## Supporting Information:

## Fluorine-Substituted Quinoidal Thiophene with F-H Hydrogen Bond Locked Conformation for High Performance n -Channel Organic Transistor

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## 1. Synthesis and Characterizations



Scheme S1. Synthetic routes of F2-QBTTE and QBTTE

## Compound 1

A three neck flask was charged with 3-bromo-4-decyltetradecanylthiophene-2-carbaldehyde ( $5.43 \mathrm{~g}, 10.29 \mathrm{mmol}$ ), $\mathrm{K}_{2} \mathrm{CO}_{3}(3.42 \mathrm{~g}, 24.76 \mathrm{mmol})$ and DMF ( 30 mL ).To this mixture, ethyl mercaptoacetate ( $1.80 \mathrm{~g}, 14.94 \mathrm{mmol}$ ) was added dropwise. The reaction mixture was stirred for 8 hours at $80^{\circ} \mathrm{C}$ under nitrogen and then cooled to room temperature. The reaction mixture was quenched with 30 mL water and extracted with ethyl acetate. The combined extracts were washed with brine and dried over anhydrous $\mathrm{MgSO}_{4}$. The solvent was removed by rotary evaporation and the crude product was purified by silica gel column chromatography (eluent: eluent: Petroleum ether/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}=5: 1$ ) to yield compound $\mathbf{1}(5.42 \mathrm{~g}, 96 \%)$. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.97(\mathrm{~s}$, $1 \mathrm{H}), 7.16(\mathrm{~s}, 1 \mathrm{H}), 4.38(\mathrm{q}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.65(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 1.84(\mathrm{br}, 1 \mathrm{H}), 1.40(\mathrm{t}, J=7.0$ $\mathrm{Hz}, 3 \mathrm{H}), 1.24(\mathrm{br}, 40 \mathrm{H}), 0.88(\mathrm{t}, J=6.1 \mathrm{~Hz}, 6 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 162.78,144.83$, 138.14, 134.50, 134.46, 126.94, 126.18, 61.34, 37.27, 34.82, 33.65, 32.03, 32.02, 30.05, 29.80, 29.79, 29.76, 29.74, 29.72, 29.47, 29.46, 26.62, 22.80, 14.46, 14.21. MS (MALDI-TOF) m/z: $549.3(\mathrm{M}+\mathrm{H})^{+}$; HRMS (MALDI) m/z: calcd for $\mathrm{C}_{33} \mathrm{H}_{57} \mathrm{O}_{2} \mathrm{~S}_{2}{ }^{+1} 549.3800$, found 549.3776 .

## Compound 2

A three neck flask was charged with THF ( 50 mL ), methanol ( 20 mL ), water ( 10 mL ), NaOH $(5.0 \mathrm{~g})$, compound $\mathbf{1}(5.40 \mathrm{~g}, 9.84 \mathrm{mmol})$ and a catalytic amount of tetrabutylammonium iodide $(0.50 \mathrm{~g}, 1.35 \mathrm{mmol})$. The mixture was refluxed for 4 hours. After cooling down to room
temperature, dilute hydrochloric acid was added to adjust PH to below 7.0. Most of the solvent was evaporated and the residue was extracted with ethyl acetate. The combined extracts were washed with brine and dried over anhydrous $\mathrm{MgSO}_{4}$. The solvent was removed by rotary evaporation and the crude product was purified by silica gel column chromatography (eluent: Petroleum ether/ $\mathrm{EA}=4: 1$ ) to yield compound $2(4.27 \mathrm{~g}, 83.4 \%) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $8.08(\mathrm{~s}, 1 \mathrm{H}), 7.22(\mathrm{~s}, 1 \mathrm{H}), 2.67(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.85(\mathrm{br}, 1 \mathrm{H}), 1.24(\mathrm{br}, 80 \mathrm{H}), 0.88(\mathrm{t}, J=6.2$ $\mathrm{Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 168.36,146.23,138.34,134.65,133.02,128.06,128.02$, $37.27,34.78,33.62,32.03,30.03,29.76,29.72,29.47,26.62,22.80,14.24$. MS (ESI) m/z: 543.35 $(\mathrm{M}+\mathrm{Na})^{+}$.

