

## Supporting Information

# Hydrogen-Bonded 3D network of d<sup>10</sup>-Metal Halide Coordination Polymer Containing N-(3-pyridinyl) nicotinamide: Influence of Ligand Conformation, Halide Anions and Solvent

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## Contents

Description	Page No.
Mechanochemical and thermal synthesis method for 1 and 2	S2
<b>Table S1.</b> Selected bond distances (Å) and angles (°) for <b>1-4</b>	S3
<b>Table S2.</b> Aromatic π···π interaction parameters (Å and °) for compound <b>L</b> and <b>1-4</b>	S3
<b>Table S3.</b> Dihedral angles of Py-C, Py-N rings and amide plane in compound <b>L</b> and <b>1-4</b>	S4
<b>Table S4.</b> Results of Cambridge Structural Database (CSD) analysis of complexes containing 3-pna ligand	S5-S8
<b>Fig.S1</b> Hirshfeld surface mapped with dnorm (top, left), decomposed fingerprint plot for compound L (top, right), showing (a) NH...N and CH...N hydrogen bonds and Hirshfeld surfaces mapped with shape index (red and blue triangles has been shown by black circle), curvedness and decomposed finger print plots to C-C contacts (bottom, left to right)showing relative contribution of π...π stacking interactions for compound L.	S9
<b>Fig. S2</b> Representation of CH...π interaction in compound <b>1</b> , showing nearest bonds to CH group	S10
<b>Fig. S3</b> Hirshfeld surface mapped with $d_{norm}$ (left) and two-dimensional fingerprint plot for compound <b>1</b> , showing (a) N-H...Cl, (b) C-H...Cl and (c) CH...π hydrogen bonds	S11
<b>Fig. S4</b> Hirshfeld surfaces mapped with shape index (red and blue triangles has been shown by black circle), curvedness and decomposed finger print plots to C-C contacts showing relative contribution of π···π stacking interactions for compound <b>1(a)</b> , <b>2(b)</b> , <b>3(c)</b> and <b>4(d)</b> (left to right)	S12
<b>Fig. S5-S13</b> IR Spectra of compounds.	S13-S17
<b>References</b>	S17- S18

## **Experimental**

### **Mechanochemical method:**

Neat grinding of 3-pna (0.04 g, 0.2 mmol) with mercury (II) chloride (0.054 g, 0.2 mmol) manually in a mortar and pestle for about 15 minutes produced a white powder in high yield. Mp 277-279 °C, IR data (KBr pellet, cm-1): 694, 803, 1053, 1121, 1290, 1333, 1427s, 1484, 1534s, 1585, 1604, 1682s, 3332. The same procedure was used for preparation of product between 3-pna and HgBr<sub>2</sub> and powder of this compounds was isolated after grinding of 1:1 stoichiometric ratio of starting material after 15 minutes. In the case of obtained powder from reaction between L and HgBr<sub>2</sub>, Mp 268-270°C, IR data (KBr pellet, cm-1): 691, 798, 1050, 1117, 1289, 1426s, 1484, 1528s, 1590, 1683s, 3347.

### **Thermal method:**

In this method, mixture of 3-pna (0.04 g, 0.2 mmol) with mercury (II) chloride (0.054 g, 0.2 mmol) was heated in the oven at 110 °C for about 4 h that produced a white powder which was characterized. Mp 279-280 °C, IR data (KBr pellet, cm-1): 697, 807, 1045, 1117, 1298s, 1333, 1427s, 1482, 1537s, 1587, 1681s, 3338. Same process was performed for reaction between L and HgBr<sub>2</sub>. The resulting powder of this reaction was isolated after heating of ligand and metal salt at 110 °C for about 4 h. In the case of product prepared from reaction between 3-pna and HgBr<sub>2</sub>, Mp 268-271°C, IR data (KBr pellet, cm-1): 693, 798, 1050, 1118, 1288, 1424s, 1482, 1525s, 1585, 1681s, and 3347.

**TableS1.** Selected bond distances ( $\text{\AA}$ ) and angles ( $^\circ$ ) for **1-4** (M= Cd / Hg, X= halogen)

<b>Compound</b>		[HgCl <sub>2</sub> (L)] <sub>n</sub> (1)	[HgBr <sub>2</sub> (L)] <sub>n</sub> (2)	[HgI <sub>2</sub> (L)] <sub>n</sub> .meth (3)	[CdI <sub>2</sub> (L)] <sub>n</sub> (4)
<b>Bond distance</b>	M-X	2.396(5)	2.490(2)	2.635(1)	2.677(1)
		2.356(6)	2.517(2)	2.633(1)	2.688(1)
	M-N	2.363(16)	2.408(14)	2.500(10)	2.275(8)
		2.315(17) <sup>a</sup>	2.362(14) <sup>b</sup>	2.446(11) <sup>c</sup>	2.251(9) <sup>d</sup>
<b>Bond angle</b>	X-M-X	140.3(2)	140.9(6)	147.1(4)	127.2(3)
	N-M-N	89.4(6) <sup>a</sup>	90.3(5) <sup>b</sup>	78.3(4) <sup>c</sup>	99.8(3) <sup>d</sup>
	X-M-N	99.4(4)	107.0(3)	101.4(3)	105.7(2)
		108.5(4)	100.7(3)	101.5(3)	111.5(2)
		109.3(4) <sup>a</sup>	108.1(3) <sup>b</sup>	100.8(3) <sup>c</sup>	100.3(2) <sup>d</sup>
		98.5(4) <sup>a</sup>	98.5 (3) <sup>b</sup>	106.8(3) <sup>c</sup>	108.5(2) <sup>d</sup>

Symmetry codes: <sup>a</sup> 3/2-x, -1/2+y, 3/2-z. <sup>b</sup> 5/2-x, -1/2+y, 3/2-z. <sup>c</sup> 3/2-x, 1/2+y , 3/2-z. <sup>d</sup> 1/2-x, -1/2+y , 1/2-z.

**Table S2.** Aromatic  $\pi\cdots\pi$  interaction parameters ( $\text{\AA}$  and  $^\circ$ ) for **1-4**

<b>Compound</b>	$\pi_{\text{py}}\cdots\pi_{\text{py}}$	$d_{Cg-Cg}$	$d_{\text{plane-plane}}$	$\alpha$	$\gamma, \beta$	$d_{\text{offset}}$
<b>L</b>	$\pi_{\text{py-C}}\cdots\pi_{\text{py-C}}^a$	3.87	3.49	0.0	25.8	1.68
	$\pi_{\text{py-N}}\cdots\pi_{\text{py-N}}^a$					
<b>[HgCl<sub>2</sub>(L)]<sub>n</sub>(1)</b>	$\pi_{\text{py-C}}\cdots\pi_{\text{py-N}}^a$	3.72	3.34, 3.44	4.6	26.3,22.2	1.65,1.40
<b>[HgBr<sub>2</sub>(L)]<sub>n</sub>(2)</b>	$\pi_{\text{py-C}}\cdots\pi_{\text{py-N}}^a$	3.82	3.44, 3.60	8.2	25.8,19.8	1.67,1.29
<b>[HgI<sub>2</sub>(L)]<sub>n</sub></b>	$\pi_{\text{py-C}}\cdots\pi_{\text{py-C}}^b$	3.54	3.37	0.0	17.8	1.09
<b>.meth(3)</b>						
<b>[CdI<sub>2</sub>(L)]<sub>n</sub> (4)</b>	$\pi_{\text{py-C}}\cdots\pi_{\text{py-N}}^a$	3.89	3.60,3.80	13.5	22.5,12.4	1.49,0.84
Symmetry codes: <sup>a</sup> -1+x, y, z. <sup>b</sup> 2-x, -y, 1-z.						

