

Electronic Supplementary Information for

**SO₂ Conversion to Sulfones:
Development and Mechanistic Insights
of a Sulfonylative Hiyama Cross-Coupling**

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1. Experimental details

1.1. General considerations

All reactions and manipulations were performed at 20 °C in a recirculating MBraun LabMaster DP inert atmosphere (Ar) drybox and/or using standard vacuum Schlenk line procedures, if not otherwise stated. Glassware was dried overnight at 60 °C before use. All NMR spectra were obtained using a Bruker AVANCE Neo 400 MHz spectrometer. Chemical shifts for ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra were referenced to solvent impurities. Reagents were purchased from commercial suppliers, stored in the glovebox and used as received. SO_2 was released by thermal degradation with a heat gun ($T \sim 500\text{-}600^\circ\text{C}$) of $\text{K}_2\text{S}_2\text{O}_5$ (purchased from Aldrich). Solvents were thoroughly dried by standard methods and distilled immediately before use. **1c**,¹ **1e-1f**,² and **8c**,³ were synthesized according to adapted literature procedures.

1.2. Experimental set-up

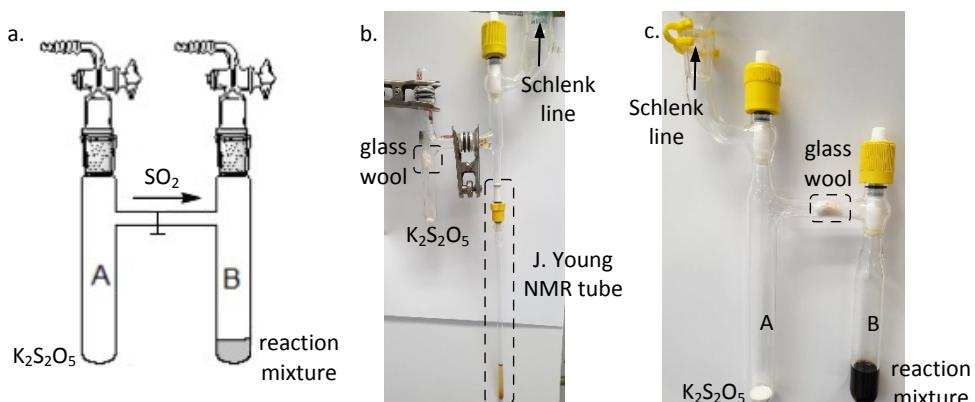


Figure S1. a. Scheme of the experimental set-up for SO_2 generation and condensation; b. Picture of the set-up for NMR-scale reactions; c. Picture of the set-up for scale-up reactions (1 mmol). Use of glass wool avoid the bumping of $\text{K}_2\text{S}_2\text{O}_5$ in the reaction mixture.

1.3. Screening of the reaction conditions

All the listed experiments were performed under argon in 2.5 mL J. Young NMR tubes at the 0.1 mmol scale, according to adapted general procedure (described below, **1.5.a**).

Fluoride source (eq.)	SO ₂ source (eq.)	Catalyst (mol%)	Ligand (mol%)	Temperature (°C)	Yield in 3a (%)
TBAF·3H ₂ O (1)	SO ₂ (1)	Pd(acac) ₂ (5)	-	80 °C	12
TBAF·3H ₂ O (1)	DABSO (0.5)	Pd(acac) ₂ (5)	-	80 °C	0
TBAT (1)	SO ₂ (1)	Pd(acac) ₂ (5)	-	80 °C	17
CsF (1)	SO ₂ (1)	Pd(acac) ₂ (5)	-	80 °C	0
CsF (3)	SO ₂ (1)	Pd(acac) ₂ (5)	-	80 °C	traces
TBAT (2)	SO ₂ (1)	Pd(acac) ₂ (5)	-	80 °C	8
TBAT (0.9)	SO ₂ (1)	Pd(acac) ₂ (5)	-	80 °C	13
TBAT (1)	SO ₂ (2)	Pd(acac) ₂ (5)	-	80 °C	26
TBAT (1)	SO ₂ (4)	Pd(acac) ₂ (5)	-	80 °C	0
TBAT (1)	SO ₂ (2)	Pd(dba) ₂ (5)	-	80 °C	5
TBAT (1)	SO ₂ (2)	Pd(OAc) ₂ (5)	-	80 °C	12
TBAT (1)	SO ₂ (2)	Pd(Cl) ₂ (5)	-	80 °C	15
TBAT (1)	SO ₂ (2)	Pd(OCOCF ₃) ₂ (5)	-	80 °C	traces
TBAT (1)	SO ₂ (2)	Pd(acac) ₂ (5)	XPhos (10)	80 °C	41
TBAT (1)	SO ₂ (2)	Pd(acac) ₂ (5)	Xantphos (10)	80 °C	78
TBAT (1)	SO ₂ (2)	Pd(acac) ₂ (5)	P(Ph) ₃ (10)	80 °C	33
TBAT (1)	SO ₂ (2)	Pd(acac) ₂ (5)	P(Cy) ₃ (10)	80 °C	30
TBAT (1)	SO ₂ (2)	Pd(acac) ₂ (5)	CataCXium A (10)	80 °C	33
TBAT (1)	SO ₂ (2)	Pd(acac) ₂ (5)	dppb (10)	80 °C	42
TBAT (1)	SO ₂ (2)	Pd(acac) ₂ (5)	Xantphos (5)	80 °C	65
TBAT (1)	SO ₂ (2)	Pd(acac) ₂ (5)	Xantphos (10)	100°C	60
TBAT (1)	SO ₂ (2)	Pd(acac) ₂ (5)	Xantphos (10)	t.a.	0
TBAF (1M, THF) (1)	SO ₂ (2)	Pd(acac) ₂ (5)	Xantphos (10)	80 °C	71
TBAT (1)	SO ₂ (2)	-	-		0

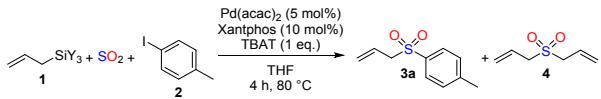
Table S1: Influence of the reaction conditions on the sulfonylative Hiyama cross-coupling

1.4. Fluoro-affinity of allylsilanes

The fluoro-affinity of different organosilanes were quantified by the Gibbs free energy variation for the fluoride transfer from Me₃SiF₂[−] to the organosilane (see equation below, computational details are described in **2.**).



Even though the poor fluorophile trimethyl(allyl)silane does not lead to the obtention of the diallyl sulfone **4**, the cross-coupling is also found to be diminished and thus despite a higher temperature or reaction time.



Y_3	Temperature (°C)	Time (h)	Yield 3a (%)	Yield 4 (%)	4/3a ratio	ΔG (kcal.mol ⁻¹)
(OMe) ₃ (1b)	80	4	51	23	0.45	2.98
(OEt) ₃ (1a)	80	4	78	10	0.13	3.60
Me ₂ (OMe) (1c)	80	4	74	7	0.09	11.94
Me ₃ (1d)	80	4	54	0	0	15.60
Me ₃ (1d)	80	24	68	0	0	15.60
Me ₃ (1d)	100	4	71	0	0	15.60

Table S2: Influence of the allylsilane on the formation of diallyl sulfone

1.5. Synthesis and characterization of allylsulfones

a) General procedure for NMR-scale experiments

In a glovebox, an oven dried 2.5 mL NMR tube equipped with a J. Young valve was charged with Pd(acac)₂ (1.52 mg, 0.005 mmol, 5 mol%), Xantphos (5.79 mg, 0.01 mmol, 10 mol%), TBAT (53.99 mg, 0.1 mmol, 1 eq.) and the electrophile (0.1 mmol, 1 eq.) followed by 400 μ L of *d*₈-THF. To the resulting homogeneous yellow solution were sequentially added the silane (0.11 mmol, 1,1 eq.) and mesitylene (5 μ L) as an internal standard. The tube was then sealed, brought out of the glovebox and connected to a fitted small vacuum line involving a flask containing K₂S₂O₅ (88.93 mg, 0.4 mmol, 4 eq.). The tube was degassed by freezing the solution in liquid nitrogen and evacuation under vacuum. After three evacuation/argon cycles, under static vacuum, K₂S₂O₅ was heated at 600 °C with a heat gun. Degradation released pale yellow SO₂ vapors which were condensed in the NMR tube cooled under liquid nitrogen (~ 2 minutes). The valve of the NMR tube was then closed, the mixture warmed to room temperature and immersed in a pre-heated oil bath at 80 °C (oil temperature). Yields of sulfones were determined by ¹H NMR integration versus mesitylene as an internal standard ($\delta_H = 6.74$ and 2.22 ppm in *d*₈-THF).

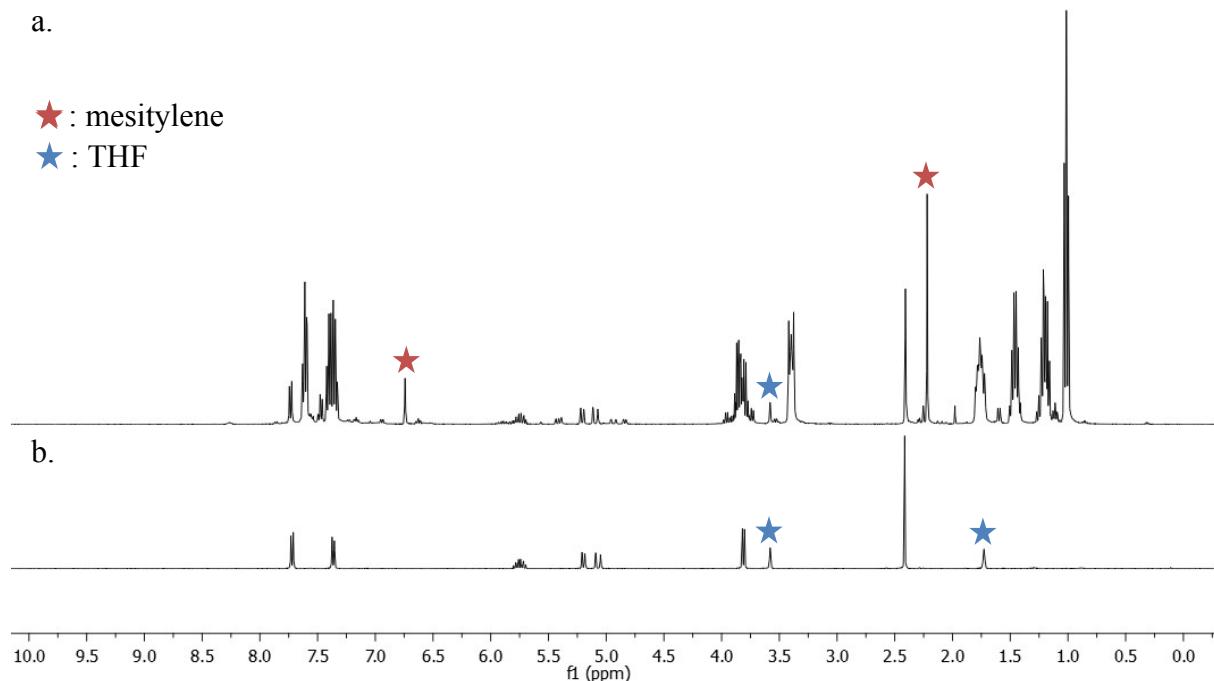
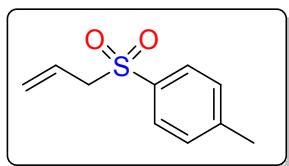


Figure S3: Representative ¹H NMR spectra obtained in d₈-THF for the formation of 3a. a. Crude reaction mixture after heating 4 h at 80 °C. Yield = 78%; b. Isolated spectrum of 3a.

b) General procedure for preparative scale experiments

In a glovebox, the catalytic chamber (chamber B, see Figure S1) of an oven dried 2*20 mL two-chambers set-up was charged with Pd(acac)₂ (15.23 mg, 0.05 mmol, 5 mol%), Xantphos (57.86 mg, 0.1 mmol, 10 mol%) and the electrophile (1 mmol, 1 eq.) followed by 3 mL of THF. To the resulting homogeneous yellow solution were added the silane (0.11 mmol, 1,1 eq.) and TBAF (1 mL, 1M solution in THF, 1 mmol, 1 eq.). The chamber B was then sealed and the two-chambers set-up brought out of the glovebox. The chamber A (see Figure S1) is then filled with K₂S₂O₅ (889.28 mg, 4 mmol, 4 eq.) and connected to a vacuum line. The whole system was degassed by freezing the solution in liquid nitrogen and evacuation under vacuum. After three evacuation/argon cycles, under static vacuum, K₂S₂O₅ was heated at 600 °C with a heat gun. Degradation released pale yellow SO₂ vapors which were condensed in the chamber B cooled under liquid nitrogen (~2 minutes). The chamber B was then sealed, the mixture warmed to room temperature and immersed in a pre-heated oil bath at 80 °C (oil temperature). After 4 h, the reaction mixture was then quenched with 5 mL H₂O. The aqueous phase was extracted with EtOAc (3 x 5 mL) and the combined organic phases were dried with MgSO₄. The volatiles were then removed under vacuum and the resulting crude mixture was purified by column chromatography on silica gel (eluent: *n*-pentane/ethyl acetate).

○ Allyl 4-methylphenyl sulfone (**3a**)

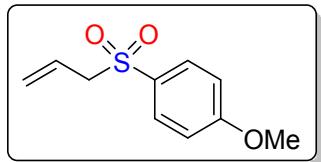


Obtained as a pale oil in 69% yield (135 mg) using 9:1 *n*-pentane/ethyl acetate for the column chromatography.

The spectroscopic data are in agreement with those reported in the literature.⁴

¹H NMR (400 MHz, *d*₈-THF) δ (ppm) = 7.72 (d, *J* = 8.2 Hz, 2H, Ar), 7.36 (d, *J* = 7.9 Hz, 2H, Ar), 5.75 (ddt, *J* = 17.5, 10.3, 7.3 Hz, 1H, CH₂=CH-CH₂), 5.24 – 5.17 (m, 1H, CH₂=CH-CH₂), 5.07 (dd, *J* = 17.1, 1.4 Hz, 1H, CH₂=CH-CH₂), 3.81 (d, *J* = 7.3 Hz, 2H, CH₂=CH-CH₂), 2.41 (s, 3H, C₆H₄-CH₃). **¹³C{¹H}-NMR** (400 MHz, *d*₈-THF) δ (ppm) = 144.8 (C(Ar)-CH₃), 137.5 (C(Ar)-SO₂), 130.0 (C(Ar) *meta*), 129.2 (C(Ar) *ortho*), 127.0 (CH₂=CH-CH₂), 123.3 (CH₂=CH-CH₂), 60.9 (CH₂=CH-CH₂), 21.3 (CH₃).

○ Allyl 4-methoxyphenyl sulfone (**3b**)

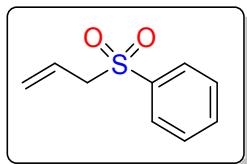


Obtained as a pale oil in 75% yield (159 mg) using 9:1 *n*-pentane/ethyl acetate for the column chromatography.

The spectroscopic data are in agreement with those reported in the literature.⁴

¹H NMR (400 MHz, *d*₈-THF) δ (ppm) = 7.76 (d, *J* = 8.9 Hz, 2H, Ar), 7.05 (d, *J* = 8.9 Hz, 2H, Ar), 5.76 (ddt, *J* = 17.5, 10.2, 7.4 Hz, 1H, CH₂=CH-CH₂), 5.26 – 5.15 (m, 1H, CH₂=CH-CH₂), 5.07 (dd, *J* = 17.1, 1.3 Hz, 1H, CH₂=CH-CH₂), 3.86 (s, 3H, C₆H₄-OCH₃), 3.79 (d, *J* = 7.3 Hz, 2H, CH₂=CH-CH₂). **¹³C{¹H}-NMR** (400 MHz, *d*₈-THF) δ (ppm) = 164.5 (C(Ar)-OCH₃), 131.9 (C(Ar)-SO₂), 131.3 (C(Ar) *ortho*), 127.2 (CH₂=CH-CH₂), 123.2 (CH₂=CH-CH₂), 114.6 (C(Ar) *meta*), 61.2 (CH₂=CH-CH₂), 55.8 (OCH₃).

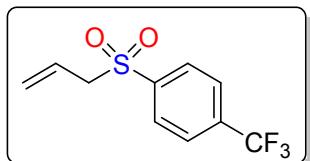
- Allyl phenyl sulfone (**3c**)



TBAT was used as the fluoride source instead of TBAF (1M). Obtained as a pale oil in 82% yield (176 mg) using 9:1 *n*-pentane/ethyl acetate for the column chromatography. The spectroscopic data are in agreement with those reported in the literature.⁴

¹H NMR (400 MHz, *d*₈-THF) δ (ppm) = 7.88 – 7.82 (m, 2H, Ar), 7.68 – 7.61 (m, 1H, Ar), 7.59 – 7.51 (m, 2H, Ar), 5.76 (ddt, *J* = 17.4, 10.2, 7.3 Hz, 1H, CH₂=CH-CH₂), 5.25 – 5.17 (m, 1H, CH₂=CH-CH₂), 5.08 (ddd, *J* = 17.1, 2.7, 1.3 Hz, 1H, CH₂=CH-CH₂), 3.88 – 3.82 (m, 2H, CH₂=CH-CH₂). **¹³C{¹H}-NMR** (400 MHz, *d*₈-THF) δ (ppm) = 140.3 (C(Ar)-SO₂), 133.9 (C(Ar) *para*), 129.5 (C(Ar) *meta*), 129.2 (C(Ar) *ortho*), 126.8 (CH₂=CH-CH₂), 123.5 (CH₂=CH-CH₂), 60.8 (CH₂=CH-CH₂).

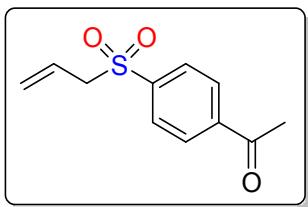
- Allyl 4-trifluoromethylphenyl sulfone (**3d**)



Obtained as a white solid in 33% yield (83 mg) using 9:1 *n*-pentane/ethyl acetate for the column chromatography.

¹H NMR (400 MHz, *d*₈-THF) δ (ppm) = 8.06 (d, *J* = 8.2 Hz, 2H, Ar), 7.92 (d, *J* = 8.3 Hz, 2H, Ar), 5.78 (ddt, *J* = 17.5, 10.2, 7.4 Hz, 1H, CH₂=CH-CH₂), 5.34 – 5.18 (m, 1H, CH₂=CH-CH₂), 5.19 – 5.05 (m, 1H, CH₂=CH-CH₂), 3.96 (d, *J* = 7.4 Hz, 2H, CH₂=CH-CH₂). **¹³C{¹H}-NMR** (400 MHz, *d*₈-THF) δ (ppm) = 143.9 (C(Ar)-SO₂), 135.2 (q, *J*_{C-F} = 32.7 Hz, C(Ar)-CF₃), 130.1 (C(Ar) *ortho*), 126.9 (q, *J*_{C-F} = 3.8 Hz C(Ar) *meta*), 126.3 (CH₂=CH-CH₂), 124.43 (q, *J*_{C-F} = 272.8 Hz, CF₃), 124.2 (CH₂=CH-CH₂), 60.5 (CH₂=CH-CH₂).

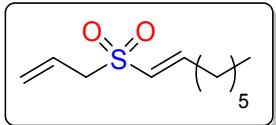
- Allyl 4-acethylphenyl sulfone (**3e**)



Obtained as a tan oil in 39% yield (87 mg) using 1:1 *n*-pentane/ethyl acetate for the column chromatography.

¹H NMR (400 MHz, *d*₈-THF) δ (ppm) = 8.18 – 8.10 (m, 2H, Ar), 8.00 – 7.92 (m, 2H, Ar), 5.77 (ddt, *J* = 17.5, 10.2, 7.4 Hz, 1H, CH₂=CH-CH₂), 5.22 (ddt, *J* = 10.2, 1.5, 0.8 Hz, 1H, CH₂=CH-CH₂), 5.09 (dq, *J* = 17.1, 1.3 Hz, 1H, CH₂=CH-CH₂), 3.95 – 3.89 (m, 2H, CH₂=CH-CH₂), 2.61 (s, 3H, C(O)CH₃). **¹³C{¹H}-NMR** (400 MHz, *d*₈-THF) δ (ppm) = 196.4 (C(O)CH₃), 143.5 (C(Ar)-SO₂), 141.7 (C(Ar)-C(O)CH₃), 129.5 (C(Ar) *meta*), 129.2 (C(Ar) *ortho*), 126.5 (CH₂=CH-CH₂), 123.9 (CH₂=CH-CH₂), 60.6 (CH₂=CH-CH₂), 26.5 (C(O)CH₃).

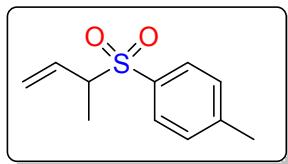
- Allyl *E*-oct-1-enyl sulfone (**3f**)



Obtained as a pale yellow oil in 52% yield (83 mg) using 9:1 *n*-pentane/ethyl acetate for the column chromatography.

