

Supplementary Information

A switch-on fluorophore by water molecules via hydrogen bonding and its application of bio-imaging of formaldehyde in living cells

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1. Fluorescence spectra of CHa1

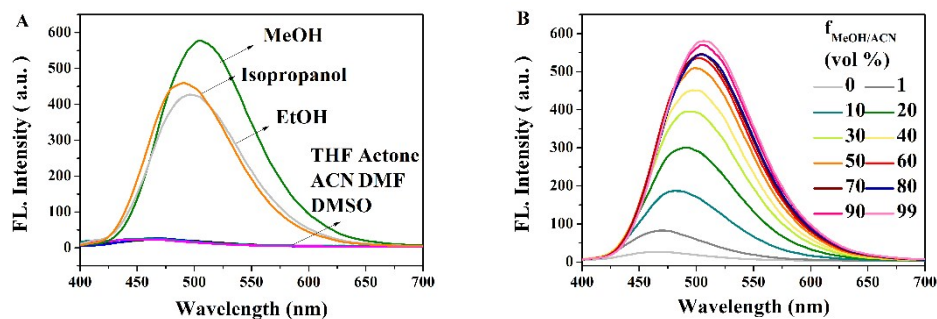


Fig S1. Fluorescence spectra of **CHa1** (10 μM) (A) in different solvents, (B) in MeOH/ACN mixtures with different MeOH fractions. ($\lambda_{\text{ex}} = 363$ nm)

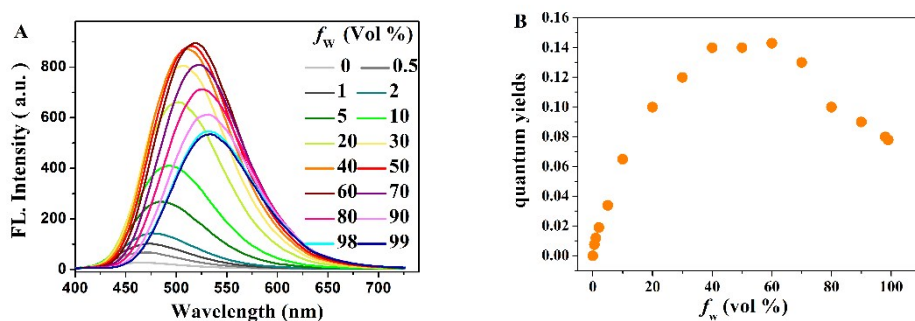


Fig S2. (A) Fluorescence spectra of **CHa1** (10 μM) in ACN/H₂O mixtures with different f_w . (B) Fluorescence quantum yields of **CHa1** (10 μM) in ACN/H₂O mixtures with different f_w . ($\lambda_{\text{ex}} = 363$ nm)

2. ¹H NMR spectra of CHa1

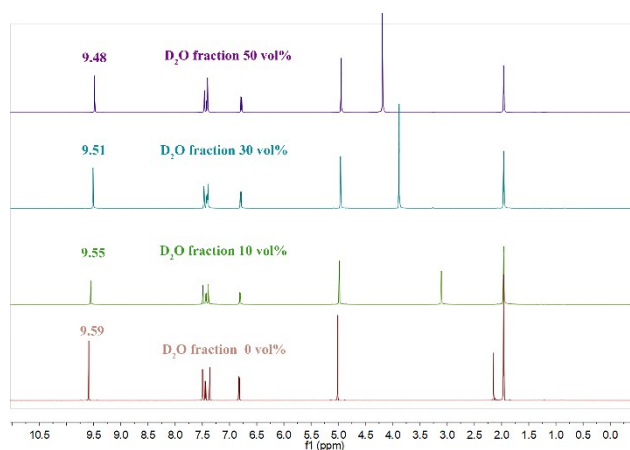


Fig S3. ¹H NMR spectra of **CHa1** (5 mg/mL) in Acetonitrile-*d*₃/D₂O mixture with different D₂O fractions (0, 10, 30, 50 vol%).

3. Temperature effect in fluorescence intensity of CHa1

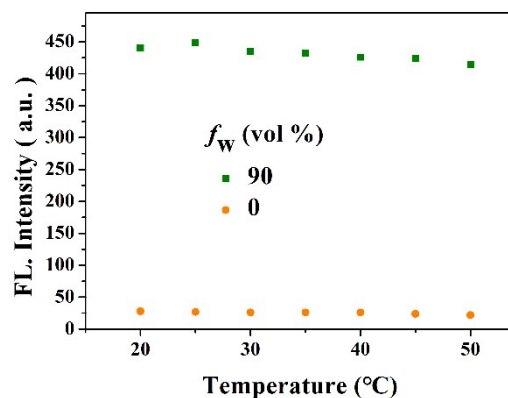


Fig S4. Fluorescence spectra of **CHa1** (10 μ M) at different temperature in H₂O/ACN (90 : 10, v/v) system. (λ_{ex} = 363 nm).

4. pH effect in fluorescence intensity of CHa1

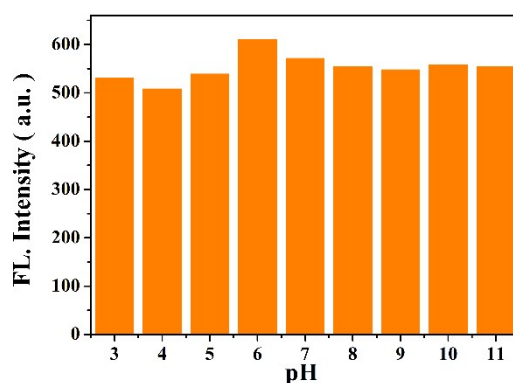


Fig S5. Fluorescence spectra of **CHa1** (10 μ M) at different pH values in H₂O/ACN (90 : 10, v/v) system. (λ_{ex} = 363 nm).

5. Fluorescence intensities and quantum yields of chromene derivatives

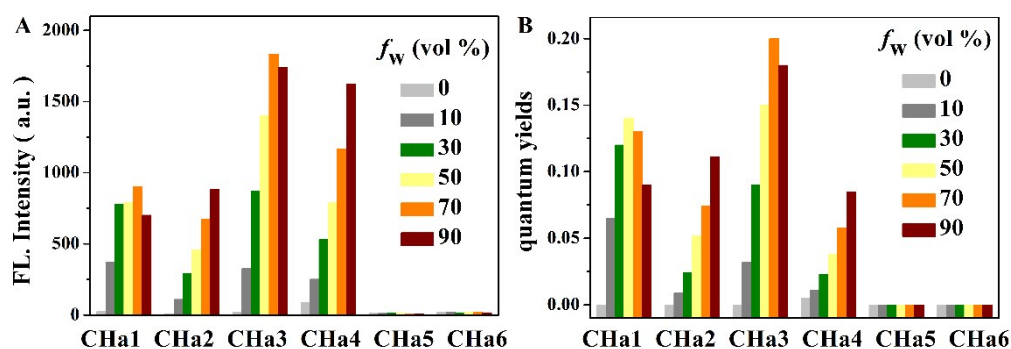


Fig S6. Fluorescence intensities and quantum yields of chromene derivatives (10 μ M) in H₂O/ACN mixtures with different water fractions (f_w : 0 vol%, 10 vol%, 30 vol%, 50 vol%, 70 vol%, 90 vol%). **CHa1** (λ_{ex} = 363 nm). **CHa2** (λ_{ex} = 358 nm). **CHa3** (λ_{ex} = 360 nm). **CHa4** (λ_{ex} = 365 nm). **CHa5** (λ_{ex} = 349 nm). **CHa6** (λ_{ex} = 363 nm).

6. Stability of CHa3

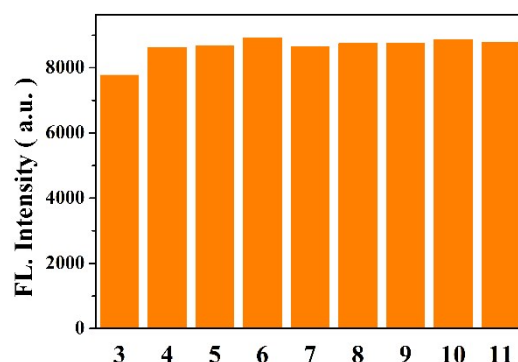


Fig S7. Fluorescence spectra of **CHa3** (100 μ M) at different pH values in H₂O/ACN (90 : 10, v/v) system. (λ_{ex} = 360 nm).

