Supporting Information

for

Nitro Reduction-based Fluorescent Probes for Carbon Monoxide Require Reactivity Involving a Ruthenium Carbonyl Moiety

Zhengnan Yuan[#], Xiaoxiao Yang[#], Ladie Kimberly De La Cruz and Binghe Wang^{*}

Department of Chemistry and Center for Diagnostics and Therapeutics, Georgia State

University, Atlanta, Georgia 30303, USA.

* To whom correspondence should be addressed: wang@gsu.edu (B. Wang)

[#]These two authors made equal contributions.

General Information

All reagents and solvents were of reagent grade from commercial suppliers (Sigma Aldrich and etc.). ¹H (400 MHz) and ¹³C (100 MHz) NMR spectra were performed on a Bruker-400 spectrometer. Mass spectral analyses were performed on an ABI API 3200 (ESI-Triple Quadruple) by the Georgia State University Mass Spectrometry Facilities. Fluorescence spectra were recorded on a Shimadzu RF-5301PC fluorometer. Absorption spectra were measured on Varian Cary 100 Bio UV-Visible spectrophotometer. **CORM-2, CORM-3, CORM-A1** and **CORM-401** were purchased from Sigma-Aldrich and were used without purification. **BW-CO-103**¹ and **COP-1**² were synthesized according to literature procedures. **iCORM-3** was prepared according to a literature procedure.³ 2-Dicyanomethylen-3-cyano-4,5,5-trimethyl-2,5-dihydrofurane was synthesized according to the reported procedures.⁴

Synthesis

5-nitro-2-phenyl-1*H***-benzo**[*de*]isoquinoline-1,3(2*H*)-dione (COFP) was synthesized according to a reported procedure.⁵ 3-Nitro-1,8-naphthalic anhydride (0.62 mmol, 150 mg) and aniline (0.62 mmol, 58 mg) were refluxed in 4 ml acetic acid for 4 h. 20 ml water was added at room temperature and the formed yellow solid was filtered and washed with water. The crude product was purified by silica gel column chromatography to yield the pure product as white solid (103 mg, 52%). ¹H NMR (DMSO-d₆) δ 9.51 (s, 1H), 8.95 (s, 1H), 8.81 (d, *J* = 8.4 Hz, 1H), 8.68 (d, *J* = 7.2 Hz, 1H), 8.08 (t, *J* = 8.0 Hz, 1H), 7.56-7.48 (m, 3H), 7.41 (d, *J* = 7.2 Hz, 2H). ¹³C NMR (DMSO-d₆) δ 163.2, 162.7, 145.9, 136.5, 135.7, 134.0, 131.1, 130.1, 129.9, 129.4, 129.1, 128.6, 124.7, 123.3, 122.9. HRMS calculated for C₁₈H₁₀N₂O₄Na [M+Na]⁺: m/z 341.0540, found 341.0538.

2-(2-morpholinoethyl)-5-nitro-1H-benzo[de]isoquinoline-1,3(2H)-dione (LysoFP-NO₂) was synthesized following a modified reported procedure.⁶ To a solution of 3-nitro-1,8-naphthalic anhydride (0.33 mmol, 80 mg) in 20 ml ethanol was added 4-(2-aminoethyl) morpholine (0.39 mmol, 51 mg) dropwise in 3 ml ethanol. The solution was allowed to stir for 40 min at r.t. and then heated to reflux for another 2 h. The formed solid was filtered and washed with 5 ml cold ethanol. The crude product was purified by silica gel column chromatography to yield the pure product (42 mg, 36%). ¹H NMR (CDCl₃) δ 9.30 (d, *J* = 2.2 Hz, 1H), 9.13 (d, *J* = 2.2 Hz, 1H), 8.93-8.61 (m, 1H), 8.43 (d, *J* = 8.2 Hz, 1H), 8.07-7.77 (m, 1H), 4.36 (t, *J* = 6.7 Hz, 2H), 3.80-3.50 (m, 4H), 2.72 (t, *J* = 6.6 Hz, 2H), 2.58 (s, 4H). ¹³C NMR (CDCl₃) δ 163.3, 162.6, 146.5, 135.7, 134.5, 131.1, 130.4, 129.3, 129.1, 124.8, 124.4, 123.3, 67.1, 56.2, 54.0, 37.7. HRMS calculated for C₁₈H₁₈N₃O₅ [M+H]⁺: m/z 356.1237, found 356.1246.

