Electronic Supplementary Material (ESI) for Analytical Methods. This journal is © The Royal Society of Chemistry 2019

Supporting Information for:

Simultaneous determination of multiclass antibiotics in sewage sludge based on QuEChERS extraction and liquid chromatography-tandem mass spectrometry

Akinranti S. Ajibola ^{a,b}, Selina Tisler^a, Christian Zwiener^{a*}

^a Eberhard Karls Universität Tübingen, Environmental Analytical Chemistry, Center for Applied Geoscience (ZAG), Hölderlinstr 12, 72074 Tübingen, Germany.

^b Department of Chemistry, Faculty of Science, University of Ibadan, Ibadan, Nigeria.

*Corresponding author

E-mail address: christian.zwiener@uni-tuebingen.de (C. Zwiener)

CONTENTS

- Table S1: Name, CAS number and some relevant physicochemical properties of the target antibiotics
- Discussion on preliminary assessment of extraction solvents and buffers
- Fig. S1: Preliminary assessment of extraction solvents and buffers
- Discussion on influence of methanol and Na2EDTA on the extraction recovery
- Fig. S2: Influence of methanol and Na₂EDTA on the extraction recovery
- Fig. S3: Preliminary assessment of d-SPE sorbents efficiency
- Fig. S4: Recovery studies with different amounts of d-SPE sorbent(s)
- Fig. S5: MS/MS chromatograms of ofloxacin and ciprofloxacin in sample blank extract

Table S1: Name, CAS number and some relevant physicochemical properties of the target antibiotics

	Compound	CAS No	Molecular Formular	Molecular Weight	Log K _{ow} ^a	Log K _d ^b	pK _a ^c
Fluoroquinolones	Ciprofloxacin	85721-33-1	C ₁₇ H ₁₈ FN ₃ O ₃	331.34	-0.28	4.3	5.8, 8.6
	Ofloxacin	82419-36-1	C ₁₈ H ₂₀ FN ₃ O ₄	361.37	-0.39	4.2	5.5, 6.2
	Norfloxacin	70458-96-7	C ₁₆ H ₁₈ FN ₃ O ₃	319.33	-1.03	4.2	5.8, 8.7
	Sulfamethoxazole	723-46-6	C ₁₀ H ₁₁ N ₃ O ₃ S	253.28	0.89	2.1 - 2.6	2.0, 7.6
Sulfonamides	Sulfadimethoxine	122-11-2	$C_{12}H_{14}N_4O_4S$	310.33	1.63	-	2.0, 6.9
	Sulfamethazine	57-68-1	$C_{12}H_{14}N_4O_2S$	278.33	0.19	-	2.0, 7.0
	Sulfadoxine	2447-57-6	C ₁₂ H ₁₄ N ₄ O ₄ S	310.33	0.70	-	2.3, 6.1
Tetracyclines	Oxytetracycline	79-57-2	C ₂₂ H ₂₄ N ₂ O ₉	460.43	-0.90	-	3.27, 4.6
	Tetracycline	60-54-8	$C_{22}H_{24}N_2O_8$	444.43	-1.30	3.9	3.3, 4.6
Macrolides	Azithromycin	83905-01-5	$C_{38}H_{72}N_2O_{12}$	748.98	4.02	2.5 -2.7	8.7
	Erythromycin	114-07-8	C ₃₇ H ₆₇ NO ₁₃	733.93	3.06	2.2	8.9
	Clarithromycin	81103-11-9	C ₃₈ H ₆₉ NO ₁₃	747.95	3.16	2.5-2.6	9.0
Bacteriostatic	Trimethoprim	738-70-5	C ₁₄ H ₁₈ N ₄ O ₃	290.32	0.91	-	7.1
Amphenicol	Thiamphenicol	15318-45-3	C ₁₂ H ₁₅ Cl ₂ NO ₅ S	356.22	-0.27	-	7.7

	Amoxicillin	26787-78-0	$C_{16}H_{19}N_3O_5S$	365.40	0.87	-	2.4, 7.4
Penicillins							
(β-Lactams)	Flucloxacillin-Na	1847-24-1	C ₁₉ H ₁₆ ClFN ₃ NaO ₅ S	475.85	-1.44	-1	-
	Penicillin G-Na	6957-8	C ₁₆ H ₁₇ N ₂ NaO ₄ S	356.37	1.83	-	-

a-KOWWIN v 1.67 (EPI Suite, USEPA), b- ref. 1, c - ref. 2, (-) - data not available