## Compound 3

To a solution of compound $2(1.58 \mathrm{~g}, 3.04 \mathrm{mmol})$ in 15 mL dry THF, n -butyllithium ( 2.8 mL , $6.72 \mathrm{mmol}, 2.4 \mathrm{M}$ in hexane) was added dropwise under nitrogen at $-78^{\circ} \mathrm{C}$. After the addition was finished, the resulting mixture was stirred for another one hour and then N -Fluorobenzenesulfonimide ( $1.16 \mathrm{~g}, 3.67 \mathrm{mmol}$ ) was added. The final mixture was allowed to warm to room temperature and was stirred overnight. The reaction was quenched with 10 mL water and then acidified with dilute HCl to $\mathrm{PH}<7$. The mixture was extracted with ethyl acetate. The combined extracts were washed with brine and dried over anhydrous $\mathrm{MgSO}_{4}$. The solvent was removed by rotary evaporation and the crude product was purified by silica gel column chromatography (eluent: Petroleum ether/ $\mathrm{EA}=6: 1$ ) to yield compound $3(1.32 \mathrm{~g}, 80.3 \%) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.24(\mathrm{~s}, 1 \mathrm{H}), 2.63(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.80(\mathrm{br}, 1 \mathrm{H}), 1.25(\mathrm{br}, 40 \mathrm{H})$, $0.88(\mathrm{td}, J=6.8,1.3 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 167.07(\mathrm{~d}, J=3.1 \mathrm{~Hz}), 154.71(\mathrm{~d}, J$ $=280.9 \mathrm{~Hz}), 142.39(\mathrm{~d}, J=7.4 \mathrm{~Hz}), 135.37,128.61,127.24(\mathrm{~d}, J=22.7 \mathrm{~Hz}), 111.74(\mathrm{~d}, J=9.9$ Hz ), 37.43, 33.93, 33.60, 32.05, 32.04, 30.02, 29.82, 29.80, 29.78, 29.77, 29.76, 29.73, 29.49, $29.48,27.01,26.62,22.81,14.24 .{ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-114.50$. MS (MALDI-TOF) m/z: $539.2(\mathrm{M}+\mathrm{H})^{+}$.

## Compound 4a

A solution of compound $6(1.32 \mathrm{~g}, 2.44 \mathrm{mmol})$ in tetrahydrofuran $(5 \mathrm{~mL})$ was added to a mixture of lithium aluminum hydride $(0.21 \mathrm{~g}, 5.4 \mathrm{mmol})$ in dry THF ( 10 mL ) under nitrogen at $0{ }^{\circ} \mathrm{C}$. The reaction mixture was stirred overnight at room temperature and was quenched slowly with water $(0.2 \mathrm{~mL}), 15 \%$ aqueous NaOH solution $(0.4 \mathrm{~mL})$ and water $(0.6 \mathrm{~mL})$ successively. The mixture was extracted with ethyl acetate. The combined extracts were washed with brine and dried over anhydrous $\mathrm{MgSO}_{4}$. The solvent was removed by rotary evaporation and the crude product was purified by silica gel column chromatography (eluent: Petroleum ether/ EA=5:1) to yield compound $4 \mathbf{4 a}(0.71 \mathrm{~g}, 55.7 \%) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 6.96(\mathrm{~s}, 1 \mathrm{H}), 4.85-4.84(\mathrm{~m}, 2 \mathrm{H})$, $2.59(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.80(\mathrm{br}, 1 \mathrm{H}), 1.24(\mathrm{br}, 40 \mathrm{H}), 0.88(\mathrm{t}, J=6.8 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 147.31(\mathrm{~d}, J=260.3 \mathrm{~Hz}), 135.43(\mathrm{~d}, J=6.4 \mathrm{~Hz}), 134.68(\mathrm{~d}, J=1.3 \mathrm{~Hz}), 125.18(\mathrm{~d}$, $J=25.5 \mathrm{~Hz}), 120.94(\mathrm{~d}, J=17.4 \mathrm{~Hz}), 111.81(\mathrm{~d}, J=1.4 \mathrm{~Hz}), 55.93,37.27,33.68,33.53,32.03$, 30.01, 29.80, 29.79, 29.77, 29.76, 29.75, 29.71, 29.47, 29.45, 26.59, 22.79, 14.21. ${ }^{19}$ F NMR (376 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$-135.19. MS (MALDI-TOF) m/z: $507.2(\mathrm{M}-\mathrm{OH})^{+}$.

