## **Electronic Supplementary Information**

# Unexpected Solvent Effect on the Binding of Positively-Charged Macrocycles to Neutral Aromatic Hydrocarbons

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### **1. General Method**

All the reagents involved in this research were commercially available and used without further purification unless otherwise noted. Solvents were either employed as purchased or dried prior to use by standard laboratory procedures. Thin-layer chromatography (TLC) was carried out on 0.25 mm Yantai silica gel plates (60F-254). Column chromatography was performed on silica gel 60 (Tsingdao 40 – 63 nm, 200 – 300 mesh). <sup>1</sup>H, <sup>13</sup>C, <sup>1</sup>H, <sup>1</sup>H-COSY, and <sup>1</sup>H, <sup>1</sup>H-ROESY NMR spectra were recorded on Bruker Avance-400, 500 spectrometers. All chemical shifts are reported in ppm with residual solvents as the internal standards. The following abbreviations were used for signal multiplicities: s, singlet; d, doublet; t triplet; m, multiplet. Electrospray-ionization time-of-flight high-resolution mass spectrometry (ESI-HRMS) experiments were conducted on an applied Q EXACTIVE mass spectrometry system. Compound **3** were synthesized according to the literature procedures.<sup>1</sup>

**Quantum chemistry calculations** were performed using Gaussian 09 package.<sup>2</sup> The structure of the host-guest complex between **9** and **1b** was optimized by employing density functional theory (DFT) with dispersion corrected method (wB97XD)<sup>3</sup> in combination with 6-31G\* basis set including the solvent effects for  $CH_2Cl_2$  using the SMD solution model.<sup>4</sup> Minima were characterized by the absence of imaginary frequencies.

### 2 .Synthetic Procedures



To the solution of imidazole (0.26 g, 3.82 mmol) in dry THF (100 mL) was added NaH (0.18 g, 7.64 mmol) at 0 °C and the resulting mixture was stirred for 30 min at 0 °C. **3** (0.5 g, 0.764 mmol) was then added. After stirring for additional 4 h at room temperature, the mixture was filtered. The filtrate was rotavapped under vacuum to remove the solvent. The solid was washed with diethyl ether to afford **4** (0.44 g, 92%). m.p.=225 °C. <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN, 298 K)  $\delta$  [ppm] = 8.68 (d, *J* = 9.4 Hz, 2H), 7.81 (d, *J* = 9.3 Hz, 2H), 7.51 – 7.46 (m, 4H), 7.16 (d, *J* = 9.2 Hz, 2H), 6.88 (s, 2H), 6.75 (s, 2H), 6.27 – 6.25 (m, 1H), 5.53 – 5.42 (m, 5H), 4.15 (m, 4H), 2.47 – 2.43 (m, 2H), 1.83 – 1.75 (m,

4H), 1.49 (m, 7.4 Hz, 4H), 0.96 (t, J = 7.4 Hz, 6H). <sup>13</sup>C NMR (126 MHz, CD<sub>3</sub>CN, 298 K)  $\delta$  [ppm] = 153.92, 149.01, 137.43, 129.19, 128.66, 126.73, 126.07, 122.99, 120.09, 119.97, 119.32, 114.83, 92.03, 69.23, 40.68, 31.68, 19.42, 13.52. ESI-HRMS: m/z calcd for [M+2H]<sup>2+</sup> C<sub>39</sub>H<sub>42</sub>N<sub>4</sub>O<sub>4</sub><sup>2+</sup>, 315.3128; found 315.3118 (error = - 3.2 ppm).



<sup>1</sup>H NMR spectrum (500 MHz, CD<sub>3</sub>CN, 298 K) of Compound 4









#### Naphthotube 2a and 2b



400 mL MeCN was added into a 1000-mL three-neck flask charged with a magnetic stirring bar. The flask was then evacuated and refilled with Ar (using a gas balloon). The solutions of **3** (1.4 g, 2.14 mmol, in 60 mL MeCN ) and **4** (1.35 g, 2.14 mmol, in 60 mL MeCN) were added dropwise during 10 h by using two separate syringes to the flask *via* a double-channel syringe pump. The reaction mixture was stirred for another 24 h at 70 °C. After removing most of the solvent in vacuum, 100 ml of methanol and 10 ml of the aqueous solution of  $NH_4PF_6(1.6 \text{ g}, 4.28 \text{ mmol})$  were added to the mixture which was stirred for another 2 h. The suspension is filtered. Collecting the filter cake was collected and washed repeatedly with acetone to afford **2b** (1.0 g, 33%) as a white solid. The solvent in the filtrate was removed undergo vacuum, and the residue was washed repeatedly with a small amount of acetone and ethyl acetate to afford **2a** (0.42 g, 14%) as a white solid.

