

## ***Supporting Information***

### ***Two-Step Template Method for Synthesis of Axis-Length- Controlled Porphyrin-Containing Hollow Structures***

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## 1. General Comments

### 1.1 Materials

Solvents were purified as follows: reaction solvents were degassed through freeze-pump-thaw (three times), before use. Dry toluene (water content : < 10 ppm), THF (water content : < 10 ppm), and DMF (water content : < 50 ppm) were purchased from Kanto Chemical and further purified by passage through activated alumina under positive argon or nitrogen pressure as described by Grubbs *et al.*<sup>1</sup> All other reagents and solvents were purchased from commercial sources (FUJIFILM Wako Pure Chemical Corporation or Tokyo Chemical Industry Co., Ltd., Grade : Extra Pure) and used without further purification. A fluorene derivative **S1**,<sup>2</sup> porphyrin derivative **S4**,<sup>3</sup> **T1**<sup>4</sup> and **S10**,<sup>5</sup> were prepared by the reported procedures. **ZnP<sub>2</sub>** was synthesized according to previous methods.<sup>6</sup>

### 1.2 Experimental

#### NMR Spectroscopy

<sup>1</sup>H NMR spectra were measured on a Bruker AVANCE-500 spectrometer. The <sup>1</sup>H NMR chemical shifts are reported relative to tetramethylsilane (TMS, 0.00 ppm) or residual protonated solvents (7.26 ppm) in CDCl<sub>3</sub>, or (5.32 ppm) in CD<sub>2</sub>Cl<sub>2</sub>. The <sup>13</sup>C NMR chemical shifts are reported relative to tetramethylsilane (TMS, 0.00 ppm), residual protonated <sup>13</sup>CDCl<sub>3</sub> (77.16 ppm), or <sup>13</sup>CD<sub>2</sub>Cl<sub>2</sub> (53.84 ppm).

#### Preparative recycling gel permeation chromatography (GPC)

Preparative recycling GPC was performed with a JAI LC9104 System equipped with JAIGEL-1H and 2H columns, a JAI UV DETECTOR 310, and a JAI RI DETECTOR RI-5, a JAI LC9130NEXT System equipped with JAIGEL-2H and 2.5H columns, a JAI UV DETECTOR 370NEXT, and a JAI RI DETECTOR RI 700NEXT, a JAI LC9140 System equipped with JAIGEL-2.5H and -3H columns, a JAI UV DETECTOR 310, and a JAI RI DETECTOR RI-5.

#### Mass Spectroscopy (MS)

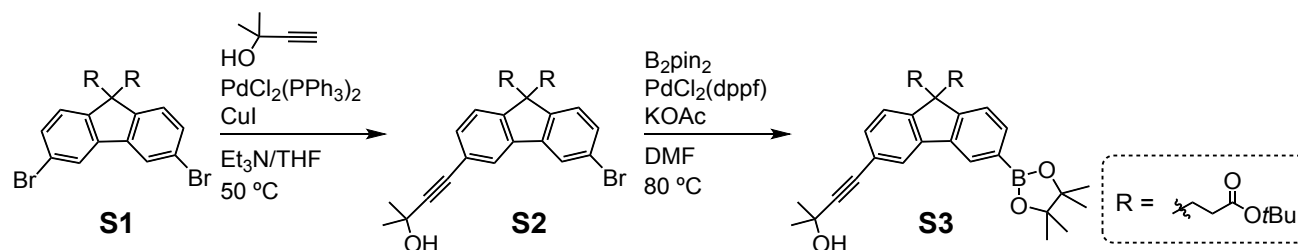
Matrix assisted laser desorption/ionization time-of-flight (MALDI-TOF) mass spectra were obtained with  $\alpha$ -cyano-4-hydroxycinnamic acid (CHCA) and *trans*-2-[3-(4-*t*-butyl-phenyl)-2-methyl-2-propenylidene]malononitrile (DCTB) as a matrix on Thermo Fisher Scientific LTQ orbitrapXL. ESI mass spectra were obtained on Thermo Fisher Scientific EXACTIVE and SHIMADZU Axima Performance.

#### Absorption spectroscopy

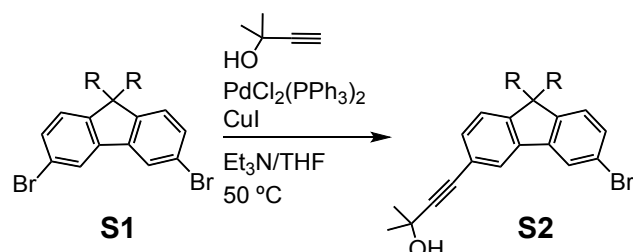
Absorption spectra were measured with SHIMADZU UV-2600 spectrophotometer. For solution measurements, concentrations of 10<sup>-6</sup> M solutions were used.

## 2. Synthetic Procedures

### Scheme S1. Synthesis of fluorene S3 from S1



#### 2.1 Synthesis of S2



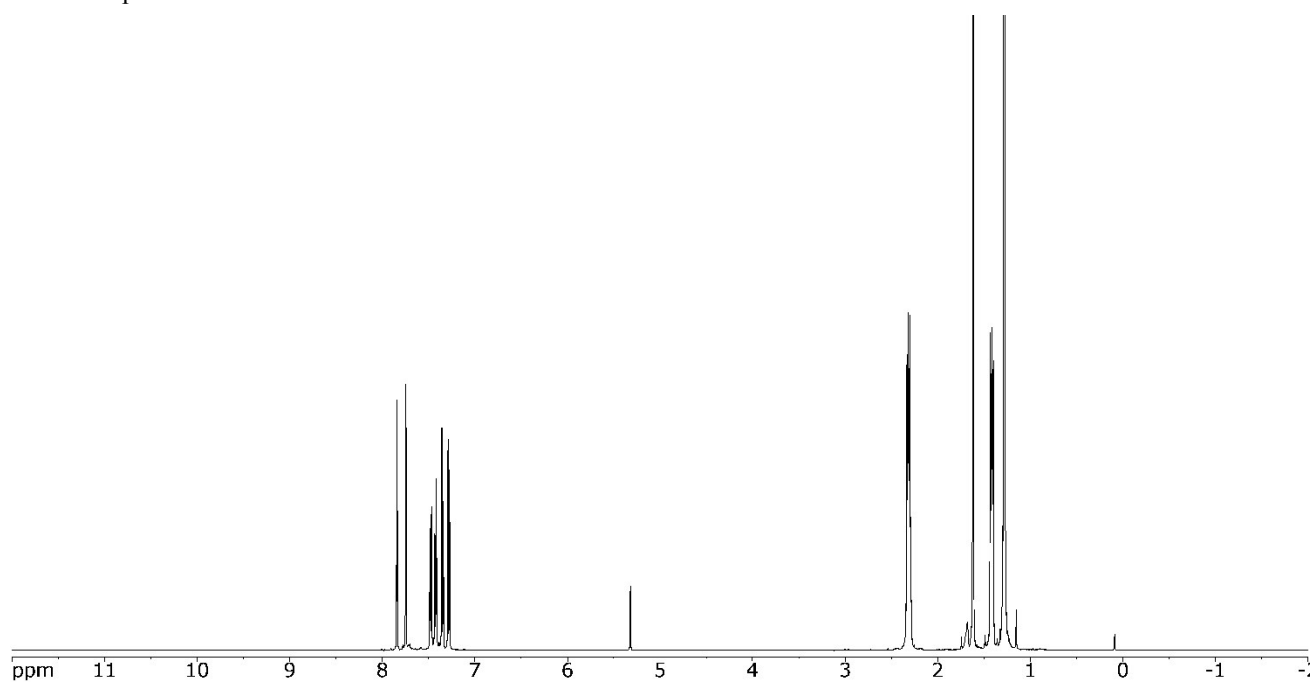
**S1** (4.69 g, 8.08 mmol),  $\text{PdCl}_2(\text{PPh}_3)_2$  (283 mg, 0.404 mmol), and  $\text{CuI}$  (38.5 mg, 0.202 mmol) were dissolved in degassed THF /  $\text{Et}_3\text{N}$  (4 / 1, v / v, 330 mL) under Ar. 2-Methyl-3-butyn-2-ol (0.706 mL, 7.27 mmol) was added. The solution was stirred at  $50^\circ\text{C}$  overnight, and passed through the Celite pad. The solvent was evaporated, after which the residue was purified by silica gel column chromatography ( $\text{EtOAc}$  / hexane = 1 / 4, v / v) to yield **S2** (1.62 g, 34%) as a yellow solid. Unreacted **S1** (2.50 g, 4.31 mmol) was recovered as a pure solid during the above purification process.

*HR-MS (ESI-MS)* ( $m/z$ ) 605.1873 ( $[\text{S1} + \text{Na}]^+$ ,  $\text{C}_{32}\text{H}_{39}\text{BrNaO}_5$ , calcd. 605.1879)

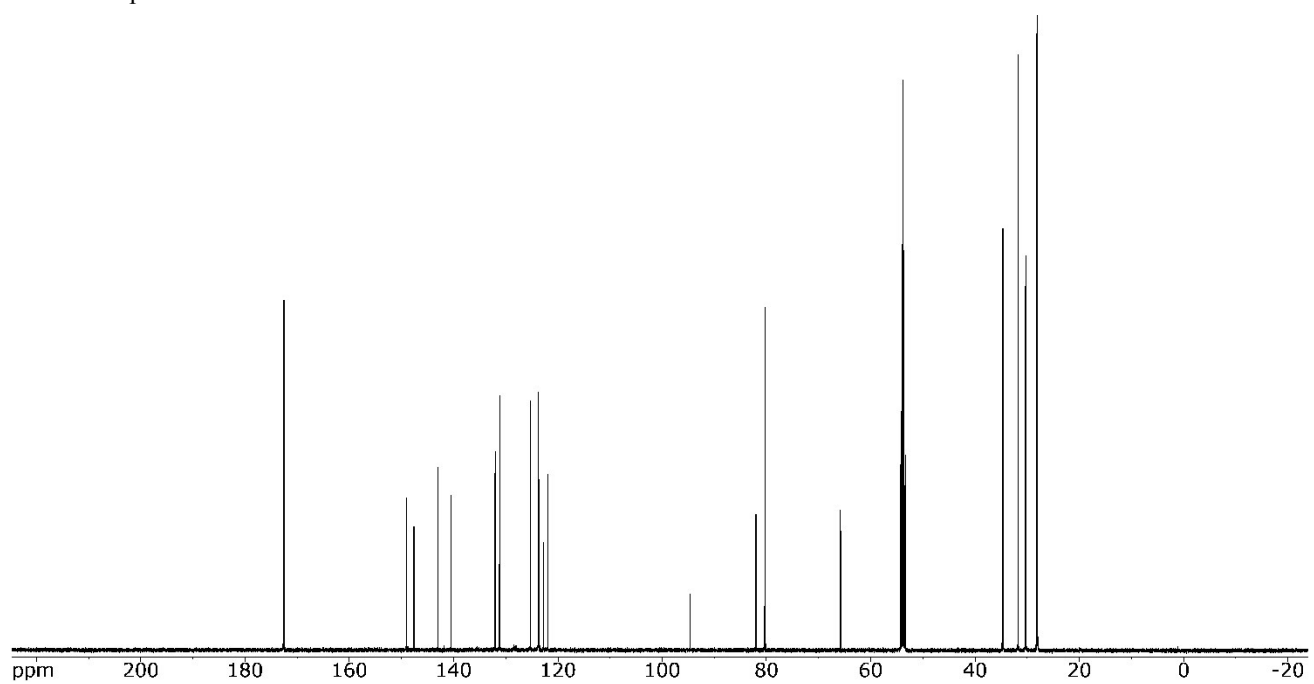
$^1\text{H NMR}$  (500 MHz,  $\text{CD}_2\text{Cl}_2$ , *r. t.*):  $\delta$  7.84 (d,  $J = 1.5$  Hz, 1H, Ar-*H*), 7.74 (s, 1H, Ar-*H*), 7.48 (dd,  $J = 8.0, 1.6$  Hz, 1H, Ar-*H*), 7.42 (dd,  $J = 7.8, 1.1$  Hz, 1H, Ar-*H*), 7.35 (d,  $J = 7.8$  Hz, 1H, Ar-*H*), 7.28 (d,  $J = 8.0$  Hz, 1H, Ar-*H*), 2.32 (t,  $J = 8.3$  Hz, 4H,  $-\text{CH}_2\text{CH}_2\text{COO}^t\text{Bu}$ ), 1.62 (s, 6H,  $-\text{C}(\text{CH}_3)_2\text{OH}$ ), 1.42 (t,  $J = 8.3$  Hz, 4H,  $-\text{CH}_2\text{CH}_2\text{COO}^t\text{Bu}$ ), 1.28 (s, 18H,  $-\text{COOC}(\text{CH}_3)_3$ ).

$^{13}\text{C NMR}$  (126 MHz,  $\text{CD}_2\text{Cl}_2$ , *r. t.*):  $\delta$  172.5, 149.1, 147.6, 143.0, 140.5, 132.0, 131.2, 125.3, 123.8, 123.69, 123.66, 122.8, 121.9, 94.7, 82.0, 80.3, 65.8, 54.12, 34.7, 31.8, 30.3, 28.1.

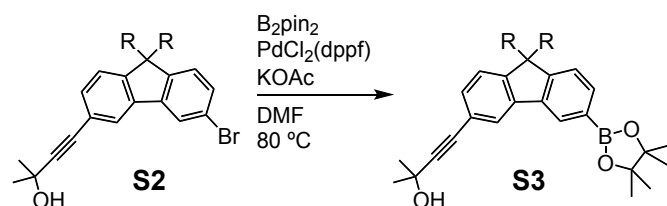
$^1\text{H}$  NMR spectrum of **S2**



$^{13}\text{C}$  NMR spectrum of **S2**



## 2.2 Synthesis of S3



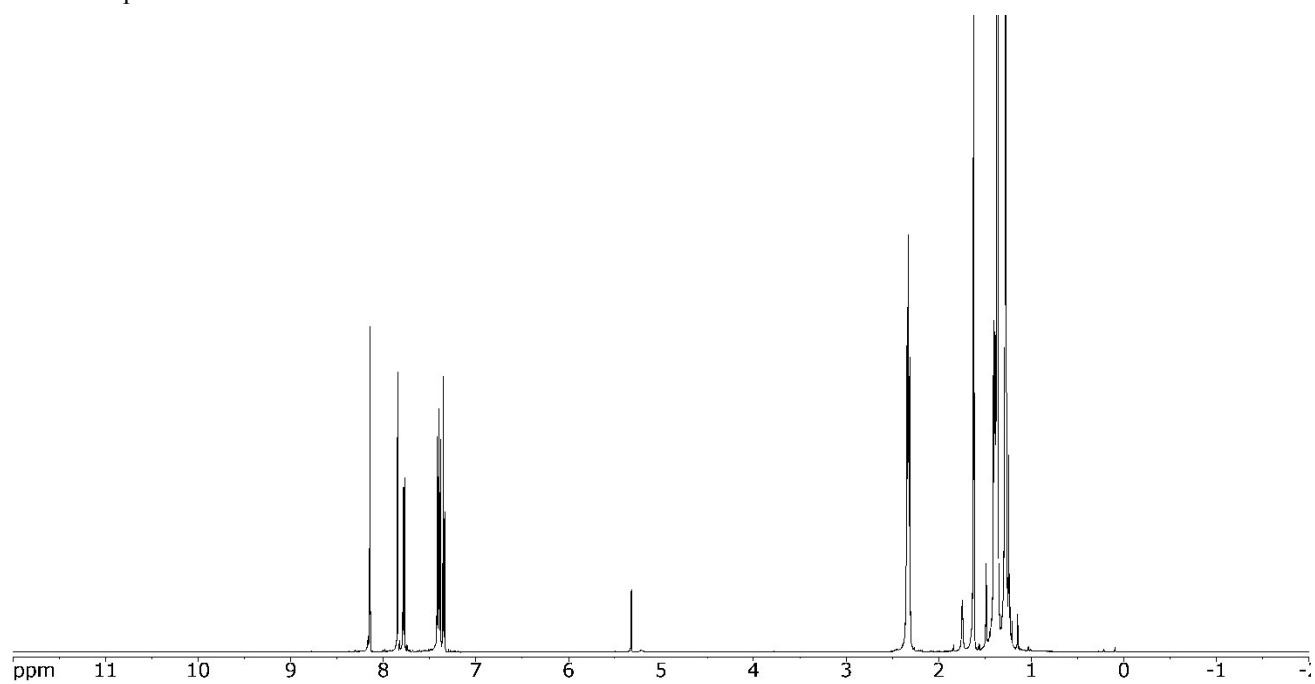
**S2** (1.34 g, 2.31 mmol), bis(pinacolato)diboron (0.706 g, 2.78 mmol),  $PdCl_2(dppf)$  (188 mg, 0.231 mmol), and  $KOAc$  (302 mg, 3.08 mmol) were dissolved in dry  $DMF$  (54.0 mL) under Ar. The solution was stirred at  $80\text{ }^{\circ}C$  overnight, after which the solvent was removed under vacuum. The residue was purified by silica gel column chromatography ( $EtOAc$  / hexane = 1 / 4, v / v) and GPC with  $CHCl_3$  as an eluent to yield **S3** (1.01 g, 69%) as a white solid.

*HR-MS (ESI-MS)* ( $m/z$ ) 653.3621 ( $[S3+Na]^+$ ,  $C_{38}H_{51}BNaO_7$ , calcd. 653.3626)

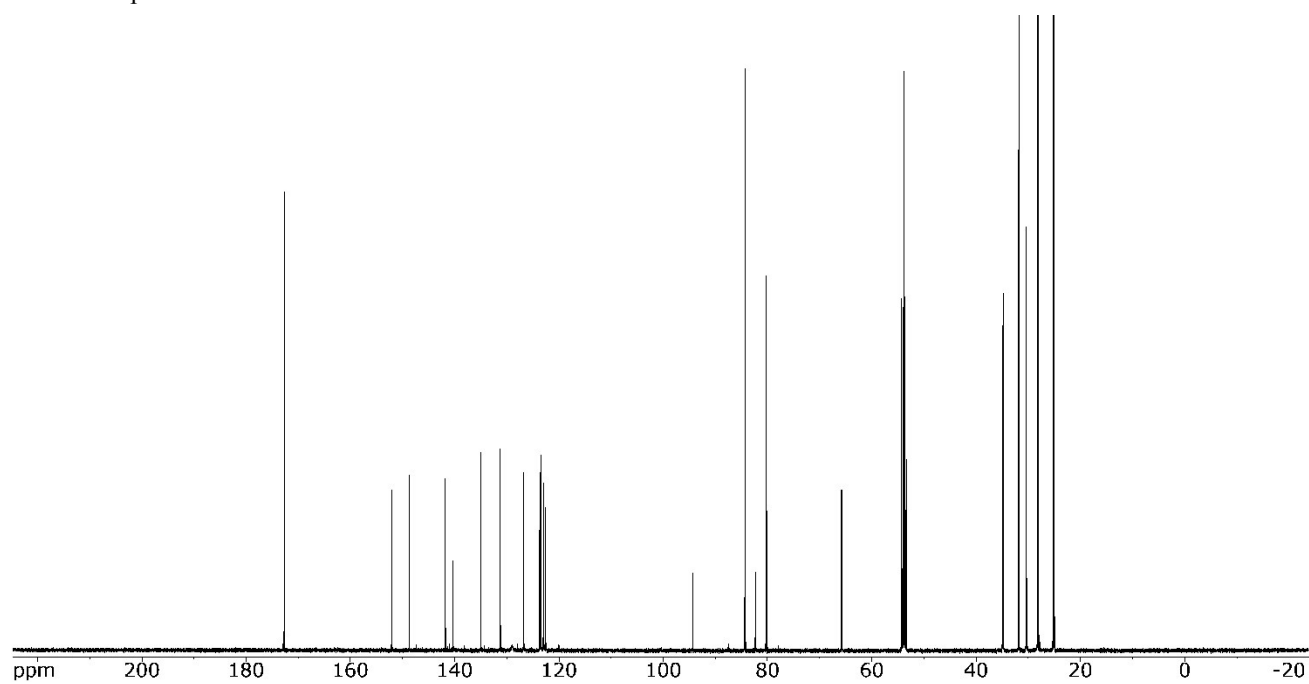
$^1H$  NMR (500 MHz,  $CD_2Cl_2$ , *r. t.*):  $\delta$  8.14 (s, 1H, Ar-*H*), 7.84 (s, 1H, Ar-*H*), 7.78 (d,  $J = 7.5$  Hz, 1H, Ar-*H*), 7.42-7.38 (m, 2H, Ar-*H*), 7.35 (s, 1H, Ar-*H*), 2.33 (t,  $J = 8.3$  Hz, 4H,  $-CH_2CH_2COO^tBu$ ), 1.62 (s, 6H,  $-C(CH_3)_2OH$ ), 1.40 (several peaks, 4H+12H,  $-CH_2CH_2COO^tBu$  and B(pin)-*H*), 1.28 (s, 18H,  $-COOC(CH_3)_3$ )

$^{13}C$  NMR (126 MHz,  $CD_2Cl_2$ , *r. t.*):  $\delta$  172.7, 152.0, 148.7, 141.8, 140.3, 135.0, 131.2, 126.8, 123.67, 123.49, 123.0, 122.5, 94.4, 84.3, 82.3, 80.2, 65.8, 54.3, 34.9, 31.8, 30.4, 28.1, 25.1.

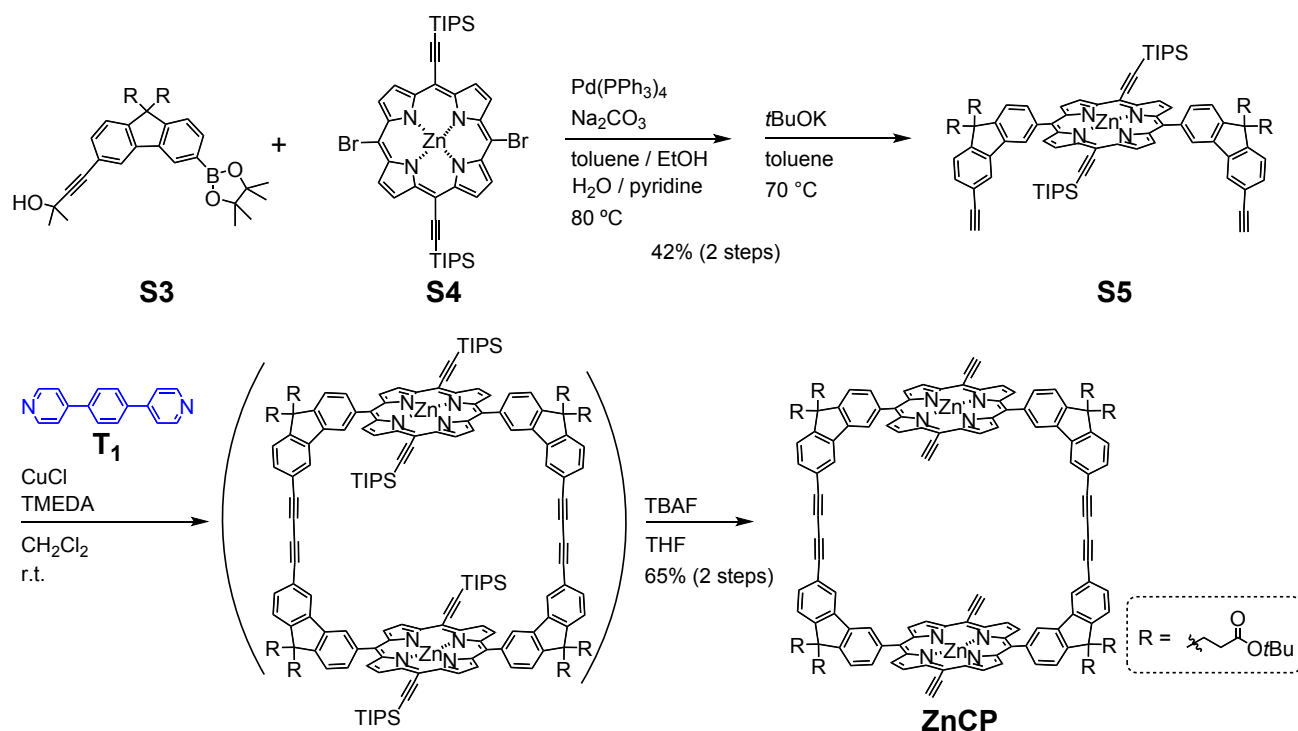
$^1\text{H}$  NMR spectrum of **S3**



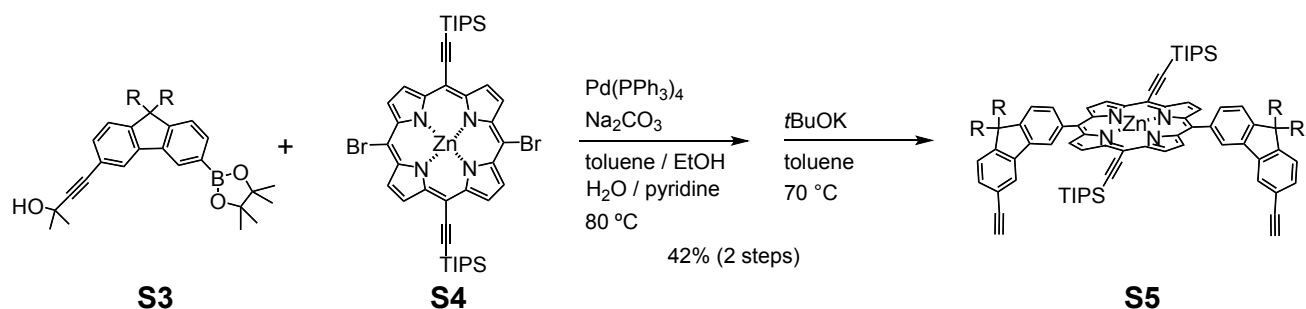
$^{13}\text{C}$  NMR spectrum of **S3**



## Scheme S2. Synthesis of ZnCP



### 2.3 Synthesis of S5



**S3** (960 mg, 1.52 mmol), **S4** (454 mg, 0.509 mmol), Pd(PPh<sub>3</sub>)<sub>4</sub> (119 mg, 0.103 mmol), and Na<sub>2</sub>CO<sub>3</sub> (699 mg, 6.59 mmol) were dissolved in degassed toluene / EtOH / H<sub>2</sub>O (5 / 1 / 1, v / v / v, 35.0 mL) and pyridine (2.5 mL) under Ar. The solution was stirred at 80 °C for 2 days. The organic layer was extracted with EtOAc, after which the residue was purified by silica gel column chromatography (EtOAc / hexane = 1 / 1, v / v) and GPC with CHCl<sub>3</sub> as an eluent. The product was dissolved in degassed toluene (34 mL) before <sup>t</sup>BuOK (568 μL, 1 M in THF, 0.568 mmol) was added. The solution was stirred at 70 °C for 30 min. The crude product was purified by silica gel column chromatography (CH<sub>2</sub>Cl<sub>2</sub> / EtOAc = 4 / 1, v / v) to yield **S5** (345 mg, 42%) as a purple solid.

