

Electronic Supporting Information for:

High pressure synthesis of quasi-one-dimensional GdFeO₃-type perovskite PrCuO₃ with nearly divalent Cu ions

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1. High pressure synthesis

We have performed high pressure oxygen annealing for the oxygen deficient perovskite $\text{PrCuO}_{2.5}$ at several temperatures 420-500 °C under a high pressure of 7.5 GPa. Since NaClO_3 does not release oxygen below 420 °C and the impurity phase of PrO_2 appears above 450 °C, we concluded that the optimum annealing temperature is about 440 °C at a high pressure of 7.5 GPa. Furthermore, we have confirmed that the oxidizer NaClO_3 should be mixed with the oxygen deficient perovskite $\text{PrCuO}_{2.5}$ to obtain the fully oxidized PrCuO_3 .

The XRD patterns of $\text{PrCuO}_{2.5}$ and PrCuO_3 , which was obtained by the high-pressure oxygen annealing for $\text{PrCuO}_{2.5}$ at 450 °C and 7.5 GPa, are shown in Fig. S1. All the peaks for $\text{PrCuO}_{2.5}$ can be indexed with an orthorhombic unit cell ($Pbam$) with $a = 5.584 \text{ \AA}$, $b = 10.353 \text{ \AA}$, $c = 3.828 \text{ \AA}$. The main peaks for PrCuO_3 can be indexed with an orthorhombic unit cell ($Pbnm$) with $a = 5.301 \text{ \AA}$, $b = 6.245 \text{ \AA}$, $c = 7.278 \text{ \AA}$. The 020 reflection with asterisk is superimposed by the main reflection from the impurity phase of PrO_2 , which is absent in the sample obtained at 440 °C and 7.5 GPa.

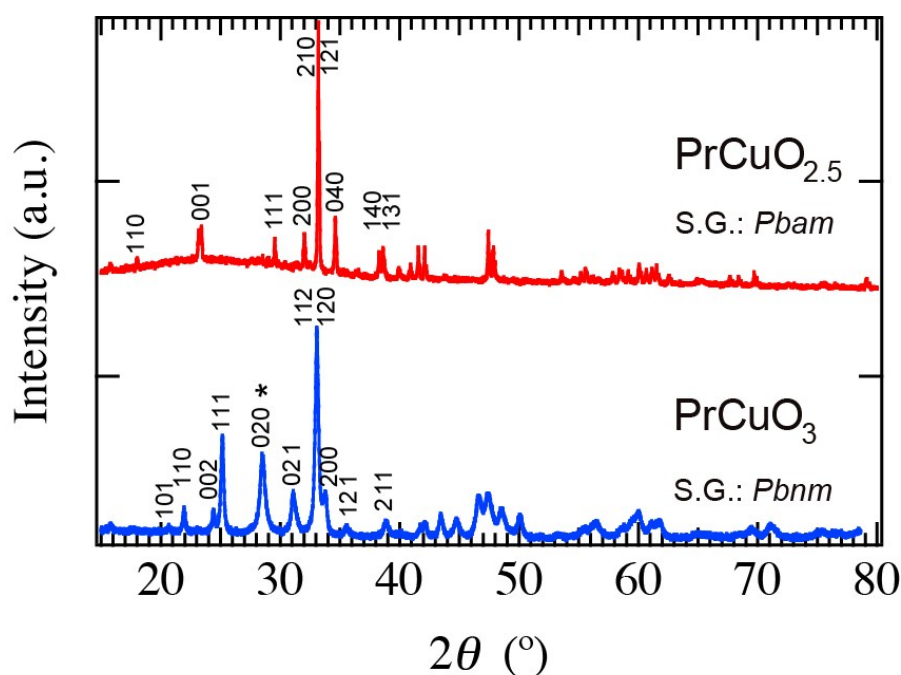


Fig. S1: XRD patterns for $\text{PrCuO}_{2.5}$ and PrCuO_3 measured by a Cu $K\alpha$ radiation.

2. Linear interpolation analyses for XANES spectra

As shown in Fig. 4, the pre-edge structure below 8995 eV, which corresponds to the quadruple transition, is prominent especially for CuO. In order to minimize the effect of the pre-edge structure on the estimation of the absorption edge energy, we chose the intersection of the horizontal line for the linear interpolation analyses as shown in the inset of Fig. 4 (the horizontal line is located at 70% of the maximum value). On the basis of the energies of intersections for CuO and LaCuO₃, we define that the absorption edge energies for Cu²⁺ and Cu³⁺ states are 8996.5 eV and 8997.6 eV, respectively. Provided that the oxidation state of Cu is linearly proportional to the absorption edge energy (the linear interpolation analysis),²⁵ the oxidation state of Cu in PrCuO₃ is calculated to be +2.2.