## Compound 4b

Compound $\mathbf{4 b}$ was synthesized according to the same procedure as for compound $\mathbf{4 a}$. Yield: $94.9 \% .^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.17(\mathrm{~s}, 1 \mathrm{H}), 6.93(\mathrm{~s}, 1 \mathrm{H}), 4.87(\mathrm{~d}, J=5.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.62(\mathrm{~d}$, $J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.88(\mathrm{br}, 1 \mathrm{H}), 1.24(\mathrm{br}, 40 \mathrm{H}), 0.88(\mathrm{t}, J=6.0 \mathrm{~Hz}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 145.40,140.19,137.85,134.34,122.33,118.41,61.07,37.32,34.92,33.66,32.04,30.09,29.78$, 29.77, 29.48, 29.47, 26.66, 22.81, 14.23. MS (MALDI-TOF) m/z: 507.3 (M+H)+; HRMS (MALDI) m/z: calcd for $\mathrm{C}_{31} \mathrm{H}_{55} \mathrm{OS}_{2}{ }^{+1} 507.3694$, found 507.3675

## Compound 5a

To a stirred suspension of pyridinium chlorochromate ( $0.42 \mathrm{~g}, 1.94 \mathrm{mmol}$ ) in dichloromethane ( 8 mL ), compound $\mathbf{4 a}(0.65 \mathrm{~g}, 1.23 \mathrm{mmol})$ was added. The mixture was stirred overnight and then filtered through funnel filled with silica gel. The gum in the flask was washed with dichloromethane three times. The combined filtrate was evaporated under reduced pressure. The crude product was flash chromatographed on silica. Light petroleum-ethyl acetate (ratio 5:1) eluted compound $5 \mathbf{5 a}(0.52 \mathrm{~g}, 80 \%) .{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 10.10(\mathrm{~s}, 1 \mathrm{H}), 7.26(\mathrm{~s}, 1 \mathrm{H}), 2.63$ (d, $J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.87(\mathrm{br}, 1 \mathrm{H}), 1.24(\mathrm{br}, 40 \mathrm{H}), 0.88(\mathrm{~d}, J=8 \mathrm{~Hz}, 6 \mathrm{H}){ }^{13} \mathrm{C}$ NMR $(101 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 8180.20,156.51(\mathrm{~d}, J=279.4 \mathrm{~Hz}), 144.60(\mathrm{~d}, J=7.5 \mathrm{~Hz}), 135.79,129.73,126.52(\mathrm{~d}, J=$ $21.9 \mathrm{~Hz}), 123.33$ (d, $J=9.2 \mathrm{~Hz}), 37.45,33.90,33.62,32.02,32.00,29.98,29.78,29.77,29.75$, 29.72, 29.68, 29.45, 29.43, 26.63, 22.78, 14.19. ${ }^{19} \mathrm{~F}$ NMR ( 376 MHz, ) $\delta-123.70 . \mathrm{MS}$ (MALDI-TOF) m/z: $523.3(\mathrm{M}+\mathrm{H})^{+}$; HRMS (MALDI) m/z: calcd for $\mathrm{C}_{31} \mathrm{H}_{52} \mathrm{FOS}_{2}{ }^{+1} 523.3438$, found 523.3431.

