Supporting Information

Dithieno[3,2-*a*:3',2'-*j*][5,6,11,12]chrysene Diimides: A Versatile

Electron-Deficient Building Block for Polymeric Semiconductors

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Experimental Section

1. Materials and General Methods

Commercially available reagents and solvents were obtained and used as received unless otherwise mentioned. Compounds M1 and 1 were synthesized according to the reported procedures.¹ ¹H NMR (300 MHz or 400 MHz) and ¹³C NMR (100 MHz) spectra were measured on Varian Mercury (300 MHz and 400 MHz) or JEOL NMR (400 MHz) instruments. Elemental analyses were performed on an Elementar Vario EL III elemental analyzer. Mass spectra (EI, DART, ESI and MALDI-TOF) were carried out on Thermo Fisher Scientific LTQ FT Ultra Mass Spectrometer, Waters Micromass GCT premier or Agilent Technologies 5973N. Optical absorption spectra were measured on a U-3900 UV-vis spectrophotometer. Thermogravimetric analysis (TGA) measurements were conducted on a TGA O500 instrument under a dry nitrogen flow at a heating rate of 10 °C/min, heating from room temperature to 500 °C. Differential scanning calorimetry (DSC) measurements were performed on a DSC O2000 instrument under a dry nitrogen flow at a heating rate of 10 °C/min, heating from room temperature to 350 °C. Cyclic voltammetry (CV) was carried out on CHI610D instruments using $Bu_4NPF_6(0.1 \text{ M})$ as supporting electrolyte and ferrocene as an internal reference at a scan rate of 100 mV/s. The CV cell consisted of a platinum button working electrode, a platinum wire counter electrode, and an Ag/AgNO₃ reference electrode. Melting point was measured on SGW X-4 or WRS-1A microscopic melting point apparatus. X-ray diffraction (XRD) measurements were carried out in the reflection mode with Cu Ka radiation using a 2-kW Rigaku X-ray diffraction system. Atomic force microscopy (AFM) was recorded on a Bruker Inova atomic microscope in tapping mode with a silicon tip. Transmission electron microscopy (TEM) measurements were performed in a JEM-2100 instrument.

2. Synthesis and characterizations



Compound 2: Compound **M1** (130 mg, 0.11 mmol) was dissolved in 20 mL DCM solution. Bromine (43 mg, 0.28 mmol) was then added to the solution and stirred at 40 °C for 3 h. 20 mL Na₂SO₃ aqueous solution was added to quench the reaction. The aqueous phase was removed and the organic phase was dried by anhydrous Na₂SO₄. After removing the DCM solvent under reduced pressure, the crude product was purified by silica gel column chromatography, using the mixture of petroleum ether and dichloromethane (4:1) as the eluent. 118 mg red solid product was obtained (yield: 80%) Mp: 120-121 °C. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 9.25 (s, 2H), 8.54-8.57 (d, J = 8.8 Hz, 2H), 7.97-7.99 (d, J = 8.8 Hz, 2H), 3.64-3.66 (d, J = 7.2 Hz, 4H), 1.94 (br, 2H), 1.22-1.26 (m, 80H), 0.85-0.86 (m, 12 H). ¹³C NMR (100 MHz, CDCl₃) δ (ppm) 167.9, 167.9, 143.5, 133.5, 131.5, 130.2, 130.2, 129.4, 127.4, 126.7, 122.1, 121.2, 115.8, 43.1, 37.0, 32.0, 31.9, 31.5, 30.1, 29.7, 29.6, 29.4, 26.3, 22.7, 14.1.



¹H NMR of compound **2**





P1: To a mixture of compound **1** (192.7 mg, 0.14 mmol) and 5,5'bis(trimethylstannyl)bithiophene (68 mg, 0.14 mmol) in 20 mL toluene solution, Pd₂(dba)₃ (3 mg, 2% mmol) and P(o-tol)₃ (3 mg, 4% mmol) was added under N₂ atmosphere. The mixture was stirred at 110 °C for 36 h. 1 mL KF solution was added to quench the reaction. The mixture was dropped into 200 mL methanol solution to get a green precipitate. The precipitate was filtered and subjected to Soxhlet extraction successively with methanol, acetone, hexane and ethyl acetate to remove the oligomers and impurities. The residue was extracted by DCM to obtain 140 mg dark-green solid (yield: 78%). M_n = 24305, PDI = 1.65. ¹H NMR (400 MHz, CDCl₃) δ (ppm) 8.85-9.76 (br, 2H), 7.42-8.29 (br, 4H), 3.02-3.99 (br, 4H), 0.46-1.47 (br, 92 H). Anal. Calcd for C₈₂H₁₁₀N₂O₄S₄: C, 74.84; H, 8.43; N, 2.13. Found: C, 74.51; H, 8.50; N, 1.76.