¹H NMR (400 MHz, *d*₈-THF) δ (ppm) = 6.75 (dt, *J* = 14.9, 6.9 Hz, 1H, CH=CH-(CH₂)₅-CH₃), 6.35 (dt, *J* = 15.1, 1.4 Hz, 1H, CH=CH-(CH₂)₅-CH₃), 5.82 (ddt, *J* = 17.6, 10.4, 7.3 Hz, 1H, CH₂=CH-CH₂), 5.34 – 5.25 (m, 2H, CH₂=CH-CH₂), 3.67 (d, *J* = 7.3 Hz, 2H, CH₂=CH-CH₂), 2.25 (td, *J* = 8.1, 1.3 Hz, 2H, CH=CH-CH₂-(CH₂)₃-CH₂-CH₃), 1.55 – 1.41 (m, 2H, CH=CH-CH₂-(CH₂)₃-CH₂-CH₃), 1.33 (m, 6H, CH=CH-CH₂-(CH₂)₃-CH₂-CH₃), 0.90 (t, *J* = 6.8 Hz, 3H, CH=CH-(CH₂)₅-CH₃). **¹³C{¹H}-NMR** (400 MHz, *d*₈-THF) δ (ppm) = 149.2 (SO₂-CH=CH), 129.4 (SO₂-CH=CH), 127.3 (CH₂=CH-CH₂), 123.0 (CH₂=CH-CH₂), 59.5 (CH₂=CH-CH₂), 32.3 (CH=CH-CH₂-(CH₂)₄-CH₃), 32.0 (CH=CH-CH₂-CH₂-(CH₂)₃-CH₃), 29.5 (CH=CH-(CH₂)₂-CH₂-(CH₂)₂-CH₃), 28.5 (CH=CH-(CH₂)₃-CH₂-CH₂-CH₃), 23.2 (CH=CH-(CH₂)₄-CH₂-CH₃), 14.2 (CH=CH-(CH₂)₅-CH₃).

o 1-Methylallyl phenyl sulfone (**3g**)

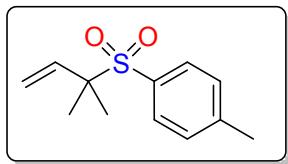


Obtained as a pale yellow oil in 72% yield (151 mg) using 9:1 *n*-pentane/ethyl acetate for the column chromatography.

The spectroscopic data are in agreement with those reported in the literature.⁵

¹H NMR (400 MHz, *d*₈-THF) δ (ppm) = 7.69 (d, *J* = 8.3 Hz, 2H, Ar), 7.36 (d, *J* = 7.9 Hz, 2H, Ar), 5.80 (ddd, *J* = 17.3, 10.4, 7.8 Hz, 1H, CH₂=CH-CH(CH₃)), 5.16 (dt, *J* = 10.4, 1.0 Hz, 1H, CH₂=CH-CH(CH₃)), 5.02 (dt, *J* = 17.2, 1.2 Hz, 1H, CH₂=CH-CH(CH₃)), 3.82 – 3.65 (m, 1H, CH₂=CH-CH(CH₃)), 2.41 (s, 3H, C₆H₄-CH₃), 1.34 (d, *J* = 6.9 Hz, 3H, CH₂=CH-CH(CH₃)).
¹³C{¹H}-NMR (400 MHz, *d*₈-THF) δ (ppm) = 144.9 (CH₂=CH-CH(CH₃)), 135.9 (C(Ar)-SO₂), 133.2 (C(Ar) *para*), 130.0 (C(Ar) *meta*), 129.9 (C(Ar) *ortho*), 120.7 (CH₂=CH-CH(CH₃)), 64.3 (CH₂=CH-CH(CH₃)), 21.3 (CH₃) 13.1 (CH₂=CH-CH(CH₃)).

o 1,1-Dimethylallyl phenyl sulfone (**3h**)

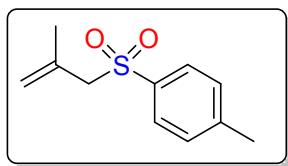


Obtained as a pale oil in 51% yield (114 mg) using 9:1 *n*-pentane/ethyl acetate for the column chromatography.

The spectroscopic data are in agreement with those reported in the literature.⁶

¹H NMR (400 MHz, *d*₈-THF) δ (ppm) = 7.66 (d, *J* = 8.2 Hz, 2H, Ar), 7.34 (d, *J* = 8.0 Hz, 2H, Ar), 6.02 (dd, *J* = 17.5, 10.7 Hz, 1H, CH₂=CH-C(CH₃)₂), 5.17 (d, *J* = 10.7 Hz, 1H, CH₂=CH-C(CH₃)₂), 5.01 (d, *J* = 17.5 Hz, 1H, CH₂=CH-C(CH₃)₂), 2.41 (s, 3H, C₆H₄-CH₃), 1.36 (s, 6H, CH₂=CH-C(CH₃)₂). **¹³C{¹H}-NMR** (400 MHz, *d*₈-THF) δ (ppm) = 144.9 (CH₂=CH-C(CH₃)₂), 138.5 (C(Ar)-CH₃), 134.1 (C(Ar)-SO₂), 131.4 (C(Ar) *meta*), 129.5 (C(Ar) *ortho*), 118.0 (CH₂=CH-C(CH₃)₂), 64.7 (CH₂=CH-C(CH₃)₂), 21.3 (CH₃), 20.8 (CH₂=CH-C(CH₃)₂).

- 2-Methylallyl phenyl sulfone (**3i**)



Obtained as a white solid in 28% yield (59 mg) using 9:1 *n*-pentane/ethyl acetate for the column chromatography.

The spectroscopic data are in agreement with those reported in the literature.⁷

¹H NMR (400 MHz, *d*₈-THF) δ (ppm) = 7.73 (d, *J* = 8.3 Hz, 2H, Ar), 7.36 (d, *J* = 8.0 Hz, 2H, Ar), 4.96 – 4.87 (m, 1H, CH₂=C(CH₃)-CH₂), 4.60 (d, *J* = 0.7 Hz, 1H, CH₂=C(CH₃)-CH₂), 3.79 (s, 2H, CH₂=C(CH₃)-CH₂), 2.41 (s, 3H, C₆H₄-CH₃), 1.83 (s, 3H, CH₂=C(CH₃)-CH₂). **¹³C{¹H}-NMR** (400 MHz, *d*₈-THF) δ (ppm) = 144.8 (CH₂=C(CH₃)-CH₂), 137.5 (C(Ar)-CH₃), 135.6 (C(Ar)-SO₂), 130.0 (C(Ar) *meta*), 129.2 (C(Ar) *ortho*), 119.7 (CH₂=C(CH₃)-CH₂), 64.3 (CH₂=C(CH₃)-CH₂), 22.6 (CH₂=C(CH₃)-CH₂), 21.3 (CH₃).

1.6. *Hammett plot and mechanistic control experiments*

1.6.1. *Hammett plot*

The Hammett plot was obtained by considering the NMR yield of the reaction at t = 30 min.

1.6.2. *Mechanistic control experiments*

As soon as sulfur dioxide is added to the reaction mixture, a transient species can be observed and attributed to allylsulfinate (see spectrum below).

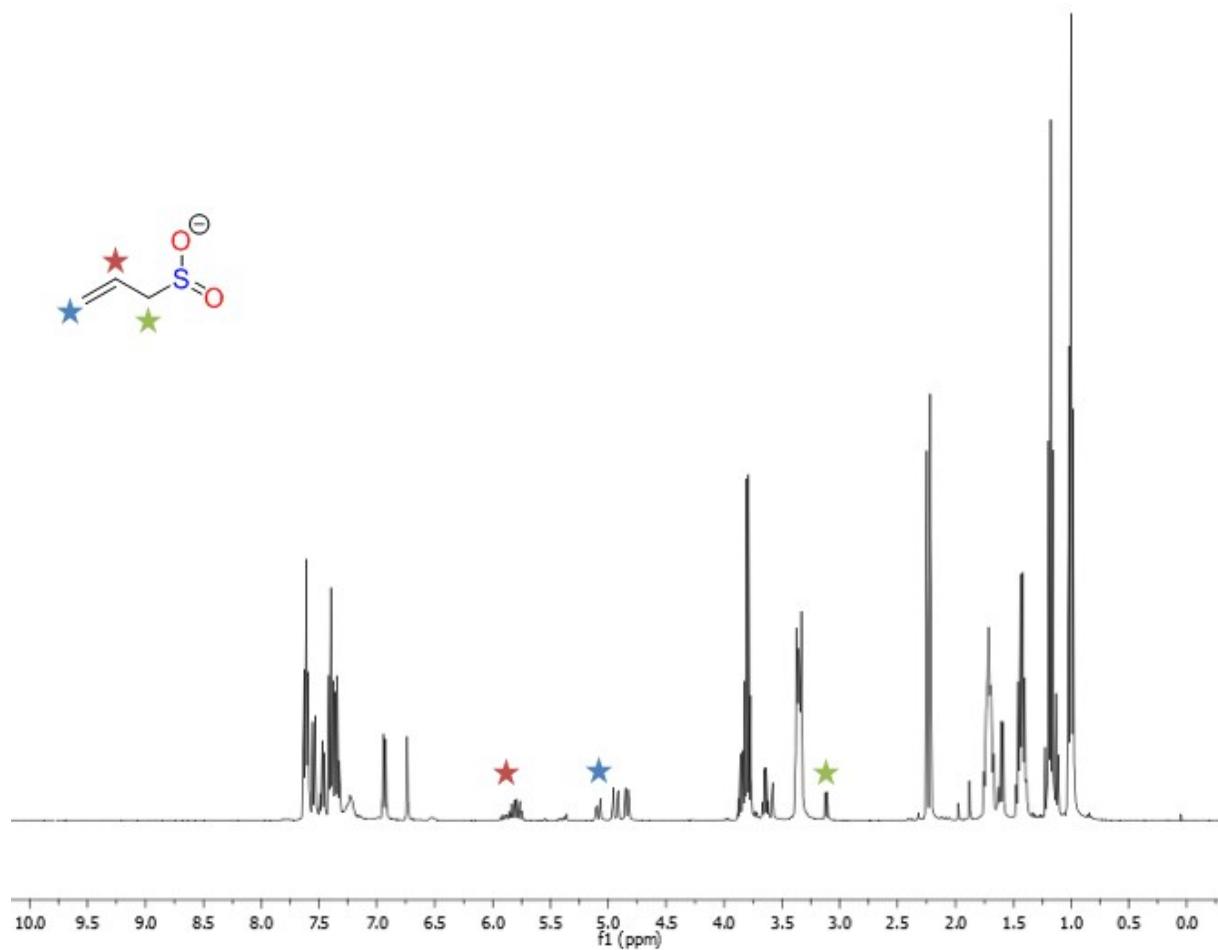


Figure S4: Representative ¹H NMR monitoring of a sulfonylative Hiyama cross-coupling in d₈-THF.

Reactions described on Scheme 4 were performed at the NMR scale according to below procedures. Yields were determined by ¹H NMR integration versus mesitylene as an internal standard.

a) Classical Hiyama cross-coupling (without SO₂)

In a glovebox, an oven dried 2.5 mL NMR tube equipped with a J. Young valve was charged with Pd(acac)₂ (1.52 mg, 0.005 mmol, 5 mol%), Xantphos (5.79 mg, 0.01 mmol, 10 mol%), TBAT (53.99 mg, 0.1 mmol, 1 eq.) and the electrophile (0.1 mmol, 1 eq.) followed by 400 μ L of *d*₈-THF. To the resulting homogeneous yellow solution were sequentially added the silane (0.11 mmol, 1,1 eq.) and mesitylene (5 μ L) as an internal standard. The tube was then sealed, brought out of the glovebox and immersed in a pre-heated oil bath at 80 °C (oil temperature).

b) Sulfonylative Hiyama cross-coupling

See general procedure for NMR-scale experiments (**1.4.a**) above.

c) Ethyl iodide trapping of sulfinate

In a glovebox, an oven dried 2.5 mL NMR tube equipped with a J. Young valve was charged with TBAT (53.99 mg, 0.1 mmol, 1 eq.) followed by 400 μ L of d_8 -THF. To the resulting homogeneous colourless solution were sequentially added the silane (0.1 mmol, 1 eq.), ethyl iodide (8 μ L, 0.1 mmol, 1 eq.) and mesitylene (5 μ L) as an internal standard. The tube was then sealed, brought out of the glovebox and connected to a fitted small vacuum line involving a flask containing $K_2S_2O_5$ (88.93 mg, 0.4 mmol, 4 eq.). The tube was degassed by freezing the solution in liquid nitrogen and evacuation under vacuum. After three evacuation/argon cycles, under static vacuum, $K_2S_2O_5$ was heated at 600 °C with a heat gun. Degradation released pale yellow SO_2 vapors which were condensed in the NMR tube cooled under liquid nitrogen (~ 2 minutes). The valve of the NMR tube was then closed, the mixture warmed to room temperature and immersed in a pre-heated oil bath at 80 °C (oil temperature).

d) Direct cross-coupling of aryl sulfinate

In a glovebox, an oven dried 2.5 mL NMR tube equipped with a J. Young valve was charged with $Pd(acac)_2$ (1.52 mg, 0.005 mmol, 5 mol%), Xantphos (5.79 mg, 0.01 mmol, 10 mol%), sodium *para*-toluenesulfinate (17.82 mg, 0.1 mmol, 1 eq.) followed by 400 μ L of d_7 -DMF. To the resulting solution were sequentially added iodobenzene (11.2 μ L, 0.1 mmol, 1 eq.) and mesitylene (5 μ L) as an internal standard. The tube was then sealed, brought out of the glovebox and immersed in a pre-heated oil bath at 80 °C (oil temperature).

The reaction was carried in DMF due to solubility issues of sodium *para*-toluenesulfinate in THF. The coupling of *para*-tolyl(triethoxy)silane with phenyl iodide in DMF did not give any of the desired sulfone.

1.7. NMR spectra of isolated compounds

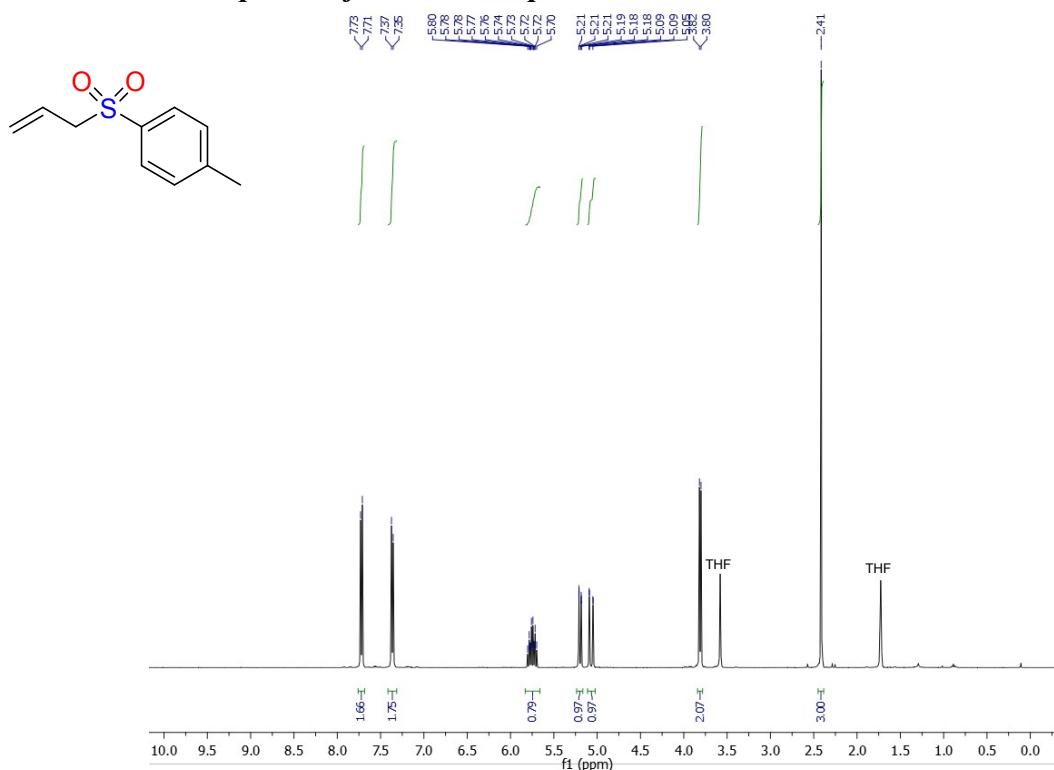


Figure S5: ^1H NMR spectrum of (3a) in d_8 -THF

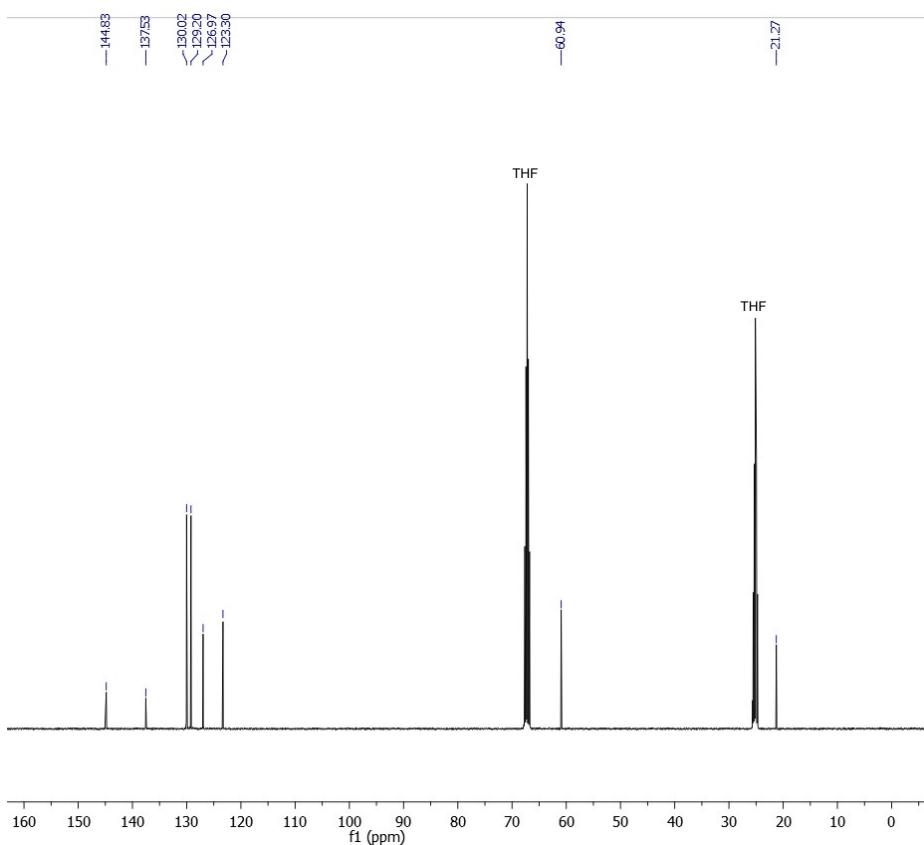


Figure S6: $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of (3a) in d_8 -THF

