

Supporting Information

Semi-bridging σ -Silyls as Z-type Ligands

Benjamin J. Frogley, Anthony F. Hill,* Manab Sharma, Arup Sinha and Jas S. Ward

Synthetic Details and Spectroscopic Data

General Considerations

All manipulations of air-sensitive compounds were carried out under a dry and oxygen-free argon atmosphere using standard Schlenk, dry-box and vacuum line techniques. Glassware was flame-dried under vacuum prior to use. Solvents were dried by conventional methods. NMR spectra were recorded at 25°C on Bruker Avance 400 MHz and 600 MHz spectrometers. The chemical shifts (δ) for ^1H and ^{13}C spectra are given in ppm relative to residual signals of the solvent, ^{29}Si relative to an internal SiMe_4 reference, and ^{31}P relative to an external H_3PO_4 reference. Low resolution mass spectra were obtained on a ZAB-SEQ4F spectrometer by positive ion ESI techniques using an acetonitrile matrix by the mass spectrometry service of the Australian National University. Assignments were made relative to M, where M is the molecular cation. Elemental microanalysis was performed by the micro analytical service of the Australian National University. Data for X-ray crystallography were collected with a Nonius Kappa CCD diffractometer. The compounds **1**,^[1] $[\text{Pd}_2(\text{dba})_3]$,^[2] $[\text{Pt}(\text{nbe})_3]$ ^[3] and $[\text{Ni}(\text{cod})_2]$ ^[4] were prepared according to published procedures. Other reagents were used as received from commercial suppliers.

Synthesis of $[\text{Pd}_2\{\text{SiPh}(\text{NCH}_2\text{PPh}_2)_2\text{C}_6\text{H}_4\}_2]$ (**4**)

To a purple solution of $[\text{Pd}_2(\text{dba})_3 \cdot \text{dba}]$ (50 mg, 0.087 mmol) in thf (10 mL) was added the pro-ligand $\text{HSiPh}(\text{CH}_2\text{PPh}_2)_2\text{C}_6\text{H}_4$ (**3**: 53 mg, 0.087 mmol). The mixture was stirred at room temperature for 20 hours. The resulting orange solution was concentrated and pentane was added to induce precipitation. The orange precipitate was collected by filtration, washed with pentane and diethyl ether to remove excess dba and H_2dba and dried *in vacuo*. Although conversion was spectroscopically near quantitative, the isolated yield was diminished due to the requisite extensive washing with diethyl ether. Yield: 48 mg (77%). ^1H NMR (600 MHz, C_6D_6): $\delta_{\text{H}} = 7.61$ (d, 4 H, $^3J_{\text{HH}} = 7.2$ Hz, C_6H_4), 7.30 (t, 4 H, $^3J_{\text{HH}} = 8.4$ Hz, C_6H_5), 7.02 (t, 2 H, $^3J_{\text{HH}} = 7.2$ Hz), 7.00 (t, 2 H, $^3J_{\text{HH}} = 7.2$ Hz), 6.92-6.89 (m, 6H, $^3J_{\text{HH}} = 7.2$ Hz), 6.87-6.79 (m, 16 H, C_6H_5), 6.74-6.70 (m, 8 H, C_6H_5), 6.65 (t, 6 H, $^3J_{\text{HH}} = 6.6$ Hz), 6.61 (t, 4 H, $^3J_{\text{HH}} = 7.2$ Hz), 6.51-6.6 (m, 6 H, C_6H_5), 4.37 (dd, 2 H, $J = 13.2, 4.9$ Hz, PCH₂), 4.02 (dt, 2 H, $J = 13.3, 4.0$ Hz, PCH₂), 3.87 (dd, 2 H, $J = 13.1, 2.9$ Hz, PCH₂), 3.45 (dt, 2 H, $J = 13.1, 4.3$ Hz, PCH₂). $^{13}\text{C}\{^1\text{H}\}$ NMR (150 MHz, C_6D_6): $\delta_{\text{C}} = 144.0$ - 143.8 (m), 142.6 (t, $J = 10.5$ Hz), 142.2 (t, $J = 3$ Hz), 139.8 (t, $J = 10.5$ Hz), 136.6 (d, $J = 24$ Hz), 136.4 (d, $J = 28.5$ Hz), 134.6 (t, $J = 9$ Hz), 134.1 (t, $J = 7.5$ Hz), 133.9, 133.0 (t, $J = 6$ Hz), 132.0 (t, $J = 6$ Hz), 129.7, 129.6, 129.1, 128.7, 128.5, 128.3, 127.9, 127.4, 118.4, 116.8, 108.6, 107.7, 48.2 (d, $J = 30$ Hz), 45.9 (t, $J = 12$ Hz). $^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, C_6D_6): $\delta_{\text{P}} = 44.91$ (t, 2 P, $J = 10.5$ Hz), 25.13 (t, 2 P, $J = 10.5$ Hz). NB: The **AA'BB'** spin system manifests as two apparent triplet resonances (*cf.* **A₂B₂**), the inference being that $J_{\text{AB}} \approx J_{\text{AB}'} \ll (\delta_{\text{A}} - \delta_{\text{B}})$. ^{29}Si NMR (79.5 MHz, 300 K, C_6D_6): $\delta_{\text{Si}} = 58.35$ (m, 2Si). ESI-MS (low-res, +ve ion, MeCN): $m/z = 1428$ [M]⁺. Anal. Found: C, 64.19; H, 4.76; N, 3.92%. Calcd. for $\text{C}_{76}\text{H}_{66}\text{N}_4\text{P}_4\text{Si}_2\text{Pd}_2$: C, 63.91; H, 4.66; N, 3.92%. Crystals of a benzene solvate **4**.(C_6H_6)₂ suitable for X-ray diffraction were grown by slow diffusion of hexane vapour into a benzene (CAUTION) solution of compound **4**. *Crystal data for 4*.(C_6H_6)₂: $\text{C}_{88}\text{H}_{78}\text{N}_4\text{P}_4\text{Pd}_2\text{Si}_2$, $M_w = 1584.52$ g mol⁻¹, orange block, $0.1903 \times 0.1403 \times 0.0611$ mm³, triclinic, space group *P*-1 (No. 2), $V = 3724.1(2)$ Å³, $Z = 2$, $D_{\text{calcd}} = 1.4129$ Mgm⁻³, $F_{000} = 1636$, Cu K α radiation, $\lambda = 1.54184$ Å, $\mu = 5.398$ mm⁻¹, $T = 150.0(1)$ K, $2\theta_{\text{max}} = 144.9^\circ$, 21950 reflections collected, 14151 unique ($R_{\text{int}} = 0.0384$), $R_1 = 0.0558$, $wR_2 = 0.1515$, R indices based on 12345 reflections with $I > 2\sigma(I)$ (refinement on F^2), 829 parameters, Lp and absorption corrections applied, CCDC 1517410.

Supporting Information

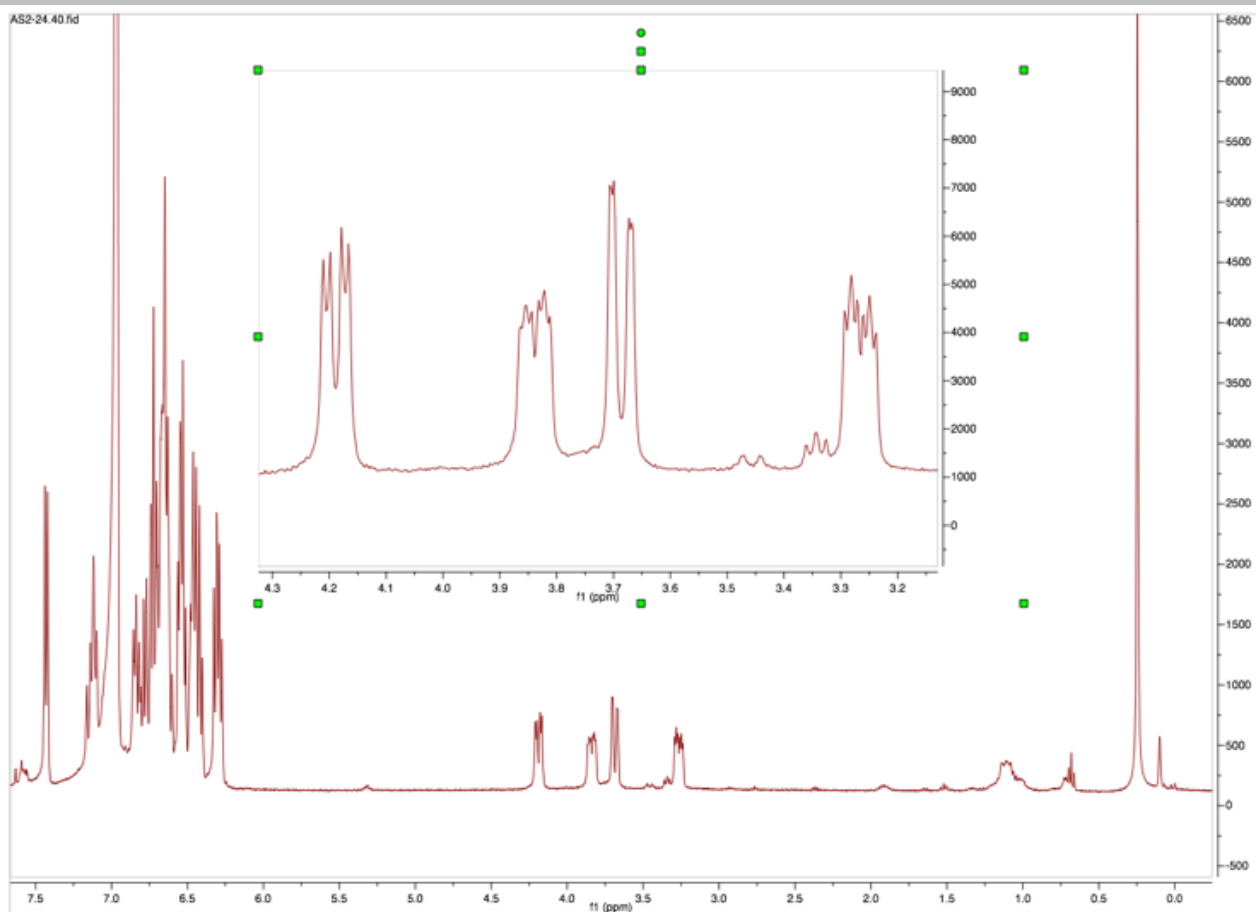


Figure S1. ^1H NMR Spectrum of 4 (400.1 MHz, 298 K, C_6D_6)

