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### **Supporting Information**

# Optically-Controlled Supramolecular Self-assembly of Antibiotic for Antibacterial Regulation

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## Experimental

**Materials:** All organic solvents were obtained from Beijing Chemical Works and used as received. 4-Phenylazobenzoyl Chloride was purchased from TCI. Norfloxacin was purchased from across.  $\alpha$  cyclodextrin was purchased from J&K Scientific Ltd. All chemicals were used without further purification.

**Measurements:** The 1H-NMR and 13C-NMR were taken from the Bruker Avance III 400MHz HD spectrometer. MS experiments were performed on an Impact HD Q-TOF mass spectrometer (Bruker Daltonics, Bremen, Germany), operated in the positive ion mode. SEM images were collected by JSM-6700F. UV-Vis absorption spectra were determined on a Thermo Scientific Evolution 201 spectrophotometer.

The diameter was collected by Dynamic Light Scatting (Malvern, Nano ZS 90). The calorimetric measurement was taken on MicroCal VP-ITC apparatus at 298.15 K.

Synthetic route of azobenzene-norfloxacin (Azo-Nor): Norfloxacin (0.5 g, 1.5 mmol), triethylamine (0.167 g, 1.65 mmol) were completely dissolved in super dry dimethyl chloride (20 mL), and the reaction system was put into ice bath for 30 min. and the solvent of 4-phenylazobenzoyl chloride (0.919 g, 3.75 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (8 mL) was added to the above system dropwisely. The reaction was conducted overnight at room temperature. Water (20 mL) was added to stop the reaction and the mixture was washed with water for three times. Azobenzene-norfloxacin (0.394 g, 0.75 mmol) was purified by column chromatography (CH<sub>2</sub>Cl<sub>2</sub>/ ethyl acetate/methanol = 1/1/0.1, v/v/v) in yield of 49.8%. <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>, ppm): 8.70 (s, 1H), 8.02 (s, 1H), 7.91 (m, 3H), 7.62(d, 2H), 7.59 (m, 3H) 6.88 (d, 1H) 4.43 (m, 2H) 3.91 (d, 4H) 3.35 (m, 4H) 1.66 (m, 3H). <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>, ppm):  $\delta$ 177.02, 169.85, 167.02, 154.50, 153.39, 152.48, 147.33, 145.62, 145.54, 137.03, 131.67, 129.22, 128.20, 123.09, 113.25, 113.07, 108.62, 104.34, 49.78, 14.52, 13.74.

#### Isothermal titration microcalorimetry (ITC) analysis:

In the following equations,

K = binding constant;

N = number of binding ratio

 $\Theta$  = fraction of sites occupied by compound;

M<sub>t</sub> and [M] are bulk and free concentration of bacterium;

 $X_t$  and [X] are bulk and free concentration of the repetitive unit of compound in  $V_0$ ;

 $V_0$  = the volume of active sample cell;

 $\Delta V_i$  = injection volume;

Q = heat content;

 $Q_i$  = heat content from the completion of the *i*<sup>th</sup> injection.

The binding process is described in a simplified fashion by equation (1).

$$X + M \Leftrightarrow XM$$
 (1)

In the process, the binding constant and the bulk concentration of the compound can be expressed as follows:

$$K = \frac{\Theta}{(1 - \Theta)[X]} (2)$$
$$X_t = [X] + N\Theta M_t (3)$$

Combining equations (2) and (3) above gives:

$$\Theta^2 - \Theta \left[ 1 + \frac{X_t}{M_t} + \frac{1}{NKM_t} \right] + \frac{X_t}{M_t} = 0(4)$$

The total heat content Q contained in  $V_0$  at fractional saturation  $\Theta$  is:

$$Q = N\Theta M_t \Delta H V_0 \quad (5)$$

Solving the quadratic equation (4) for  $\Theta$  and then substituting this into equation (5) gives:

$$Q = \frac{M_t \Delta HV_0}{2} \left[ 1 + \frac{X_t}{M_t} + \frac{1}{NKM_t} - \sqrt{\left(1 + \frac{X_t}{M_t} + \frac{1}{NKM_t}\right)^2 - \frac{4X_t}{M_t}} \right] (6)$$