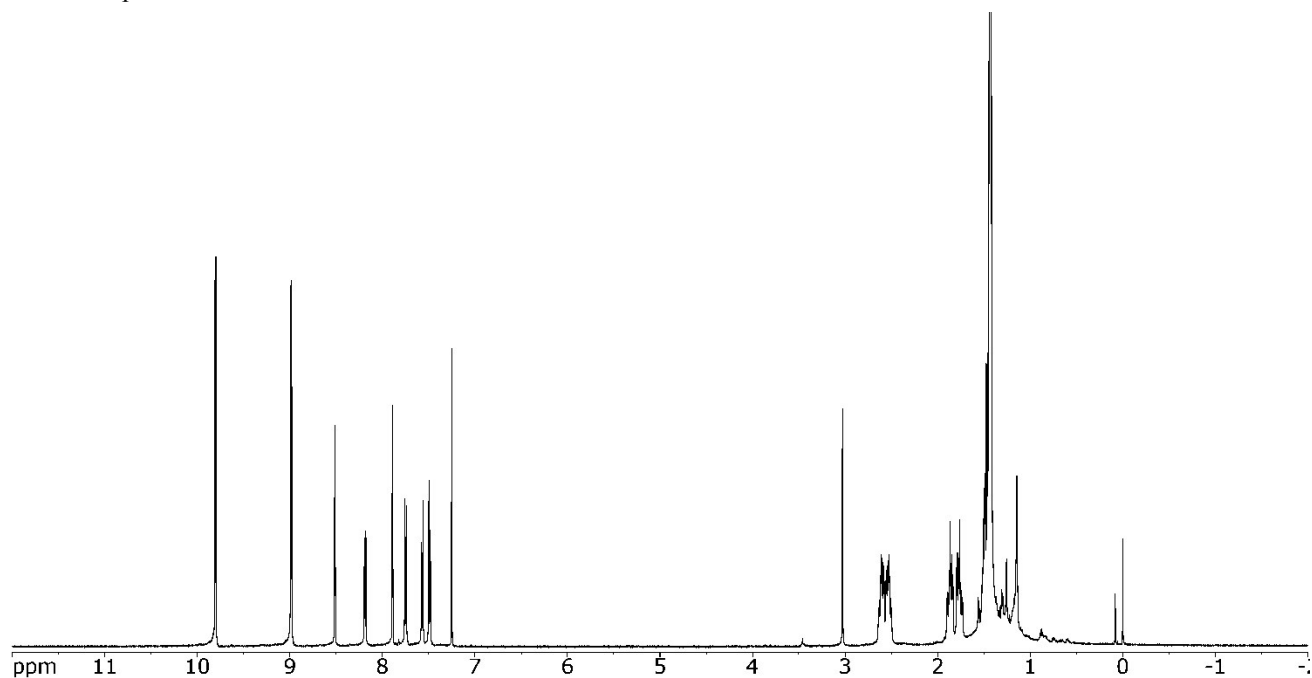
HR-MS (APCI-MS) (m/z) 1621.770 ([**S5**+H]<sup>+</sup>, C<sub>100</sub>H<sub>117</sub>N<sub>4</sub>O<sub>8</sub>Si<sub>2</sub>Zn, calcd. 1620.770)

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, r. t.): δ 9.80 (d, *J* = 4.6 Hz, 4H, β-*H*), 8.98 (d, *J* = 4.6 Hz, 4H, β-*H*), 8.51 (s, 2H, Ar-*H*), 8.19 (d, *J* = 7.6 Hz, 2H, Ar-*H*), 7.89 (s, 2H, Ar-*H*), 7.75 (d, *J* = 7.5 Hz, 2H, Ar-*H*), 7.57 (d, *J* = 8.1 Hz, 2H, Ar-*H*), 7.49 (d, *J* = 8.0 Hz, 2H, Ar-*H*), 3.03 (s, 2H, -CCH), 2.63-2.51 (m, 8H, -CH<sub>2</sub>CH<sub>2</sub>COO<sup>t</sup>Bu), 1.90-1.73 (m, 8H, -CH<sub>2</sub>CH<sub>2</sub>COO<sup>t</sup>Bu), 1.51-

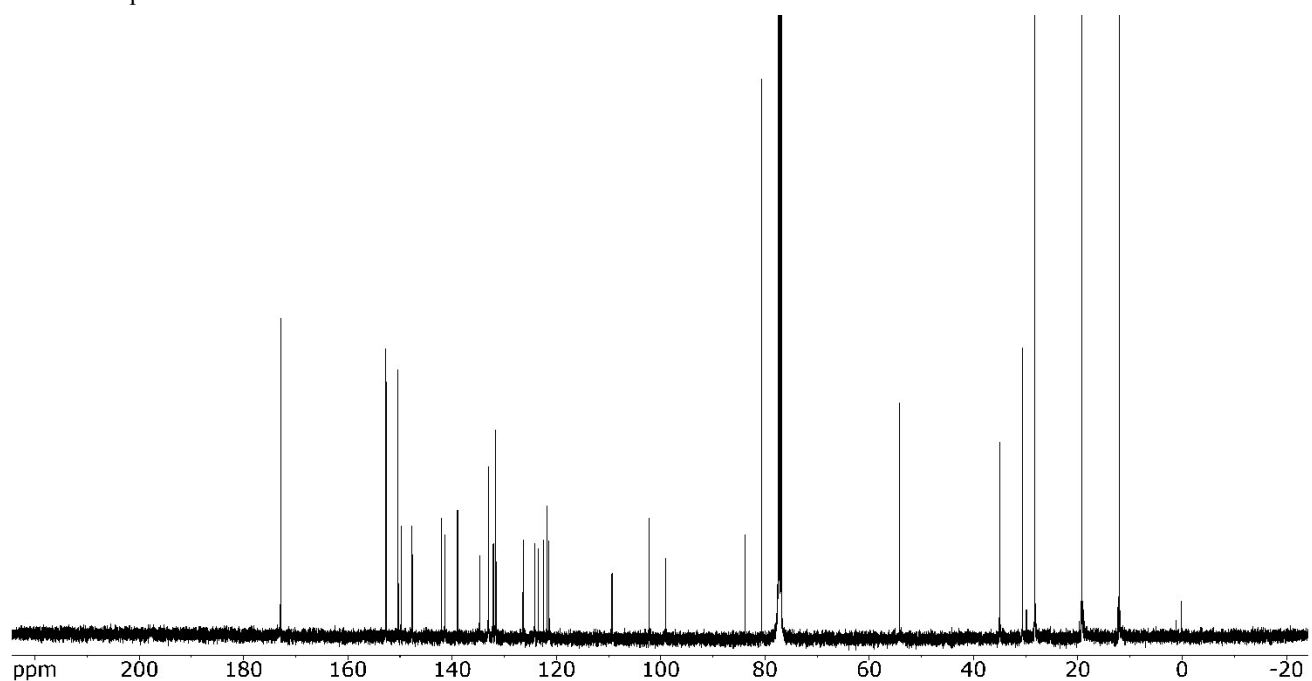
1.43 (m, 36H+42H, -COOC(CH<sub>3</sub>)<sub>3</sub>, -Si(CH(CH<sub>3</sub>)<sub>2</sub>)<sub>3</sub>).

<sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>, r. t.): δ 172.7, 152.5, 150.2, 149.6, 147.5, 141.9, 141.2, 138.8, 134.6, 132.9, 132.0, 131.5, 126.2, 124.0, 123.3, 122.3, 121.6, 121.3, 109.2, 102.0, 98.9, 83.7, 80.5, 54.0, 34.8, 30.5, 28.1, 19.1.

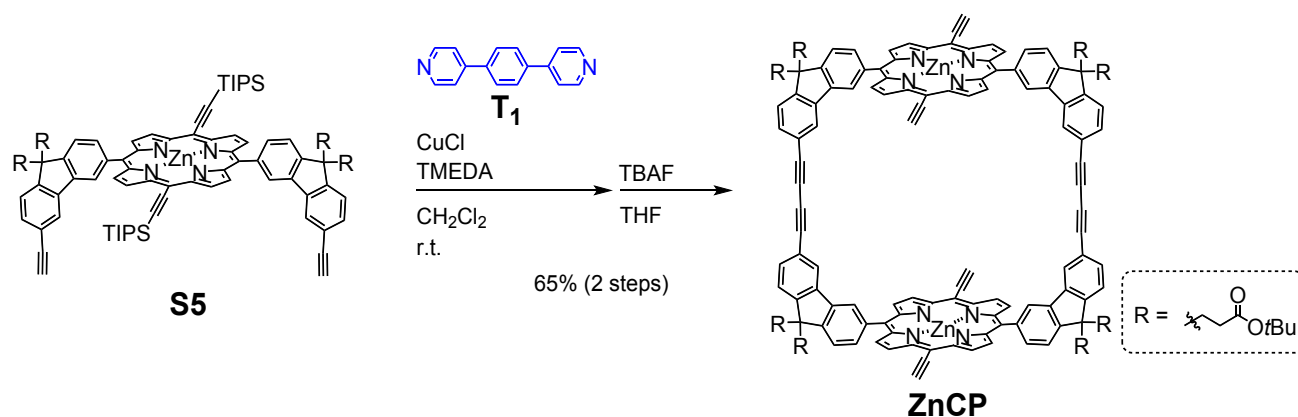
<sup>1</sup>H NMR spectrum of **S5**



<sup>13</sup>C NMR spectrum of **S5**



## 2.4 Synthesis of ZnCP



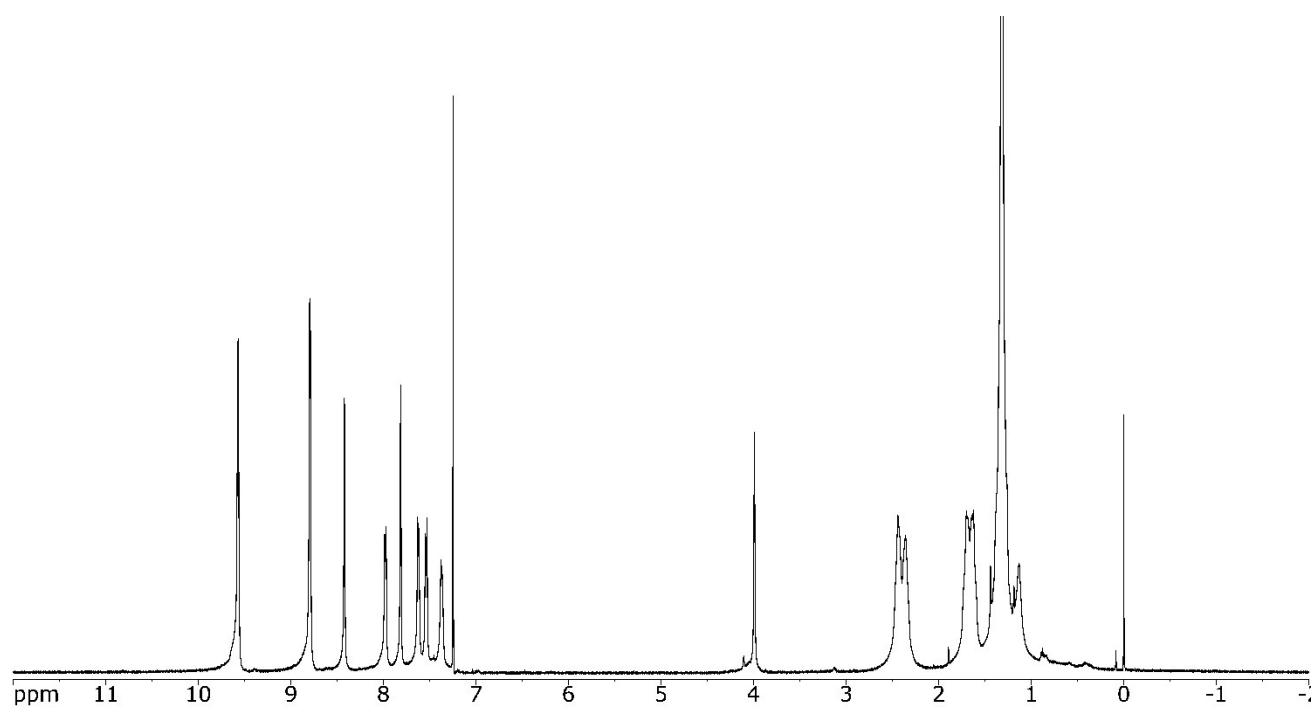
**S5** (325 mg, 0.200 mmol) and **T1** (281 mg, 1.21 mmol) were dissolved in dry CH<sub>2</sub>Cl<sub>2</sub> (509 mL), and CuCl (2.46 g, 24.8 mmol) was added to the solution under Ar before TMEDA (1.8 mL) was added slowly. Ar was substituted to dry air, after which the solution was stirred at r. t. for 15 h. The solution was passed through the Celite pad before extracted with EtOAc. The residue was purified by silica gel column chromatography (EtOAc / hexane / pyridine = 1 / 1 / 0.01, v / v / v) and GPC using CHCl<sub>3</sub> as an eluent. Pyridine was added as an eluent to remove **T1** from **ZnCP**. The product was dissolved in THF (83 mL) before addition of TBAF (0.400 mL, 0.400 mmol, 1M in THF). The solution was stirred at r. t. for 30 min. the crude mixture was purified by GPC using CHCl<sub>3</sub> as an eluent to afford **ZnCP** (169 mg, 65%) as a purple solid.

*HR-MS (MALDI-TOF MS)* (m/z) 2613.965 ([**ZnCP**+H]<sup>+</sup>, C<sub>164</sub>H<sub>149</sub>N<sub>8</sub>O<sub>16</sub>Zn<sub>2</sub>, calcd. 2613.967)

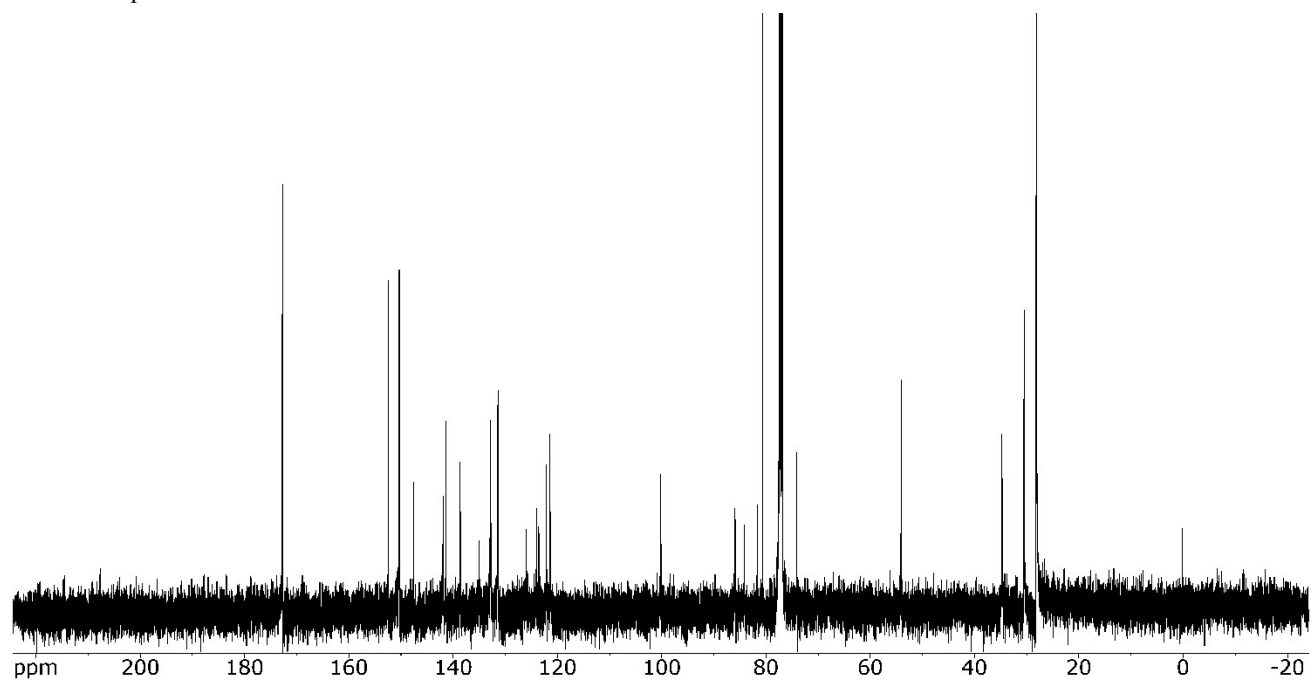
<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, r. t.): δ 9.57 (d, *J* = 3.5 Hz, 8H, β-*H*), 8.79 (d, *J* = 3.9 Hz, 8H, β-*H*), 8.42 (s, 4H, Ar-*H*), 7.98 (d, *J* = 6.7 Hz, 4H, Ar-*H*), 7.81 (s, 4H, Ar-*H*), 7.62 (d, *J* = 6.8 Hz, 4H, Ar-*H*), 7.54 (d, *J* = 7.9 Hz, 4H, Ar-*H*), 7.37 (d, *J* = 6.9 Hz, 4H, Ar-*H*), 3.99 (s, 4H, -CCH), 2.44-2.35 (m, 16H, -CH<sub>2</sub>CH<sub>2</sub>COO<sup>t</sup>Bu), 1.70-1.60 (m, 16H, -CH<sub>2</sub>CH<sub>2</sub>COO<sup>t</sup>Bu), 1.32 (s, 72H, -CH<sub>2</sub>CH<sub>2</sub>COOC(CH<sub>3</sub>)<sub>3</sub>).

<sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>, r. t.): δ 172.7, 152.4, 150.3, 150.2, 147.6, 141.9, 141.4, 138.7, 135.0, 132.9, 131.4, 125.9, 124.0, 123.6, 122.1, 121.4, 121.3, 100.2, 100.1, 85.9, 84.2, 81.7, 80.6, 74.1, 54.1, 34.7, 30.5, 28.1.

$^1\text{H}$  NMR spectrum of **ZnCP**



$^{13}\text{C}$  NMR spectrum of **ZnCP**

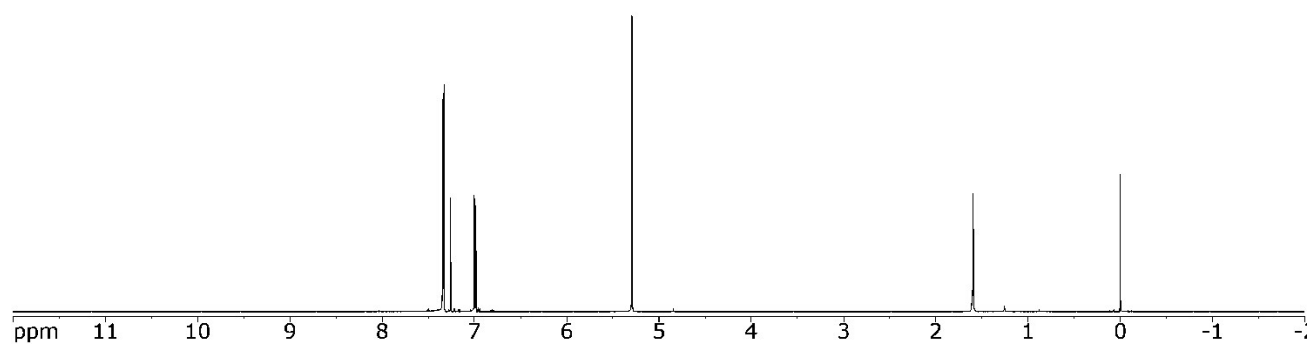


*HR-MS (EI MS)* (m/z) 345.8348 ( $[\mathbf{S6}]^+$ , C<sub>6</sub>H<sub>4</sub>I<sub>2</sub>O, calcd. 345.8352)

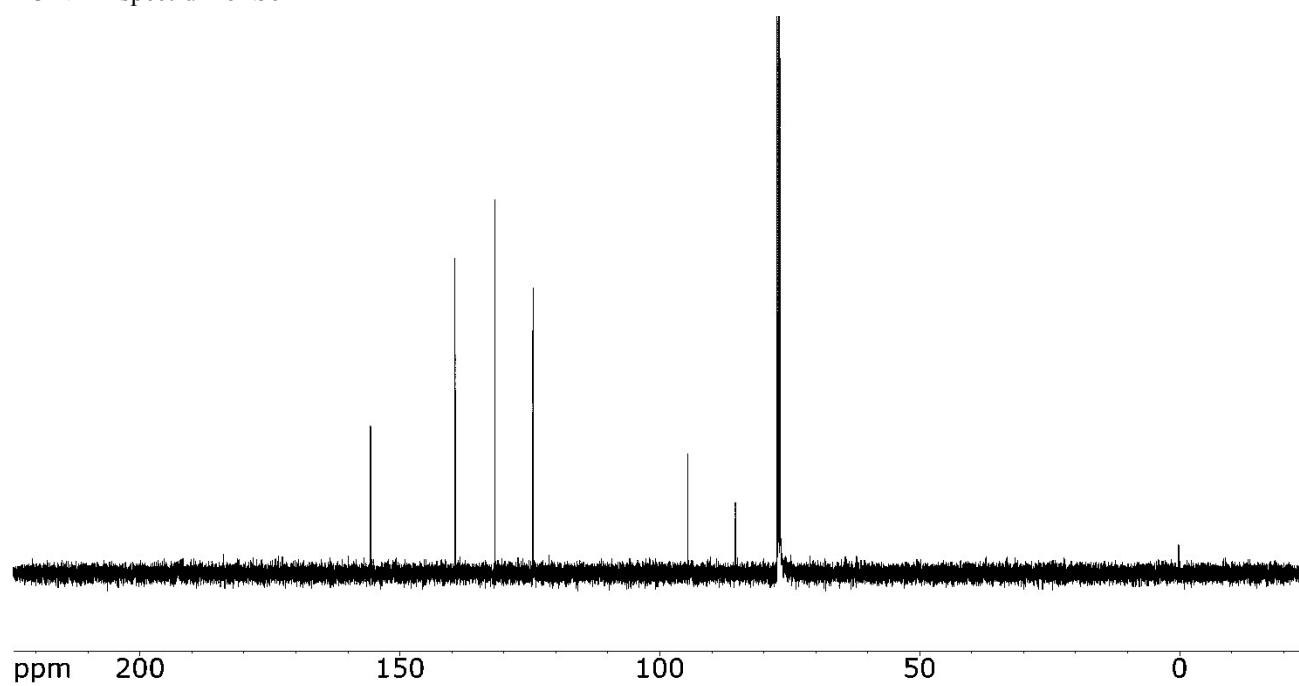
<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, r. t.): δ 7.34 (d, *J* = 2.0 Hz, 1H, Ar-*H*), 7.34 (d, *J* = 8.3 Hz, 1H, Ar-*H*), 7.00 (dd, *J* = 8.3, 2.0 Hz, 1H, Ar-*H*), 5.29 (s, 1H, -OH).

$^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ , r. t.):  $\delta$  155.6, 139.4, 131.7, 124.4, 94.6, 85.4.

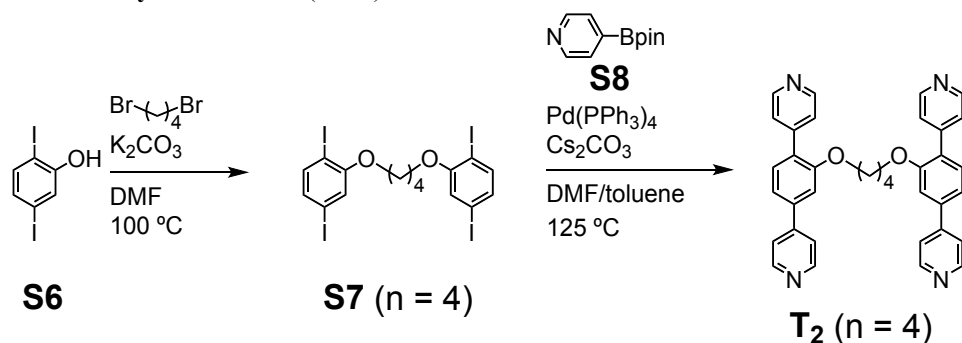
$^1\text{H}$  NMR spectrum of **S6**



$^{13}\text{C}$  NMR spectrum of **S6**



## 2.6 Synthesis of **T**<sub>2</sub> (n = 4)



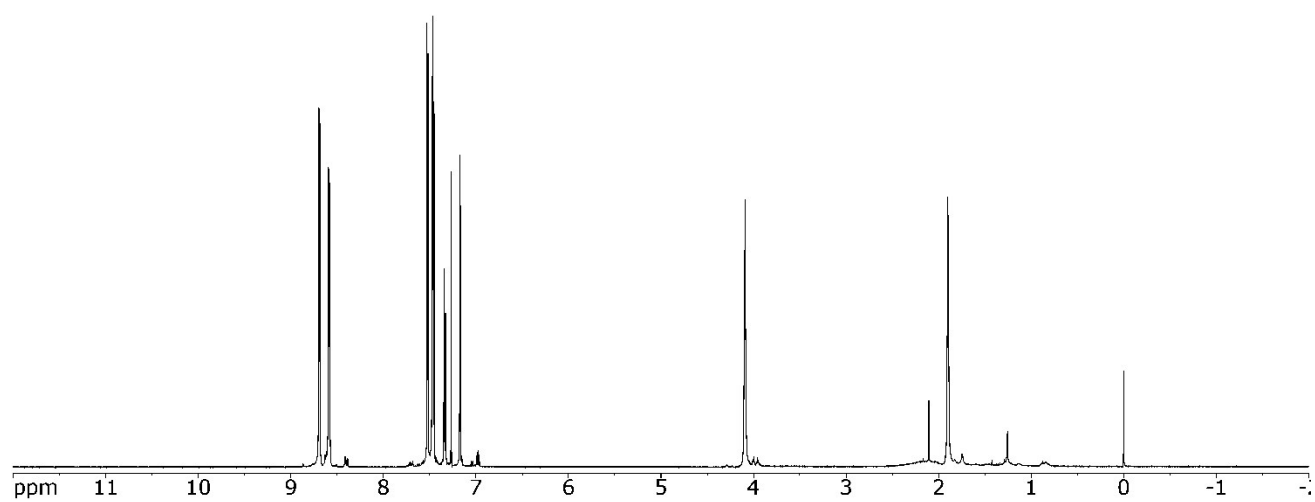
**S6** (300 mg, 1.00 mmol),  $K_2CO_3$  (222 mg, 1.61 mmol), and 1,4-dibromobutane (57.8  $\mu$ L, 0.485 mmol) were dissolved in dry DMF (15 mL) under Ar. The solution was stirred at 100 °C overnight, after which the residue was washed by water twice and brine once. The organic layer was dried by  $Na_2SO_4$  and evaporated. The residue was purified by silica gel column chromatography ( $CH_2Cl_2$  / hexane = 1 / 9, v / v) to afford **S7** (n = 4) (202 mg, 0.271 mmol) as a white solid. Half amount of **S7** (n = 4) (101 mg, 0.136 mmol), **S8** (333 mg, 1.63 mmol),  $Cs_2CO_3$  (530 mg, 1.63 mmol), and  $Pd(PPh_3)_4$  (94.0 mg, 81.3  $\mu$ mol) were dissolved in DMF / toluene (1 / 1, v / v, 6.6 mL) under Ar. The solution was stirred at 125 °C overnight. The solvents were removed under vacuum, after which the residue was passed through a Celite pad. The crude mixture was purified by silica gel column chromatography ( $CH_2Cl_2$  / MeOH = 100 / 1 to 10 / 1, v / v) to afford **T**<sub>2</sub> (n = 4) (34.0 mg, 25% for 2 steps) as a yellow solid.

*HR-MS (APCI MS)* (m/z) 551.2429 ( $[T_2$  (n = 4) + H]<sup>+</sup>,  $C_{36}H_{31}N_4O_2$ , calcd. 551.2427)

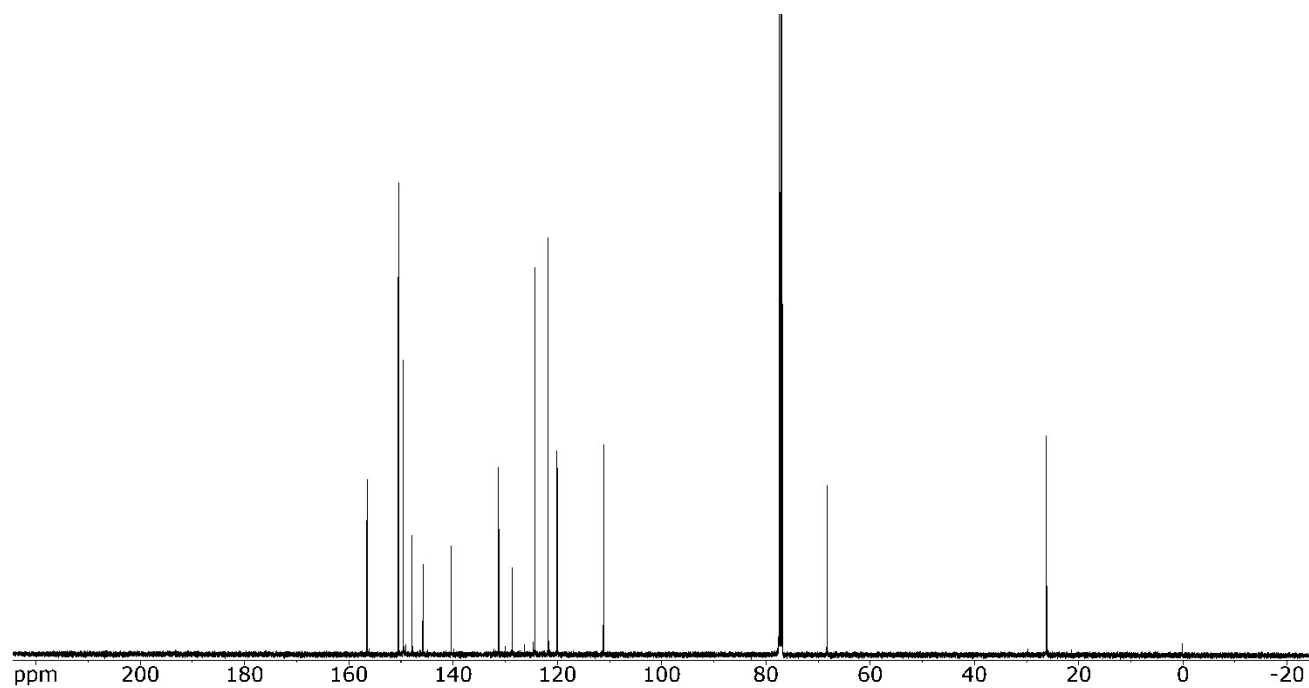
<sup>1</sup>H NMR (500 MHz,  $CDCl_3$ , r. t.):  $\delta$  8.69 (dd,  $J$  = 4.5, 1.5 Hz, 4H, Ar-*H*), 8.59 (dd,  $J$  = 4.5, 1.5 Hz, 4H, Ar-*H*), 7.52 (dd,  $J$  = 4.5, 1.6 Hz, 4H, Ar-*H*), 7.47-7.45 (m, 6H, Ar-*H*), 7.33 (dd,  $J$  = 7.9, 1.6 Hz, 2H, Ar-*H*), 7.17 (d,  $J$  = 1.6 Hz, 2H, Ar-*H*), 4.09 (t,  $J$  = 4.5 Hz, 4H, -OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-), 1.90 (m, 4H, -OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-).

<sup>13</sup>C NMR (126 MHz,  $CDCl_3$ , r. t.):  $\delta$  156.5, 150.5, 149.6, 147.8, 145.8, 140.4, 131.3, 128.7, 124.3, 121.7, 120.0, 111.1, 68.3, 26.2.