Preliminary assessment of extraction solvents and buffers

From the preliminary experiments almost all the extraction solvents in combination with a buffer or non-buffered salt provided good recoveries for trimethoprim, thiamphenicol and the four sulfonamides with the exception of methanol/acetonitrile/0.1M Na₂EDTA-McIlvaine buffer (2 mL:8 mL:10 mL) + non-buffered salt with which a reduction in recovery was observed (J and K in Fig. S1). Modification of organic solvent (acetonitrile) with 2 mL methanol slightly reduced their recoveries using 0.1 M Na₂EDTA-McIlvaine buffer. Though comparable extraction recoveries were observed for sulfonamide antibiotics with all the extraction solvents and buffer combination, 0.1 M Na₂EDTA-McIlvaine buffer generally presented much lower recoveries. Modification of organic solvent (acetonitrile) with 1% acetic acid however, slightly improved the recoveries of sulfonamides using the 0.1 M Na₂EDTA-McIlvaine buffer. Citrate and acetate buffers provided highest recoveries for sulfonamides.

For the fluoroquinone norfloxacin, extraction solvents involving 0.1 M Na₂EDTA McIlvaine buffer yielded good recoveries (Fig. S1, experiments G to K). The presence of Na₂EDTA and ultrasonication obviously improved the recovery of norfloxacin. Na₂EDTA was found to improve the extraction recovery of fluoroquinolones from sewage sludge and other related environmental matrices in previous studies.^{2,3} (Gago-Ferrero et al 2015, Huang et al. 2013). Generally low recoveries were obtained for norfloxacin with acetate buffer. Peysson and Vulliet also found low recoveries for fluoroquinolones with acetate buffer in their work.⁴ However, the authors discontinued to test further the possibility of obtaining better recoveries for fluoroquinolones with citrate buffer. Based on our preliminary studies, citrate buffer proved to be a prospective better candidate for the extraction of fluoroquinolones than acetate buffer (experiments B and E in Fig. S1).

Extraction with acetonitrile/water only using any of the buffer or non-buffered salts yielded good recoveries for the macrolides (clarithromycin, azithromycin and erythromycin). Overall, citrate buffer provided best recoveries for azithromycin and clarithromycin (exp. B), for erythromycin it was non-buffered salt (exp. C). Moreover, citrate buffer, non-buffered salt and 0.1 M Na₂EDTA-McIlvaine buffer gave good recoveries for tetracycline. Modification of acetonitrile with methanol when using 0.1 M Na₂EDTA-McIlvaine buffer apparently improved tetracycline recovery (experiment K). Modification of extraction solvents with Na₂EDTA when using citrate buffer and its addition to McIlvaine buffer also led to improvement of tetracycline recovery. Bourdat-Deschamps and co-workers also added EDTA to extraction solvents in order to improve the recovery of tetracyclines.⁵ It is worthy to note that β-lactam

antibiotics (amoxicillin, penicillin and flucloxacillin) were not recovered by any of the combination of extraction solvents and buffers. Further attempts were made in subsequent experiments in order to improve the recoveries of β -lactam and other target antibiotics. Therefore, based on the results of these preliminary studies, extraction solvent composition (acetonitrile/methanol and Na₂EDTA in water) with citrate buffer and ultrasonication were selected for further experiments.



Description of extraction parameters for the preliminary experiments (Fig. S1); Ciprofloxacin and Ofloxacin values were not included due to very high

enhancement resulting in corresponding very high recovery values

Experiment	Organic solvent	Aqueous solvent	Buffer salt	Ultrasonication
Α	10 mL acetonitrile	10 mL H ₂ O	acetate	Х
В	10 mL acetonitrile	10 mL H ₂ O	citrate	Х
С	10 mL acetonitrile	10 mL H ₂ O	non-buffered	Х
L	10 mL acetonitrile	10 mL 0.1 M Na ₂ EDTA in H ₂ O	non-buffered	Х
D	10 mL acetonitrile modified with 1% acetic acid	10 mL 0.1 M Na ₂ EDTA in H ₂ O	acetate	Х
E	10 mL acetonitrile modified with 1% acetic acid	10 mL 0.1 M Na ₂ EDTA in H ₂ O	citrate	Х
F	10 mL acetonitrile modified with 1% acetic acid	10 mL 0.1 M Na ₂ EDTA in H ₂ O	non-buffered	Х
G	10 mL acetonitrile modified with 1% acetic acid	10 mL 0.1 M Na ₂ EDTA-Mcllvaine buffer	non-buffered	Х
Н	10 mL acetonitrile	10 mL 0.1 M Na ₂ EDTA-Mcllvaine buffer	non-buffered	Х
Ι	10 mL acetonitrile	10 mL 0.1 M Na ₂ EDTA-Mcllvaine buffer	non-buffered	ultrasonication
J	8 mL acetonitrile/2 mL methanol	10 mL 0.1 M Na ₂ EDTA-Mcllvaine buffer	non-buffered	Х
K	8 mL acetonitrile/2 mL methanol	10 mL 0.1 M Na ₂ EDTA-Mcllvaine buffer	non-buffered	ultrasonication