**Table S3.** Dihedral angles of Py-C, Py-N rings and amide plane in **1-4**

Compound	< Py-C and Amide plane (°)	<Py-N and Amide plane(°)	<Py-N and Py-C(°)
<b>L</b>	19.3(1)	4.2 (1)	22.4(8)
[HgCl <sub>2</sub> (L)] <sub>n</sub> ( <b>1</b> )	29.9(1)	32.7(1)	4.6(6)
[HgBr <sub>2</sub> (L)] <sub>n</sub> ( <b>2</b> )	26.7(7)	28. 2(9)	8.2(5)
[HgI <sub>2</sub> (L)] <sub>n</sub> .meth( <b>3</b> )	6.3(8)	44.2(6)	38.1(3)
[CdI <sub>2</sub> (L)] <sub>n</sub> ( <b>4</b> )	30.3(4)	31.5(7)	13.5(3)

**TableS4.** Results of Cambridge Structural Database (CSD) analysis of complexes containing 3-pna ligand

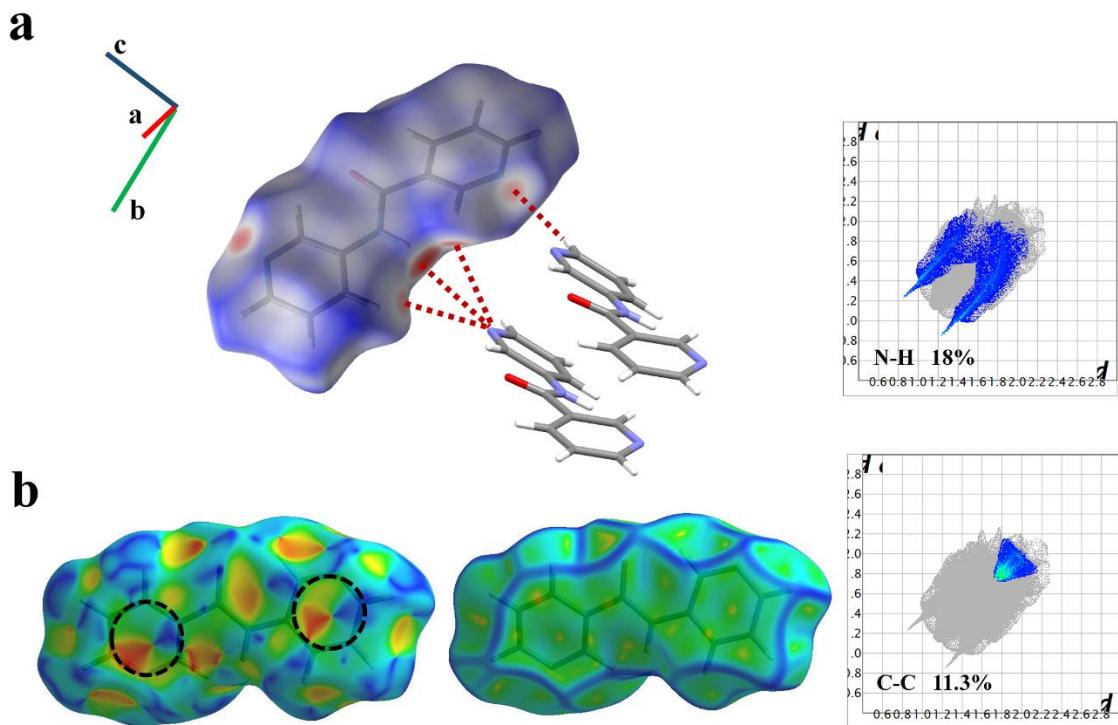
NO.	Refcode	Compounds	Coligand	Confor mation on 3-pna	Amide Synthon	Guest Mol.	Geometry coordination	Architecture	Ref.
1	EVOKUX	[Ni(3-pna)(mip)]·H <sub>2</sub> O	H2mip = 5-methylisophthalic acid]	Anti	$D_1^1(2)$ NH...O (H <sub>2</sub> O)	H <sub>2</sub> O	distorted octahedral	2D networks, ladder-like double chains.	1
2	EVOLAE	[Ni(3-pna)(mip)(H <sub>2</sub> O)]·4H <sub>2</sub>	H2mip = 5-methylisophthalic acid]	Anti	$R_2^2(18)$ NH...O (mip)	H <sub>2</sub> O	distorted octahedral	2D, 4-connected grid structure consisting of 1D [Ni-3-pna] <sub>n</sub> and [Ni-mip] <sub>n</sub> chains.	1
3	EVOLEI	[Cd <sub>2</sub> (3pna)(mip) <sub>2</sub> (H <sub>2</sub> O)]·2H <sub>2</sub> O	H2mip = 5-methylisophthalic acid]	Anti	$D_1^1(2)$ NH...O (H <sub>2</sub> O)	H <sub>2</sub> O	distorted pentagonal bipyramidal	3D framework constructed from Cd <sub>4</sub> (COO) <sub>4</sub> subunits and $\mu^2$ -bridging 3-pna.	1
4	EVOLIM	[Cd(3-pna)(mip)] <sub>2</sub> ·3H <sub>2</sub> O	H2mip = 5-methylisophthalic acid]	Anti	-	H <sub>2</sub> O	distorted octahedral	2D networks, ladder-like double chains	1
5	EVOLOS	[Co(3-pna)(mip)]·H <sub>2</sub> O	H2mip = 5-methylisophthalic acid]	Anti	$D_1^1(2)$ NH...O (H <sub>2</sub> O)	H <sub>2</sub> O	distorted octahedral	2D networks, ladder-like double chains.	1
6	AVUBAU	[Co(NO <sub>3</sub> ) <sub>2</sub> (3-pna) <sub>2</sub> ] <sub>n</sub>	-	Syn	$C_1^1(4)$ NH...O (3-pna)	-	octahedral	honeycomb-shaped grid	2
7	AVUBEY	[CoBr <sub>2</sub> (3-pna) <sub>2</sub> ] <sub>n</sub>	-	Syn	-	-	distorted octahedral	Honeycomb-shaped grid.	2
8	BIPCIO	{[Zn(H <sub>3</sub> -pna)(sip)].1.5H <sub>2</sub> O} <sub>n</sub>	metal 5-sulfoisophthalate (sip)	Syn	$D_1^1(2)$ NH...O (H <sub>2</sub> O)	H <sub>2</sub> O	A slightly distorted tetrahedral	3-connected 1D ladder structure.	3
9	BIPCUA	{[Cd <sub>3</sub> (3pna) <sub>2</sub> (sip) <sub>2</sub> (H <sub>2</sub> O) <sub>6</sub> ] <sub>n</sub> .2H <sub>2</sub> O} <sub>n</sub>	metal 5-sulfoisophthalate (sip)	Syn	$C_1^1(17)$ NH...O (sip)	H <sub>2</sub> O	distorted pentagonal bipyramid/octahedron	2D network topology, ladder motifs	3
10	CICSAK	[Cd(D-cam)(3-pna)(H <sub>2</sub> O) <sub>2</sub> ] <sub>n</sub>	D-camphoric acid (H <sub>2</sub> D-cam)	Syn	-	-	Octahedral	1-D coordination polymer chain structure.	4
11	DOLSEE	[Cu(ip)(3-pna)] <sub>n</sub>	ip = isophthalate	Anti	-	-	distorted square pyramidal	(4,4) layered grid structure based on {Cu <sub>2</sub> (OCO) <sub>2</sub> } dimeric units.	5
12	FIGQOD	{[Cu(glu)(3-pna)(H <sub>2</sub> O)].H <sub>2</sub> O} <sub>n</sub>	glu = glutarate,	Syn	$D_1^1(2)$ NH...O (H <sub>2</sub> O)	H <sub>2</sub> O	distorted octahedron	Simple (4,4) grid-like layer structure.	6
13	FUHPIJ	{[Cu(adp)(3-pna)]·2.5H <sub>2</sub> O} <sub>n</sub>	adipic acid (H <sub>2</sub> adp)	Anti	$D_1^1(2)$ NH...O (H <sub>2</sub> O)	H <sub>2</sub> O	square pyramidal	Dimer-based 3-D network	7