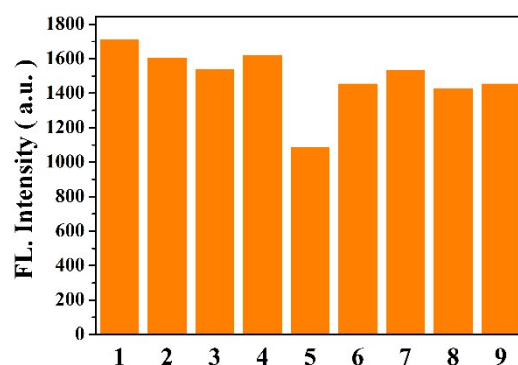


Fig S8. Fluorescence intensity of probe **CHa3** (10 μ M) in the presence of 10 equiv. various analytes in PBS buffer (pH 7.4, containing 10% ACN) for 120 min at 37 °C. (1) blank; (2) cysteine; (3) Homocysteine; (4) glutathione; (5) NaHSO₃; (6) NaHSO₄ (7) Na₂S; (8) ascorbic acid; (9) H₂O₂. (λ_{ex} = 360 nm)

7. HPLC spectra and pH stability of the response

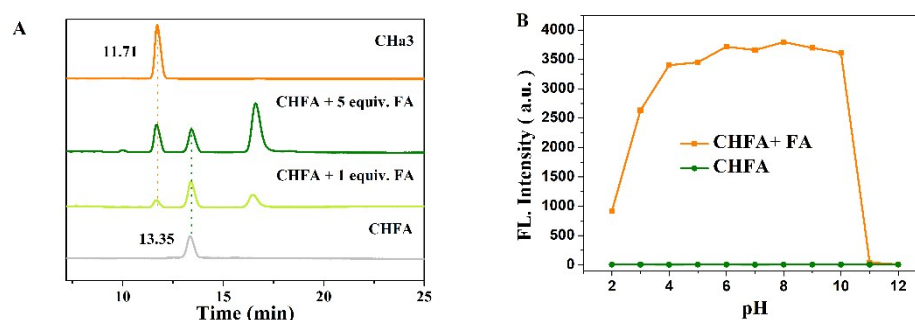


Fig S9. (A) Reverse-phase HPLC chromatogram analysis. The light gray: **CHFA** (20 μ M), The light green line: **CHFA** (20 μ M) + FA (1.0 equiv.), The olive line: **CHFA** (20 μ M) + FA (5.0 equiv.), The orange line: **CHa3** (20 μ M) Conditions: Incubation for 180 min in 10 mM PBS buffer at pH 7.4 containing 10% ACN at 37 °C. The eluent is MeOH/H₂O (v/v, 70: 30) mixed solvent. The flow rate is 0.5 mL/min⁻¹, the detection wavelength is 275 nm. (B) Fluorescence intensity of **CHFA** (20 μ M) at 517 nm under different pH values in the absence and the presence of 100 equiv. of FA in PBS buffer (10 mM, containing 10% ACN). Spectra are recorded after incubation of FA for 180 min. (λ_{ex} = 360 nm)

8. Photostability and cytotoxicity assay of CHa3 and CHFA

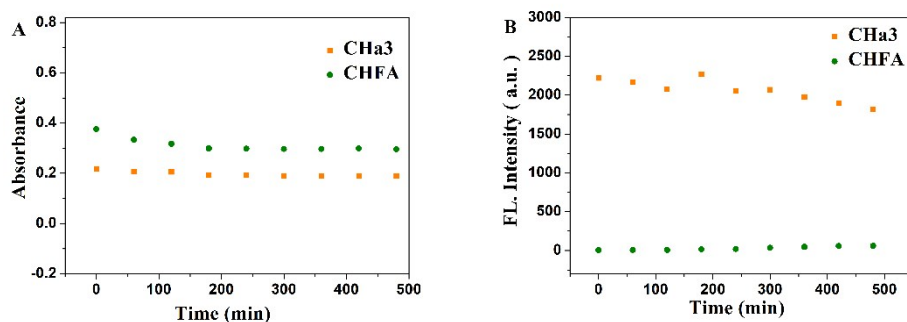


Fig. S10. Time-dependent of **CHa3** (20 μM) and **CHFA** (20 μM) in $\text{H}_2\text{O}/\text{ACN} = 90:10$ (v/v) solution. Under sustained illumination (254 nm). (A) UV absorption spectra. (B) Fluorescence spectra.

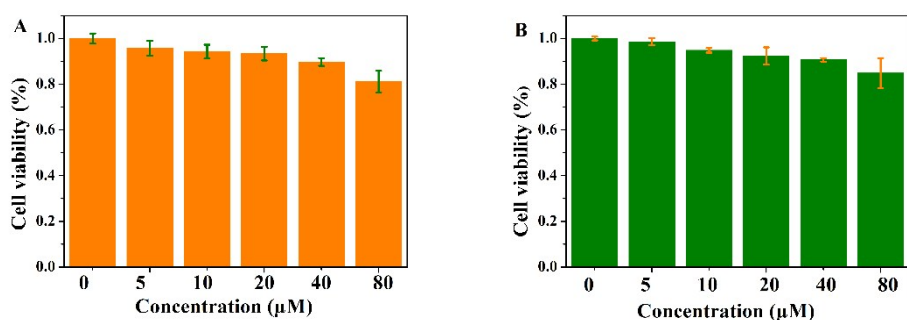


Fig. S11. Cell viabilities of HeLa cells after incubation with different concentrations of **CHa3** (A) and **CHFA** (B). The error bars are standard deviation.

9. Detection limits of CHFA responds to formaldehyde

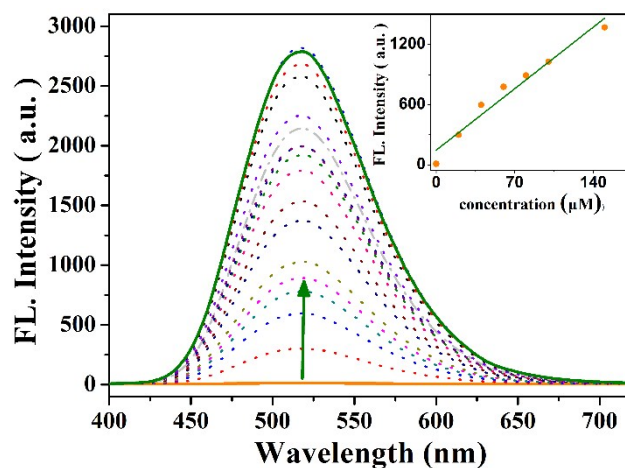


Fig. S12. Fluorescence spectra of 20 μM **CHFA** reaction with FA (0–3.0) mM for 180 min in pH 7.4 PBS buffer (10 mM containing 10% ACN). Excitation at 360 nm. Insert: The linear relationship between the fluorescence intensity and the concentration of FA.

10. Fluorescence lifetime of CHa1

Table 1 Fluorescence lifetime of **CHa1** in the ACN/H₂O mixture with different water fraction.

f_w (vol %)	A_1	T_1/ns	A_2	T_2/ns	T/ns
0	38600000	534.36536	680.63534	4938.55138	535.0833688
10	728.73216	4957.81339	28600000	552.22872	553.2360096
30	3140000	726.64389	510.63283	7882.58709	739.2510779
50	2430000	750.55805	548.16295	7505.85319	765.7796211
70	1890000	784.83447	531.83332	8593.56926	808.8822789

11. Characterization of compounds

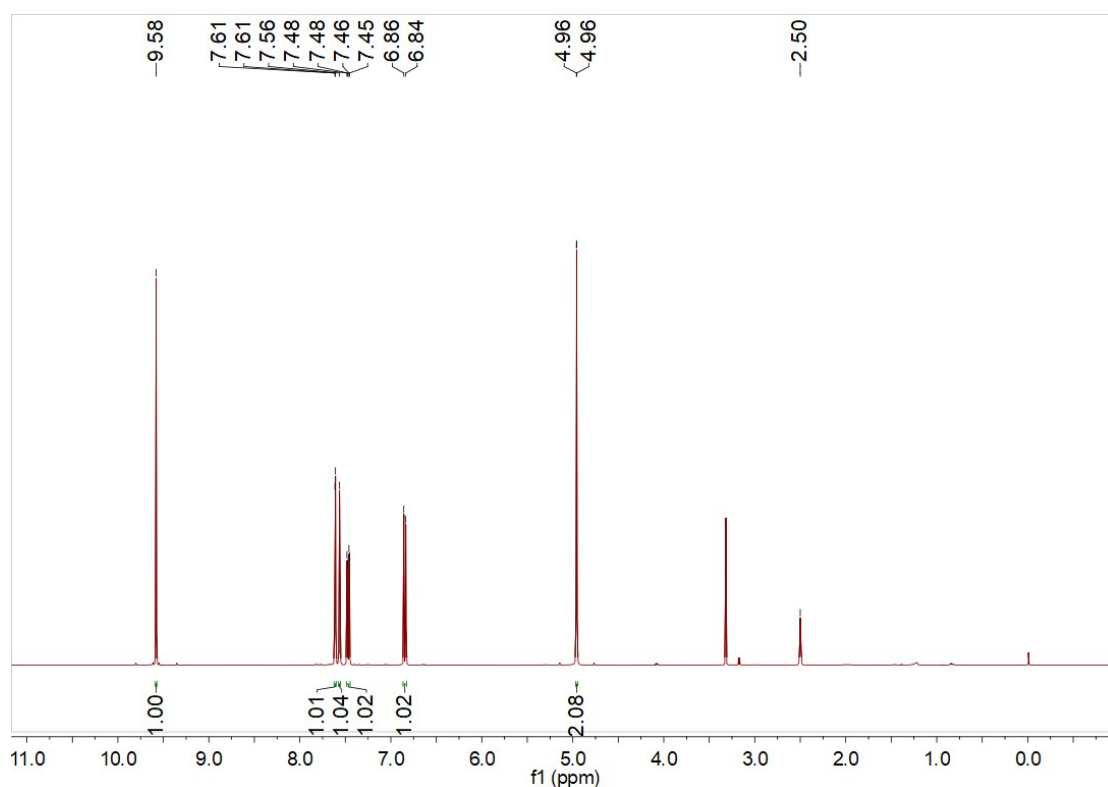


Fig S13a. ¹H NMR spectra of **CHa1** in DMSO-*d*₆.