**2a:** m.p. >320 °C. <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN, 298 K)  $\delta$  [ppm] = 9.19 (s, 1H), 8.70 (d, *J* = 9.4 Hz, 2H), 7.65 (d, *J* = 9.1 Hz, 2H), 7.48 (d, *J* = 9.4 Hz, 2H), 7.23 (d, *J* = 9.1 Hz, 2H), 6.61 (s, 2H), 6.30 (s, 1H), 5.79 (d, *J* = 14.9 Hz, 2H), 5.58 (d, *J* = 14.9 Hz, 2H), 5.40 (s, 1H), 4.19 (m, 4H), 2.41 (s, 2H), 1.80 (m, 4H), 1.49 (m, 4H), 0.95 (t, *J* = 7.3 Hz, 6H). <sup>13</sup>C NMR (126 MHz, CD<sub>3</sub>CN, 298 K)  $\delta$  [ppm] = 154.77, 149.48, 136.37, 129.35, 128.02, 127.26, 122.81, 120.70, 115.05, 114.70, 92.30, 69.87, 44.36, 32.05, 26.56, 23.06, 19.93,

14.06. ESI-HRMS: m/z calcd for  $[M-2PF_6^-]^{2+} C_{72}H_{74}N_4O_8^{2+}$ , 561.2748; found 561.2746 (error = - 0.4 ppm).



<sup>1</sup>H NMR spectrum (500 MHz, CD<sub>3</sub>CN, 298 K) of Naphthotube **2a** 





**2b:** m.p. >320 °C. <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN, 298 K)  $\delta$  [ppm] = 8.99 (s, 1H), 8.71 (d, J = 9.5 Hz, 2H), 7.80 (d, J = 9.3 Hz, 2H), 7.41 (d, J = 9.5 Hz, 2H), 7.18 (d, J = 9.2 Hz, 2H), 6.59 (d, J = 1.5 Hz, 2H), 6.28 (s, 1H), 5.73 – 5.60 (m, 4H), 5.45 (s, 1H), 4.16 (m, 4H), 2.46 (s, 2H), 1.80 (m, 4H), 1.49 (m, 4H), 0.96 (t, J = 7.4 Hz, 6H). <sup>13</sup>C NMR (126 MHz, CD<sub>3</sub>CN, 298 K)  $\delta$  [ppm] = 155.04, 149.64, 136.74, 129.36, 127.95, 127.19, 122.77, 122.48, 120.96, 120.70, 114.96, 114.88, 92.35, 69.74, 44.28, 32.04, 26.49, 23.12, 19.90, 14.10. ESI-HRMS: m/z calcd for [M-2PF<sub>6</sub><sup>-</sup>]<sup>2+</sup> C<sub>72</sub>H<sub>74</sub>N<sub>4</sub>O<sub>8</sub><sup>2+</sup>, 561.2748; found 561.2746 (error = - 0.4 ppm).



<sup>1</sup>H NMR spectrum (500 MHz, CD<sub>3</sub>CN, 298 K) of Naphthotube **2b** 



ESI-HRMS mass spectrum of Naphthotube 2b

Naphthotube 1a and 1b



To the solution of **2a** or **2b** (300 mg, 0.21 mmol) in  $CH_2Cl_2$  (100 mL) and  $H_2O$  (100 mL) was added NaBArF (558 mg, 0.63 mmol) at room temperature. The resulting mixture was stirred for 2 h, and was then extracted with  $CH_2Cl_2$ . The combined organic layers ( $CH_2Cl_2$ ) were washed with saturated solution of NaHCO<sub>3</sub>. After removing solvent under vacuum, the residue was washed with diethyl ether to afford **1a** or **1b** (600 mg, 99%).