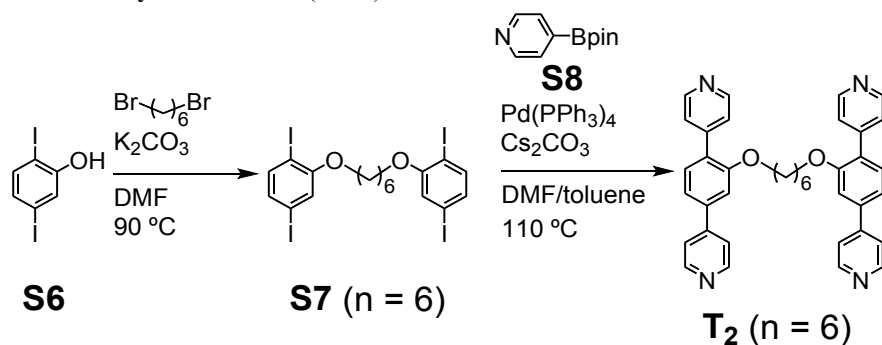
$^1\text{H}$  NMR spectrum of  $\text{T}_2$  ( $n = 4$ )



$^{13}\text{C}$  NMR spectrum of  $\text{T}_2$  ( $n = 4$ )



## 2.7 Synthesis of **T**<sub>2</sub> (n = 6)



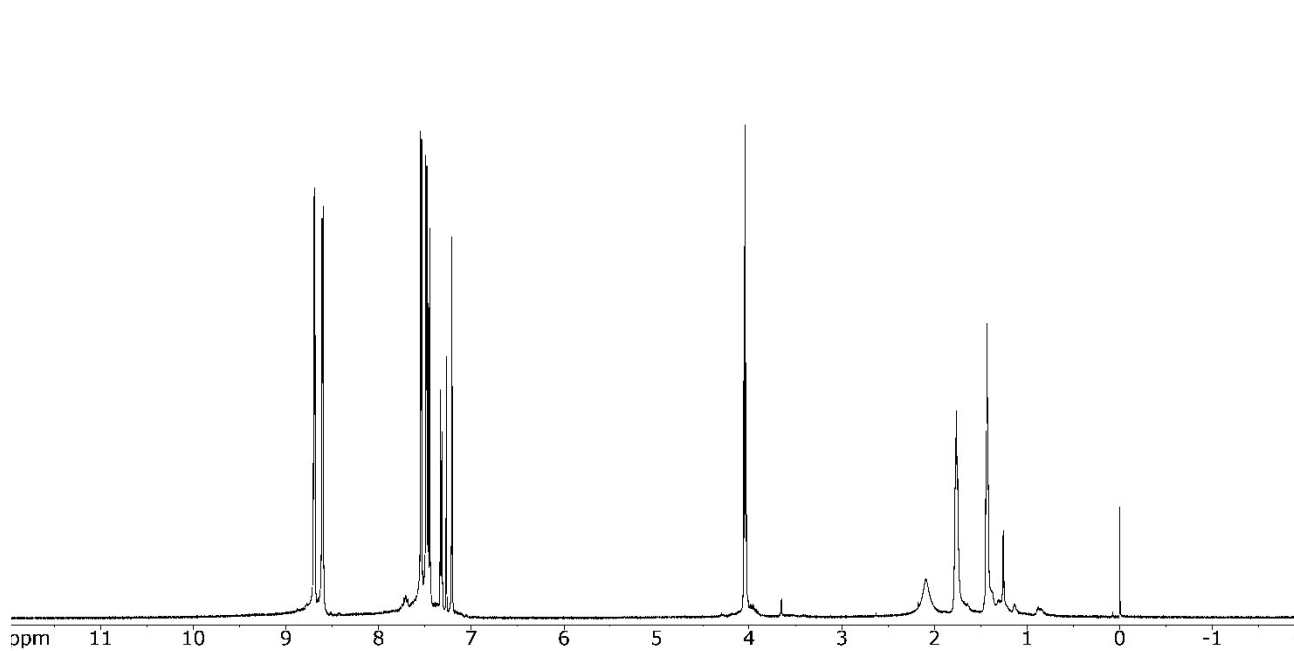
**S5** (200 mg, 0.669 mmol),  $\text{K}_2\text{CO}_3$  (105 mg, 0.760 mmol), and 1,6-dibromohexane (46.1  $\mu\text{L}$ , 0.304 mmol) were dissolved in dry DMF (10 mL). The solution was stirred at 90  $^{\circ}\text{C}$  overnight under Ar, after which the solution was extracted by EtOAc and washed by water four times and brine once. The residue was purified by silica gel column chromatography ( $\text{CH}_2\text{Cl}_2$  / hexane = 1 / 10, v / v) to afford **S7** (n = 6) as a white solid (153 mg, 0.198 mmol). **S7** (n = 6) (153 mg, 0.198 mmol), **S8** (406 mg, 1.98 mmol),  $\text{Cs}_2\text{CO}_3$  (538 mg, 1.98 mmol) and  $\text{Pd(PPh}_3)_4$  (68.6 mg, 59.4  $\mu\text{mol}$ ) were dissolved in DMF / toluene (1 / 1, v / v, 9.6 mL) under Ar. The solution was stirred at 110  $^{\circ}\text{C}$  overnight. The solvents were removed under vacuum, after which the residue was passed through a Celite pad. The crude mixture was purified by silica gel column chromatography ( $\text{CH}_2\text{Cl}_2$  / MeOH = 100 / 1 to 20 / 1, v / v) to afford **T**<sub>2</sub> (n = 6) (101 mg, 57% for 2 steps) as a yellow solid.

*HR-MS (APCI MS)* (m/z) 579.2739 ( $[\text{T}_2\text{ (n = 6)} + \text{H}]^+$ ,  $\text{C}_{38}\text{H}_{35}\text{N}_4\text{O}_2$ , calcd. 579.2760)

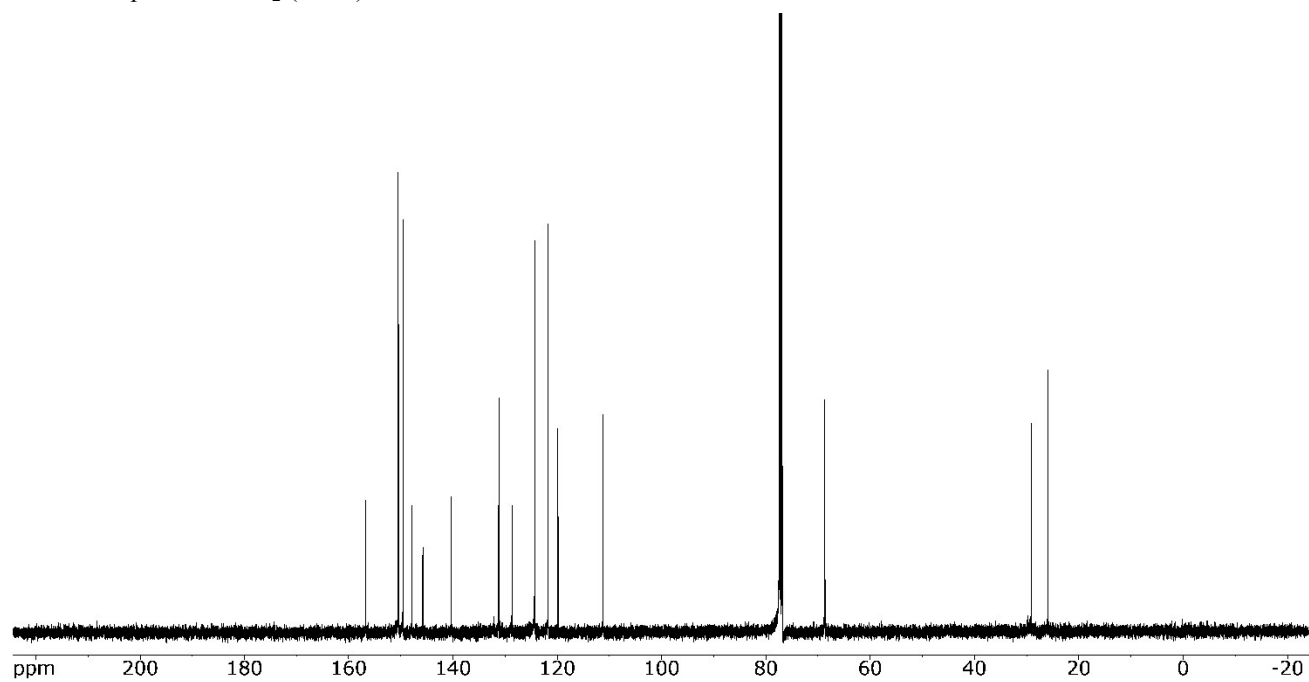
*$^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3$ , r. t.):*  $\delta$  8.69 (dd,  $J = 4.5, 1.3$  Hz, 4H, Ar-*H*), 8.61 (dd,  $J = 4.3, 1.3$  Hz, 4H, Ar-*H*), 7.54 (dd,  $J = 4.6, 1.3$  Hz, 4H, Ar-*H*), 7.48 (dd,  $J = 4.5, 1.3$  Hz, 4H, Ar-*H*), 7.46 (d,  $J = 7.9$  Hz, 2H, Ar-*H*), 7.32 (dd,  $J = 7.8, 1.1$  Hz, 2H, Ar-*H*), 7.21 (d,  $J = 1.1$  Hz, 2H, Ar-*H*), 4.05 (t,  $J = 6.2$  Hz, 4H,  $-\text{OCH}_2\text{CH}_2\text{CH}_2-$ ), 1.79-1.74 (m, 4H,  $-\text{OCH}_2\text{CH}_2\text{CH}_2-$ ), 1.45-1.42 (m, 4H,  $-\text{OCH}_2\text{CH}_2\text{CH}_2-$ ).

*$^{13}\text{C NMR}$  (126 MHz,  $\text{CDCl}_3$ , r. t.):*  $\delta$  156.7, 150.5, 149.6, 147.9, 145.8, 140.3, 131.2, 128.7, 124.3, 121.8, 119.9, 111.2, 68.7, 29.1, 25.9.

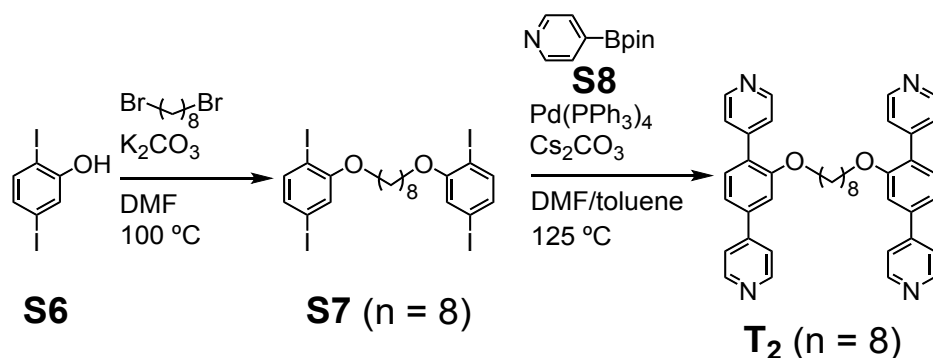
$^1\text{H}$  NMR spectrum of  $\text{T}_2$  ( $n = 6$ )



$^{13}\text{C}$  NMR spectrum of  $\text{T}_2$  ( $n = 6$ )



## 2.8 Synthesis of **T**<sub>2</sub> (n = 8)



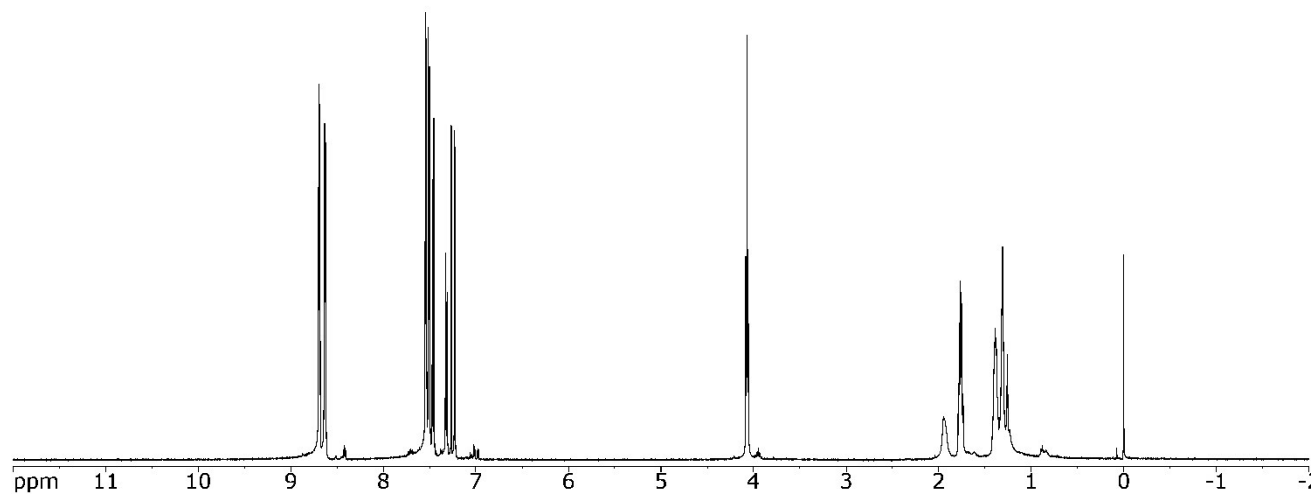
**S6** (400 mg, 1.16 mmol),  $K_2CO_3$  (216 mg, 1.56 mmol), and 1,8-dibromooctane (72.0  $\mu$ L, 0.387 mmol) were dissolved in dry DMF (20 mL) under Ar. The solution was stirred at 100 °C for 36 h, after which DMF was removed under vacuum. The residue was purified by silica gel column chromatography ( $CH_2Cl_2$  / hexane = 1 / 4 to 1 / 2, v / v) to afford **S7** (n = 8) (305 mg, 0.380 mmol) as a white solid. **S7** (n = 8) (305 mg, 0.380 mmol), **S8** (624 mg, 3.04 mmol),  $Cs_2CO_3$  (990 mg, 3.04 mmol) and  $Pd(PPh_3)_4$  (87.8 mg, 76.0  $\mu$ mol) were dissolved in DMF / toluene (1 / 1, v / v, 26.0 mL). The solution was stirred at 125 °C overnight. The solvents were removed under vacuum, after which the residue was passed through a Celite pad. The crude mixture was purified silica gel column chromatography ( $CH_2Cl_2$  / MeOH = 100 / 1 to 20 / 1, v / v) to afford **T**<sub>2</sub> (n = 8) (168 mg, 72% for 2 steps) as a yellow solid.

*HR-MS (APCI MS)* (m/z) 607.3062 ( $[T_2$  (n = 8) + H]<sup>+</sup>,  $C_{40}H_{39}N_4O_2$ , calcd. 607.3073)

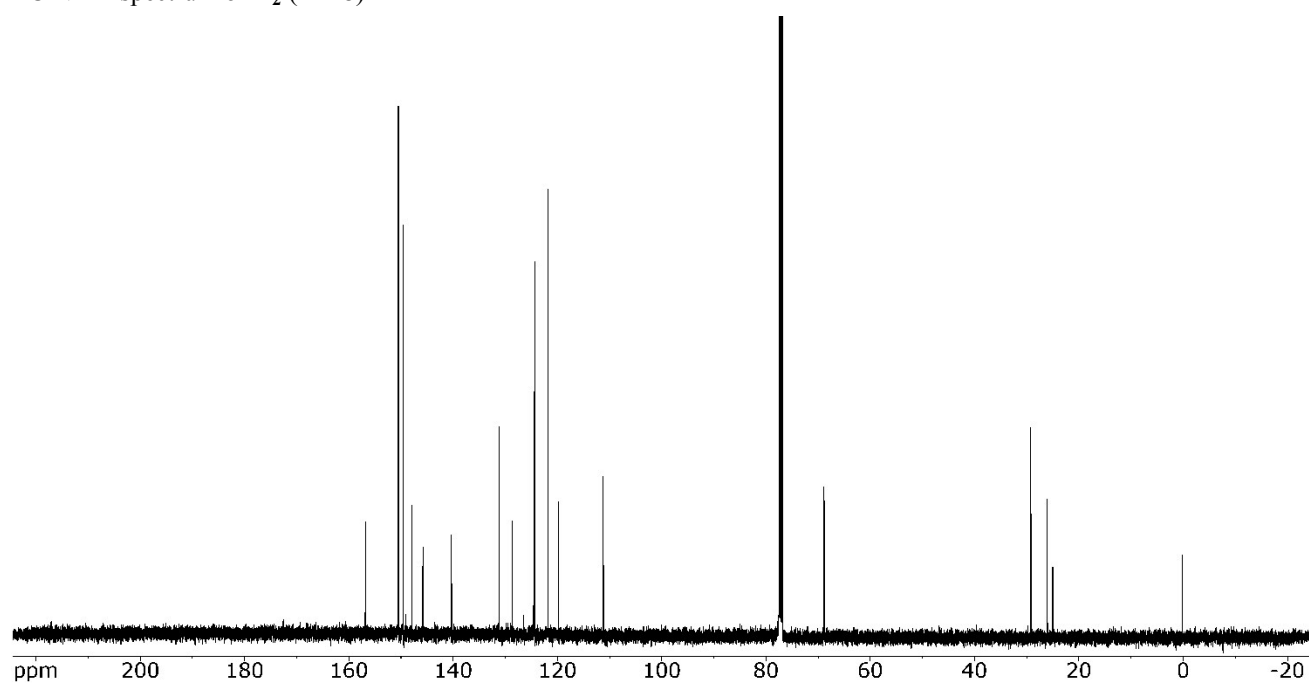
*<sup>1</sup>H NMR* (500 MHz,  $CDCl_3$ , r. t.):  $\delta$  8.69 (dd,  $J$  = 4.7, 1.3 Hz, 4H, Ar-*H*), 8.63 (d,  $J$  = 5.9 Hz, 4H, Ar-*H*), 7.54 (dd,  $J$  = 4.6, 1.4 Hz, 4H, Ar-*H*), 7.51 (dd,  $J$  = 4.7, 1.3 Hz, 4H, Ar-*H*), 7.46 (d,  $J$  = 7.8 Hz, 2H), 7.32 (dd,  $J$  = 7.9, 1.3 Hz, 2H, Ar-*H*), 7.23 (d,  $J$  = 1.2 Hz, 2H, Ar-*H*), 4.07 (t,  $J$  = 6.3 Hz, 4H, -OCH<sub>2</sub>CH<sub>2</sub>-), 1.78-1.73 (m, 4H, -OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-), 1.40-1.37 (m, 4H, -OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-), 1.31 (d,  $J$  = 6.3 Hz, 4H, -OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-).

*<sup>13</sup>C NMR* (126 MHz,  $CDCl_3$ , r. t.):  $\delta$  156.8, 150.5, 149.6, 148.0, 145.8, 140.3, 131.2, 128.6, 124.4, 121.8, 119.8, 111.2, 68.9, 29.2, 26.1, 25.0.

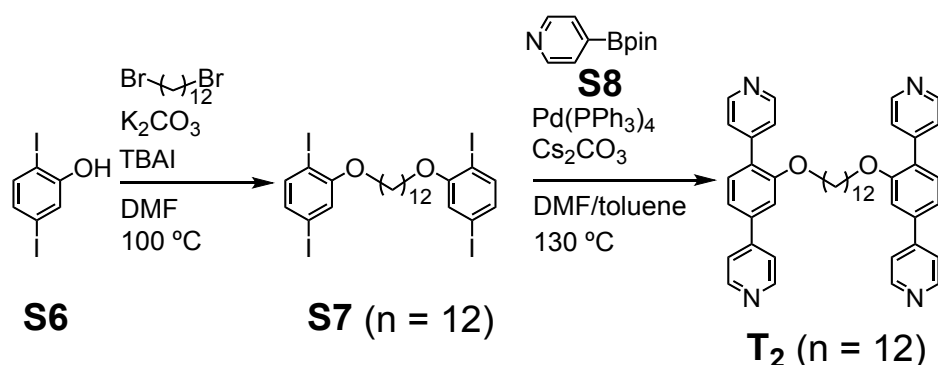
$^1\text{H}$  NMR spectrum of  $\text{T}_2$  ( $n = 8$ )



$^{13}\text{C}$  NMR spectrum of  $\text{T}_2$  ( $n = 8$ )



## 2.9 Synthesis of **T**<sub>2</sub> (n = 12)



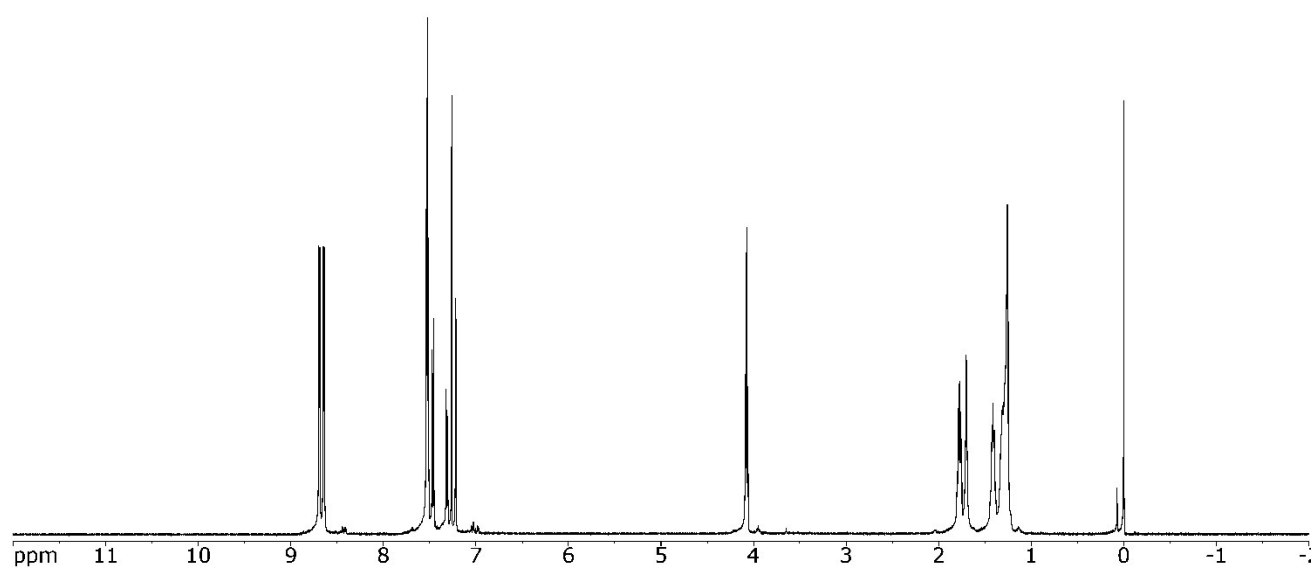
**S5** (500 mg, 1.52 mmol),  $K_2CO_3$  (210 mg, 1.52 mmol), 1,12-dibromododecane (176 mg, 0.508 mmol), and tetrabutylammonium iodide (9.38 mg, 25.4  $\mu$ mol) were dissolved in dry DMF (23.0 mL) under  $N_2$ . The solution was stirred at 100 °C overnight, after which the solution was filtered through the Celite pad. The residue was purified by silica gel column chromatography (EtOAc / hexane = 1 / 40, v / v) to afford **S7** (n = 12) (192 mg, 0.224 mmol) as a white solid. **S7** (n = 12) (172 mg, 0.200 mmol), **S8** (328 mg, 1.60 mmol),  $Cs_2CO_3$  (521 mg, 1.60 mmol) and  $Pd(PPh_3)_4$  (46.2 mg, 40.0  $\mu$ mol) were dissolved in DMF / toluene (1 / 1, v / v, 13.6 mL) under  $N_2$ . The solution was stirred at 130 °C overnight. The solvents were removed under vacuum, after which the residue was passed through a Celite pad. The crude mixture was purified by silica gel column chromatography ( $CH_2Cl_2$  / MeOH = 100 / 1 to 10 / 1, v / v) to afford **T**<sub>2</sub> (n = 12) (92 mg, 31% for 2 steps) as a pale yellow solid.

*HR-MS (APCI MS)* (m/z) 663.3675 ( $[T_2$  (n = 12)+H]<sup>+</sup>,  $C_{44}H_{47}N_4O_2$ , calcd. 663.3699)

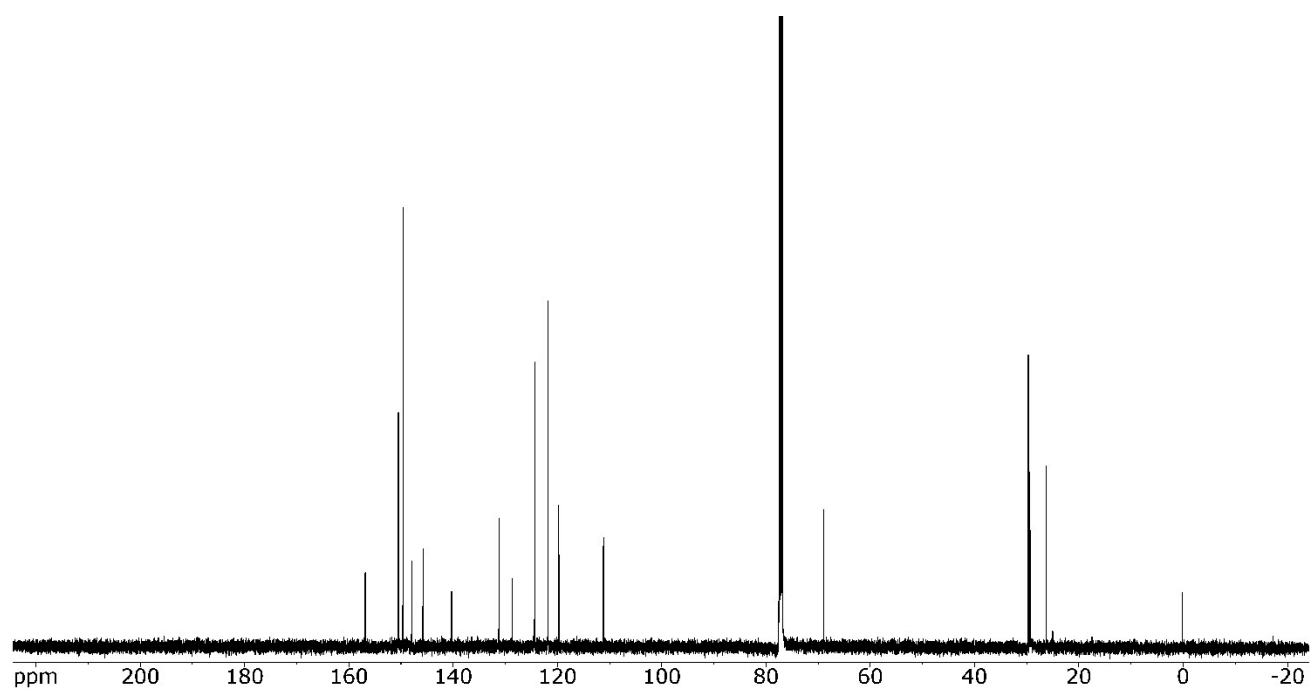
<sup>1</sup>H NMR (500 MHz,  $CDCl_3$ , r. t.):  $\delta$  8.69 (dd,  $J$  = 4.6, 1.3 Hz, 4H, Ar-*H*), 8.64 (dd,  $J$  = 4.7, 1.2 Hz, 4H, Ar-*H*), 7.54-7.51 (two double doublet peaks overlapped, 8H, Ar-*H*), 7.46 (d,  $J$  = 7.8 Hz, 2H, Ar-*H*), 7.31 (dd,  $J$  = 7.9, 1.0 Hz, 2H, Ar-*H*), 7.22 (d,  $J$  = 1.1 Hz, 2H, Ar-*H*), 4.07 (t,  $J$  = 6.4 Hz, 4H, -OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-), 1.77 (dt,  $J$  = 14.3, 6.9 Hz, 4H, -OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-), 1.41 (dt,  $J$  = 14.5, 7.1 Hz, 4H, -OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-), 1.33-1.26 (several peaks overlapped, 4H+4H+4H, -OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-).

<sup>13</sup>C NMR (126 MHz,  $CDCl_3$ , r. t.):  $\delta$  156.9, 150.5, 149.6, 148.0, 145.8, 140.3, 131.2, 128.7, 124.4, 121.8, 119.7, 111.2, 68.9, 29.7 (two methylene carbons were overlapped), 29.4, 29.3, 26.2.

