# **Electronic Supplementary Information** *for*

## Halogenated Tetraphenylethene with Enhanced Aggregation-Induced Emission: Anomalous Anti-Heavy-Atom Effect and Self-Reversible Mechanochromism

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#### **1. Experimental Section**

Synthesized of 1,1,2,2-Tetraphenylethene (TPE). 1,1,2,2-Tetraphenylethylene was synthesized and purified according to the previous procedure in the literature.<sup>7</sup> <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.10(m, J = 12 Hz, 12H), 7.03(m, J = 6 Hz, 8H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 143.91, 141.12, 131.52, 127.83, 126.60. MS (ESI) m/z: [M<sup>+</sup>] 332.2 (calcd. for C<sub>26</sub>H<sub>20</sub>, 332.2).

Synthesized of 1,1,2,2-Tetra(4-fluorophenyl)ethene (TFPE). A certain amounts of Zn dust (1.84 g, 28 mmol) and TiCl<sub>4</sub> (1.58 mL, 14 mmol) were refluxed for 3 h in 100 mL of dry THF under N<sub>2</sub> atmosphere. A solution of 4,4'-Difluorobenzophenone (1.53 g, 7 mmol) in dry THF (30 mL) was added to the preceding suspension, and then the reaction was refluxed at 80 °C for 8 h. An aqueous solution containing 10% K<sub>2</sub>CO<sub>3</sub> (50 mL) was added after the reaction mixture was cooled down to room temperature. The resulting product was extracted with ethyl acetate. The solvent was evaporated under vacuum and the crude product was purified by a silica gel column using pure petroleum ether as the eluent. TFPE was obtained in 90% yield (1.37 g). <sup>1</sup>H NMR(600 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 6.94(d, J = 12 Hz, 8H), 6.81(d, J = 18 Hz, 8H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 162.56, 160.92, 139.30, 133.01, 115.23. HRMS (ESI) m/z [M+Cl<sup>-</sup>] 439.0906 (calcd. for C<sub>26</sub>H<sub>16</sub>F<sub>4</sub>, 404.1188).

Synthesis of 1,1,2,2-Tetra(4-chlorophenyl)ethene (TCPE). The synthesis of TCPE followed the synthesis procedure of TFPE with replacing 4,4'-difluorobenzophenone with 4,4'-dichlorobenzophenone. The crude product was finally purified by column chromatography to give a white solid (Yield 86%). Molecular formula:  $C_{26}H_{16}Cl_4$ . <sup>1</sup>H NMR(600 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.10(d, J = 12 Hz, 8H), 6.90(d, J = 12 Hz, 8H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 141.12, 139.56, 132.99, 132.45, 128.33. HRMS (ESI) m/z [M+Cl<sup>-</sup>] 502.9719 (calcd. for  $C_{26}H_{16}Cl_4$ , 468.0006).

**Synthesis of 1,1,2,2-Tetra(4-Bromophenyl)ethene (TBPE).** A similar procedure was applied to synthesis of TBPE using 4,4'-dibromobenzophenone as the initial reactant. A white powder in a yield of 80% was acquired after further purification using column

chromatography. Molecular formula:  $C_{26}H_{16}Br_4$ . <sup>1</sup>H NMR(600 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.23(d, J = 6 Hz, 8H), 6.82(d, J = 12 Hz, 8H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 141.67, 139.83, 132.96, 131.53, 121.53. HRMS (ESI) m/z: [M+Cl<sup>-</sup>] 678.7687 (calcd. for  $C_{26}H_{16}Br_4$ , 643.7985).

Synthesis of 1,1,2,2-Tetra(4-Iodophenyl)ethene (TIPE). A given amount of tetraphenylethylene (1.028 g, 3.1 mmol) was first dissolved in 20 ml THF, and then 1.718 g of iodate (9.8 mmol), 1.914 g of iodine (7.5 mmol), 4 mL of water and 80 mL of acetic acid were added. The resulting mixture was stirred at 80 °C for 12 h. Then the mixture was cooled to room temperature and iced water was added. The product was extracted with dichloromethane, and washed with sodium thiosulfate. The dichloromethane solution was dried with anhydrous magnesium sulfate, and further concentrated to solid under vacuum. The obtained crude product was recrystallized by ethyl acetate and petroleum ether. Yield: 0.68 g (66%). <sup>1</sup>H NMR(600 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.44(d, J = 6 Hz, 8H), 6.69(d, J = 12 Hz, 8H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 142.26, 140.00, 137.68, 133.16, 93.24.