14	GOBTUO	$[\text{Cu}_2(\text{pyro})(\text{pyroH}_2)(3\text{pnaH})_2(\text{H}_2\text{O})_2]n$	Pyro = 1,2,4,5-benzenetetracarboxylate	Syn	$C_1^1(10)$ NH...O (pyro)	-	distorted octahedral	2-D coordination polymer built from $\{\text{Cu}_2\text{O}_2(\text{OCO})_2\}$ dimeric units.	8
15	GOCSOI	$\{[\text{Cu}(\text{pyro})(3\text{-pnaH})_2(\text{H}_2\text{O})_2]\cdot 4\text{H}_2\text{O}\}n$	Pyro = 1,2,4,5-benzenetetracarboxylate	Anti	$C_1^1(14)$ NH...O (Pyro)	$\text{H}_2\text{O}$	distorted octahedral	Simple neutral 1-D chain polymer.	8
16	IMEXAB	$[\text{Zn}_2(2,4\text{-pdc})_2(\text{H}_2\text{O})_4(3\text{pna})]\cdot 3\text{H}_2\text{O}n$	2,4-pyridinedicarboxylic acid (2,4-pdcH <sub>2</sub> )	Anti	$D_1^1(2)$ NH...O ( $\text{H}_2\text{O}$ )	$\text{H}_2\text{O}$	octahedral	Two-fold parallel interpenetrated herring bone layer net.	9
17	JAFDEC	$[\text{Co}(\text{L})(\text{ADTZ})]\cdot \text{H}_2\text{O}$	$\text{H}_2\text{ADTZ}=2,5\text{-}(s\text{-acetic acid})\text{dimercapto-1,3,4-thiadiazole}$	Anti	$D_1^1(2)$ NH...O ( $\text{H}_2\text{O}$ )	$\text{H}_2\text{O}$	distorted octahedral geometry	2D double-layer networks based on 1D $[\text{M-L}]_n$ zigzag chains and 1D $[\text{M-ADTZ}]_{2n}$ double-chains.	10
18	PADLAK	$[\text{Cd}(\text{L})(\text{ADTZ})]\cdot \text{H}_2\text{O}$	$\text{H}_2\text{ADTZ}=2,5\text{-}(s\text{-acetic acid})\text{dimercapto-1,3,4-thiadiazole}$	Anti	$C_1^1(17)$ NH...O (ADTZ)	$\text{H}_2\text{O}$	distorted octahedral geometry	2D double-layer networks based on 1D $[\text{M-L}]_n$ zigzag chains and 1D $[\text{M-ADTZ}]_{2n}$ double-chains	10
19	PADLEO	$[\text{Zn}(\text{L})(\text{ADTZ})]\cdot \text{H}_2\text{O}$	$\text{H}_2\text{ADTZ}=2,5\text{-}(s\text{-acetic acid})\text{dimercapto-1,3,4-thiadiazole}$	Anti	$D_1^1(2)$ NH...O ( $\text{H}_2\text{O}$ )	$\text{H}_2\text{O}$	distorted octahedral geometry	2D double-layer networks based on 1D $[\text{M-L}]_n$ zigzag chains and 1D $[\text{M-ADTZ}]_{2n}$ double-chains.	10
20	KIJWUX	$[\text{Cd}(\text{suc})(3\text{-pna})\cdot 2.5\text{H}_2\text{O}]n$	succinic acid ( $\text{H}_2\text{suc}$ )	Syn	$D_1^1(2)$ NH...O ( $\text{H}_2\text{O}$ )	$\text{H}_2\text{O}$	distorted pentagonal bipyramidal	simple 3-D non-interpenetrated network	11
21	KIJWUX 01	$[\text{Cd}(\text{suc})(3\text{-pna})\cdot 2.5\text{H}_2\text{O}]n$	succinic acid ( $\text{H}_2\text{suc}$ )	Syn	$D_1^1(2)$ NH...O ( $\text{H}_2\text{O}$ )	$\text{H}_2\text{O}$	distorted pentagonal bipyramid	simple 3-D non-interpenetrated network	11
22	MOVPUI	$\{[\text{Co}(\text{NCS})_2(3\text{-pna})_2]\}n$	-	Anti	-	-	octahedral	2D sheet composed of a nearly square grid.	12
23	MULQOB	$[\text{Co}(\text{tpdc})(3\text{-pna})]n$	2,5-thiophenedicarboxylic acid ( $\text{H}_2\text{tpdc}$ )	Anti	-	-	distorted trigonal bipyramidal	2-D layer structure containing $\{\text{Co}_2(\text{OCO})_2\}$ dimers linked by tpdc and 3-pna ligands.	13
24	MULQU H	$[\text{Co}(\text{D-cam})(3\text{-pna})(\text{H}_2\text{O})_2]n$	chiral D-camphorate (D-cam)	Syn	-	-	distorted octahedron	Acentric ribbon coordination polymer motifs.	13
25	MULRA O	$[\text{Ni}(\text{D-cam})(3\text{-pna})(\text{H}_2\text{O})_2]n$	chiral D-camphorate (D-cam)	Syn	-	-	distorted octahedron	Acentric ribbon coordination polymer motifs.	13
26	PORXIF	$\{[\text{Cd}(\text{msuc})(3\text{-pna})]\cdot 2\text{H}_2\text{O}\}n$	(msuc = 2-methylsuccinate)	Anti	$D_1^1(2)$ NH...O ( $\text{H}_2\text{O}$ )	$\text{H}_2\text{O}$	distorted pentagonal bipyramidal	Layer motifs that contain $\{\text{Cd}_2(\text{OCO})_2\}$ dimeric units, linked into a non-interpenetrated 3D network.	14