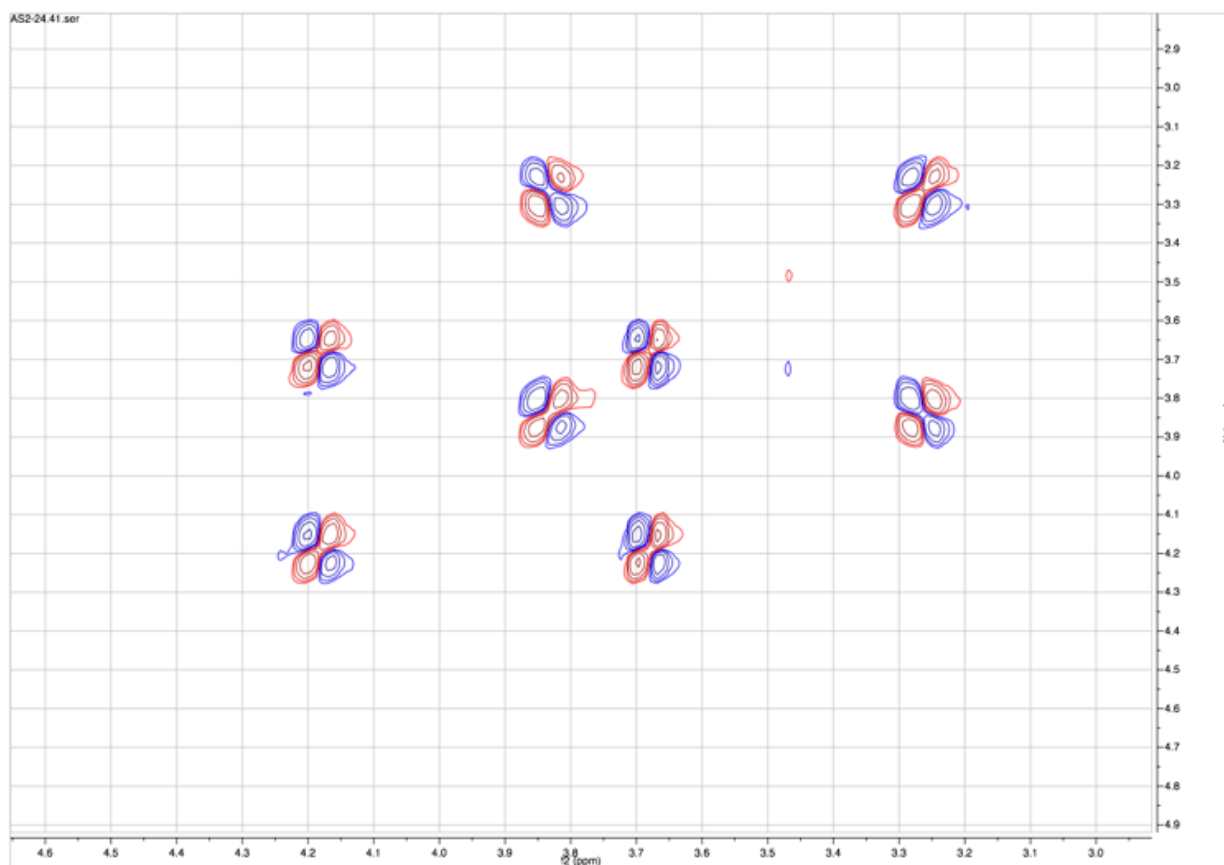
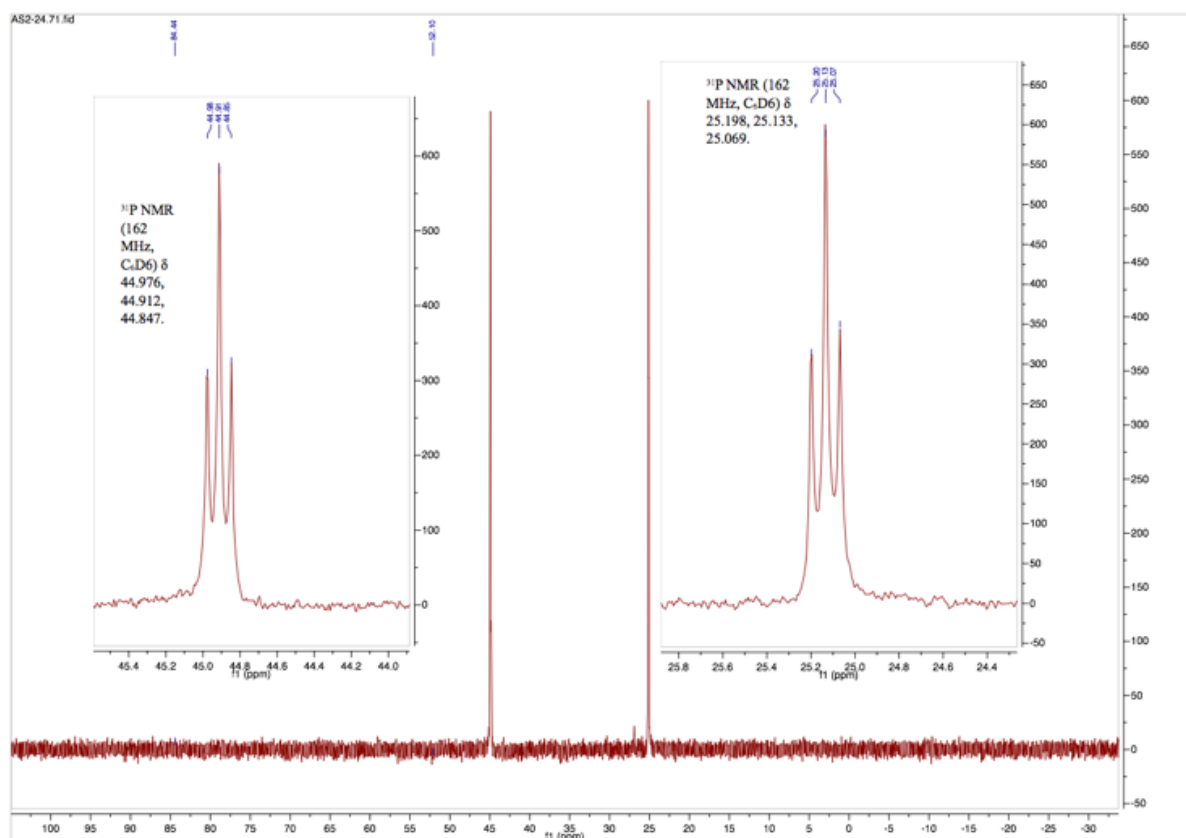
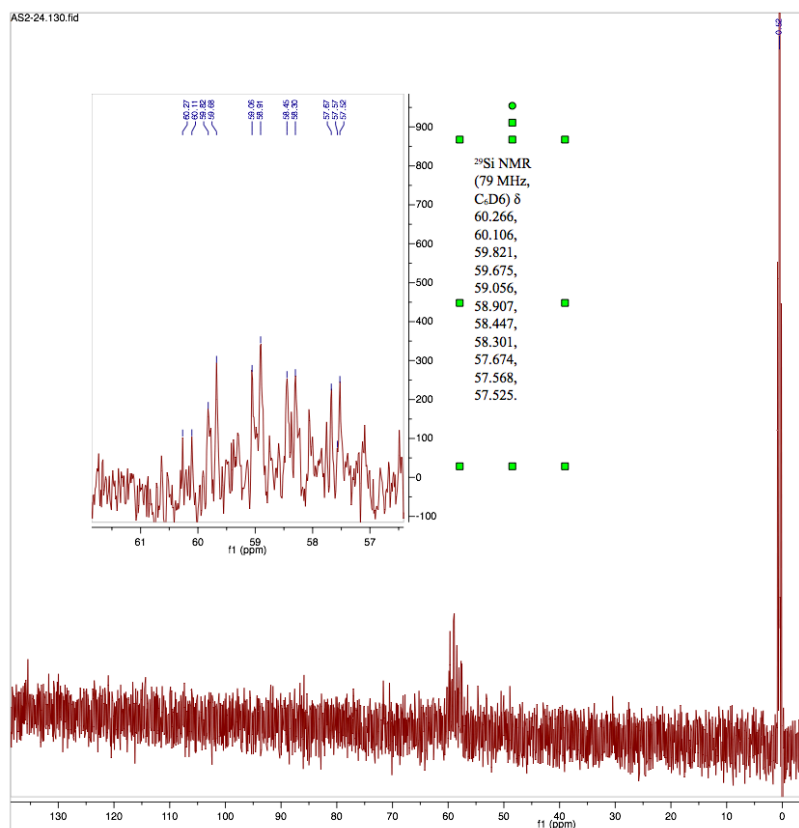


Figure S2. ^1H - ^1H COSY NMR Spectrum of 4 (PCH_2 Region).

Figure S3. $^{31}\text{P}\{^1\text{H}\}$ NMR Spectrum of **4** (162 MHz, C_6D_6 , 298 K)Figure S4. $^{29}\text{Si}\{^1\text{H}\}$ NMR Spectrum of **4** (162 MHz, C_6D_6 , 298 K)

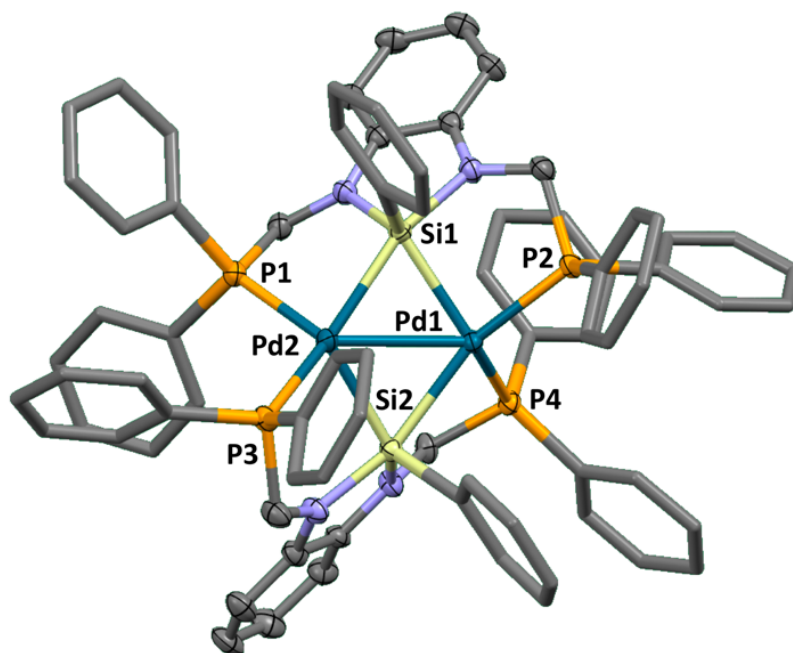


Figure S5. Molecular structure of **4** in a crystal of $4 \cdot (\text{C}_6\text{H}_6)_2$ (50% displacement ellipsoids, hydrogen atoms and solvent omitted, phenyl groups simplified). Selected bond lengths (Å) and angles ($^\circ$): Pd1–Pd2 2.7856(4), Pd1–P1 2.3484(11), Pd1–P3 2.3507(11), Pd1–Si1 2.4730(11), Pd1–Si2 2.4068(11), Pd2–Pd2–P4 2.3458(10), Pd2–Si1 2.4128(11), Pd2–Si2 2.4732(11), P1–Pd1–Pd2 23.05(3), P1–Pd1–P3 105.41(4), P1–Pd1–Si1 81.73(4), P1–Pd1–Si2 130.60(4), P3–Pd1–Pd2 130.19(3), P3–Pd1–Si1 154.65(4), P3–Pd1–Si2 84.63(4), Si1–Pd1–Pd2 54.23(3), Si2–Pd1–Pd2 56.32(3), Si2–Pd1–Si1 109.51(4), P2–Pd2–Pd1 130.91(3), P2–Pd2–Si1 84.58(4), P2–Pd2–Si2 155.87(4), P4–Pd2–Pd1 122.45(3), P4–Pd2–P2 104.85(4), P4–Pd2–Si1 129.77(4), P4–Pd2–Si2 81.79(4), Si1–Pd2–Pd1 56.26(3), Si1–Pd2–Si2 109.31(4), Si2–Pd2–Pd1 54.08(3).