(E)-2-(3-cyano-4-(3-hydroxy-4-nitrostyryl)-5,5-dimethylfuran-2(5H)-ylidene)malononitrile

(NIR-CO) was synthesized following a literature procedure.⁷ To a solution of 3-hydroxy-4nitrobenzaldehyde (105 mg, 0.63 mmol) in 5 ml ethanol was added piperidine (53 mg, 0.63 mmol) and 2-dicyanomethylen-3-cyano-4,5,5-trimethyl-2, 5-dihydrofurane (125 mg, 0.63 mmol). The mixture was allowed to stir at r.t. for 12 h and then filtered. The residue was purified by silica gel column chromatography to yield the pure product (80 mg, 36%). ¹H NMR (DMSO-d₆) δ 7.98 (d, *J* = 8.8 Hz, 1H), 7.81 (d, *J* = 16.8 Hz, 1H), 7.51 (d, *J* = 8.4 Hz, 2H), 7.24 (d, *J* = 16.4 Hz, 1H), 1.78 (s, 6H). ¹³C NMR (DMSO-d₆) δ 177.0, 174.2, 152.31, 144.1, 140.4, 138.1, 126.1, 119.7, 119.3, 112.6, 111.9, 110.7, 102.2, 99.9, 55.6, 25.1. HRMS calculated for C₁₈H₁₁N₄O₄ [M-H]⁻: m/z 347.0780, found 347.0774.

cis-RuCl₂(DMSO)₄ (Complex D) was synthesized according to a reported procedure.⁸ Briefly, 100 mg RuCl₃.xH₂O was dissolved in 4 ml DMSO and stirred at reflux for 1 h. The color of the reaction mixture changed from deep brown to light yellow. After cooling to room temperature, the reaction mixture was cooled at -20 °C; the crystalized bright yellow solid was filtered and washed with acetone followed by drying in vacuum to afford 147 mg (63%) of a bright yellow solid. NMR shown the aqua species *cis,fac*-[RuCl₂(dmso-S)₃(H₂O)] after dissolved in D₂O as suggested. ¹H NMR (D₂O) δ 3.48 (s, 6H), 3.46 (s, 6H), 3.37 (s, 6H), 2.70 (s, 6H). ¹³C NMR (D₂O) δ 46.73, 45.70, 44.31, 38.67.

N-isobutyl-4-nitrobenzamide (PNB) To a solution of 4-nitrobenzoic acid (200 mg, 1.2 mmol) in 5 ml DCM was added EDC (343 mg, 1.8 mmol), DMAP (219 mg, 1.8 mmol), triethylamine (181 mg, 1.8 mmol) and isobutylamine (87.6 mg, 1.2 mmol). The mixture was allowed to stir at r.t. for 12 h. The reaction solution was diluted with 15 ml DCM and washed with 10 ml 0.1 M HCl. The organic layer was dried over anhydrous Na₂SO₄ and concentrated under reduced pressure. The residue was purified by silica gel column chromatography to yield the pure product as a white solid (205 mg, 77%). ¹H NMR (CDCl₃) δ 8.15 (d, *J* = 8.8 Hz, 2H), 7.90 (d, *J* = 8.4 Hz, 2H), 7.09 (s, 1H), 3.21 (t, *J* = 6.4 Hz, 2H), 1.87-1.84 (m, 1H), 0.90 (d, *J* = 6.8 Hz, 6H). ¹³C NMR (CDCl₃) δ 165.8, 149.6, 140.6, 128.2, 123.9, 47.78, 28.7 20.3. HRMS calcd for C₁₁H₁₄N₂O₃ [M+H]⁺ 223.1083, found 223.1093.

4-amino-N-isobutylbenzamide (PAB) To a solution of **PNB** (45 mg, 0.18 mmol) in 1.8 ml acetone/water (5:1) was added zinc powder (117 mg, 1.8 mmol) and ammonium chloride (144 mg, 2.7 mmol). The mixture was vigorously shaken at room temperature for 1 min and then filtered. The filtrate was diluted with 10 ml ethyl acetate and washed with 5 ml saturated NaHCO₃ solution. The organic layer was dried over anhydrous Na₂SO₄ and concentrated under reduced pressure. The residue was purified by silica gel column chromatography to yield the pure product as a white solid (23 mg, 67%). ¹H NMR (Methanol-d₄) δ 8.13 (s, 1H), 7.59 (d, *J* = 8.4 Hz, 2H), 6.66 (d, *J* = 8.8 Hz, 2H), 3.15 (t, *J* = 6.4 Hz, 2H), 1.92-1.86 (m, 1H), 0.94 (d, *J* = 6.8 Hz, 6H). ¹³C NMR (Methanol-d₄) δ 170.6, 153.0, 129.8, 123.5, 114.7, 48.3, 29.9, 20.6. HRMS calcd for C₁₁H₁₆N₂O [M-H]⁻: m/z 191.1186, found 191.1184.