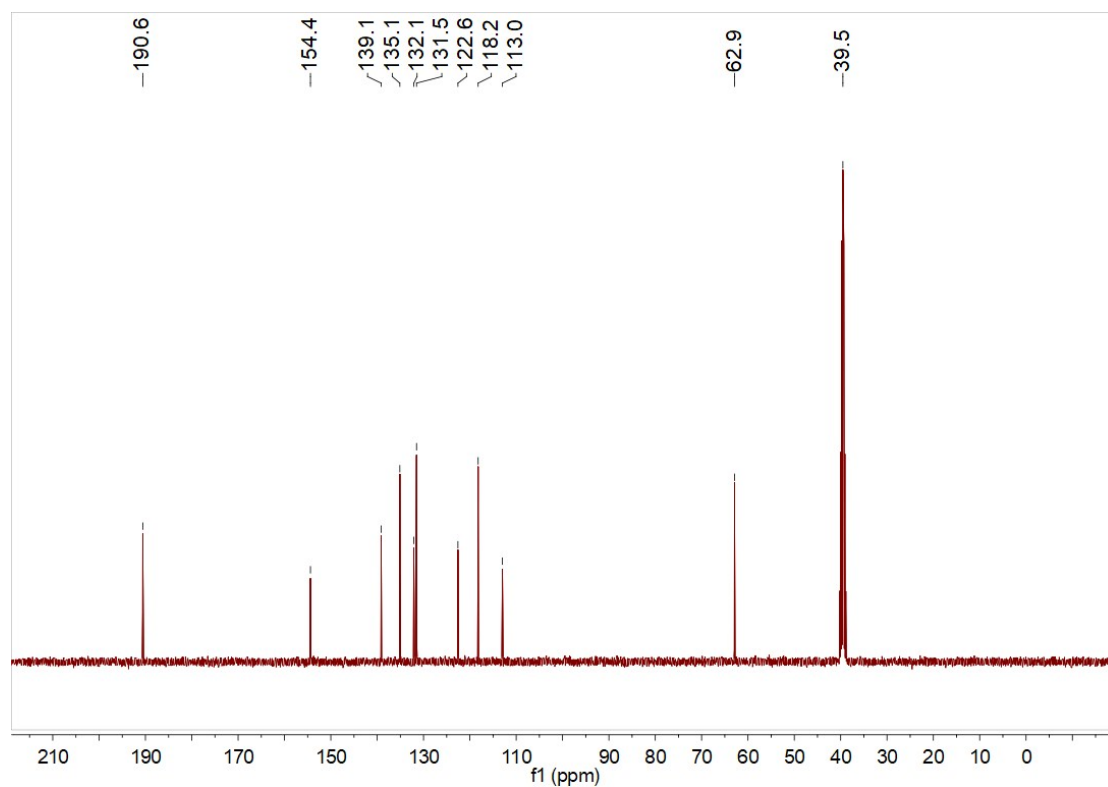


Fig S13b. ^{13}C NMR spectra of CHa1 in $\text{DMSO-}d_6$.

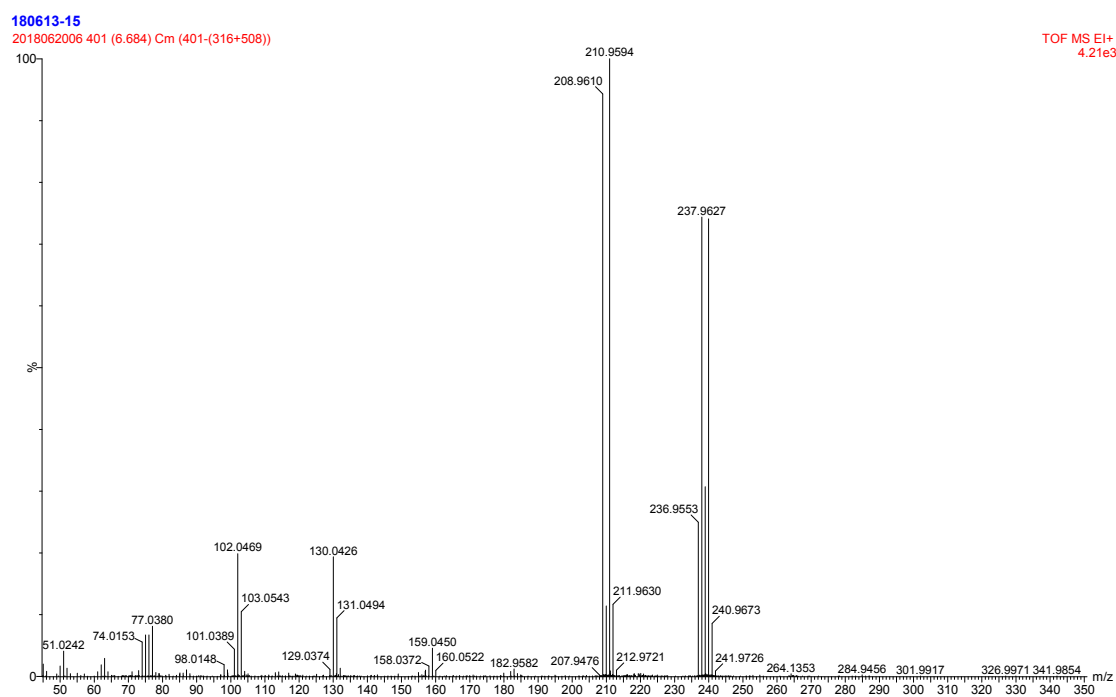


Fig S13c. Mass spectrum of CHa1.

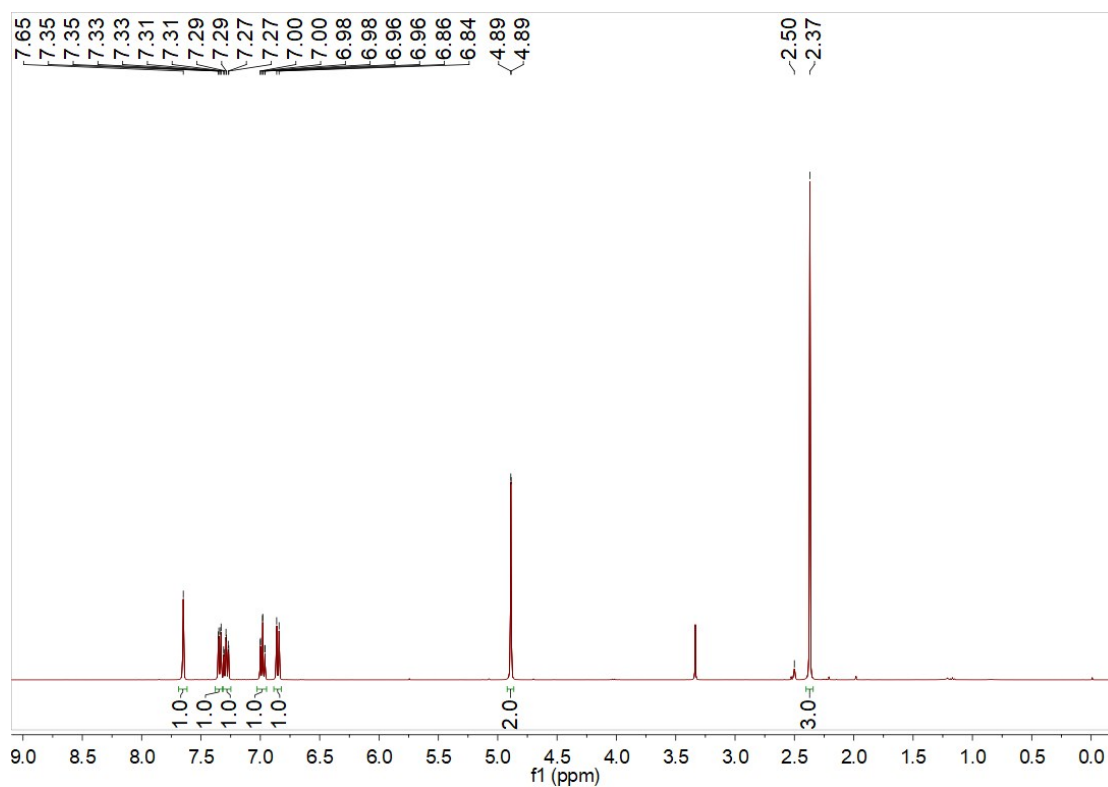


Fig S14a. ^1H NMR spectra of **CHa2** in $\text{DMSO}-d_6$.