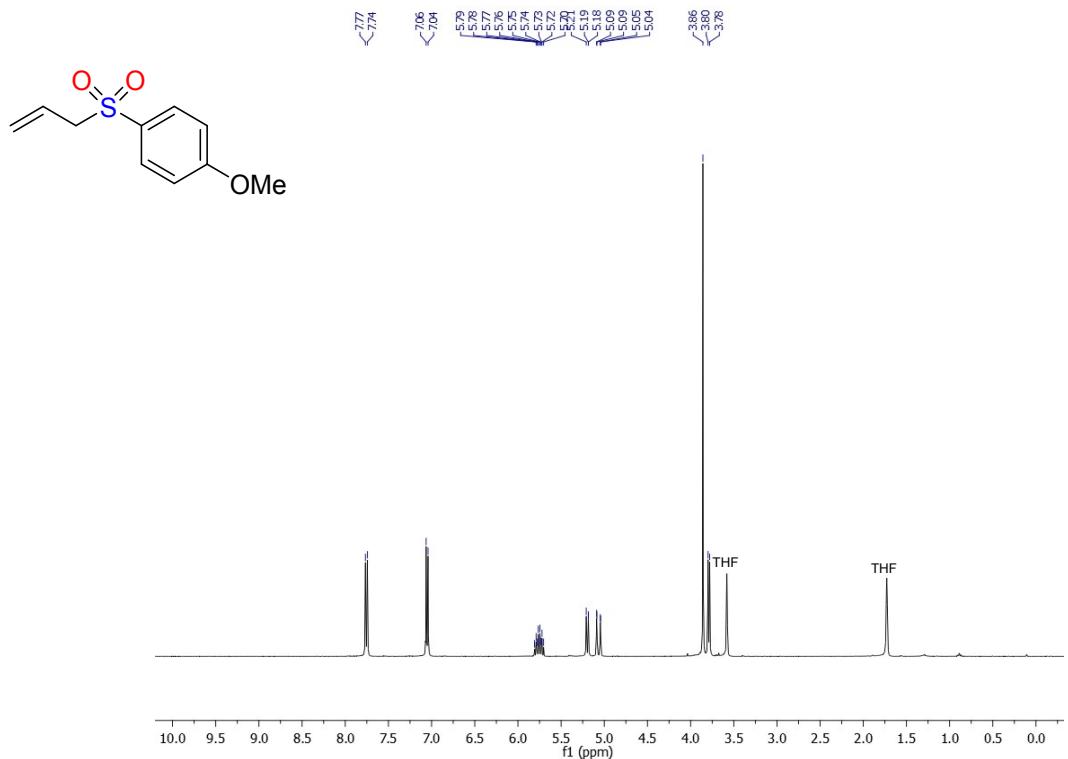


Figure S7: ¹H NMR spectrum of (3b) in *d*₈-THF

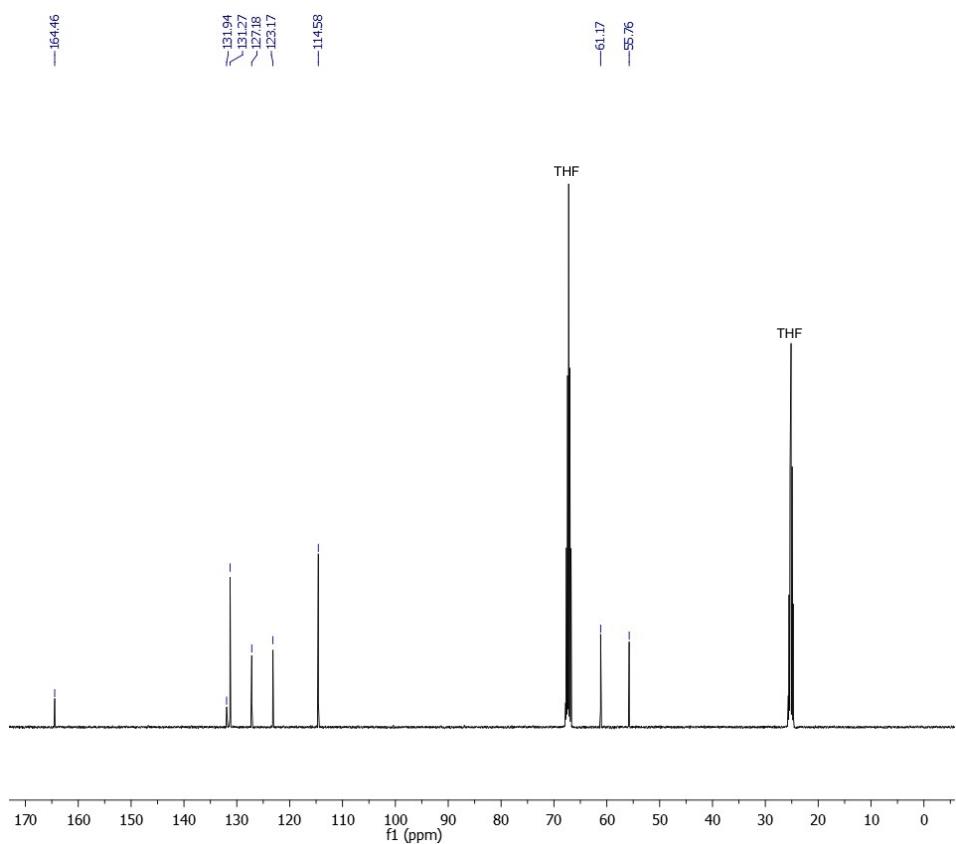
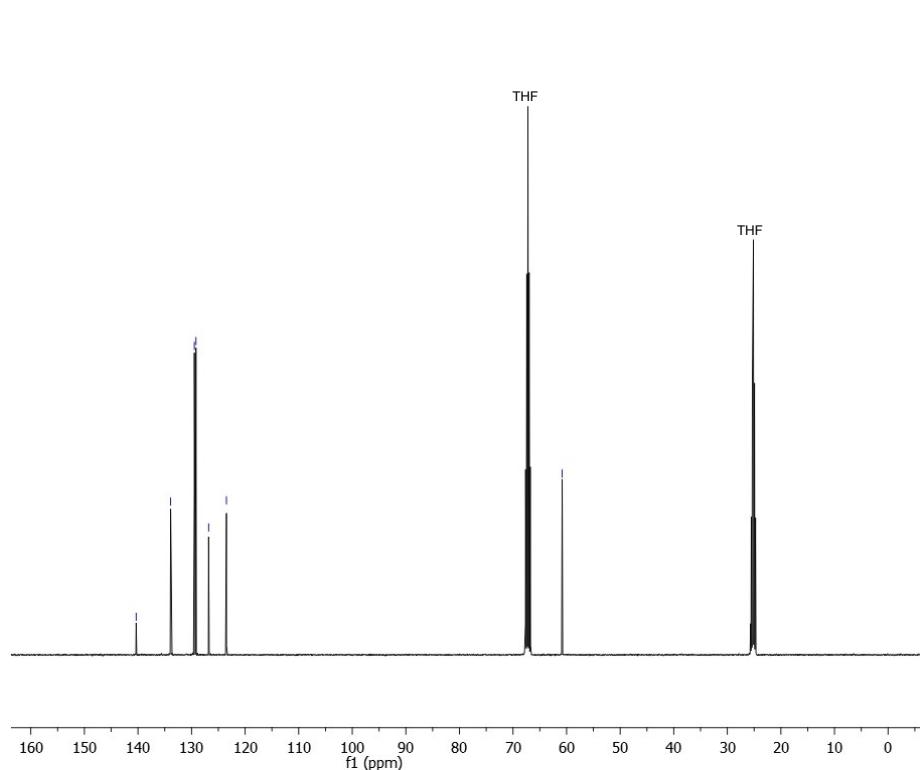
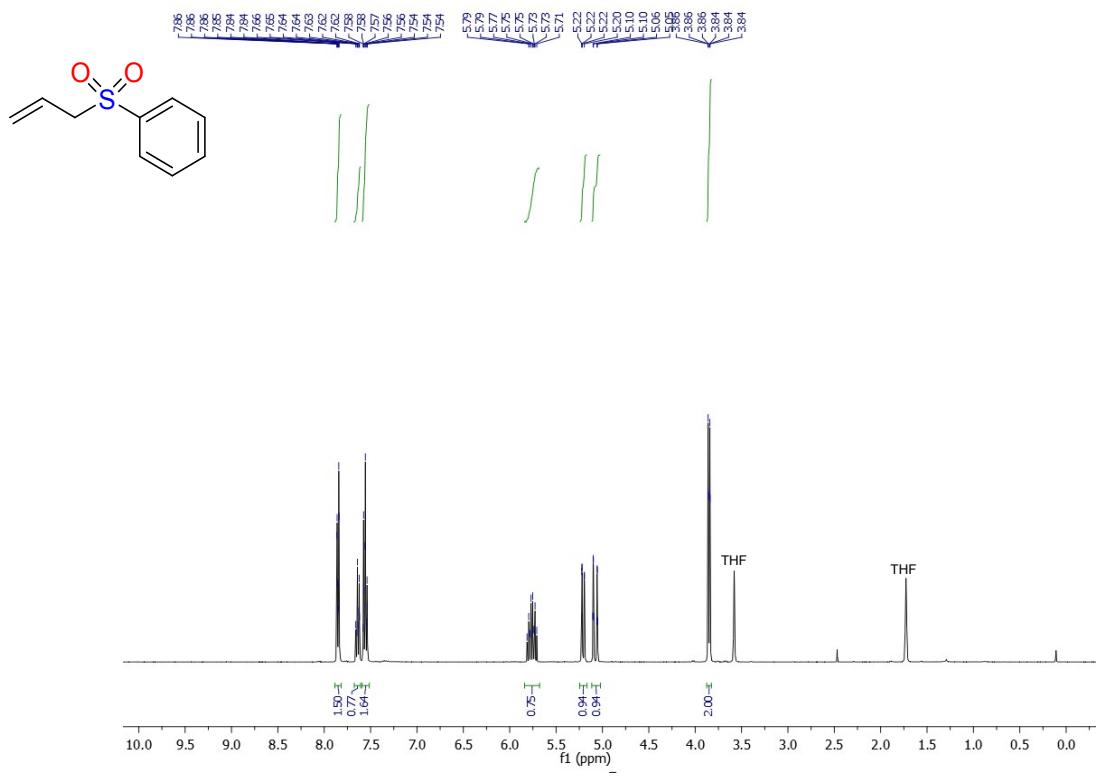


Figure S8: ¹³C{¹H} NMR spectrum of (3b) in *d*₈-THF



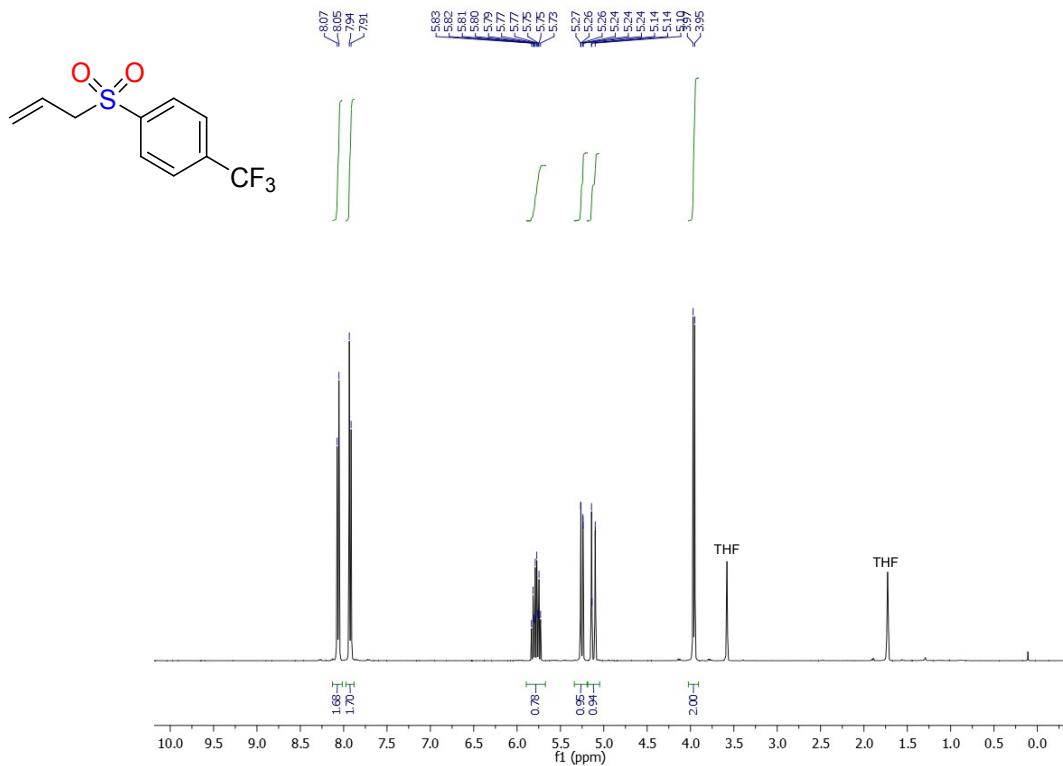


Figure S11: ^1H NMR spectrum of (3d) in d_8 -THF

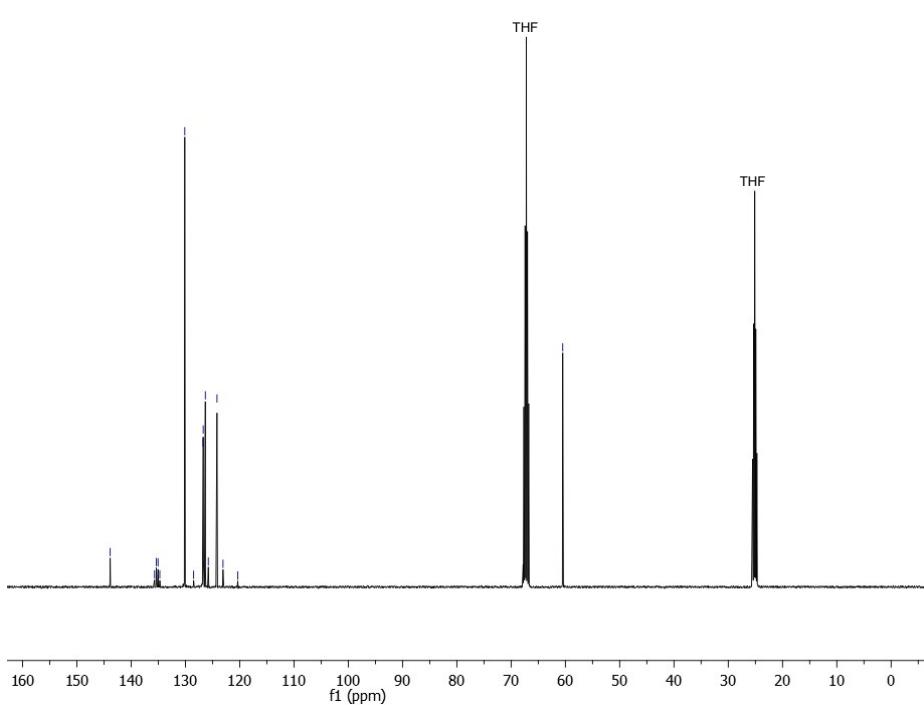


Figure S12: $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of (3d) in d_8 -THF

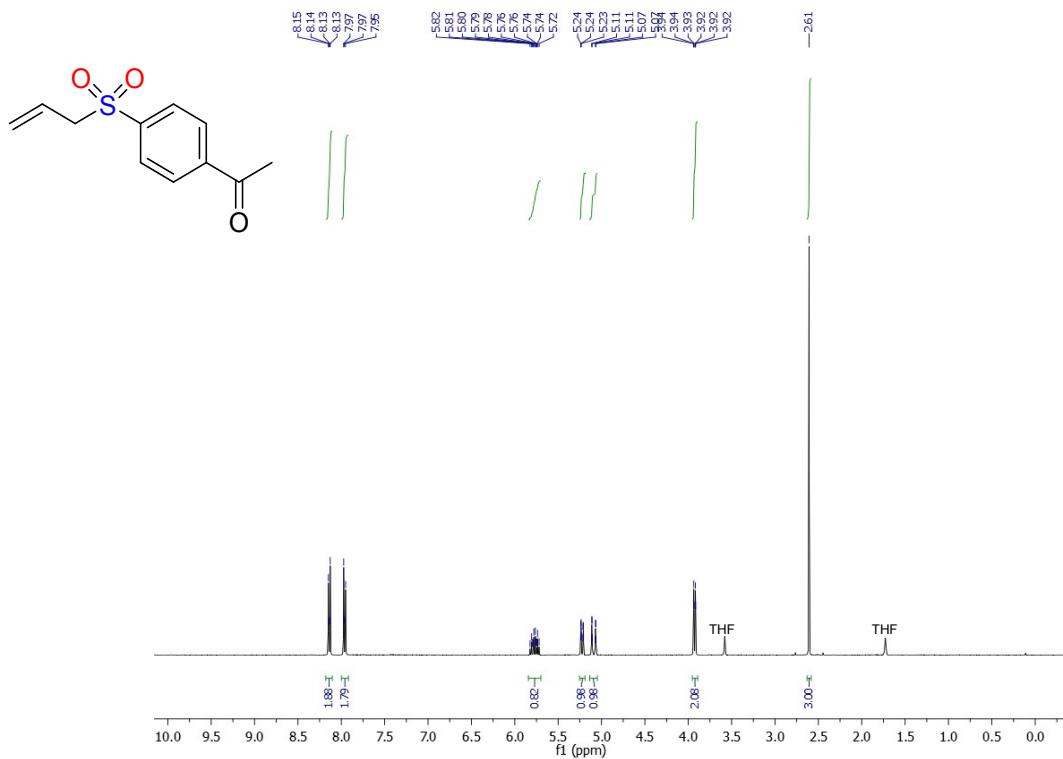


Figure S13: ^1H NMR spectrum of (3e) in d_8 -THF

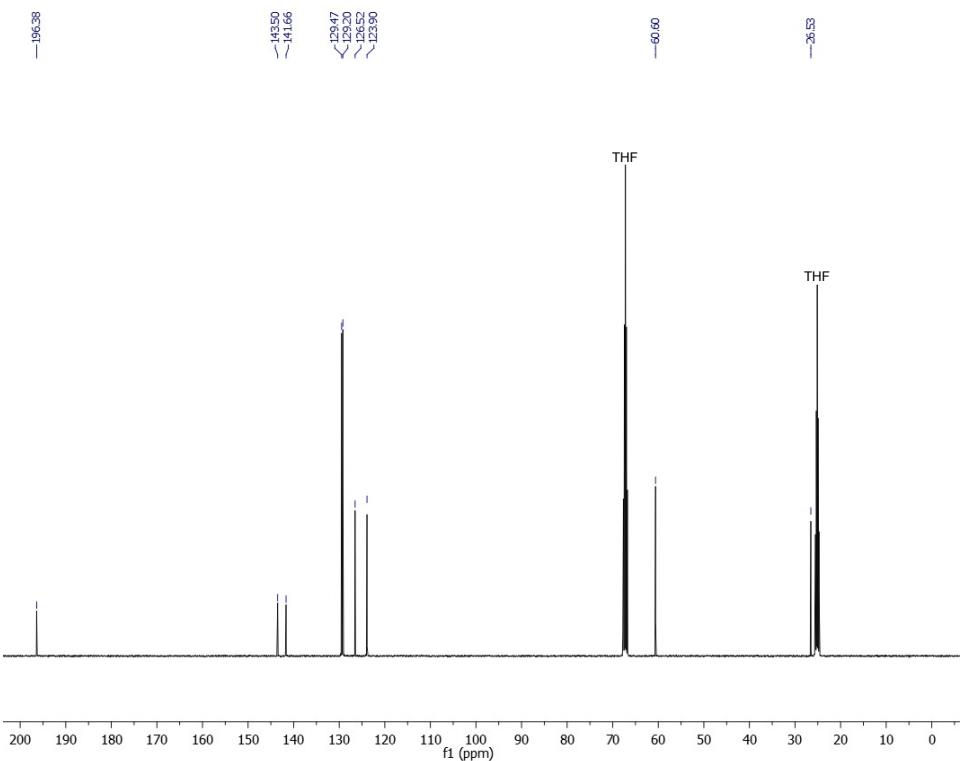


Figure S14: $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of (3e) in d_8 -THF

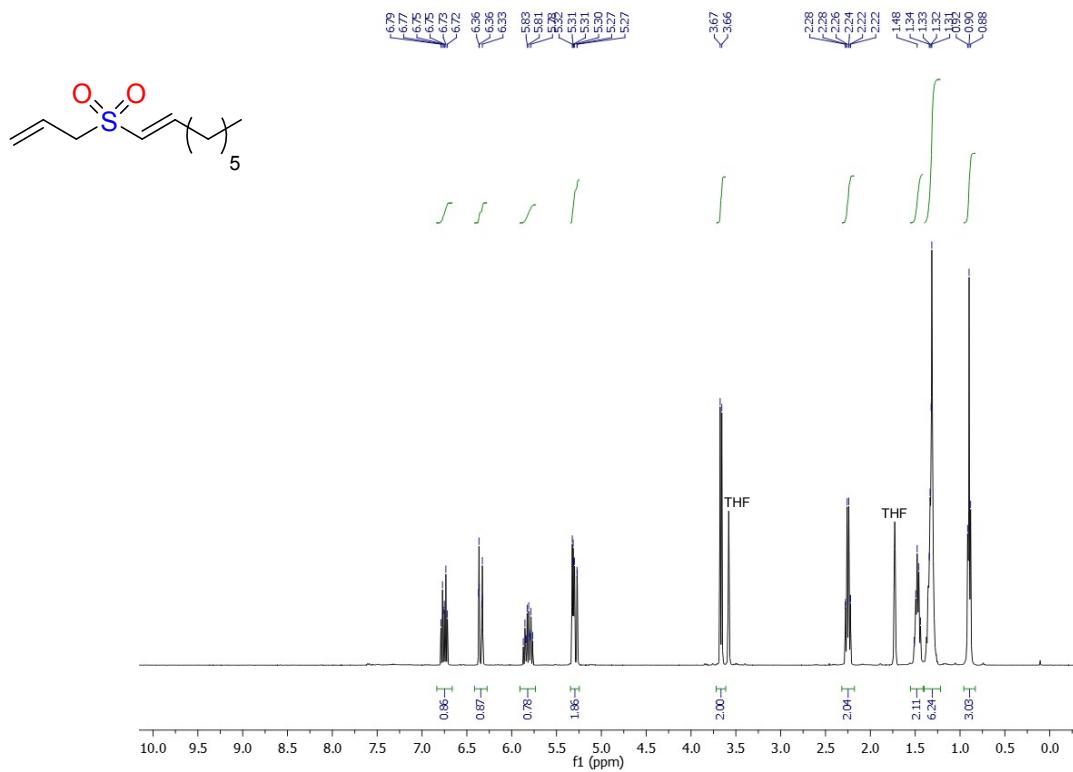


Figure S15: ^1H NMR spectrum of (3f) in d_8 -THF

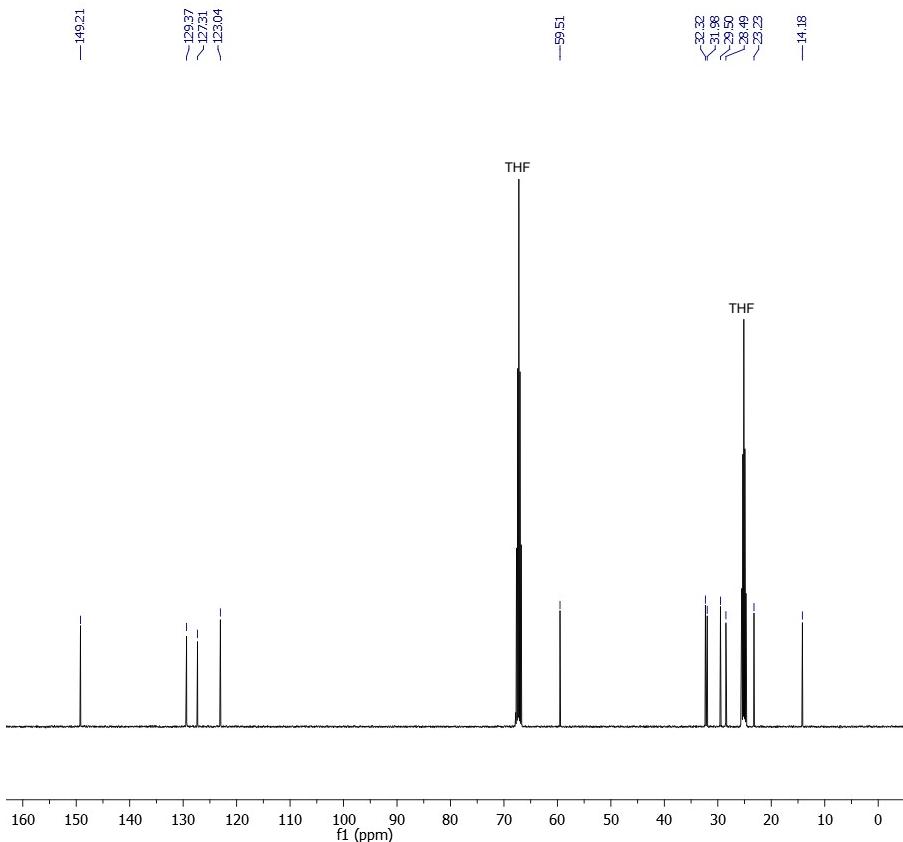


Figure S16: $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of (3f) in d_8 -THF