**Characterization of Fluorescence Properties of All Samples.** Steady PL spectra of all samples were performed on an Edinburgh Instruments model FLS980 fluorescence spectrophotometer equipped with a xenon arc lamp using a front face sample holder. Time-resolved fluorescence measurements were conducted with EPL-series lasers. The fluorescence quantum yields of all samples were determined using an integrating sphere equipped in FLS980 spectrophotometer for at least three times.

#### 2. Computational Details

All the calculations were performed with density functional theory (DFT) and timedependent density functional theory (TDDFT) implemented in Gaussian 09 program package.<sup>1</sup> The ground state equilibrium geometries and the normal modes of vibration of the TPE derivatives were computed using density functional theory (DFT) with the hybrid B3LYP functional.<sup>2</sup> The basis set 6-31+G(d,p) level was used for C, H, F and Cl, and the basis set LanL2DZ was used for Br and I. NBO calculations were performed at the same level, and the visualization of HOMOs and LUMOs were conducted with Gview program. Excited state calculations were run at PBE0 functional with time-dependent density functional theory (TD-DFT).<sup>3,4</sup> The basis set 6-31+G(d,p) level was used for C, H, F and Cl, and the basis set LanL2DZ was used for Br and I.

### 3. Supplementary Schemes and Figures



**Scheme S1**. (a) Synthesis routes of TPE, TFPE, TCPE and TBPE, and (b) synthesis route of TIPE.



**Figure S1**. Time-resolved fluorescence decay curves of TPE (a), TFPE (b), TCPE (c), TBPE (d) and TIPE (e) in solid state.



**Figure S2**. Aggregation-induced emission behaviors of TPE (a), TFPE (b), TCPE (c), TBPE (d) and TIPE (e) using water as the poor solvent.



**Figure S3**. Crystal structures and molecular packing with labeled distances between halogen-halogen atoms of TFPE (a), TCPE (b), TBPE (c) and TIPE (d). Carbon, hydrogen, fluorine, chlorine, bromine and iodine atoms are represented in gray, white, green, olive, red and purple respectively.



Figure S4. PL spectra of TPE, TFPE, TCPE, TBPE and TIPE dispersed in SiO<sub>2</sub>.



**Figure S5**. Time-resolved fluorescence decay curves of TPE (a), TFPE (b), TCPE (c), TBPE (d) and TIPE (e) dispersed in SiO<sub>2</sub>.



**Figure S6**. PL spectra and fluorescent images of TFPE, TCPE, TBPE and TIPE in THF at 77 K.



**Figure S7**. Time-resolved fluorescence decay curves of TFPE (a), TCPE (b), TBPE (c) and TIPE (d) in THF at 77 K.



**Figure S8**. (a) PL spectra of pristine, ground and self-recovered TCPE solid within 8 min. Inset: photos of pristine crystals, ground crystals and self-recovered crystals of TCPE. (b) XRD patterns of pristine, grinding and self-recovered crystals of TCPE.



**Figure S9**. (a) PL spectra of pristine, ground and self-recovered TIPE solid within 8 min. Inset: photos of pristine crystals, ground crystals and annealed crystals of TIPE. (b) XRD patterns of pristine, grinding and self-recovered crystals of TIPE.

### 4. Supplementary Tables

Compound	Center	Contribution to HOMO (%)	Contribution to LUMO (%)
TDE	С	98.86	95.77
IPE	Н	0.29	0.20
	С	94.59	94.40
TFPE	Н	0.15	0.19
	F	4.76	2.11
	С	89.61	93.72
ТСРЕ	Н	0.14	0.18
	Cl	9.75	2.51
	С	82.74	84.98
TBPE	Н	4.82	5.20
	Br	12.07	2.22
	С	82.16	94.26
TIPE	Н	0.14	0.21
	Ι	17.30	2.30

Table S1. Contribution ratioes of different atoms to HOMOs and LUMOs of TPE, TFPE, TCPE, TBPE and TIPE

Table S2. Excitation energies of first 20 states for TPE, TFPE, TCPE, TBPE andTIPE calculated with TDDFT at PBE0/6-31+G(d,p) level.