## Compound 5b

Compound $\mathbf{5 b}$ was synthesized according to the same procedure as for compound $\mathbf{5 a}$. Yield: $80 \% .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 10.10(\mathrm{~s}, 1 \mathrm{H}), 7.29(\mathrm{~s}, 1 \mathrm{H}), 7.26(\mathrm{~s}, 1 \mathrm{H}), 2.62(\mathrm{~d}, J=7.1 \mathrm{~Hz}$, 2H), $1.79(\mathrm{br}, 1 \mathrm{H}), 1.24(\mathrm{br}, 40 \mathrm{H}), 0.90-0.85(\mathrm{~m}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 183.50$, $146.73,144.84,138.63,134.91,129.65,129.26,37.31,34.72,33.65,32.02,32.01,30.01,29.78$, 29.77, 29.75, 29.73, 29.70, 29.46, 29.44, 26.64, 22.78, 14.21. MS (MALDI-TOF) m/z: 505.3 $(\mathrm{M}+\mathrm{H})^{+} ;$HRMS (MALDI) m/z: calcd for $\mathrm{C}_{31} \mathrm{H}_{53} \mathrm{OS}_{2}{ }^{+1} 505.3538$, found 505.3526 .

## Compound 6a

To a 50 mL three-neck flask containing zinc powder ( $1.34 \mathrm{~g}, 20.72 \mathrm{mmol}$ ) and THF ( 20 mL ) was added $\mathrm{TiCl}_{4}(2.24 \mathrm{~g}, 11.8 \mathrm{mmol})$ dropwise under a nitrogen stream. The solution was stirred for 2 h at $0^{\circ} \mathrm{C}$ and compound $5 \mathrm{a}(1.08 \mathrm{~g}, 2.06 \mathrm{mmol})$ was added in one portion. The mixture was refluxed for 5 h and then cooled to room temperature. The reaction mixture was quenched with water and extracted with ethyl acetate. Then the organic layer was washed with brine and dried over $\mathrm{MgSO}_{4}$. After removing the solvent by rotary evaporation, the crude product was purified by silica gel column chromatography (eluent: Petroleum ether) to yield $\mathbf{6 a}(0.92 \mathrm{~g}, 87.7 \%) .{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 6.97$ (s, 2H), 6.96 (s, 2H), 2.59 (d, $J=6.9 \mathrm{~Hz}, 4 \mathrm{H}$ ), 1.80 (br, 2H), 1.25 (br, $80 \mathrm{H}), 0.87(\mathrm{~m}, 12 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 148.58(\mathrm{~d}, J=265.7 \mathrm{~Hz}), 135.37(\mathrm{~d}, J=7.1$ $\mathrm{Hz}), 135.19,127.51(\mathrm{~d}, J=24.5 \mathrm{~Hz}), 123.70,121.84(\mathrm{~d}, J=14.5 \mathrm{~Hz}), 116.23,37.45,34.08$, $33.60,32.04,30.05,29.81,29.79,29.77,29.74,29.48,26.63,14.24 .{ }^{19} \mathrm{~F}$ NMR ( $282 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$-133.44. MS (MALDI-TOF) m/z: $1013.7(\mathrm{M}-\mathrm{H})^{+}$; HRMS (MALDI) $\mathrm{m} / \mathrm{z}$ : calcd for