Synthesis of $[\text{Ni}_2\{\text{SiPh}(\text{NCH}_2\text{PPh}_2)_2\text{C}_6\text{H}_4\}_2]$ (**5**)

The pro-ligand $\text{HSiPh}(\text{CH}_2\text{PPh}_2)_2\text{C}_6\text{H}_4$ (**3**; 110 mg, 0.182 mmol) was added to a thf (10 mL solution of $[\text{Ni}(\text{cod})_2]$ (50.0 mg, 0.182 mmol). Immediately the colour of the solution changed to dark reddish brown. After stirring at room temperature for 10 mins the solution was concentrated and pentane was added to induce precipitation. The resulting dark red-brown precipitate was washed with pentane and dried *in vacuo*. Yield: 105 mg (87%). ^1H NMR (600 MHz, C_6D_6): $\delta_{\text{H}} = 7.798$ (d, 4H, $J = 7.2$ Hz), 7.122 (dt, 4H, $J = 7.8$ Hz, $J = 1.2$ Hz), 7.074–7.026 (m, 10H), 6.934 (t, 4H, 7.2 Hz), 6.823–6.647 (m, 20H), 6.495–6.466 (m 8H), 6.419 (dt, 4H, $J = 7.8$ Hz, $J = 1.2$ Hz), 6.118 (t, 4H, $J = 7.8$ Hz), 4.111 (dd, 2H, $J = 12$ Hz, $J = 6.2$ Hz), 3.922 (dd, 2H, $J = 13.1$ Hz, $J = 2.9$ Hz), 3.716 (dd, 2H, $J = 11.9$ Hz, $J = 2.5$ Hz), 3.605 (dd, 2H, $J = 13$ Hz, $J = 7.5$ Hz). ^{13}C NMR (150 MHz, C_6D_6): $\delta = 143.5$ (d, $J = 16.5$ Hz), 142.98 (d, $J = 16.5$ Hz), 140.21 (dd, $J = 26$ Hz, 13 Hz), 139.01, 138.85, 138.59, 138.34, 136.23, 135.23 (d, $J = 30$ Hz), 132.83, 132.75, 132.45, 132.41, 132.38, 132.18, 132.15, 132.12, 129.62, 129.29, 129.16, 129.03, 128.12, 128.00, 127.91, 127.86, 127.79, 127.77, 124.49 (d, $J = 13$ Hz), 118.59, 117.89, 109.20, 108.39, 48.07 (d, $J = 27$ Hz), 45.83 (d, $J = 36$ Hz). ^{31}P NMR (162 MHz, C_6D_6): $\delta_{\text{P}} = 32.398$ (dd, 2P, $J = 11.7$, 6.6 Hz), 23.768 (dd, 2P, $J = 11.4$, 6.3 Hz). ^{29}Si NMR (79.5 MHz, C_6D_6): $\delta_{\text{Si}} = 55.99$ (dddd, $J = 60.1$, 43.2, 16.2, 5.2 Hz, 2Si). Anal. Found: C, 65.91; H, 5.31; N, 4.07%. Calcd. for $\text{C}_{76}\text{H}_{66}\text{N}_4\text{P}_4\text{Si}_2\text{Ni}_2$: C, 68.49; H, 4.99; N, 4.20%. Satisfactory elemental data not acquired due to the compound undergoing rapid oxidation upon exposure to air. Calcd. for $\text{C}_{76}\text{H}_{66}\text{N}_4\text{O}_4\text{P}_4\text{Si}_2\text{Ni}_2$: C, 65.35; H, 4.76; N, 4.01%. ESI-MS (low-res., +ve ion, MeCN): $m/z = 1273.32$ $[(\text{C}_{38}\text{H}_{33}\text{N}_2\text{P}_2\text{Si})_2\text{Ni} + \text{H}]^+$, 1332.24 $[\text{M} + \text{H}]^+$. Crystals of a benzene solvate **5**·(C_6H_6)_{1.5} suitable for X-ray diffraction were grown by slow diffusion of hexane vapour into benzene (CAUTION) solution of compound **5**. *Crystal data for 5*·(C_6H_6)_{1.5}: $\text{C}_{85}\text{H}_{75}\text{N}_4\text{Ni}_2\text{P}_4\text{Si}_2$, $M_w = 1450.06$ g mol^{-1} , dark red-brown prism, $0.210 \times 0.090 \times 0.085$ mm^3 , triclinic, space group $P-1$ (No. 2), $a = 12.6509(3)$, $b = 16.9249(6)$, $c = 17.6665(6)$ Å, $\alpha = 81.1890(12)$, $\beta = 77.4286(16)$, $\gamma = 73.2272(17)^\circ$, $V = 3518.3(2)$ Å³, $Z = 2$, $D_{\text{calcd}} = 1.369$ Mg m^{-3} , $F_000 = 1514$, Nonius KappaCCD, Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å, $\mu = 0.711$ mm^{-1} , $T = 200(1)$ K, $2\theta_{\text{max}} = 50.3^\circ$, 46219 reflections collected, 12476 unique ($R_{\text{int}} = 0.069$). $R_1 = 0.0438$, $wR_2 = 0.1071$, R indices based on 9240 reflections with $I > 2.0\sigma(I)$ (refinement on F^2), 874 parameters, 0 restraints. Lp and absorption corrections applied, CCDC 1517411.

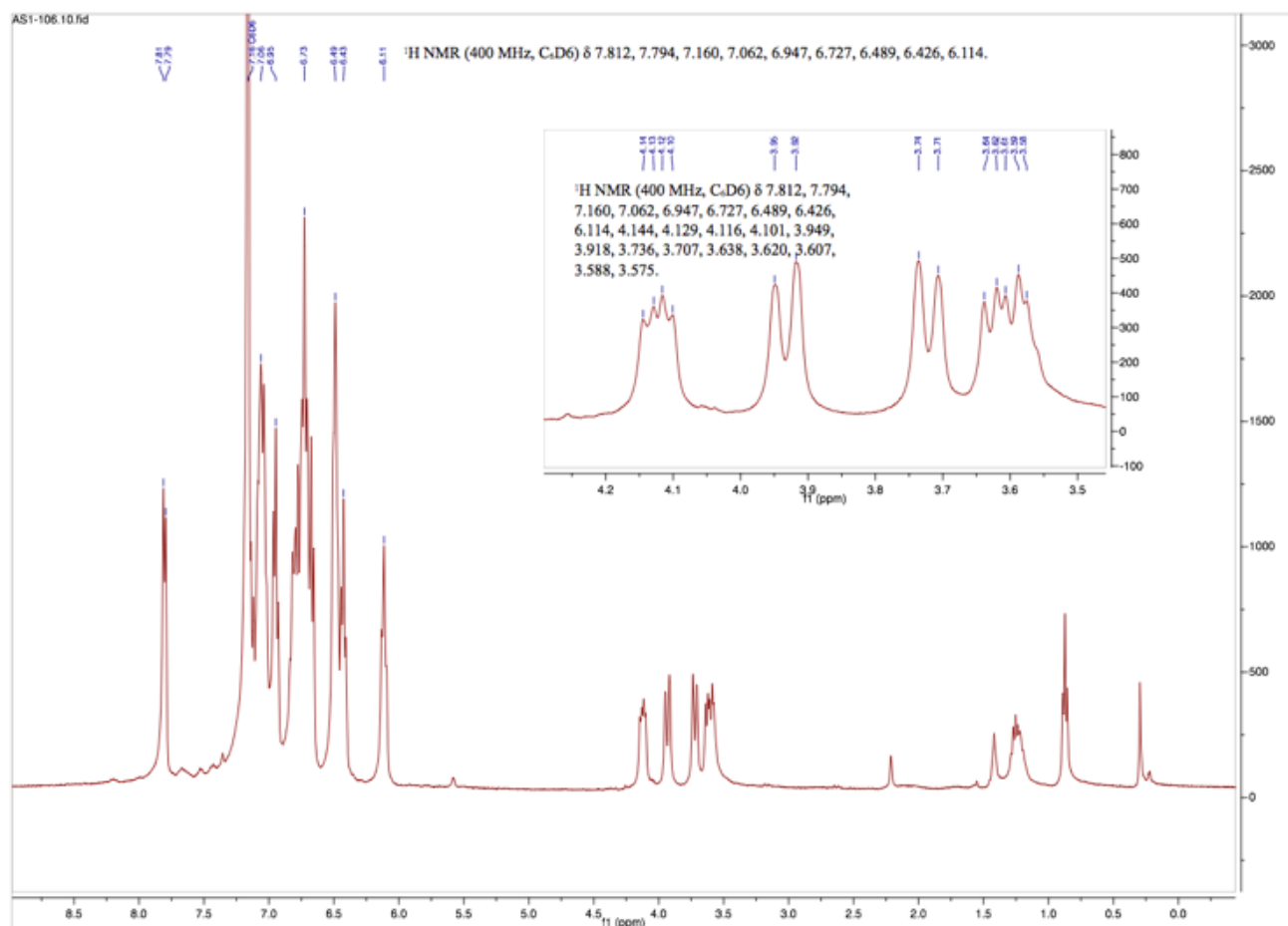


Figure S6. $^1\text{H NMR}$ spectrum of **5** (400 MHz, C_6D_6 , 298 K)

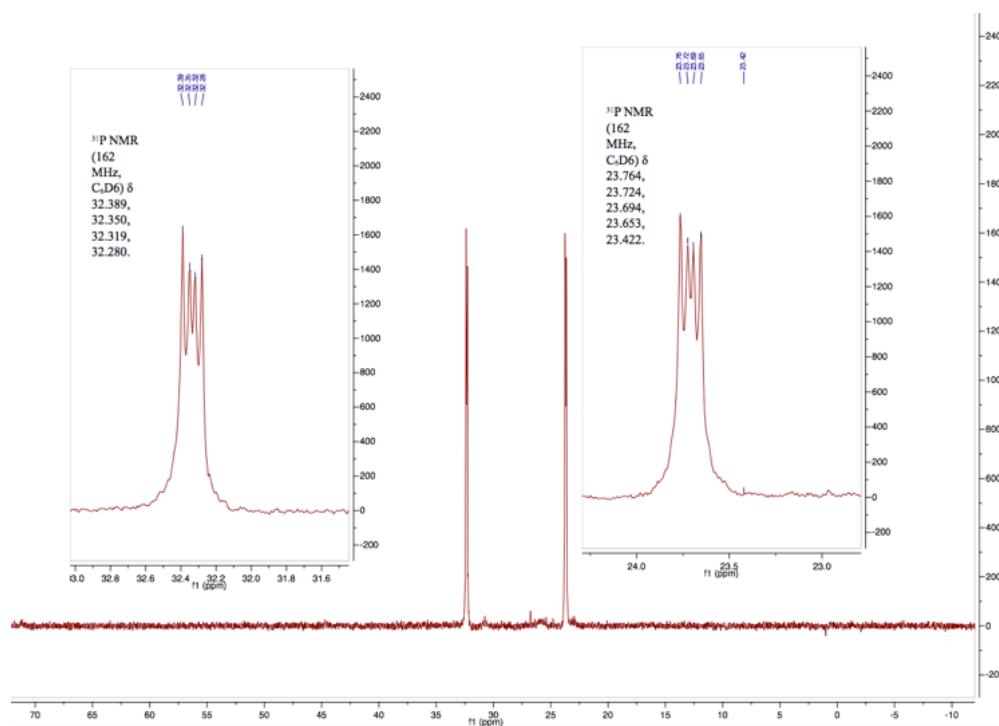
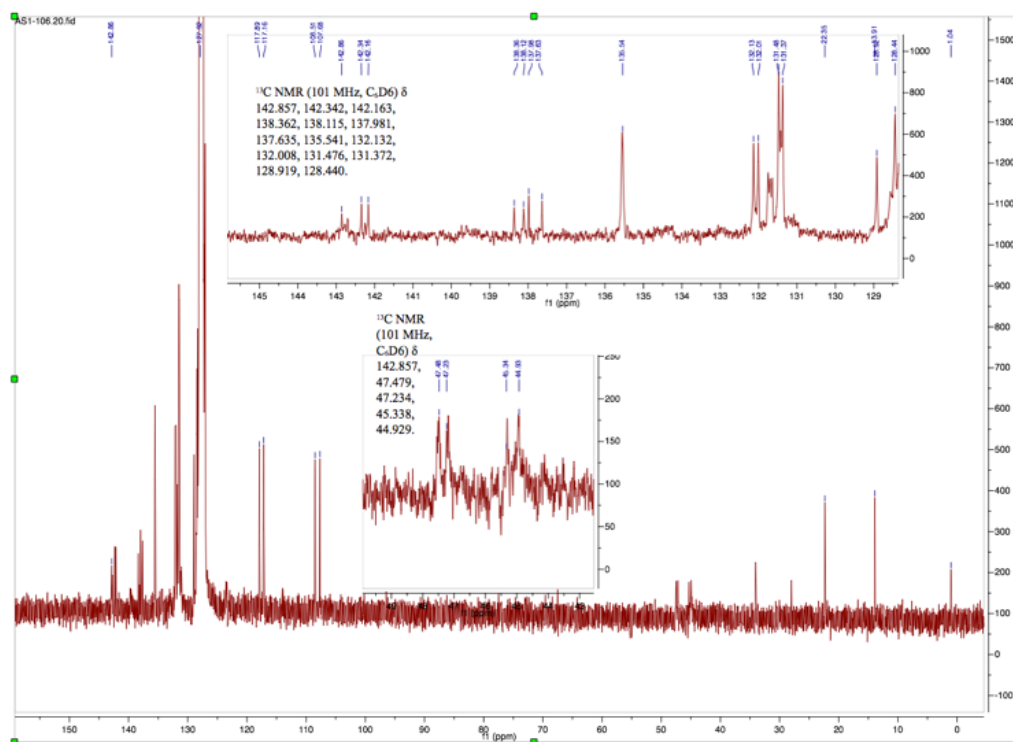
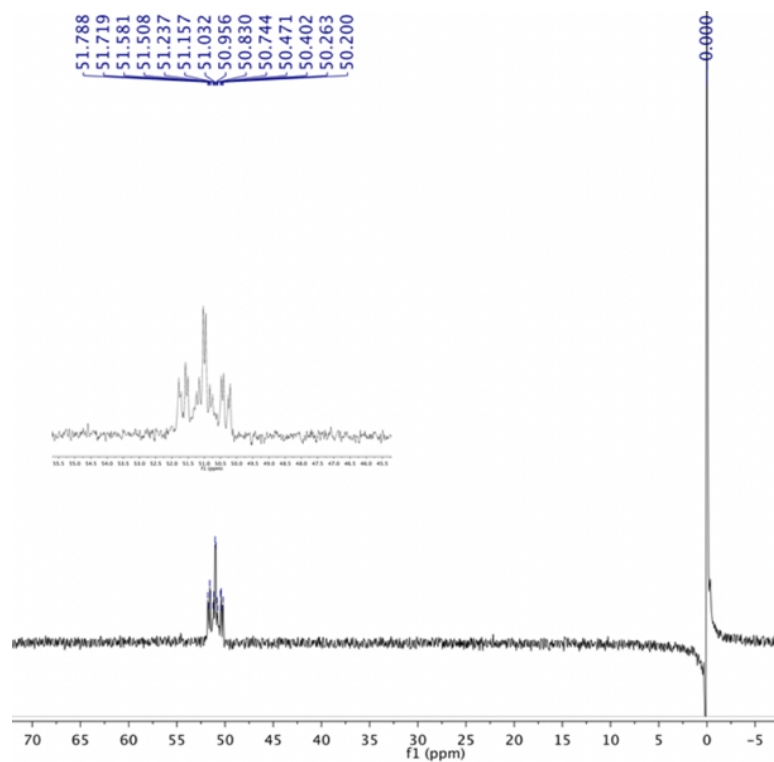