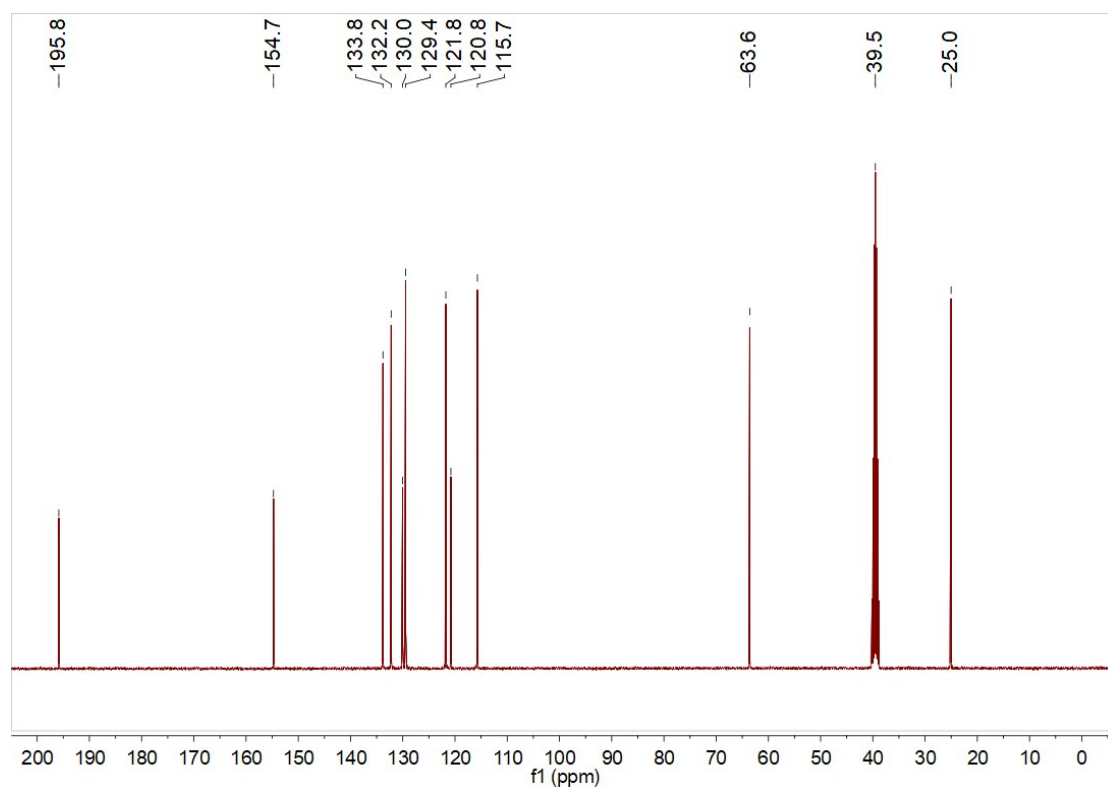


Fig S14b. ^{13}C NMR spectra of **CHa2** in $\text{DMSO}-d_6$.

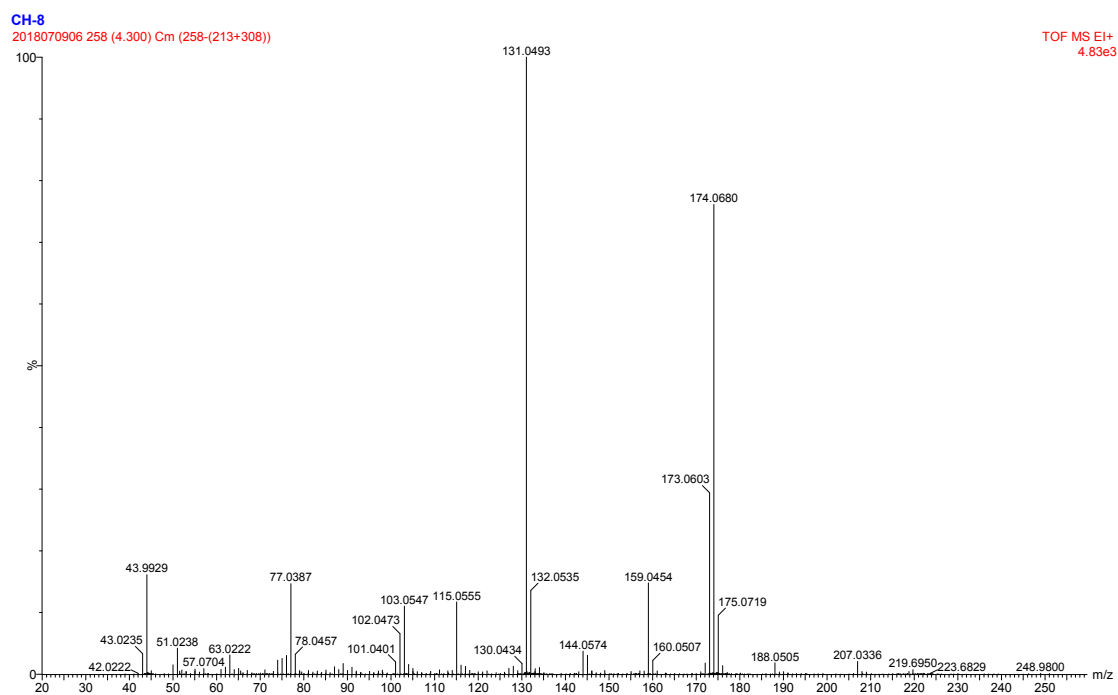


Fig S14c. Mass spectrum of CHa2.

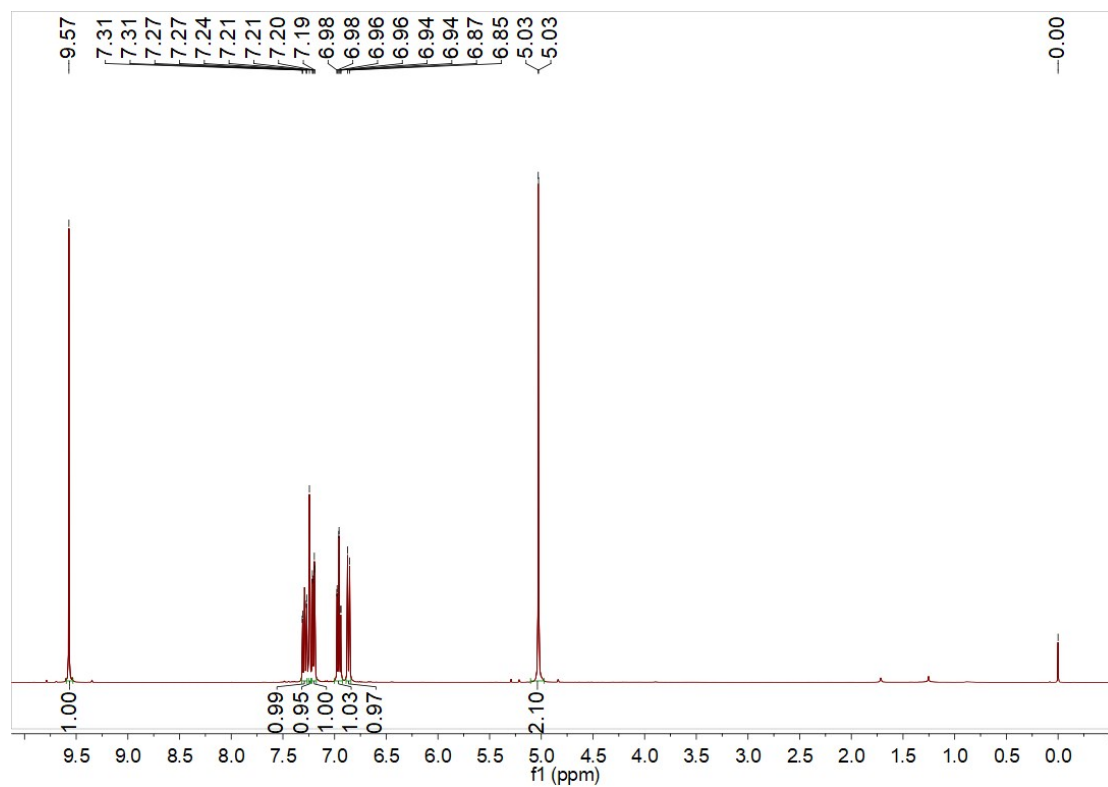


Fig S15a. ^1H NMR spectra of CHa3 in CDCl_3

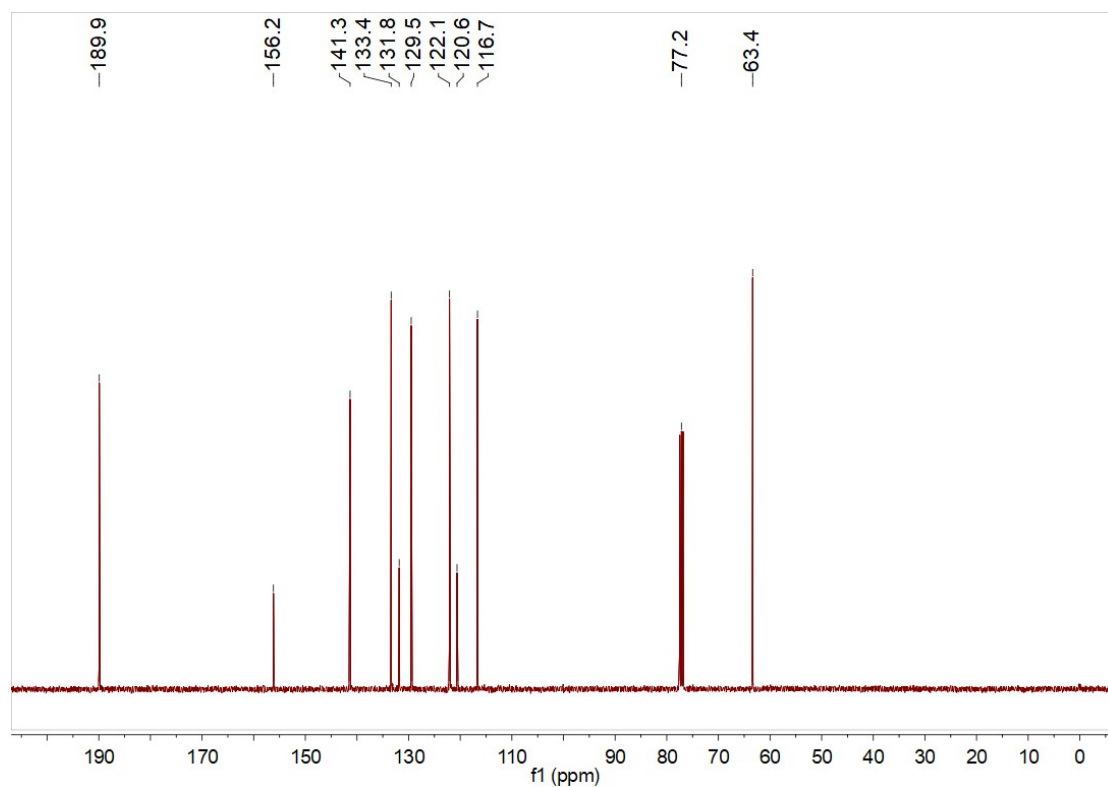


Fig S15b. ^{13}C NMR spectra of **CHa3** in CDCl_3 .