¹³C NMR of polymer **P1**



P2: To a mixture of compound **2** (107 mg, 0.081 mmol) and 5,5'bis(trimethylstannyl)bithiophene (40 mg, 0.081 mmol) in the 5 mL toluene solution, Pd₂(dba)₃ (3 mg, 4% mmol)and P(o-tol)₃ (3 mg, 8% mmol) was added under N₂ atmosphere. The mixture was stirred at 100 °C for 10 minutes. 1 mL KF solution was added to quench the reaction. The mixture was dropped into 200 mL methanol solution to get a green precipitate. The precipitate was filtered and subjected to Soxhlet extraction successively with methanol, acetone, hexane and ethyl acetate to remove the oligomers and impurities. The residue was extracted by dichlorobenzene to obtain 86 mg dark-green solid (yield: 80%). M_n = 65775, PDI = 3.13. Anal. Calcd for $C_{82}H_{110}N_2O_4S_4$: C, 74.84; H, 8.43; N, 2.13. Found: C, 74.11; H, 8.13; N, 1.96.

3. Fabrication and measurement of BGTC OFETs

The Si/SiO₂ substrates with a capacitance of 10 nF cm⁻² was used as the gate electrode and dielectric layer. Thin films of **P1** and **P2** were deposited on octadecyltrichlorosilane (OTS)-treated Si/SiO₂ substrates by spin-coating their respectively chloroform or chlorobenzene solutions (5 mg mL⁻¹). Gold, used as source and drain electrode, were deposited on the top of the active layer through a shadow mask under high vacuum (~10⁻⁷ Torr), affording a bottom-gate top-contact (BGTC) device configuration. The channel length (L) and width (W) were 31 µm and 273 µm, respectively. The electric characteristics of the OFET devices were measured by using a Keithley 4200-SCS semiconductor parameter analyzer in a glovebox with a nitrogen atmosphere. The fieldeffect mobility was calculated in the saturation regime according to the equation I_{DS} = $(\mu WC_i/2L)(V_G - V_T)^2$, where I_{DS} is the drain-source current, V_G is the gate voltage, V_T is the threshold voltage, µ is the field-effect mobility, W is the channel width, L is the channel length, and C₁ is the capacitance per unit area of the gate dielectric layer.

4. Fabrication and measurement of all-polymer solar cells

PTB7-Th and **P1** with a weight ratio of 1:1 were co-dissolved in chlorobenzene, the total material concentration was 18 mg mL⁻¹, different amounts of DIO was added as solvent additive for the blend films. The all-polymer solar cells were fabricated with a structure of ITO/ZnO/active layer/MoO₃/Ag. A thin layer (~30 nm) of ZnO sol-gel was spin-coated onto ITO glass and annealed at 180 °C in ambient condition for 30 min then transferred into glovebox immediately. Active layers (the as prepared solutions) were then spin-coated in the glovebox. Finally, MoO₃ (~10 nm) and Ag (80 nm) anodes were thermal evaporated through a shadow mask in glovebox at a chamber pressure of ~2.0

 $\times 10^{-6}$ mbar. The active area of the device was 4 mm². Current density-voltage (*J-V*) curves were detected on a Keithley 2400 source meter. Photocurrent was obtained upon irradiation using a solar simulator (XES-70S1, SAN-EI ELECTRIC CO., LTD.), with AM 1.5 G filter. Light intensity was calibrated to 100 mW cm⁻² using a NREL-certified standard silicon cell (AK-200, Konica Minolta, INC.). External quantum efficiency (EQE) was recorded by a Newport quantum efficiency test model 77890 (Newport Co. Ltd.) with a Xe lamp.

5. Density functional theory (DFT) calculations

To investigate the relationship between the molecular structure and optoelectronic performance, the geometry optimization was performed using the Gaussian 16 package with the B3LYP hybrid density functional and the 6-31G(d,p) basis set. The long side chains were replaced by methyl groups, and the trimers were utilized to represent the polymeric backbone in order to simplify the calculations.²

The full optimization was carried out based on assumed planar conformation. Strong geometric twist taken place around the inter-ring C-C bonds. The non-planar structures at 298.15 K and 1 atm were estimated from the gas-phase studies. Harmonic vibration frequency calculations at the same level were performed to verify all stationary points as local minima (with no imaginary frequency).