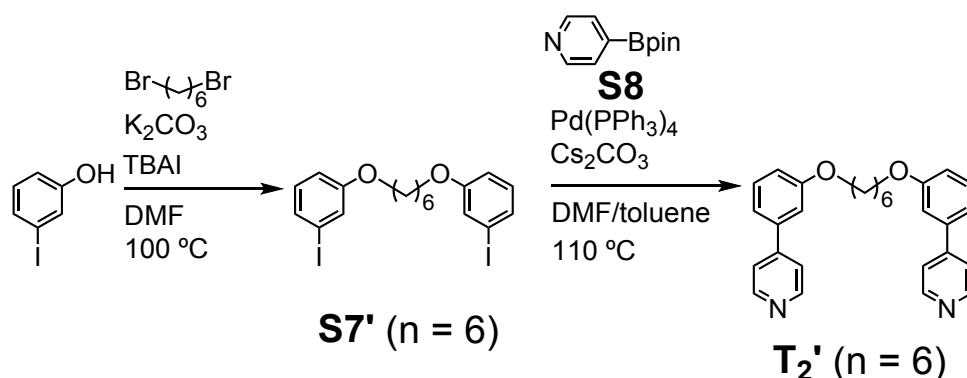
$^1\text{H}$  NMR spectrum of  $\text{T}_2$  ( $n = 12$ )



$^{13}\text{C}$  NMR spectrum of  $\text{T}_2$  ( $n = 12$ )



## 2.10 Synthesis of **T<sub>2</sub>'** (n = 6)



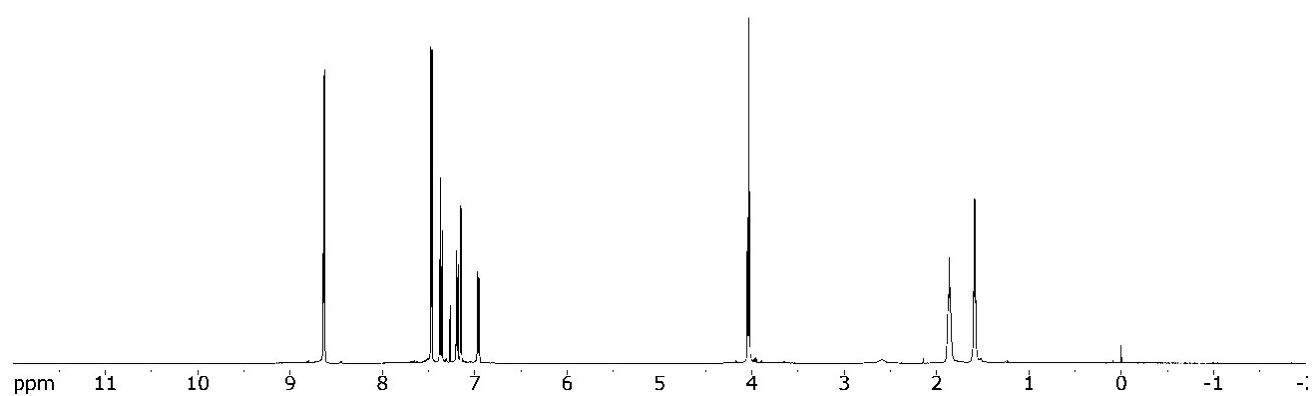
3-Iodophenol (400 mg, 1.82 mmol),  $K_2CO_3$  (315 mg, 2.28 mmol), 1,6-dibromohexane (138  $\mu$ L, 0.910 mmol), and tetrabutylammonium iodide (16.8 mg, 45.5  $\mu$ mol) were dissolved in dry DMF (20.0 mL) under  $N_2$ . The solution was stirred at 100 °C overnight, after which the solution was filtered through the Celite pad. The residue was purified by silica gel column chromatography ( $CH_2Cl_2$  / hexane = 1 / 8, v / v) to afford **S7'** as a white solid (393 mg, 0.753 mmol). The half amount of **S7'** (197 mg, 0.377 mmol), **S8** (385 mg, 1.88 mmol),  $Cs_2CO_3$  (612 mg, 1.88 mmol) and  $Pd(PPh_3)_4$  (65.1 mg, 56.3  $\mu$ mol) were dissolved in DMF / toluene (1 / 1, v / v, 10.6 mL) under  $N_2$ . The solution was stirred at 110 °C overnight. The solvents were removed under vacuum, after which the residue was passed through a Celite pad. The crude mixture was purified by silica gel column chromatography ( $CH_2Cl_2$  / EtOAc = 1 / 2, v / v) to afford **T<sub>2</sub>'** (n = 6) (96.4 mg, 50% for 2 steps) as a pale yellow solid.

*HR-MS (MALDI-TOF MS)* (m/z) 425.2218 ( $[T_2' (n = 6) + H]^+$ ,  $C_{28}H_{29}N_2O_2$ , calcd. 425.2229)

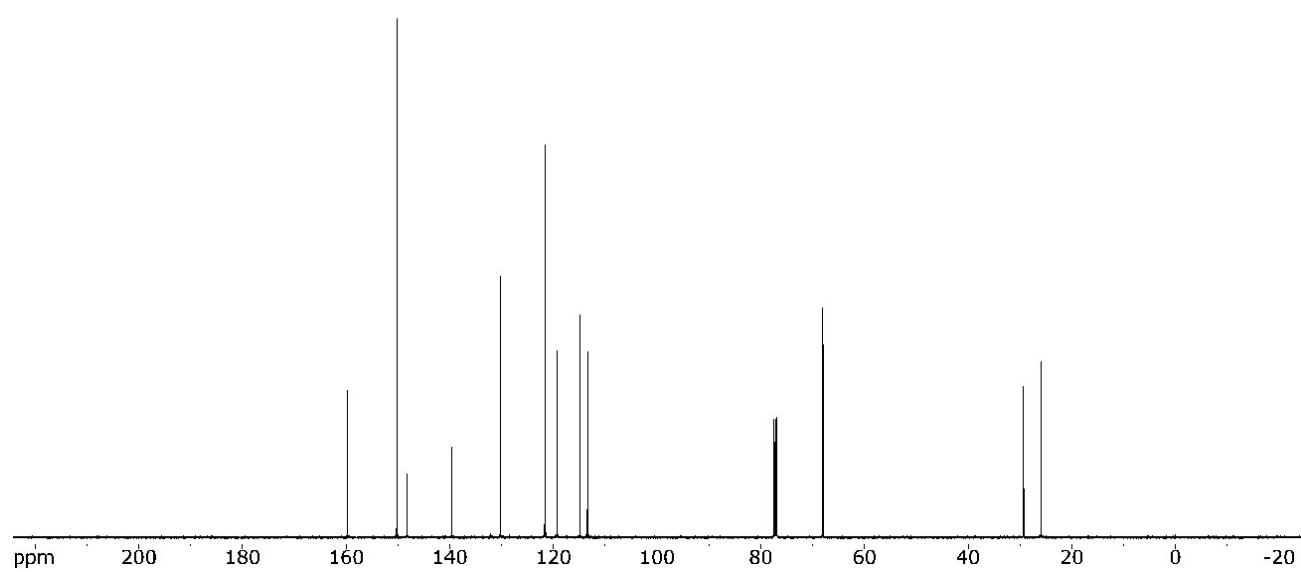
*<sup>1</sup>H NMR (500 MHz,  $CDCl_3$ , r. t.):*  $\delta$  8.63 (dd,  $J = 4.5, 1.6$  Hz, 4H, Ar-*H*), 7.47 (dd,  $J = 4.5, 1.7$  Hz, 4H, Ar-*H*), 7.37 (t,  $J = 7.9$  Hz, 2H, Ar-*H*), 7.20-7.18 (m, 2H, Ar-*H*), 7.15 (t,  $J = 2.1$  Hz, 2H, Ar-*H*), 6.96 (ddd,  $J = 8.2, 2.5, 0.7$  Hz, 2H, Ar-*H*), 4.04 (t,  $J = 6.4$  Hz, 4H, Ar- $OCH_2CH_2CH_2-$ ), 1.89-1.84 (m, 4H, Ar- $OCH_2CH_2CH_2-$ ), 1.59 (m, 4H, Ar- $OCH_2CH_2CH_2-$ ).

*<sup>13</sup>C NMR (126 MHz,  $CDCl_3$ , r. t.):*  $\delta$  159.7, 150.3, 148.3, 139.6, 130.2, 121.7, 119.3, 114.9, 113.5, 68.0, 29.3, 25.9.

$^1\text{H}$  NMR spectrum of  $\text{T}_2'$  ( $n = 6$ )

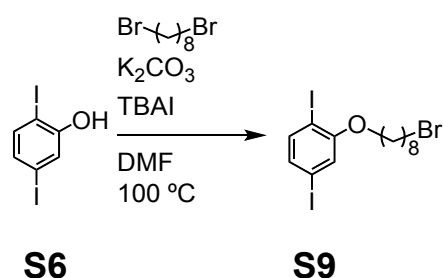


$^{13}\text{C}$  NMR spectrum of  $\text{T}_2'$  ( $n = 6$ )



## Scheme S4. Synthesis of T<sub>3</sub> (n = 8)

### 2.11 Synthesis of S9



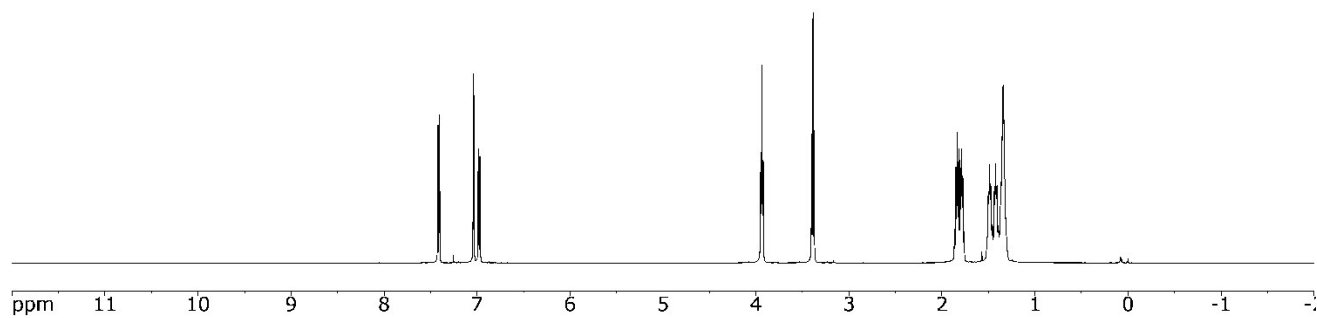
S6 (1.00 g, 2.89 mmol), K<sub>2</sub>CO<sub>3</sub> (912 mg, 6.60 mmol), 1,8-dibromooctane (2.20 mL, 11.9 mmol), and tetrabutylammonium iodide (54.5 mg, 147 μmol) were dissolved in dry DMF (100 mL). The solution was stirred at 100 °C overnight, after which DMF was removed under vacuum. The residue was filtered through the Celite pad and purified by silica gel column chromatography (CH<sub>2</sub>Cl<sub>2</sub> / hexane = 1 / 4, v / v) to afford S9 (638 mg, 41%) as a white solid.

HR-MS (APCI-TOF MS): (m/z) 535.8697 ([S9+H]<sup>+</sup>, C<sub>14</sub>H<sub>19</sub>BrI<sub>2</sub>O, calcd. 535.8709)

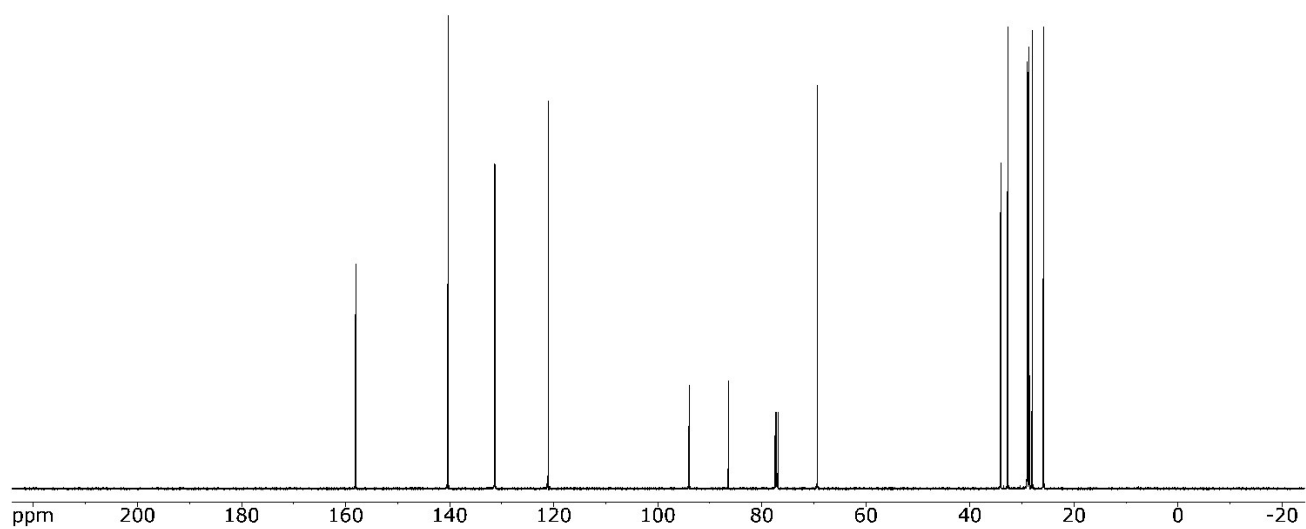
<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, r. t.): δ 7.41 (dd, *J* = 8.1, 1.1 Hz, 1H, Ar-*H*), 7.04 (s, 1H, Ar-*H*), 6.98 (dt, *J* = 8.1, 1.5 Hz, 1H, Ar-*H*), 3.94 (t, *J* = 6.24 Hz, 2H, Ar-OCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-), 3.39 (t, *J* = 6.94 Hz, 2H, -CH<sub>2</sub>-Br), 1.86-1.76 (m, 4H, methylene protons), 1.52-1.34 (m, 8H, methylene protons).

<sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>, r. t.): δ 158.1, 140.3, 131.3, 121.1, 94.0, 86.4, 69.3, 34.1, 32.7, 29.0, 28.9, 28.6, 28.1, 25.9.

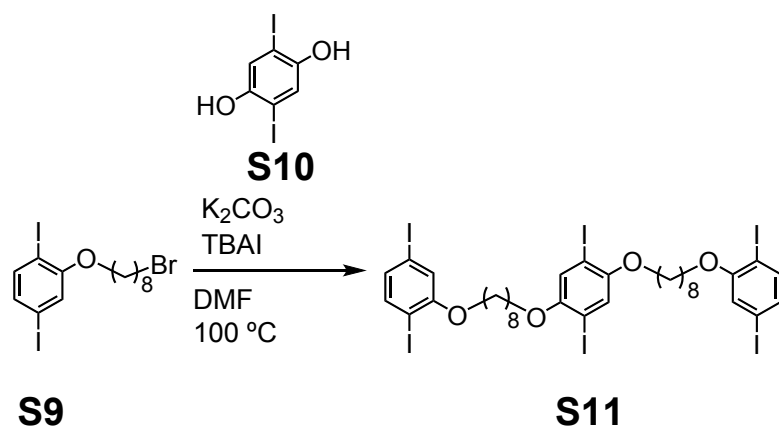
$^1\text{H}$  NMR spectrum of **S9**



$^{13}\text{C}$  NMR spectrum of **S9**



## 2.12 Synthesis of S11



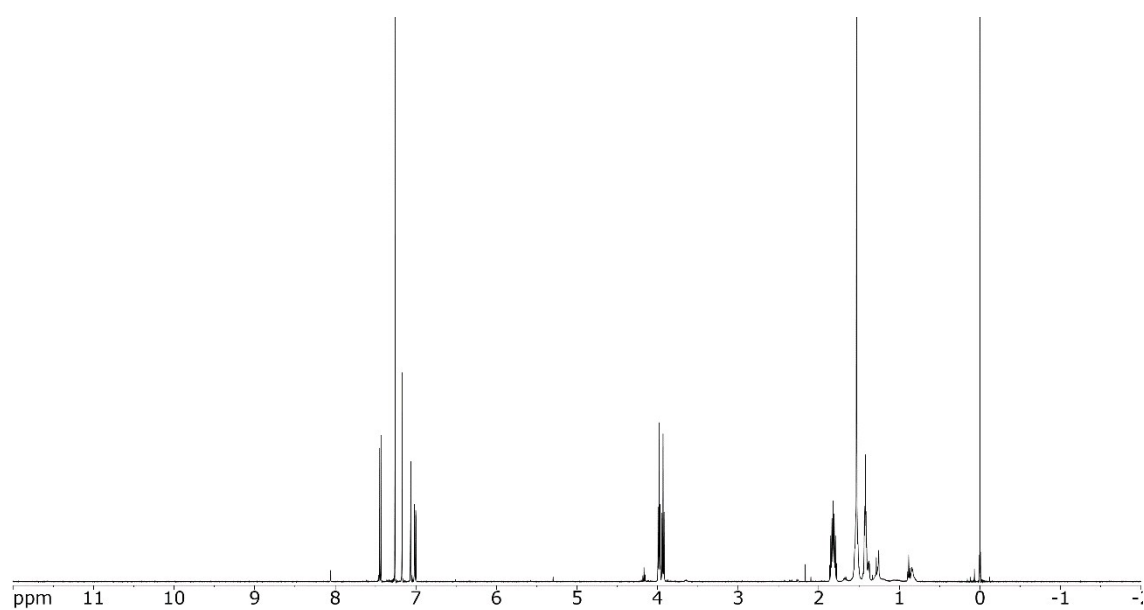
**S9** (337 mg, 0.628 mmol),  $\text{K}_2\text{CO}_3$  (107 mg, 0.775 mmol), **S10** (90.9 mg, 0.251 mmol), and tetrabutylammonium iodide (4.64 mg, 12.6  $\mu\text{mol}$ ) were dissolved in dry DMF (2.2 mL). The solution was stirred at 100  $^{\circ}\text{C}$  overnight, after which DMF was removed under vacuum. The residue was filtered through the Celite pad and purified by silica gel column chromatography ( $\text{CH}_2\text{Cl}_2$  / hexane = 1 / 1, v / v) to afford **S11** (295 mg, 92%) as a white solid.

*HR-MS (ESI-TOF MS)*: ( $m/z$ ) 1273.718 ( $[\text{S11}+\text{H}]^+$ ,  $\text{C}_{34}\text{H}_{41}\text{I}_6\text{O}_4$ , calcd. 1274.720)

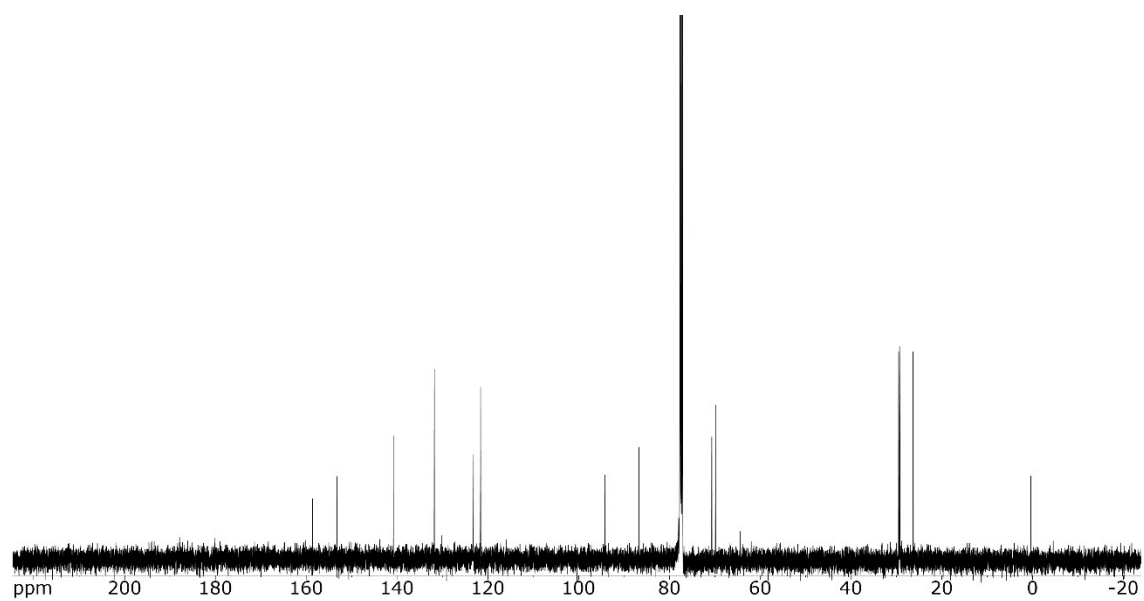
$^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3$ , *r. t.*):  $\delta$  7.44 (d,  $J$  = 8.2 Hz, 2H), 7.17 (s, 2H), 7.06 (d,  $J$  = 1.8 Hz, 2H), 7.01 (dd,  $J$  = 8.2, 1.8 Hz, 2H), 3.98 (t,  $J$  = 6.4, 4H,  $-\text{O}-\text{CH}_2-\text{CH}_2-$ ), 3.93 (t,  $J$  = 6.4 Hz, 4H,  $-\text{O}-\text{CH}_2-\text{CH}_2-$ ), 1.87-1.78 (m, 8H, methylene protons), 1.57-1.50 (m, 8H, methylene protons), 1.42 (m, 8H, methylene protons).

$^{13}\text{C NMR}$  (126 MHz,  $\text{CDCl}_3$ , *r. t.*):  $\delta$  158.6, 153.12, 140.8, 131.8, 123.1, 121.6, 94.2, 86.7 (two peaks overlapped), 70.7, 69.8, 29.5, 29.5, 29.46, 29.3, 26.3, 26.3.

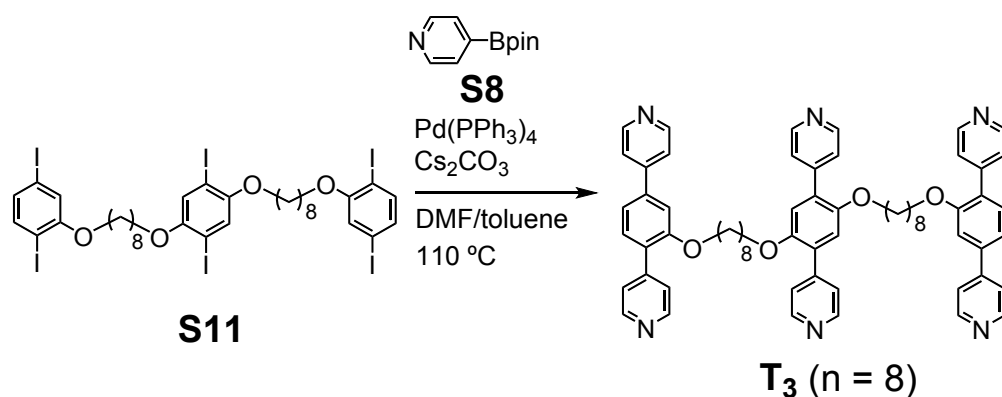
$^1\text{H}$  NMR spectrum of **S11**



$^{13}\text{C}$  NMR spectrum of **S11**



## 2.13 Synthesis of **T**<sub>3</sub> (n = 8)



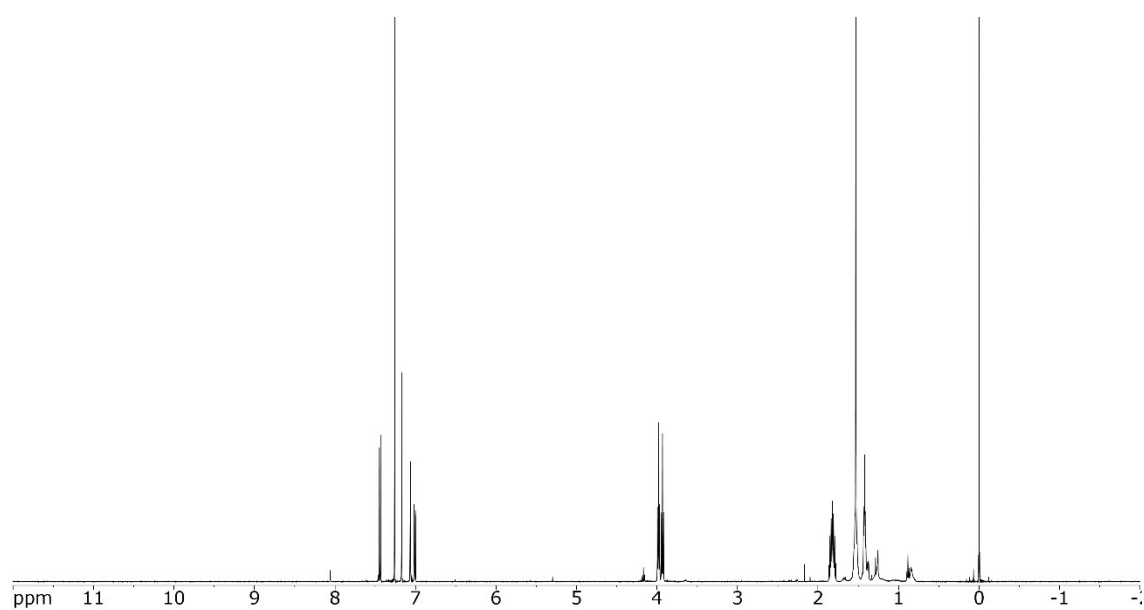
**S****11** (40.9 mg, 32.1  $\mu$ mol), **S****8** (186.8 mg, 0.910 mmol), Cs<sub>2</sub>CO<sub>3</sub> (206 mg, 0.628 mmol), and Pd(PPh<sub>3</sub>)<sub>4</sub> (7.7 mg, 6.7  $\mu$ mol) were dissolved in DMF / toluene (1 / 1, v / v, 2.4 mL). The solution was stirred at 110 °C overnight. The solvent was removed under vacuum, after which the residue was passed through a Celite pad and membrane filter. The crude mixture was purified by silica gel column chromatography (EtOAc / CH<sub>2</sub>Cl<sub>2</sub> = 10 / 3, v / v to EtOAc / MeOH / Et<sub>3</sub>N = 30 / 9 / 1, v / v / v) to afford **T**<sub>3</sub> (n = 8) (26.6 mg, 86%) as a white solid.

*HR-MS (ESI-TOF MS)*: (m/z) 981.5045 ([**T**<sub>3</sub> (n = 8) + H]<sup>+</sup>, C<sub>64</sub>H<sub>65</sub>N<sub>6</sub>O<sub>4</sub>, calcd. 981.5067)

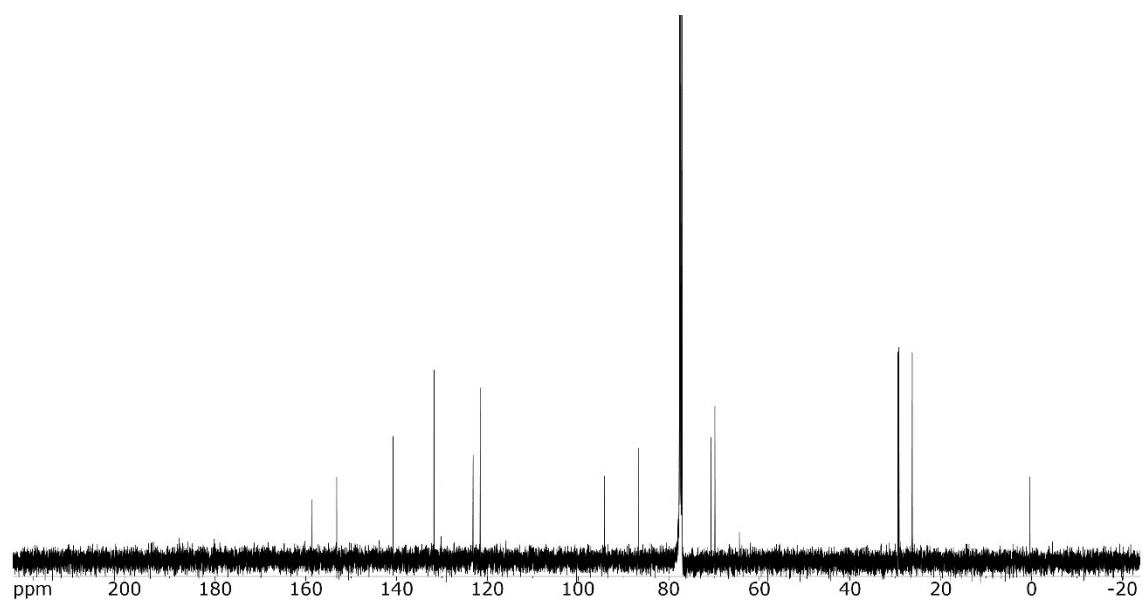
<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, r. t.):  $\delta$  8.70-8.67 (broad, 4H, Ar-H), 8.65-8.62 (broad, 8H, Ar-H), 7.56-7.53 (broad, 4H, Ar-H), 7.53-7.50 (broad, 8H, Ar-H), 7.46 (d, J = 7.8, 2H, Ar-H), 7.32 (d, J = 7.8, 2H, Ar-H), 7.23 (s, 2H, Ar-H), 6.99 (s, 2H, Ar-H), 4.07 (t, J = 6.3 Hz, 4H), 3.95 (t, J = 6.2 Hz, 4H), 1.79-1.73 (m, 4H), 1.73-1.67 (m, 4H), 1.38-1.24 (m, 16H).

<sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>, r. t.):  $\delta$  156.8 (two peaks overlapped), 150.53, 150.51, 149.65, 149.61, 147.9, 145.93, 145.78, 140.3, 131.2, 129.2, 128.6, 124.4, 121.8, 119.8, 115.6, 111.2, 69.7, 68.8, 29.35, 29.28 (two peaks overlapped), 29.20, 26.14, 26.12.