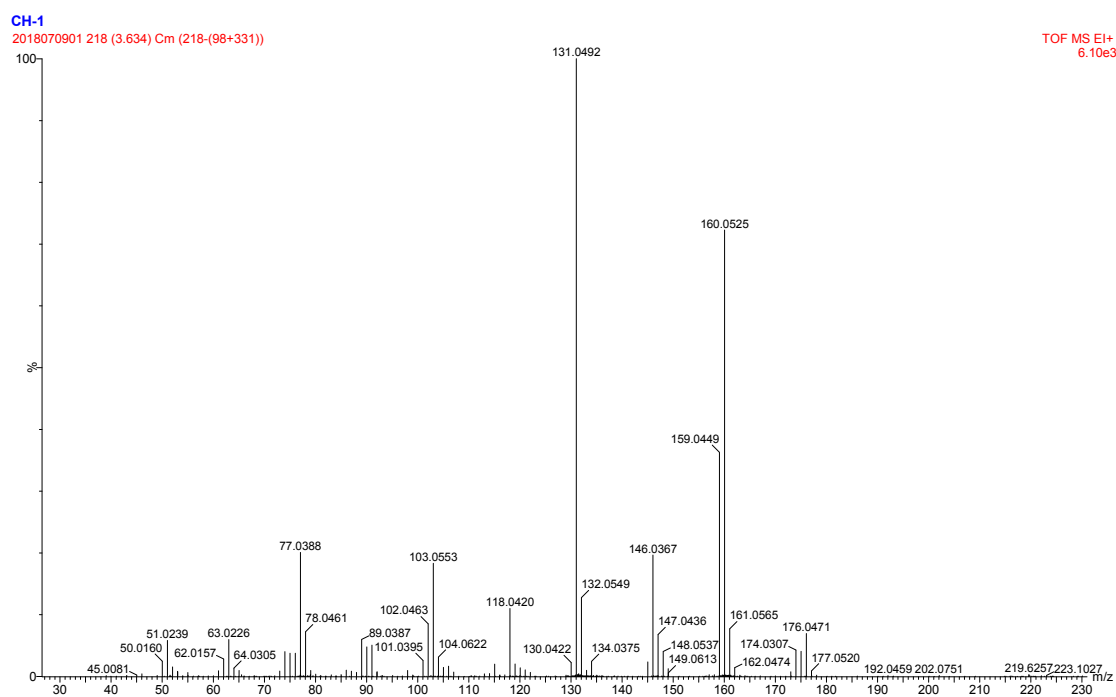


Fig S15c. Mass spectrum of **CHa3**.

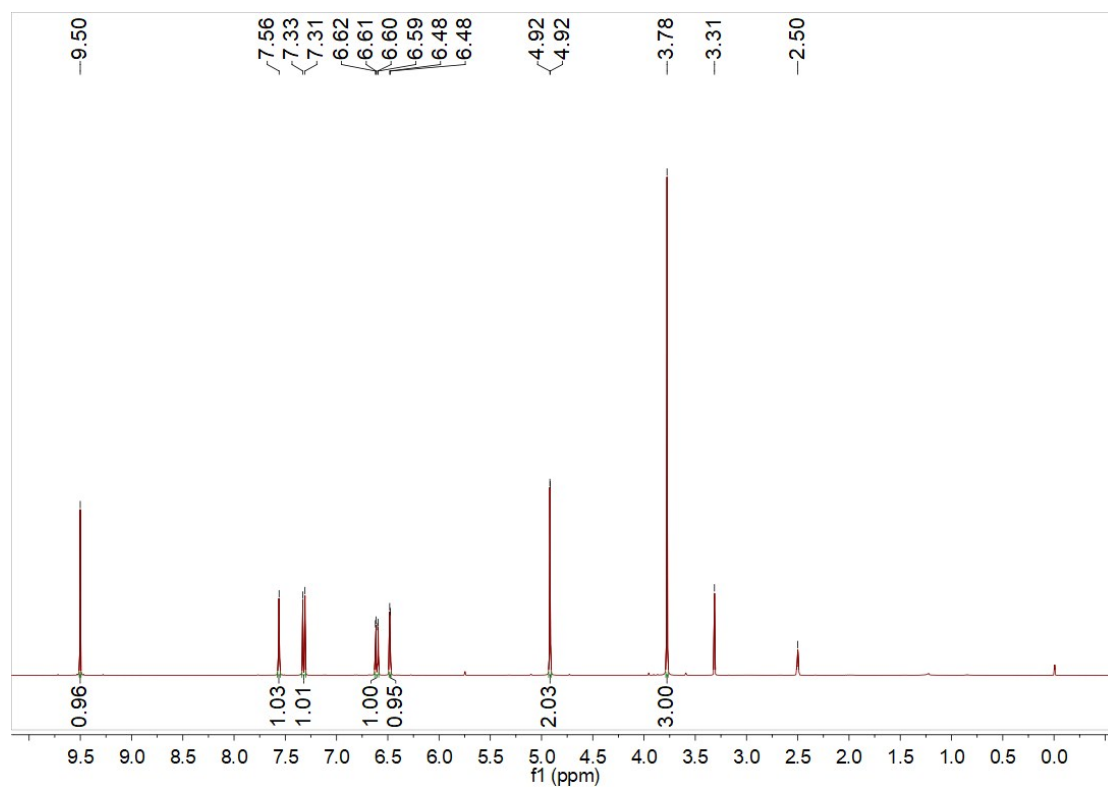


Fig S16a. ¹H NMR spectra of CHa4 in DMSO-*d*₆

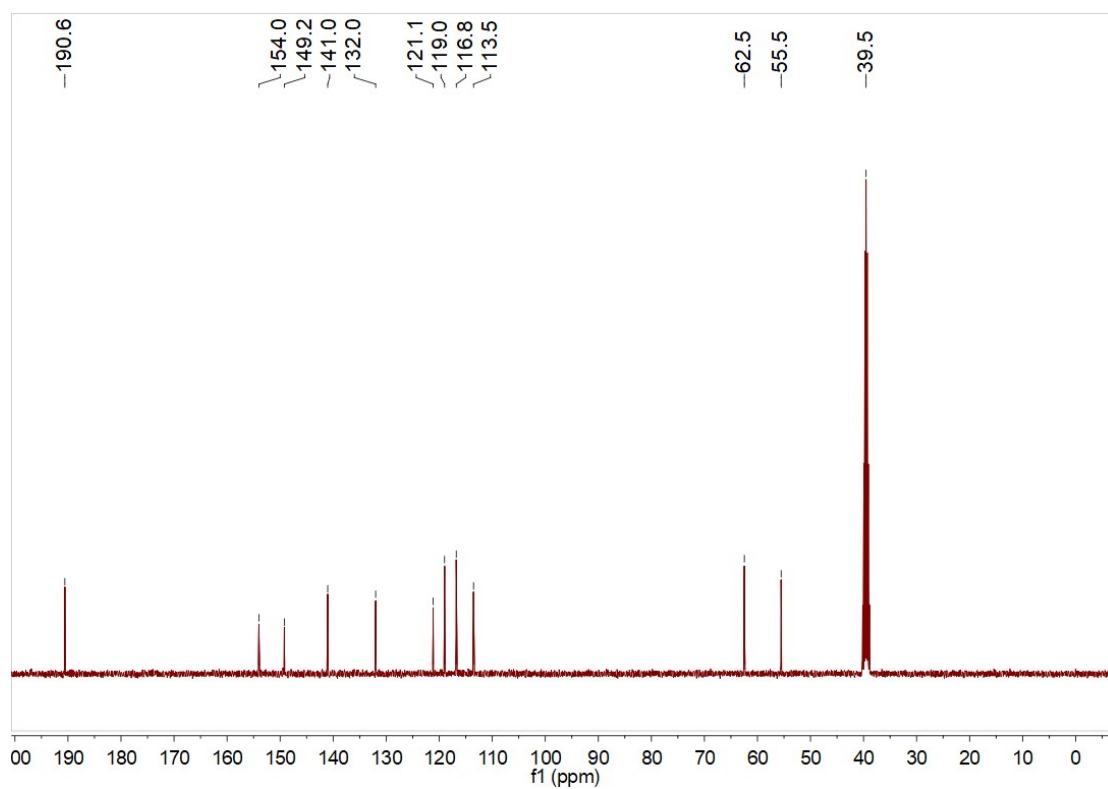


Fig S16b. ¹³C NMR spectra of CHa4 in DMSO-*d*₆.