Figure S7. $^{31}\text{P}\{^1\text{H}\}$ NMR Spectrum of **5** (162 MHz, C_6D_6 , 298 K)

Figure S8. ¹³C{¹H} NMR Spectrum of **5** (101 MHz, C₆D₆, 298 K)Figure S9. ²⁹Si{¹H} NMR Spectrum of **5** (C₆D₆, 298 K)

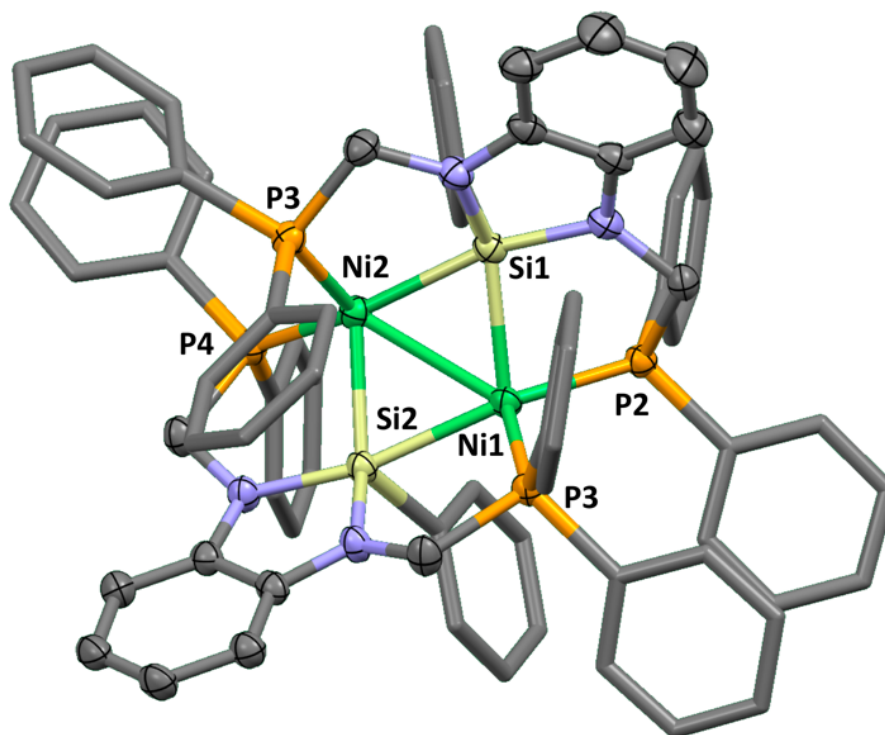


Figure S10. Molecular structure of **5** in a crystal of $5 \cdot (\text{C}_6\text{H}_6)_{1.5}$ (50% displacement ellipsoids, hydrogen atoms and solvent omitted, phenyl groups simplified). Selected bond lengths (Å) and angles (°): Ni1–Ni2 2.6020(5), Ni1–P1 2.2358(8), Ni1–P2 2.2189(9), Ni1–Si1 2.2906(9), Ni1–Si2 2.3131(9), Ni2–P3 2.2015(9), Ni2–P4 2.2013(9), Ni2–Si1 2.3213(9), Ni2–Si2 2.2937(9), P1–Ni1–Ni2 124.16(3), P1–Ni1–Si1 117.63(3), P1–Ni1–Si2 87.40(3), P2–Ni1–Ni2 123.61(3), P2–Ni1–P1 107.81(3), P2–Ni1–Si1 82.51(3), P2–Ni1–Si2 154.45(4), Si1–Ni1–Ni2 56.21(2), Si1–Ni1–Si2 109.02(3), Si2–Ni1–Ni2 55.26(2), P3–Ni2–Ni1 118.99(3), P3–Ni2–Si1 85.54(3), P3–Ni2–Si2 113.33(3), P4–Ni2–Ni1 126.52(3), P4–Ni2–P3 108.37(4), P4–Ni2–Si1 156.08(3), P4–Ni2–Si2 84.18(3), Si1–Ni2–Ni1 55.10(2), Si2–Ni2–Ni1 55.96(2), Si2–Ni2–Si1 108.62(3).

Synthesis of $[\text{Pt}_2\{\text{SiPh}(\text{NCH}_2\text{PPh}_2)_2\text{C}_6\text{H}_4\}_2]$ (**6**)

The pro-ligand $\text{HSiPh}(\text{CH}_2\text{PPh}_2)_2\text{C}_6\text{H}_4$ (**3**; 64 mg, 0.105 mmol) was added to a thf (10 mL) solution of $[\text{Pt}(\text{nbe})_3]$ (50 mg, 0.105 mmol). The mixture was then stirred at room temperature for 6 hours. The resulting orange solution was concentrated and pentane was added to induce precipitation. The yellow-orange precipitate was collected by filtration, washed with pentane and dried under vacuo. Yield: 48 mg (92%). ^1H NMR (600 MHz, C_6D_6 , 298 K): $\delta_{\text{H}} = 7.88$ (dd, 4 H, $^3,^4J_{\text{HH}} = 7.8, 1.2$ Hz, C_6H_4), 7.64 (t, 4 H, $^3J_{\text{HH}} = 9$ Hz), 7.05–6.98 (m, 8 H, C_6H_5), 6.93–6.85 (m, 8 H, C_6H_5), 6.81 (t, 6 H, $J = 7.8$ Hz, C_6H_5), 6.77–6.71 (m, 14 H, C_6H_5), 6.64 (td, 2 H, $^3,^4J_{\text{HH}} = 7.5, 1.8$ Hz), 6.59–6.56 (m, 6 H, C_6H_5), 6.49–6.45 (m, 6 H, C_6H_5), 4.63–4.53 (m, 2 H, PCH_2), 4.53–4.49 (dd, 2 H, $J = 12, 10.2$ Hz, PCH_2), 4.42–4.39 (dd, 2 H, $J = 12, 6$ Hz, PCH_2), 3.28–3.24 (m, 2H). $^{13}\text{C}\{^1\text{H}\}$ NMR (150 MHz, C_6D_6 , 298 K): $\delta_{\text{C}} = 147.4, 142.9$ (d, $J = 15$ Hz), 142.7, 140.1 (d, $J = 36$ Hz), 138.4 (t, $J = 18$ Hz), 135.0 (d, $J = 15$ Hz), 134.4 (t, $J = 9$ Hz), 132.7, 132.6, 132.6, 132.5, 130.0 (d, $J = 4.5$ Hz), 129.3 (d, $J = 4.5$ Hz), 128.8 (t, $J = 1.5$ Hz), 128.7, 128.4, 128.3, 128.13, 128.10, 128.07, 127.9, 127.8 (d, $J = 10.5$ Hz), 127.6, 127.4, 118.0, 117.1, 109.0, 108.5, 51.55 (m), 46.40 (d, $J = 36$ Hz). $^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, C_6D_6 , 298 K): $\delta_{(\text{P}_{\text{A,A}'})} = 68.39$ ($^1J(\text{PtP}_{\text{A}}) = 3500$ Hz, $^2J(\text{PtP}_{\text{A}}) = 522$ Hz, $^3J(\text{P}_{\text{A}}\text{P}_{\text{A}'}) = 101$ Hz), 43.83 ($^1J(\text{PtP}_{\text{B}}) = 2224$ Hz, $^2J(\text{P}_{\text{B}}\text{P}_{\text{B}'})$ and $^2J(\text{P}_{\text{A}}\text{P}_{\text{B}})$ not resolved), See Figure S12 for *g*NMR simulation. ^{29}Si NMR (79.5 MHz, C_6D_6 , 298 K): $\delta_{\text{Si}} = 60.54$ (t, $^2J_{\text{SiP}} = 4.9$ Hz), 58.44 (t, $^2J_{\text{SiP}} = 5.6$ Hz). Anal. Found: C, 56.86; H, 4.18; N, 3.55%. Calcd. for $\text{C}_{76}\text{H}_{66}\text{N}_4\text{P}_4\text{Si}_2\text{Pt}_2$: C, 56.85; H, 4.14; N, 3.49%. ESI-MS (low-res, +ve ion, MeCN): $m/z = 1605.31$ [M] $^+$, 803.66 [($\text{M}/2$)+ H] $^+$. Crystals of a benzene solvate $6 \cdot (\text{C}_6\text{H}_6)_2$ suitable for X-ray diffraction were grown by slow diffusion of hexane vapour into benzene (CAUTION) solution of compound **6**. *Crystal data for 6* (C_6H_6): $\text{C}_{88}\text{H}_{78}\text{N}_4\text{P}_4\text{Pt}_2\text{Si}_2$, $M_w = 1761.78$ g mol^{-1} , orange plate, 0.120 x 0.110 x 0.080 mm^3 , triclinic, space group *P*-1 (No. 2), $a = 12.9341(2)$, $b = 15.4591(3)$, $c = 20.0358(3)$ Å, $\alpha = 70.8529(9)$, $\beta = 79.0293(10)$, $\gamma = 88.9368(10)^\circ$, $V = 3711.21(11)$ Å 3 , $Z = 2$, $D_{\text{calcd}} = 1.577$ M gm^{-3} , $F_{000} = 1756$, Nonius KappaCCD, Mo $\text{K}\alpha$ radiation, $\lambda = 0.71073$ Å, $\mu = 3.934$ mm^{-1} , $T = 200(1)$ K, $2\theta_{\text{max}} = 55.0^\circ$, 84751 reflections collected, 16970 unique ($R_{\text{int}} = 0.085$), $R_1 = 0.0368$, $wR_2 = 0.0875$, R indices based on 13109 reflections with $I > 2.0\sigma(I)$ (refinement on F^2), 847 parameters, 0 restraints. Lp and absorption corrections applied, CCDC 1517308.