## Compound 6b

Compound $\mathbf{6 b}$ was synthesized according to the same procedure as for compound $9 \mathbf{a}$. Yield: $58.3 \% .^{1}{ }^{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.15(\mathrm{~s}, 2 \mathrm{H}), 7.06(\mathrm{~s}, 2 \mathrm{H}), 6.94(\mathrm{~s}, 2 \mathrm{H}), 2.63(\mathrm{~d}, J=6.9 \mathrm{~Hz}$, $4 \mathrm{H}), 1.84(\mathrm{br}, 2 \mathrm{H}), 1.25(\mathrm{br}, 80 \mathrm{H}), 0.87(\mathrm{t}, J=6.6 \mathrm{~Hz}, 12 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $143.89,138.90,138.89,134.41,123.10,122.14,119.12,37.34,34.92,33.68,32.05,30.09,29.84$, 29.83, 29.81, 29.80, 29.78, 29.77, 29.50, 26.67, 22.82, 14.23. MS (MALDI-TOF) m/z: 977.6 $(\mathrm{M}-\mathrm{H})^{+} ;$HRMS (MALDI) m/z: calcd for $\mathrm{C}_{62} \mathrm{H}_{105} \mathrm{~S}_{4}{ }^{+1} 977.7099$, found 977.7083.

## Compound 7a

$N$-Bromosuccinimide ( $0.32 \mathrm{~g}, 1.8 \mathrm{mmol}$ ) was added in portions to a solution of compound $\mathbf{6 a}$ $(0.87 \mathrm{~g}, 0.86 \mathrm{mmol})$ in $\mathrm{AcOH}(6 \mathrm{~mL})$ and $\mathrm{CHCl}_{3}(6 \mathrm{~mL})$ at room temperature. After the solution was stirred at room temperature for 1.5 h , water $(10 \mathrm{~mL})$ and sat. $\mathrm{NaHCO}_{3}$ (aq) was added to adjust pH to neutral. The aqueous layer was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and combined organic layer was washed with brine and dried over $\mathrm{MgSO}_{4}$. After removing the solvent by rotary evaporation, the crude product was purified by silica gel column chromatography (eluent: Petroleum ether) to yield compound 7a ( $0.84 \mathrm{~g}, 83.4 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 6.93(\mathrm{~s}, 1 \mathrm{H}), 2.59(\mathrm{~d}, J=7.5$ $\mathrm{Hz}, 2 \mathrm{H}), 1.88(\mathrm{br}, 1 \mathrm{H}), 1.25(\mathrm{br}, 40 \mathrm{H}), 0.87(\mathrm{~m}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3} \delta 147.88(\mathrm{~d}, J=$ $266.8 \mathrm{~Hz}), 134.82,134.04(\mathrm{~d}, J=6.9 \mathrm{~Hz}), 125.98(\mathrm{~d}, J=24.6 \mathrm{~Hz}), 121.33(\mathrm{~d}, J=14.2 \mathrm{~Hz})$, $116.08,112.53,37.25,33.66,33.53,32.04,30.03,29.82,29.79,29.77,29.73,29.49,29.48,26.59$, $22.81,14.24 .{ }^{19} \mathrm{~F}$ NMR ( $282 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-133.01$. MS (MALDI-TOF) m/z: $1171.5(\mathrm{M}-\mathrm{H})^{+}$; HRMS (MALDI) m/z: calcd for $\mathrm{C}_{62} \mathrm{H}_{101} \mathrm{Br}_{2} \mathrm{~F}_{2} \mathrm{~S}_{4}{ }^{+1} 1169.5115$, found 1169.5092.

## Compound 7b

Compound 7b was synthesized according to the same procedure as for compound 10a. Yield: $91.2 \% .^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.07(\mathrm{~s}, 2 \mathrm{H}), 7.03(\mathrm{~s}, 2 \mathrm{H}), 2.62(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 4 \mathrm{H}), 1.92(\mathrm{br}$, $2 \mathrm{H}), 1.25(\mathrm{br}, 80 \mathrm{H}), 0.87(\mathrm{t}, J=5.7 \mathrm{~Hz}, 12 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.98,137.37$, 137.17, 133.97, 121.87, 118.49, 111.32, 37.08, 34.07, 33.60, 31.95, 29.96, 29.74, 29.73, 29.70, 29.68, 29.65, 29.40, 26.93, 26.52, 22.72, 14.14. MS (MALDI-TOF) m/z: 1134.5 (M) ${ }^{+}$; HRMS (MALDI) m/z: calcd for $\mathrm{C}_{62} \mathrm{H}_{104} \mathrm{Br}_{2} \mathrm{~S}_{4}{ }^{+1} 1132.5227$, found 1132.5237.