Considering one of the most ideal molecular configuration, the structural planarization was very important factor to improve the charge transfer property. The planar structures were also optimized by constraining the molecules to planar conformations (thus, freezing out he torsional motions along the long polymer chains). All the optimized geometry and frontier molecular orbitals are listed in the Table S5-8 and Figure S8.



Figure S1 GPC distribution plots of polymer P1 with tetrahydrofuran as eluent at room temperature.



Figure S2 GPC distribution plots of polymer P2 with 1,2,3-trichlorobenzene as eluent at 150 °C.



Figure S3 TGA plots of polymers P1 and P2 under a nitrogen flow with a heating rate of 10 °C/min.



Figure S4 DSC plots of polymer P1.



Figure S5 DSC plots of polymer P2.



Figure S6 UV-vis absorption spectra of polymer P1 in dichloromethane solution and in thin film.



Figure S7 UV-vis absorption spectra of polymer P2 in chlorobenzene solution and in thin films.



Torsion angle (C1-C2-C3-C4): 23.45°

Figure S8 Molecular geometries and torsion angles of model molecules of P1 and P2 obtained by DFT calculations.



Figure S9 Cyclic voltammograms of thin films of P1 and P2 in 0.1 M $Bu_4N^+PF_6^-$ solution in acetonitrile at a scan rate 100 mV/s.



Figure S10 Typical transfer characteristics of OFETs based on P2.



Figure S11 X-ray diffraction (XRD) patterns of as-spun and after thermal annealing at 80 °C, 120 °C, 160 °C, 200 °C thin films of P1 (a) and P2 (b).



Figure S12 AFM images of thin films of **P1** (a-e) and **P2** (f-j), as-spun (a, f),after thermal annealing at 80 °C (b, g), 120 °C (c, h), 160 °C (d, i), 200 °C (e, j), respectively.



PTB7-Th

Figure S13 Chemical structure of PTB7-Th.



Figure S14 SCLC fittings of the hole-only device based on the PTB7-Th/P1 blend films with 1% (v/v) DIO.



Figure S15 SCLC fittings of the electron-only device based on the PTB7-Th/P1 blend films with 1% (v/v) DIO.



Figure S16 X-ray diffraction (XRD) patterns of PTB7-Th/**P1** as-cast, with 0.5% (v/v) DIO, with 1.0% (v/v) DIO, with 1.5% (v/v) DIO films.



Figure S17 AFM height and phase images $(2 \times 2 \mu m^2)$ of PTB7-Th/P1 blend films with 0.5% (v/v) DIO (a,c) and with 1.5% (v/v) DIO (b,d).



Figure S18 TEM images of PTB7-Th/P1 as-cast (a), with 0.5% (v/v) DIO (b), with 1.0% (v/v) DIO (c) and with 1.5% (v/v) DIO (d) films, respectively.

 Table S1. Summarize of experimental and calculated optoelectronic parameters of P1 and P2.

Polymer	λ _{max} (nm) soln/film	E _{LUMO} ^a (eV)	E _{HOMO} ^b (eV)	<i>E_g^{opt c}</i> (eV) soln/film	<i>E_g^{cv}</i> (eV)
P1	607/610	- 4.01	- 6.05	1.76/1.71	2.04
P2	638/640	- 3.96	- 5.76	1.63/1.59	1.80

 ${}^{a}E_{\text{LUMO}} = -(E_{\text{onset}}, {}^{\text{red}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{\text{ox}} - E_{1/2}(\text{Fc/Fc}^{+}) + 5.10) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{2}{}^{b}E_{\text{HOMO}}) \text{ (eV)}. {}^{2}{}^{b}E_{\text{HOMO}} = -(E_{\text{onset}}, {}^{2}{}^{b}E_{\text{HOMO}})$

5.10) (eV).³ ^cObtained from the onset wavelength of solution/film absorption.

Polymer	Annealing Temperature (°C)	<i>V</i> _T (V) (Ave.)	$I_{ m on}/I_{ m off}$	μ _e (cm ² /V.s) Max. (Ave.)
P1	As-spun	31.6 (35.3)	104~105	1.8 (1.6) ×10 ⁻²
	80	24.4 (27.8)	~105	2.9 (2.1) ×10 ⁻²
	120	28.3 (29.7)	105~106	4.7 (4.6) ×10 ⁻²
	160	23.8 (24.5)	10 ⁵ ~10 ⁶	6.4 (6.3) ×10 ⁻²
	200	24.1 (25.0)	~10 ⁶	0.11 (0.10)

Table S2 Characteristics of OFET device based on P1 at different annealingtemperatures.