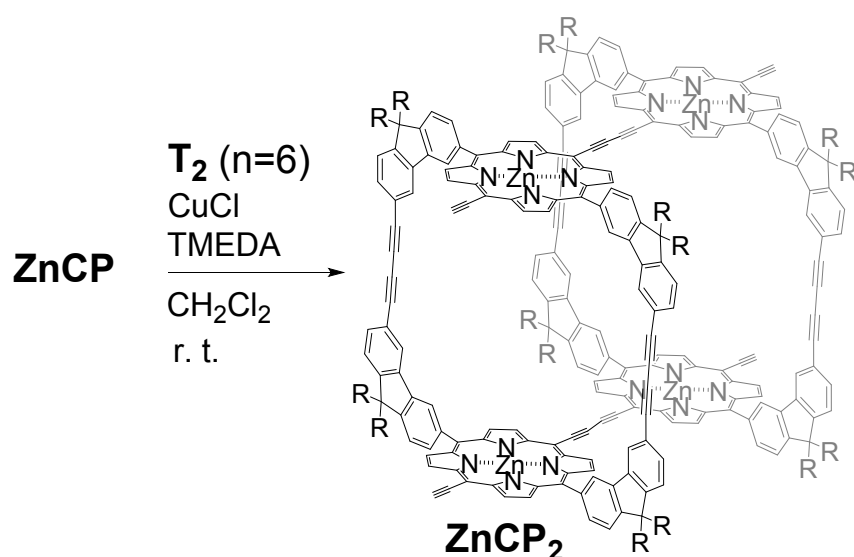
$^1\text{H}$  NMR spectrum of  $\text{T}_3$  ( $n = 8$ )



$^{13}\text{C}$  NMR spectrum of  $\text{T}_3$  ( $n = 8$ )



## 2.14 Synthesis of $\text{ZnCP}_2$

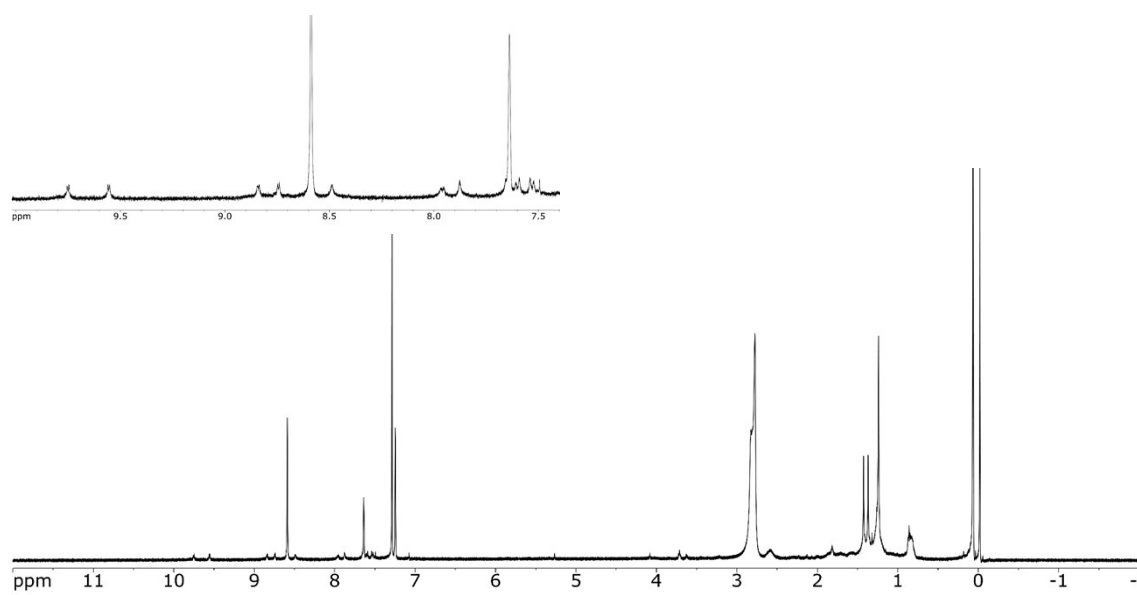


$\text{ZnCP}$  (2.17 mg, 0.83  $\mu\text{mol}$ ) and  $\text{T}_2$  ( $n = 6$ ) (0.248 mg, 0.428  $\mu\text{mol}$ ) were dissolved in  $\text{CH}_2\text{Cl}_2$  (217 mL) followed by slow additions of  $\text{CuCl}$  (171 mg, 1.73 mmol) and  $\text{TMEDA}$  (190  $\mu\text{L}$ , 1.30 mmol) under air. The solution was stirred at r. t. for 18 h under air. Another tubulation of  $\text{ZnCP}$  was conducted under the same condition as the above. The combined reaction mixture was carefully washed with  $\text{NH}_4\text{Cl}$  aq. with  $\text{TMEDA}$  once and  $\text{NH}_4\text{Cl}$  aq. once, water once and brine once to remove copper residue from the solution. The organic layer was dried over  $\text{Na}_2\text{SO}_4$ , then the solvent was removed *in vacuo*. The residue was filtered through silica gel ( $\text{CH}_2\text{Cl}_2$  /  $\text{EtOAc}$  / pyridine = 100 / 10 / 1, v / v / v) to remove  $\text{T}_2$  ( $n=6$ ) from  $\text{ZnCP}_2$ . The crude product was purified by GPC (toluene / pyridine = 100 / 1, v / v) to afford  $\text{ZnCP}_2$  (1.1 mg, 25%).

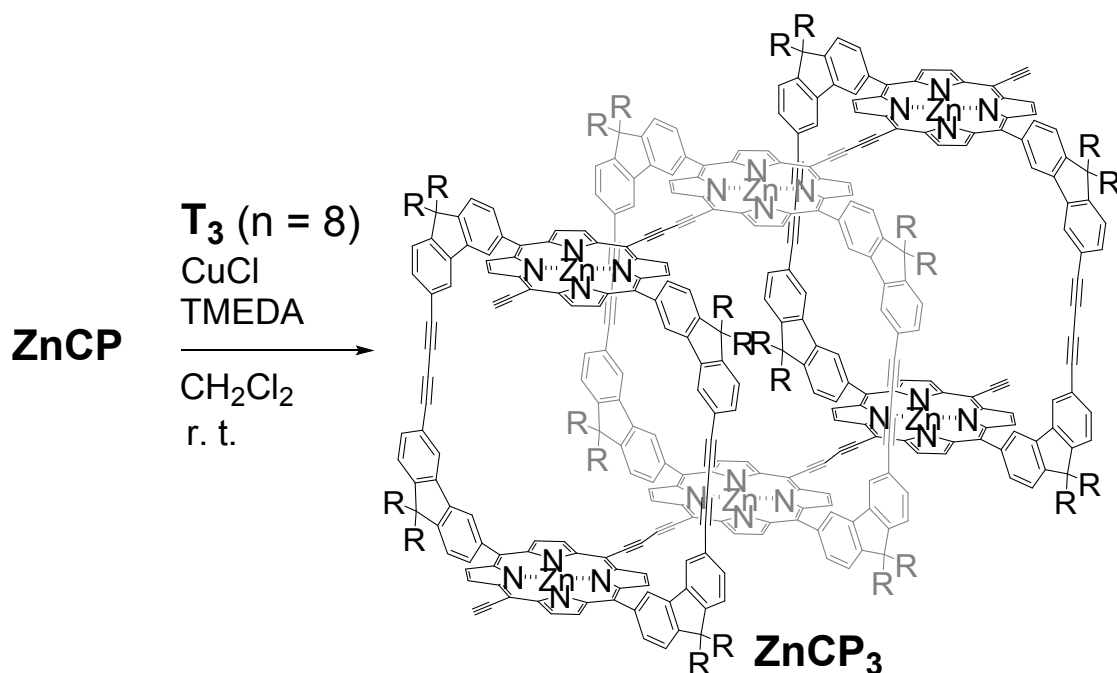
*HR-MS (MALDI-TOF MS)* ( $m/z$ ) 5231.37 ( $[\text{ZnCP}_2 + \text{H}]^+$ ,  $\text{C}_{328}\text{H}_{293}\text{N}_{16}\text{O}_{32}\text{Zn}_4$ , calcd. 5231.8984)

$^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3$  (+ 1 % pyridine- $d_5$ ), r. t.)  $\delta$ : 9.76 (d,  $J = 4.3$  Hz, 8H,  $\beta\text{-H}$ ), 9.56 (d,  $J = 4.4$  Hz, 8H,  $\beta\text{-H}$ ), 8.85 (d,  $J = 4.4$  Hz, 8H,  $\beta\text{-H}$ ), 8.75 (d,  $J = 4.4$  Hz, 8H,  $\beta\text{-H}$ ), 8.51 (s, 8H, Ar-H), 7.97 (d,  $J = 6.9$  Hz, 8H, Ar-H), 7.67 (s, 8H, Ar-H), 7.62 (d,  $J = 7.9$  Hz, 8H, Ar-H), 7.55 (d,  $J = 8.5$  Hz, 8H, Ar-H), 4.09 (s, 4H), 2.62-2.57 (m, 32H), 1.89-1.68 (m, 32H), 1.45 (s,  $-\text{CH}_2\text{CH}_2\text{COOC}(\text{CH}_3)_3$ ), 1.42 (s,  $-\text{CH}_2\text{CH}_2\text{COOC}(\text{CH}_3)_3$ ).

$^1\text{H}$  NMR spectrum of  $\text{ZnCP}_2$



## 2.15 Synthesis of $\text{ZnCP}_3$

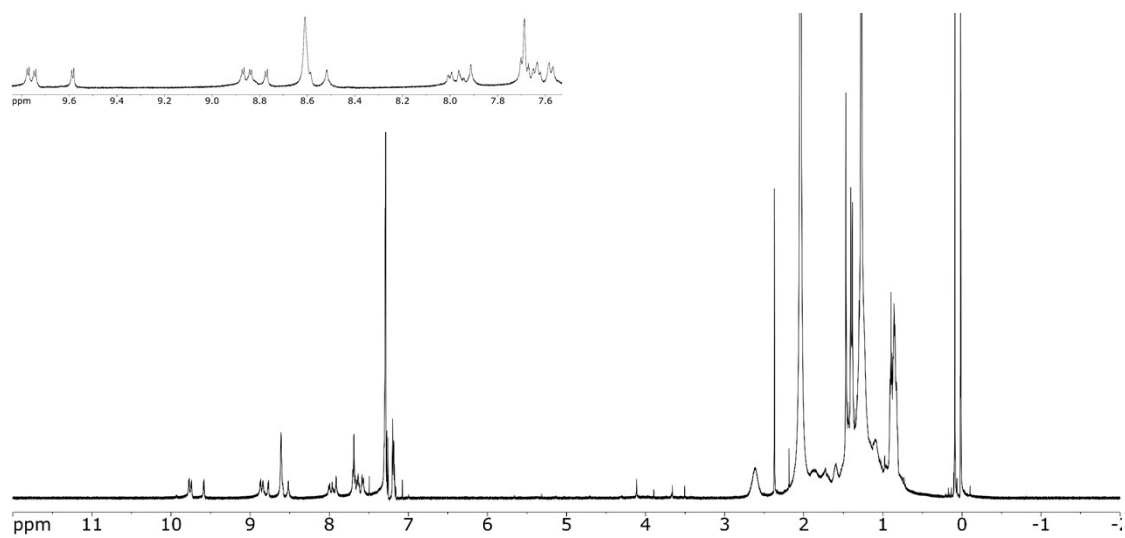


**ZnCP** (5.0 mg, 1.9  $\mu\text{mol}$ ) and **T<sub>3</sub>** ( $n = 8$ ) (0.624 mg, 0.637  $\mu\text{mol}$ ) from a stock solution was added to a round bottom flask, after which a solvent was removed under vacuum. These compounds were dissolved in  $\text{CH}_2\text{Cl}_2$  (500 mL) followed by slow addition of  $\text{CuCl}$  (680 mg, 6.87 mmol) and  $\text{TMEDA}$  (790  $\mu\text{L}$ , 5.15 mmol). The solution was stirred at r.t. for 45 min under air. Another tubulation of **ZnCP** (11.0 mg, 4.2  $\mu\text{mol}$ ) was conducted under the same condition (concentration, equivalents of reagents to **ZnCP**, and temperature) as the above. The combined reaction mixture was carefully extracted by  $\text{NH}_4\text{Cl}$  aq. with  $\text{TMEDA}$  once and  $\text{NH}_4\text{Cl}$  aq. once, water once, and brine once to remove copper residue from the solution. The organic layer was dried by  $\text{Na}_2\text{SO}_4$ , then the solvent was removed *in vacuo*. The residue was filtered through silica gel ( $\text{CH}_2\text{Cl}_2$  /  $\text{EtOAc}$  / pyridine = 100 / 10 / 1, v / v / v) to remove **T<sub>3</sub>** ( $n = 8$ ) from **ZnCP<sub>3</sub>**. The crude product was purified by GPC (toluene / pyridine = 100 / 3, v / v) to afford **ZnCP<sub>3</sub>** (2.4 mg, 10%).

*HR-MS(MALDI-TOF MS)* ( $m/z$ ) 7846.62 ( $[\text{ZnCP}_3 + \text{H}]^+$ ,  $\text{C}_{492}\text{H}_{437}\text{N}_{24}\text{O}_{48}\text{Zn}_6$ , calcd. 7844.8304)

$^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3$  (+ 1 % pyridine- $d_5$ ), r. t.)  $\delta$ : 9.76 (d,  $J = 4.1$  Hz, 8H,  $\beta\text{-H}$ ), 9.73 (d,  $J = 4.3$  Hz, 8H,  $\beta\text{-H}$ ), 9.57, (d,  $J = 4.3$  Hz, 8H,  $\beta\text{-H}$ ), 8.86 (d,  $J = 4.3$  Hz, 8H,  $\beta\text{-H}$ ), 8.83 (d,  $J = 4.1$  Hz, 8H,  $\beta\text{-H}$ ), 8.76 (d,  $J = 4.2$  Hz, 8H,  $\beta\text{-H}$ ), 8.57 (s, 4H, Ar-H), 8.50 (s, 8H, Ar-H), 7.98 (d,  $J = 7.2$ , 8H, Ar-H), 7.95-7.92 (a double doublet peak and a singlet peak overlapped, 8H, Ar-H), 7.90 (s, 8H, Ar-H), 7.69-7.65 (two doublet peaks and residual peaks overlapped, 12H, Ar-H), 7.63-7.60 (two doublet peaks overlapped, 12H, Ar-H), 7.57-7.50 (two doublet peaks overlapped, 12H, Ar-H), 4.09 (s, 4H), 2.67-2.53(m, methylene protons), 1.92-1.53(m, methylene protons), 1.45 (s,  $-\text{CH}_2\text{CH}_2\text{COOC}(\text{CH}_3)_3$ ), 1.39 (s,  $-\text{CH}_2\text{CH}_2\text{COOC}(\text{CH}_3)_3$ ), 1.3 (s,  $-\text{CH}_2\text{CH}_2\text{COOC}(\text{CH}_3)_3$ ).

$^1\text{H}$  NMR spectrum of  $\text{ZnCP}_3$



### 3. Complexation study

#### 3.1 2:1 complexation between ZnCP and $T_2$ ( $n = x$ ) by NMR titrations

Scheme S1. Complexation between **ZnCP'** and  $T_2$  ( $n = x$ ).

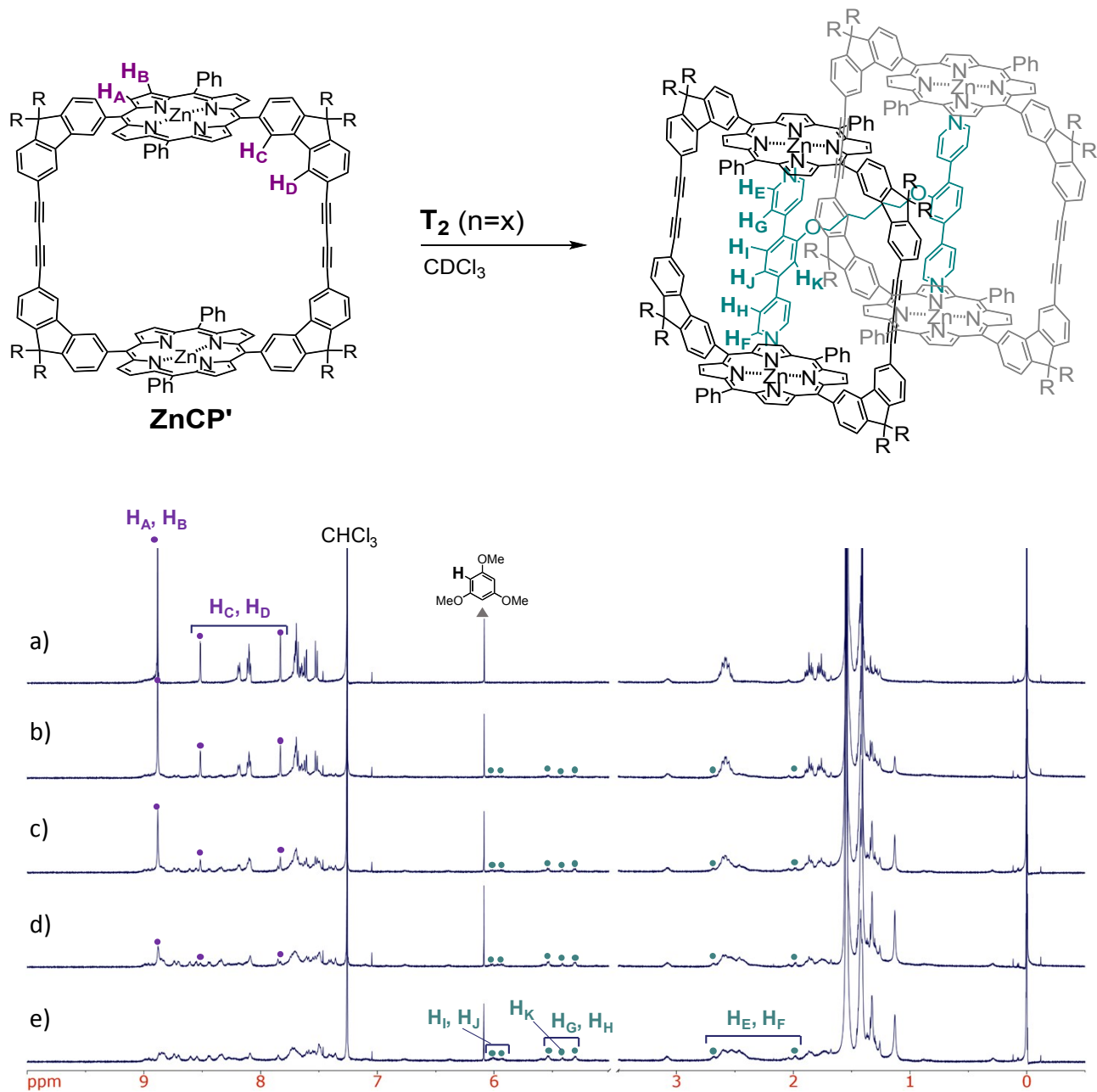


Figure S1.  $^1H$  NMR titration of  $T_2$  ( $n = 4$ ) to **ZnCP'** in  $CDCl_3$ .

a) 0 equiv., b) 0.125 equiv., c) 0.250 equiv., d) 0.375 equiv., e) 0.500 equiv.

1,3,5-trimethoxybenzene (gray) was added as an internal standard.

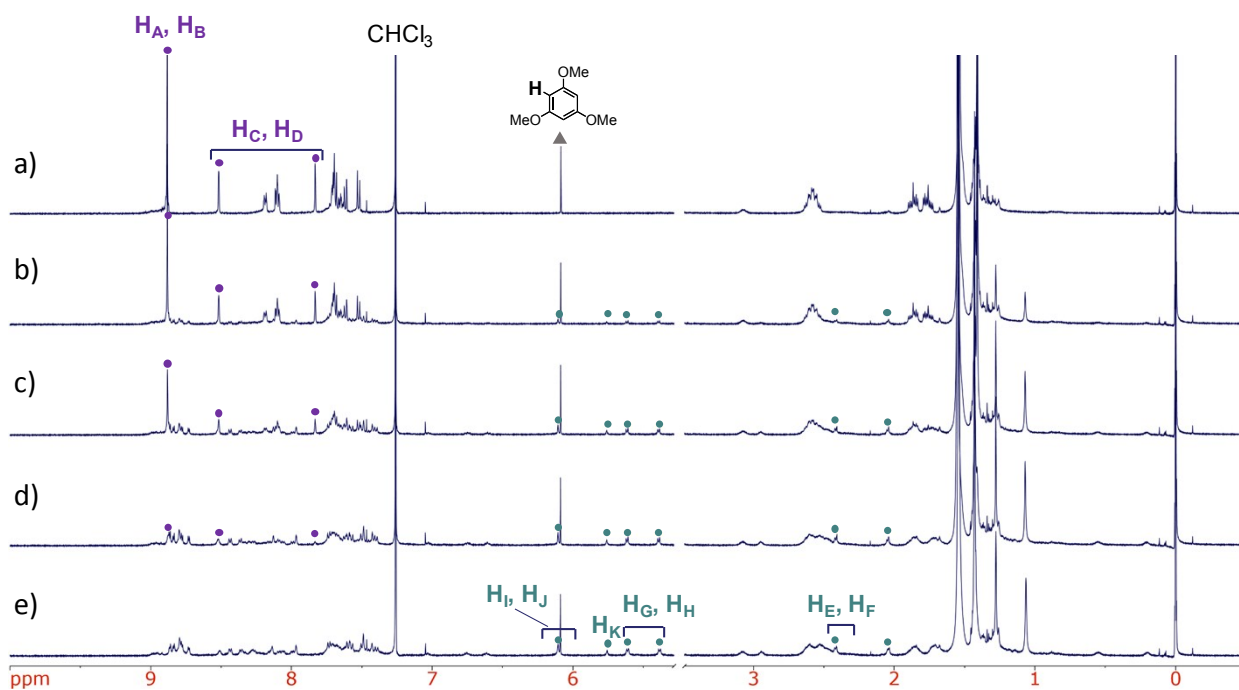


Figure S2.  $^1\text{H}$  NMR titration of  $\text{T}_2$  ( $n = 6$ ) to  $\text{ZnCP}'$  in  $\text{CDCl}_3$ .

a) 0 equiv., b) 0.125 equiv., c) 0.250 equiv., d) 0.375 equiv., e) 0.500 equiv.

1,3,5-trimethoxybenzene (gray) was added as an internal standard.

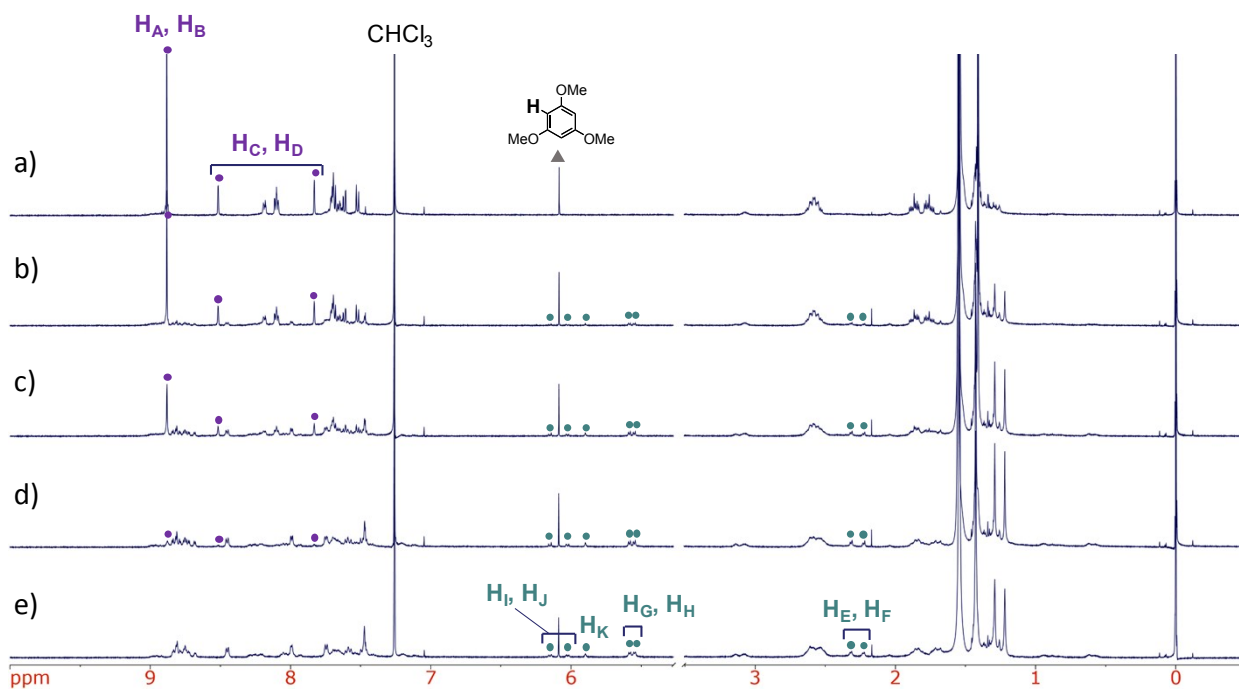


Figure S3.  $^1\text{H}$  NMR titration of  $\text{T}_2$  ( $n = 8$ ) to  $\text{ZnCP}'$  in  $\text{CDCl}_3$ .

a) 0 equiv., b) 0.125 equiv., c) 0.250 equiv., d) 0.375 equiv., e) 0.500 equiv.

1,3,5-trimethoxybenzene (gray) was added as an internal standard.

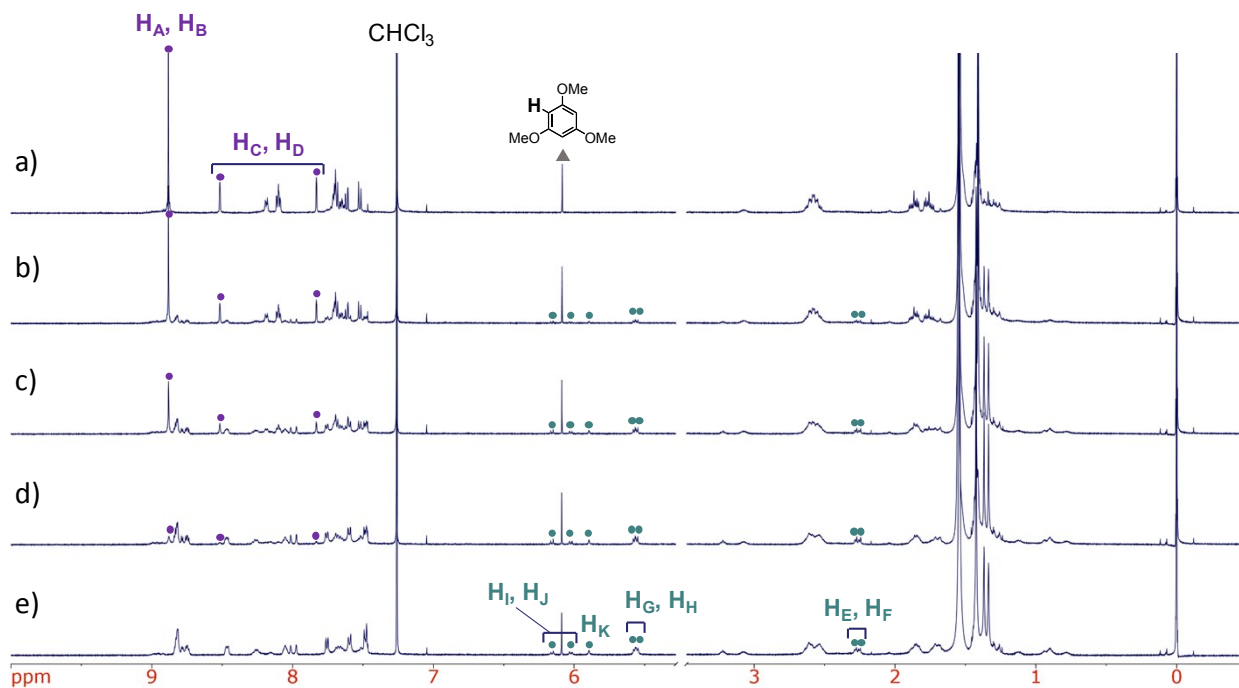


Figure S4.  $^1\text{H}$  NMR titration of  $\text{T}_2$  ( $n = 12$ ) to  $\text{ZnCP}'$  in  $\text{CDCl}_3$ .

a) 0 equiv., b) 0.125 equiv., c) 0.250 equiv., d) 0.375 equiv., e) 0.500 equiv.

1,3,5-trimethoxybenzene (gray) was added as an internal standard.

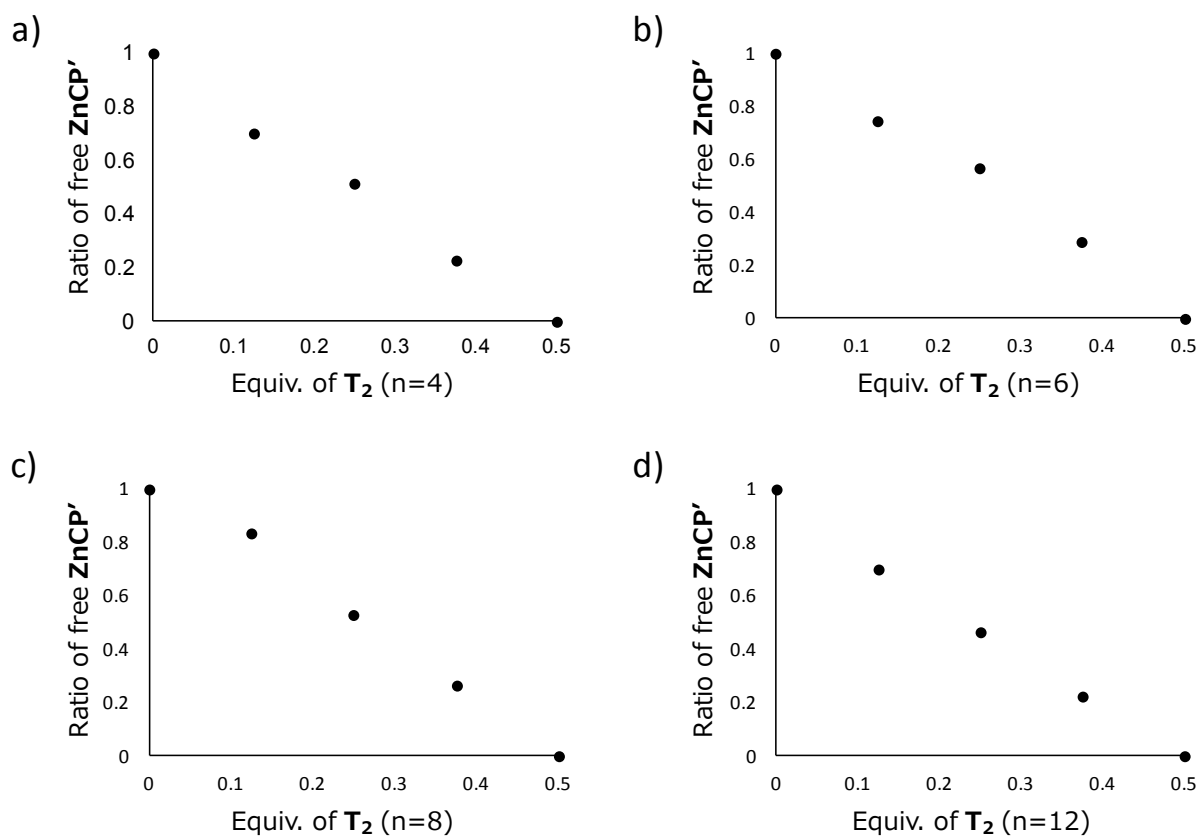


Figure S5. Conversion ratio of free  $\text{ZnCP}'$  in  $^1\text{H}$  NMR titration of  $\text{T}_2$  ( $n = x$ ) to  $\text{ZnCP}'$  in  $\text{CDCl}_3$ .

a)  $\text{T}_2$  ( $n = 4$ ), b)  $\text{T}_2$  ( $n = 6$ ), c)  $\text{T}_2$  ( $n = 8$ ), d)  $\text{T}_2$  ( $n = 12$ )

### 3.2 3:1 complexation between ZnCP and $T_3$ ( $n = 8$ ) by NMR titrations

Scheme S2. Complexation between **ZnCP'** and  $T_3$  ( $n = 8$ ).