**1a:** m.p. =130 °C. <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN, 298 K) δ [ppm] = 9.15 (s, 1H), 8.70 (d, *J* = 9.5 Hz, 2H), 7.69 (s, 12H), 7.64 (d, *J* = 9.2 Hz, 2H), 7.47 (d, *J* = 9.5 Hz, 2H), 7.23 (d, *J* = 9.2 Hz, 2H), 6.60 (s, 2H), 6.30 (s, 1H), 5.68 (m, 4H), 5.40 (s, 1H), 4.19 (t, *J* = 6.3 Hz, 4H), 2.40 (s, 2H), 1.79 (m, 4H), 1.49 (m, 4H), 0.95 (t, *J* = 7.4 Hz, 6H). <sup>13</sup>C NMR (126 MHz, CD<sub>3</sub>CN, 298 K) δ [ppm] = 162.20, 161.81, 161.41, 161.01, 153.85, 148.57, 135.40, 134.66, 129.33, 129.31, 129.28, 129.26, 129.08, 129.06, 129.03, 129.01, 128.83, 128.80, 128.78, 128.76, 128.57, 128.55, 128.53, 128.51, 128.44, 127.71, 127.11, 126.34, 125.55, 123.39, 121.87, 121.85, 121.22, 119.79, 119.76, 114.11, 113.76, 91.39, 68.94, 43.45, 31.12, 25.63, 22.14, 18.99, 13.10. ESI-HRMS: m/z calcd for [M-2BArF<sup>-</sup>]<sup>2+</sup>  $C_{72}H_{74}N_4O_8^{2+}$ , 561.2748; found 561.2747(error = - 0.2 ppm).



<sup>13</sup>C NMR spectrum (126 MHz, CD<sub>3</sub>CN, 298 K) of Naphthotube 1a



ESI-HRMS mass spectrum of Naphthotube 1a

**1b:** m.p. =189 °C. <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN,298 K) δ [ppm] = 8.94 (s, 1H), 8.74 (d, J = 9.5 Hz, 2H), 7.81 (d, J = 9.3 Hz, 2H), 7.71 (d, J = 12.0 Hz, 12H), 7.43 (d, J = 9.5 Hz, 2H), 7.21 (d, J = 9.2 Hz, 2H), 6.61 (s, 2H), 6.31 (s, 1H), 5.79 – 5.57 (m, 4H), 5.48 (s, 1H), 4.18 (t, J = 6.6 Hz, 4H), 2.49 (s, 2H), 1.90 – 1.77 (m, 4H), 1.62 – 1.40 (m, 4H), 0.99 (t, J = 7.4 Hz, 6H). <sup>13</sup>C NMR (126 MHz, CD<sub>3</sub>CN, 298 K) δ [ppm] = 162.20, 161.80, 161.41, 161.01, 154.12, 148.73, 135.74, 134.66, 129.33, 129.31, 129.28, 129.26, 129.08, 129.05, 129.03, 129.01, 128.83, 128.80, 128.78, 128.76, 128.57, 128.55, 128.53, 128.51, 128.43, 127.71, 127.04, 126.28, 125.55, 123.38, 121.82, 121.57, 121.22, 120.05, 119.78, 114.01, 113.93, 91.43, 68.81, 43.37, 31.13, 25.56, 22.21, 18.97, 13.16. ESI-HRMS: m/z calcd for [M-2BArF<sup>-</sup>]<sup>2+</sup> C<sub>72</sub>H<sub>74</sub>N<sub>4</sub>O<sub>8</sub><sup>2+</sup>, 561.2746; found 561.2746(error = - 0.2 ppm).



<sup>13</sup>C NMR spectrum (126 MHz, CD<sub>3</sub>CN, 298 K) of Naphthotube **1b** 



ESI-HRMS mass spectrum of Naphthotube 1b

## 3. 2D NMR of Naphthotubes 1a, 1b, and 2b



Fig. S1<sup>1</sup>H,<sup>1</sup>H-COSY NMR spectrum (500 MHz, CD<sub>3</sub>CN, 298 K) of 1a



*Fig. S2* <sup>1</sup>H,<sup>1</sup>H-ROESY NMR spectrum (500 MHz, CD<sub>3</sub>CN, 298 K) of **1a.** The absolute configurations (syn or anti) cannot be assigned based on ROESY NMR spectrum, but were assigned from X-ray single crystal structure.



Fig. S3 <sup>1</sup>H, <sup>1</sup>H-COSY NMR spectrum (500 MHz, CD<sub>3</sub>CN, 298 K) of Naphthotube 1b



*Fig. S4* <sup>1</sup>H,<sup>1</sup>H-ROESY NMR spectrum (500 MHz, CD<sub>3</sub>CN, 298 K) of **1b**. The absolute configurations (syn or anti) cannot be assigned based on ROESY NMR spectrum, but were assigned from X-ray single crystal structure.