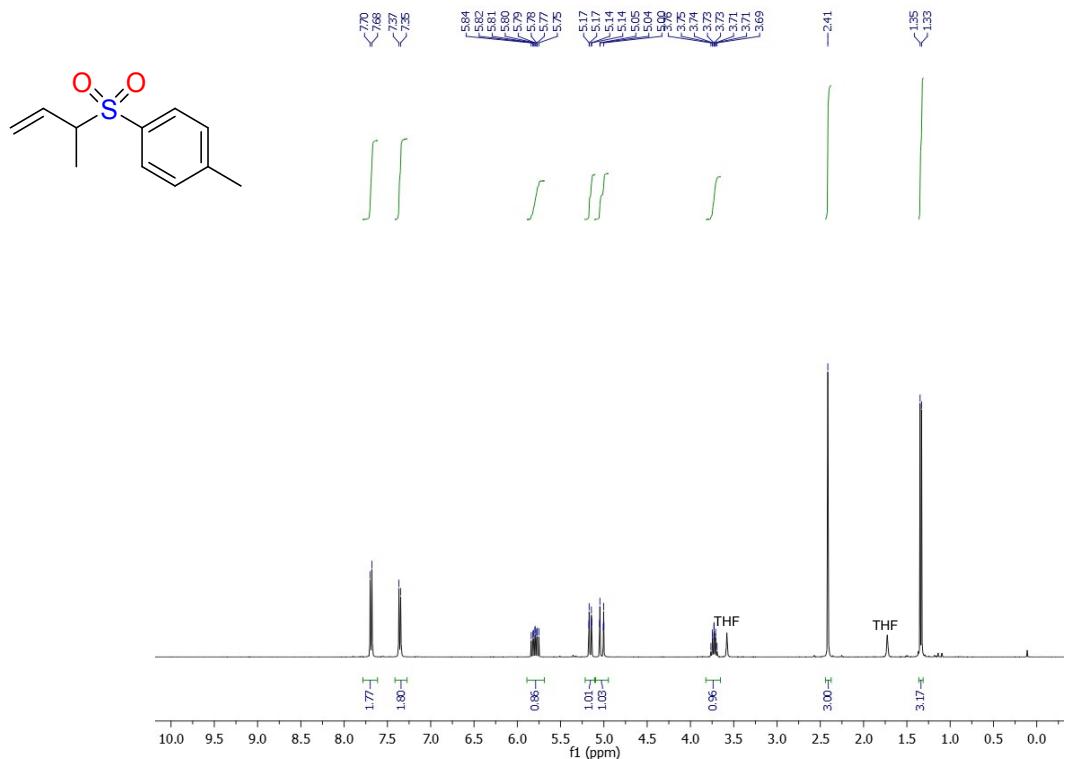


Figure S17: ^1H NMR spectrum of (3g) in d_8 -THF

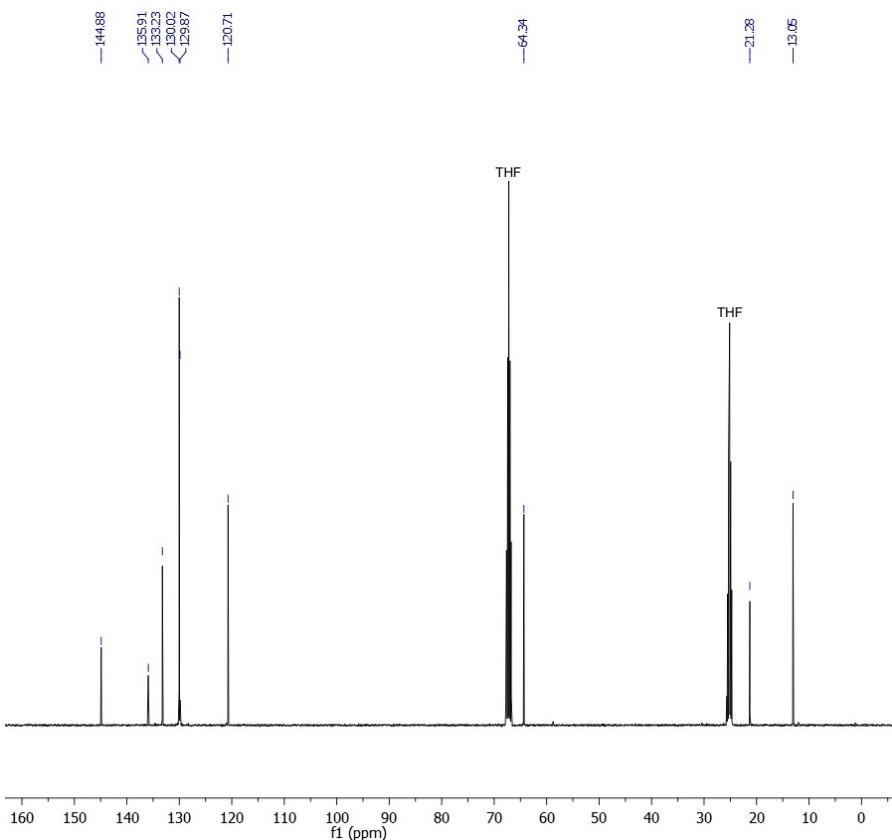


Figure S18: $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of (3g) in d_8 -THF

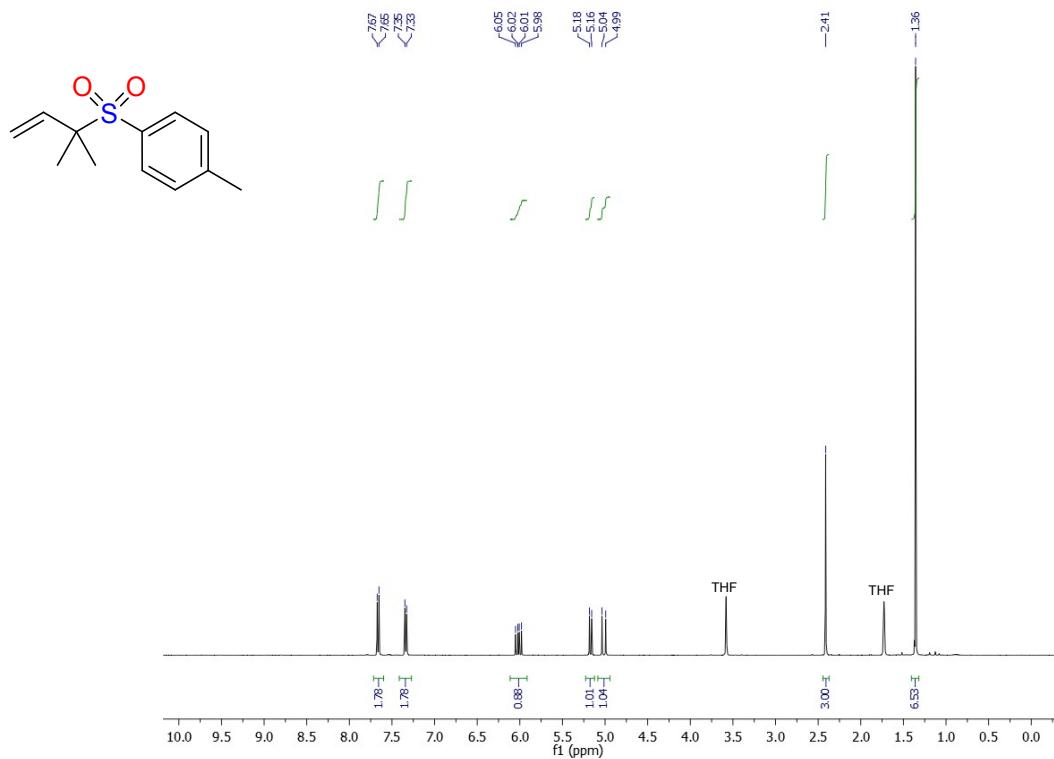


Figure S19: ^1H NMR spectrum of (3h) in d_8 -THF

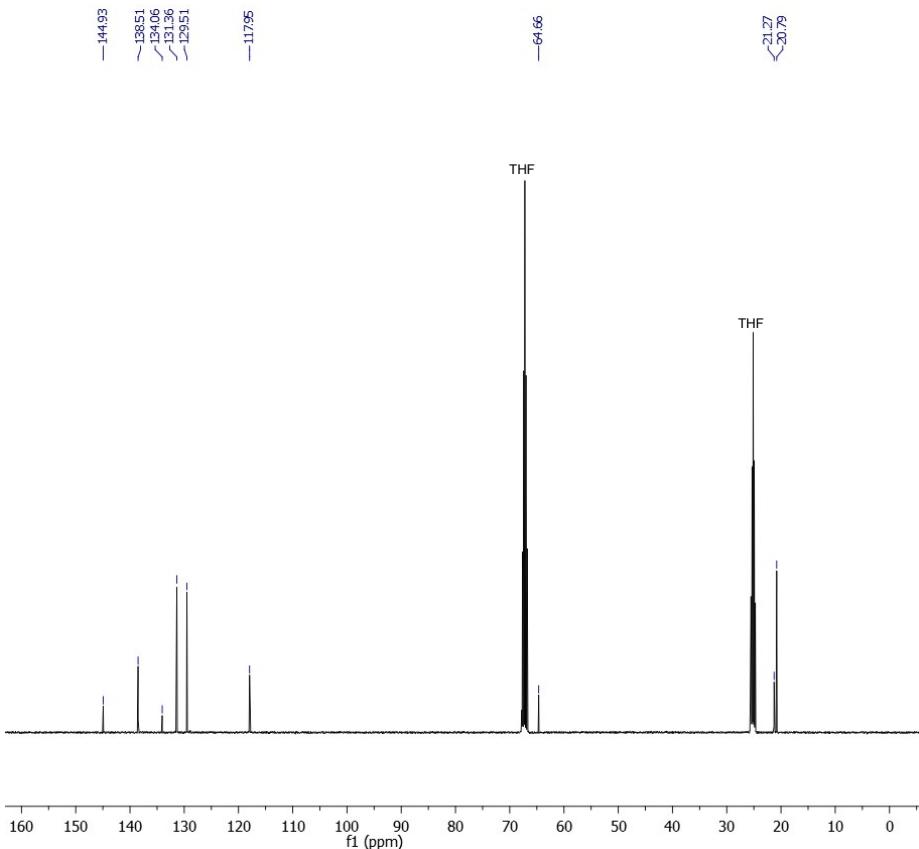


Figure S20: $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of (3h) in d_8 -THF

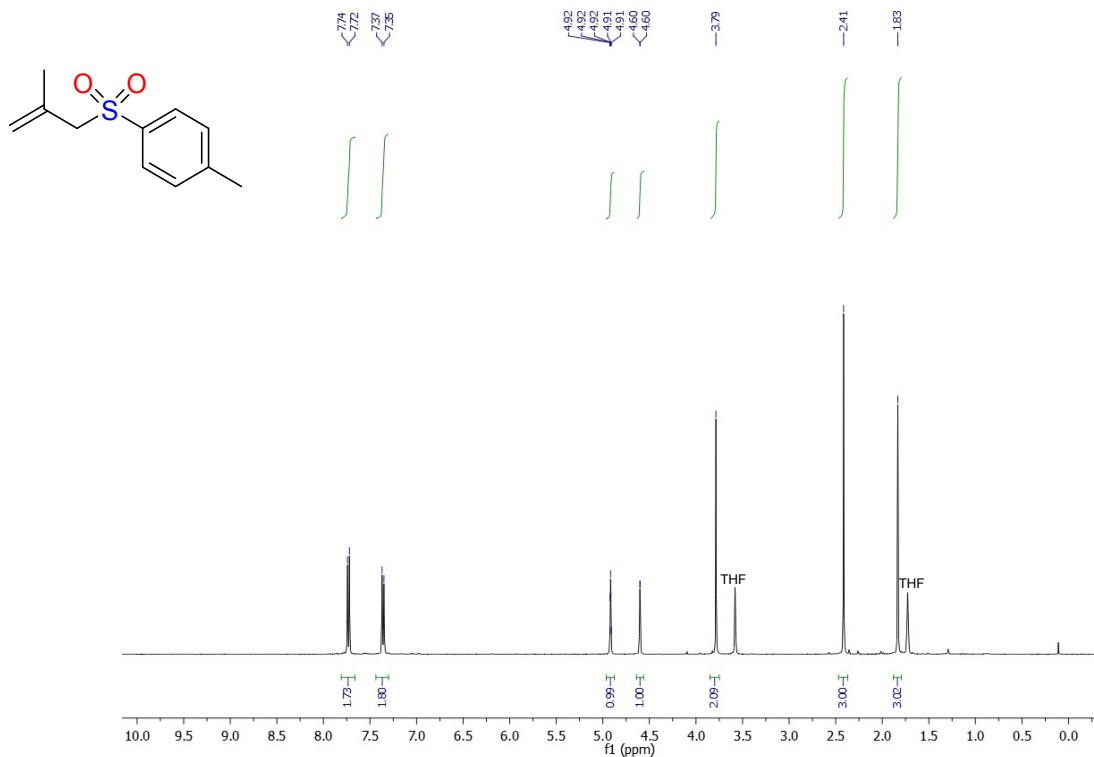


Figure S21: ^1H NMR spectrum of (3i) in d_8 -THF

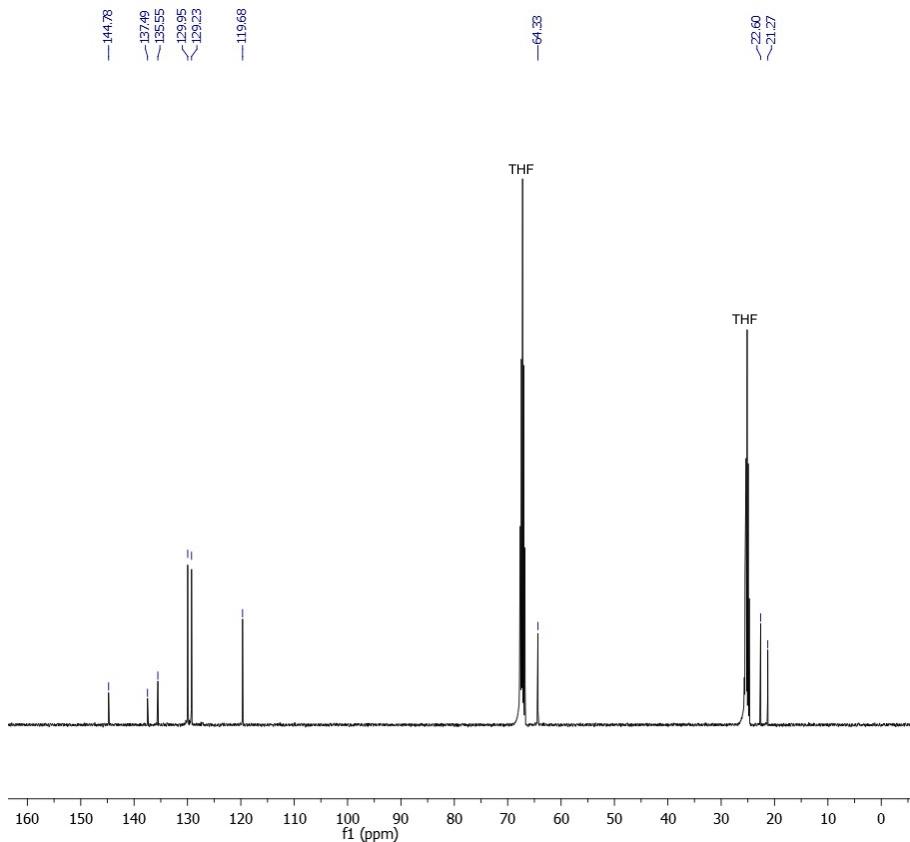


Figure S22: $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of (3i) in d_8 -THF

2. Computational Details

Density functional theory⁸ was applied to determine the structural and energetic features of the intermediates and transition states described herein. Calculations were performed using Gaussian09 version D.01 code.⁹ The 6-311+G(d) basis set was used for atoms C, H, N, and O whereas 6-311++G(d,p) was used for P, Si, S and F (unless for the CF₃ moiety for which 6-311+G(d) was used). The Becke three-parameter hybrid functional B3LYP¹⁰ was used with Grimme's D3 empirical dispersion.¹¹ All structures were optimized without geometrical constraint; stationary points were characterized by frequency calculations (one negative frequency for a transition state, no negative frequency for minima). Solvent effects (THF) were included in structure-optimization and frequency calculations using the PCM model implemented in the Gaussian code.¹² All relative energies (corrected for ZPE contributions) and Gibbs free energies (T= 298 K, P= 1 atm) are reported in kcal.mol⁻¹.

2.1. Computed potential energy surface for the formation of the sulfinate

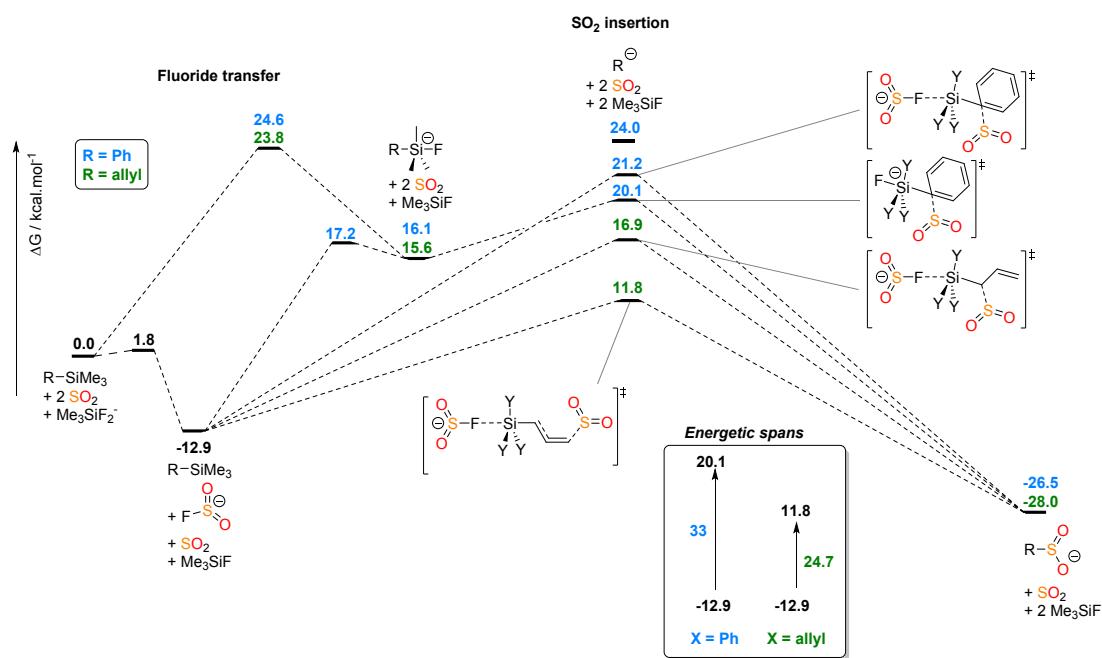


Figure S23. Potential energy surface with different reaction pathways for the formation of **8a** and **8b**. Values given correspond to Gibbs free energies with respect to the starting material ($G=0.0 \text{ kcal.mol}^{-1}$).