State	ТРЕ	TFPE	ТСРЕ	TBPE	TIPE
No					
1	2.3532 eV (T <sub>1</sub> )	2.3352 eV (T <sub>1</sub> )	2.2806 eV (T <sub>1</sub> )	2.2852 eV (T <sub>1</sub> )	2.2798 eV (T <sub>1</sub> )
2	3.3232 eV (T <sub>2</sub> )	3.3317 eV (T <sub>2</sub> )	3.2235 eV (T <sub>2</sub> )	3.2338 eV (T <sub>2</sub> )	3.2157 eV (T <sub>2</sub> )
3	3.3424 eV (T <sub>3</sub> )	3.3529 eV (T <sub>3</sub> )	3.2369 eV (T <sub>3</sub> )	3.2471 eV (T <sub>3</sub> )	3.2275 eV (T <sub>3</sub> )
4	3.4676 eV (T <sub>4</sub> )	3.4831 eV (T <sub>4</sub> )	3.3768 eV (T <sub>4</sub> )	3.3840 eV (T <sub>4</sub> )	3.3625 eV (T <sub>4</sub> )
5	3.7072 eV (S <sub>1</sub> )	3.6830 eV (S <sub>1</sub> )	3.5553 eV (S <sub>1</sub> )	3.5401 eV (S <sub>1</sub> )	3.5043 eV (S <sub>1</sub> )
6	3.8032 eV (T <sub>5</sub> )	3.8001 eV (T <sub>5</sub> )	3.7564 eV (T <sub>5</sub> )	3.7561 eV (T <sub>5</sub> )	3.7414 eV (T <sub>5</sub> )
7	4.0436 eV (T <sub>6</sub> )	3.8263 eV (T <sub>6</sub> )	3.8731 eV (T <sub>6</sub> )	3.8912 eV (T <sub>6</sub> )	3.8587 eV (T <sub>6</sub> )
8	4.1787 eV (T <sub>7</sub> )	3.9770 eV (T <sub>7</sub> )	4.0343 eV (T <sub>7</sub> )	4.0501 eV (T <sub>7</sub> )	3.9471 eV (T <sub>7</sub> )
9	4.2503 eV (T <sub>8</sub> )	4.1109 eV (T <sub>8</sub> )	4.1585 eV (T <sub>8</sub> )	4.1679 eV (T <sub>8</sub> )	3.9488 eV (T <sub>8</sub> )
10	4.2655 eV (T <sub>9</sub> )	4.1473 eV (S <sub>2</sub> )	4.1814 eV (S <sub>2</sub> )	4.1893 eV (S <sub>2</sub> )	3.9498 eV (T <sub>9</sub> )
11	4.3123 eV	4.1513 eV (T <sub>9</sub> )	4.1862 eV (T <sub>9</sub> )	4.1908 eV (T <sub>9</sub> )	3.9619 eV
	(T <sub>10</sub> )				(T <sub>10</sub> )
12	4.3450 eV (S <sub>2</sub> )	4.3556 eV (S <sub>3</sub> )	4.2885 eV	4.2681 eV	4.0386 eV
			(T <sub>10</sub> )	(T <sub>10</sub> )	(T <sub>11</sub> )
13	4.3906 eV	4.3622 eV	4.2999 eV (S <sub>3</sub> )	4.2991 eV (S <sub>3</sub> )	4.1494 eV (S <sub>2</sub> )
	(T <sub>11</sub> )	$(T_{10})$			

14	4.4467 eV	$(S_3)$	4.3971	eV	4.3415	eV	4.3328	eV	4.1494	eV
			(T <sub>11</sub> )		$(T_{11})$		(T <sub>11</sub> )		$(T_{12})$	
15	4.4721	eV	4.4909	eV	4.4154	eV	4.3932	eV	4.1741	eV
	(T <sub>12</sub> )		$(T_{12})$		$(T_{12})$		$(T_{12})$		(T <sub>13</sub> )	
16	4.4745	eV	4.5388	eV	4.4167	eV	4.3963	eV	4.2216	eV
	(T <sub>13</sub> )		(T <sub>13</sub> )		(T <sub>13</sub> )		(T <sub>13</sub> )		(T <sub>14</sub> )	
17	4.5107	eV	4.6051	eV	4.4313	eV	4.4178	eV	4.2607 eV	$V(S_3)$
	(T <sub>14</sub> )		(T <sub>14</sub> )		$(T_{14})$		(T <sub>14</sub> )			
18	4.5964	eV	4.6433 eV	V (S <sub>4</sub> )	4.4887 eV	V (S <sub>4</sub> )	4.4449 eV	V (S <sub>4</sub> )	4.2966	eV
	(T <sub>15</sub> )								(T <sub>15</sub> )	
19	4.6342	eV	4.6824 eV	$V(S_5)$	4.5130	eV	4.4598	eV	4.3052 eV	V (S <sub>4</sub> )
	$(T_{16})$				$(T_{15})$		(T <sub>15</sub> )			
20	4.6483	eV	4.6957 eV	$V(S_6)$	4.5633	eV	4.5361	eV	4.3329	eV
	(T <sub>17</sub> )				(T <sub>16</sub> )		(T <sub>16</sub> )		(T <sub>16</sub> )	