## Compound F2-QBTTE

Malononitrile ( $0.22 \mathrm{~g}, 3.36 \mathrm{mmol}$ ) was added to an ice-water cooled suspension of sodium hydride ( $0.16 \mathrm{~g}, 6.7 \mathrm{mmol}$ ) in 1,2-dimethoxyethane $(15 \mathrm{~mL})$ under a nitrogen stream. The mixture was stirred at room temperature for another 30 min . Then compound $\mathbf{7 a}(0.70 \mathrm{~g}$, 0.60 mmol ) and tetrakis(triphenylphosphine)palladium ( $42 \mathrm{mg}, 0.036 \mathrm{mmol}$ ) were added. The reaction mixture was refluxed for 6 h . The resulting solution was quenched with water carefully and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. After dried over $\mathrm{MgSO}_{4}$, the solvent was removed by rotary evaporation to afford the product which was used without further purification.

2,3-dichloro-5,6-dicyano-1,4-benzoquinone ( $0.14 \mathrm{~g}, 0.61 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ was added dropwise to the solution of crude product above in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(15 \mathrm{~mL})$. After stirring for 30 min , the reaction mixture was filtrated. Hexane ( 10 mL ) was added to the filtrate to precipitate crude product after removing the most of solvent by rotary evaporation. The crude product was purified
by silica gel column chromatography (eluent: Petroleum ether/ $\mathrm{CH}_{2} \mathrm{Cl}_{2}=3: 1$ ) to yield F2-QBTTE $(0.15 \mathrm{~g}, 22.6 \%) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 6.95(\mathrm{~s}, 2 \mathrm{H}), 2.78(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 4 \mathrm{H}), 1.88(\mathrm{br}, 2 \mathrm{H})$, $1.25(\mathrm{br}, 80 \mathrm{H}), 0.87(\mathrm{t}, J=6.2 \mathrm{~Hz}, 12 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 170.42,152.40,152.33$, $137.99,133.52,133.45,125.78,119.75,114.50,113.14,38.31,33.30,33.04,32.02,32.00,30.06$, $29.78,29.75,29.73,29.72,29.69,29.45,29.43,26.21,22.78,22.77,14.20 .{ }^{19}$ F NMR ( 376 MHz , $\mathrm{CDCl}_{3}$ ) $\delta$-119.54. MS (MALDI-TOF) m/z: $1139.7(\mathrm{M}+\mathrm{H})^{+}$; HRMS (MALDI) m/z: calcd for $\mathrm{C}_{68} \mathrm{H}_{101} \mathrm{~F}_{2} \mathrm{~N}_{4} \mathrm{~S}_{4}{ }^{+1}$ 1139.6872, found 1139.6827. Anal. Calcd for $\mathrm{C}_{68} \mathrm{H}_{100} \mathrm{~F}_{2} \mathrm{~N}_{4} \mathrm{~S}_{4}$ : C, 73.99; H, 9.31; N, 5.08; Found: C, 73.99; H, 9.28; N, 5.00.