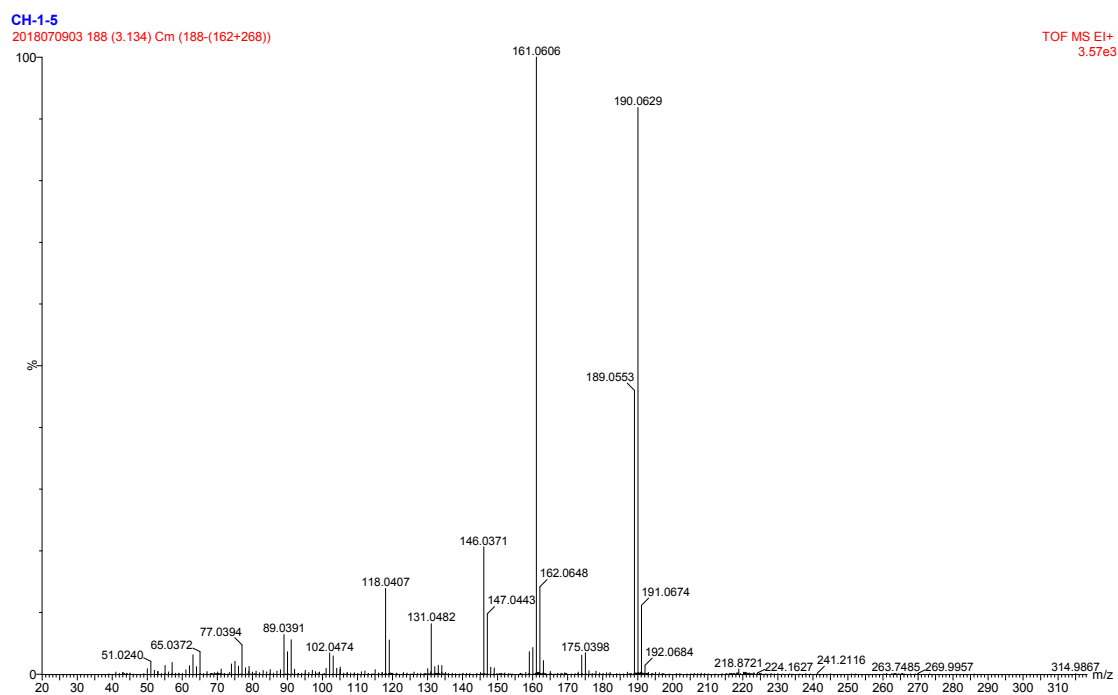


Fig S16c. Mass spectrum of CHa4.

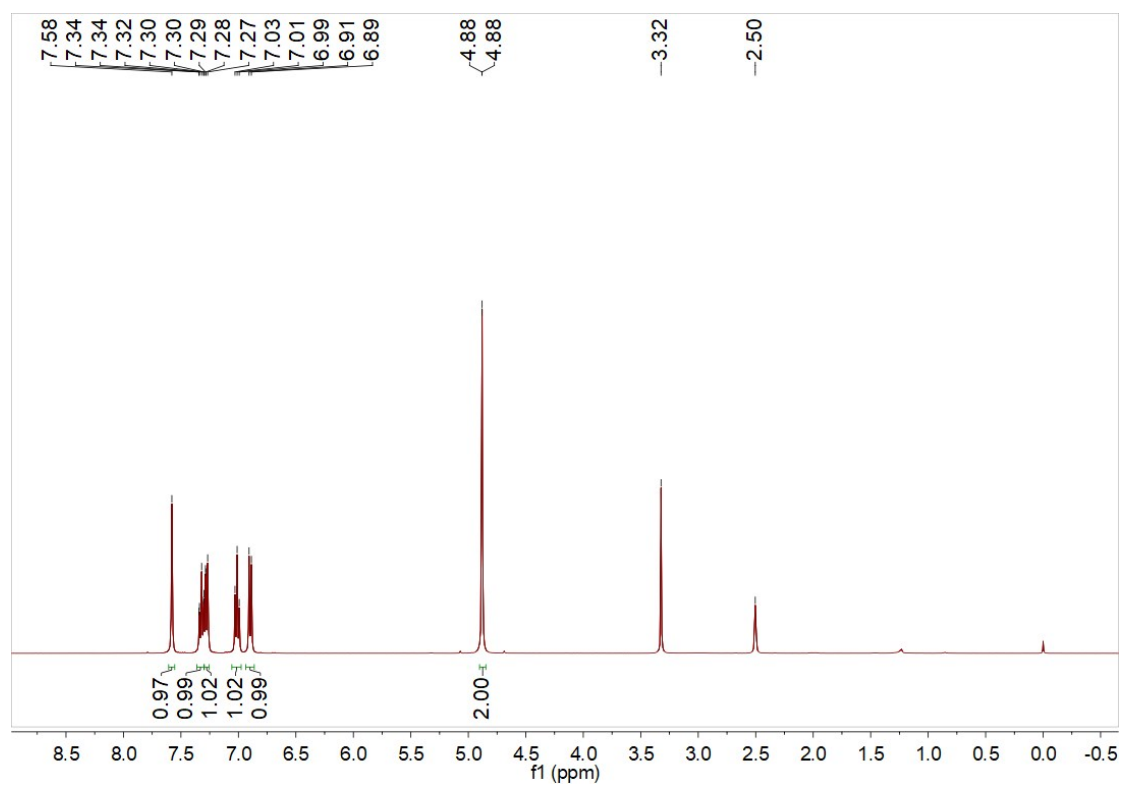


Fig S17a. ¹H NMR spectra of CHa5 in DMSO-*d*₆.

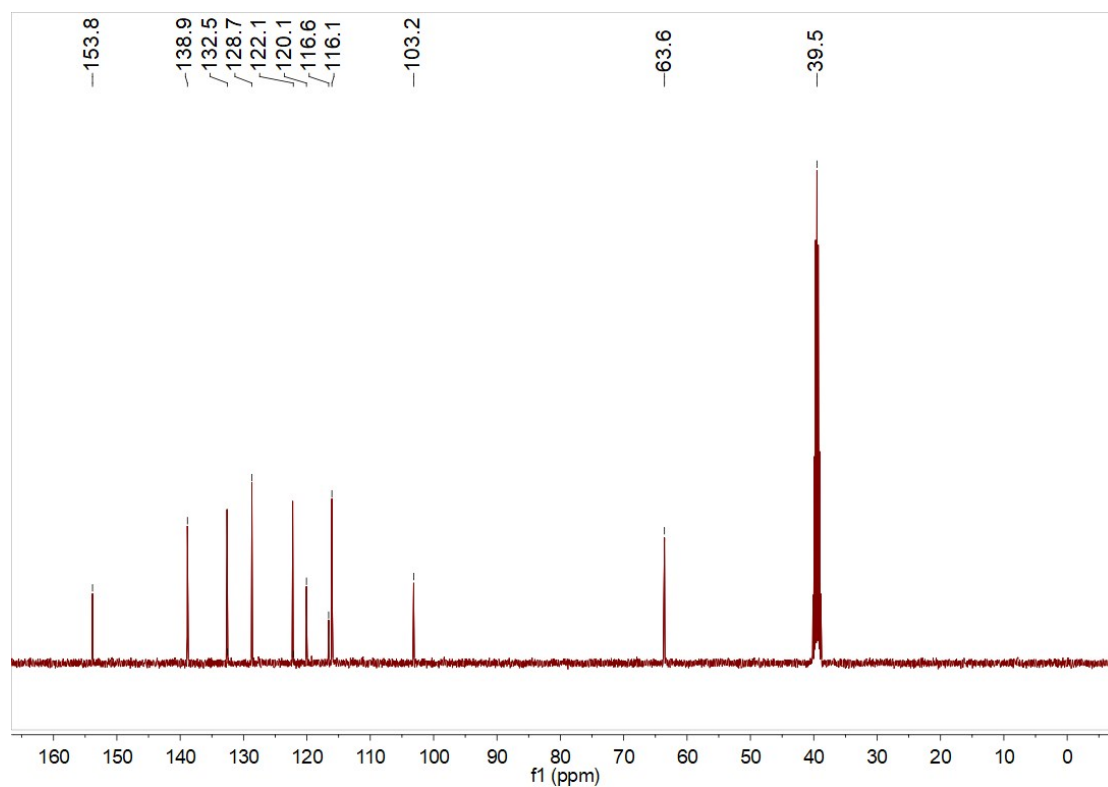


Fig S17b. ^{13}C NMR spectra of CHa5 in $\text{DMSO}-d_6$.