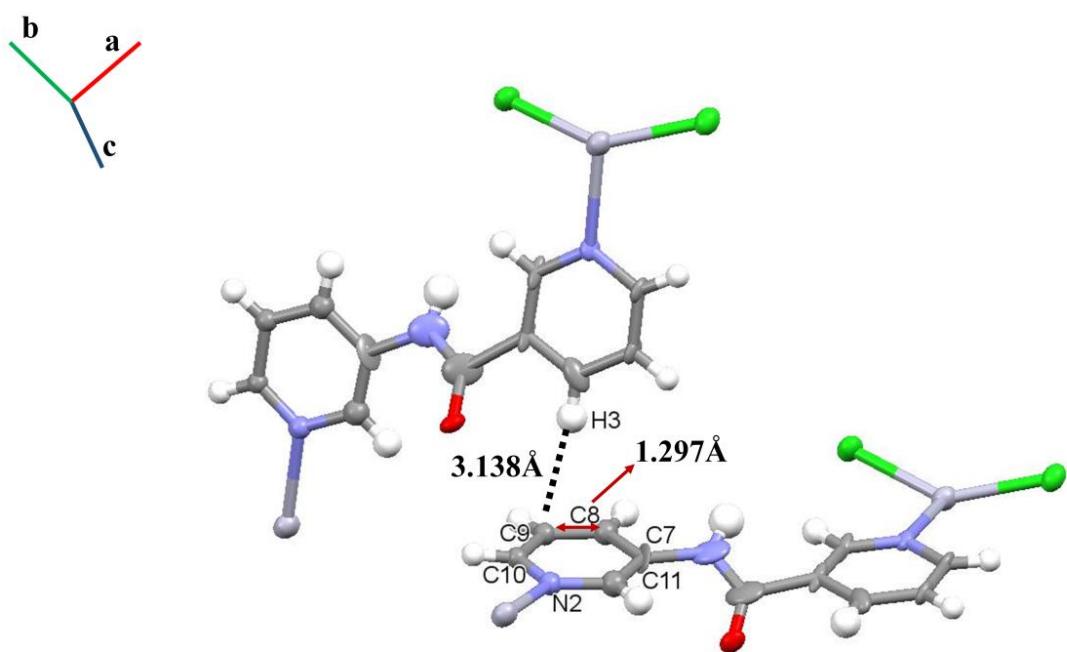
27	PORXOL	$[\text{Cd}2(\text{dmsuc})2(3\text{-pna})_2 \cdot 1.5\text{H}_2\text{O}]n$	(dmsuc = 2,2-dimethylsuccinate)	Anti	$S_1^1(10)$ $\text{NH}\dots\text{O}$ (dmsuc)	$\text{H}_2\text{O}$	distorted octahedral coordination	1D $[\text{Cd}(\text{dmsuc})]_n$ ribbons containing $\{\text{Cd}_2\text{O}_2\}$ dimeric units.	14
28	QISYEY	$\{[\text{Zn}2(\text{hmph})2(3\text{-pna})]\cdot\text{H}_2\text{O}\}n$	hmph = homophthalate	Syn	$D_1^1(2)$ $\text{NH}\dots\text{O}$ ( $\text{H}_2\text{O}$ )	$\text{H}_2\text{O}$	distorted square-pyramidal / distorted tetrahedral	(4,4) grid coordination polymer layers.	15
29	QIYHAJ	$\{[\text{Zn}(\text{tere})(3\text{-pna})_2(\text{H}_2\text{O})_2] \cdot 2\text{H}_2\text{O}\}n$	potassium terephthalate (K2tere)	Syn	$D_1^1(2)$ $\text{NH}\dots\text{O}$ ( $\text{H}_2\text{O}$ )	$\text{H}_2\text{O}$	octahedral	1-D coordination polymer chains.	16
30	QIXWUR	$\{[\text{Zn}(\text{tere})(3\text{-pna})] \cdot 3\text{H}_2\text{O}\}n$	potassium terephthalate (K2tere)	Syn	$D_1^1(2)$ $\text{NH}\dots\text{O}$ ( $\text{H}_2\text{O}$ )	$\text{H}_2\text{O}$	pseudo-tetrahedral	2-D (6,3) hexagonal grid topology	16
31	SIXTUP	$\{(\text{CH}_3\text{COO})(\mu\text{-OOCH}_3)\text{Zn}\}_2(3\text{-pna})_2\}n$	-	Anti	-	-	slightly distorted trigonal bipyramidal	1-D zigzag polymeric chain / ladder type architecture.	17
32	SIXVAX	$\{(\text{H}_2\text{O})_2\text{Zn}\}(3\text{-pna})_4\text{BF}_4\}n$	-	Syn	$C_1^1(9)$ $\text{O}\dots\text{HO}$ (coordinated $\text{H}_2\text{O}$ )	$\text{BF}_4$	slightly distorted octahedral	2-D nonporous corrugated sheet.	17
33	SIXVEB	$\{(\text{Cl})_2\text{Zn}\}(3\text{-pna})_n$	-	Anti	$R_2^2(14)$ $\text{NH}\dots\text{Cl}$	-	distorted tetrahedral	1-D zigzag coordination polymeric chain.	17
34	SIXVOL	$\{(\text{NO}_3)_2\text{Cd}(3\text{-pna})_2\} \cdot \text{nitrobenzene}\}n$	-	Anti	-	$\text{Ph}\text{-NO}_2$	slightly distorted octahedral	2-D grid architecture 2-D porous sheet.	17
35	VIGYUH	$[\text{Cu}(\text{t-14cdc})(3\text{-pna})]n$	(t-14cdc) = trans1,4-cyclohexanedicarboxylate	Anti	$C_1^1(9)$ $\text{NH}\dots\text{O}$ (t-14cdc)	-	Jahn-Teller distorted square pyramidal	Non-interpenetrated 3-D coordination polymer network	18
36	VIPTIY	$\{(\text{H}_2\text{O})_4\text{Co}(3\text{-pna})_2\} \cdot \text{fumarate.2H}_2\text{O}\}n$	-	Syn	$C_1^1(9)$ $\text{O}\dots\text{HO}$ (coordinated $\text{H}_2\text{O}$ )	$\text{H}_2\text{O}$ fumarate	slightly distorted octahedral	1-D zigzag polymeric chain.	19
37	VIPTOE	$\{(\text{H}_2\text{O})_3(\mu\text{-fumarate})\text{Co}(\mu\text{-L1})_2\} \cdot \text{fumarate.2H}_2\text{O}\}n$	L1= 3-pna	Anti	$D_1^1(2)$ $\text{NH}\dots\text{O}$ (fumarate)	$\text{H}_2\text{O}$ fumarate	slightly distorted octahedral	ladder type 1-D polymeric network.	19
38	WIZMUP	$[\text{CuMoO}_4(3\text{-pna})]n$	MoO4	Anti	$R_2^2(18)$ $\text{NH}\dots\text{O}$ ( $\text{MoO}_4$ )	-	distorted square pyramid/	2-D $[\text{CuMoO}_4(3\text{-pna})]n$ coordination polymer	20