 Table S3
 Characteristics of OFET device based on P2 at different annealing temperatures.

Polymer	Annealing Temperature (°C)	V _T (V) (Ave.)	I _{on} /I _{off}	$\mu_{\rm h}({\rm cm}^2/{\rm V.s})$ Max. (Ave.)	μ _e (cm ² /V.s) Max. (Ave.)
As-spun 80 P2 120 160 200	A a <i>a</i> mum	-46.7 (-48.9)	~10 ³	1.84 (1.66) ×10 ⁻⁴	
	As-spun	55.0 (57.8)	~10 ³		3.59 (3.50) ×10 ⁻⁴
	80	-46.2 (-48.0)	10 ³ ~10 ⁴	2.86 (2.71) ×10 ⁻⁴	
	45.2 (56.0)	10 ³ ~10 ⁴		6.16 (5.92) ×10 ⁻⁴	
	120	-40.5 (-42.7)	10 ³ ~10 ⁴	4.81 (4.54) ×10 ⁻⁴	
	120	42.5 (45.7)	10 ³ ~10 ⁴		9.82 (9.06) ×10 ⁻⁴
	160	-38.7 (-41.9)	~104	6.25 (5.97) ×10 ⁻⁴	
	160	39.9 (43.8)	~104		1.16 (1.15) ×10 ⁻³
	200	-30.8 (-32.4)	~104	8.70 (8.28) ×10 ⁻⁴	
	200	35.9 (40.1)	~104		1.79 (1.71) ×10 ⁻³

Active Layer (1:1)	DIO (v/v,%)	V _{oc} (V)	J _{SC} (mA cm ⁻²)	FF (%)	PCE (%)
	0	1.01	9.66	42.5	4.15
	0.5	1.00	11.18	49.5	5.53
PTB7-Th/P1	1	0.99	12.43	61.1	7.52
	1.5	0.98	12.00	43.3	5.09

 Table S4. Characteristics of all-polymer solar cells based on PTB7-Th/P1 blend films.

Table	S5.	Optimized	geometry	and	frontier	molecular	orbitals	of P1	(non-planar
structu	re)								





Table S6. Optimized geometry and frontier molecular orbitals of P2 (non-planar structure)



 Table S7. Optimized geometry and frontier molecular orbitals of P1 (planar structure)



 Table S8. Optimized geometry and frontier molecular orbitals of P2 (planar structure)

Cartesian Coordinates (in Å)

P1 (non-p	lanar	structure)	
\			~ /	

C	15.975296	-0.377302	1.656441
С	15.331470	-0.379075	0.424261
С	16.032956	-0.211438	-0.809799
С	17.445566	-0.330875	-0.720201
С	15.413024	0.053546	-2.096028
С	16.136950	-0.157986	-3.319505
С	17.457263	-0.685915	-3.166730
С	18.099764	-0.706305	-1.934266
С	14.968817	-0.817782	2.690390
N	13.782936	-1.056111	2.005388
С	13.916108	-0.848446	0.634841
Ċ	18.359127	-1.375909	-4.159480
Ň	19.495104	-1.760494	-3.450850
С	19.411495	-1.426686	-2.104223
0	20.262048	-1.741296	-1.290852
Õ	18 199241	-1 631682	-5 339057
0	15 075393	-1 003232	3 889273
Ő	13 026845	-1 072636	-0 166698
C	13 471379	0.964677	-3 311062
C	14 174984	0 730275	-4 519898
C	15 485617	0 197498	-4 552715
C	14 128335	0.634971	-2 134809
Н	13 643110	0.885282	-1 206234
C	20 189901	0.552683	1 706409
C	19 464382	0.530739	2 924769
C	18 080976	0.239528	2.921709
C	17 368541	-0.080121	1 778421
C	18 110481	-0.086676	0 547370
C	19 477543	0.260903	0.552512
н	19 993362	0.354487	-0 388463
C	17 590115	0.345409	4 337441
C	15 981110	0.124979	-5 903286
C	15 082206	0.124777	-6.827554
C	18 556058	0.667714	5 235037
ç	20 12/0/7	0.869/20	1 510023
S	13 576877	1 070015	-6 130056
ы	16 566551	0.152965	-0.130030 1 611700
и П	16 05/1172	0.152705	6 160/22
Н	15 209996	0.584272	-7 900760
C	12 582458	1 567481	-7.500700
с н	12.302450	-2 /00370	2.042343
и П	12.800700	-2.499370	3.109473
и П	12.192517	-0.042210	1.857621
	11.04/393 20 502720	-1.743323	-/ 020567
с ц	20.373/30	-2.3199/3	-4.020307
и П	21.320003	-1.755149	-3.333133
п U	20./10393	-3.404340 2711264	-J.40J27/ 5 065561
п	20.331404	-2.711304	-3.003304 6 202075
п	10.444200	0.190333	0.3020/3