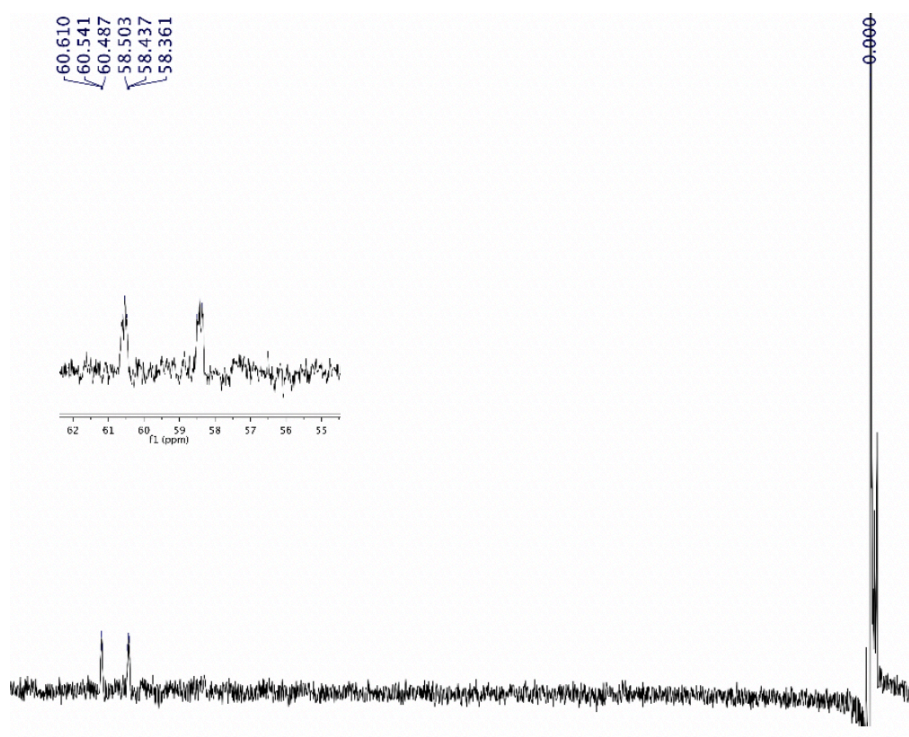


Figure S11. $^{29}\text{Si}\{^1\text{H}\}$ NMR Spectrum of **6** (C_6D_6 , 298 K)

Supporting Information

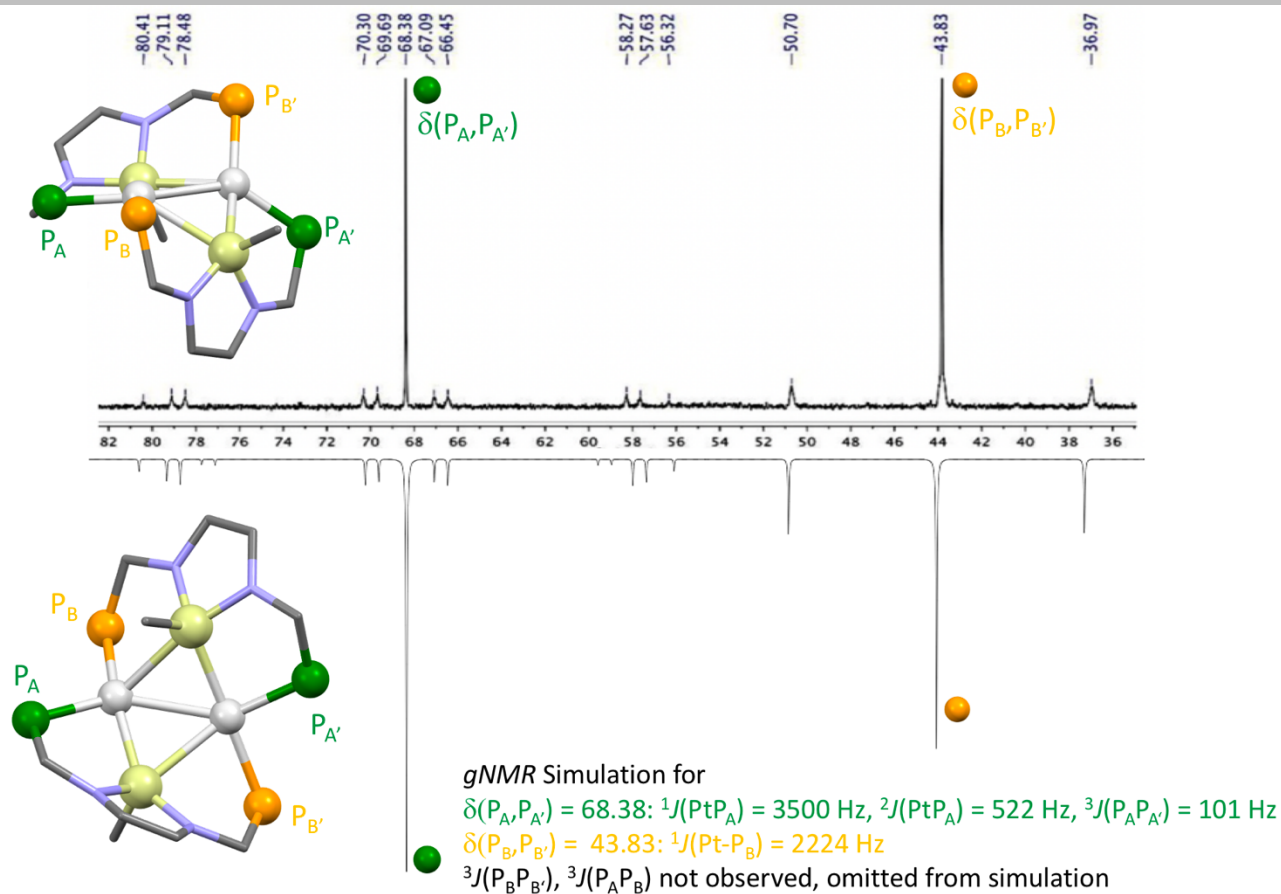


Figure S12. $^{31}\text{P}\{^1\text{H}\}$ NMR Spectrum of **6** (C_6D_6 , 162 MHz, 298 K) (Simulated using gNMR: P. H. M. Budzelaar, gNMR, version 5.1, <http://www.home.cc.umanitoba.ca/~budzelaar/gNMR/gNMR.html>).

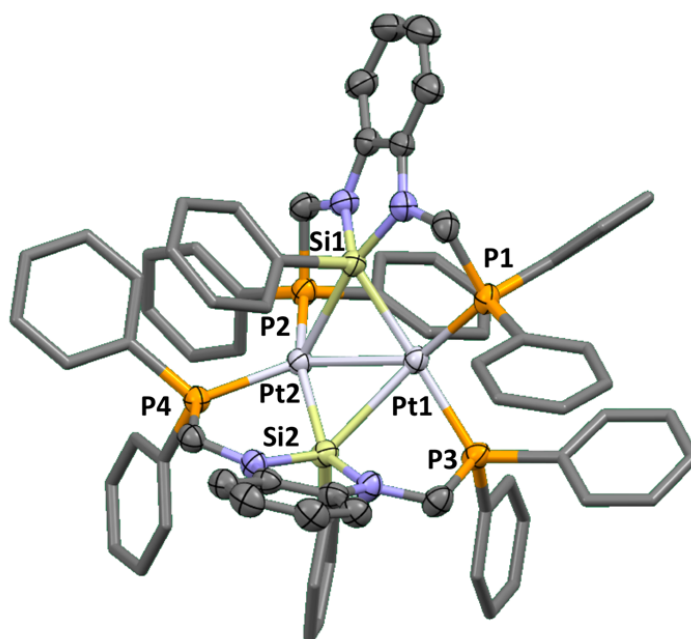


Figure S13. Molecular structure of **6** in a crystal of **6**.(C_6H_6)₂ (50% displacement ellipsoids, hydrogen atoms and solvent omitted, phenyl groups simplified). Selected bond lengths (Å) and angles (°): Pt1–P1 2.2841(12), Pt1–P3 2.3451(12), Pt1–Si1 2.3821(12), Pt1–Si2 2.7133(13), Pt2–P2 2.2805(11), Pt2–P4 2.2836(11), Pt2–Si1 2.6421(12), Pt2–Si2 2.3782(12), P1–Pt1–Pt2 145.24(3), P1–Pt1–P3 96.95(4), P1–Pt1–Si1 83.30(4), P3–Pt1–Pt2 117.74(3), P3–Pt1–Si1 178.21(4), P3–Pt1–Si2 76.12(4), Si1–Pt1–Pt2 61.96(3), Si1–Pt1–Si2 102.67(4), Si2–Pt1–Pt2 51.91(3), P2–Pt2–Pt1 107.93(3), P2–Pt2–P4 111.18(4), P2–Pt2–Si1 81.96(4), P2–Pt2–Si2 158.40(4), P4–Pt2–Pt1 140.19(3), P4–Pt2–Si1 126.31(4), P4–Pt2–Si2 81.62(4), Si1–Pt2–Pt1 52.73(3), Si2–Pt2–Pt1 63.90(3), Si2–Pt2–Si1 104.92(4).

(a) Pd₂ (Compound **4**)

(b) Ni₂ (Compound **5**)

(c) Pt₂ (Compound **6**)

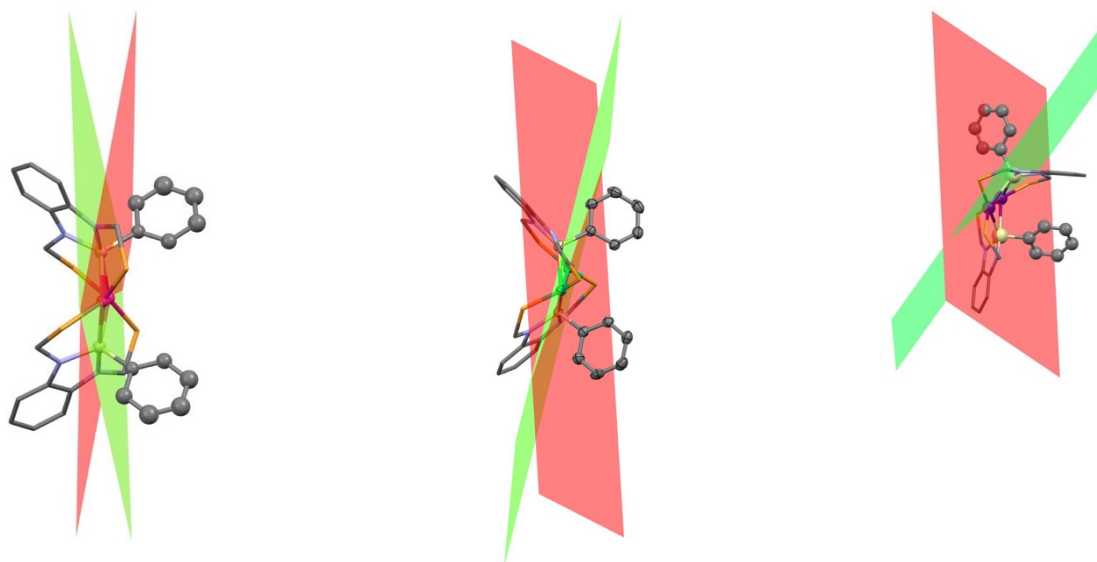
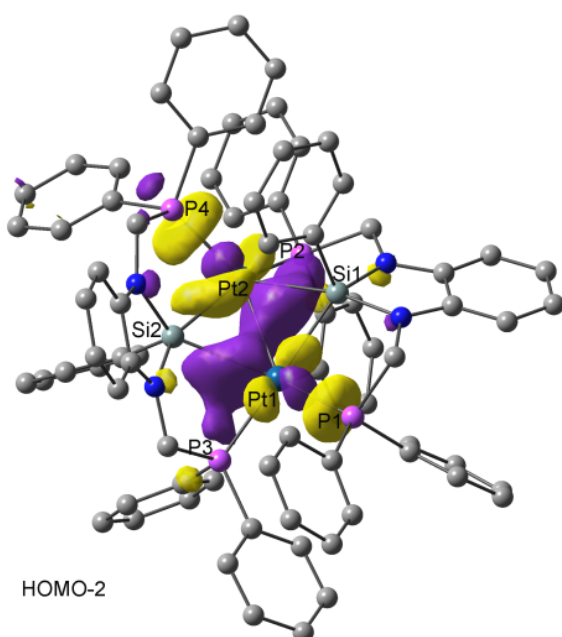


Figure S14. Folding of two M_2Si planes across M-M vector. (a) M = Pd; (b) M = Ni; (c) M = Pt.