2.2. Energies and frequencies of computed compounds and transition states

All energies are in Hartree, imaginary frequencies are in cm^{-1} , the 3 displayed frequencies being the lowest ones.¹³

2.2.1. Fluoro-affinity of allylsilanes



Element	X	Y	Z
Si	-0.00039800	-0.00028300	0.00678600
C	-1.78839099	0.13964500	-0.52003801
H	-1.87244499	0.13820601	-1.61166704
H	-2.38248110	-0.69743198	-0.14163300
H	-2.24080396	1.06627500	-0.15475300
C	0.77372801	-1.61909997	-0.51701301
H	1.79871595	-1.70989096	-0.14558101
H	0.20276000	-2.47399211	-0.14293700
H	0.80928200	-1.69711006	-1.60850894
C	1.01639700	1.47877204	-0.51552802
H	0.58548301	2.41194892	-0.14098001
H	2.04286790	1.40847194	-0.14366600
H	1.06414795	1.55097198	-1.60698497
F	-0.00281200	-0.00075200	1.66680801

	1	2	3
	A	A	A
Frequencies --	133.2221	150.0940	150.7969
Red. masses --	1.0099	1.0398	1.0441
Frc consts --	0.0106	0.0138	0.0140
IR Inten --	0.0012	0.0037	0.0020
Sum of electronic and zero-point Energies=		-509.161188	
Sum of electronic and thermal Energies=		-509.152799	
Sum of electronic and thermal Enthalpies=		-509.151855	
Sum of electronic and thermal Free Energies=		-509.193018	

HF = -509.274188643



Element	X	Y	Z
Si	0.00002300	-0.00374000	0.00017400
F	1.82993305	-0.00779900	-0.00021700
F	-1.82990003	-0.00667100	0.00038200
C	0.00016900	-0.40796900	1.86227906
H	0.00024400	0.52507699	2.44042206
H	-0.89006698	-0.96823102	2.15952492
H	0.89037102	-0.96834201	2.15942097
C	0.00049700	1.81463504	-0.57021999
H	0.00042600	1.85983098	-1.66680098
H	-0.88933599	2.34917307	-0.22719900
H	0.89068800	2.34867191	-0.22734299

C	-0.00064800	-1.40778697	-1.28868306
H	-0.00117300	-2.37882304	-0.77684897
H	-0.89167899	-1.37954795	-1.92166102
H	0.89042199	-1.38047802	-1.92164397

	1	2	3
	A	A	A
Frequencies --	-59.1633	58.8019	77.2376
Red. masses --	1.0447	1.0555	1.0637
Frc consts --	0.0022	0.0022	0.0037
IR Inten --	0.0174	0.1991	0.0683
Sum of electronic and zero-point Energies=		-609.187295	
Sum of electronic and thermal Energies=		-609.178045	
Sum of electronic and thermal Enthalpies=		-609.177101	
Sum of electronic and thermal Free Energies=		-609.221013	

HF = -609.301004925

allylSiMe₃

Element	X	Y	Z
C	-1.44365597	0.69515097	-1.07747304
H	-1.09742904	0.34229699	-2.04608107
H	-2.51353192	0.65618497	-0.89814502
C	-0.59480202	1.15516400	-0.15485600
H	-1.00266504	1.49807501	0.79618502
C	0.88665301	1.29493999	-0.31443399
H	1.24518001	0.70304102	-1.16501403
H	1.40360796	0.91935402	0.57796001
Si	1.44796395	3.10161090	-0.57927698
C	0.79319000	4.17225790	0.82978803
H	1.12179196	5.21095085	0.72204798
H	-0.30085900	4.17432308	0.85578698
H	1.14714599	3.81346607	1.80176103
C	0.76701897	3.71486592	-2.22658205
H	1.03652704	4.76087189	-2.40428901
H	1.15870702	3.12561107	-3.06205297
H	-0.32429901	3.64228606	-2.25439095
C	3.33434010	3.14177108	-0.59949601
H	3.73826909	2.50727606	-1.39493001
H	3.70720792	4.15747976	-0.76669800
H	3.75103688	2.78912306	0.34939301

	1	2	3
	A	A	A
Frequencies --	45.8756	70.3601	134.8255
Red. masses --	2.4757	2.8714	1.0447
Frc consts --	0.0031	0.0084	0.0112
IR Inten --	0.1231	0.0650	0.0044
Sum of electronic and zero-point Energies=		-526.496092	
Sum of electronic and thermal Energies=		-526.484597	
Sum of electronic and thermal Enthalpies=		-526.483653	
Sum of electronic and thermal Free Energies=		-526.532821	

HF = -526.676627637

allylSiMe₃F⁻

Element	X	Y	Z
C	-0.20242301	-0.04070200	-1.06762397
H	0.00664000	-0.08573500	-2.13468504
H	-1.03511500	-0.64058697	-0.71052700
C	0.52529198	0.72984701	-0.23934400
H	0.24591000	0.71441603	0.81671298
C	1.60681498	1.66693294	-0.59084803
H	2.01527500	1.45289302	-1.58831203
H	2.42996693	1.59631598	0.13393401
C	0.45963201	3.60594392	1.23009706
H	0.45873600	4.59039688	1.70236099
H	-0.57205898	3.23275709	1.24625301
H	1.04914498	2.91989589	1.84832895
C	-0.13093100	3.52047706	-2.04227710
H	-1.10293698	3.95155907	-1.78207302
H	0.25059599	4.10877609	-2.88455009
H	-0.29305300	2.49477291	-2.38329911
C	2.89028406	4.22866583	-0.92740399
H	3.59242702	3.40743303	-1.10166705
H	2.91037512	4.89032507	-1.79947305
H	3.26461697	4.82013798	-0.08400100
Si	1.09222102	3.66228390	-0.57802498
F	0.65478402	5.53539181	-0.58197898

	1	2	3
	A	A	A
Frequencies --	88.8054	98.9997	111.5108
Red. masses --	1.6080	1.8254	1.7937
Frc consts --	0.0075	0.0105	0.0131
IR Inten --	0.5295	2.1026	1.2362

Sum of electronic and zero-point Energies= -626.498718
 Sum of electronic and thermal Energies= -626.486094
 Sum of electronic and thermal Enthalpies= -626.485150
 Sum of electronic and thermal Free Energies= -626.535962

HF = -626.680284135

allylSiMe₂OMe

Element	X	Y	Z
C	-1.44664097	0.78712302	-1.05934405
H	-1.09314299	0.31428301	-1.97262895
H	-2.52216697	0.83801299	-0.92171502
C	-0.60150498	1.28250802	-0.15272100
H	-1.01147497	1.74987996	0.74166399
C	0.89229298	1.29914701	-0.26620799
H	1.23054802	0.63668603	-1.07216799
H	1.35121500	0.93361199	0.66119897
Si	1.60210204	3.02170205	-0.59438097

C	0.92610502	3.74100494	-2.19128799
H	1.30025804	4.75349617	-2.37142897
H	1.22166705	3.12774611	-3.04890609
H	-0.16652900	3.78276205	-2.17480898
C	3.47969890	2.96492505	-0.63925201
H	3.83319092	2.33036089	-1.45845795
H	3.90907788	3.96022201	-0.79136199
H	3.88860011	2.56413007	0.29316300
O	1.08436894	3.93935204	0.71567899
C	1.33574998	5.33184290	0.86168897
H	0.88354701	5.66252708	1.79852998
H	2.41030312	5.54666519	0.90163600
H	0.89373302	5.90911198	0.04120600

	1	2	3
	A	A	A
Frequencies --	45.0064	58.3862	73.5298
Red. masses --	2.8099	2.3850	2.8128
Frc consts --	0.0034	0.0048	0.0090
IR Inten --	0.0291	5.8949	0.5001

Sum of electronic and zero-point Energies= -601.762247
 Sum of electronic and thermal Energies= -601.749612
 Sum of electronic and thermal Enthalpies= -601.748668
 Sum of electronic and thermal Free Energies= -601.801268

HF = -601.948683982

allylSiMe₂OMeF⁻

Element	X	Y	Z
C	-0.07571800	-0.03125500	-1.05911505
H	0.09804000	-0.02718200	-2.13344908
H	-0.83331001	-0.71887600	-0.69297302
C	0.60663801	0.77972102	-0.23292200
H	0.36904499	0.70920902	0.83092499
C	1.59699500	1.81441295	-0.59505200
H	1.99003398	1.65124297	-1.60652006
H	2.44501591	1.79412401	0.10103900
C	0.30366901	3.59916592	1.27241695
H	0.07968900	4.56740808	1.72454906
H	-0.61524802	3.00006890	1.29437602
H	1.02914202	3.07506990	1.90506995
C	-0.28830001	3.53803611	-1.99413204
H	-1.25168705	4.01334286	-1.78465605
H	0.12991600	4.05570412	-2.86638308
H	-0.46890801	2.49711609	-2.27327800
Si	0.92939502	3.73358393	-0.53212798
F	0.46803600	5.52477407	-0.52055699
O	2.56393790	4.19144392	-0.84440601
C	3.07585096	5.49755192	-1.01072395
H	2.94592190	6.11317682	-0.11316000
H	4.14895010	5.41323280	-1.21968305
H	2.60179591	6.02812576	-1.84423697

	1	2	3
	A	A	A
Frequencies --	64.1790	83.8809	112.4479
Red. masses --	2.1892	2.2714	1.5106
Frc consts --	0.0053	0.0094	0.0113
IR Inten --	1.5016	6.2682	0.1706
Sum of electronic and zero-point Energies=		-701.771460	
Sum of electronic and thermal Energies=		-701.758003	
Sum of electronic and thermal Enthalpies=		-701.757059	
Sum of electronic and thermal Free Energies=		-701.810236	

HF = -701.959120013

allylSiOMe₃

Element	X	Y	Z
C	-3.61613989	-0.36729699	0.04063500
H	-3.66644597	-1.03211701	0.89936298
H	-4.38688517	-0.48521799	-0.71440101
C	-2.65382791	0.54764700	-0.08305300
H	-2.64436793	1.18477297	-0.96588600
C	-1.53248203	0.76823598	0.89135700
H	-1.72383106	0.24726900	1.83563995
H	-1.43306506	1.83442402	1.12833405
Si	0.11917300	0.19137999	0.24587700
O	0.19008200	-1.46184397	0.28358701
O	1.31209099	0.80908000	1.21436799
O	0.29398701	0.73376399	-1.30930400
C	0.73068100	-2.32071900	-0.71864599
H	1.82355905	-2.35088801	-0.65778202
H	0.34233099	-3.32557797	-0.54760402
H	0.43924901	-1.99471700	-1.72117496
C	2.56822300	0.19093899	1.49547899
H	2.98906589	0.66946602	2.38067389
H	2.44827199	-0.87827098	1.69242299
H	3.26377797	0.32270801	0.66041899
C	1.52782094	1.00505197	-1.97278798
H	2.17379189	1.64296305	-1.36211598
H	2.06223392	0.07891600	-2.20859098
H	1.30113995	1.52372599	-2.90504909

	1	2	3
	A	A	A
Frequencies --	27.2697	46.3004	60.1739
Red. masses --	2.8692	3.0630	2.6643
Frc consts --	0.0013	0.0039	0.0057
IR Inten --	5.3245	1.0778	2.3730
Sum of electronic and zero-point Energies=		-752.297607	
Sum of electronic and thermal Energies=		-752.282691	
Sum of electronic and thermal Enthalpies=		-752.281747	
Sum of electronic and thermal Free Energies=		-752.341339	

HF = -752.496295063

allylSiOMe₃F⁻

Element	X	Y	Z
C	-0.03704200	-0.02022600	-1.24291801
H	0.40411201	-0.11158600	-2.23371196
H	-0.85414898	-0.69832599	-1.01038098
C	0.39205000	0.89819801	-0.36649299
H	-0.10307600	0.94289899	0.60407901
C	1.46000898	1.91070998	-0.57492399
H	2.00504494	1.71461105	-1.50543594
H	2.18500710	1.84792697	0.24736001
Si	0.87976098	3.78083396	-0.58113497
F	0.38081601	5.47853804	-0.55906898
O	2.52100492	4.26756001	-0.72601497
C	3.01859188	5.58460188	-0.85713297
H	2.78513908	6.20477009	0.01590500
H	4.10804605	5.52696323	-0.95911998
H	2.61581397	6.09315920	-1.74078798
O	-0.03186200	3.53090191	-2.00807309
O	0.17571300	3.56714106	0.97088999
C	-0.87784702	4.44656515	-2.67318511
H	-0.33085901	5.31658888	-3.05524111
H	-1.33679795	3.92762399	-3.52225804
H	-1.67937303	4.81723213	-2.02359009
C	-0.51740700	4.51953411	1.75201595
H	-0.85033703	4.02501822	2.67157197
H	0.11747800	5.36940193	2.02879596
H	-1.39853597	4.91766024	1.23565102

	1	2	3
	A	A	A
Frequencies --	39.5420	59.4832	71.6783
Red. masses --	2.8786	2.9544	3.0763
Frc consts --	0.0027	0.0062	0.0093
IR Inten --	4.7772	2.8647	1.6293

Sum of electronic and zero-point Energies= -852.322224
 Sum of electronic and thermal Energies= -852.306918
 Sum of electronic and thermal Enthalpies= -852.305974
 Sum of electronic and thermal Free Energies= -852.364591

HF = -852.522162510

allylSiOEt₃

Element	X	Y	Z
C	-1.39342296	0.38068399	-0.66662699
H	-1.28114104	0.04532200	-1.69474900
H	-2.38785100	0.31286100	-0.23686400
C	-0.36208799	0.85694802	0.03217600
H	-0.53070903	1.18355894	1.05771303
C	1.03975201	1.01640296	-0.47857699
H	1.17339802	0.49603099	-1.43328094
H	1.75397694	0.57498598	0.22815600
Si	1.54977405	2.80125594	-0.70978600

O	1.43763399	3.57123208	0.74883401
C	0.84645301	4.84035778	1.05541396
H	0.25998500	4.70424318	1.96831405
H	0.15583400	5.12875509	0.25709200
O	3.10867000	2.83816195	-1.25881100
O	0.55217499	3.52374601	-1.80958402
C	0.89016098	4.34033918	-2.93720102
H	0.29013899	3.98400593	-3.77924109
H	1.94400203	4.19775915	-3.19580603
C	4.21818686	3.60468507	-0.77283800
H	5.07198381	2.92423892	-0.70943701
H	4.00459290	3.96543503	0.23805600
C	0.59667200	5.80587721	-2.66081405
H	-0.45282200	5.94584322	-2.38991189
H	1.21358299	6.18716478	-1.84289598
H	0.80493802	6.40764523	-3.55068493
C	1.90866101	5.90774393	1.26177800
H	2.48348498	6.07827616	0.34765601
H	1.44203699	6.85471916	1.54955602
H	2.60441899	5.61423922	2.05190492
C	4.53264809	4.76457596	-1.70328200
H	5.41362190	5.30710888	-1.34701598
H	4.73696280	4.40525198	-2.71500206
H	3.69888592	5.46974516	-1.75394201

	1	2	3
	A	A	A
Frequencies --	39.0007	43.9306	54.1521
Red. masses --	3.5662	3.5481	3.3802
Frc consts --	0.0032	0.0040	0.0058
IR Inten --	0.8384	1.9554	2.3788

Sum of electronic and zero-point Energies= -870.202222
 Sum of electronic and thermal Energies= -870.184259
 Sum of electronic and thermal Enthalpies= -870.183315
 Sum of electronic and thermal Free Energies= -870.249226

HF = -870.486948951

allylSiOEt₃F⁻

Element	X	Y	Z
C	-0.06599000	0.15781701	-1.22751606
H	0.27510700	0.14078900	-2.26100898
H	-0.82663798	-0.56957501	-0.95538300
C	0.41035900	1.04784405	-0.34629899
H	0.01497000	1.01739502	0.66952699
C	1.40457594	2.12185597	-0.59872299
H	1.85654104	2.00779796	-1.59088504
H	2.21348906	2.05066299	0.14015000
Si	0.73911101	3.95808291	-0.43603200
F	0.16284300	5.62454700	-0.27808800
O	2.34471989	4.53189421	-0.65128303
C	2.80557394	5.86541224	-0.77699399
H	2.30207705	6.52215481	-0.05965500
H	3.87371612	5.85750818	-0.52231401

O	-0.26686800	3.73883796	-1.80604196
O	0.16602300	3.61258602	1.14584005
C	-1.21140206	4.62777710	-2.37400603
H	-0.79866499	5.63988400	-2.45471406
H	-1.40613401	4.27132416	-3.39418793
C	-0.44637200	4.47602510	2.08670592
H	-1.01697302	3.83997893	2.77668309
H	-1.15852404	5.14833689	1.59654999
C	0.57665098	5.29110718	2.87157893
H	1.28718495	4.63080597	3.37827802
H	1.13587999	5.94521284	2.19854093
H	0.08550900	5.91202116	3.62966895
C	-2.51906705	4.66584015	-1.58924901
H	-2.94214702	3.66066194	-1.50116503
H	-2.34429789	5.05455923	-0.58371198
H	-3.25599599	5.30712080	-2.08648491
C	2.62397099	6.40868092	-2.19100595
H	3.13070297	5.76618290	-2.91766310
H	1.56335497	6.44802999	-2.44982004
H	3.03921199	7.41933918	-2.27849102

	1	2	3
	A	A	A
Frequencies --	53.6373	56.4138	62.7441
Red. masses --	3.0930	3.2301	3.0814
Frc consts --	0.0052	0.0061	0.0071
IR Inten --	1.5931	1.0935	1.4010
Sum of electronic and zero-point Energies=		-970.225090	
Sum of electronic and thermal Energies=		-970.206519	
Sum of electronic and thermal Enthalpies=		-970.205575	
Sum of electronic and thermal Free Energies=		-970.271487	

HF = -970.510805356

2.2.2. Formation of the sultinate

SO₂

Element	X	Y	Z
S	0.36723700	2.31060505	7.78685093
O	-0.32170501	3.59135199	7.92644501
O	1.01494706	1.75963497	8.97487831

	1	2	3
	A	A	A
Frequencies --	499.1558	1127.5937	1278.5302
Red. masses --	18.5513	18.0149	20.2749
Frc consts --	2.7233	13.4955	19.5268
IR Inten --	57.1188	52.6412	381.9765
Sum of electronic and zero-point Energies=		-548.664618	
Sum of electronic and thermal Energies=		-548.661526	
Sum of electronic and thermal Enthalpies=		-548.660582	

Sum of electronic and thermal Free Energies= -548.689503

HF = -548.671236713

FSO₂⁻

Element	X	Y	Z
S	0.11493900	-1.63376606	-0.00039500
F	-1.77339005	-1.56509101	0.00084900
O	0.39902601	-0.86115998	-1.24072504
O	0.40061200	-0.86229998	1.24027205

	1	2	3
	A	A	A
Frequencies --	287.6111	300.2414	428.4482
Red. masses --	17.8874	18.2373	22.1383
Frc consts --	0.8718	0.9686	2.3944
IR Inten --	11.5757	87.1137	269.4793

Sum of electronic and zero-point Energies= -648.710736
Sum of electronic and thermal Energies= -648.706515
Sum of electronic and thermal Enthalpies= -648.705571
Sum of electronic and thermal Free Energies= -648.737981

HF = -648.719284708

allyl-

Element	X	Y	Z
C	-1.51088405	0.40489000	-0.71583498
H	-1.30340505	-0.64169800	-0.93701297
H	-2.55363202	0.70898902	-0.71040398
C	-0.49151900	1.31902099	-0.45063800
H	-0.84383798	2.33732390	-0.24223500
C	0.89519298	1.17227697	-0.40225300
H	1.54242802	2.01403499	-0.17398600
H	1.37955499	0.21476300	-0.59193498

	1	2	3
	A	A	A
Frequencies --	259.7708	357.7850	432.7241
Red. masses --	1.3085	1.1773	1.8598
Frc consts --	0.0520	0.0888	0.2052
IR Inten --	1.1952	741.2045	4.7677

Sum of electronic and zero-point Energies= -117.323988
Sum of electronic and thermal Energies= -117.319541
Sum of electronic and thermal Enthalpies= -117.318597
Sum of electronic and thermal Free Energies= -117.348929

HF = -117.386842748

allylSO₂⁻

Element	X	Y	Z
C	-1.35658300	0.25180399	-0.27944300
H	-0.81307298	-0.52442700	0.25387201
H	-2.44027805	0.19034401	-0.25715899
C	-0.72156602	1.22856998	-0.93504298
H	-1.30738699	1.98729002	-1.45002699
C	0.74953097	1.41574597	-0.96423000
H	1.12961304	1.69563794	-1.95080400
H	1.29581499	0.54724801	-0.58868700
S	1.31382000	2.87515903	0.11825600
O	0.61651200	4.05869389	-0.57215297
O	2.82459402	2.86813307	-0.16837899

	1	2	3
	A	A	A
Frequencies --	74.1650	92.6784	189.8145
Red. masses --	3.1430	3.8312	3.1377
Frc consts --	0.0102	0.0194	0.0666
IR Inten --	4.3849	4.1295	1.3150
Sum of electronic and zero-point Energies=	-666.070430		
Sum of electronic and thermal Energies=	-666.063545		
Sum of electronic and thermal Enthalpies=	-666.062601		
Sum of electronic and thermal Free Energies=	-666.101980		

HF = -666.147650988

PhSiMe₃

Element	X	Y	Z
Si	1.47111201	-0.00969200	0.00000000
C	2.07206392	0.90138298	-1.53980005
H	1.74324000	0.39654499	-2.45330405
H	1.68729603	1.92557395	-1.57192802
H	3.16519189	0.95853698	-1.56414104
C	2.07208705	0.90152800	1.53970599
H	1.68734205	1.92572999	1.57173300
C	2.11965895	-1.77929294	0.00008000
H	1.78961694	-2.33339405	0.88414901
H	1.78963101	-2.33346796	-0.88394701
H	3.21427298	-1.78358400	0.00008900
C	-0.42352501	-0.02298200	0.00001300
C	-1.13973296	1.18760097	-0.00004500
C	-2.53401709	1.20937002	-0.00003800
H	-0.60574001	2.13495803	-0.00009600
C	-2.56307292	-1.20213401	0.00008500
C	-3.25040388	0.01105300	0.00002800
H	-3.06196499	2.15821695	-0.00008200
H	-3.11307001	-2.13839602	0.00013500
H	-4.33583498	0.02416400	0.00003300
H	3.16521597	0.95866001	1.56403697
H	1.74325597	0.39679000	2.45326209
C	-1.16647804	-1.21473503	0.00007700
H	-0.65440202	-2.17266107	0.00012200

	1	2	3
	A	A	A
Frequencies --	26.0169	84.3145	121.5663
Red. masses --	3.3381	3.1953	2.7327
Frc consts --	0.0013	0.0134	0.0238
IR Inten --	0.0021	0.2383	0.2377

Sum of electronic and zero-point Energies= -640.844456
 Sum of electronic and thermal Energies= -640.832186
 Sum of electronic and thermal Enthalpies= -640.831242
 Sum of electronic and thermal Free Energies= -640.882867

HF = -641.045509403

PhSiMe₃F⁻

Element	X	Y	Z
Si	1.36168206	0.02922300	0.00057300
C	1.56696498	1.81885803	-0.65019602
H	0.77521300	2.10801196	-1.34794700
H	1.54966497	2.54079700	0.17555401
H	2.53170800	1.93133199	-1.14819598
C	1.27300501	-0.33600399	1.88529098
H	0.68843901	0.44488299	2.38673401
C	1.61503005	-1.37042499	-1.28286099
H	1.66038895	-2.34987903	-0.79076701
H	0.81185001	-1.41738605	-2.02425599
H	2.56539202	-1.23538804	-1.80327797
F	3.29566002	0.01203100	0.21387900
C	-0.66612399	0.01405400	-0.08036600
C	-1.43979001	1.16498303	0.16783500
C	-2.83732891	1.14640999	0.21291700
H	-0.94143701	2.11779904	0.33553699
C	-2.79897499	-1.21533501	-0.22907400
C	-3.52829099	-0.04944300	0.01060600
H	-3.38795209	2.06473804	0.40489900
H	-3.31940889	-2.15768790	-0.38546699
H	-4.61409283	-0.07268900	0.04074800
H	2.24989796	-0.40462601	2.36791992
H	0.73086798	-1.27379704	2.05720592
C	-1.40191197	-1.17232096	-0.27009100
H	-0.87255198	-2.10489202	-0.45634699

	1	2	3
	A	A	A
Frequencies --	30.6652	87.1161	108.5669
Red. masses --	3.1465	3.8335	2.7561
Frc consts --	0.0017	0.0171	0.0191
IR Inten --	0.7985	0.7998	3.2065

Sum of electronic and zero-point Energies= -740.845215
 Sum of electronic and thermal Energies= -740.831491
 Sum of electronic and thermal Enthalpies= -740.830546
 Sum of electronic and thermal Free Energies= -740.885196