								distorted tetrahedron		
39	WUDGU Z	[Zn(ip)(3-pna)] <i>n</i>	ip= isophthalate	Anti	-	-	distorted trigonal bipyramidal	Dimer-based 3,5-connected layer motif	21	
40	WUDHA G	{[Zn(mip)(3-pna)].2H <sub>2</sub> O} <i>n</i>	mip = 5-methylisophthalate	Syn	<i>D</i> <sub>1</sub> <sup>1</sup> (2) NH...O (H <sub>2</sub> O)	H <sub>2</sub> O	distorted tetrahedral	4-connected 1-D ribbon Motif.	21	
41	XAFBUD	[(Cd(3-pna )( $\mu$ 3- SO <sub>4</sub> )(H <sub>2</sub> O). 2H <sub>2</sub> O] <i>n</i>	-	Anti	<i>S</i> <sub>1</sub> <sup>1</sup> (10) NH...O ( $\mu$ 3SO <sub>4</sub> )	H <sub>2</sub> O	slightly distorted octahedral	2D coordination polymer	22	
42	XIFJAZ	{[Cd <sub>2</sub> (adp)(nic)2(3-pna)2] .6H <sub>2</sub> O} <i>n</i>	H <sub>2</sub> adp = adipic acid nic = nicotinate	Syn	<i>D</i> <sub>1</sub> <sup>1</sup> (2) NH...O (H <sub>2</sub> O)	H <sub>2</sub> O	pentagonal bipyramidally	3-D network 3-D coordination polymer	23	

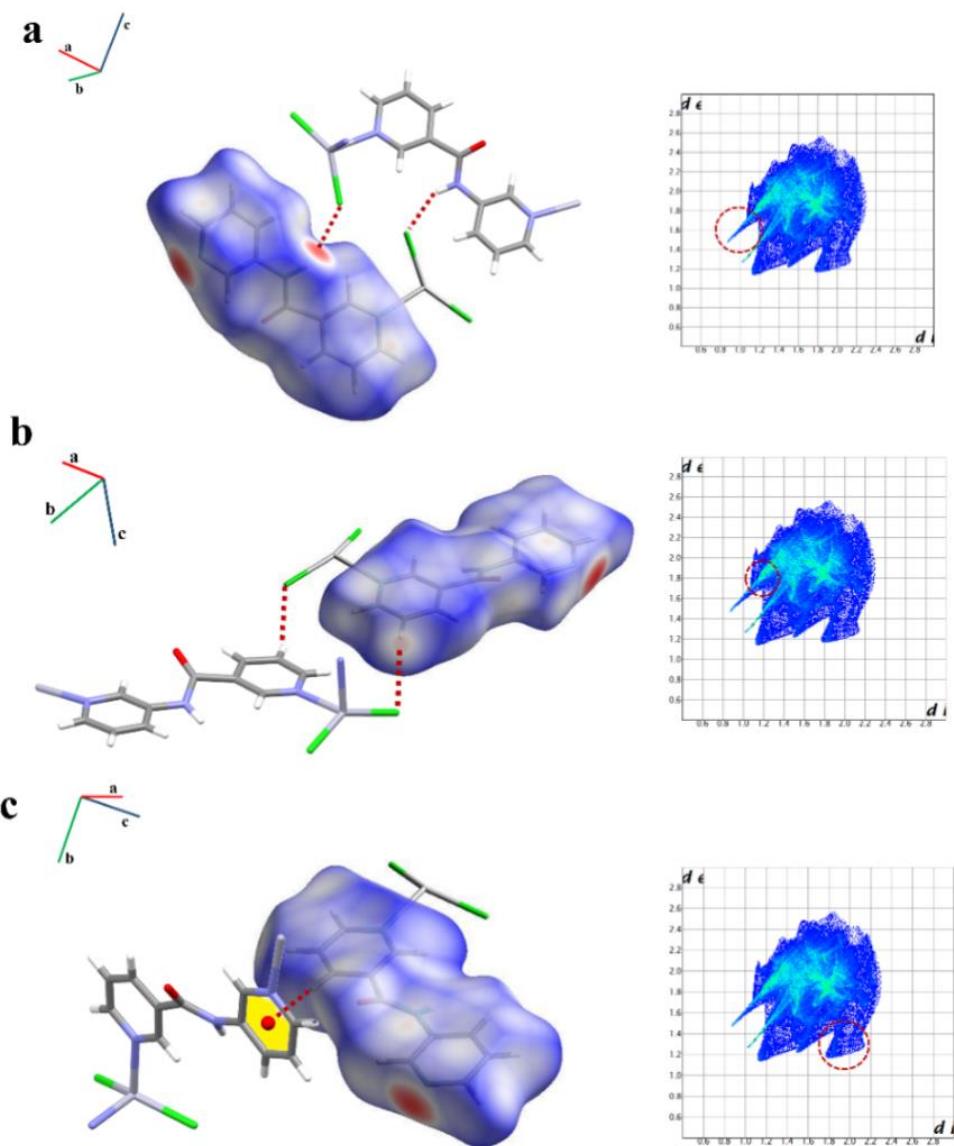
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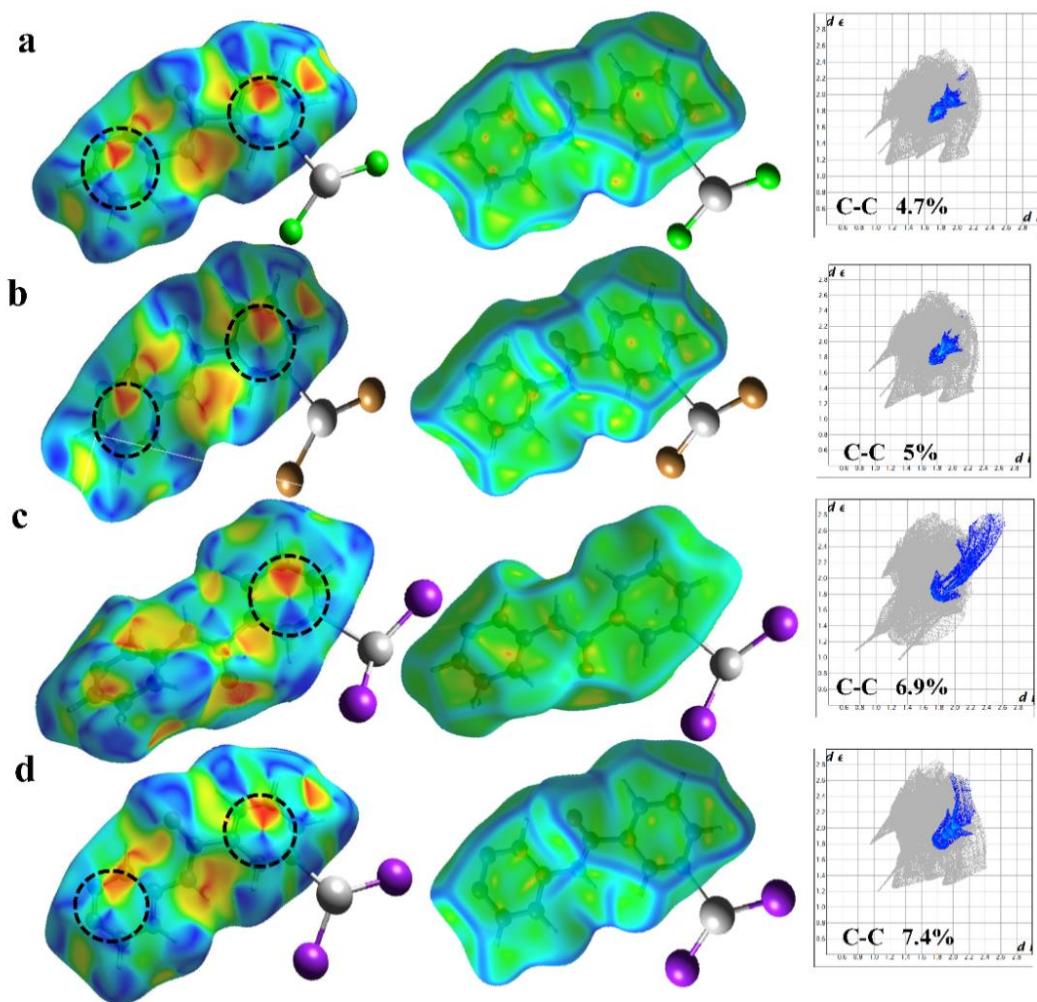
**Fig. S1** Hirshfeld surface mapped with  $d_{norm}$  (top, left), decomposed fingerprint plot for compound L (top, right), showing (a) NH $\cdots$ N and CH $\cdots$ N hydrogen bonds and Hirshfeld surfaces mapped with shape index (red and blue triangles has been shown by black circle), curvedness and decomposed finger print plots to C-C contacts (bottom, left to right) showing relative contribution of  $\pi\cdots\pi$  stacking interactions for compound L.