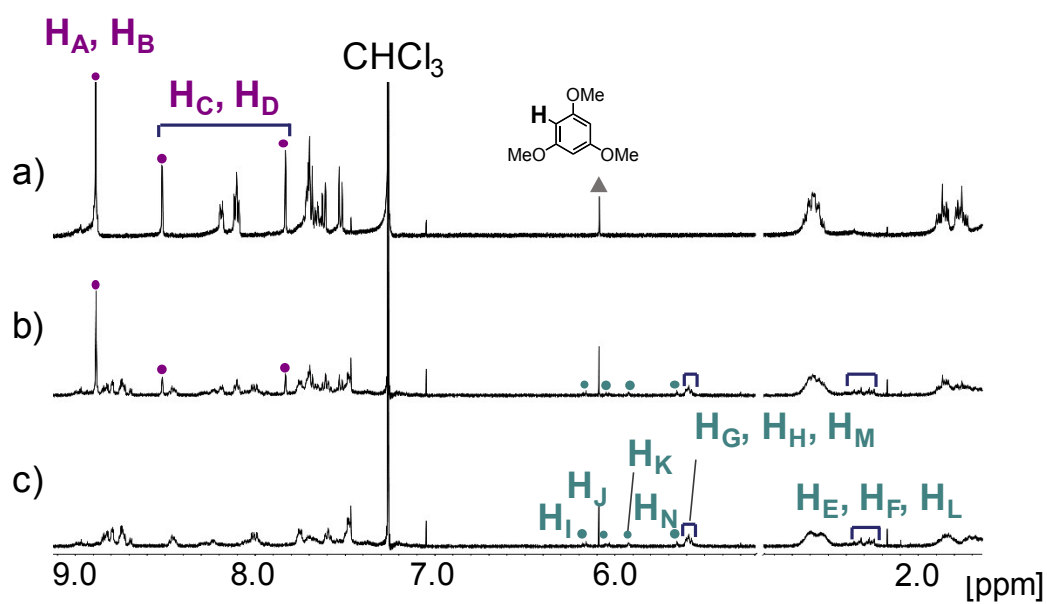
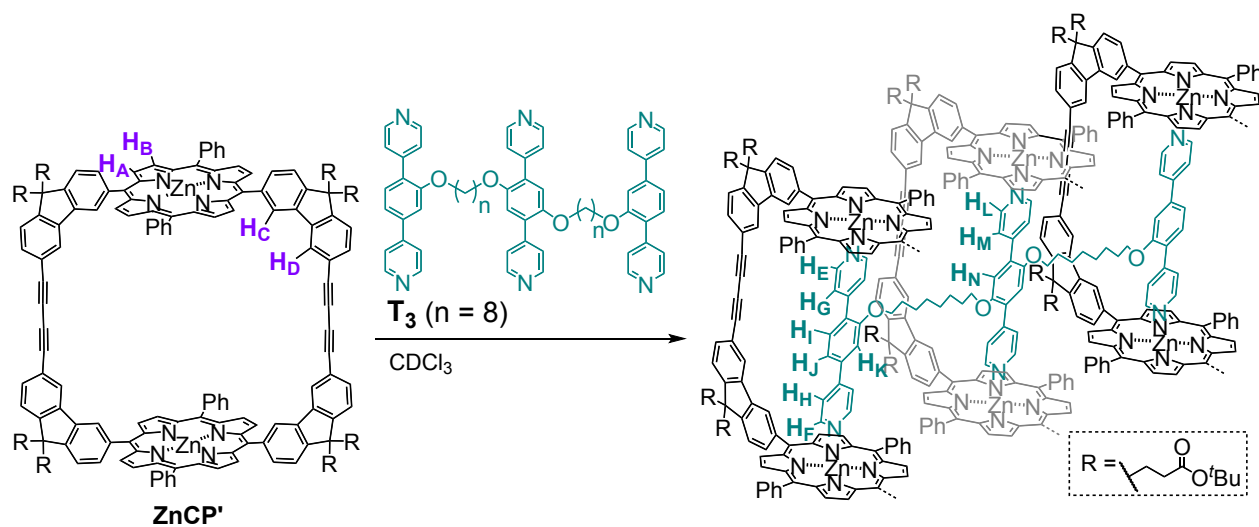


Figure S6.  $^1H$  NMR titration of  $T_3$  ( $n = 8$ ) to **ZnCP'** in  $CDCl_3$ .

a) 0 equiv., b) 0.13 equiv., c) 0.33 equiv.

1,3,5-trimethoxybenzene (gray) was added as an internal standard.

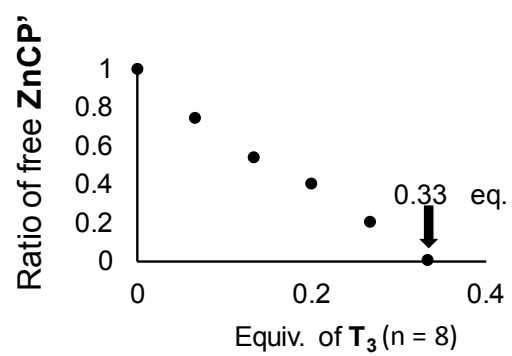


Figure S7. Conversion of **ZnCP'** in <sup>1</sup>H NMR titration of T<sub>3</sub> (n = 8) to **ZnCP'** in CDCl<sub>3</sub>.

#### 4. Investigation of Template Synthesis

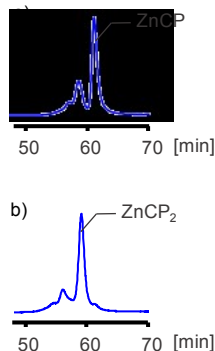
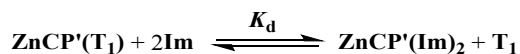


Figure S8. SEC chromatograms for synthesis of **ZnCP<sub>2</sub>** in the presence of templates (Eluent : THF, detection : 420 nm).  
Templates: (a) **T<sub>2</sub>'** (n = 6) and (b) **T<sub>2</sub>** (n = 6)

#### 5. Binding study

The simple two-state 2-site model was applied to indirectly determine the binding constant  $K_f$  for **ZnCP'(T<sub>1</sub>)**. The titration of **ZnCP'(T<sub>1</sub>)** with imidazole (**Im**) was analyzed assuming that each pyridyl unit of **T<sub>1</sub>** binds independently, so that the equilibrium can be treated as the displacement of a 2-site ligand **T<sub>1</sub>** from a 2-site receptor **ZnCP'**.



$$K_d = \frac{[\text{ZnCP}'(\text{Im})_2][\text{T}_1]}{[\text{ZnCP}'(\text{T}_1)][\text{Im}]^2} \quad (\text{S1})$$

The denaturation equilibrium constant,  $K_d$ , defined by equation (S1) was determined by fitting the binding isotherm to equation (S2).

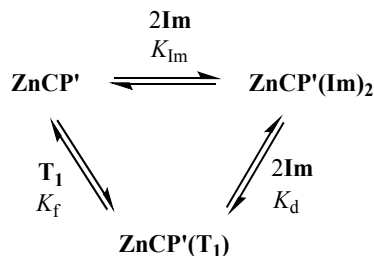
$$\Delta A = \frac{L(-K_d[\text{Im}]^2 + \sqrt{K_d^2[\text{Im}]^2 + 4K_d[\text{Im}]^2[\text{ZnCP}'(\text{T}_1)]_0})}{2[\text{ZnCP}'(\text{T}_1)]_0} \quad (\text{S2})$$

where  $\Delta A$  is the change of absorption at any point in the titration,  $L$  is  $\Delta A$  at 100 % complexation,  $[\text{Im}]$  is the total concentration of imidazole at each point in the titration and  $[\text{ZnCP}'(\text{T}_1)]_0$  is the initial concentration of **ZnCP'(T<sub>1</sub>)**. To determine the 1 : 1 formation constant between **ZnCP'** and **T<sub>1</sub>**, denoted  $K_f$ , the following thermodynamic cycle was used:

**Comment [A]:** K の後の下付き文字は非斜体  
全体的にチェックを

**Comment [A]:** + - × は全て両側にスペースを入れる

**Comment [A]:** A の前に文字化け  
K<sub>d</sub> が定義されていない  
ZnCP ' (T<sub>1</sub>) の ' が全角?



**Comment [A]:** The xxx equilibrium constant, Kf can be....

The addition of excess imidazole, **Im** to **ZnCP'( $T_1$ )** results in the displacement of the ligand **T<sub>1</sub>** from **ZnCP'**, to generate **ZnCP'(**Im**)<sub>2</sub>**. The formation constant  $K_f$  can be calculated using equation (S3),

$$K_f = \frac{K_{\text{Im}}^2}{K_d} \quad (\text{S3})$$

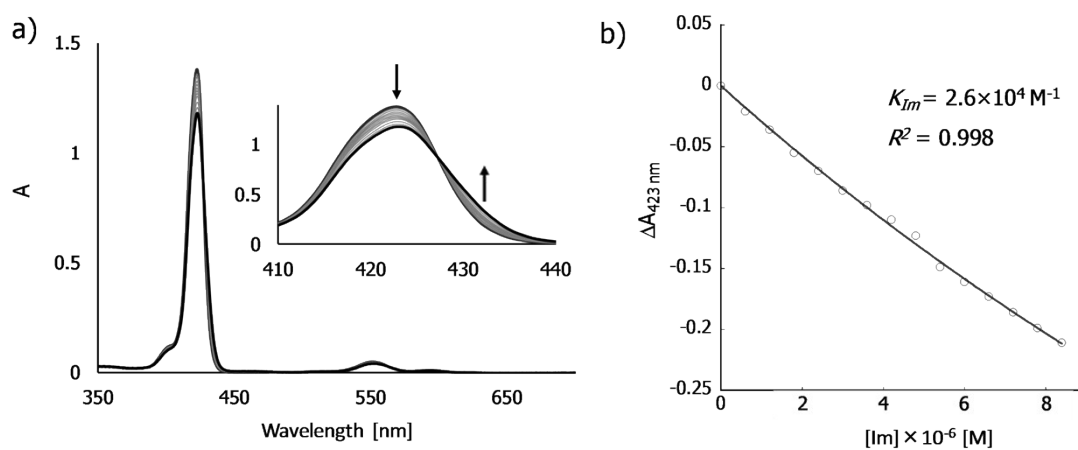
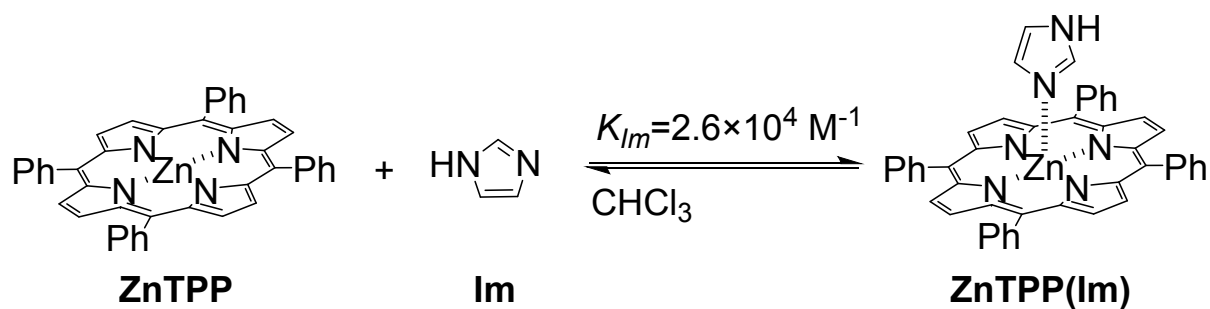
The binding constant,  $K_{\text{Im}}$  was determined by fitting the binding isotherm to equation (S4).

$$\Delta A = \frac{L(1 + K_{\text{Im}}[\text{Im}] + K_{\text{Im}}[\text{ZnCP}'(\text{T}_1)]_0 - \sqrt{L^2(K_{\text{Im}}[\text{Im}] + K_{\text{Im}}[\text{ZnCP}'(\text{T}_1)]_0)^2 - 4K_{\text{Im}}^2[\text{Im}][\text{ZnCP}'(\text{T}_1)]_0L^2})}{2K_{\text{Im}}[\text{Im}]} \quad (\text{S4})$$

The value of  $K_{\text{Im}}$  obtained from fitting the binding isotherms (S4) at 423 nm to a 1 : 1 binding model for **ZnTPP** and **Im** (Scheme S3, Figure S9) instead of **ZnCP'** and **Im** is  $K_{\text{Im}} = 2.6 \times 10^4 \text{ M}^{-1}$ .

In the case of denaturation equilibrium constant  $K_d$  obtained from fitting the binding isotherms (S2) at 436 nm is  $K_d = 3.9 \text{ M}^{-1}$  (Scheme S4, Figure S10). The value of  $K_f$  obtained from equation (S3) was  $1.7 \times 10^8 \text{ M}^{-1}$ .

Scheme S3. Complexation of **ZnTPP** with imidazole (**Im**) in  $\text{CHCl}_3$



Scheme S4. Displacement of the ligand **T<sub>1</sub>** from **ZnCP'** by imidazole in CHCl<sub>3</sub>

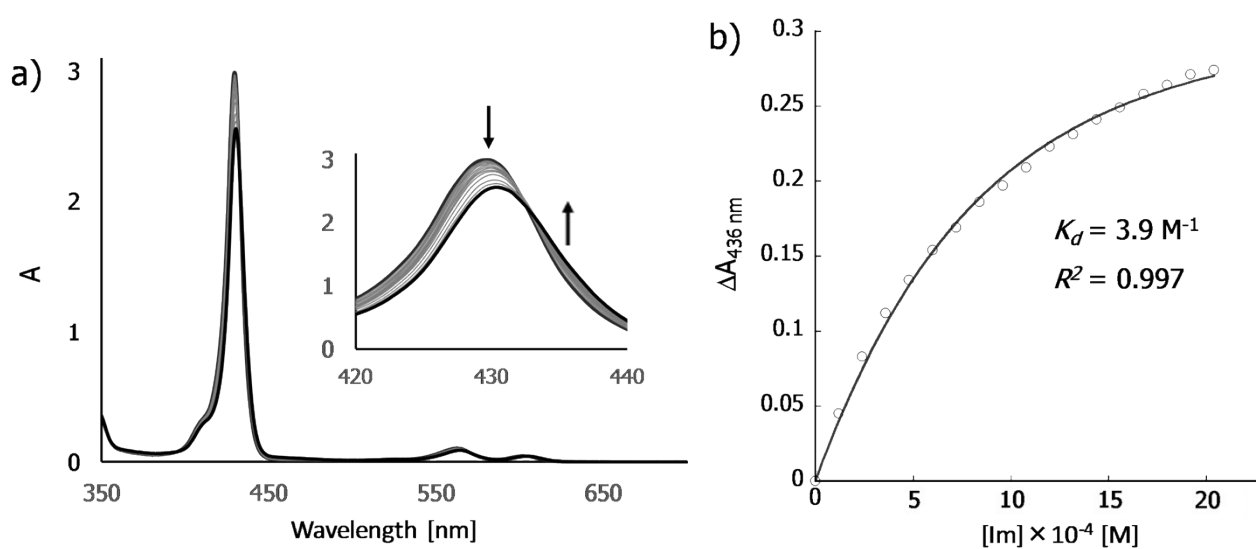
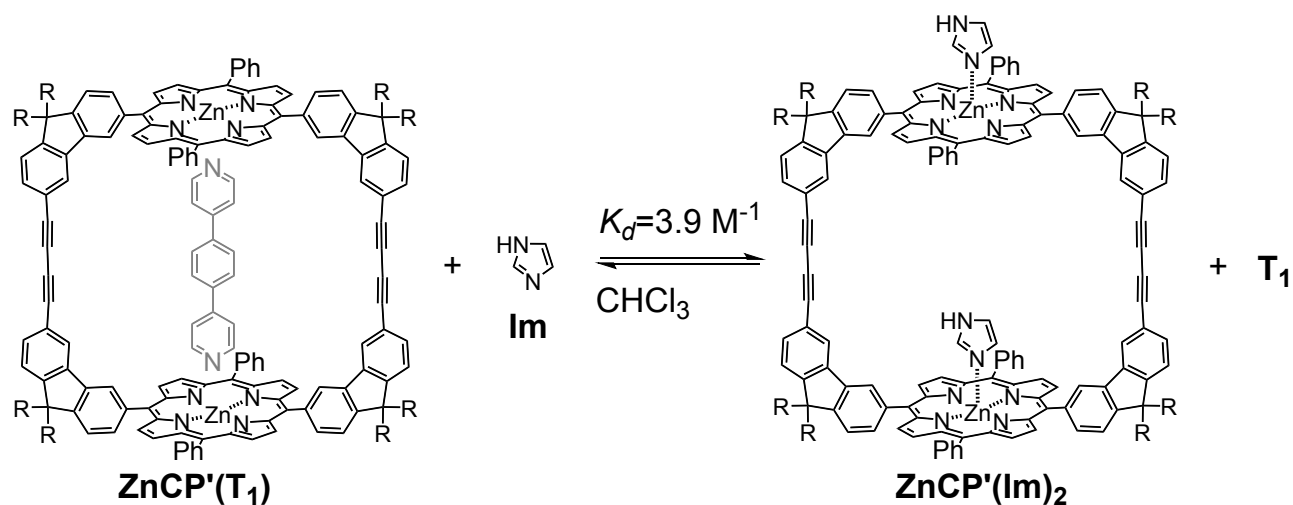


Figure S10. a) UV titration of imidazole, **Im** with **ZnCP'** ( $3.0 \times 10^{-6} \text{ M}$ , r. t., CHCl<sub>3</sub>). Arrows indicate areas of increasing and decreasing absorption during the titration. b) The data at 436 nm were fitted to a 1 : 1 binding isotherm (S2) to give a binding constant of  $K_d = 3.9 \text{ M}^{-1}$ .

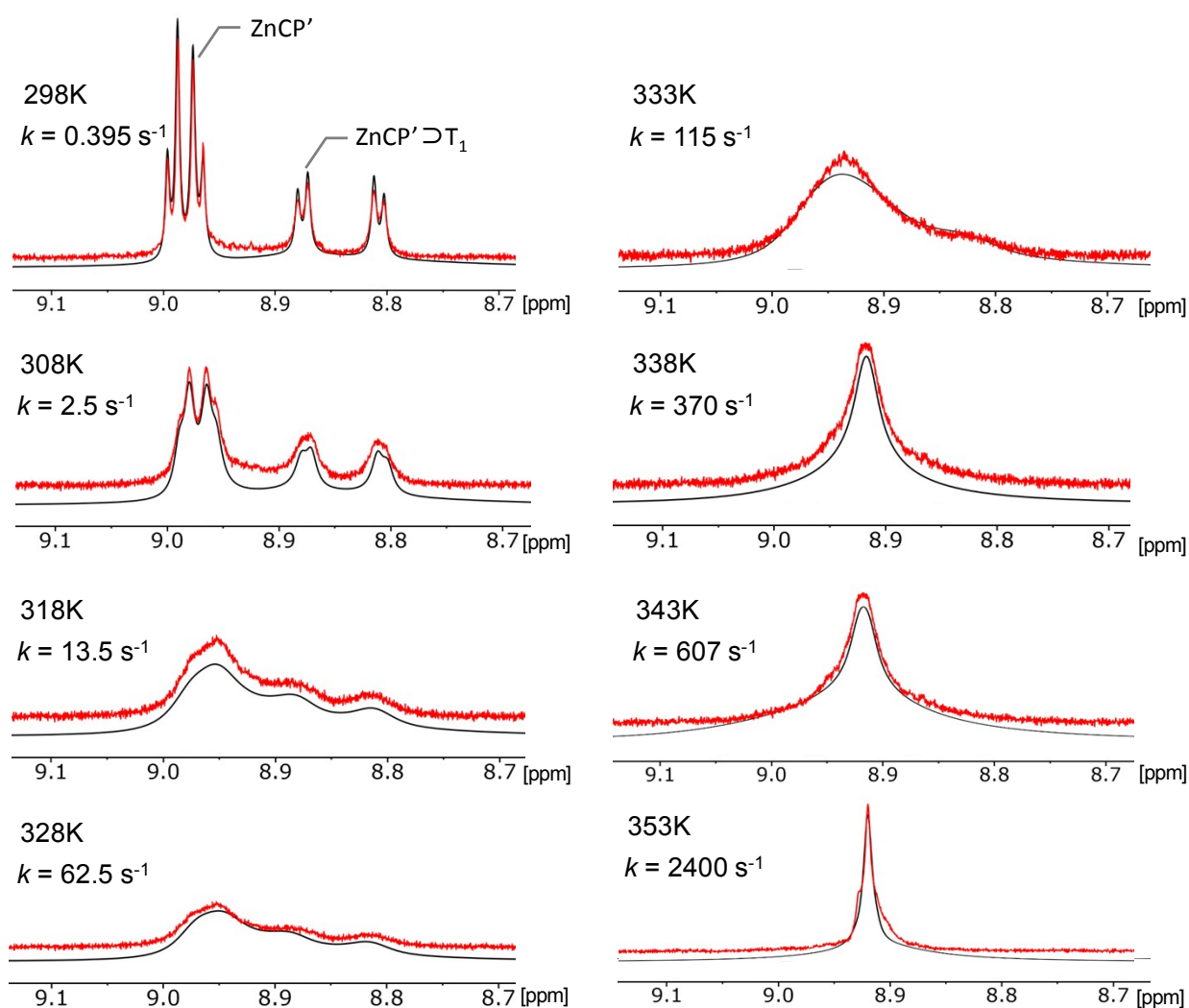


Figure S10. Experimental (red) and theoretical (black) splitting of  $\beta\text{-H}$  of  $\text{ZnCP}'$  and  $\text{ZnCP}' \rightleftharpoons \text{T}_1$  in  $^1\text{H}$  VT-NMR in toluene- $d_8$ .  $k$  means rate constants of ligand exchange at different temperatures.

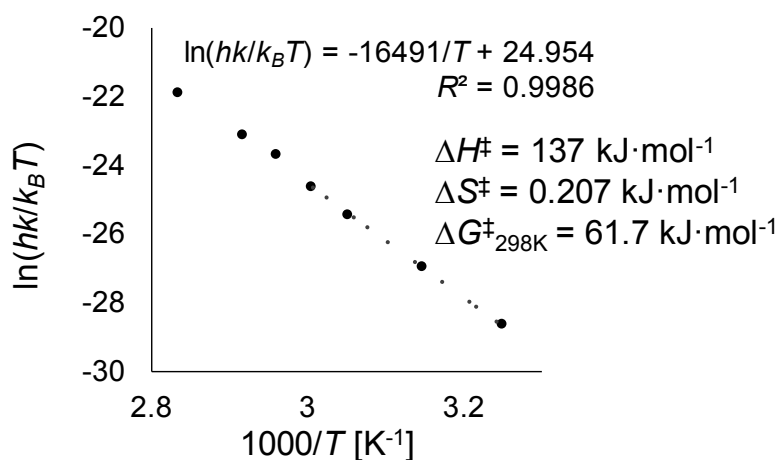


Figure S11. Eyring plot for the ligand exchange reaction between  $\text{ZnCP}'$  and  $\text{ZnCP}' \rightleftharpoons \text{T}_1$ .

## 6. Computational Study

### Method

The optimized geometries of **ZnCP** and **ZnCP<sub>2</sub>** were calculated at the B3LYP level of theory employing the 6-31G(d,p) for carbons, hydrogens, and oxygens and 6-31G+(d) for nitrogens and Lanl2DZ for zincs, neglecting ester groups and terminal alkynes of **ZnCP** and ester groups of **ZnCP<sub>2</sub>** in the calculation. Stable structures of **T<sub>2</sub>** (n = x) were calculated at the B3LYP level of theory using the 6-31G(d) for carbons, hydrogens and nitrogens. All DFT calculations were carried out using Gaussian 09.<sup>7</sup>

### DFT optimized structure

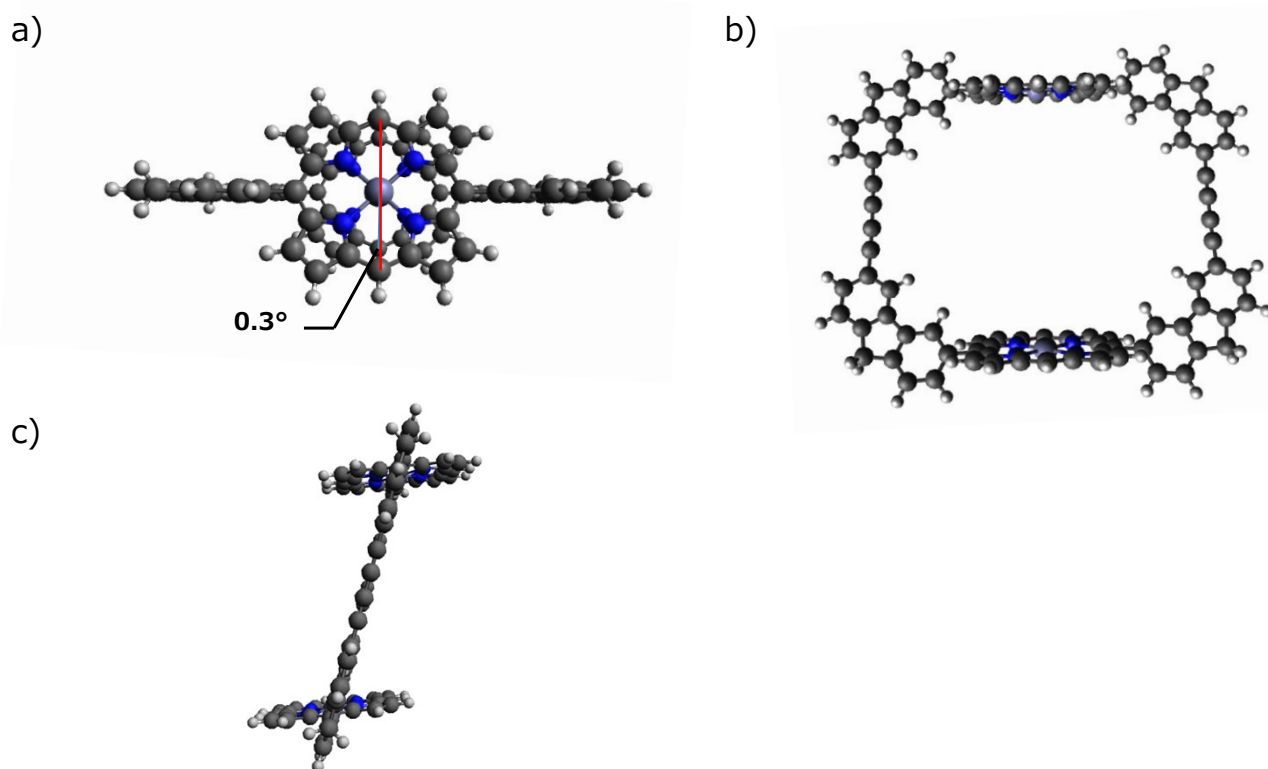


Figure S12. The optimized structure of **ZnCP**.

a) top view, b) front view, c) side view

C = dark gray; H = light gray; N = blue; Zn = purple. Hydrogen atoms are used in the calculation instead of allyl ester groups of fluorenes for simplicity.