## Compound QBTTE

Compound QBTTE was synthesized according to the same procedure as for compound F2-QBTTE. Yield: 63.5\%. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.02(\mathrm{~s}, 2 \mathrm{H}), 7.00(\mathrm{~s}, 2 \mathrm{H}), 2.83(\mathrm{~d}, J$ $=7.3 \mathrm{~Hz}, 4 \mathrm{H}), 1.91(\mathrm{br}, 2 \mathrm{H}), 1.25(\mathrm{br}, 80 \mathrm{H}), 0.87(\mathrm{t}, J=6.3 \mathrm{~Hz}, 12 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 101 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 170.82,158.38,151.70,145.27,131.97,126.18,125.24,124.06,123.50,115.90,114.90$, $113.55,67.65,38.17,33.96,33.02,31.94,31.62,29.99,29.71,29.67,29.66,29.62,29.38,29.37$, 26.20, 22.71, 14.14. MS (MALDI-TOF) m/z: 1103.6 (M+H) ${ }^{+}$; HRMS (MALDI) m/z: calcd for $\mathrm{C}_{68} \mathrm{H}_{103} \mathrm{~N}_{4} \mathrm{~S}_{4}{ }^{+1}$ 1103.7066, found 1103.7047. Anal. Calcd for $\mathrm{C}_{68} \mathrm{H}_{102} \mathrm{~N}_{4} \mathrm{~S}_{4}: \mathrm{C}, 71.66 ; \mathrm{H}, 8.84 ; \mathrm{N}$, 4.92; Found: C, 71.77; H, 8.82; N, 4.72.

## 2. DFT MO Calculations

(a)
F2-QBTTE







$-5.90 \mathrm{eV}$

$-5.76 \mathrm{eV}$

Figure S1. Molecular geometry and frontier molecular orbitals of F2-QBTTE and QBTTE at the DFT-B3LYP/6-31G(d,p) level.

## 3. TGA and DSC Analysis

TGA and DSC was measured on a TAQ500 instruments under a dry N2 flow at a heating rate of $10^{\circ} \mathrm{C} / \mathrm{min}$


Figure S2. Thermal gravity analysis (TGA) of F2-QBTTE


Figure S3. Thermal gravity analysis (TGA) of QBTTE


Figure S4. DSC thermograms of F2-QBTTE and QBTTE
4. HNMR spectra of F2-QBTTE and QBTTE.


Figure S5. HNMR spectra of F2-QBTTE.


Figure S6. HNMR spectra of QBTTE.

Table S1. F2-QBTTE and QBTTE-based FETs annealed at varied temperatures.

|  | $\mathrm{T}_{\text {anneal }}\left({ }^{\circ} \mathrm{C}\right)$ | $\mu_{\mathrm{e}}\left(\mathrm{cm}^{2} \mathrm{~V}^{-1} \mathrm{~s}^{-1}\right)$ | $\mathrm{V}_{\text {th }}(\mathrm{V})$ | $\mathrm{I}_{\text {on }} / \mathrm{I}_{\text {off }}$ |
| :---: | :---: | :---: | :---: | :---: |
| F2-QBTTE | As deposited | 0.05 (0.04 $\pm 0.008)$ | -0.3 | $10^{4}-10^{5}$ |
|  | 80 | 0.16 (0.12 $\pm 0.04)$ | 2.1 | $10^{5}-10^{6}$ |
|  | 100 | $0.4(0.31 \pm 0.05)$ | 0.2 | $10^{4}-10^{5}$ |
|  | 120 | 0.27 (0.21 $\pm 0.05)$ | 1.1 | $10^{4}-10^{5}$ |
| QBTTE | As deposited | $\begin{gathered} 1.6 \times 10^{-3} \\ \left(7.0 \times 10^{-4} \pm 4.4 \times 10^{-4}\right) \end{gathered}$ | 4.2 | $10^{3}-10^{4}$ |
|  | 80 | $0.013(0.01 \pm 0.003)$ | 1.3 | $10^{4}-10^{5}$ |
|  | 100 | $0.021(0.017 \pm 0.003)$ | 3.4 | $10^{4}-10^{5}$ |
|  | 120 | $0.021(0.016 \pm 0.004)$ | 10.9 | $10^{4}-10^{5}$ |