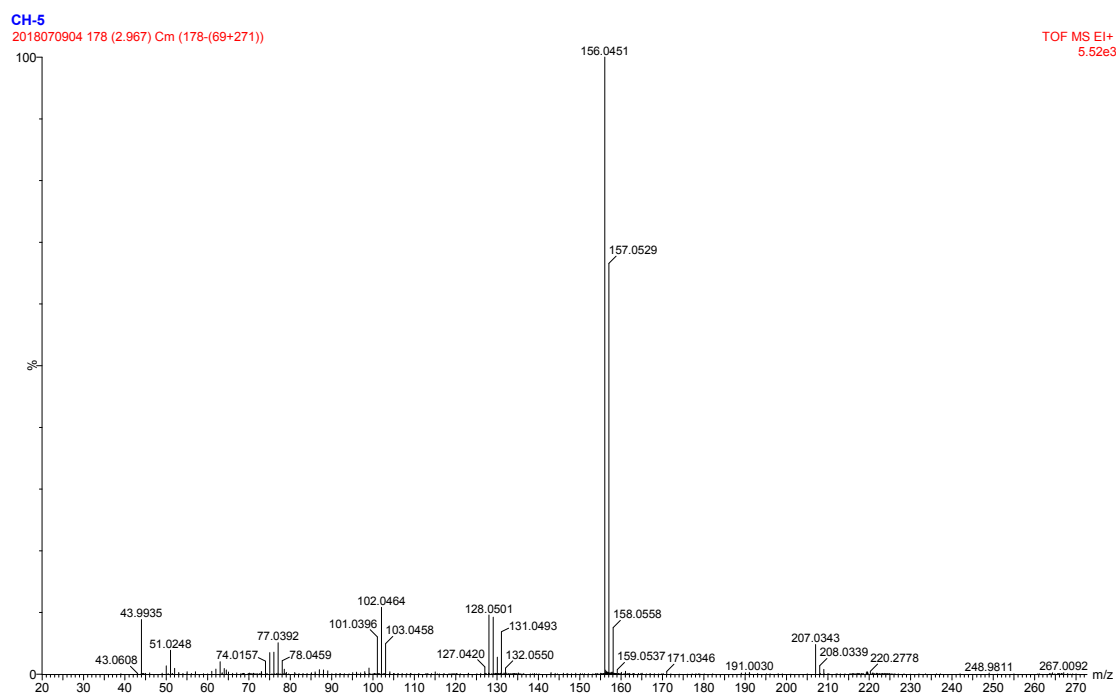


Fig S17c. Mass spectra of CHa5.

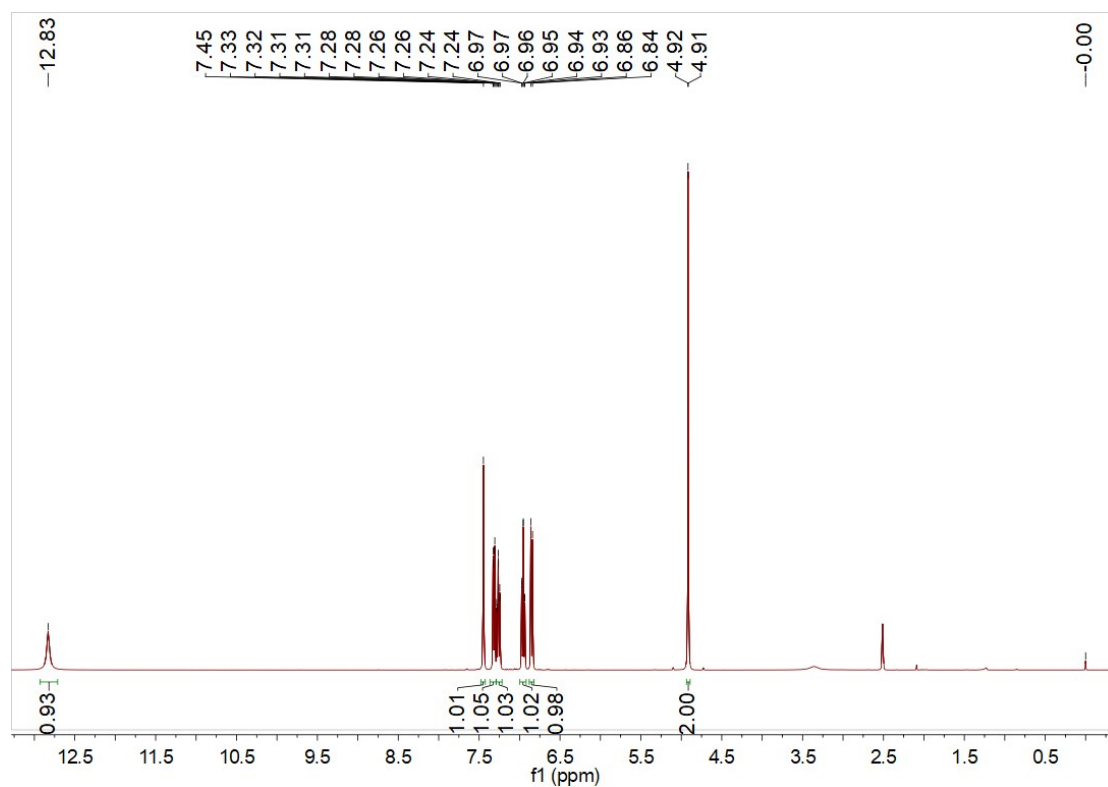


Fig S18a. ^1H NMR spectra of **CHa6** in $\text{DMSO}-d_6$.

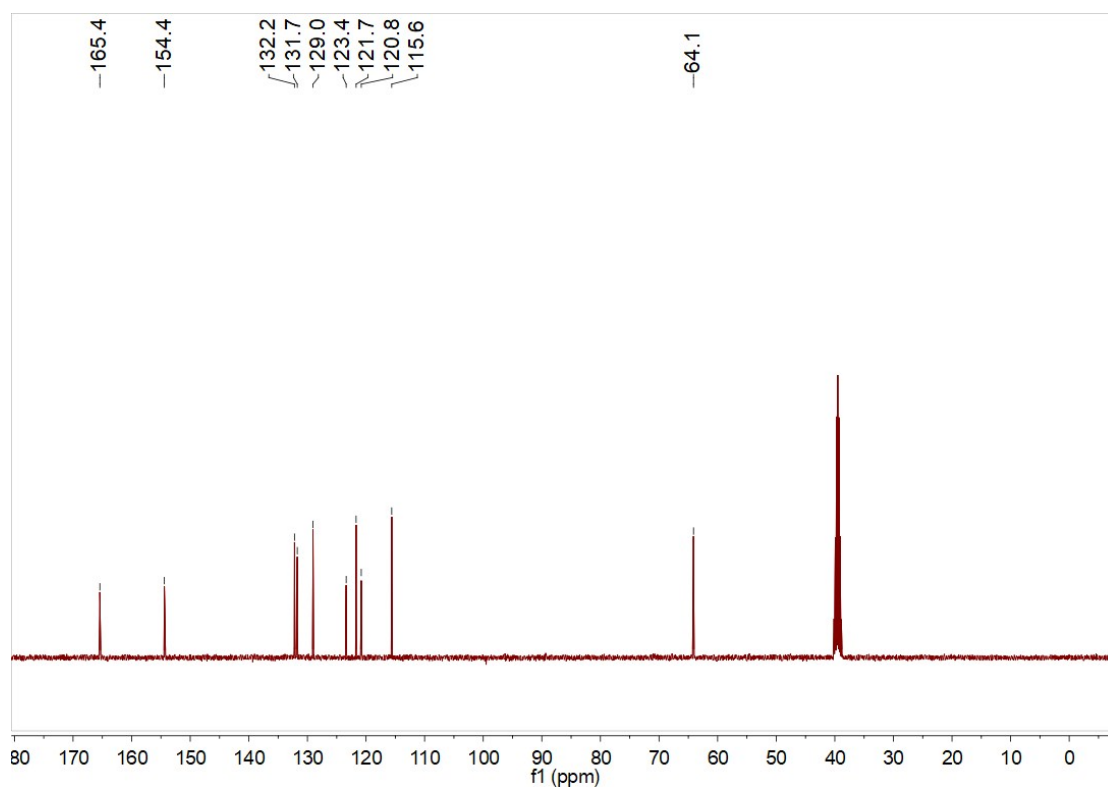


Fig S18b. ^{13}C NMR spectra of **CHa6** in $\text{DMSO}-d_6$.

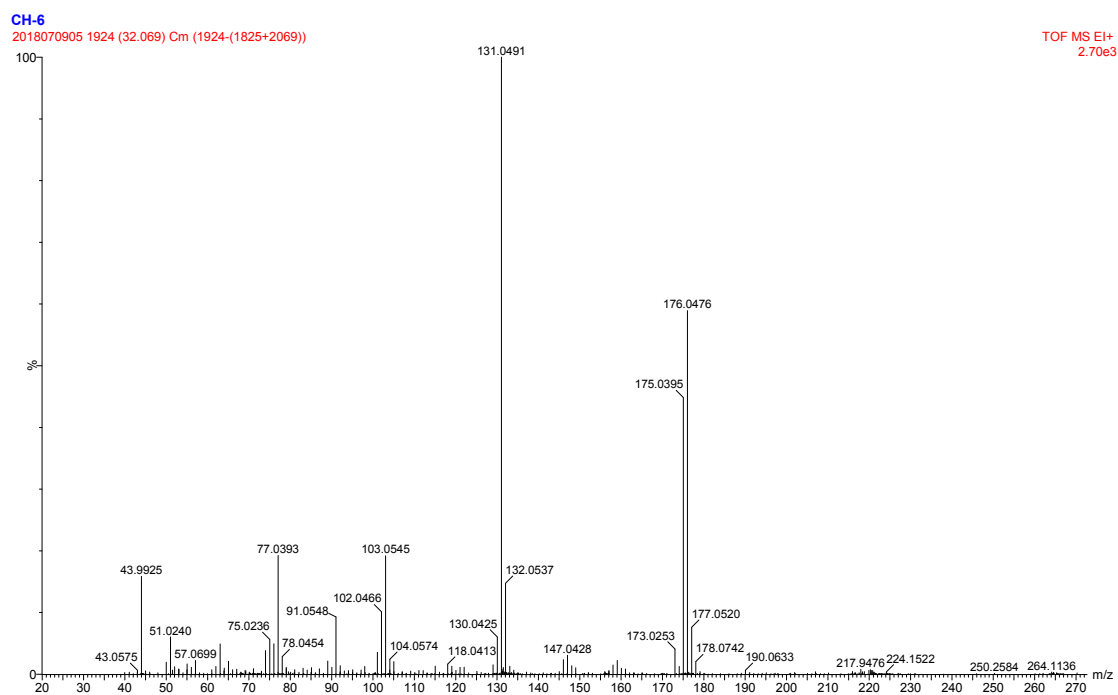


Fig S18c. Mass spectrum of CHa6.