General procedures for spectral studies

Stock solutions preparation: COFP, **LysoFP-NO**₂ and **NIR-CO** were dissolved in DMSO to afford a 500-μM stock solution; **COP-1** was dissolved in DMSO to afford a 200-μM stock solution. For BW-**CO-103**, **CORM-2**, **CORM-401** and **Complex D**, the compounds were freshly prepared in DMSO to afford 5-mM stock solutions. **CORM-3** and **CORM-A1** were dissolved in distilled water to afford 5-mM stock solutions. According to the literature procedure,³ **CORM-3** was dissolved in PBS and incubated overnight to afford the 5-mM **iCORM-3** stock solution.

CO gas treatment: To a 6-ml headspace vial, 2 ml of the probe solution was added. 10 ml pure CO gas was directly bubbled into the solution by a headspace gas syringe. The resulting solution was further incubated at 37 °C for 60 min followed by the acquisition of the fluorescence spectra.



Figure S1. Fluorescence spectra of **COP-1** (1 μ M) upon treatment with **CO-103** (100 μ M) over 2 h in DMSO/PBS (pH = 7.4) 5:1 at 37 °C (λ_{ex} = 475 nm, slit widths: W_{ex} = 5 nm, W_{em} = 3 nm).



Figure S2. Fluorescence spectra of CP-103 formation during incubation with COP-1 (1 μ M) in DMSO/PBS (pH = 7.4) 5:1 at 37 °C (λ_{ex} = 373 nm, slit widths: W_{ex} = W_{em} = 5 nm).



Figure S3. Fluorescent spectra of **COP-1** (1 μ M) upon incubation with **CORM-A1** (100 μ M) over 60 min in PBS (pH = 7.4, 4% DMSO) at 37 °C. (λ_{ex} = 475 nm, slit widths: W_{ex} = 5 nm, W_{em} = 3 nm).



Figure S4. Fluorescent spectra of **COP-1** (1 μ M) upon incubation with **CORM-401** (100 μ M) over 60 min in PBS (pH = 7.4, 4% DMSO) at 37 °C. (λ_{ex} = 475 nm, slit widths: W_{ex} = 5 nm, W_{em} = 3 nm).



Figure S5. Fluorescent spectral changes of **COFP** (10 μ M) upon incubation with **CORM-2** (100 μ M) in PBS (pH = 7.4, 4% DMSO) at 37 °C. (λ_{ex} = 440 nm, slit widths: W_{ex} = W_{em} = 10 nm)



Figure S6. Fluorescent spectral changes of **COFP** (10 μ M) upon incubation with reaction products between **CORM-2** and DMSO (100 μ M) in PBS (pH = 7.4, 4% DMSO) at 37 °C. (λ_{ex} = 440 nm, slit widths: W_{ex} = W_{em} = 10 nm)



Figure S7. Fluorescent spectra of COFP (10 μ M) upon incubation with CORM-3 (100 μ M) in PBS (pH = 7.4, 4% DMSO) at 37 °C. (λ_{ex} = 440 nm, slit widths: W_{ex} = W_{em} = 10 nm)



Figure S8. Fluorescent spectra of COFP (10 μ M) upon incubation with iCORM-3 (100 μ M) in PBS (pH = 7.4, 4% DMSO) at 37 °C. (λ_{ex} = 440 nm, slit widths: W_{ex} = W_{em} = 10 nm)



Figure S9. Fluorescence intensity changes from **COFP** (10 μ M) upon treatment with **CORM-3** and **iCORM-3** in PBS (pH = 7.4, 4% DMSO) at 37 °C. (λ_{ex} = 440 nm, λ_{em} = 522 nm, slit widths: W_{ex} = W_{em} = 10 nm)



Figure S10. Fluorescent spectra of LysoFP-NO₂ (10 μ M) upon incubation with CORM-3 (100 μ M) in PBS (pH = 7.4, 4% DMSO) at 37 °C. (λ_{ex} = 440 nm, slit widths: W_{ex} = W_{em} = 10 nm)



Figure S11. Fluorescent spectra of LysoFP-NO₂ (10 μ M) upon treatment with CO gas in PBS (2% DMSO) at 37 °C for 1 h. (λ_{ex} = 440 nm, slit widths: W_{ex} = W_{em} = 10 nm)