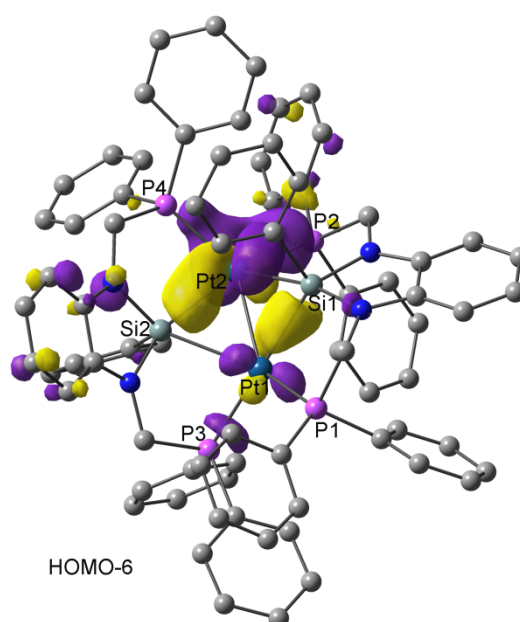
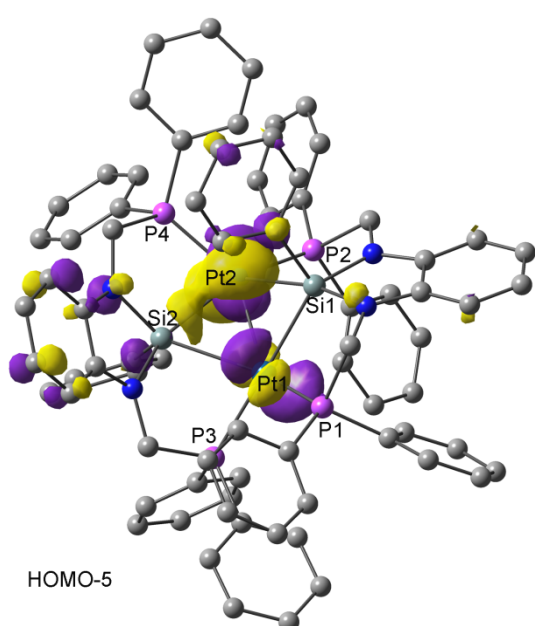
Computational Details

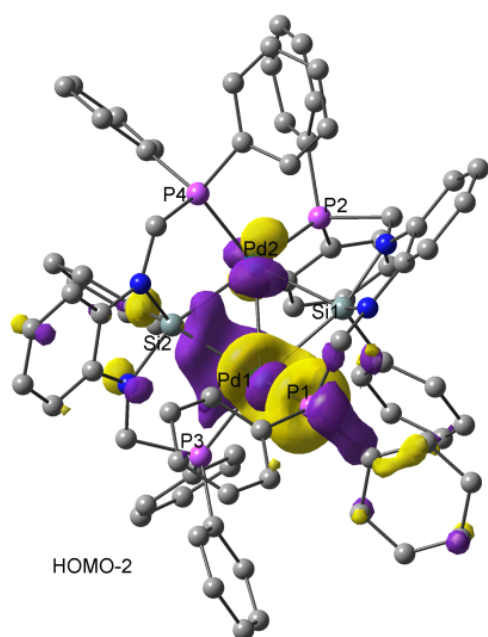
All computational work was performed by using the Gaussian 09 suite of programs.⁵ The geometries of complexes **11-13** were optimized at the DFT level of theory using the exchange functional of Becke⁶ in conjunction with the correlation functional of Perdew^{7,8} (BP86). The Stuttgart basis set in combination with the 60-core-electron relativistic effective core potential (SDD)⁹ was used for Ni, Pd and Pt; 6-31G(d)¹⁰ basis sets were used for all other atoms. Frequency calculations were performed to confirm that optimized structures were minimal or saddle points at the BP86 level of theory. Single point energy calculations for all the optimized structures were carried out with a larger basis set viz the quadruple- ζ valence def2-QZVP¹¹ basis set on Ni, Pd and Pt along with the corresponding ECP and the 6-311+G(2d,p) basis set on other atoms. Single point calculations were also carried out at the M06, B97D and B3LYP level of theories, and found to follow similar trends. Natural bond orbital (NBO) analyses were carried out at the BP86 level of theory.

[Pt₂{SiPh(NCH₂PPh₂)₂C₆H₄}₂] (6)

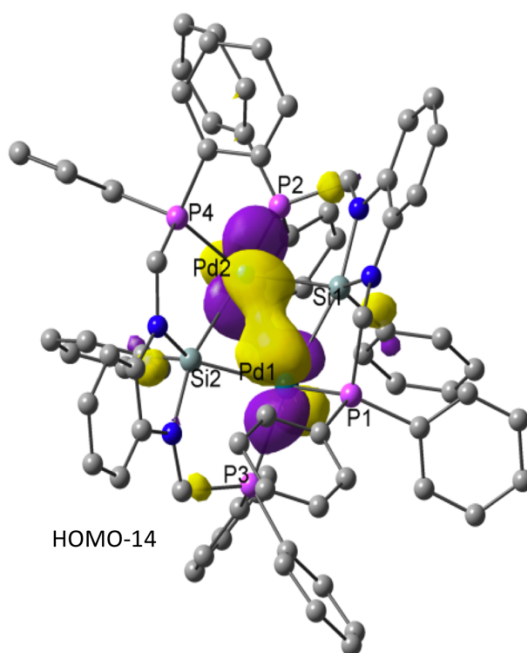
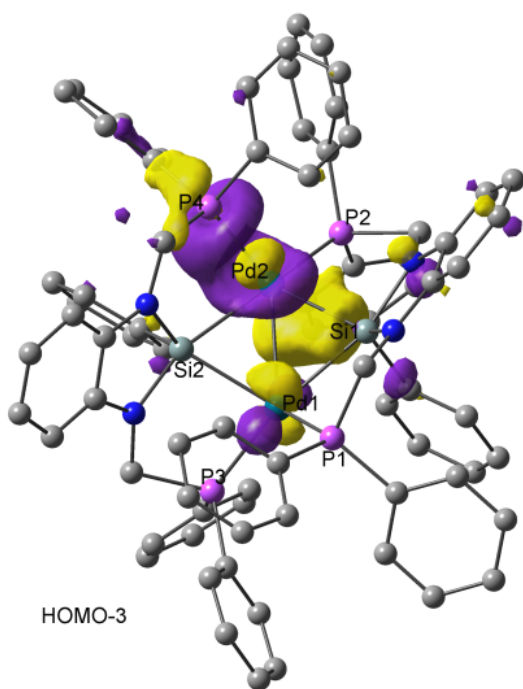


For this complex, NBO analysis indicates that both the Pt-centers have five lone pairs with an electronic configuration of [Xe]5d⁹5s^{0.5}6p^{0.5} and natural charges of -0.5. The Wiberg bond order and HOMO-2 closely represent a conventional Pt-Pt bond. NBO second order perturbation theory analysis (donor-acceptor) suggests that this bonding interaction comprises primarily the Pt2-P2 bond electron pair to the antibonding Pt1-P3 orbital, but hardly any contribution is found from either of the Si atoms. Therefore, this corresponds a 4-centered-2-electron type of bonding as described by Nova and Nazari which here involves P-Pt-Pt-P centers rather than Si-Pt-Pt-Si. NBO analysis shows that each of the Pt-centers have three different types of interactions with both the Si atoms. The almost degenerate HOMO-5 and HOMO-6 (332-331) orbitals reflect the polar-covalent bonding interaction which primarily comprises σ -back donation from a Pt-d lone pair to an orbital on the Si atom, while the other Si atom donates electron density to the of Pt-d orbital.



[Pd₂{SiPh(NCH₂PPh₂)₂C₆H₄}₂] (4)

In contrast to the observation of Nova and Nazari, NBO-NLMO analysis of the Pd₂-complex did not show any obvious 4c-2e bonding. Both the Pd-centers interact in two different ways to each of the Si-atoms. The almost degenerate HOMO-2 and HOMO-3 sets of orbitals reflect the bonding interaction due to the Pd-P bonding orbital interaction with the Si-*p* orbital, while HOMO-4 represent back donation from a Pd-*d* orbital to the Si-*p* orbital. The metal-metal bonding orbital is very low in energy and can be seen only in the HOMO-14. Each Pd^I centre is thus a d⁹ system due to this interaction).

**[Ni₂{SiPh(NCH₂PPh₂)₂C₆H₄}₂] (5)**

Each nickel has five lone pairs, corresponding to HOMO, HOMO-4, HOMO-6, HOMO-7 and HOMO-8 the with electronic configuration of 3d^{9.4}4p^{0.7}5s^{0.3} and natural charges of -0.5. Each Ni-center is found to form covalent bonds only with the P-atoms of ligands while the Ni-Ni bond (HOMO-3; Ni-Ni bonding with from the Ni-P bonding to the Ni-*d* orbital) or Ni-Si bonds involve extended delocalization (HOMO-5 represent Ni to Si σ-back donation; (326) and HOMO-11 involves Si to Ni σ-donation). The primary component of the Ni-Ni σ-bond arises from HOMO-12.

Cartesian Coordinates for [Pd₂{SiPh(NCH₂PPh₂)₂C₆H₄}₂] (4)

Optimised structure at the MO6L level of theory (SDD basis set for palladium with SDD core potential and 6-31g(d) for remaining elements)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	46	0	1.373476	-0.032203	0.131778
2	46	0	-1.403038	0.013830	0.178978
3	15	0	2.664128	-1.105069	-1.560233
4	15	0	-2.684533	1.349636	-1.350026
5	15	0	-2.859471	-1.594695	1.169458
6	15	0	2.835966	1.371394	1.402657
7	14	0	-0.128464	-1.956267	-0.434376
8	14	0	0.092867	2.012456	-0.064434
9	7	0	0.395718	-2.465655	-2.076474
10	7	0	-0.447666	2.783746	-1.601597
11	7	0	1.544598	3.099615	-0.179495
12	6	0	0.317261	4.442106	-3.306047
13	6	0	-0.354764	-3.839005	-4.015924
14	6	0	-0.474979	-3.331897	-2.726466
15	6	0	0.428067	3.740574	-2.110053
16	6	0	2.737180	1.253329	-2.993542
17	6	0	2.504319	4.911924	-1.605549
18	6	0	-3.562990	0.307513	-2.565120
19	7	0	-1.593705	-2.999947	-0.722042
20	6	0	-2.030290	-1.104868	3.784395
21	6	0	2.155435	0.253416	3.851254
22	6	0	-3.200257	-3.192092	3.485482
23	6	0	-2.698947	-2.001192	2.945432
24	6	0	-1.273404	2.586387	3.843911
25	6	0	1.285748	5.398884	-3.642215
26	6	0	1.551840	3.954487	-1.261707
27	6	0	0.788268	-2.640858	2.215703
28	6	0	3.294925	2.255354	-3.779562
29	6	0	-0.793895	2.726164	1.453654
30	6	0	3.482598	0.116289	-2.646266
31	6	0	3.034599	2.493421	4.007468
32	6	0	4.638436	1.173907	1.155817
33	6	0	-1.597416	-3.678199	-1.924855
34	6	0	-1.857384	-1.391686	5.135753
35	6	0	5.360347	1.010561	-3.881237
36	6	0	-2.541034	-4.579102	-2.413908
37	6	0	4.801715	0.004498	-3.094955
38	6	0	4.520204	-3.226586	-2.033721
39	6	0	-2.589579	-3.216191	0.286125
40	6	0	-0.690211	2.068459	2.688576
41	6	0	2.680694	1.388712	3.224014
42	6	0	-2.390786	-5.110167	-3.701102
43	6	0	2.361758	5.634648	-2.795506
44	6	0	4.612398	2.134597	-4.221878
45	6	0	2.835895	2.471265	5.385131
46	6	0	-5.410522	-0.782542	2.046068
47	6	0	-1.480864	2.233136	-2.438575
48	6	0	1.446673	-3.409862	3.174393
49	6	0	1.549887	-1.967927	-2.772827
50	6	0	0.821569	-2.978989	0.855083
51	6	0	1.971882	0.224761	5.230349
52	6	0	-4.662679	-1.330789	0.994959
53	6	0	-1.315914	-4.737904	-4.499306
54	6	0	-2.341252	-2.588168	5.659188
55	6	0	5.456016	0.553354	2.108778
56	6	0	5.942611	-4.107964	-0.294192
57	6	0	3.957896	-2.317835	-1.128297