Fig. S5 <sup>1</sup>H, <sup>1</sup>H-COSY NMR spectrum (500 MHz, CD<sub>3</sub>CN, 298 K) of Naphthotube 2b



*Fig. S6* <sup>1</sup>H, <sup>1</sup>H-ROESY NMR spectrum (500 MHz, CD<sub>3</sub>CN, 298 K) of **2b.** The absolute configurations (syn or anti) cannot be assigned based on ROESY NMR spectrum, but were assigned from X-ray single crystal structure.

## 4. NMR Spectra of Host-Guest Complexes



*Fig. S7* <sup>1</sup>H NMR spectra (400 MHz,  $CD_3CN$ , 298 K) of (a) guest **5**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **5**.



*Fig. S8* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **5**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **5**.



*Fig. S9* <sup>1</sup>H NMR spectra (400 MHz,  $CD_3CN$ , 298 K) of (a) guest **5**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **5**.



*Fig. S10* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **5**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **5**.



*Fig. S11* <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of (a) guest **6**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **6**.



*Fig. S12* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **6**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **6**.



*Fig. S13* <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of (a) guest **6**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **6**.



*Fig. S14* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **6**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **6**.



*Fig. S15* <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of (a) guest **7**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **7**.



*Fig. S16* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **7**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **7**.



*Fig. S17* <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of (a) guest **7**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **7**.



*Fig. S18* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **7**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **7**.



*Fig. S19* <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of (a) guest **8**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **8**.



*Fig. S20* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **8**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **8**.



*Fig. S21* <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of (a) guest **8**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **8**.


*Fig. S22* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **8**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **8**.



*Fig. S23* <sup>1</sup>H NMR spectra (400 MHz,  $CD_3CN:CD_2Cl_2= 1:1, 298$  K) of (a) guest **8**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **8**.



*Fig. S24* <sup>1</sup>H NMR spectra (400 MHz,  $CD_3CN:CD_2Cl_2=1:1$ , 298 K) of (a) guest **8**, (c) **2b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **2b** and guest **8**.



*Fig. S25* <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of (a) guest **9**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **9**.



*Fig. S26* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **9**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **9**.



*Fig. S27* <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of (a) guest **9**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **9**.



*Fig. S28* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **9**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **9**.



*Fig. S29* <sup>1</sup>H NMR spectra (400 MHz, DMSO- $d_6$ , 298 K) of (a) guest **9**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **9**.



*Fig. S30* <sup>1</sup>H NMR spectra (400 MHz,  $CD_3CN:CD_2Cl_2= 1:1$ , 298 K) of (a) guest **9**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **9**.



*Fig. S31* <sup>1</sup>H NMR spectra (400 MHz,  $CD_3CN:CD_2Cl_2= 1:1, 298$  K) of (a) guest **9**, (c) **2b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **2b** and guest **9**.



*Fig. S32* <sup>1</sup>H NMR spectra (400 MHz,  $CD_3CN$ , 298 K) of (a) guest **10**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **10**.



*Fig. S33* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **10**, (c) **1a** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1a** and guest **10**.



*Fig. S34* <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of (a) guest **10**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **10**.



*Fig. S35* <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of (a) guest **10**, (c) **1b** and (b) its equimolar mixture. The proton of the guest experiences the upfield shift, the proton 9 of the host experiences the upfield shift, suggesting that the complexation between **1b** and guest **10**.

## 5. Binding Constants Determined by NMR Titration.



*Fig. S36* Job's plot obtained by plotting the chemical shift change ( $\Delta\delta$ ) of the Host's proton (10) in <sup>1</sup>H NMR spectra by varying the ratio of the host and the guest against the mole fraction of **1b**. The total concentration of the host and the guest is fixed: [Host] + [Guest] = 2.0 mM. This experiment supports the 1:1 binding stoichiometry between **9** and **1b** in CD<sub>3</sub>CN.



*Fig. S37* Molar ratio plot of **1b** titrated by **9**. These experiments support the 1:1 binding stoichiometry between **9** and **1b** in  $CD_3CN$ .



*Fig.* S38 Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of **1a** (0.5 mM) titrated by the **5** (0~13.33 mM).