Figure S13. Fluorescent spectra of **COFP** (10 μ M) upon incubation with **CORM-A1** (100 μ M) over 60 min in PBS (pH = 7.4, 4% DMSO) at 37 °C. (λ_{ex} = 440 nm, slit widths: W_{ex} = W_{em} = 10 nm)



Figure S14. Fluorescent spectra of **COFP** (10 μ M) upon incubation with **CORM-401** (100 μ M) over 60 min in PBS (pH = 7.4, 4% DMSO) at 37 °C. (λ_{ex} = 440 nm, slit widths: W_{ex} = W_{em} = 10 nm)



Figure S15. Fluorescent spectra of **COFP** (10 μ M) upon incubation with **Complex D** (100 μ M) for 60 min in PBS (pH = 7.4, 4% DMSO) at 37 °C. (λ_{ex} = 440 nm, slit widths: W_{ex} = W_{em} = 10 nm)

HPLC and LC-MS analysis of the reduction of PNB by CORM-3



Stock solution preparation: PNB was dissolved in DMSO to afford a 10 mM stock solution. **CORM-3** was dissolved in distilled water to afford a 100 mM stock solution.

20 μ L of **PNB** stock solution was diluted with 360 μ L PBS (pH 7.4), followed by addition of 20 μ L of the CORM-3 stock solution. The final concentrations of PNB and CORM-3 were 500 μ M and 5 mM respectively. The reaction mixture was incubated at 37°C and monitored by HPLC (gradient: ACN in water (0.1%TFA) 5-95% in 10 min, C18 4.6*150 mm column). For CO gas treatment, pure CO gas was bubbled through the **PNB** solution (500 μ M in PBS, pH 7.4 with 5% DMSO) for 2 h at 37 °C. The formation of **PAB** was identified by HPLC (with **PAB** reference compound) and LC/MS.



Figure S16. HPLC of the PNB/CORM-3 reaction.





Figure S17. LC/MS analysis of the HPLC sample: (**A**) Before addition of CORM-3, PNB peak at 5.9 min with $m/z=223.3 [M+H]^+$; (**B**) **PAB** reference peak at 4.8 min with $m/z=193.0 [M+H]^+$; (**C**) Reaction mixture with CORM-3 showed **PAB** peak at 4.8 min with $m/z=193.4 [M+H]^+$; (**D**) Bubbling PNB with CO showed only **PNB** peak at 5.9 min with $m/z=223.2 [M+H]^+$



Figure S18. A literature proposed mechanism for the reduction of a nitro group to an amine by ruthenium-CO complexes.⁹



Figure S19. The requirement of the ruthenium carbonyl complex in CO sources for probe reduction.

Spectroscopic data of synthesized compounds















References

1. X. Ji, C. Zhou, K. Ji, R. E. Aghoghovbia, Z. Pan, V. Chittavong, B. Ke and B. Wang, *Angew. Chem. Int. Ed.*, 2016, **55**, 15846-15851.

- 2. B. W. Michel, A. R. Lippert and C. J. Chang, J. Am. Chem. Soc., 2012, **134**, 15668-15671.
- 3. E. Clark James, P. Naughton, S. Shurey, J. Green Colin, R. Johnson Tony, E. Mann Brian, R. Foresti and R. Motterlini, *Circ. Res.*, 2003, **93**, e2-e8.
- 4. D. Villemin and L. Liao, *Synth. Commun.*, 2001, **31**, 1771-1780.
- 5. Y.-C. Yuan, R. Kamaraj, C. Bruneau, T. Labasque, T. Roisnel and R. Gramage-Doria, *Org. Lett.*, 2017, **19**, 6404-6407.
- 6. K. Dhara, S. Lohar, A. Patra, P. Roy, S. K. Saha, G. C. Sadhukhan and P. Chattopadhyay, *Anal. Chem.*, 2018, **90**, 2933-2938.
- 7. Z. Wang, C. Liu, X. Wang, Q. Duan, P. Jia, H. Zhu, Z. Li, X. Zhang, X. Ren, B. Zhu and W. Sheng, *Sens. Actuators B Chem.*, 2019, **291**, 329-336.
- 8. T. B. Rauchfuss, in *Inorganic Syntheses*, ed. T. B. Rauchfuss, John Wiley & Sons, Hoboken, N, J, 1 edn., 2010, vol. 35, ch. 8, pp. 148-163.
- 9. K. Nomura, J. Mol. Catal. A: Chem., 1995, **95**, 203-210.