 Table S3. Photophysical properties of TPE, TFPE, TCPE, TBPE and TIPE in different states

	$\lambda_{ab} (nm)$	$\lambda_{em}$ (nm)	τ (ns)	Ф (%)	$k_{\rm r} (10^8  {\rm s}^{-1})$	$k_{\rm nr} (10^8  {\rm s}^{-1})$
			Crystal			
TPE	-	445	1.5	32.6	2.2	4.5
TFPE	-	438	3.0	58.6	2.0	1.3
TCPE	-	445	2.4	75.5	3.1	1.1
TBPE	-	452	2.0	64.4	3.2	1.8
TIPE	-	467	2.1	49.3	2.3	2.5
		Di	ispersed in Si	$\mathcal{O}_2$		
TPE	-	470	3.9	22.2	0.57	2.0
TFPE	-	470	3.6	5.2	0.14	2.6
TCPE	-	480	6.6	47.0	0.71	0.81
TBPE	-	485	6.7	54.3	0.81	0.69
TIPE	-	493	3.8	41.9	1.10	1.53
			In solution			
TPE	240 310	466	6.7	< 1	-	-
TFPE	231 304	460	6.5	< 1	-	-
TCPE	250 320	475	6.1	< 1	-	-
TBPE	253 323	510	5.3	< 1	-	-
TIPE	253 327	515	4.0	< 1	-	-

source	TFPE	TCPE	TBPE	TIPE
CCDC No.	1948854	1948855	1948856	1958228
Chemical formula	$C_{26}H_{16}F_4$	$C_{26}H_{16}Cl_4$	C <sub>26</sub> H <sub>16</sub> Br <sub>4</sub> 2(C <sub>4</sub> H <sub>8</sub> O)	$C_{26}H_{16}I_4$
Formula weight	404.39	470.19	792.24	836.02
Temperature (K)	289	298	293	273
wavelength (Å)	0.71073	1.54178	0.71073	0.71073
crystal system	orthorhombic	monoclinic	orthorhombic	orthorhombic
space group	Fdd2	P 21/n	P 21 21 21	P 21 21 21
a (Å)	18.5763	20.1695	9.3392	9.6191
b (Å)	40.241	9.8843	15.1764	17.2316
c (Å)	5.6042	23.3874	21.938	18.9386
α	90.00	90.00	90.00	90.00
β	90.00	92.701	90.00	90.00
γ	90.00	90.00	90.00	90.00
volume (Å <sup>3</sup> )	4189.3	4657.4	3109.4	3139.1
Ζ	8	8	4	4
no. reflections collected	7647	60267	37711	33612
no. unique reflections	2344	8465	7120	8873
R <sub>int</sub>	0.0224	0.0699	0.0964	0.0286
no. parameters	136	541	361	308
GOF on F <sup>2</sup>	1.057	1.051	1.017	1.050
R1	0.0396	0.0708	0.0457	0.0427
wR2	0.1034	0.2078	0.0971	0.1383

Table S4. Crystallographic data for TFPE, TCPE, TBPE and TIPE

# 5. Spectra of Compounds



<sup>1</sup>H NMR spectrum of 1,1,2,2-tetraphenylethene (TPE) in CDCl<sub>3</sub>



<sup>13</sup>C NMR spectrum of 1,1,2,2-tetraphenylethene (TPE) in CDCl<sub>3</sub>



<sup>1</sup>H NMR spectrum of 1,1,2,2-tetra(4-fluorophenyl)ethylene (TFPE) in CDCl<sub>3</sub>



<sup>13</sup>C NMR spectrum of 1,1,2,2-tetra(4-fluorophenyl)ethylene (TFPE) in CDCl<sub>3</sub>



<sup>1</sup>H NMR spectrum of 1,1,2,2-tetra(4-chlorophenyl)ethylene (TCPE) in CDCl<sub>3</sub>



<sup>13</sup>C NMR spectrum of 1,1,2,2-tetra(4-chlorophenyl)ethylene (TCPE) in CDCl<sub>3</sub>



<sup>1</sup>H NMR spectrum of 1,1,2,2-tetra(4-bromophenyl)ethylene (TBPE) in CDCl<sub>3</sub>