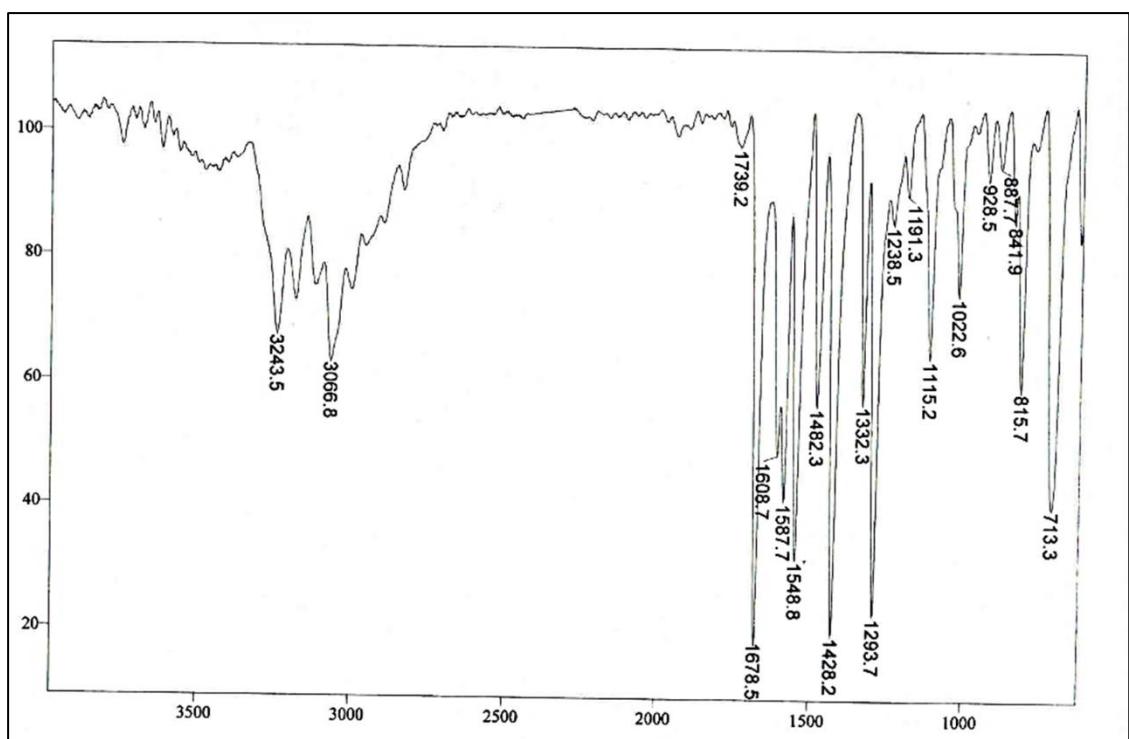
**Fig. S2** Representation of CH... $\pi$  interaction in compound 1, showing nearest bonds to CH group.



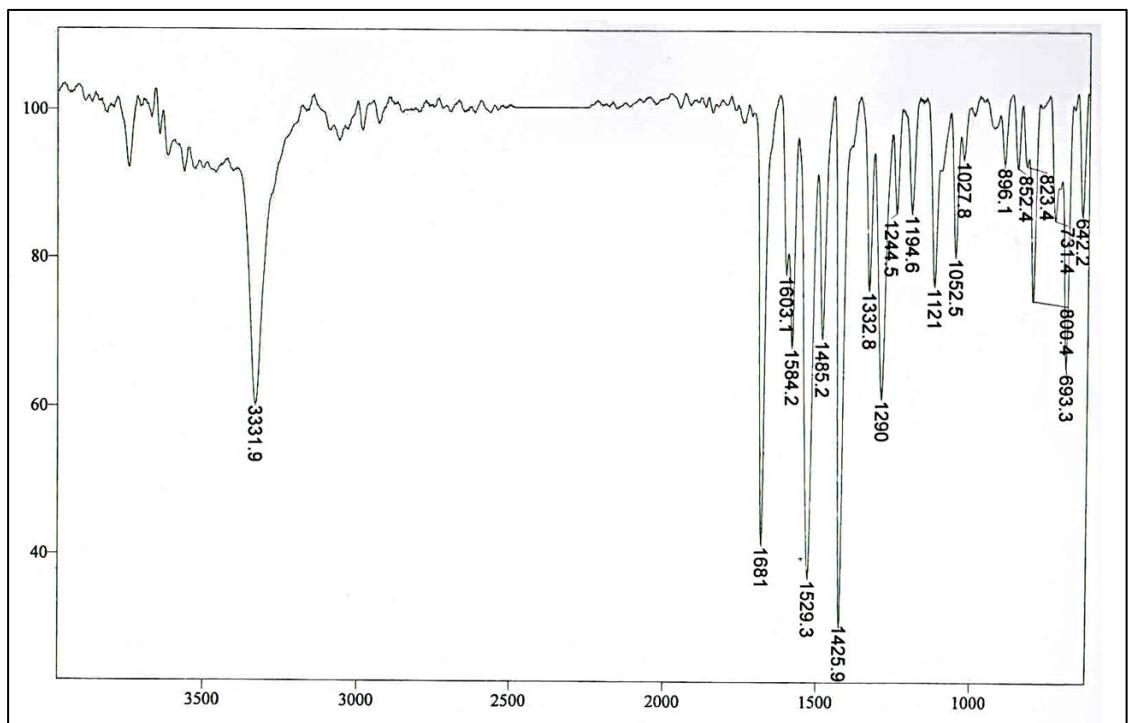
**Fig. S3** Hirshfeld surface mapped with  $d_{norm}$  (left) and two-dimensional fingerprint plot for compound 1, showing (a) N-H...Cl, (b) C-H...Cl and (c) CH... $\pi$  hydrogen bonds.