С	12.116231	1.513274	-3.215479
С	11.167431	1.154968	-2.278516
S	11.495970	2.792591	-4.246209
С	9.954705	1.873979	-2.390596
Н	11.341739	0.366289	-1.555383
С	9.953658	2.802221	-3.413417
Н	9.091596	1.695842	-1.759256
С	8 899775	3 702668	-3 824856
Č	8.761022	4.394843	-5.012058
S	7 554684	4 050623	-2 756201
Č	7 589977	5 185468	-5 074219
H	9 469979	4 308660	-5 827696
C	6 806422	5 110256	-3 939767
Н	7 298346	5 773124	-5 937361
C	29 968513	9 288693	3 642147
C	29.879369	7 915399	3 464413
C C	30 665165	7 215509	2 497403
C C	31 742662	7 960340	1 946153
C C	30 440507	5 845723	2 069785
C	31 466648	5 111420	1 382883
C	32 709519	5 801611	1.202003
C	32.709519	7 169360	1.222731
C C	20 10/701	9.635657	1.455557
C N	29.194791	8 437600	5 378307
C N	28.085555	7 3/7000	1 603567
C	29.070802	5 275025	4.005507
U N	34.093092	5.275925	0.923007
IN C	34.943038	0.374282	0.979103
C	34.272014	7.343224 9.605779	1.515/41
0	24.03/940 24.405214	0.00 <i>3</i> 770 4 142550	1.303162
0	34.463314	4.142339	0./15088
0	29.023338	6 102267	J.4J/290
0 C	28.803010	0.19330/	4.888034
C	28.833801	3.994343	1.808432
C	29.854062	3.261283	1.1506//
C	31.145/10	5./89388	0.914841
U U	29.159594	5.2/58/4	2.227630
П	28.3/1801	5.8/6484	2.650822
C	32.412966	11.502633	0.90/982
C	31.523535	12.19/2/2	1./408/6
C	30.683853	11.546795	2.6/1801
C	30.760006	10.113/03	2.781364
C	31.692078	9.410654	1.939888
C	32.485977	10.135588	1.011927
H	33.142731	9.599335	0.346303
H	33.016304	12.030499	0.176723
C	29.820415	12.494915	3.330835
C	31.958388	2.858934	0.174063
C	31.321852	1.692015	-0.100409
С	30.020236	13.775609	2.922973
S	31.262259	13.923597	1.709233

S	29.695561	1.637992	0.513413
Н	29.108452	12.225181	4.093985
Н	32.981113	3.053168	-0.104739
Н	31.716884	0.837236	-0.632039
С	27.932741	8.336759	6.615985
Н	28.501890	8,770895	7.441483
Н	26 981236	8 867955	6 529668
Н	27 751868	7 278465	6 802643
C	36 387571	6 297197	0.826265
Н	36 717411	6 940628	0.006780
Н	36 885900	6 617858	1 744647
Н	36 638335	5 259164	0.609465
H	29 509075	14 661814	3 272898
C II	27.307073	3 500853	2 081583
C	26 730551	3 790516	2.001303
c c	26.750551	2 508115	0.070817
S C	20.331811	2.308113	2 100181
С U	23.439007	3.214190	3.190181
П	27.123723	4.577170	4.023002
	23.1/3003	2.4/3814	2.033301
П	24./31333	3.310343	4.003810
C	23.9/5/00	1./4/190	1.099971
C	23.805246	0.764341	0.744587
5	22.461250	2.066335	2.522528
C	22.484518	0.263061	0.6/5915
H	24.619569	0.393883	0.132268
C	21.619570	0.846808	1.580184
H	22.169758	-0.531236	0.008504
С	2.067970	8.343952	-3.266224
С	2.195457	9.691666	-3.582524
С	3.385115	10.237647	-4.156635
С	4.311782	9.278914	-4.646405
С	3.685299	11.653791	-4.271795
С	4.701830	12.119185	-5.175163
С	5.332494	11.105659	-5.962579
С	5.193839	9.754728	-5.665462
С	0.620937	8.098738	-2.917255
Ν	-0.026172	9.321851	-3.049647
С	0.835359	10.329521	-3.476261
С	6.134664	11.209568	-7.235705
Ν	6.432189	9.902753	-7.615464
С	5.881485	8.966155	-6.748536
0	5.954666	7.765183	-6.939613
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0	0.027794	7.079215	-2.613980
0	0.467353	11.461457	-3.734251
С	3.303538	13.929448	-3.404240
С	4.286561	14.392402	-4.315613
С	4.997356	13.527734	-5.181552
С	3.051990	12.565354	-3.401278
Н	2.371250	12.186278	-2.657398