*Fig. S39* Non-linear curve-fitting for the complexation between 1a and 5 in CD<sub>3</sub>CN at 298 K.



*Fig. S40* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>, 298 K) of **1a** (0.5 mM) titrated by **5** (0~22.22 mM).



*Fig. S41* Non-linear curve-fitting for the complexation between 1a and 5 in CD<sub>2</sub>Cl<sub>2</sub> at 298 K.



*Fig. S42* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of **1b** (0.5 mM) titrated by **5** (0~13.33 mM).



*Fig. S43* Non-linear curve-fitting for the complexation between 1b and 5 in CD<sub>3</sub>CN at 298 K.



*Fig. S44* Partial <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of **1b** (0.5 mM) titrated by **5** (0~22.22 mM).



*Fig. S45* Non-linear curve-fitting for the complexation between **1b** and **5** in  $CD_2Cl_2$  at 298 K.



*Fig. S46* Partial <sup>1</sup>H NMR spectra (400 MHz, CDCN<sub>3</sub>, 298 K) of **1a** (0.5 mM) titrated by **6** (0~6.22 mM).



*Fig. S47* Non-linear curve-fitting for the complexation between 1a and 6 in CDCN<sub>3</sub> at 298 K.



*Fig. S48* Partial <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of **1a** (0.5 mM) titrated by **6** (0~22.2 mM).



*Fig. S49* Non-linear curve-fitting for the complexation between 1a and 6 in CD<sub>2</sub>Cl<sub>2</sub> at 298 K.



*Fig. S50* Partial <sup>1</sup>H NMR spectra (400 MHz, CDCN<sub>3</sub>, 298 K) of **1b** (0.5 mM) titrated by **6** (0~6.22 mM).



*Fig. S51* Non-linear curve-fitting for the complexation between 1b and 6 in CDCN<sub>3</sub> at 298 K.



*Fig. S52* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>, 298 K) of **1b** (0.5 mM) titrated by **6** (0~22.2 mM).



*Fig. S53* Non-linear curve-fitting for the complexation between **1b** and **6** in  $CD_2Cl_2$  at 298 K.



*Fig. S54* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of **1a** (0.5 mM) titrated by **7** (0~13.3 mM).



*Fig. S55* Non-linear curve-fitting for the complexation between 1a and 7 in CD<sub>3</sub>CN at 298 K.



*Fig. S56* Partial <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of **1a** (0.5 mM) titrated by **7** (0~22.2 mM).



*Fig. S57* Non-linear curve-fitting for the complexation between 1a and 7 in CD<sub>2</sub>Cl<sub>2</sub> at 298 K.



*Fig. S58* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of **1b** (0.5 mM) titrated by **7** (0~13.33 mM).



*Fig. S59* Non-linear curve-fitting for the complexation between **1b** and **7** in  $CD_3CN$  at 298 K.



*Fig. S60* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>, 298 K) of **1b** (0.5 mM) titrated by **7** (0~22.2 mM).



*Fig. S61* Non-linear curve-fitting for the complexation between **1b** and **7** in  $CD_2Cl_2$  at 298 K.



*Fig. S62* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of **1a** (0.5 mM) titrated by **8** (0~13.33 mM).



*Fig. S63* Non-linear curve-fitting for the complexation between 1a and 8 in CD<sub>3</sub>CN at 298 K.



*Fig. S64* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>, 298 K) of **1a** (0.5 mM) titrated by **8** (0~22.2 mM).



*Fig. S65* Non-linear curve-fitting for the complexation between 1a and 8 in CD<sub>2</sub>Cl<sub>2</sub> at 298 K.



*Fig. S66* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of **1b** (0.5 mM) titrated by **8** (0~13.33 mM).



*Fig. S67* Non-linear curve-fitting for the complexation between 1b and 8 in CD<sub>3</sub>CN at 298 K.



*Fig. S68* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>, 298 K) of **1b** (0.5 mM) titrated by **8** (0~22.2 mM).



*Fig. S69* Non-linear curve-fitting for the complexation between **1b** and **8** in  $CD_2Cl_2$  at 298 K.



*Fig. S70* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of **1a** (0.5 mM) titrated by **9** (0~13.33 mM).



*Fig. S71* Non-linear curve-fitting for the complexation between 1a and 9 in CD<sub>3</sub>CN at 298 K.