HF = -741.047123814

Ph⁻

Element	X	Y	Z
C	-3.24354196	0.66071898	-0.22072200
C	-1.84621596	0.67988002	-0.22029801
C	-1.16179597	1.90538001	-0.22093400
C	-1.79063594	3.17667294	-0.22199200
C	-3.20609808	3.08575392	-0.22236800
C	-3.92511511	1.88038695	-0.22177000
H	-3.78755307	-0.28096700	-0.22027799
H	-1.29363406	-0.26027799	-0.21952400
H	-0.06893300	1.84953403	-0.22055800
H	-3.80139589	4.00375509	-0.22314499
H	-5.01558113	1.88946199	-0.22211100

	1	2	3
	A	A	A
Frequencies --	359.6685	398.6895	603.7144
Red. masses --	2.6289	3.9032	7.0779
Frc consts --	0.2004	0.3655	1.5199
IR Inten --	0.0011	23.3412	0.7004
Sum of electronic and zero-point Energies=		-231.652171	
Sum of electronic and thermal Energies=		-231.647737	
Sum of electronic and thermal Enthalpies=		-231.646793	
Sum of electronic and thermal Free Energies=		-231.679633	

HF = -231.737903992

PhSO₂⁻

Element	X	Y	Z
C	-3.04858303	1.68083096	0.07317900
C	-1.65701699	1.73153806	-0.03074600
C	-1.00349104	2.96581101	-0.02299100
C	-1.73905003	4.14655876	0.09169200
C	-3.12945104	4.09185314	0.18862200
H	0.07812000	3.00828099	-0.11113200
H	-3.56003308	0.72283500	0.05961500
H	-1.08447504	0.81364298	-0.12135200
H	-1.24192500	5.11106110	0.08278000
S	-4.10193491	5.65522909	0.43406999
O	-3.30264997	6.66352606	-0.40956700
O	-5.45307922	5.29028177	-0.20509300
C	-3.78480196	2.86126089	0.18917900
H	-4.86726284	2.83769798	0.25440001

	1	2	3
	A	A	A
Frequencies --	46.6995	135.1530	190.8861
Red. masses --	6.6791	5.5625	7.1521

Frc consts -- 0.0086 0.0599 0.1535
 IR Inten -- 1.3604 7.0461 0.4178

Sum of electronic and zero-point Energies= -780.416242
 Sum of electronic and thermal Energies= -780.408539
 Sum of electronic and thermal Enthalpies= -780.407594
 Sum of electronic and thermal Free Energies= -780.449561

HF = -780.513319086

TS: F⁻ transfer from F₂SiMe₃⁻ to SO₂

Element	X	Y	Z
Si	2.32052207	-1.77358401	0.14601301
C	1.34828699	-0.17533900	-0.13734201
H	1.40064895	0.49458000	0.72527897
H	1.72683501	0.34885001	-1.01861596
H	0.29424801	-0.39854899	-0.33346301
C	4.13026905	-1.86145496	-0.40524000
H	4.32626677	-2.79620910	-0.93846601
H	4.35517883	-1.04716599	-1.10121405
H	4.82028580	-1.78877699	0.43967199
C	1.40317297	-3.37688804	0.55555499
H	2.08672094	-4.08671999	1.03546202
H	0.60254598	-3.17686892	1.27609897
H	0.97109699	-3.85266590	-0.32474601
F	1.88766599	-2.16165805	-1.83792400
F	2.68458891	-1.44054604	1.83995295
S	2.88990402	-1.14573097	-3.96170402
O	2.99821997	0.24890199	-3.51920104
O	1.77451098	-1.44354296	-4.86683702

1	2	3	
A	A	A	
Frequencies --	-102.8094	32.5263	49.9668
Red. masses --	9.3387	7.3313	4.6765
Frc consts --	0.0582	0.0046	0.0069
IR Inten --	295.6750	2.2848	0.1591

Sum of electronic and zero-point Energies= -1157.865200
 Sum of electronic and thermal Energies= -1157.851041
 Sum of electronic and thermal Enthalpies= -1157.850097
 Sum of electronic and thermal Free Energies= -1157.907687

HF = -1157.98686694

TS: F⁻ transfer from F₂SiMe₃⁻ to allylSiMe₃

Element	X	Y	Z
Si	-3.55786705	-0.05041500	-0.01259000
C	-2.87130690	-1.71875298	0.50628197
H	-2.84271097	-2.40559793	-0.34630299
H	-1.86511397	-1.63120306	0.90975302
H	-3.52301192	-2.16851807	1.26300597
C	-3.20383596	0.55962402	-1.75205696

H	-2.18458891	0.92609102	-1.84848297
H	-3.35621905	-0.24892101	-2.47529697
H	-3.89999604	1.36385703	-2.01413608
C	-3.75439310	1.23818302	1.33635497
H	-2.78732610	1.63759899	1.63679194
H	-4.38573694	2.06288910	0.98839098
H	-4.24583387	0.79919797	2.21143293
F	-5.19817209	-0.48835000	-0.22196600
F	-1.15105796	0.61834502	0.28189400
Si	0.79664600	0.20039500	0.04211400
C	2.77780604	-0.21844600	-0.16625001
C	0.70364702	-0.63945401	1.75672901
H	-0.17305200	-0.32668301	2.32522702
H	0.64587700	-1.72786903	1.62935495
H	1.60181403	-0.44119000	2.35182691
C	0.25114900	-0.80280697	-1.49038398
H	0.30214500	-0.17031699	-2.38493896
H	0.91040897	-1.65903294	-1.66957903
H	-0.77455002	-1.16010797	-1.40541101
C	0.97995299	2.09740305	-0.05338600
H	1.99610496	2.42067790	-0.29476699
H	0.29952699	2.51591802	-0.80087298
H	0.69464397	2.54119110	0.90665603
H	3.30701303	0.30118200	0.64399201
H	2.86150002	-1.29895699	0.01864600
C	3.35461402	0.12530001	-1.48207402
C	4.14080000	1.17214406	-1.78380096
H	4.45557404	1.37850499	-2.80303788
H	4.48757219	1.86351705	-1.01828206
H	3.05212307	-0.51954699	-2.30995488

	1	2	3
	A	A	A
Frequencies --	-97.2073	31.5560	43.0801
Red. masses --	8.4256	2.8026	3.7532
Frc consts --	0.0469	0.0016	0.0041
IR Inten --	258.6122	0.0956	0.7071

Sum of electronic and zero-point Energies= -1135.665281
 Sum of electronic and thermal Energies= -1135.643193
 Sum of electronic and thermal Enthalpies= -1135.642249
 Sum of electronic and thermal Free Energies= -1135.715890

HF = -1135.96154511

TS: F⁻ transfer from F₂SiMe₃⁻ to PhSiMe₃

Element	X	Y	Z
Si	-3.61382389	-0.16883400	0.06591000
C	-2.67241001	-1.73846197	0.47383499
H	-2.65986991	-2.41322708	-0.38836500
H	-1.64445400	-1.52678299	0.75610900
H	-3.16829991	-2.26633906	1.29549205
C	-3.47769094	0.52296901	-1.67221296
H	-2.57541299	1.12108600	-1.78287995
H	-3.45275688	-0.28628999	-2.40968204

H	-4.35088015	1.14440000	-1.89847803
C	-3.80869198	1.08262098	1.44777906
H	-2.89393806	1.66055703	1.57094896
H	-4.63687897	1.76608300	1.23182094
H	-4.03742886	0.57983702	2.39366293
F	-5.19691706	-0.79982901	-0.00065700
F	-1.21404600	0.80555099	0.14246801
Si	0.75219798	0.38859701	-0.04817300
C	2.72118497	-0.02007200	-0.10663700
C	5.01421785	0.59855700	0.55372101
C	4.64105797	-1.49180806	-0.57233500
C	5.52274895	-0.58777899	0.01971100
H	5.68608904	1.31464303	1.02127802
H	5.01870298	-2.42140603	-0.99203998
H	6.58669376	-0.80251801	0.06540600
C	0.64352500	-0.17618300	1.77949095
H	-0.33987501	-0.01790000	2.22345805
H	0.89253700	-1.24218798	1.85067701
H	1.39257801	0.35113499	2.38102889
C	0.23405600	-0.79005897	-1.45779204
H	0.91964501	-0.71637702	-2.30882502
H	0.25142500	-1.83450496	-1.12601805
H	-0.77731699	-0.57603902	-1.79985404
C	0.88949603	2.24504209	-0.48306400
H	1.84484696	2.51734805	-0.93999100
C	3.64471507	0.86391503	0.48713800
H	3.28632593	1.79797399	0.91725200
H	0.08108400	2.52040005	-1.16429496
H	0.75985801	2.85967493	0.41543600
C	3.27229190	-1.20551300	-0.62804300
H	2.62205195	-1.93970597	-1.09678304

	1	2	3
	A	A	A
Frequencies --	-88.7008	30.2314	34.8347
Red. masses --	8.0137	3.6640	4.6599
Frc consts --	0.0371	0.0020	0.0033
IR Inten --	213.6639	0.0130	2.2142

Sum of electronic and zero-point Energies= -1250.012179
 Sum of electronic and thermal Energies= -1249.989102
 Sum of electronic and thermal Enthalpies= -1249.988158
 Sum of electronic and thermal Free Energies= -1250.064735

HF = -1250.32882891

TS: F⁻ transfer from FSO₂⁻ to PhSiMe₃

Element	X	Y	Z
C	-0.60013998	-0.36954600	0.00720400
C	-1.50818598	-0.40142101	1.08185506
C	-1.71569502	-2.40308595	-0.82504803
C	-2.49302912	-1.38715601	1.20998704
H	-1.45443404	0.36232799	1.85385001
C	-2.60380507	-2.39562702	0.25286800
H	-1.78471100	-3.18359089	-1.57919300

H	-3.17331195	-1.36763501	2.05828094
H	-3.36684108	-3.16334605	0.34502301
Si	0.91304898	0.94285601	-0.15049300
C	0.15188700	2.31272507	0.94383001
H	0.30750501	2.09763598	2.00779700
H	0.61470002	3.27860308	0.73806399
H	-0.92828602	2.40066910	0.78984201
C	0.81287998	1.20453203	-2.04300809
H	-0.11682200	0.84354198	-2.48996401
H	0.91527402	2.26958299	-2.26737499
H	1.64666998	0.69986498	-2.54412889
C	2.14984202	-0.32880500	0.57491899
H	3.18023491	0.02894800	0.60045898
H	1.84689403	-0.60072702	1.59333301
H	2.11331105	-1.25462306	-0.01067100
S	4.49133301	2.87962508	-1.71253598
O	5.24909592	3.58685589	-0.67414403
F	2.48578906	2.15337110	-0.27438799
O	4.98429108	1.54396403	-2.06655407
C	-0.74012399	-1.40951896	-0.93342900
H	-0.06343700	-1.45252800	-1.78537703

	1	2	3
	A	A	A
Frequencies --	-66.5330	39.6955	43.0729
Red. masses --	10.1570	8.5604	7.9235
Frc consts --	0.0265	0.0079	0.0087
IR Inten --	268.3546	3.6007	1.7901

Sum of electronic and zero-point Energies= -1289.526077
 Sum of electronic and thermal Energies= -1289.508395
 Sum of electronic and thermal Enthalpies= -1289.507451
 Sum of electronic and thermal Free Energies= -1289.572901

HF = -1289.73602407

TS: SO₂ insertion in allylSiMe₃ (S_{E2'})

Element	X	Y	Z
C	-0.44593200	-0.64998502	0.55826002
Si	1.63514698	-0.25486201	0.15521599
C	1.66282105	1.12338996	1.45388198
H	1.02430403	1.95529699	1.13886404
H	1.27581501	0.76472598	2.41247392
H	2.67232203	1.50613499	1.60839498
C	1.34782004	0.22279599	-1.65604699
H	0.35348099	0.64772499	-1.81660199
H	2.09020209	0.94493300	-1.99973094
H	1.43174303	-0.66468799	-2.29196095
C	2.26080394	-1.97400606	0.62623203
H	2.67387390	-2.46465993	-0.25932801
S	-3.48505497	-2.92207909	0.55173600
H	3.07950807	-1.88095999	1.34414005
H	1.49074399	-2.61936212	1.05190396
S	4.94786978	-0.43309999	-1.77717900
O	6.19888115	-0.42359599	-0.99949902

F	3.61122489	0.15526401	-0.18717000
O	4.40857077	-1.77490997	-2.05752611
H	-0.44441500	-0.94282597	1.60986197
H	-0.85075498	0.35113701	0.40171599
C	-0.93169802	-1.63236594	-0.34246799
H	-1.17924595	-1.29676294	-1.34850502
C	-1.14420795	-2.97071910	-0.07181000
H	-0.81843299	-3.39448094	0.87361801
H	-1.30193496	-3.67460203	-0.88234800
O	-4.14740276	-2.33828998	-0.64719099
O	-3.68627906	-4.39161301	0.69458598

	1	2	3
	A	A	A
Frequencies --	-226.9814	23.3071	26.5453
Red. masses --	11.3253	11.9835	8.8847
Frc consts --	0.3438	0.0038	0.0037
IR Inten --	2716.2545	2.9755	6.8961
Sum of electronic and zero-point Energies=		-1723.866465	
Sum of electronic and thermal Energies=		-1723.845478	
Sum of electronic and thermal Enthalpies=		-1723.844534	
Sum of electronic and thermal Free Energies=		-1723.919877	

HF = -1724.06542740

TS: SO₂ insertion in allylSiMe₃ (S_{E2})

Element	X	Y	Z
C	-0.36277801	-0.52551401	0.59381300
Si	1.82117701	-0.24270099	0.15784299
C	1.84642196	1.18696904	1.40548694
H	2.86688399	1.49508595	1.63973999
H	1.31726205	2.05412507	0.99565500
H	1.34143198	0.91605198	2.33690596
C	1.46097803	0.13376100	-1.66734505
H	0.52062398	0.67407298	-1.80035305
H	2.26394105	0.72446299	-2.11357403
H	1.38071597	-0.80076301	-2.23217607
C	2.25873303	-1.99459100	0.72276098
H	2.85036111	-1.93396103	1.64271700
S	-2.82948089	-0.08113700	1.71247494
O	-3.57588696	0.55282700	0.60066801
O	-2.59364796	0.80381101	2.87901402
H	1.39303195	-2.63123202	0.90581399
H	2.89051104	-2.47218108	-0.03046100
S	5.08544588	-0.70303798	-1.79836702
O	6.33875513	-0.54769999	-1.04753101
F	3.68737292	0.03439400	-0.17221200
O	4.58815289	-2.08128095	-1.91125202
H	-0.26824099	-0.81175703	1.63958204
H	-0.61963600	0.51878202	0.43058199
C	-0.94785702	-1.49958003	-0.31424001
H	-1.16276598	-1.13547301	-1.31812000
C	-1.21768999	-2.79135203	-0.04805200
H	-1.04100895	-3.22071910	0.93519503

H -1.61108398 -3.45807910 -0.80837798

	1	2	3
	A	A	A
Frequencies --	-247.0341	22.0915	26.9192
Red. masses --	10.6069	7.7820	7.4194
Frc consts --	0.3814	0.0022	0.0032
IR Inten --	2949.4755	3.6137	2.9784

Sum of electronic and zero-point Energies= -1723.858783
Sum of electronic and thermal Energies= -1723.837326
Sum of electronic and thermal Enthalpies= -1723.836382
Sum of electronic and thermal Free Energies= -1723.912929

HF = -1724.05669580

TS: SO₂ insertion in PhSiMe₃

Element	X	Y	Z
C	-0.52568001	-0.33981401	0.00310200
C	-1.05511105	-0.70589399	1.25342703
C	-1.99081004	-2.02473593	-1.00549901
C	-2.00485206	-1.71550298	1.38921702
H	-0.72756797	-0.18159799	2.14707208
C	-2.47482896	-2.38230610	0.25512800
H	-2.35780907	-2.53272295	-1.89294696
H	-2.38572311	-1.97875702	2.37204790
H	-3.22075295	-3.16535306	0.35190099
Si	1.71234500	-0.02264300	-0.09442400
C	1.81163204	1.05217898	1.46181095
H	2.45150089	0.54155999	2.18797803
H	2.30280495	1.99863899	1.21970606
H	0.86213702	1.27879500	1.94408906
C	1.67990804	0.72849101	-1.83392298
H	0.69479102	0.91004801	-2.26234508
H	2.22689295	1.67517805	-1.82583106
H	2.22543693	0.05479400	-2.50091100
C	1.89044201	-1.90181506	0.09190900
H	2.93003893	-2.18489289	0.26509100
S	-0.87594700	2.07133198	-0.20513400
O	-1.44246900	2.37157798	1.13683605
O	-1.88370502	2.07876301	-1.29869294
H	1.27812898	-2.28052998	0.91317600
H	1.54526603	-2.39956093	-0.81905800
S	5.24794197	-0.86956197	-1.39057302
O	5.63537121	-1.87872398	-0.39543200
F	3.62304497	0.14952300	-0.18670200
O	4.43024302	-1.37093902	-2.50383711
C	-1.04032302	-1.01316297	-1.12133002
H	-0.68869102	-0.74902701	-2.11456990

	1	2	3
	A	A	A
Frequencies --	-223.9299	23.1763	29.8522
Red. masses --	7.9496	9.9111	12.6123

Frc consts -- 0.2349 0.0031 0.0066
 IR Inten -- 990.9600 7.7003 0.6263

Sum of electronic and zero-point Energies= -1838.202398
 Sum of electronic and thermal Energies= -1838.180390
 Sum of electronic and thermal Enthalpies= -1838.179445
 Sum of electronic and thermal Free Energies= -1838.256114

HF = -1838.42109766

TS: SO₂ insertion in PhSiMe₃F⁻

Element	X	Y	Z
C	0.40062201	-0.54039299	-0.02594100
C	1.06648803	-0.77102900	-1.24608004
C	2.21353292	-1.80059099	1.05951202
C	2.25515890	-1.49468303	-1.33137095
H	0.64989197	-0.36620700	-2.16530800
C	2.83762193	-2.01256204	-0.17263301
H	2.65857601	-2.19722605	1.96845198
H	2.73283505	-1.64928401	-2.29543805
H	3.76905203	-2.56833506	-0.22774599
Si	-1.77301097	-0.29070899	-0.00711000
C	-1.81496203	0.85213500	-1.53376198
H	-2.41741896	0.37343600	-2.31256795
H	-2.33760500	1.77700198	-1.26874197
H	-0.85437602	1.12521505	-1.96997905
C	-1.81161797	0.44428700	1.75558496
H	-0.85007203	0.71715301	2.19270897
H	-2.44969702	1.33310199	1.75469804
H	-2.28985000	-0.27564600	2.42904902
F	-3.60873699	-0.21797501	0.00497800
C	-1.84267604	-2.18709111	-0.24970400
H	-2.87208390	-2.55000591	-0.29903799
S	0.77217400	2.07822299	0.23774099
O	1.33625400	2.46218300	-1.07360804
O	1.73850095	2.11998296	1.35588706
H	-1.32456398	-2.47480297	-1.17007303
H	-1.32787395	-2.70076108	0.56826800
C	1.02583301	-1.07438600	1.12003899
H	0.57222801	-0.92117202	2.09630489

	1	2	3
	A	A	A
Frequencies --	-105.2075	23.6674	57.0629
Red. masses --	5.6578	9.3110	2.1559
Frc consts --	0.0369	0.0031	0.0041
IR Inten --	227.8497	3.5019	0.2480

Sum of electronic and zero-point Energies= -1289.522418
 Sum of electronic and thermal Energies= -1289.504951
 Sum of electronic and thermal Enthalpies= -1289.504007
 Sum of electronic and thermal Free Energies= -1289.568327

HF = -1289.73234055

2.2.3. Reductive elimination

allyl–Pd(PMe₃)₂–Ph

Element	X	Y	Z
Pd	-1.45955396	-0.11114700	0.14976101
C	0.60236299	-0.21547499	0.24187499
C	1.29027200	-0.14600100	1.46661997
C	1.38933206	-0.35128900	-0.91477603
C	2.68389297	-0.22495200	1.53848100
H	0.73089302	-0.03589500	2.39292502
C	2.78440499	-0.43302500	-0.85505199
H	0.91134799	-0.38755500	-1.89010298
C	3.44127893	-0.37328500	0.37481999
H	3.17906308	-0.17180000	2.50514793
H	3.35854793	-0.54004401	-1.77232695
H	4.52431011	-0.43631801	0.42559499
C	-1.12036896	1.97732699	-0.03581300
H	-1.82923198	2.51419902	0.60979301
H	-0.13303401	2.22065902	0.37334201
C	-1.20149100	2.55110407	-1.41883695
H	-1.08877397	3.63945293	-1.48319304
C	-1.38253701	1.89829397	-2.57028008
H	-1.41330004	2.41996193	-3.52344990
H	-1.49485004	0.81763798	-2.59527802
P	-1.43765402	-2.50011396	0.25686201
P	-3.80464792	0.22584800	-0.08088100
C	-4.49445009	-0.46073300	-1.64714706
H	-4.27740479	-1.52742004	-1.72195196
H	-5.57583284	-0.31114101	-1.70368695
H	-4.02009201	0.04186500	-2.49185705
C	-2.94752002	-3.52241397	0.55078298
H	-3.72476006	-3.28294706	-0.17644501
H	-2.71035194	-4.58637524	0.46800101
H	-3.33559990	-3.33055305	1.55253804
C	-0.82413799	-3.20684505	-1.32888496
H	0.15615700	-2.78402805	-1.55120897
H	-0.74469101	-4.29562283	-1.27155101
H	-1.50663805	-2.93919897	-2.13800097
C	-0.28462401	-3.24075198	1.48518097
H	-0.26663399	-4.32961082	1.39121401
H	0.71693301	-2.84155607	1.32895005
H	-0.60351199	-2.97452998	2.49497700
C	-4.92357302	-0.48774499	1.20037103
H	-5.96782398	-0.25020400	0.98051399
H	-4.81139612	-1.56976295	1.24934804
H	-4.66429186	-0.07311400	2.17648792
C	-4.42535496	1.95732105	-0.14340000
H	-3.90055490	2.50480700	-0.92626703
H	-5.49926805	1.97070897	-0.34549600
H	-4.23939705	2.45306396	0.81088102

	1	2	3
	A	A	A
Frequencies --	23.1467	29.8216	35.8658
Red. masses --	4.1193	3.3640	3.3807