Fig. S1 Preliminary assessment of extraction solvents and buffers

Influence of methanol and Na2EDTA on the extraction recovery

Considering the tremendous influence of methanol and Na₂EDTA on the recoveries of most target antibiotics in the preliminary experiments (Fig. S1), some experiments were further carried out to investigate possible improvement in the recoveries by increasing the concentration of Na₂EDTA in the aqueous solution to 0.2M. Results are presented in Fig. S2. In comparison with 0.1M Na₂EDTA slightly better recoveries were obtained with 0.2M Na₂EDTA aqueous solution for all target antibiotics using 2mL methanol as organic modifier, except the β -lactam antibiotics which were not recovered. However, modification of organic solvent with 1mL methanol gave rise to a general decrease in the recoveries of sulfonamides, macrolides and tetracyclines thus indicating the necessity of optimizing methanol contents for the recoveries of most target antibiotics. Finally, taking into account the desirable influence of methanol and Na₂EDTA on the recoveries of most target antibiotics, 2mL methanol + 8 mL acetonitrile + 0.2M Na₂EDTA aqueous solution was chosen as the extraction solvents, employing citrate buffer salt for partitioning followed by ultrasonication.



Extraction involved the use of citrate buffer salt for each extraction solvent composition followed by ultrasonication.

Extraction solvent composition: 10 mL organic solvent + 10 mL water at varied composition of organic solvent, acetic acid and Na, EDTA;

VA: 8 mL ACN + 2 mL MeOH+10 mL 0.1M Na₂EDTA; VB: 8 mL ACN + 2 mL MeOH + 10 mL 0.2M Na₂EDTA; VC: 9 mL ACN + 1 mL MeOH + 10 mL 0.2M Na₂EDTA; VD: 9 mL ACN + 1 mL

MeOH + 10 mL 0.1M Na_EDTA; VE: 8 mL 1% acetic acid in ACN + 2 mL MeOH + 10 mL 0.1M Na_EDTA; UB: 6 mL ACN + 4 mL MeOH + 10 mL 0.1M Na_EDTA

Fig. S2 Influence of methanol and Na2EDTA on the extraction recovery



Extraction solvent composition: 9 mL ACN + 1 mL MeOH + 10 mL H_2O ; + citrate buffer followed by ultrasonication

Fig. S3 Preliminary assessment of d-SPE sorbents efficiency



Extraction solvent composition: 8 mL ACN + 2 mL MeOH + 10 mL 0.1M Na₂EDTA; + citrate buffer followed by ultrasonication. Each d-SPE sorbent composition contained 200 mg MgSO₄

Fig. S4 Recovery studies with different amounts of d-SPE sorbent(s)



Fig. S5 MS/MS chromatograms of ofloxacin and ciprofloxacin in sample blank: high values of peak areas depicting the presence of the two antibiotics in significant amounts

References

÷

- P. Verlicchi, M. Al Aukidy, A. Jelic, M. Petrović and D. Barceló, *Sci. Total Environ.*, 2014, 470, 844–854. https://dx.doi.org/10.1016/j.scitotenv.2013.10.026.
- P. Gago-Ferrero, V. Borova, M. E. Dasenaki and N. S. Thomaidis, *Anal. Bioanal. Chem.*, 2015, 407, 4287- 4297. https://doi.org/10.1007/s00216-015-8540-6.
- Y. J. Huang, M. M. Cheng, W. H. Li, L. H. Wu, Y. S. Chen, Y. M. Luo, P. Christie and H. B. Zhang, *Anal. Methods*, 2013, 5, 3721–3731. <u>https://doi.org/10.1039/c3ay40220g.</u>
- 4. W. Peysson and E. Vulliet, J. Chromatogr A., 2013, 1290, 46-61. http://dx.doi.org/10.1016/j.chroma.2013.03.057.
- 5. M. Bourdat-Deschamps, S. Leang, N. Bernet, J. Daudin, S. Nelieu, *J. Chromatogr A*, 2014, **1349**, 11-23. <u>http://dx.doi.org/10.1016/j.chroma.2014.05.006</u>.