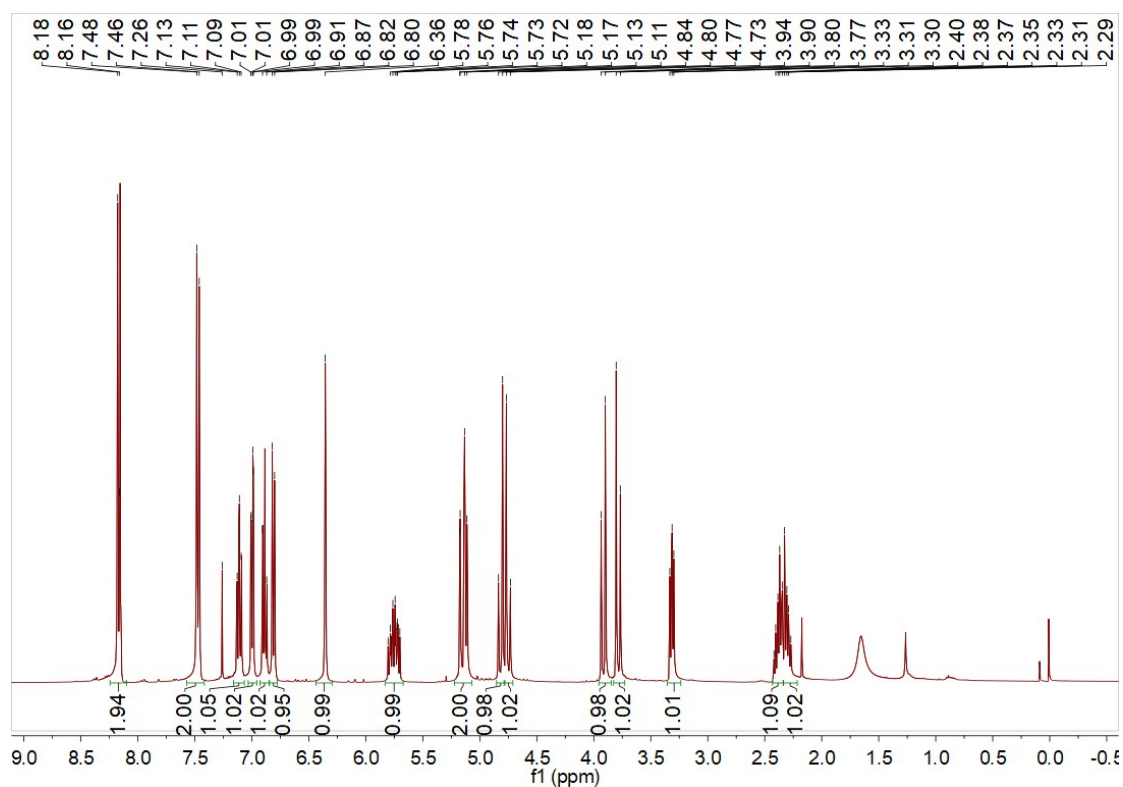


Fig S19a. ¹H NMR spectra of CHa6 in CDCl₃

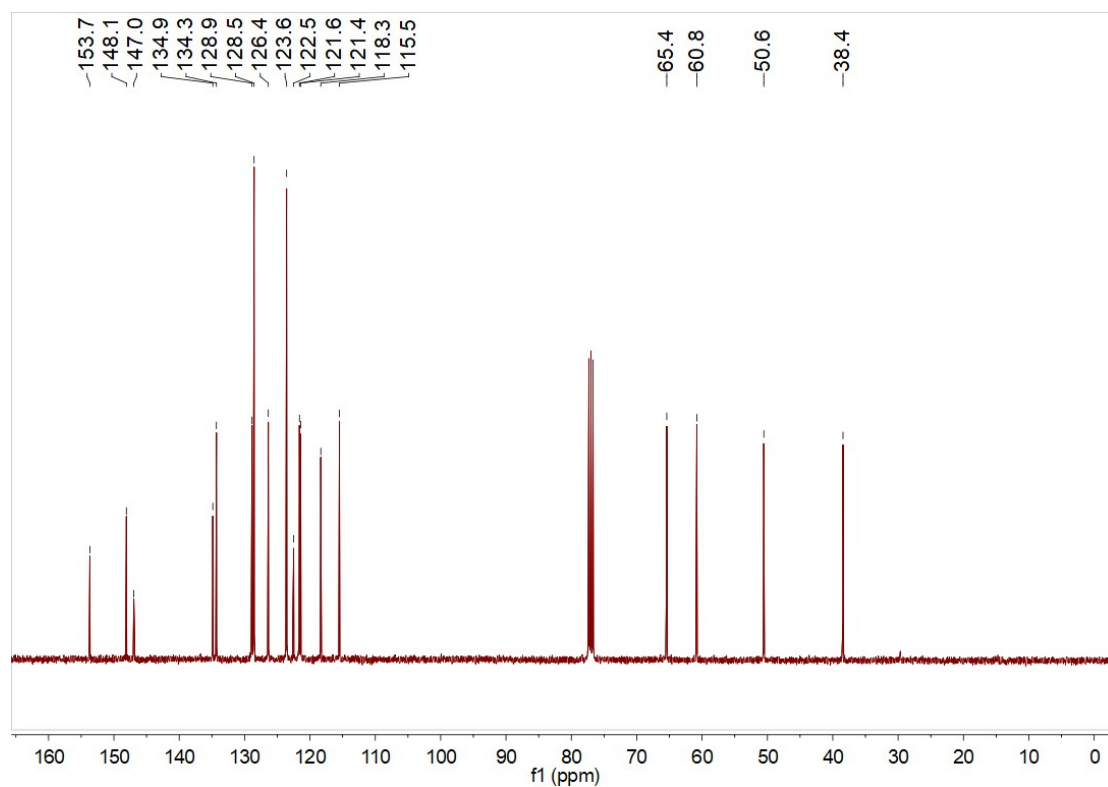


Fig S19b. ^{13}C NMR spectra of CHFA in CDCl_3 .

CHFA #159 RT: 0.87 AV: 1 NL: 6.27E6
T: FTMS - p ESI Full ms [50.0000-750.0000]

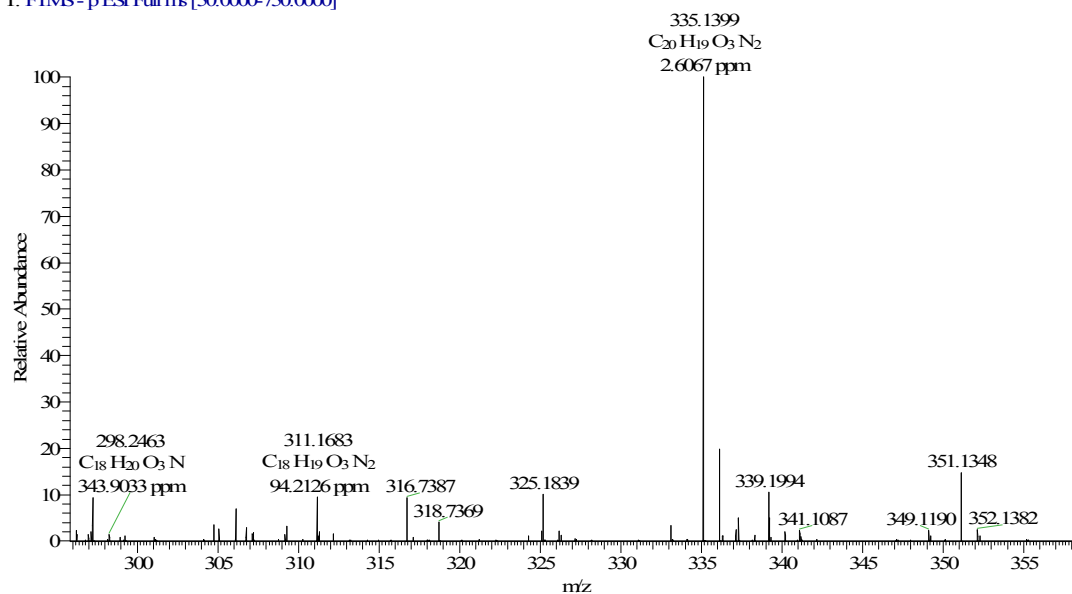


Fig S19c. Mass spectrum of CHFA.