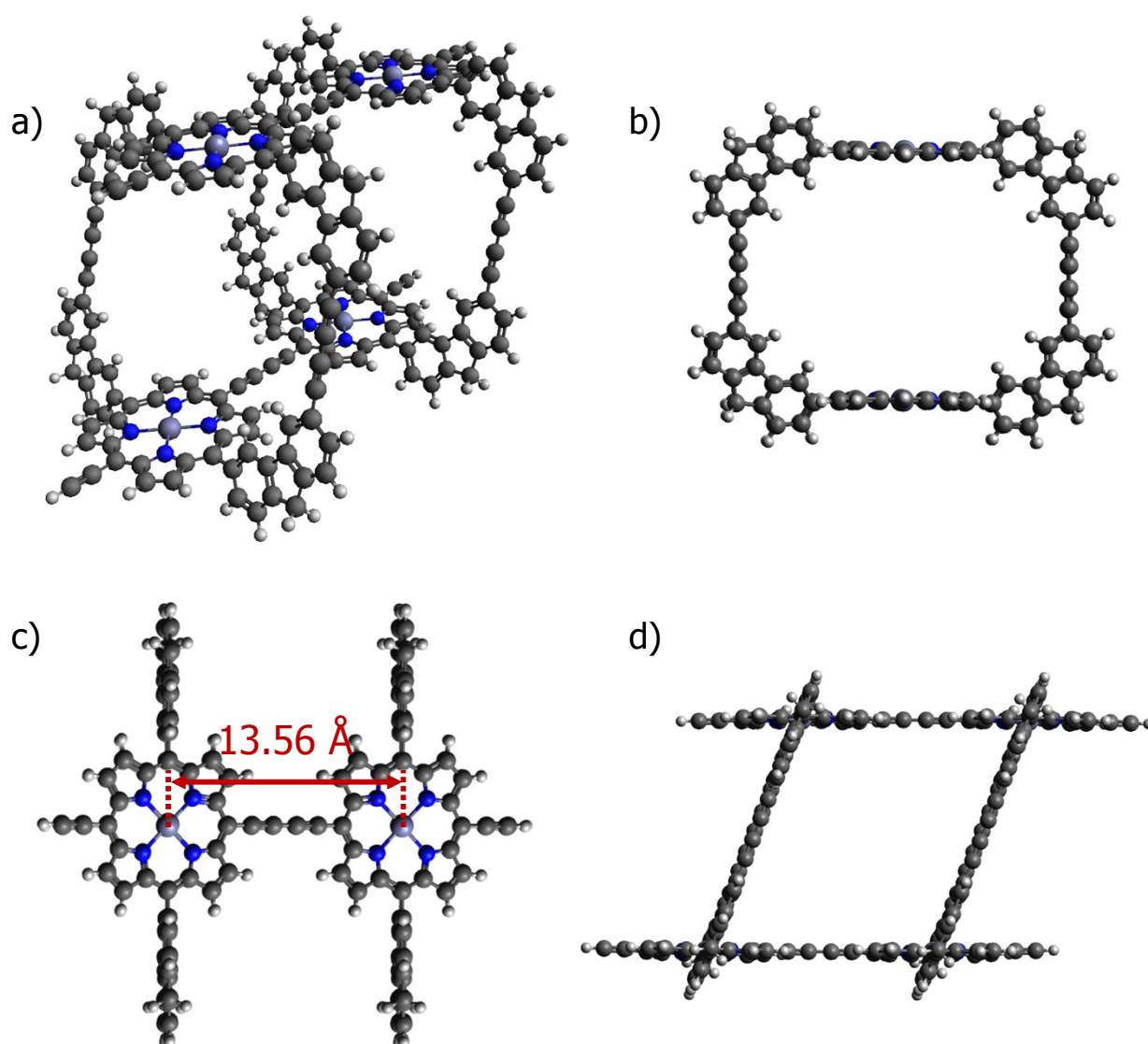


Figure S13. The optimized structure of **ZnCP<sub>2</sub>**.

a) slanting view, b) front view, c) top view, d) side view

Arrow indicates the distance between two zincs. C = dark gray; H = light gray; N = blue; Zn = purple. Hydrogen atoms are used in the calculation instead of allyl ester groups of fluorenes for simplicity.

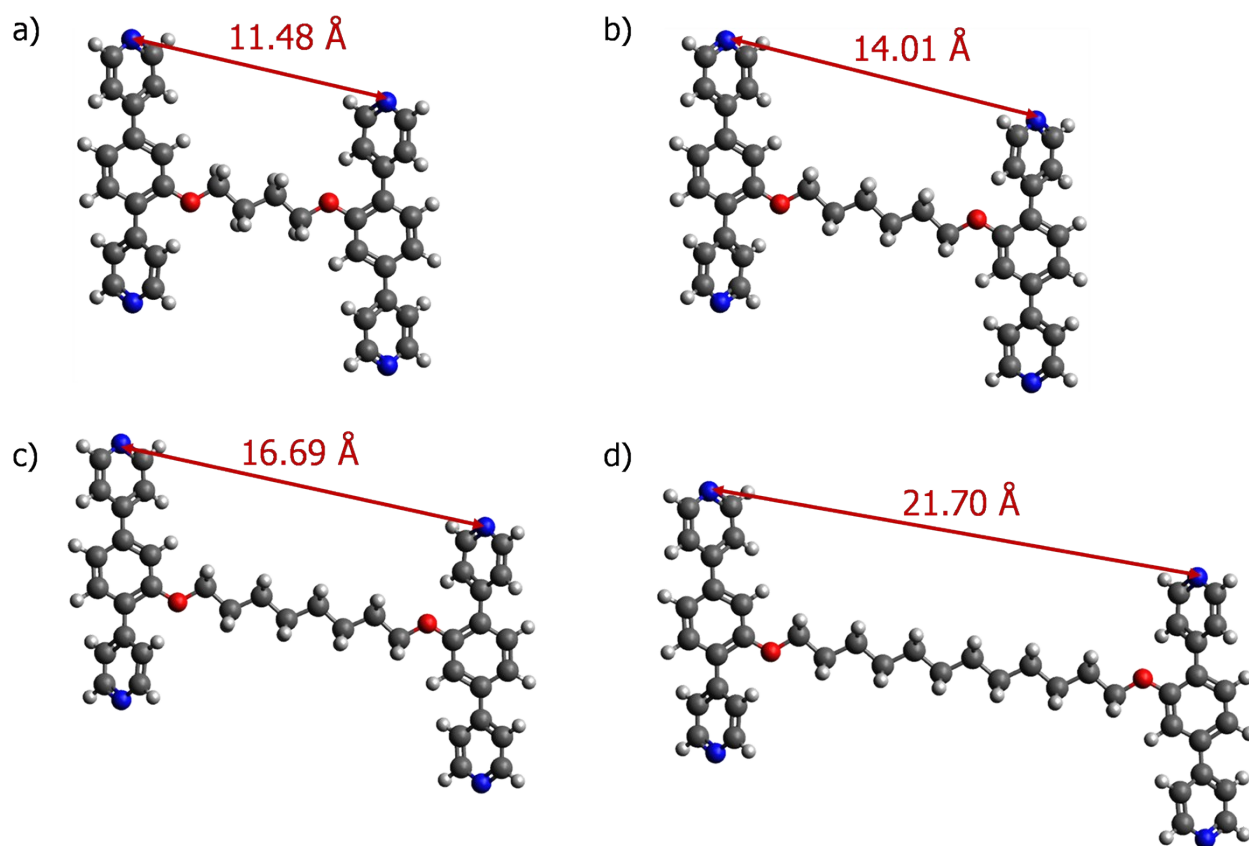


Figure S14. The optimized structures of  $T_2$  ( $n = x$ ).

a)  $T_2$  ( $n = 4$ ), b)  $T_2$  ( $n = 6$ ), c)  $T_2$  ( $n = 8$ ), d)  $T_2$  ( $n = 12$ )

Arrow indicates the distance between two nitrogens. C = dark gray; H = light gray; N = blue; O = red.

## Computed Cartesian coordinates

### 1) ZnCP

B3LYP/6-31G(d) for C, H, O; 6-31G+(d) for N; Lanl2DZ for Zn

|   |          |          |         |
|---|----------|----------|---------|
| C | -2.48706 | -6.58334 | 3.46489 |
| C | -3.34218 | -6.17822 | 2.48554 |
| C | -2.67548 | -6.40441 | 1.22766 |
| N | -1.42529 | -6.93109 | 1.44569 |
| C | -1.28178 | -7.04809 | 2.80963 |
| C | 3.09207  | -8.84282 | 2.51074 |
| C | 2.2038   | -8.49658 | 3.48333 |
| C | 1.00563  | -8.03042 | 2.81643 |
| N | 1.18352  | -8.09328 | 1.45343 |
| C | 2.44824  | -8.5891  | 1.24632 |
| C | -0.14964 | -7.56128 | 3.47015 |

|   |          |          |          |
|---|----------|----------|----------|
| C | 2.23794  | -8.47745 | -3.48734 |
| C | 3.11702  | -8.82813 | -2.508   |
| C | 2.46064  | -8.58181 | -1.2486  |
| N | 1.19767  | -8.08564 | -1.46539 |
| C | 1.03282  | -8.01599 | -2.82968 |
| C | -3.32024 | -6.17023 | -2.53071 |
| C | -2.45614 | -6.57124 | -3.50383 |
| C | -1.25578 | -7.03611 | -2.83967 |
| N | -1.41146 | -6.92391 | -1.47665 |
| C | -2.66424 | -6.39951 | -1.26781 |
| C | -0.11698 | -7.54525 | -3.49204 |
| C | -2.2038  | 8.49658  | 3.48333  |
| C | -3.09207 | 8.84282  | 2.51074  |
| C | -2.44824 | 8.5891   | 1.24632  |
| N | -1.18352 | 8.09328  | 1.45343  |
| C | -1.00563 | 8.03042  | 2.81643  |
| C | 3.34218  | 6.17822  | 2.48554  |
| C | 2.48706  | 6.58334  | 3.46489  |
| C | 1.28178  | 7.04809  | 2.80963  |
| N | 1.42529  | 6.93109  | 1.44569  |
| C | 2.67548  | 6.40441  | 1.22766  |
| C | 0.14964  | 7.56128  | 3.47015  |
| C | 2.45614  | 6.57124  | -3.50383 |
| C | 3.32024  | 6.17023  | -2.53071 |
| C | 2.66424  | 6.39951  | -1.26781 |
| N | 1.41146  | 6.92391  | -1.47665 |
| C | 1.25578  | 7.03611  | -2.83967 |
| C | -3.11702 | 8.82813  | -2.508   |
| C | -2.23794 | 8.47745  | -3.48734 |
| C | -1.03282 | 8.01599  | -2.82968 |
| N | -1.19767 | 8.08564  | -1.46539 |
| C | -2.46064 | 8.58181  | -1.2486  |
| C | 0.11698  | 7.54525  | -3.49204 |
| C | -0.13222 | -7.60479 | -4.989   |
| C | -0.11629 | -6.43089 | -5.7578  |
| C | -0.13578 | -6.51912 | -7.15066 |
| C | -0.17011 | -7.77497 | -7.79038 |
| C | -0.18466 | -8.9416  | -7.03598 |

|   |          |          |           |
|---|----------|----------|-----------|
| C | -0.16587 | -8.85073 | -5.64169  |
| C | -0.12274 | -5.47112 | -8.18251  |
| C | -0.15117 | -6.09088 | -9.45049  |
| C | -0.09032 | -4.08331 | -8.07129  |
| C | -0.08615 | -3.29972 | -9.24401  |
| C | -0.11539 | -3.93353 | -10.50732 |
| C | -0.14786 | -5.32337 | -10.61002 |
| C | -0.05209 | -1.89336 | -9.1722   |
| C | -0.0203  | -0.6442  | -9.14855  |
| C | 0.13222  | 7.60479  | -4.989    |
| C | 0.16587  | 8.85073  | -5.64169  |
| C | 0.18466  | 8.9416   | -7.03598  |
| C | 0.17011  | 7.77497  | -7.79038  |
| C | 0.13578  | 6.51912  | -7.15066  |
| C | 0.11629  | 6.43089  | -5.7578   |
| C | 0.15117  | 6.09088  | -9.45049  |
| C | 0.12274  | 5.47112  | -8.18251  |
| C | 0.14786  | 5.32337  | -10.61002 |
| C | 0.11539  | 3.93353  | -10.50732 |
| C | 0.08615  | 3.29972  | -9.24401  |
| C | 0.09032  | 4.08331  | -8.07129  |
| C | 0.05209  | 1.89336  | -9.1722   |
| C | 0.0203   | 0.6442   | -9.14855  |
| C | 0.18069  | 7.60265  | 4.99514   |
| C | 0.15954  | 6.42887  | 5.76393   |
| C | 0.1901   | 6.51693  | 7.15661   |
| C | 0.24284  | 7.77228  | 7.79602   |
| C | 0.26366  | 8.93875  | 7.04158   |
| C | 0.23198  | 8.84813  | 5.6475    |
| C | 0.17279  | 5.46908  | 8.18848   |
| C | 0.2184   | 6.08843  | 9.45616   |
| C | 0.12192  | 4.08181  | 8.07729   |
| C | 0.11795  | 3.29799  | 9.24988   |
| C | 0.16446  | 3.93139  | 10.5129   |
| C | 0.21435  | 5.32071  | 10.61553  |
| C | 0.06773  | 1.89132  | 9.17869   |
| C | 0.02298  | 0.64397  | 9.15642   |
| C | -0.18069 | -7.60265 | 4.99514   |

|   |          |          |           |
|---|----------|----------|-----------|
| C | -0.23198 | -8.84813 | 5.6475    |
| C | -0.26366 | -8.93875 | 7.04158   |
| C | -0.24284 | -7.77228 | 7.79602   |
| C | -0.1901  | -6.51693 | 7.15661   |
| C | -0.15954 | -6.42887 | 5.76393   |
| C | -0.2184  | -6.08843 | 9.45616   |
| C | -0.17279 | -5.46908 | 8.18848   |
| C | -0.21435 | -5.32071 | 10.61553  |
| C | -0.16446 | -3.93139 | 10.5129   |
| C | -0.11795 | -3.29799 | 9.24988   |
| C | -0.12192 | -4.08181 | 8.07729   |
| C | -0.06773 | -1.89132 | 9.17869   |
| C | -0.02298 | -0.64397 | 9.15642   |
| H | -2.65171 | -6.5773  | 4.53187   |
| H | -4.33943 | -5.77292 | 2.59562   |
| H | 4.09583  | -9.22906 | 2.63001   |
| H | 2.34263  | -8.53847 | 4.5532    |
| H | 2.38729  | -8.51364 | -4.556    |
| H | 4.12227  | -9.21283 | -2.61957  |
| H | -4.31722 | -5.7664  | -2.64841  |
| H | -2.61198 | -6.56271 | -4.57208  |
| H | -2.34263 | 8.53847  | 4.5532    |
| H | -4.09583 | 9.22906  | 2.63001   |
| H | 4.33943  | 5.77292  | 2.59562   |
| H | 2.65171  | 6.5773   | 4.53187   |
| H | 2.61198  | 6.56271  | -4.57208  |
| H | 4.31722  | 5.7664   | -2.64841  |
| H | -4.12227 | 9.21283  | -2.61957  |
| H | -2.38729 | 8.51364  | -4.556    |
| H | -0.08131 | -5.46654 | -5.25957  |
| H | -0.21503 | -9.91625 | -7.51591  |
| H | -0.18509 | -9.75559 | -5.04235  |
| H | -0.06775 | -3.59371 | -7.10313  |
| H | -0.1121  | -3.31802 | -11.40051 |
| H | -0.17022 | -5.79297 | -11.58956 |
| H | 0.18509  | 9.75559  | -5.04235  |
| H | 0.21503  | 9.91625  | -7.51591  |
| H | 0.08131  | 5.46654  | -5.25957  |

|    |          |          |           |
|----|----------|----------|-----------|
| H  | 0.17022  | 5.79297  | -11.58956 |
| H  | 0.1121   | 3.31802  | -11.40051 |
| H  | 0.06775  | 3.59371  | -7.10313  |
| H  | 0.11216  | 5.46475  | 5.26623   |
| H  | 0.30857  | 9.91292  | 7.52134   |
| H  | 0.25483  | 9.75279  | 5.04801   |
| H  | 0.08432  | 3.59312  | 7.10911   |
| H  | 0.1603   | 3.31589  | 11.40608  |
| H  | 0.24919  | 5.78992  | 11.5949   |
| H  | -0.25483 | -9.75279 | 5.04801   |
| H  | -0.30857 | -9.91292 | 7.52134   |
| H  | -0.11216 | -5.46475 | 5.26623   |
| H  | -0.24919 | -5.78992 | 11.5949   |
| H  | -0.1603  | -3.31589 | 11.40608  |
| H  | -0.08432 | -3.59312 | 7.10911   |
| C  | 3.23751  | 6.14577  | -0.00126  |
| H  | 4.23885  | 5.72582  | -0.00703  |
| C  | -3.02682 | 8.82068  | 0.03496   |
| H  | -4.03663 | 9.21982  | 0.04076   |
| C  | 3.02682  | -8.82068 | 0.03496   |
| H  | 4.03663  | -9.21982 | 0.04076   |
| C  | -3.23751 | -6.14577 | -0.00126  |
| H  | -4.23885 | -5.72582 | -0.00703  |
| C  | 0.17986  | 7.63741  | -9.26659  |
| H  | 1.07438  | 8.07936  | -9.72575  |
| H  | -0.68301 | 8.12131  | -9.74367  |
| C  | 0.26668  | 7.59309  | 9.29937   |
| H  | 1.1702   | 8.02367  | 9.75171   |
| H  | -0.58643 | 8.08397  | 9.78676   |
| C  | -0.26668 | -7.59309 | 9.29937   |
| H  | 0.58643  | -8.08397 | 9.78676   |
| H  | -1.1702  | -8.02367 | 9.75171   |
| C  | -0.17986 | -7.63741 | -9.26659  |
| H  | -1.07438 | -8.07936 | -9.72575  |
| H  | 0.68301  | -8.12131 | -9.74367  |
| Zn | 0.11473  | 7.50309  | 0.01673   |
| Zn | -0.11473 | -7.50309 | 0.01673   |

## 2) ZnCP<sub>2</sub>

B3LYP/6-31G(d) for C, H, O; 6-31G+(d) for N; Lanl2DZ for Zn

|   |          |          |          |
|---|----------|----------|----------|
| C | -2.19721 | -6.50079 | -3.50517 |
| C | -3.09489 | -6.17936 | -2.53013 |
| C | -2.45921 | -6.45392 | -1.26289 |
| N | -1.19594 | -6.94649 | -1.48224 |
| C | -1.00287 | -6.99026 | -2.8428  |
| C | 3.42421  | -8.70946 | -2.50581 |
| C | 2.53966  | -8.37028 | -3.48698 |
| C | 1.34833  | -7.86341 | -2.8325  |
| N | 1.53159  | -7.90967 | -1.47114 |
| C | 2.78707  | -8.41824 | -1.24314 |
| C | 0.17844  | -7.41862 | -3.4894  |
| C | 2.55069  | -8.27207 | 3.51225  |
| C | 3.43148  | -8.63905 | 2.53773  |
| C | 2.79003  | -8.38259 | 1.2699   |
| N | 1.53568  | -7.86698 | 1.48748  |
| C | 1.35748  | -7.78225 | 2.8481   |
| C | -3.08374 | -6.10139 | 2.51121  |
| C | -2.18314 | -6.39439 | 3.49248  |
| C | -0.99195 | -6.90604 | 2.84118  |
| N | -1.18978 | -6.90341 | 1.48013  |
| C | -2.45314 | -6.41595 | 1.25088  |
| C | 0.19076  | -7.31705 | 3.49624  |
| C | -12.6047 | -17.4435 | -3.34052 |
| C | -13.4895 | -17.0725 | -2.37115 |
| C | -12.8527 | -17.3223 | -1.0995  |
| N | -11.5971 | -17.838  | -1.31048 |
| C | -11.4136 | -17.9287 | -2.66956 |
| C | -6.97043 | -19.602  | -2.31132 |
| C | -7.86788 | -19.3126 | -3.29655 |
| C | -9.06237 | -18.8017 | -2.6508  |
| N | -8.86961 | -18.8011 | -1.28948 |
| C | -7.60639 | -19.2862 | -1.05387 |
| C | -10.2435 | -18.3947 | -3.31132 |
| C | -7.88355 | -19.1904 | 3.70084  |
| C | -6.98273 | -19.5153 | 2.72986  |
| C | -7.61303 | -19.242  | 1.45979  |

|   |          |          |          |
|---|----------|----------|----------|
| N | -8.87645 | -18.7474 | 1.67271  |
| C | -9.07459 | -18.7003 | 3.0329   |
| C | -13.4979 | -16.9781 | 2.67199  |
| C | -12.6174 | -17.3131 | 3.65819  |
| C | -11.424  | -17.8243 | 3.01066  |
| N | -11.6019 | -17.7841 | 1.64796  |
| C | -12.8562 | -17.2758 | 1.41336  |
| C | -10.2575 | -18.2681 | 3.67392  |
| C | 0.21983  | -7.23894 | 4.99208  |
| C | -0.59431 | -8.07313 | 5.775    |
| C | -0.54535 | -7.97455 | 7.16682  |
| C | 0.31241  | -7.04606 | 7.79153  |
| C | 1.12134  | -6.21685 | 7.02337  |
| C | 1.07113  | -6.31779 | 5.6302   |
| C | -1.27123 | -8.71463 | 8.21091  |
| C | -0.85537 | -8.23255 | 9.47121  |
| C | -2.22132 | -9.72854 | 8.11617  |
| C | -2.76669 | -10.2718 | 9.29855  |
| C | -2.34186 | -9.77985 | 10.55414 |
| C | -1.39048 | -8.76475 | 10.63995 |
| C | -3.73819 | -11.3088 | 9.24472  |
| C | -4.57133 | -12.2022 | 9.24024  |
| C | -10.2869 | -18.2973 | 5.1715   |
| C | -11.1383 | -19.1971 | 5.83917  |
| C | -11.1888 | -19.2525 | 7.23488  |
| C | -10.3801 | -18.3987 | 7.97574  |
| C | -9.52218 | -17.4911 | 7.32123  |
| C | -9.4729  | -17.438  | 5.92695  |
| C | -9.21269 | -17.1579 | 9.61604  |
| C | -8.79653 | -16.7173 | 8.34076  |
| C | -8.67785 | -16.5879 | 10.7669  |
| C | -7.72644 | -15.5761 | 10.6482  |
| C | -7.30132 | -15.1254 | 9.37731  |
| C | -7.84642 | -15.707  | 8.21318  |
| C | -6.3298  | -14.0907 | 9.28986  |
| C | -5.49666 | -13.198  | 9.2564   |
| C | -10.2668 | -18.4784 | -4.80685 |
| C | -9.45511 | -17.6418 | -5.58966 |

|   |          |          |          |
|---|----------|----------|----------|
| C | -9.49843 | -17.7456 | -6.98125 |
| C | -10.3478 | -18.6819 | -7.60573 |
| C | -11.1543 | -19.5135 | -6.83763 |
| C | -11.11   | -19.4071 | -5.44467 |
| C | -8.77386 | -17.0043 | -8.02533 |
| C | -9.18175 | -17.4936 | -9.28542 |
| C | -7.8315  | -15.9833 | -7.93046 |
| C | -7.28569 | -15.44   | -9.11257 |
| C | -7.70242 | -15.9392 | -10.368  |
| C | -8.64618 | -16.9615 | -10.454  |
| C | -6.32183 | -14.3958 | -9.05822 |
| C | -5.49308 | -13.4984 | -9.0517  |
| C | 0.2021   | -7.38389 | -4.98687 |
| C | 1.04544  | -6.47645 | -5.65448 |
| C | 1.09006  | -6.41564 | -7.05016 |
| C | 0.28367  | -7.27184 | -7.79089 |
| C | -0.56582 | -8.18727 | -7.13635 |
| C | -0.60946 | -8.24559 | -5.74212 |
| C | -0.88197 | -8.51437 | -9.43113 |
| C | -1.29015 | -8.96225 | -8.15583 |
| C | -1.41728 | -9.08441 | -10.5818 |
| C | -2.36105 | -10.1033 | -10.4627 |
| C | -2.77808 | -10.5613 | -9.19174 |
| C | -2.23254 | -9.97961 | -8.02788 |
| C | -3.74195 | -11.6031 | -9.10354 |
| C | -4.5707  | -12.4999 | -9.06791 |
| H | -2.32453 | -6.40977 | -4.57369 |
| H | -4.09382 | -5.78115 | -2.65261 |
| H | 4.41617  | -9.12678 | -2.62131 |
| H | 2.672    | -8.46339 | -4.55474 |
| H | 2.68662  | -8.3357  | 4.58172  |
| H | 4.4236   | -9.05381 | 2.66099  |
| H | -4.08153 | -5.69758 | 2.6247   |
| H | -2.30632 | -6.2703  | 4.55813  |
| H | -12.7368 | -17.3853 | -4.41078 |
| H | -14.4815 | -16.6591 | -2.50044 |
| H | -5.97146 | -20.004  | -2.42051 |
| H | -7.74031 | -19.4385 | -4.3615  |

|   |          |          |          |
|---|----------|----------|----------|
| H | -7.76061 | -19.2796 | 4.77     |
| H | -5.98496 | -19.9152 | 2.85671  |
| H | -14.4901 | -16.5596 | 2.78141  |
| H | -12.7536 | -17.2146 | 4.72497  |
| H | -1.24735 | -8.79185 | 5.2885   |
| H | 1.78389  | -5.49341 | 7.4916   |
| H | 1.69331  | -5.67005 | 5.02012  |
| H | -2.5494  | -10.1088 | 7.15411  |
| H | -2.76996 | -10.2072 | 11.45474 |
| H | -1.07551 | -8.39876 | 11.61368 |
| H | -11.7604 | -19.8645 | 5.25042  |
| H | -11.8515 | -19.9603 | 7.72634  |
| H | -8.81975 | -16.7356 | 5.41738  |
| H | -8.99304 | -16.9219 | 11.75198 |
| H | -7.29854 | -15.1196 | 11.53446 |
| H | -7.51812 | -15.3583 | 7.23929  |
| H | -8.8084  | -16.9174 | -5.10313 |
| H | -11.8106 | -20.2427 | -7.30579 |
| H | -11.7305 | -20.0563 | -4.83438 |
| H | -7.50991 | -15.5975 | -6.96839 |
| H | -7.27421 | -15.5118 | -11.2686 |
| H | -8.95487 | -17.3327 | -11.4277 |
| H | 1.66577  | -5.80764 | -5.06558 |
| H | 1.7464   | -5.70209 | -7.54174 |
| H | -1.25629 | -8.95373 | -5.23234 |
| H | -1.10836 | -8.74512 | -11.567  |
| H | -2.78906 | -10.5599 | -11.3489 |
| H | -2.55435 | -10.3338 | -7.0538  |
| C | -7.03336 | -19.4759 | 0.20814  |
| C | -13.4245 | -17.0695 | 0.1518   |
| C | 3.35857  | -8.63004 | 0.01588  |
| C | -3.03253 | -6.22307 | -0.00787 |
| C | -10.2456 | -18.2557 | 9.47734  |
| H | -9.91772 | -19.1901 | 9.95275  |
| H | -11.1982 | -17.9852 | 9.95287  |
| C | -10.2079 | -18.5928 | -9.111   |
| H | -9.87154 | -19.5416 | -9.55052 |
| H | -11.1601 | -18.3465 | -9.59995 |

|    |          |          |          |
|----|----------|----------|----------|
| C  | 0.1441   | -7.41011 | -9.29247 |
| H  | 1.0965   | -7.67221 | -9.7729  |
| H  | -0.1921  | -6.47609 | -9.76282 |
| C  | 0.1776   | -7.13995 | 9.29697  |
| H  | -0.15041 | -6.19047 | 9.74154  |
| H  | 1.13003  | -7.39477 | 9.78131  |
| Zn | -10.2351 | -18.2921 | 0.17898  |
| Zn | 0.16921  | -7.4072  | 0.00238  |
| C  | -14.8504 | -16.488  | 0.13958  |
| C  | -15.9626 | -16.0344 | 0.13004  |
| C  | -5.59942 | -20.0374 | 0.22215  |
| C  | -4.48095 | -20.4753 | 0.23307  |
| H  | -3.48465 | -20.8654 | 0.2428   |
| C  | 4.78451  | -9.21163 | 0.02298  |
| C  | 5.89675  | -9.66527 | 0.02852  |
| H  | 6.8875   | -10.0694 | 0.03345  |
| C  | -4.46647 | -5.66146 | -0.01254 |
| C  | -5.58494 | -5.2234  | -0.01618 |
| C  | -7.01887 | -4.66179 | -0.02085 |
| C  | -8.13734 | -4.22374 | -0.02449 |
| C  | -9.57128 | -3.66213 | -0.02916 |
| C  | -10.1411 | -3.4411  | -1.28737 |
| C  | -10.1478 | -3.43988 | 1.22591  |
| C  | -9.49526 | -3.70601 | -2.55142 |
| N  | -11.4027 | -2.94706 | -1.51326 |
| C  | -9.50753 | -3.7047  | 2.49259  |
| N  | -11.4101 | -2.94491 | 1.44567  |
| C  | -10.3809 | -3.36583 | -3.53129 |
| H  | -8.49712 | -4.10779 | -2.66878 |
| C  | -11.5816 | -2.88471 | -2.87455 |
| Zn | -12.7774 | -2.48398 | -0.03829 |
| C  | -10.397  | -3.36349 | 3.46862  |
| H  | -8.51014 | -4.10719 | 2.61401  |
| C  | -11.5945 | -2.88136 | 2.80662  |
| H  | -10.2433 | -3.44242 | -4.5997  |
| C  | -12.7552 | -2.44736 | -3.52984 |
| N  | -14.1435 | -2.0221  | -1.52174 |
| N  | -14.1488 | -2.01958 | 1.44094  |