*Fig. S72* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>, 298 K) of **1a** (0.5 mM) titrated by **9** (0~22.2 mM).



*Fig. S73* Non-linear curve-fitting for the complexation between 1a and 9 in CD<sub>2</sub>Cl<sub>2</sub> at 298 K.



*Fig. S74* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of **1b** (0.5 mM) titrated by **9** (0~13.33 mM).



*Fig.* S75 Non-linear curve-fitting for the complexation between 1b and 9 in  $CD_3CN$  at 298 K.



*Fig. S76* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>, 298 K) of **1b** (0.5 mM) titrated by **9** (0~22.2 mM).



*Fig.* S77 Non-linear curve-fitting for the complexation between **1b** and **9** in  $CD_2Cl_2$  at 298 K.



*Fig. S78* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of **1a** (0.4 mM) titrated by **10** (0~6.22 mM).



*Fig. S79* Non-linear curve-fitting for the complexation between 1a and 10 in CD<sub>3</sub>CN at 298 K.


*Fig. S80* Partial <sup>1</sup>H NMR spectra (400 MHz,  $CD_2Cl_2$ , 298 K) of **1a** (0.5 mM) titrated by **10** (0~22.2 mM).



*Fig. S81* Non-linear curve-fitting for the complexation between 1a and 10 in  $CD_2Cl_2$  at 298 K.



*Fig. S82* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 298 K) of **1b** (0.4 mM) titrated by **10** (0~6.22 mM).



*Fig. S83* Non-linear curve-fitting for the complexation between **1b** and **10** in CD<sub>3</sub>CN at 298 K.



*Fig. S84* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>, 298 K) of **1b** (0.5 mM) titrated by **10** (0~22.2 mM).



*Fig. S85* Non-linear curve-fitting for the complexation between 1b and 10 in  $CD_2Cl_2$  at 298 K.

## 6. <sup>1</sup>H NMR Spectra of 1b titrated by Bu<sub>4</sub>NBArF



*Fig. S86* <sup>1</sup>H NMR spectra (400 MHz, 298 K) of **1b** (0.5 mM) titrated by Bu<sub>4</sub>NBArF; (a) in  $CD_2Cl_2$  and (b) in  $CD_3CN$ ; (c) chemical shift changes of proton 9 of **1b** with increasing concentrations of Bu<sub>4</sub>NBArF (black square,  $CD_2Cl_2$ ; red circle,  $CD_3CN$ ).

## 7. Control Experiments.



*Fig. S87* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN, 0.5 mM, 298 K) of a) **1b**, b) the equimolar mixture of **1b** and CH<sub>2</sub>Cl<sub>2</sub>, c) CH<sub>2</sub>Cl<sub>2</sub>, d) the equimolar mixture of **1a** and CH<sub>2</sub>Cl<sub>2</sub>, and e) **1a**. In the presence of **1a/1b**, the signals of CH<sub>2</sub>Cl<sub>2</sub> undergo slight upfield shift, suggesting that CH<sub>2</sub>Cl<sub>2</sub> is a guest and should be encapsulated in the cavity of **1a/1b**. Consequently, a solvent molecule may sit in the cavity in CD<sub>2</sub>Cl<sub>2</sub>, and the incoming aromatic hydrocarbon guest has to compete with the solvent molecule. This is also in line with the single crystal structure obtained from CH<sub>2</sub>Cl<sub>2</sub>, in which a CH<sub>2</sub>Cl<sub>2</sub> molecule is encapsulated in the cavity. This may be another reason why the binding affinity is smaller in CD<sub>2</sub>Cl<sub>2</sub>.



*Fig. S88* Partial <sup>1</sup>H NMR spectra (400 MHz,  $CD_3CN:CD_2Cl_2 = 1:1$ , 298 K) of **1b** (0.4 mM) titrated by **8** (0~13.33 mM).



*Fig. S89* Non-linear curve-fitting for the complexation between **1b** and **8** in  $CD_3CN:CD_2Cl_2 = 1:1$  at 298 K.



*Fig. S90* Partial <sup>1</sup>H NMR spectra (400 MHz,  $CD_3CN:CD_2Cl_2 = 1:1 298 \text{ K}$ ) of **1b** (0.5 mM) titrated by **9** (0~13.33 mM).



*Fig. S91* Non-linear curve-fitting for the complexation between **1b** and **9** in  $CD_3CN:CD_2Cl_2 = 1:1$  at 298 K.