Supporting Information

58	6	0	-2.849747	-0.774880	-3.103258
59	6	0	2.163800	-4.538563	2.782782
60	6	0	5.194444	1.544869	-0.076814
61	6	0	-3.449039	-1.641984	-4.010140
62	6	0	-3.013367	-3.487058	4.832527
63	6	0	2.539343	3.132475	0.853341
64	6	0	-2.001081	3.771724	3.777391
65	6	0	-5.675093	4.551466	0.126328
66	6	0	6.800541	0.317313	1.836526
67	6	0	-5.286012	-1.535593	-0.245026
68	6	0	7.346936	0.697334	0.613474
69	6	0	2.302327	1.339451	5.998341
70	6	0	-1.507111	3.934907	1.418240
71	6	0	-4.891373	0.506080	-2.954100
72	6	0	-5.862967	3.200227	0.412807
73	6	0	-2.118157	4.444337	2.560122
74	6	0	2.203659	-4.899315	1.435596
75	6	0	-4.588515	4.946923	-0.648479
76	6	0	-4.778726	-1.443035	-4.381406
77	6	0	-7.362972	-0.665070	0.629559
78	6	0	-3.693945	3.999846	-1.141696
79	6	0	1.537179	-4.130326	0.486251
80	6	0	-3.874420	2.640501	-0.861291
81	6	0	-6.627116	-1.207923	-0.421806
82	6	0	5.500951	-4.121354	-1.616545
83	6	0	-4.964282	2.254948	-0.066704
84	6	0	6.540346	1.313543	-0.341070
85	6	0	5.394219	-3.205625	0.612915
86	6	0	4.398135	-2.327127	0.200101
87	6	0	-5.493074	-0.368836	-3.856909
88	6	0	-6.750888	-0.456319	1.864366
89	1	0	5.402145	-0.859389	-2.812145
90	1	0	6.390717	0.915601	-4.222852
91	1	0	5.056486	2.923510	-4.826947
92	1	0	2.705606	3.137879	-4.025233
93	1	0	1.717359	1.363946	-2.614423
94	1	0	3.926605	-1.650482	0.916045
95	1	0	5.722829	-3.199335	1.651118
96	1	0	6.711898	-4.807806	0.028707
97	1	0	5.925781	-4.828460	-2.327073
98	1	0	4.196617	-3.232396	-3.075085
99	1	0	4.565154	2.004026	-0.842487
100	1	0	6.954418	1.611486	-1.303063
101	1	0	8.398843	0.511901	0.403292
102	1	0	7.425098	-0.164205	2.587883
103	1	0	5.038600	0.252606	3.069891
104	1	0	3.469731	3.377491	3.541335
105	1	0	3.102385	3.340961	5.983118
106	1	0	2.147970	1.326326	7.076134
107	1	0	1.561998	-0.667306	5.703032
108	1	0	1.879151	-0.607552	3.240724
109	1	0	1.576801	-4.420447	-0.566430
110	1	0	2.764165	-5.780977	1.124959
111	1	0	2.683756	-5.141993	3.526069
112	1	0	1.396593	-3.129251	4.227456
113	1	0	0.221703	-1.756767	2.520129
114	1	0	-1.581884	4.484824	0.478995
115	1	0	-2.682169	5.375419	2.504769
116	1	0	-2.468173	4.178839	4.673443
117	1	0	-1.159242	2.064276	4.795460
118	1	0	-0.121499	1.135873	2.731780
119	1	0	-5.461671	1.338776	-2.543609
120	1	0	-6.529849	-0.206263	-4.150309
121	1	0	-5.255558	-2.127391	-5.081302
122	1	0	-2.878471	-2.480644	-4.407823
123	1	0	-1.816001	-0.951672	-2.794710
124	1	0	-5.107656	1.200579	0.172912
125	1	0	-6.707576	2.878061	1.020271

Supporting Information

126	1	0	-6.371415	5.294567	0.511267
127	1	0	-4.429115	6.001344	-0.869673
128	1	0	-2.836917	4.334689	-1.725309
129	1	0	-3.745814	-3.891022	2.850721
130	1	0	-3.397297	-4.420801	5.239847
131	1	0	-2.198387	-2.821432	6.713184
132	1	0	-1.339068	-0.680067	5.777777
133	1	0	-1.636843	-0.180119	3.361047
134	1	0	-4.718315	-1.940388	-1.085569
135	1	0	-7.092080	-1.370362	-1.392893
136	1	0	-8.410879	-0.406300	0.486906
137	1	0	-7.319254	-0.035046	2.692515
138	1	0	-4.936867	-0.611038	3.012593
139	1	0	-3.388497	-4.877184	-1.798386
140	1	0	-3.127005	-5.820716	-4.072980
141	1	0	-1.206830	-5.150680	-5.500520
142	1	0	0.490118	-3.550732	-4.639811
143	1	0	3.355065	5.100955	-0.952521
144	1	0	3.106000	6.386233	-3.053733
145	1	0	1.183618	5.959399	-4.569653
146	1	0	-0.522339	4.258529	-3.974846
147	1	0	-2.282142	-3.978815	1.024433
148	1	0	-3.539796	-3.549845	-0.153238
149	1	0	-1.085859	1.503389	-3.171365
150	1	0	-1.995792	3.011609	-3.021056
151	1	0	1.279082	-1.241541	-3.561963
152	1	0	2.089591	-2.789957	-3.271319
153	1	0	2.213985	3.730859	1.723116
154	1	0	3.484342	3.558487	0.490508

Cartesian Coordinates for $[\text{Ni}_2\{\text{SiPh}(\text{NCH}_2\text{PPh}_2)_2\text{C}_6\text{H}_4\}_2]$ (5)

Optimised structure at the MO6L level of theory (SDD basis set for nickel with SDD core potential and 6-31g(d) for remaining elements)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.041181	0.102229	-0.028176
2	6	0	0.010670	0.106424	1.374271
3	6	0	1.185309	0.163050	2.114313
4	6	0	2.418759	0.216137	1.468657
5	6	0	2.463591	0.233853	0.078164
6	6	0	1.285888	0.185578	-0.663702
7	6	0	-2.461253	-1.141102	0.055924
8	6	0	-3.707689	-0.844465	0.602129
9	6	0	-4.401486	-1.784641	1.364074
10	6	0	-3.844194	-3.041819	1.579547
11	6	0	-2.593160	-3.350901	1.041271
12	6	0	-1.903577	-2.406760	0.289686
13	6	0	-1.108242	-0.944486	-2.447950
14	6	0	-1.344466	-0.871377	-4.861433
15	6	0	-0.736021	-2.096792	-5.119052
16	6	0	-0.739317	-2.605167	-6.424063
17	6	0	-1.358429	-1.906037	-7.453956
18	6	0	-1.967525	-0.667645	-7.205349
19	6	0	-1.936093	-0.136601	-5.921062
20	6	0	-3.443578	1.735399	-6.330372
21	6	0	-5.607322	3.672062	-6.228298
22	6	0	-6.525238	4.478586	-5.546940
23	6	0	-7.639892	4.992202	-6.200357
24	6	0	-7.869735	4.674779	-7.536572
25	6	0	-6.977372	3.847860	-8.215421
26	6	0	-5.844229	3.358110	-7.571248
27	6	0	-2.950072	4.476631	-5.687388
28	6	0	-3.325422	5.803530	-5.445830

Supporting Information

29	6	0	-2.414872	6.840822	-5.622325
30	6	0	-1.117703	6.567925	-6.052037
31	6	0	-0.738817	5.252303	-6.304963
32	6	0	-1.644085	4.212455	-6.118584
33	6	0	-0.897943	2.565995	-3.340630
34	6	0	-1.110178	3.922617	-3.031741
35	6	0	-0.049206	4.813545	-2.869132
36	6	0	1.262239	4.363313	-2.989230
37	6	0	1.505407	3.019120	-3.279946
38	6	0	0.441353	2.139783	-3.452519
39	6	0	-0.822582	3.400628	1.132102
40	6	0	0.331686	3.289802	0.350068
41	6	0	1.588362	3.436588	0.927446
42	6	0	1.702509	3.667965	2.295627
43	6	0	0.557514	3.762454	3.085075
44	6	0	-0.701868	3.642948	2.505159
45	6	0	-2.852354	5.123969	0.277522
46	6	0	-1.905750	6.046396	-0.184624
47	6	0	-2.217794	7.399306	-0.272064
48	6	0	-3.476464	7.853420	0.118329
49	6	0	-4.418192	6.945723	0.595494
50	6	0	-4.113622	5.589861	0.668640
51	6	0	-3.637663	2.682880	1.517282
52	6	0	-5.827798	1.629492	1.401438
53	6	0	-5.981090	1.422804	2.767719
54	6	0	-7.037673	0.628172	3.229961
55	6	0	-7.919657	0.039225	2.331154
56	6	0	-7.756322	0.214806	0.950984
57	6	0	-6.703583	0.996226	0.481152
58	6	0	-7.111848	0.719277	-1.922334
59	6	0	-7.351955	0.767918	-4.733178
60	6	0	-7.534868	-0.231453	-5.695366
61	6	0	-8.537791	-0.111279	-6.655731
62	6	0	-9.381852	0.994709	-6.659966
63	6	0	-9.217401	1.988834	-5.697853
64	6	0	-8.204804	1.881633	-4.753743
65	6	0	-5.283899	-0.871965	-3.472658
66	6	0	-4.661110	-1.341870	-4.637659
67	6	0	-3.981368	-2.554072	-4.644618
68	6	0	-3.885465	-3.304753	-3.473797
69	6	0	-4.476289	-2.837184	-2.302819
70	6	0	-5.174602	-1.631938	-2.301441
71	6	0	-5.653405	3.908238	-1.755063
72	6	0	-4.865784	4.955459	-2.268699
73	6	0	-5.417674	6.187054	-2.624858
74	6	0	-6.787753	6.394602	-2.502894
75	6	0	-7.601971	5.361611	-2.029622
76	6	0	-7.042251	4.142408	-1.662487
77	7	0	-6.327252	1.218801	-0.831266
78	7	0	-4.824908	2.354878	0.766713
79	7	0	-2.508645	1.048195	-5.479956
80	7	0	-1.509257	-0.254614	-3.637470
81	15	0	-1.550472	0.095430	-0.957309
82	15	0	-2.436169	3.331793	0.280748
83	15	0	-6.047925	0.783718	-3.443662
84	15	0	-4.124945	3.112570	-5.318349
85	14	0	-4.889139	2.286293	-1.058047
86	14	0	-2.339493	1.341190	-3.687423
87	28	0	-2.641329	1.983137	-1.466325
88	28	0	-4.410174	2.255609	-3.312469
89	1	0	-0.947954	0.051788	1.892624
90	1	0	1.135908	0.165902	3.202265
91	1	0	3.339572	0.254441	2.048739
92	1	0	3.421250	0.286315	-0.438865
93	1	0	1.344754	0.219716	-1.750121
94	1	0	-4.146080	0.129649	0.405211
95	1	0	-5.375003	-1.522812	1.779759
96	1	0	-4.379930	-3.783352	2.170369