|   |          |          |          |
|---|----------|----------|----------|
| H | -10.2641 | -3.43988 | 4.53764  |
| C | -12.77   | -2.44148 | 3.45654  |
| C | -13.9448 | -2.0445  | -2.88202 |
| C | -12.7265 | -2.39177 | -5.02659 |
| C | -15.4143 | -1.55032 | -1.30057 |
| C | -15.418  | -1.54675 | 1.21348  |
| C | -13.956  | -2.03811 | 2.80258  |
| C | -12.7475 | -2.38349 | 4.9534   |
| C | -15.1434 | -1.56275 | -3.54225 |
| C | -13.5231 | -3.2543  | -5.79668 |
| C | -11.8936 | -1.46354 | -5.67847 |
| C | -16.0491 | -1.26732 | -2.56639 |
| C | -15.9954 | -1.34475 | -0.04475 |
| C | -16.0576 | -1.25837 | 2.47554  |
| C | -15.1566 | -1.55217 | 3.4562   |
| C | -11.9116 | -1.45935 | 5.60734  |
| C | -13.5527 | -3.23966 | 5.7217   |
| H | -15.2681 | -1.45887 | -4.60991 |
| C | -13.4752 | -3.17626 | -7.1898  |
| H | -14.1618 | -3.97836 | -5.29913 |
| C | -11.8447 | -1.38298 | -7.07301 |
| H | -11.2849 | -0.79425 | -5.07809 |
| H | -17.0531 | -0.8815  | -2.68728 |
| C | -17.4371 | -0.80326 | -0.04729 |
| H | -17.0614 | -0.87016 | 2.59068  |
| H | -15.2854 | -1.44445 | 4.52296  |
| C | -11.868  | -1.37684 | 7.00194  |
| H | -11.2964 | -0.79466 | 5.00858  |
| C | -13.5103 | -3.1595  | 7.11491  |
| H | -14.1938 | -3.9607  | 5.2229   |
| C | -12.6362 | -2.24014 | -7.82848 |
| C | -14.1848 | -3.94724 | -8.22257 |
| H | -11.1966 | -0.65363 | -7.55224 |
| C | -18.5616 | -0.38089 | -0.04927 |
| C | -12.668  | -2.22776 | 7.75556  |
| H | -11.2174 | -0.65069 | 7.48258  |
| C | -14.2295 | -3.92392 | 8.14598  |
| C | -12.7683 | -2.35973 | -9.33234 |

|   |          |          |          |
|---|----------|----------|----------|
| C | -13.7783 | -3.47624 | -9.49005 |
| C | -15.1133 | -4.97936 | -8.11218 |
| H | -19.5633 | -0.00466 | -0.05103 |
| C | -12.807  | -2.34416 | 9.25905  |
| C | -13.8251 | -3.4534  | 9.41435  |
| C | -15.1649 | -4.94964 | 8.03372  |
| H | -13.1159 | -1.42415 | -9.79118 |
| H | -11.8106 | -2.60186 | -9.81261 |
| C | -14.3013 | -4.03788 | -10.6505 |
| C | -15.6463 | -5.5526  | -9.28598 |
| H | -15.4337 | -5.35137 | -7.14429 |
| H | -13.1499 | -1.40555 | 9.71528  |
| H | -11.8529 | -2.59224 | 9.74354  |
| C | -14.357  | -4.00902 | 10.57363 |
| C | -15.707  | -5.51669 | 9.20644  |
| H | -15.4841 | -5.32125 | 7.06529  |
| C | -15.2311 | -5.07142 | -10.549  |
| H | -13.9935 | -3.68082 | -11.6298 |
| C | -16.5958 | -6.60893 | -9.21575 |
| C | -15.2937 | -5.03607 | 10.47027 |
| H | -14.0508 | -3.65206 | 11.55347 |
| C | -16.6636 | -6.56635 | 9.13469  |
| H | -15.6494 | -5.52175 | -11.443  |
| C | -17.412  | -7.51757 | -9.19559 |
| H | -15.7191 | -5.48169 | 11.36328 |
| C | -17.4841 | -7.47114 | 9.11478  |
| C | -18.3203 | -8.52906 | -9.1966  |
| C | -18.3954 | -8.47987 | 9.11375  |
| C | -19.1363 | -9.43781 | -9.21858 |
| C | -19.216  | -9.38455 | 9.13179  |
| C | -20.0852 | -10.4945 | -9.29092 |
| C | -20.1732 | -10.4339 | 9.20133  |
| C | -20.4902 | -10.9823 | -10.5547 |
| C | -20.6276 | -11.0616 | -8.11847 |
| C | -20.7058 | -11.007  | 8.02726  |
| C | -20.5966 | -10.9079 | 10.46428 |
| C | -21.4192 | -12.0164 | -10.6583 |
| H | -20.0648 | -10.5366 | -11.4476 |

|   |          |          |          |
|---|----------|----------|----------|
| C | -21.5552 | -12.0943 | -8.23093 |
| H | -20.315  | -10.6846 | -7.14999 |
| C | -21.6421 | -12.0322 | 8.13735  |
| H | -20.3789 | -10.6405 | 7.0595   |
| C | -21.5342 | -11.9344 | 10.56547 |
| H | -20.1784 | -10.4576 | 11.35835 |
| C | -21.9515 | -12.5719 | -9.49916 |
| H | -21.7191 | -12.3785 | -11.6381 |
| C | -22.2731 | -12.8599 | -7.19989 |
| C | -22.353  | -12.802  | 7.10454  |
| C | -22.0567 | -12.4961 | 9.40487  |
| H | -21.8482 | -12.2862 | 11.54467 |
| C | -22.9627 | -13.6876 | -9.34375 |
| C | -23.1069 | -13.7994 | -7.84039 |
| C | -22.2364 | -12.7746 | -5.80685 |
| C | -22.2994 | -12.7291 | 5.71131  |
| C | -23.2005 | -13.7304 | 7.74325  |
| C | -23.0736 | -13.6061 | 9.24718  |
| H | -23.9166 | -13.4481 | -9.83296 |
| H | -22.6115 | -14.6256 | -9.79488 |
| C | -23.9045 | -14.6526 | -7.08685 |
| C | -23.0392 | -13.6331 | -5.0387  |
| H | -21.6017 | -12.0479 | -5.30797 |
| C | -23.0984 | -13.5893 | 4.94105  |
| H | -21.6543 | -12.0107 | 5.21392  |
| C | -23.9945 | -14.5852 | 6.98767  |
| H | -22.7343 | -14.5423 | 9.71104  |
| H | -24.0316 | -13.3555 | 9.72268  |
| C | -23.8669 | -14.5648 | -5.6924  |
| H | -24.5487 | -15.3845 | -7.56746 |
| C | -23.0225 | -13.5697 | -3.54208 |
| C | -23.0639 | -13.5391 | 3.44413  |
| C | -23.9396 | -14.51   | 5.59306  |
| H | -24.6489 | -15.3089 | 7.46685  |
| H | -24.4804 | -15.2309 | -5.09344 |
| C | -21.8382 | -13.9692 | -2.88261 |
| C | -24.2013 | -13.129  | -2.89854 |
| C | -24.2341 | -13.1027 | 2.78249  |

|    |          |          |          |
|----|----------|----------|----------|
| C  | -21.8727 | -13.9459 | 2.80185  |
| H  | -24.55   | -15.1778 | 4.99285  |
| C  | -20.6343 | -14.4543 | -3.53066 |
| N  | -21.6505 | -13.9845 | -1.52069 |
| C  | -25.3967 | -12.6513 | -3.56741 |
| N  | -24.3912 | -13.0595 | -1.53907 |
| C  | -25.4369 | -12.6171 | 3.43232  |
| N  | -24.4076 | -13.0462 | 1.41979  |
| C  | -20.6774 | -14.4284 | 3.46763  |
| N  | -21.669  | -13.9715 | 1.44192  |
| C  | -19.7365 | -14.7446 | -2.54602 |
| H  | -20.501  | -14.5638 | -4.59673 |
| C  | -20.3814 | -14.455  | -1.28684 |
| Zn | -23.0284 | -13.5149 | -0.0507  |
| C  | -26.2902 | -12.306  | -2.59649 |
| H  | -25.5257 | -12.5803 | -4.63728 |
| C  | -25.6546 | -12.5643 | -1.32592 |
| C  | -26.3185 | -12.281  | 2.4474   |
| H  | -25.5784 | -12.5351 | 4.49982  |
| C  | -25.668  | -12.5524 | 1.18731  |
| C  | -19.7685 | -14.7273 | 2.49579  |
| H  | -20.5571 | -14.5305 | 4.53593  |
| C  | -20.3979 | -14.4455 | 1.22714  |
| H  | -18.7316 | -15.1311 | -2.65682 |
| C  | -19.8104 | -14.654  | -0.02533 |
| H  | -27.2874 | -11.9049 | -2.72396 |
| C  | -26.2345 | -12.3368 | -0.07351 |
| H  | -27.3168 | -11.8779 | 2.55869  |
| H  | -18.7657 | -15.1148 | 2.62103  |
| C  | -18.3688 | -15.1955 | -0.01345 |
| C  | -27.6684 | -11.7752 | -0.0833  |
| C  | -17.2444 | -15.6179 | -0.00419 |
| C  | -28.7868 | -11.3371 | -0.09094 |
| H  | -29.7831 | -10.9469 | -0.09775 |

3)  $T_2$  (n = 4)

B3LYP/6-31G(d)

|   |          |         |          |
|---|----------|---------|----------|
| C | -4.63793 | 8.25676 | 12.13563 |
|---|----------|---------|----------|

|   |          |          |          |
|---|----------|----------|----------|
| C | -5.56985 | 8.82411  | 11.26617 |
| C | -5.16416 | 9.78103  | 10.33627 |
| C | -3.82584 | 10.17189 | 10.27633 |
| C | -2.89411 | 9.60494  | 11.14578 |
| H | -4.95798 | 7.50192  | 12.86843 |
| H | -6.62431 | 8.5156   | 11.31352 |
| H | -3.50608 | 10.92653 | 9.54303  |
| H | -1.83936 | 9.91274  | 11.09864 |
| C | -6.19264 | 10.40679 | 9.37602  |
| C | -5.78685 | 11.36371 | 8.44602  |
| C | -7.53106 | 10.01652 | 9.43666  |
| C | -6.71867 | 11.92996 | 7.57566  |
| H | -4.73219 | 11.67121 | 8.39787  |
| C | -8.46259 | 10.58298 | 8.56671  |
| H | -7.85072 | 9.26189  | 10.16996 |
| C | -8.05639 | 11.53949 | 7.63581  |
| H | -9.51745 | 10.27557 | 8.61415  |
| C | -9.08508 | 12.16456 | 6.67535  |
| C | -10.4227 | 11.77409 | 6.7354   |
| C | -8.67878 | 13.12138 | 5.74445  |
| C | -11.3547 | 12.34104 | 5.86574  |
| H | -10.7433 | 11.02036 | 7.46919  |
| C | -9.61051 | 13.68773 | 4.8747   |
| H | -7.62402 | 13.42888 | 5.69711  |
| H | -12.4094 | 12.03354 | 5.91359  |
| H | -9.29065 | 14.44186 | 4.1409   |
| N | -3.30001 | 8.64692  | 12.07518 |
| N | -10.9487 | 13.29786 | 4.93554  |
| O | -6.30217 | 12.91068 | 6.62196  |
| C | -4.89722 | 13.14105 | 6.75663  |
| H | -4.68663 | 13.49231 | 7.74508  |
| H | -4.36774 | 12.2279  | 6.58218  |
| C | -4.4498  | 14.19806 | 5.72993  |
| H | -4.97898 | 15.11151 | 5.90468  |
| H | -4.66079 | 13.84709 | 4.74148  |
| C | -2.93667 | 14.4458  | 5.87448  |
| H | -2.40749 | 13.53245 | 5.69982  |
| H | -2.72558 | 14.79686 | 6.86303  |

|   |          |          |          |
|---|----------|----------|----------|
| C | -2.48915 | 15.50291 | 4.84798  |
| H | -3.01843 | 16.41626 | 5.02273  |
| H | -2.70034 | 15.15194 | 3.85953  |
| O | -1.08416 | 15.73286 | 4.98221  |
| C | -0.66791 | 16.71434 | 4.02914  |
| C | 0.66959  | 17.10561 | 3.96956  |
| C | -1.59968 | 17.28016 | 3.15848  |
| C | 1.07553  | 18.06303 | 3.03949  |
| C | -1.19401 | 18.2377  | 2.22892  |
| H | -2.6541  | 16.97117 | 3.20541  |
| C | 0.14398  | 18.6295  | 2.16968  |
| H | 2.12992  | 18.37185 | 2.99329  |
| H | 0.46375  | 19.38461 | 1.43698  |
| C | 1.69829  | 16.48077 | 4.93027  |
| C | 1.2927   | 15.52356 | 5.86021  |
| C | 3.03614  | 16.87184 | 4.87002  |
| C | 2.22485  | 14.95745 | 6.73031  |
| H | 0.23832  | 15.21503 | 5.90754  |
| C | 3.96804  | 16.30626 | 5.7402   |
| H | 3.35571  | 17.62628 | 4.13642  |
| H | 1.90463  | 14.20339 | 7.46386  |
| H | 5.02251  | 16.61479 | 5.69347  |
| C | -2.22233 | 18.86331 | 1.26831  |
| C | -3.55988 | 18.4721  | 1.32704  |
| C | -1.81603 | 19.82187 | 0.3396   |
| C | -4.49136 | 19.03916 | 0.45683  |
| H | -3.88026 | 17.71684 | 2.05922  |
| C | -2.74714 | 20.38849 | -0.53074 |
| H | -0.76143 | 20.13043 | 0.29381  |
| H | -5.54565 | 18.72996 | 0.50287  |
| H | -2.42711 | 21.14355 | -1.26339 |
| N | 3.56215  | 15.34901 | 6.67087  |
| N | -4.08517 | 19.99672 | -0.47231 |

4)  $T_2$  (n = 6)

B3LYP/6-31G(d)

|   |         |         |        |
|---|---------|---------|--------|
| C | 0.1289  | 17.0106 | 5.3341 |
| H | -0.0246 | 16.4126 | 6.208  |

|   |          |         |        |
|---|----------|---------|--------|
| H | -0.0224  | 16.4113 | 4.4607 |
| C | -0.87    | 18.1826 | 5.3319 |
| H | -0.7165  | 18.7806 | 4.458  |
| H | -0.7187  | 18.7819 | 6.2053 |
| C | -2.308   | 17.6314 | 5.3305 |
| H | -2.4616  | 17.0333 | 6.2045 |
| H | -2.4595  | 17.032  | 4.4572 |
| C | -3.3071  | 18.8032 | 5.3284 |
| H | -3.1535  | 19.4013 | 4.4545 |
| H | -3.1557  | 19.4026 | 6.2017 |
| C | -4.7451  | 18.252  | 5.327  |
| H | -4.8987  | 17.654  | 6.2009 |
| H | -4.8964  | 17.6528 | 4.4536 |
| C | -5.7441  | 19.424  | 5.3248 |
| H | -5.5905  | 20.022  | 4.4509 |
| H | -5.5927  | 20.0233 | 6.1982 |
| O | -7.0794  | 18.9122 | 5.3236 |
| C | -8.0076  | 19.9999 | 5.3215 |
| C | -7.5417  | 21.3145 | 5.3211 |
| C | -9.3797  | 19.7461 | 5.3204 |
| C | -8.4474  | 22.3758 | 5.3183 |
| H | -6.4604  | 21.5149 | 5.3216 |
| C | -10.2852 | 20.8071 | 5.3181 |
| C | -9.8189  | 22.122  | 5.3166 |
| H | -11.3666 | 20.6073 | 5.3172 |
| H | -10.5329 | 22.9583 | 5.3146 |
| C | -7.9327  | 23.8271 | 5.3172 |
| C | -8.8378  | 24.8883 | 5.3144 |
| C | -6.5605  | 24.0808 | 5.3185 |
| C | -8.3715  | 26.2032 | 5.3142 |
| H | -9.9193  | 24.6889 | 5.3137 |
| C | -6.0944  | 25.3954 | 5.3179 |
| H | -5.8471  | 23.244  | 5.3207 |
| H | -9.0852  | 27.0397 | 5.3125 |
| H | -5.013   | 25.5955 | 5.319  |
| C | -9.8938  | 18.2944 | 5.321  |
| C | -11.2651 | 18.0399 | 5.3198 |
| C | -8.9873  | 17.2334 | 5.323  |

|   |          |         |        |
|---|----------|---------|--------|
| C | -11.7306 | 16.7248 | 5.3195 |
| H | -11.9795 | 18.8758 | 5.3178 |
| C | -9.4529  | 15.9186 | 5.3233 |
| H | -7.9062  | 17.4343 | 5.324  |
| H | -12.8118 | 16.5243 | 5.3181 |
| H | -8.7389  | 15.0822 | 5.3249 |
| O | 1.4642   | 17.5225 | 5.3354 |
| C | 2.3924   | 16.4346 | 5.3374 |
| C | 3.7639   | 16.6879 | 5.3385 |
| C | 1.9258   | 15.1195 | 5.3379 |
| C | 4.6694   | 15.6265 | 5.3413 |
| C | 2.8312   | 14.0585 | 5.3402 |
| H | 0.8444   | 14.9201 | 5.3369 |
| C | 4.2031   | 14.3119 | 5.3423 |
| H | 5.7507   | 15.8264 | 5.3427 |
| H | 4.9166   | 13.4753 | 5.3443 |
| C | 2.3167   | 12.6069 | 5.3407 |
| C | 3.222    | 11.5459 | 5.343  |
| C | 0.9445   | 12.3532 | 5.3384 |
| C | 2.7558   | 10.231  | 5.3442 |
| H | 4.3033   | 11.7456 | 5.3451 |
| C | 0.4785   | 11.0385 | 5.3392 |
| H | 0.231    | 13.1899 | 5.3365 |
| H | 3.4697   | 9.3946  | 5.3465 |
| H | -0.6028  | 10.8382 | 5.3374 |
| C | 4.2792   | 18.1391 | 5.3385 |
| C | 3.3746   | 19.2006 | 5.3358 |
| C | 5.6515   | 18.392  | 5.341  |
| C | 3.8415   | 20.5153 | 5.3366 |
| H | 2.2931   | 19.0016 | 5.3342 |
| C | 6.1183   | 19.7064 | 5.3413 |
| H | 6.3646   | 17.5549 | 5.3431 |
| H | 3.1282   | 21.3521 | 5.3348 |
| H | 7.1998   | 19.9061 | 5.3432 |
| N | -6.9999  | 26.4567 | 5.3161 |
| N | -10.8247 | 15.6642 | 5.3211 |
| N | 5.2133   | 20.7682 | 5.3395 |
| N | 1.3842   | 9.9773  | 5.3425 |

5) T<sub>2</sub> (n = 8)

B3LYP/6-31G(d)

|   |          |         |        |
|---|----------|---------|--------|
| C | -5.4872  | 5.4123  | 8.8367 |
| C | -6.4195  | 5.9805  | 7.9682 |
| C | -6.014   | 6.9375  | 7.0383 |
| C | -4.6755  | 7.3276  | 6.9774 |
| C | -3.7434  | 6.7598  | 7.8459 |
| H | -5.8071  | 4.6574  | 9.5695 |
| H | -7.4741  | 5.6726  | 8.0163 |
| H | -4.3559  | 8.0823  | 6.2441 |
| H | -2.6885  | 7.067   | 7.798  |
| C | -7.0429  | 7.5642  | 6.0791 |
| C | -6.6373  | 8.5212  | 5.1491 |
| C | -8.3815  | 7.1747  | 6.1407 |
| C | -7.5695  | 9.0883  | 4.2797 |
| H | -5.5825  | 8.8281  | 5.1002 |
| C | -9.3134  | 7.742   | 5.2717 |
| H | -8.701   | 6.42    | 6.874  |
| C | -8.9074  | 8.6986  | 4.3408 |
| H | -10.3684 | 7.4352  | 5.3199 |
| C | -9.9365  | 9.3246  | 3.3814 |
| C | -11.2743 | 8.9349  | 3.4424 |
| C | -9.5304  | 10.2815 | 2.4505 |
| C | -12.2067 | 9.5027  | 2.5737 |
| H | -11.5947 | 8.1811  | 4.1762 |
| C | -10.4625 | 10.8487 | 1.5817 |
| H | -8.4755  | 10.5884 | 2.4024 |
| H | -13.2615 | 9.1958  | 2.6223 |
| H | -10.1428 | 11.6029 | 0.8479 |
| N | -4.1491  | 5.8017  | 8.7753 |
| N | -11.8009 | 10.4596 | 1.6435 |
| O | -7.1532  | 10.0691 | 3.326  |
| C | -5.748   | 10.2986 | 3.4596 |
| H | -5.5364  | 10.6494 | 4.448  |
| H | -5.2192  | 9.3852  | 3.2844 |
| C | -5.3008  | 11.3557 | 2.4329 |
| H | -5.8293  | 12.2694 | 2.6084 |

|   |         |         |         |
|---|---------|---------|---------|
| H | -5.5128 | 11.0052 | 1.4445  |
| C | -3.7874 | 11.6025 | 2.5763  |
| H | -3.2589 | 10.6889 | 2.4009  |
| H | -3.5753 | 11.9531 | 3.5648  |
| C | -3.3401 | 12.6597 | 1.5498  |
| H | -3.8687 | 13.5733 | 1.7253  |
| H | -3.5523 | 12.3092 | 0.5614  |
| C | -1.8268 | 12.9064 | 1.6932  |
| H | -1.2983 | 11.9927 | 1.5177  |
| H | -1.6147 | 13.2569 | 2.6816  |
| C | -1.3794 | 13.9636 | 0.6666  |
| H | -1.908  | 14.8772 | 0.8422  |
| H | -1.5917 | 13.6131 | -0.3219 |
| C | 0.1339  | 14.2102 | 0.81    |
| H | 0.6623  | 13.2966 | 0.6344  |
| H | 0.346   | 14.5608 | 1.7985  |
| C | 0.5812  | 15.2674 | -0.2166 |
| H | 0.0527  | 16.1811 | -0.041  |
| H | 0.369   | 14.917  | -1.205  |
| O | 1.9865  | 15.4965 | -0.0834 |
| C | 2.4025  | 16.4781 | -1.0364 |
| C | 3.7401  | 16.8686 | -1.0971 |
| C | 1.47    | 17.0457 | -1.9056 |
| C | 4.146   | 17.8257 | -2.0275 |
| C | 1.8758  | 18.0029 | -2.8354 |
| H | 0.4154  | 16.7376 | -1.8576 |
| C | 3.214   | 18.3927 | -2.8966 |
| H | 5.2007  | 18.1331 | -2.0754 |
| H | 3.5338  | 19.1472 | -3.6298 |
| C | 0.8471  | 18.6295 | -3.7949 |
| C | 1.2528  | 19.5867 | -4.7246 |
| C | -0.4915 | 18.2399 | -3.7337 |
| C | 0.3209  | 20.1538 | -5.5943 |
| H | 2.3077  | 19.8937 | -4.7732 |
| C | -1.4232 | 18.8072 | -4.6029 |
| H | -0.8112 | 17.4852 | -3.0006 |
| H | 0.6411  | 20.9081 | -6.3275 |
| H | -2.4782 | 18.5003 | -4.555  |

|   |        |         |         |
|---|--------|---------|---------|
| C | 4.7696 | 16.2421 | -0.1384 |
| C | 4.3645 | 15.2852 | 0.792   |
| C | 6.108  | 16.6321 | -0.2002 |
| C | 5.2974 | 14.7171 | 1.6599  |
| H | 3.3099 | 14.9774 | 0.8407  |
| C | 7.0405 | 16.0645 | 0.6679  |
| H | 6.4271 | 17.3867 | -0.9338 |
| H | 4.9779 | 13.9624 | 2.393   |
| H | 8.0954 | 16.3717 | 0.6194  |
| N | 6.6353 | 15.1066 | 1.5978  |
| N | -1.017 | 19.7639 | -5.5335 |

**6) T<sub>2</sub> (n = 12)**

B3LYP/6-31G(d)

|   |          |          |          |
|---|----------|----------|----------|
| C | -4.58067 | 8.45105  | 12.36072 |
| C | -5.33452 | 9.61081  | 12.17911 |
| C | -4.92909 | 10.56802 | 11.2494  |
| C | -3.76838 | 10.36611 | 10.50143 |
| C | -3.01446 | 9.20677  | 10.68317 |
| H | -4.90082 | 7.69624  | 13.09351 |
| H | -6.24928 | 9.76941  | 12.7685  |
| H | -3.44877 | 11.12104 | 9.76837  |
| H | -2.09978 | 9.04739  | 10.09374 |
| C | -5.76155 | 11.84788 | 11.04836 |
| C | -5.35608 | 12.80508 | 10.11851 |
| C | -6.92171 | 12.05011 | 11.79701 |
| C | -6.11083 | 13.96418 | 9.93601  |
| H | -4.44205 | 12.64584 | 9.52827  |
| C | -7.67596 | 13.20912 | 11.61471 |
| H | -7.24125 | 11.29514 | 12.53002 |
| C | -7.2708  | 14.16612 | 10.68386 |
| H | -8.59038 | 13.36877 | 12.20447 |
| C | -8.10386 | 15.44554 | 10.4827  |
| C | -9.2638  | 15.64747 | 11.23041 |
| C | -7.69847 | 16.40275 | 9.55182  |
| C | -10.018  | 16.807   | 11.04866 |
| H | -9.58331 | 14.89346 | 11.96438 |
| C | -8.45283 | 17.56169 | 9.36978  |

|   |          |          |          |
|---|----------|----------|----------|
| H | -6.78402 | 16.24319 | 8.9622   |
| H | -10.9322 | 16.96635 | 11.63864 |
| H | -8.13362 | 18.31633 | 8.63622  |
| N | -3.42084 | 8.24888  | 11.6125  |
| N | -9.61255 | 17.7641  | 10.11853 |
| O | -5.69507 | 14.94513 | 8.98223  |
| C | -4.48726 | 14.51393 | 8.34949  |
| H | -3.72332 | 14.38771 | 9.0879   |
| H | -4.65859 | 13.58307 | 7.85072  |
| C | -4.03973 | 15.5714  | 7.32331  |
| H | -3.86789 | 16.50228 | 7.82219  |
| H | -4.8038  | 15.69803 | 6.58514  |
| C | -2.73932 | 15.10684 | 6.64142  |
| H | -2.91107 | 14.17601 | 6.14263  |
| H | -1.97509 | 14.98021 | 7.37963  |
| C | -2.29159 | 16.16428 | 5.61536  |
| H | -2.11987 | 17.09513 | 6.11429  |
| H | -3.0558  | 16.291   | 4.8773   |
| C | -0.99131 | 15.69969 | 4.93352  |
| H | -1.16315 | 14.7688  | 4.43463  |
| H | -0.22716 | 15.57301 | 5.67163  |
| C | -0.54354 | 16.75712 | 3.90732  |
| H | -0.37178 | 17.68793 | 4.40633  |
| H | -1.30787 | 16.88392 | 3.16923  |
| C | 0.75669  | 16.29244 | 3.22547  |
| H | 0.58478  | 15.36172 | 2.72656  |
| H | 1.52093  | 16.16581 | 3.96367  |
| C | 1.20438  | 17.34991 | 2.19933  |
| H | 0.44015  | 17.47663 | 1.46123  |
| H | 1.37621  | 18.28073 | 2.69827  |
| C | 2.50466  | 16.88529 | 1.51743  |
| H | 2.33283  | 15.95446 | 1.0185   |
| H | 3.26889  | 16.75858 | 2.25553  |
| C | 2.95234  | 17.94275 | 0.49129  |
| H | 2.18811  | 18.06947 | -0.24682 |
| H | 3.12417  | 18.87358 | 0.99022  |
| C | 4.25262  | 17.47814 | -0.19062 |
| H | 4.08079  | 16.54731 | -0.68954 |

|   |          |          |          |
|---|----------|----------|----------|
| H | 5.01685  | 17.35143 | 0.54749  |
| C | 4.7003   | 18.5356  | -1.21676 |
| H | 3.93608  | 18.66231 | -1.95487 |
| H | 4.87213  | 19.46643 | -0.71783 |
| O | 5.90771  | 18.10417 | -1.84995 |
| C | 6.3238   | 19.08553 | -2.80321 |
| C | 5.56961  | 20.24463 | -2.9856  |
| C | 7.48453  | 18.88376 | -3.55047 |
| C | 5.97587  | 21.20205 | -3.91553 |
| H | 4.65492  | 20.40388 | -2.39644 |
| C | 7.8905   | 19.8407  | -4.48053 |
| C | 7.13569  | 21.00003 | -4.66334 |
| H | 8.80497  | 19.68166 | -5.07022 |
| H | 7.45563  | 21.75435 | -5.39678 |
| C | 8.31724  | 17.60417 | -3.34839 |
| C | 7.91121  | 16.64728 | -2.4183  |
| C | 9.47757  | 17.40186 | -4.09612 |
| C | 8.66563  | 15.488   | -2.23555 |
| H | 6.9965   | 16.80651 | -1.82916 |
| C | 10.23212 | 16.2431  | -3.91313 |
| H | 9.79724  | 18.1563  | -4.82967 |
| H | 8.34568  | 14.73416 | -1.50166 |
| H | 11.14716 | 16.08363 | -4.50182 |
| C | 5.1423   | 22.48113 | -4.11734 |
| C | 5.54793  | 23.4387  | -5.04689 |
| C | 3.981    | 22.68179 | -3.37066 |
| C | 4.79249  | 24.5973  | -5.22972 |
| H | 6.46304  | 23.28046 | -5.63568 |
| C | 3.22595  | 23.84031 | -3.55305 |
| H | 3.6612   | 21.92642 | -2.63811 |
| H | 5.11297  | 25.3524  | -5.96209 |
| H | 2.31092  | 23.99915 | -2.96416 |
| N | 9.82617  | 15.28606 | -2.98227 |
| N | 3.63206  | 24.79851 | -4.48264 |

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