Frc consts --	0.0013	0.0018	0.0026
IR Inten --	0.0900	0.3881	0.6840

Sum of electronic and zero-point Energies= -1398.971002
 Sum of electronic and thermal Energies= -1398.944858
 Sum of electronic and thermal Enthalpies= -1398.943914
 Sum of electronic and thermal Free Energies= -1399.027675

HF = -1399.35969134

TS: reductive elimination from allyl–Pd(PMe₃)₂–Ph to allylPh

Element	X	Y	Z
C	-3.96363711	0.13088199	-0.44205701
C	-2.62624907	-0.09306700	-0.74511403
C	-1.61816597	0.84085101	-0.39019099
C	-2.03696704	1.97363102	0.34824401
C	-3.37875295	2.19208407	0.65042001
C	-4.36028814	1.27945495	0.25471199
H	-4.70914698	-0.59580600	-0.75572997
H	-2.34902692	-0.99685401	-1.27999699
H	-1.29594803	2.69567609	0.67962098
H	-3.66132498	3.08630490	1.20056403
H	-5.40512705	1.44920397	0.49362200
Pd	-0.00850500	1.11903203	-1.68959999
P	2.28312993	1.27150702	-2.41652107
P	-1.32984197	1.89987600	-3.49814010
C	2.89504409	2.94544697	-2.90211511
H	3.93895888	2.91449404	-3.22775102
H	2.28033400	3.33512092	-3.71568608
H	2.80701399	3.62812805	-2.05449700
C	2.73417497	0.28049499	-3.90956998
H	2.10827994	0.58269203	-4.75169992
H	3.78538799	0.41490301	-4.18093491
H	2.55002809	-0.77815902	-3.71527100
C	3.59917498	0.74092501	-1.23120499
H	4.59917593	0.83190501	-1.66521001
H	3.54664803	1.35415995	-0.32885599
H	3.43340206	-0.29912201	-0.94057697
C	-2.15764403	0.53793401	-4.43028879
H	-2.83671093	0.92791498	-5.19426107
H	-1.40507603	-0.09191300	-4.90897989
H	-2.72529697	-0.07655200	-3.72880507
C	-0.61241901	2.90028811	-4.87689590
H	0.19359900	2.34156990	-5.35741806
H	-1.36614799	3.15032506	-5.62955379
H	-0.19325501	3.82584405	-4.47647190
C	-2.77740502	2.92957091	-3.00589800
H	-3.35673094	2.39165401	-2.25436902
H	-2.42855692	3.86175609	-2.55708599
H	-3.41346312	3.16146708	-3.86505294
C	0.02691400	0.11056200	0.42865500
H	0.80830401	-0.44248801	-0.10515600
H	0.47690600	0.93577302	0.98133099
C	-0.66166300	-0.83915699	1.33540106
C	-0.89996099	-0.63836002	2.63541102
H	-0.99696702	-1.76860797	0.87805599
H	-1.40994596	-1.38082504	3.24102592

H	-0.59876299	0.27920201	3.13457894
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	1	2	3
	A	A	A
Frequencies --	-401.2722	22.3188	28.9055
Red. masses --	5.2527	3.4058	3.3999
Frc consts --	0.4983	0.0010	0.0017
IR Inten --	158.9664	0.4133	0.0567

Sum of electronic and zero-point Energies= -1398.933307
 Sum of electronic and thermal Energies= -1398.907108
 Sum of electronic and thermal Enthalpies= -1398.906164
 Sum of electronic and thermal Free Energies= -1398.990701

HF = -1399.31988685

allylSO₂–Pd(PMe₃)₂–Ph

Element	X	Y	Z
Pd	-1.89008498	-1.23164201	0.06765000
C	0.14401700	-1.07490504	0.31515500
C	0.71887702	-1.18243802	1.58749294
C	0.98483801	-0.79597002	-0.76952398
C	2.09258008	-1.00197995	1.77410495
H	0.09474900	-1.39365995	2.45142293
C	2.35807610	-0.61396098	-0.58859003
H	0.57153797	-0.70775902	-1.77078295
C	2.91773200	-0.71515203	0.68590403
H	2.51596498	-1.08160806	2.77150011
H	2.98987508	-0.38986099	-1.44367194
H	3.98416495	-0.57167101	0.82929701
P	-1.49775100	-3.49136305	-0.38137099
P	-4.30147791	-1.24836004	-0.28351101
C	-5.28120422	-2.80895209	-0.23998800
H	-5.27986479	-3.21841311	0.77141398
H	-6.31512308	-2.59787107	-0.52358598
H	-4.88390589	-3.55856991	-0.92351198
C	-2.31830096	-4.65179920	0.78154600
H	-3.39557600	-4.50269222	0.78779000
H	-2.10151696	-5.68478823	0.49898201
H	-1.93935394	-4.47083712	1.78879595
C	-2.05499506	-4.02132988	-2.05052495
H	-1.49504006	-3.46379089	-2.80324197
H	-1.87580502	-5.08989620	-2.19163704
H	-3.11486912	-3.81603909	-2.19456601
C	0.23094000	-4.10271692	-0.35194400
H	0.24287000	-5.17399216	-0.56673700
H	0.82812297	-3.57284093	-1.09297097
H	0.67122602	-3.91971493	0.62783903
C	-5.28753710	-0.20419399	0.86208302
H	-6.34608221	-0.25393501	0.59398001
H	-5.15774012	-0.55807102	1.88586998
H	-4.94121408	0.82522702	0.80386698
C	-4.69796515	-0.55527300	-1.93647599
H	-4.25412893	-1.17422795	-2.71843505
H	-5.77888584	-0.50214100	-2.08955193

H	-4.26779890	0.44504899	-1.98698401
S	-2.04408908	1.10745800	0.58699399
C	-0.41709200	2.01660490	0.45486799
H	-0.08081100	1.87891304	-0.57280600
H	0.25363499	1.48106098	1.12689900
C	-0.58400100	3.44544005	0.83171499
H	-0.84149200	3.63575912	1.87043798
C	-0.45345101	4.46520376	-0.01669900
H	-0.20629101	4.30940199	-1.06337595
H	-0.58513701	5.49300909	0.30640700
O	-2.94285607	1.81746197	-0.40003899
O	-2.44804692	1.29139698	2.03141689

	1	2	3
	A	A	A
Frequencies --	26.3423	38.1403	44.3951
Red. masses --	4.4319	3.8383	3.1741
Frc consts --	0.0018	0.0033	0.0037
IR Inten --	0.6694	4.8445	0.6704
Sum of electronic and zero-point Energies=		-1947.687052	
Sum of electronic and thermal Energies=		-1947.657824	
Sum of electronic and thermal Enthalpies=		-1947.656880	
Sum of electronic and thermal Free Energies=		-1947.746606	

HF = -1948.08849329

TS: reductive elimination from allylSO₂–Pd(PMe₃)₂–Ph to allylSO₂Ph

Element	X	Y	Z
C	-3.77048492	0.37590900	-0.18020400
C	-2.42296696	0.07101200	-0.33419499
C	-1.47251701	1.10915399	-0.40086901
C	-1.88308597	2.43926001	-0.18402401
C	-3.23670697	2.72966194	-0.03458600
C	-4.18568277	1.70446599	-0.03647800
H	-4.50226212	-0.42599499	-0.17096400
H	-2.09739709	-0.95894301	-0.43136999
H	-1.14192998	3.23033500	-0.14102399
H	-3.55038500	3.76063490	0.09601700
H	-5.23747396	1.93489397	0.09374800
Pd	0.00391200	1.03590298	-1.98599303
P	2.34429193	1.39709198	-2.37310195
P	-1.53995299	0.83306003	-3.75586390
C	3.51800799	0.22674100	-1.56685197
H	4.55686903	0.51083797	-1.75899506
H	3.32968211	0.22149099	-0.49317199
H	3.34545588	-0.78292900	-1.94449401
C	2.97262001	3.02493596	-1.77807701
H	2.76208210	3.10093999	-0.71096998
H	4.04803085	3.12628698	-1.95189095
H	2.44962406	3.83284307	-2.29406595
C	2.94034195	1.37640202	-4.12093782
H	4.01389122	1.57885301	-4.18040895
H	2.73703790	0.39965001	-4.56463718
H	2.40273499	2.13006806	-4.69977283

C	-2.39586401	-0.79598302	-3.89270806
H	-3.11677289	-0.79923201	-4.71541977
H	-1.65976202	-1.58483303	-4.05901194
H	-2.92024708	-1.00757396	-2.95932102
C	-0.90783298	1.06047595	-5.47540522
H	-0.11627100	0.33431199	-5.67106724
H	-1.70155203	0.93335599	-6.21768904
H	-0.48195601	2.06078601	-5.57664585
C	-2.97196603	1.99462795	-3.72100592
H	-3.51777411	1.86293304	-2.78609395
H	-2.61031294	3.02397895	-3.75928092
H	-3.64776111	1.82060003	-4.56348801
S	0.26318201	0.67935598	0.64525402
O	1.20711303	1.81807005	0.86716598
O	0.80174500	-0.69571102	0.42579499
C	-0.63241100	0.55689502	2.28795695
H	-1.18384004	1.49440205	2.37120891
H	-1.32908404	-0.27399799	2.18352890
C	0.34850001	0.36398000	3.38944101
C	0.50162601	-0.78122401	4.05444908
H	0.97405899	1.21970201	3.62776995
H	1.23273695	-0.88098902	4.85007286
H	-0.10129000	-1.65689695	3.83087397

	1	2	3
	A	A	A
Frequencies --	-242.4457	20.5945	23.0018
Red. masses --	8.9700	4.5777	3.1356
Frc consts --	0.3107	0.0011	0.0010
IR Inten --	242.4618	0.0639	1.3092

Sum of electronic and zero-point Energies= -1947.625106
 Sum of electronic and thermal Energies= -1947.595348
 Sum of electronic and thermal Enthalpies= -1947.594404
 Sum of electronic and thermal Free Energies= -1947.687441

HF = -1948.02304827

allylSO₂–Pd(PMe₃)₂–(C₆H₄)CF₃

Element	X	Y	Z
Pd	-1.88749802	-1.30296195	-0.00416400
C	0.15903600	-1.21579504	0.11414100
C	0.81580198	-1.40520406	1.33765697
C	0.93787301	-0.91234499	-1.01104605
C	2.20062208	-1.28550804	1.44147599
H	0.24609400	-1.63381100	2.23307395
C	2.32228589	-0.78862602	-0.92157501
H	0.46503201	-0.75745201	-1.97617495
C	2.95537090	-0.97199398	0.30950499
H	2.68823695	-1.42515302	2.40003109
H	2.90486097	-0.54231697	-1.80258703
P	-1.60563099	-3.55368805	-0.57664901
P	-4.31255388	-1.23794401	-0.20353800
C	-5.33396292	-2.77122211	-0.17111699
H	-5.28306723	-3.22845006	0.81826800

H	-6.37596893	-2.51542807	-0.37759200
H	-5.00620985	-3.49841189	-0.91291797
C	-2.37040091	-4.72750711	0.60938799
H	-3.43866301	-4.54713202	0.70493102
H	-2.20891094	-5.75642920	0.27972800
H	-1.90838003	-4.59045696	1.58845603
C	-2.29852605	-4.00081205	-2.21851301
H	-1.76081896	-3.44687009	-2.99001908
H	-2.18208194	-5.07167816	-2.40222597
H	-3.35359406	-3.74033499	-2.28655696
C	0.10145900	-4.21239901	-0.70174402
H	0.06446700	-5.27210712	-0.96514601
H	0.65897399	-3.66744208	-1.46246696
H	0.61843902	-4.09065723	0.24975599
C	-5.19120884	-0.22536400	1.05189705
H	-6.26688623	-0.24742900	0.85969901
H	-4.99468613	-0.62207800	2.04886007
H	-4.83029699	0.79968399	1.00778699
C	-4.78260517	-0.45141801	-1.79339302
H	-4.41139793	-1.04606903	-2.62999105
H	-5.86798620	-0.35311100	-1.87539101
H	-4.31900597	0.53464401	-1.82253098
S	-1.94403505	1.01093996	0.63508099
C	-0.30386400	1.88390803	0.43754500
H	-0.03225300	1.77623999	-0.61268097
H	0.39097500	1.31028104	1.05126500
C	-0.41187099	3.30212903	0.87242299
H	-0.59555900	3.46201205	1.93166399
C	-0.31530401	4.34653711	0.05009400
H	-0.14185500	4.22086287	-1.01523900
H	-0.40164000	5.36532307	0.41418499
O	-2.88381004	1.78857100	-0.25600401
O	-2.24580193	1.12976599	2.10966706
C	4.44660616	-0.89537901	0.40128499
F	5.04387999	-2.10327196	0.16831900
F	4.87837887	-0.49954501	1.62426698
F	4.98474216	-0.03770000	-0.50149000

	1	2	3
	A	A	A
Frequencies --	14.1901	21.2633	24.2111
Red. masses --	17.1081	4.4185	8.0479
Frc consts --	0.0020	0.0012	0.0028
IR Inten --	0.5034	0.1747	1.6999

Sum of electronic and zero-point Energies= -2284.842877
 Sum of electronic and thermal Energies= -2284.809878
 Sum of electronic and thermal Enthalpies= -2284.808934
 Sum of electronic and thermal Free Energies= -2284.909110

HF = -2285.24844083

TS: reductive elimination from allylSO₂–Pd(PMe₃)₂–(C₆H₄)CF₃ to allylSO₂(C₆H₄)CF₃

Element	X	Y	Z
C	-3.75109696	0.04012200	-0.41251299

C	-2.38201594	-0.12954400	-0.51991397
C	-1.52335501	0.99583101	-0.58015102
C	-2.07974505	2.28220296	-0.37683299
C	-3.45241189	2.44293308	-0.27152199
C	-4.29739904	1.32731795	-0.30357200
H	-4.40171003	-0.82735300	-0.41921899
H	-1.96022499	-1.12540197	-0.59475303
H	-1.42607296	3.14551806	-0.32056999
H	-3.86942101	3.43766809	-0.16363899
Pd	-0.00001100	1.02630603	-2.08119893
P	2.36915398	1.40946400	-2.39628911
P	-1.47832596	0.77086502	-3.88145208
C	3.50274396	0.23059900	-1.54767096
H	4.54899979	0.52017999	-1.68147802
H	3.25863504	0.20495901	-0.48549700
H	3.35349011	-0.77299500	-1.95081902
C	2.96610498	3.03534508	-1.76769197
H	2.70746803	3.11013007	-0.71093702
H	4.04781103	3.13895202	-1.89267600
H	2.46534991	3.84363198	-2.30464602
C	3.02769709	1.39486694	-4.12012386
H	4.10200214	1.60008204	-4.13802481
H	2.84431195	0.41855201	-4.57320118
H	2.51029110	2.14885712	-4.71645021
C	-2.21537495	-0.90743297	-4.07064819
H	-2.91373897	-0.93978399	-4.91153479
H	-1.42250001	-1.63945699	-4.23474789
H	-2.74766207	-1.17538202	-3.15655589
C	-0.77463400	1.06922400	-5.55956507
H	0.07339800	0.40138900	-5.72242594
H	-1.51959205	0.90008700	-6.34250689
H	-0.41452599	2.09774494	-5.62516594
C	-2.97657609	1.84243298	-3.88898897
H	-3.57384396	1.63719106	-3.00025892
H	-2.67914295	2.89224505	-3.86303401
H	-3.58431101	1.66332304	-4.78033209
S	0.13994800	0.73148298	0.61908901
O	0.98731399	1.94525504	0.80907899
O	0.79215002	-0.59724599	0.45286500
C	-0.82832098	0.60048097	2.21442890
H	-1.44448400	1.50021505	2.23787808
H	-1.46106398	-0.27922699	2.10246110
C	0.10992100	0.50791502	3.36556005
C	0.32548901	-0.61245102	4.05489588
H	0.64822000	1.41669405	3.61966491
H	1.02168202	-0.63989800	4.88663006
H	-0.19009900	-1.53847301	3.81673694
C	-5.77027321	1.49471200	-0.13436501
F	-6.48218679	0.55831802	-0.81377703
F	-6.16903591	1.38525605	1.17002797
F	-6.21211910	2.70599890	-0.55496401

	1	2	3
	A	A	A
Frequencies --	-221.9987	13.3125	17.3272
Red. masses --	10.2004	5.1141	17.1033
Frc consts --	0.2962	0.0005	0.0030

IR Inten -- 218.0250 0.1058 0.0405

Sum of electronic and zero-point Energies= -2284.780441
Sum of electronic and thermal Energies= -2284.746897
Sum of electronic and thermal Enthalpies= -2284.745953
Sum of electronic and thermal Free Energies= -2284.849389

HF = -2285.18241921

allylSO₂-Pd(PMe₃)₂-(C₆H₄)NMe₂

Element	X	Y	Z
Pd	-1.92783499	-1.30543900	0.06100000
C	0.11074800	-1.22501898	0.30954400
C	0.69212198	-1.33913302	1.57468796
C	0.97275800	-0.96758002	-0.75946301
C	2.06553602	-1.18915904	1.77429295
H	0.07042500	-1.52577806	2.44651294
C	2.34773493	-0.81397802	-0.58219898
H	0.57719201	-0.86248499	-1.76675701
C	2.93728900	-0.93088299	0.69491899
H	2.44707990	-1.27158201	2.78407502
H	2.95495605	-0.59867001	-1.45232499
P	-1.61020803	-3.56889892	-0.40734801
P	-4.33761311	-1.22485304	-0.29101101
C	-5.37700176	-2.74713302	-0.26533699
H	-5.38737106	-3.17076993	0.74023002
H	-6.40326691	-2.49415588	-0.54247898
H	-5.00968695	-3.50147891	-0.96058702
C	-2.47558808	-4.71065712	0.74246001
H	-3.54752493	-4.52612305	0.74305099
H	-2.29113412	-5.74854898	0.45486799
H	-2.09745598	-4.54854918	1.75327504
C	-2.18216491	-4.06519079	-2.08245802
H	-1.59120500	-3.53183699	-2.82904506
H	-2.05450797	-5.14052820	-2.22858000
H	-3.22983408	-3.80726910	-2.23156500
C	0.09531900	-4.24190807	-0.37801200
H	0.07155100	-5.31070089	-0.60454899
H	0.71405900	-3.72377992	-1.10953903
H	0.53661299	-4.08322906	0.60554498
C	-5.28662491	-0.15183800	0.86012602
H	-6.34578991	-0.15813100	0.58961302
H	-5.17327213	-0.51650798	1.88213301
H	-4.89950514	0.86347300	0.80816299
C	-4.70433092	-0.50028998	-1.93761897
H	-4.27917480	-1.12554204	-2.72495294
H	-5.78203917	-0.40664500	-2.09432697
H	-4.23844290	0.48457599	-1.97463596
S	-1.98411703	1.03205705	0.60731101
C	-0.32397601	1.87774003	0.45705801
H	-0.00662400	1.72810698	-0.57493198
H	0.33384600	1.31380999	1.11859298
C	-0.42967701	3.31163406	0.83643597
H	-0.66823399	3.51139402	1.87800300
C	-0.26770699	4.32686377	-0.01226700
H	-0.03828700	4.16277409	-1.06175196

H	-0.35520399	5.35869789	0.31310999
O	-2.87012291	1.79033899	-0.35634100
O	-2.35805011	1.21662796	2.06079602
N	4.31998920	-0.81587797	0.87817103
C	4.81406498	-0.59966099	2.23150492
C	5.10058880	-0.19971099	-0.18648399
H	6.14868116	-0.17116100	0.10841300
H	5.04018021	-0.78465998	-1.10672402
H	4.77559280	0.82749498	-0.41149399
H	5.90058279	-0.52681601	2.20853806
H	4.41434717	0.31832200	2.68852401
H	4.56072187	-1.44055498	2.88053012

	1	2	3
	A	A	A
Frequencies --	22.8449	25.8910	38.0057
Red. masses --	4.0059	4.5356	3.0980
Frc consts --	0.0012	0.0018	0.0026
IR Inten --	0.6084	2.1642	2.3181

Sum of electronic and zero-point Energies= -2081.619571
 Sum of electronic and thermal Energies= -2081.586064
 Sum of electronic and thermal Enthalpies= -2081.585120
 Sum of electronic and thermal Free Energies= -2081.684641

HF = -2082.09401290

TS: reductive elimination from allylSO₂–Pd(PMe₃)₂–(C₆H₄)NMe₂ to allylSO₂(C₆H₄)NMe₂

Element	X	Y	Z
C	-3.68787098	0.33041799	0.01451300
C	-2.33437109	0.08574700	-0.14844500
C	-1.41303504	1.14432800	-0.21378700
C	-1.87864494	2.44815111	0.02310100
C	-3.23162293	2.70149803	0.18879300
C	-4.18426085	1.65204096	0.16055299
H	-4.36308002	-0.51490402	0.02436700
H	-1.98444104	-0.93615901	-0.25093001
H	-1.17211103	3.27062988	0.06839800
H	-3.54521704	3.72538400	0.34223300
Pd	0.03953700	1.07653296	-1.84620500
P	2.34091210	1.27016497	-2.40905309
P	-1.53830099	1.00326204	-3.62848401
C	3.47720003	-0.00134900	-1.70800197
H	4.52098083	0.22034000	-1.94960105
H	3.34471798	-0.03516600	-0.62665701
H	3.21748710	-0.98341602	-2.10827589
C	3.13254690	2.83327198	-1.83433104
H	2.98831391	2.90855789	-0.75617701
H	4.20065594	2.85330009	-2.07082891
H	2.64540505	3.68796897	-2.30857301
C	2.81573510	1.24420500	-4.19409084
H	3.89478707	1.36945605	-4.32463408
H	2.51149607	0.29497299	-4.63993216
H	2.29719210	2.04736495	-4.72147799
C	-2.45662999	-0.59166598	-3.78462911