<sup>13</sup>C NMR spectrum of 1,1,2,2-tetra(4-bromophenyl)ethylene (TBPE) in CDCl<sub>3</sub>



<sup>1</sup>H NMR spectrum of 1,1,2,2-tetra(4-iodophenyl)ethylene (TIPE) in CDCl<sub>3</sub>



<sup>13</sup>C NMR spectrum of 1,1,2,2-tetra(4-iodophenyl)ethylene (TIPE) in CDCl<sub>3</sub>

## 6. Cartesian Coordinates

T	וי	P	E
			<u> </u>

С	5.61546800	12.44776300	14.27613600
С	6.39406700	12.73758800	13.03157700
С	5.12501800	12.81464100	16.71282500
С	6.68423700	11.71092900	12.11463500
Н	6.35301500	10.69841600	12.32678300
С	7.37670900	13.43820700	15.77011800
С	3.48049800	12.75589900	19.00689700
С	6.01533300	12.87387700	15.51172200
С	3.00716900	13.30964600	17.81413600
Н	2.00368200	13.72309500	17.76531500
С	9.80502700	13.28717900	15.61559300
Н	10.68751700	12.75099000	15.27793600
С	7.39571300	11.97432200	10.94333000
Н	7.62014200	11.16300800	10.25641800
С	4.36513200	11.65500900	14.06030500
С	7.52873400	14.63054800	16.50099800
Н	6.64544300	15.15054100	16.86034200

С	8.53761700	12.77061600	15.34202700
Н	8.44268900	11.83918000	14.79325000
С	6.81004700	14.04339800	12.71856400
Н	6.58084500	14.85263700	13.40451800
С	5.59562700	12.28116300	17.92642100
Н	6.60823900	11.89167500	17.97945500
С	4.77948800	12.24119100	19.05778500
Н	5.16030700	11.81285200	19.98080700
С	3.82045100	13.33732300	16.68009500
Н	3.44473900	13.76953600	15.75816900
С	4.08581600	10.50754900	14.82300900
Н	4.77994400	10.20468800	15.60047400
С	3.46105700	12.01166400	13.04320000
Н	3.66897600	12.88201800	12.42751600
С	9.93979100	14.48551900	16.32218200
С	8.79484400	15.15520200	16.76434100
Н	8.88714100	16.08431000	17.31993700
С	2.93309900	9.75520500	14.59180900
Н	2.74087200	8.86989400	15.19148100
С	7.81213200	13.27660100	10.65082600
С	2.03187000	10.13550200	13.59346200
С	2.30155100	11.26811500	12.81932900
Н	1.61118200	11.57068000	12.03684500
С	7.50983300	14.31092500	11.54088000
Н	7.81761600	15.32887100	11.31868800
Н	8.35969500	13.48385500	9.73590800
Н	1.13330900	9.55171100	13.41563300
Н	2.84714200	12.73209600	19.88902800
Н	10.92595700	14.88922500	16.53284100
TFPE			
С	5.61602100	12.44663400	14.27540300
С	6.39079200	12.74372500	13.03097000
С	5.12260800	12.82051400	16.70970100
С	6.67881300	11.72396700	12.10531900
Н	6.34913700	10.70944400	12.30792400
С	7.37950900	13.43076900	15.77036400
С	3.48768300	12.77233600	18.97001600
С	6.01619900	12.87311700	15.51148400
С	2.99282900	13.32516200	17.79546400
Н	1.98775100	13.73222100	17.76996200

Cl	2.45068200	12.72619400	20.40996000
Cl	11.52666200	15.12356000	16.65102300
ТСРЕ			
Г	11.13422900	14.73438000	10.37724300
r F	2.09041700	12.74732300	20.07104800
Г Г	0.9455/400	9.40144000	13.3924/400
Г Б	8.4/290/00	13.33843100	9.5552/500
H	7.82618500	15.34414600	11.31058400
C II	7.50699000	14.33594900	11.55175800
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#### 7. References

M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K.Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajiama, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Jr. Montgomery, J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Borthers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J.Normand, K. Rahavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V.Bakken, C. Adamo, Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski and D. J. Fox, Gaussian, Inc. Wallingford CT, 2009.

(2) A. D. Becke, J. Chem. Phys. 1993, 98, 5648-5652.

(3) F. J. A. Ferrer, J. Cerezo, E. Stendardo, R. Improta, F. Santoro, J. Chem. Theory Comput. 2013, 9, 2072–2082.

(4) C. Adamo, D. Jacquemin, Chem. Soc. Rev. 2013, 42, 845-856.