The calculated heat released from the  $i^{th}$  injection,  $\Delta Q_{(i)}$  can be expressed as:

$$\Delta Q(i) = Q(i) + \frac{dV_t}{V_0} \left[ \frac{Q(i) + Q(i-1)}{2} \right] - Q(i-1) \quad (7)$$

The practical heat released is determined in the *i*<sup>th</sup> injection. The fitting process ITC data involves 1) initial guesses of N, K, and  $\Delta H$ ; 2) calculation of  $\Delta Q_{(i)}$  for each

injection and comparison of these values with the measured heat for the corresponding experimental injection; 3) improvement in the initial values of N, K, and  $\Delta H$  by standard Marquardt methods; 4) iteration of the above procedure until no further significant improvement in fit occurs with continued iteration.

The characterization of photoisomerization behavior of Azo-Nor in DMSO: All irradiation experiments were performed with M-Ultra violet Light Source (365 nm, MUA-165) and M-Visual light source (white light, MVL-210). All the concentration of Azo-Nor was set as 25 µM when testing the absorption spectra. To measure the time of reaching the balance of cis state, the absorption spectra of Azo-Nor was texted each 10s upon exposed to 365 nm light irradiation at the intensity of 5mM/cm<sup>2</sup>. By alternating the irradiation between 365 nm (1min, 5mM/cm<sup>2</sup>) and white light (1min, 10mM/cm<sup>2</sup>), the photoswitching cycles was obtained. The absorbance was measured at 325 nm. After exposed to 365 nm light (1min, 5mM/cm<sup>2</sup>), the half-life of thermal cis-trans isomerization was obtained by monitoring the absorbance at 325 nm at intervals of 30mins under dark condition. <sup>1</sup>H NMR was employed to study the ratio of trans and cis isomers changing in different situations. The concentration of Azo-Nor was set as 500 µM. Firstly, the ratio of *trans* and *cis* isomers in thermal state was texted by <sup>1</sup>H NMR. Under the irradiation of 365 nm (1min, 5mM/cm2), the Azo-Nor transferred to 365 nm thermodynamically photostationary state, then measured the <sup>1</sup>H NMR to obtained the ratio of *trans* and *cis* isomers. Upon the irradiation of white light 10mM/cm<sup>2</sup>), the ratio of *trans* and *cis* isomers in white light (1min,

thermodynamically photostationary state was also obtained through <sup>1</sup>H NMR.

**Bacterial strains and growth solutions:** The bacterial strains were *S. aureus* (ATCC6538) and *E. coli* (Top 10). A single colony of *S. aureus* were grown in Luria-Bertani (LB) at 37 °C. *E. coli* were grown in Nutrient Broth (NB) medium supplemented with required antibiotic, 10  $\mu$ g/ml penicillin at 37 °C.

Antibacterial experiments of Azo-Nor/aCD and Azo-Nor/aCD (1/1): After the overnight growth of bacterial, *S. aureus* and *E. coli* were diluted to an OD<sub>600</sub> of 0.1 and 100  $\mu$ L of solution was added to 96-well plate.<sup>[1]</sup> To test the antibacterial activity after photo-activation, the solutions containing antibiotics were irradiated at 365 nm firstly for 10 min before adding to the cell suspension. The mixture solution of antibiotic and bacterial were grown in a microtiter plate at 37 °C under dark condition. The cell density (650 nm) was measured after 12 h with a 10 seconds shaking step before each measurement in a microplate reader. The background graphs were corrected by subtracting the OD<sub>650</sub> at time 0.

# Reference

W. A. Velema, M. J. Hansen, M. M. Lerch, A. J. M. Driessen, W. Szymanshi, B.
L. Feringa, *Bioconjugate Chem.* 2015, 26, 2592-2597.



**Figure S1.** SEM images of Azo-Nor and Azo-Nor/αCD.



**Figure S2.** (a-d) Growth rates of *E. coli* at increasing concentrations of Azo-Nor and Azo-Nor/αCD before and after irradiation.



Figure S3. (a) Growth rates of S. aureus at increasing concentrations of  $\alpha$ CD and IC<sub>50</sub> value. (b) Growth rates of E. coli at increasing concentrations of  $\alpha$ CD and IC<sub>50</sub> value.



Figure S4. Mass spectrum of azobenzene-norfloxacin.