H	-3.19300389	-0.55246198	-4.59290218
H	-1.75373495	-1.40378594	-3.98077989
H	-2.96899891	-0.80234897	-2.84421706
C	-0.95380300	1.25359702	-5.36326790
H	-0.19463000	0.50633502	-5.60303783
H	-1.77388000	1.17350805	-6.08334017
H	-0.49575099	2.24074888	-5.45434284
C	-2.93761396	2.20480394	-3.53492808
H	-3.46351409	2.06869507	-2.58888888
H	-2.54881310	3.22482800	-3.56000590
H	-3.63907194	2.06881404	-4.36367607
S	0.39303100	0.76767099	0.70635402
O	1.32600403	1.92648995	0.90962303
O	0.93694299	-0.61804003	0.53144801
C	-0.42736101	0.66238397	2.39716697
H	-0.95989001	1.60863304	2.50280094
H	-1.14172804	-0.15775301	2.33087993
C	0.59195399	0.45948201	3.46040797
C	0.73799902	-0.67412901	4.14721680
H	1.25423098	1.29936004	3.65240598
H	1.49971795	-0.77997899	4.91301298
H	0.09933700	-1.53519595	3.97027206
N	-5.53545523	1.90260196	0.27766699
C	-5.99056005	3.24224210	0.62306499
C	-6.45825911	0.79298401	0.47530699
H	-5.62884903	3.56529093	1.60848701
H	-7.07841206	3.25908589	0.63655901
H	-5.66042614	3.97322798	-0.11964000
H	-6.42178202	0.09455600	-0.36497599
H	-7.47411299	1.17793000	0.53570098
H	-6.24684477	0.23418801	1.39667797

	1	2	3
	A	A	A
Frequencies --	-225.5983	14.4861	24.0754
Red. masses --	9.5127	3.8853	3.3030
Frc consts --	0.2853	0.0005	0.0011
IR Inten --	266.6476	0.0282	1.7597

Sum of electronic and zero-point Energies= -2081.561591
 Sum of electronic and thermal Energies= -2081.527379
 Sum of electronic and thermal Enthalpies= -2081.526435
 Sum of electronic and thermal Free Energies= -2081.630196

HF = -2082.03252078

allylS-Pd(PMe₃)₂-Ph

Element	X	Y	Z
Pd	-1.88349700	-1.30377197	-0.01281000
C	0.15956300	-1.19686604	0.10774300
C	0.82122397	-1.36984599	1.33168995
C	0.94325000	-0.93337399	-1.02426898
C	2.21282601	-1.27726698	1.42469597
H	0.24935600	-1.57398605	2.23301196
C	2.33496094	-0.84052497	-0.93914902

H	0.46788800	-0.79102898	-1.99129105
C	2.97786808	-1.01240206	0.28773299
H	2.69886708	-1.41133904	2.38759494
H	2.91714597	-0.63097900	-1.83277297
P	-1.59569001	-3.57078695	-0.48183301
P	-4.29908800	-1.22767401	-0.20797300
C	-5.35962200	-2.73623300	-0.23672201
H	-5.32318783	-3.23273993	0.73413700
H	-6.39611578	-2.45695901	-0.44154701
H	-5.03358984	-3.43935609	-1.00323296
C	-2.44620609	-4.74451780	0.65134901
H	-3.51999998	-4.56946278	0.65747797
H	-2.25555992	-5.77620697	0.34516001
H	-2.06790090	-4.59821796	1.66460800
C	-2.18845701	-4.06222582	-2.15370798
H	-1.61361599	-3.51862502	-2.90556598
H	-2.05808711	-5.13580513	-2.31174302
H	-3.24017406	-3.80834007	-2.28301811
C	0.10908200	-4.25272894	-0.48353800
H	0.08251700	-5.31792021	-0.72603297
H	0.72046697	-3.72381210	-1.21388900
H	0.56326401	-4.11093283	0.49699301
C	-5.15808916	-0.24707900	1.08970594
H	-6.22459984	-0.16467801	0.86524302
H	-5.03234720	-0.73797703	2.05641794
H	-4.72140980	0.74926603	1.15231800
C	-4.79218817	-0.37891600	-1.76299906
H	-4.44814110	-0.95716500	-2.62265396
H	-5.87695122	-0.25783300	-1.82148004
H	-4.31231594	0.59990400	-1.79311597
S	-2.02235508	1.05865204	0.41040900
C	-0.34469199	1.85819101	0.37774000
H	0.07861000	1.78357899	-0.62471497
H	0.30892500	1.30780697	1.05595601
C	-0.46877900	3.28383088	0.80850899
H	-0.77678698	3.44135499	1.84162498
C	-0.24590100	4.34010887	0.02470100
H	0.05139900	4.22438908	-1.01480603
H	-0.35091600	5.35674095	0.39121300
H	4.05894709	-0.93924201	0.35713199

	1	2	3
	A	A	A
Frequencies --	23.1098	29.5151	41.9375
Red. masses --	4.3265	3.5932	2.9755
Frc consts --	0.0014	0.0018	0.0031
IR Inten --	0.2592	1.6804	3.4153

Sum of electronic and zero-point Energies= -1797.234204
 Sum of electronic and thermal Energies= -1797.206488
 Sum of electronic and thermal Enthalpies= -1797.205544
 Sum of electronic and thermal Free Energies= -1797.293233

HF = -1797.62688980

TS: reductive elimination from allylS–Pd(PMe₃)₂–Ph to allylSPh

Element	X	Y	Z
C	-4.11248589	1.08916795	0.15093499
C	-2.84121203	0.54348898	0.01392900
C	-1.68157899	1.35126102	0.15786199
C	-1.88219905	2.71406293	0.51861101
C	-3.16027308	3.24590993	0.62747997
C	-4.29442596	2.44422698	0.44854701
H	-4.97711277	0.44753399	0.00139000
H	-2.74300909	-0.49794000	-0.26550400
H	-1.02142596	3.34446001	0.72145301
H	-3.27394009	4.29775381	0.87626803
H	-5.29061508	2.86115408	0.54976702
Pd	-0.08765500	1.34940195	-1.33094203
P	2.19907188	1.69460905	-2.02231097
P	-1.48891103	1.40262699	-3.23613000
C	2.66995597	3.06778002	-3.16641498
H	3.74329710	3.06748009	-3.37946105
H	2.12310290	2.96258211	-4.10546494
H	2.39711308	4.02666187	-2.72083998
C	2.92009902	0.23800500	-2.90175796
H	2.36239195	0.05308900	-3.82217503
H	3.97415900	0.39557901	-3.14891601
H	2.83136201	-0.64703900	-2.26833105
C	3.42119002	1.92505801	-0.65719903
H	4.45143795	1.92299700	-1.02523696
H	3.22608089	2.87242007	-0.14978300
H	3.29780006	1.12180805	0.07243200
C	-2.46094608	-0.12950200	-3.58226991
H	-3.13926196	0.01008800	-4.42930603
H	-1.78059304	-0.95478100	-3.80185509
H	-3.04604101	-0.38980800	-2.69890189
C	-0.77106100	1.75160205	-4.90217304
H	-0.01658100	0.99962699	-5.14340401
H	-1.53807497	1.74803603	-5.68240213
H	-0.28491199	2.72921205	-4.88942099
C	-2.82820201	2.66569996	-3.13738489
H	-3.40029407	2.50731492	-2.22237706
H	-2.38554406	3.66267610	-3.09163404
H	-3.49642897	2.60962892	-4.00181723
S	0.01486500	0.56225002	0.97308302
C	-0.52434802	-1.20897603	1.05344796
H	-1.49001801	-1.23294306	1.56833196
H	-0.65347201	-1.59140003	0.04133800
C	0.50576502	-1.99390399	1.80232298
C	1.25096297	-2.95937896	1.26516402
H	0.63331598	-1.73570502	2.85211706
H	1.98169100	-3.50661302	1.85193300
H	1.15491998	-3.23896909	0.21921501

1	2	3	
A	A	A	
Frequencies --	-234.9236	5.9195	18.0369
Red. masses --	9.3538	3.0535	4.5239
Frc consts --	0.3042	0.0001	0.0009
IR Inten --	254.3414	0.0624	2.4693

Sum of electronic and zero-point Energies= -1797.189361
 Sum of electronic and thermal Energies= -1797.161479
 Sum of electronic and thermal Enthalpies= -1797.160535
 Sum of electronic and thermal Free Energies= -1797.251168

HF = -1797.57882877

allylS-Pd(PMe₃)₂-(C₆H₄)CF₃

Element	X	Y	Z
Pd	-1.88598204	-1.29953206	-0.03005500
C	0.15171300	-1.17875898	0.08615500
C	0.81773800	-1.36504996	1.30731106
C	0.93463302	-0.88863099	-1.04154301
C	2.20361900	-1.26199496	1.40552902
H	0.25078201	-1.58613205	2.20654202
C	2.32107210	-0.78339303	-0.95989197
H	0.46015701	-0.73164201	-2.00560188
C	2.95957899	-0.96716499	0.26870099
H	2.69320798	-1.40274704	2.36326694
H	2.90234995	-0.55076599	-1.84579206
P	-1.59449100	-3.56011009	-0.53855699
P	-4.30073977	-1.23582602	-0.19844700
C	-5.34816313	-2.75245404	-0.22561000
H	-5.29298878	-3.25750399	0.73988497
H	-6.38952112	-2.47824407	-0.41114900
H	-5.03005123	-3.44547796	-1.00430000
C	-2.41776705	-4.75579786	0.59024698
H	-3.49242496	-4.58883905	0.61694300
H	-2.22493100	-5.78083086	0.26405400
H	-2.02326608	-4.62295485	1.59911394
C	-2.20594811	-4.02685881	-2.20982289
H	-1.64377797	-3.46787310	-2.95990205
H	-2.07077193	-5.09692192	-2.38615990
H	-3.26064897	-3.77780890	-2.32195592
C	0.11666300	-4.22524595	-0.57547098
H	0.09555600	-5.28704309	-0.83236802
H	0.71233600	-3.68335509	-1.30937302
H	0.58540100	-4.09540892	0.39993501
C	-5.14515305	-0.27586800	1.12318301
H	-6.21613979	-0.20391101	0.91772598
H	-4.99688101	-0.77415299	2.08282995
H	-4.71903896	0.72482002	1.18730998
C	-4.81933403	-0.37437800	-1.73736095
H	-4.48726606	-0.94335097	-2.60780811
H	-5.90526390	-0.25688800	-1.77715099
H	-4.34378386	0.60660398	-1.76547503
S	-2.03727198	1.05550206	0.41832799
C	-0.36381099	1.86497295	0.40722400
H	0.07453200	1.79163599	-0.58880299
H	0.28409201	1.32334197	1.09810495
C	-0.50465298	3.29110289	0.83193099
H	-0.82806599	3.44976711	1.86002398
C	-0.27786499	4.34551716	0.04709200
H	0.03515100	4.22794008	-0.98752701
H	-0.39470699	5.36267424	0.40831900
C	4.45007086	-0.91304600	0.35371399

F	5.03195095	-2.13266993	0.13303600
F	4.89507723	-0.51046503	1.57114303
F	5.00077820	-0.07444100	-0.56060100

	1	2	3
	A	A	A
Frequencies --	14.6651	20.3316	24.6052
Red. masses --	16.3402	4.5762	7.2250
Frc consts --	0.0021	0.0011	0.0026
IR Inten --	0.2201	0.3964	0.8101

Sum of electronic and zero-point Energies= -2134.391125
 Sum of electronic and thermal Energies= -2134.359765
 Sum of electronic and thermal Enthalpies= -2134.358821
 Sum of electronic and thermal Free Energies= -2134.456514

HF = -2134.78821559

TS: reductive elimination from allylS-Pd(PMe₃)₂-(C₆H₄)CF₃ to allylS(C₆H₄)CF₃

Element	X	Y	Z
C	-3.96485400	-0.31793100	-0.60724998
C	-2.59341192	-0.17916100	-0.48494399
C	-1.99688303	1.09389400	-0.21020301
C	-2.90186310	2.18618798	0.01870300
C	-4.26032495	2.03137302	-0.10967800
C	-4.82283497	0.77714199	-0.44082800
H	-4.37537003	-1.29089200	-0.85651797
H	-1.96323001	-1.04521894	-0.64960200
H	-2.49508190	3.16029501	0.26964101
H	-4.90808201	2.89074492	0.03307200
Pd	-0.25380901	1.39175403	-1.46189499
P	2.06266308	1.00961900	-2.05083990
P	-1.42605400	2.21298099	-3.33108997
C	2.85535908	1.88377404	-3.47031212
H	3.91485906	1.62592995	-3.55708098
H	2.34754300	1.61109102	-4.39704323
H	2.76223803	2.96341896	-3.33610010
C	2.48272896	-0.74830300	-2.43452311
H	1.92939103	-1.07436299	-3.31744409
H	3.55382395	-0.87349600	-2.61857796
H	2.19066000	-1.38526595	-1.59731805
C	3.24449611	1.37682199	-0.68241900
H	4.27027178	1.10876405	-0.95142603
H	3.20500708	2.44143391	-0.44221899
H	2.94575310	0.82107598	0.20853300
C	-2.94417691	1.26809800	-3.76968002
H	-3.46499395	1.72929800	-4.61340284
H	-2.67816997	0.24200299	-4.03041506
H	-3.60890794	1.23543894	-2.90601397
C	-0.57218802	2.35407495	-4.95971298
H	-0.20398401	1.37311006	-5.26724005
H	-1.24520504	2.73936105	-5.73082495
H	0.28292301	3.02600002	-4.86681986
C	-2.09779406	3.91590500	-3.11378789
H	-2.72276402	3.94032502	-2.21947503

H	-1.27548504	4.62165880	-2.98159599
H	-2.69681001	4.22025394	-3.97677898
S	-0.38162601	1.12444699	0.93237501
C	-0.02615500	-0.66719103	1.28006196
H	-0.97614902	-1.11727202	1.58243704
H	0.31883299	-1.15487504	0.36926800
C	0.99586803	-0.74808300	2.36724710
C	2.21602011	-1.26214194	2.21136808
H	0.69855100	-0.34741801	3.33452511
H	2.92431998	-1.30130100	3.03230405
H	2.54860401	-1.66956997	1.26046205
C	-6.28787804	0.62272602	-0.52518100
F	-6.90868998	0.48837200	0.70180601
F	-6.90655088	1.69800401	-1.09982502
F	-6.67438221	-0.47135401	-1.23551297

	1	2	3
	A	A	A
Frequencies --	-152.8133	16.7599	20.1588
Red. masses --	9.4638	15.6754	4.6474
Frc consts --	0.1302	0.0026	0.0011
IR Inten --	170.6936	0.0616	0.7069
Sum of electronic and zero-point Energies=		-2134.353101	
Sum of electronic and thermal Energies=		-2134.321923	
Sum of electronic and thermal Enthalpies=		-2134.320979	
Sum of electronic and thermal Free Energies=		-2134.417698	

HF = -2134.74730529

allylS-Pd(PMe₃)₂-(C₆H₄)NMe₂

Element	X	Y	Z
Pd	-1.93573403	-1.26496601	-0.00551200
C	0.10598700	-1.11288798	0.11890200
C	0.78236300	-1.25912201	1.33390701
C	0.89544898	-0.82866400	-0.99939400
C	2.16855693	-1.12590694	1.43762302
H	0.22596900	-1.47000301	2.24386191
C	2.28261399	-0.69301897	-0.91957998
H	0.43003300	-0.69171500	-1.97253394
C	2.96322608	-0.85168898	0.30532300
H	2.62109804	-1.23951900	2.41489697
H	2.82644892	-0.45974001	-1.82655299
P	-1.58329105	-3.52161288	-0.46852100
P	-4.35111189	-1.24818695	-0.21743900
C	-5.37650394	-2.78124404	-0.25290501
H	-5.33653879	-3.27632809	0.71860301
H	-6.41763496	-2.52722192	-0.46724299
H	-5.02671814	-3.47693205	-1.01585400
C	-2.41716003	-4.71852589	0.65371799
H	-3.49499011	-4.56995916	0.64777100
H	-2.19754100	-5.74484491	0.34864599
H	-2.05428004	-4.56449890	1.67150497
C	-2.14679909	-4.02885914	-2.14644098
H	-1.57877600	-3.47066092	-2.89285707

H	-1.98772204	-5.09880114	-2.30344796
H	-3.20338702	-3.80133390	-2.28577900
C	0.13700999	-4.16254997	-0.45228401
H	0.13786200	-5.22878408	-0.69209403
H	0.74224102	-3.61978102	-1.17745399
H	0.57810801	-4.00628614	0.53192002
C	-5.24620199	-0.28422001	1.06870496
H	-6.31255579	-0.22840600	0.83528203
H	-5.11677980	-0.76786900	2.03861308
H	-4.83420897	0.72265899	1.13097501
C	-4.85217285	-0.41638201	-1.77956605
H	-4.48903513	-0.99090999	-2.63389301
H	-5.93891001	-0.31958401	-1.84676802
H	-4.39370012	0.57263398	-1.81092095
S	-2.12946796	1.09548199	0.41565999
C	-0.46572000	1.92420995	0.39626500
H	-0.03362800	1.85643101	-0.60285503
H	0.19148400	1.38280797	1.07808399
C	-0.61533803	3.34735990	0.82654500
H	-0.93509197	3.49969912	1.85696304
C	-0.40146500	4.40787506	0.04571900
H	-0.09333500	4.29754591	-0.99126202
H	-0.52548403	5.42255020	0.41184101
N	4.36229515	-0.76235998	0.38956001
C	4.94823694	-0.56009799	1.70785105
H	4.72940397	-1.40218198	2.36766601
H	6.03186703	-0.49937001	1.61289001
H	4.59066582	0.35938099	2.19685888
C	5.06560421	-0.11569200	-0.71008301
H	4.74098921	0.92409903	-0.87116498
H	6.13453007	-0.11304900	-0.49949101
H	4.92264700	-0.66406101	-1.64346099

	1	2	3
	A	A	A
Frequencies --	17.7454	26.3104	29.1465
Red. masses --	3.9822	4.3095	3.7039
Frc consts --	0.0007	0.0018	0.0019
IR Inten --	0.2243	1.6559	2.1578

Sum of electronic and zero-point Energies= -1931.165748
 Sum of electronic and thermal Energies= -1931.133730
 Sum of electronic and thermal Enthalpies= -1931.132785
 Sum of electronic and thermal Free Energies= -1931.230584

HF = -1931.63136409

TS: reductive elimination from allylS-Pd(PMe₃)₂-(C₆H₄)NMe₂ to allylS(C₆H₄)NMe₂

Element	X	Y	Z
C	-4.12468576	0.83014500	-0.24117599
C	-2.78791809	0.48860499	-0.06813700
C	-1.75829101	1.43858600	-0.20732699
C	-2.15818405	2.77100611	-0.43509400
C	-3.49264598	3.11250401	-0.63078701
C	-4.51993990	2.14759707	-0.56100702

H	-4.86048985	0.04107200	-0.15086800
H	-2.54624295	-0.54911000	0.12652400
H	-1.41307294	3.56027102	-0.44527900
H	-3.72453809	4.15330076	-0.81722802
Pd	-0.04180900	1.08582604	-1.47204101
P	2.09845591	1.83938801	-2.25494695
P	-1.19675803	0.04569200	-3.27138591
C	3.15623307	0.58089501	-3.09971499
H	4.12427521	0.99879402	-3.39219403
H	3.32157397	-0.26552299	-2.43001294
H	2.64345407	0.21309100	-3.99057388
C	3.26286602	2.47424197	-0.96987700
H	3.39427590	1.71606100	-0.19497500
H	4.23891592	2.73047090	-1.39254296
H	2.83338499	3.36094904	-0.49863300
C	2.14266992	3.21490502	-3.48963404
H	3.16805506	3.47090602	-3.77365398
H	1.59014904	2.91982603	-4.38411283
H	1.65867698	4.09881878	-3.06939006
C	-2.03231692	-1.57754397	-2.97763491
H	-2.60846806	-1.89481497	-3.85227489
H	-1.28528905	-2.33995104	-2.74735808
H	-2.70624804	-1.48669600	-2.12463188
C	-0.33086500	-0.29959401	-4.86807585
H	0.46639499	-1.02745903	-4.70251703
H	-1.01656795	-0.69270301	-5.62483692
H	0.12156000	0.62060398	-5.24391317
C	-2.62074590	1.07316005	-3.84062099
H	-3.27255011	1.27646303	-2.98928809
H	-2.25126505	2.02838802	-4.21897793
H	-3.19246006	0.57245803	-4.62773180
S	0.02867900	1.30075097	0.93338799
C	-0.58772898	-0.12046200	1.96610796
H	-1.61734402	0.10286000	2.26079988
H	-0.58124697	-1.03132701	1.36713398
C	0.29567900	-0.25509399	3.16474104
C	1.10606003	-1.28838503	3.39419007
H	0.25918499	0.56720400	3.87745690
H	1.72599101	-1.33952403	4.28367281
H	1.17635798	-2.12251711	2.70075297
N	-5.86167717	2.47343111	-0.81059700
C	-6.88411808	1.55582297	-0.32511401
H	-7.86773205	1.94756496	-0.58161998
H	-6.79309177	0.57795799	-0.80228698
H	-6.84173584	1.41024101	0.76537102
C	-6.23118877	3.88199711	-0.79612899
H	-5.68871117	4.43575001	-1.56508398
H	-7.29315519	3.97583795	-1.02020204
H	-6.03750992	4.36445904	0.17465000

	1	2	3
	A	A	A
Frequencies --	-262.0804	13.1910	22.1750
Red. masses --	10.8889	4.1636	3.9207
Frc consts --	0.4407	0.0004	0.0011
IR Inten --	278.5000	0.3548	1.1991

Sum of electronic and zero-point Energies= -1931.119599
Sum of electronic and thermal Energies= -1931.087375
Sum of electronic and thermal Enthalpies= -1931.086430
Sum of electronic and thermal Free Energies= -1931.186193

HF = -1931.58228456

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13. Structures can be found in the cif document attached. Reports were directly generated from the .log file using the GEAC program (link: <https://github.com/LaTruelle/Geac>).
E. Nicolas,
<http://doi.org/10.5281/zenodo.57101>)