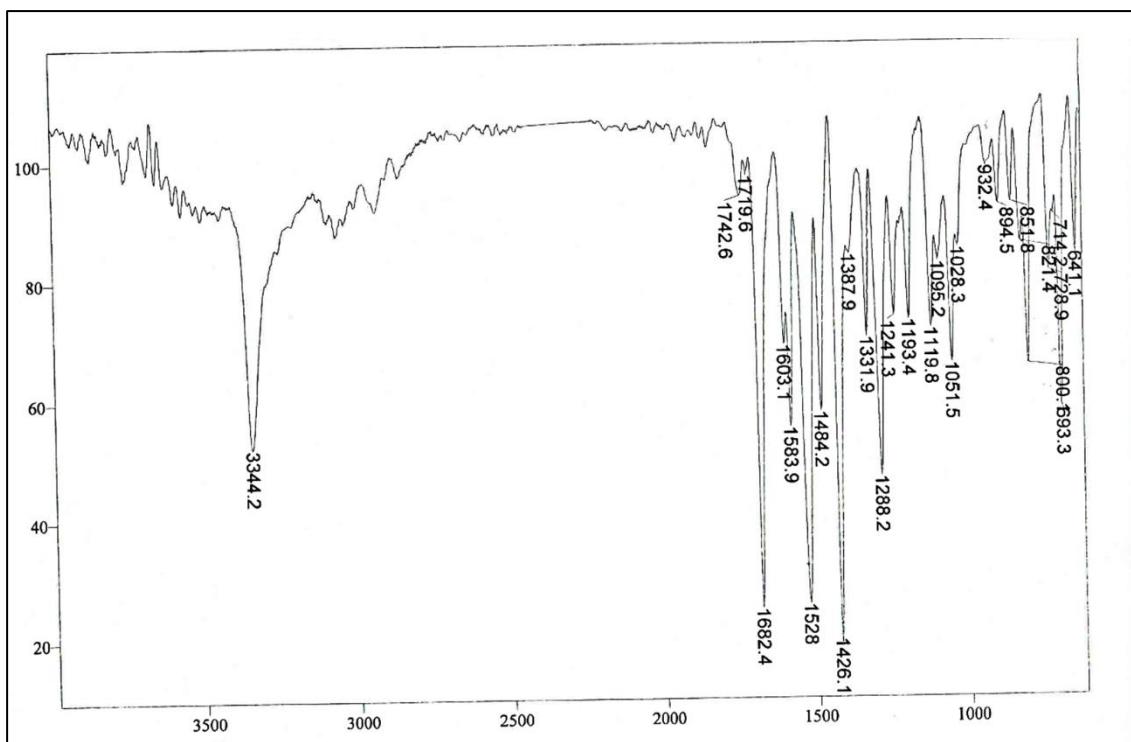
**Fig. S4** Hirshfeld surfaces mapped with shape index (red and blue triangles has been shown by black circle), curvedness and decomposed finger print plots to C-C contacts showing relative contribution of  $\pi \cdots \pi$  stacking interactions for compound 1(a), 2(b), 3(c) and 4(d) (left to right).