Supporting Information

97	1	0	-2.152398	-4.331871	1.213095
98	1	0	-0.919093	-2.650017	-0.112758
99	1	0	-0.025715	-1.158386	-2.446306
100	1	0	-1.621853	-1.918088	-2.364856
101	1	0	-0.289289	-2.665592	-4.304269
102	1	0	-0.260000	-3.561824	-6.625665
103	1	0	-1.366200	-2.312665	-8.463731
104	1	0	-2.453697	-0.119111	-8.011545
105	1	0	-2.971030	2.109691	-7.255876
106	1	0	-4.282702	1.080778	-6.640322
107	1	0	-6.367838	4.688358	-4.488676
108	1	0	-8.340283	5.622837	-5.652575
109	1	0	-8.749422	5.062256	-8.048626
110	1	0	-7.158350	3.589397	-9.257607
111	1	0	-5.142402	2.732791	-8.122829
112	1	0	-4.343600	6.025676	-5.125350
113	1	0	-2.723803	7.867763	-5.430321
114	1	0	-0.404150	7.378754	-6.189666
115	1	0	0.275402	5.027000	-6.631553
116	1	0	-1.315269	3.184942	-6.272285
117	1	0	-2.135960	4.290341	-2.980475
118	1	0	-0.253506	5.865456	-2.670038
119	1	0	2.093723	5.056668	-2.867171
120	1	0	2.530160	2.660456	-3.383495
121	1	0	0.640900	1.098688	-3.714570
122	1	0	0.237160	3.072083	-0.715338
123	1	0	2.478888	3.343332	0.307012
124	1	0	2.686459	3.767732	2.751812
125	1	0	0.645673	3.940627	4.155858
126	1	0	-1.594555	3.748246	3.122742
127	1	0	-0.911693	5.698223	-0.466103
128	1	0	-1.469780	8.103112	-0.635298
129	1	0	-3.720843	8.912517	0.053078
130	1	0	-5.407286	7.288637	0.896009
131	1	0	-4.880153	4.886967	0.994382
132	1	0	-3.204579	1.786305	2.001986
133	1	0	-3.841413	3.417733	2.315437
134	1	0	-5.289680	1.885596	3.470732
135	1	0	-7.168102	0.481352	4.300500
136	1	0	-8.747617	-0.566167	2.696140
137	1	0	-8.443754	-0.257588	0.250830
138	1	0	-7.462647	-0.311521	-1.745951
139	1	0	-8.012964	1.330409	-2.110283
140	1	0	-6.906180	-1.119278	-5.686957
141	1	0	-8.665505	-0.898289	-7.397489
142	1	0	-10.166123	1.083172	-7.410019
143	1	0	-9.867265	2.862649	-5.694823
144	1	0	-8.059904	2.681957	-4.024068
145	1	0	-4.690602	-0.749765	-5.552864
146	1	0	-3.493923	-2.889506	-5.559221
147	1	0	-3.331027	-4.242658	-3.472732
148	1	0	-4.385776	-3.398512	-1.372619
149	1	0	-5.612083	-1.279154	-1.368928
150	1	0	-3.787094	4.813984	-2.331849
151	1	0	-4.766790	6.987701	-2.976837
152	1	0	-7.224044	7.356842	-2.769116
153	1	0	-8.675778	5.518925	-1.928187
154	1	0	-7.687496	3.362615	-1.254297

Cartesian Coordinates for $[\text{Pt}_2\{\text{SiPh}(\text{NCH}_2\text{PPh}_2)_2\text{C}_6\text{H}_4\}_2]$ (6)

Optimised structure at the MO6L level of theory (SDD basis set for platinum with 60 electron SDD core potential and 6-31g(d) for remaining elements)

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z

Supporting Information

1	6	0	8.475641	2.393452	1.875566
2	6	0	7.798470	1.779814	0.827162
3	6	0	8.303833	0.609029	0.263832
4	6	0	9.482814	0.058851	0.758713
5	6	0	10.160243	0.673216	1.808725
6	6	0	9.667293	1.852671	2.373935
7	6	0	9.457903	2.085676	5.143282
8	6	0	8.326002	2.838495	5.473570
9	6	0	7.462215	2.419704	6.478725
10	6	0	7.735990	1.249266	7.182816
11	6	0	8.868248	0.501053	6.869809
12	6	0	9.721403	0.910223	5.849428
13	6	0	12.158651	1.929971	3.821930
14	6	0	14.425640	2.649228	4.353783
15	6	0	15.089243	1.738376	3.538549
16	6	0	16.489975	1.763336	3.473018
17	6	0	17.212850	2.688948	4.213020
18	6	0	16.554064	3.607722	5.042467
19	6	0	15.165779	3.588232	5.121747
20	6	0	14.928936	5.567612	6.518641
21	6	0	13.445605	6.213771	8.919393
22	6	0	13.722198	4.909497	9.345094
23	6	0	13.510058	4.543874	10.670901
24	6	0	13.022225	5.473708	11.585092
25	6	0	12.749486	6.775935	11.169485
26	6	0	12.955719	7.144768	9.844455
27	6	0	14.397313	8.305337	7.244692
28	6	0	15.602423	8.468831	7.942120
29	6	0	16.215939	9.714602	8.000407
30	6	0	15.628242	10.814957	7.375370
31	6	0	14.428740	10.663061	6.688992
32	6	0	13.821378	9.411948	6.621505
33	6	0	11.784595	3.428731	7.049928
34	6	0	12.369671	2.288543	7.622687
35	6	0	11.878372	1.737804	8.804443
36	6	0	10.782883	2.320375	9.441005
37	6	0	10.176633	3.443966	8.883100
38	6	0	10.672278	3.982617	7.699356
39	6	0	12.670837	3.405054	0.565376
40	6	0	13.298733	2.352919	-0.092321
41	6	0	12.611961	1.635408	-1.070842
42	6	0	11.295828	1.970438	-1.377884
43	6	0	10.663869	3.023516	-0.721764
44	6	0	11.348613	3.761646	0.248687
45	6	0	8.980458	5.544828	0.648032
46	6	0	8.639774	5.656665	-0.706228
47	6	0	7.378338	6.109496	-1.079503
48	6	0	6.454299	6.488075	-0.107226
49	6	0	6.792280	6.408983	1.240790
50	6	0	8.044782	5.932049	1.614825
51	6	0	11.551885	6.664510	0.598990
52	6	0	10.403943	8.768076	1.249080
53	6	0	9.933485	9.197730	0.011665
54	6	0	9.134744	10.342947	-0.068930
55	6	0	8.816869	11.059564	1.076497
56	6	0	9.277332	10.632152	2.328212
57	6	0	10.050616	9.480078	2.426801
58	6	0	10.314722	9.519911	4.856156
59	6	0	8.428263	7.712823	5.989792
60	6	0	7.405602	8.552699	6.450320
61	6	0	6.071800	8.211629	6.254939
62	6	0	5.742362	7.027053	5.597108
63	6	0	6.752201	6.184456	5.142191
64	6	0	8.089989	6.522789	5.341062
65	6	0	10.215255	9.053284	7.750777
66	6	0	10.666395	10.366209	7.931376
67	6	0	10.739071	10.919643	9.208197

Supporting Information

68	6	0	10.349900	10.176388	10.318781
69	6	0	9.893478	8.869464	10.149764
70	6	0	9.837939	8.309623	8.879969
71	6	0	13.377991	7.820377	3.380494
72	6	0	13.796108	9.158295	3.303097
73	6	0	15.147095	9.493535	3.285073
74	6	0	16.115719	8.491088	3.320695
75	6	0	15.725192	7.153956	3.367215
76	6	0	14.369968	6.827131	3.405364
77	7	0	11.220958	7.689382	1.532194
78	7	0	10.546049	8.881082	3.591092
79	7	0	13.058010	2.804198	4.518334
80	7	0	14.346529	4.433019	5.856088
81	15	0	10.665285	5.142403	1.221058
82	15	0	10.192455	8.195882	6.140765
83	15	0	10.489024	2.710805	3.769472
84	15	0	13.578111	6.674645	7.160387
85	14	0	11.533818	7.389765	3.303446
86	14	0	12.571805	4.236486	5.518319
87	78	0	10.977207	5.017493	3.531118
88	78	0	11.733561	6.512054	5.747581
89	1	0	8.080795	3.313008	2.310510
90	1	0	6.882861	2.229240	0.443378
91	1	0	7.780710	0.128251	-0.561219
92	1	0	9.886020	-0.853363	0.321058
93	1	0	11.081859	0.225428	2.178246
94	1	0	8.142817	3.775717	4.945813
95	1	0	6.589617	3.022335	6.727633
96	1	0	7.074920	0.926937	7.985669
97	1	0	9.097752	-0.403023	7.431702
98	1	0	10.606048	0.316681	5.621035
99	1	0	12.451631	1.832165	2.760281
100	1	0	12.154664	0.907366	4.243866
101	1	0	14.523863	1.006790	2.961086
102	1	0	17.007865	1.044144	2.840388
103	1	0	18.300017	2.702396	4.158915
104	1	0	17.122450	4.329975	5.627647
105	1	0	15.579941	5.263940	7.356938
106	1	0	15.547772	6.169461	5.829347
107	1	0	14.083078	4.168464	8.632154
108	1	0	13.720129	3.522447	10.984526
109	1	0	12.856143	5.185279	12.621897
110	1	0	12.374156	7.511977	11.879725
111	1	0	12.738001	8.165270	9.525198
112	1	0	16.054393	7.617423	8.452404
113	1	0	17.154654	9.830679	8.539737
114	1	0	16.109071	11.790525	7.427371
115	1	0	13.967028	11.514787	6.188927
116	1	0	12.894491	9.274243	6.065085
117	1	0	13.234635	1.834646	7.131752
118	1	0	12.350572	0.853367	9.233324
119	1	0	10.398152	1.894275	10.367291
120	1	0	9.308610	3.893882	9.366049
121	1	0	10.183814	4.855634	7.255353
122	1	0	13.203220	3.945955	1.352850
123	1	0	14.323159	2.092481	0.169219
124	1	0	13.099950	0.809519	-1.585600
125	1	0	10.745172	1.397300	-2.122996
126	1	0	9.623528	3.250152	-0.946773
127	1	0	9.371357	5.407064	-1.475308
128	1	0	7.122572	6.187087	-2.134929
129	1	0	5.475464	6.860789	-0.404123
130	1	0	6.087405	6.730612	2.006202
131	1	0	8.328006	5.883078	2.669303
132	1	0	12.635037	6.459858	0.576146
133	1	0	11.232751	6.900544	-0.423823
134	1	0	10.191317	8.649846	-0.894008
135	1	0	8.777700	10.677536	-1.041546

Supporting Information

136	1	0	8.213547	11.963125	1.010895
137	1	0	9.028735	11.209744	3.216722
138	1	0	11.121833	10.230647	5.119001
139	1	0	9.373582	10.089886	4.849503
140	1	0	7.658073	9.479421	6.967722
141	1	0	5.285757	8.872990	6.615745
142	1	0	4.697427	6.762313	5.442305
143	1	0	6.497128	5.258174	4.624057
144	1	0	8.900368	5.874940	4.985509
145	1	0	10.967465	10.967965	7.074993
146	1	0	11.098040	11.940153	9.331572
147	1	0	10.405406	10.612730	11.314656
148	1	0	9.597600	8.275821	11.013102
149	1	0	9.510796	7.274891	8.758943
150	1	0	13.042909	9.949437	3.255657
151	1	0	15.449011	10.540092	3.244516
152	1	0	17.173024	8.752128	3.303302
153	1	0	16.472217	6.359412	3.382502
154	1	0	14.070660	5.776266	3.461876

References

- 1 L. S. H. Dixon, A. F. Hill, M. Sinha, J. S. Ward, J. S. *Organometallics* **2014**, 33, 653.
- 2 a) I. J. S. Fairlamb, A. R. Kapdi, A. F. Lee, *Org. Lett.* **2004**, 6, 4435; b) A. R. Kapdi, A. C. Whitwood, D. C. Williamson, J. M. Lynam, M. J. Burns, T. J. Williams, A. J. Reay, J. Holmes, I. J. S. Fairlamb, *J. Am. Chem. Soc.* **2013**, 135, 8388.
- 3 a) L. E. Crascall, J. L. Spencer, *Inorg. Synth.* **1990**, 28, 126; b) D. B. Dell'Amico, L. Labella, F. Marchetti, S. Samaritani, *S. J. Organomet. Chem.* **2011**, 696, 1349.
- 4 a) R. A. Schunn, *Inorg. Synth.* **1974**, 15, 5; b) D. J. Krysan, J. Mackenzie, *J. Org. Chem.* **1990**, 55, 4229.
- 5 M. J. Frisch et al., *Gaussian 09*®, revision A.02; Gaussian, Inc.: Wallingford, CT, 2009.
- 6 A. D. Becke, *Phys. Rev. A* **1988**, 38, 3098.
- 7 J. P. Perdew, *Phys. Rev. B* **1986**, 34, 7406.
- 8 J. P. Perdew, *Phys. Rev. B* **1986**, 33, 8822.
- 9 (a) W. Kuechle, M. Dolg, H. Stoll and H. Preuss, *J. Chem. Phys.* **1994**, 100, 7535. (b) M. Dolg, H. Stoll, H. Preuss and R. M. Pitzer, *J. Phys. Chem.* **1993**, 97, 5852.
- 10 P. C. Hariharan and J. A. Pople, *Theor. Chim. Acta* **1973**, 28, 213.
- 11 F. Weigend, F. Furche and R. Ahlrichs, *J. Chem. Phys.* **2003**, 119, 12753.