3 µM	4.2 nM	5 mins	3
Antibodies/BSA-glutathione	Cold-vapor atomic absorption	2.5-49.9 nM	2.99 nM	few weeks	4
DNA/GO	Fluorescence	0-8 nM	1.5 nM	10 mins	5
AuNP/ Iron-porphyric MOF	UV-vis spectra	200- 400 pM	103 pM	2 s	This work

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