С	5.562297	5.849404	-3.710887
С	4.433650	5.353043	-3.010136
С	3.255391	6.113149	-2.819195
С	3.167279	7.436480	-3.378320
С	4.300111	7.928127	-4.113706
С	5.461237	7.134533	-4.222875
Н	6.337801	7.557242	-4.684585
С	2.295877	5.390389	-2.024514
С	5.976146	14.254049	-5.949144
С	5.975290	15.587340	-5.694452
С	2.720316	4.151395	-1.667740
S	4.309719	3.774007	-2.263666
S	4.795498	16.056886	-4.506594
Н	1.323451	5.778191	-1.768236
Н	6.621789	13.793505	-6.678728
Н	6.604692	16.342542	-6.144853
С	-1.456645	9.491691	-2.867774
Н	-2.005444	8.805574	-3.517413
Н	-1.736836	9.290426	-1.830513
Н	-1.697745	10.522854	-3.124870
С	7.139226	9.548040	-8.833037
Н	8.063325	9.013600	-8.597265
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С	2.538647	14.776260	-2.485303
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S	3.209582	16.150090	-1.620320
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Н	0.613013	13.798372	-2.555758
С	1.689091	16.462690	-0.807638
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С	1.560494	17.572092	0.114866
С	2.351710	18.694532	0.239272
S	0.262367	17.613086	1.297348
С	1.920180	19.583413	1.263178
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P2 (non-planar structure)

С	11.110521	13.807364	-10.628859
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С	14.543103	12.390055	-9.759447
С	15.875335	12.398641	-10.303217
С	16.046757	13.170801	-11.495003

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С	11.891133 13.703863 -8.421266
С	17.298164 13.734229 -12.121447
Ν	16.879110 14.491053 -13.211229
С	15.492690 14.503848 -13.339238
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C	18.201930 11.398/48 -9.984239
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Н	5.937892 14.971941 -11.795393
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N	28 640130	1 098137	-6 429942
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C	29.651653	6.223236	-9.278601
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С	-6.102748 13.758232 -14.984229
Н	-7.074470 13.896477 -14.538579
С	-8.976168 8.346110 -18.704952
С	-10.270092 8.698654 -18.303402
С	-10.574484 9.950107 -17.720566
С	-9.512758 10.909305 -17.558634
С	-8.190189 10.547708 -17.996148
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Н	-6.959256 8.973433 -18.822871
Н	-5.109034 14.747922 -13.368955
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С	-11.946601 10.007115	-17.315160
С	-2.142135 13.132176	-16.353422
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С	-12.666347 8.875324	-17.605053
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Н	-1.766799 12.599480	-17.211539
С	-14.074420 8.659074	-17.351358
С	-15.050683 9.617381	-17.165058
S	-14.765492 7.051678	-17.214944
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Н	-20.755243 5.690505	-15.841831
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Н	-2.521814 11.301704	-21.287127

P1 (planar structure)

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С	20.399735	5.804432	-2.663949
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С	20.097209	4.826229	2.830753
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Н	21.750792	6.399637	-5.125435
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Н	9.876102	7.036534	0.000124
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С	8.513442	7.721836	-3.938997
S	7.568999	7.598584	-1.545910
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Н	9.211774	7.700635	-4.766672
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Н	6.611862	8.243124	-5.001234
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Ĥ	29.365465	3.254599	4.103442

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S	24.472159	4.371359	2.320990
С	24.005588	4.671182	-0.210949
Н	26.100325	4.364550	-0.909828
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Н	23.500367	4.851962	-1.150555
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H	-2.330/26	9.568054	-4.070600
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