*Fig. S92* Partial <sup>1</sup>H NMR spectra (400 MHz, DMSO-*d*<sub>6</sub>, 298 K) of **1b** (0.5 mM) titrated by **9** (0~22.2 mM).



*Fig. S93* Non-linear curve-fitting for the complexation between **1b** and **9** in DMSO- $d_6$  at 298 K.



*Fig. S94* Partial <sup>1</sup>H NMR spectra (400 MHz,  $CD_3CN:CD_2Cl_2 = 1:1$ , 298 K) of **2b** (0.4 mM) titrated by **8** (0~13.33 mM).



*Fig. S95* Non-linear curve-fitting for the complexation between 2b and 8 in CD<sub>3</sub>CN:CD<sub>2</sub>Cl<sub>2</sub> =1:1 at 298 K.



*Fig. S96* Partial <sup>1</sup>H NMR spectra (400 MHz, CD<sub>3</sub>CN:CD<sub>2</sub>Cl<sub>2</sub>=1:1 298 K) of **2b** (0.4 mM) titrated by **9** (0~13.33 mM).



*Fig. S97* Non-linear curve-fitting for the complexation between 2b and 9 in CD<sub>3</sub>CN:CD<sub>2</sub>Cl<sub>2</sub> =1:1 at 298 K.

## 8. X-Ray Single Crystallography

Crystals were obtained by slow evaporation of dichloromethane or diethyl ether solutions of **1b**. Crystal data was collected on a Bruker D8 VENTURE with Cu K $\alpha$  radiation ( $\lambda$  = 1.54178 Å) at 173(2) K. The structures were solved by the direct method and different Fourier syntheses. All calculations were performed by full-matrix least-squares methods on  $F^2$  by using the SHELX-97 program<sup>5,6</sup>, all non-hydrogen atoms were refined with anisotropic thermal parameters and the hydrogen atoms were fixed at calculated positions and refined by a riding mode. SQUEEZE routine implemented on PLATON was used to remove electron densities corresponding to disordered solvent molecules in Crystal data.

Crystal Data: **1b** obtained from dichloromethane-d<sub>2</sub>: CCDC: 1945708.  $C_{137}H_{98}B_2Cl_2D_2F_{48}N_4O_8$  (*M*=2936.74 g/mol): triclinic, space group P-1 (no. 2), *a*= 13.9528(5) Å, *b*= 15.9387(5) Å, *c*= 16.2647(6) Å, *a*= 75.310(2)°, *β*= 84.666(2)°, *γ*= 67.739(2)°, *V*= 3238.1(2) Å<sup>3</sup>,*Z*= 1, *T*= 100.04 K, µ(CuK*α*)= 1.605 mm<sup>-1</sup>, *Dcalc*= 1.506 g/cm<sup>3</sup>, 45782 reflections measured (5.618° ≤ 2 $\Theta$  ≤ 120.414°), 9579 unique (*R*<sub>int</sub>= 0.0457, *R*<sub>sigma</sub>= 0.0355) which were used in all calculations. The final *R*<sub>1</sub> was 0.0572 (I > 2 $\sigma$ (I)) and *wR*<sub>2</sub> was 0.1814 (all data).



Crystal Data: **1b** obtained from diethyl ether: CCDC: 1945707.  $C_{136}H_{98}B_2F_{48}N_4O_8$ (*M*=2849.80 g/mol): triclinic, space group P-1 (no. 2), *a*= 13.9780(5) Å, *b*= 15.6198(6) Å, *c*= 16.4650(6) Å, *a*= 75.3050(10)°, *β*= 85.4450(10)°, *γ*= 68.590(2)°, *V*= 3237.0(2) Å<sup>3</sup>, *Z*= 1, *T*= 100.04 K,  $\mu$ (CuK $\alpha$ )= 1.217 mm<sup>-1</sup>, *Dcalc*= 1.462 g/cm<sup>3</sup>, 42286 reflections measured (5.55° ≤ 2 $\Theta$  ≤ 120.408°), 9575 unique (*R*<sub>int</sub>= 0.0350, *R*<sub>sigma</sub>= 0.0304) which were used in all calculations. The final *R*<sub>1</sub> was 0.0381 (I > 2 $\sigma$ (I)) and *wR*<sub>2</sub> was 0.1000 (all data).



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