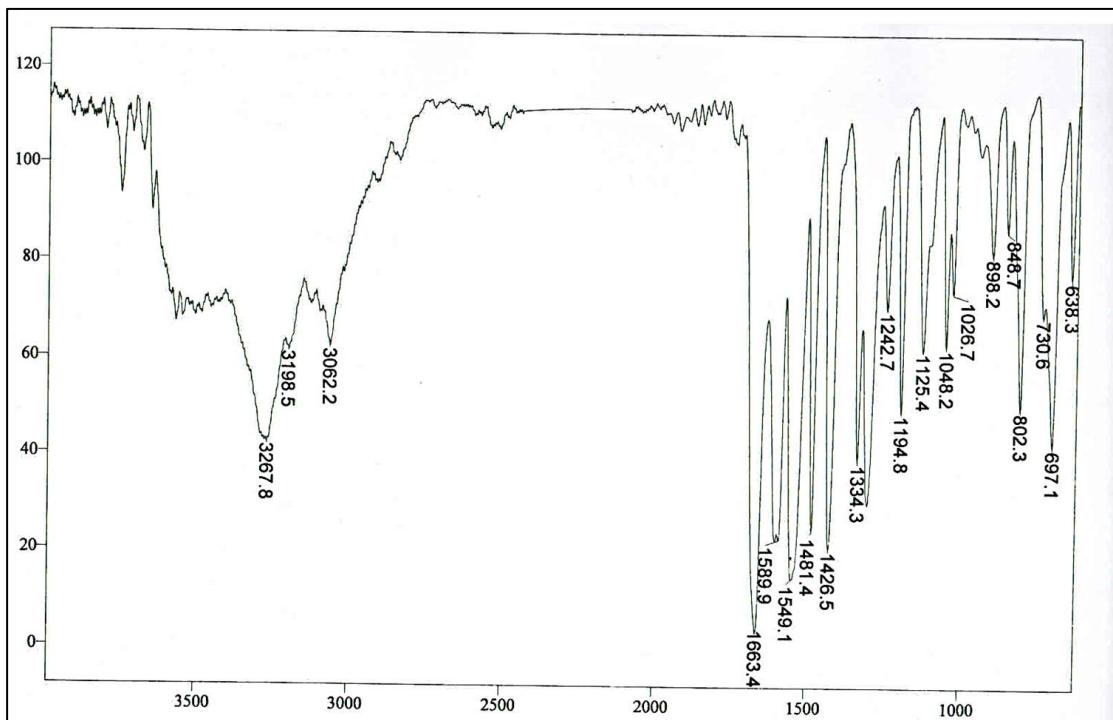
**Fig. S5** IR Spectrum of ligand



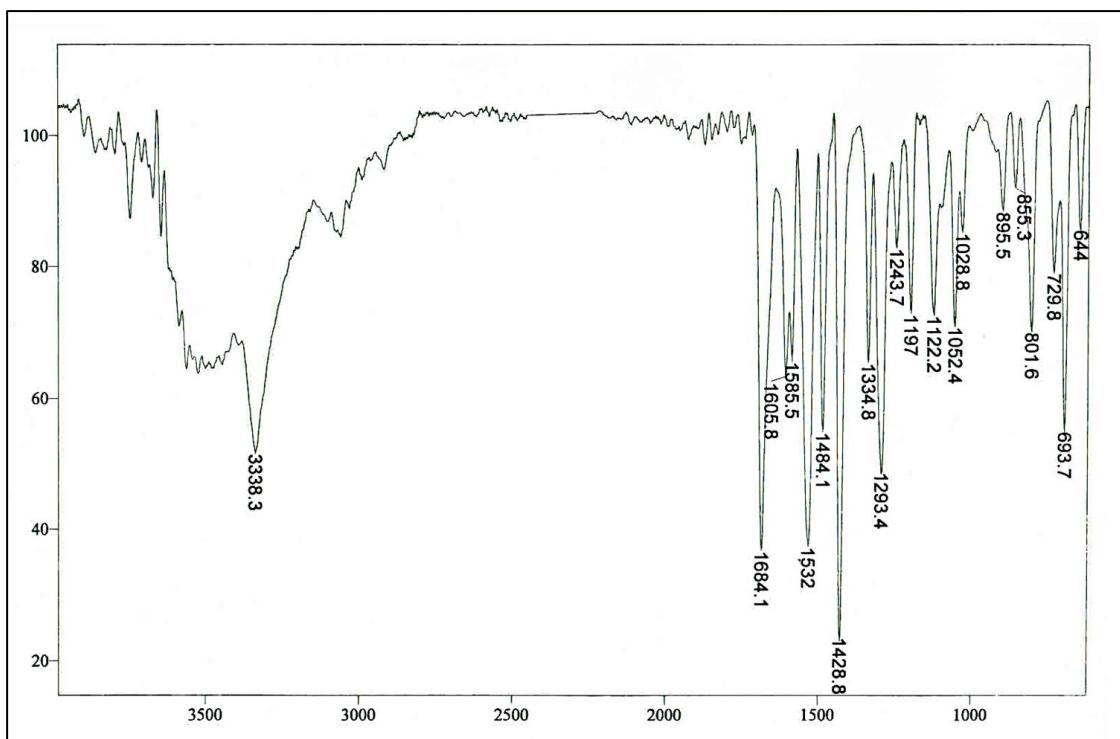
**Fig. S6** IR Spectrum of compound 1 (solution-method)



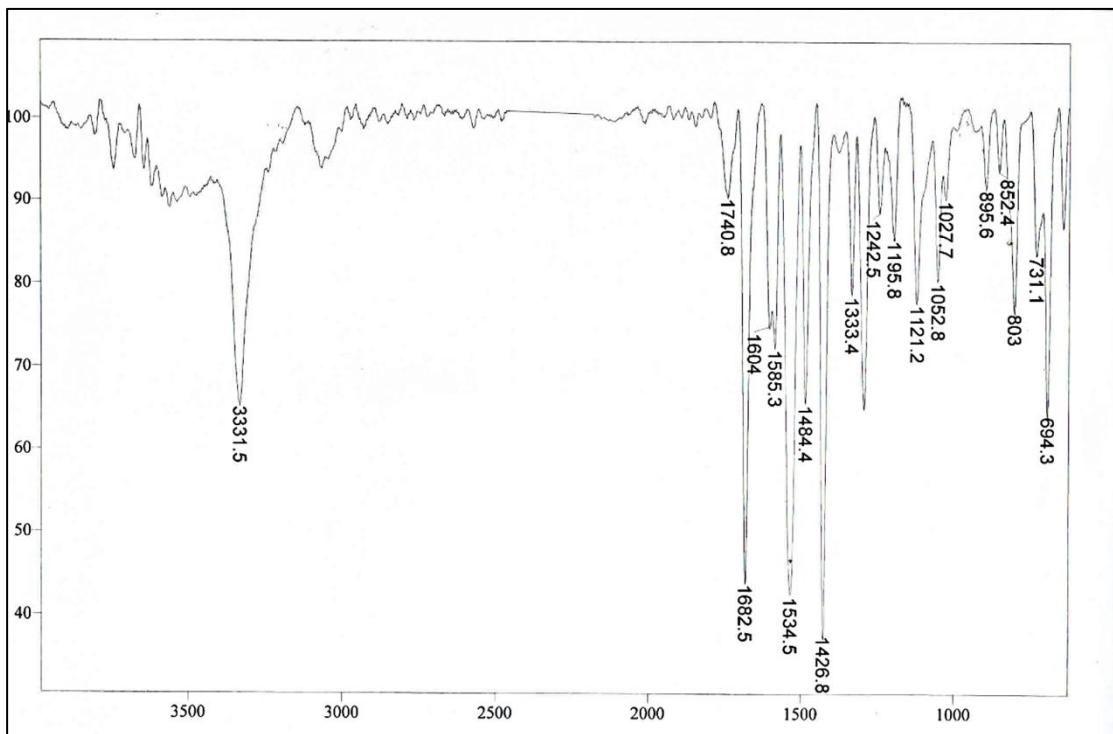
**Fig. S7** IR Spectrum of compound 2 (solution-method)



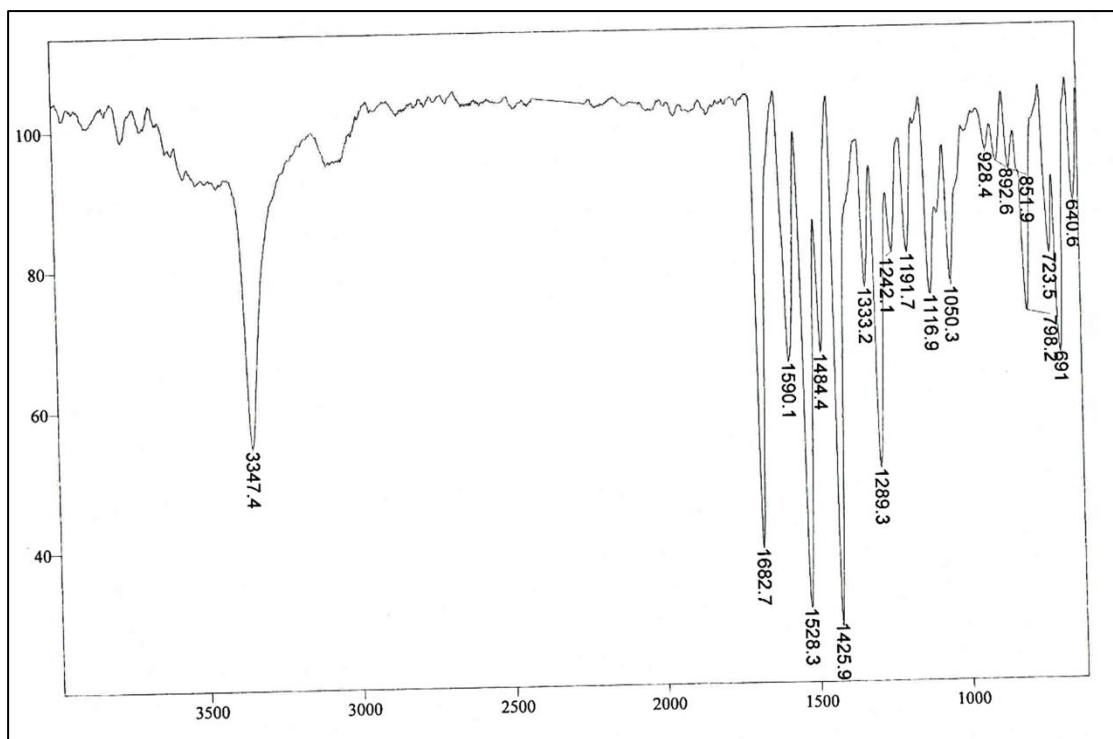
**Fig. S8** IR Spectrum of compound 3 (solution-method)



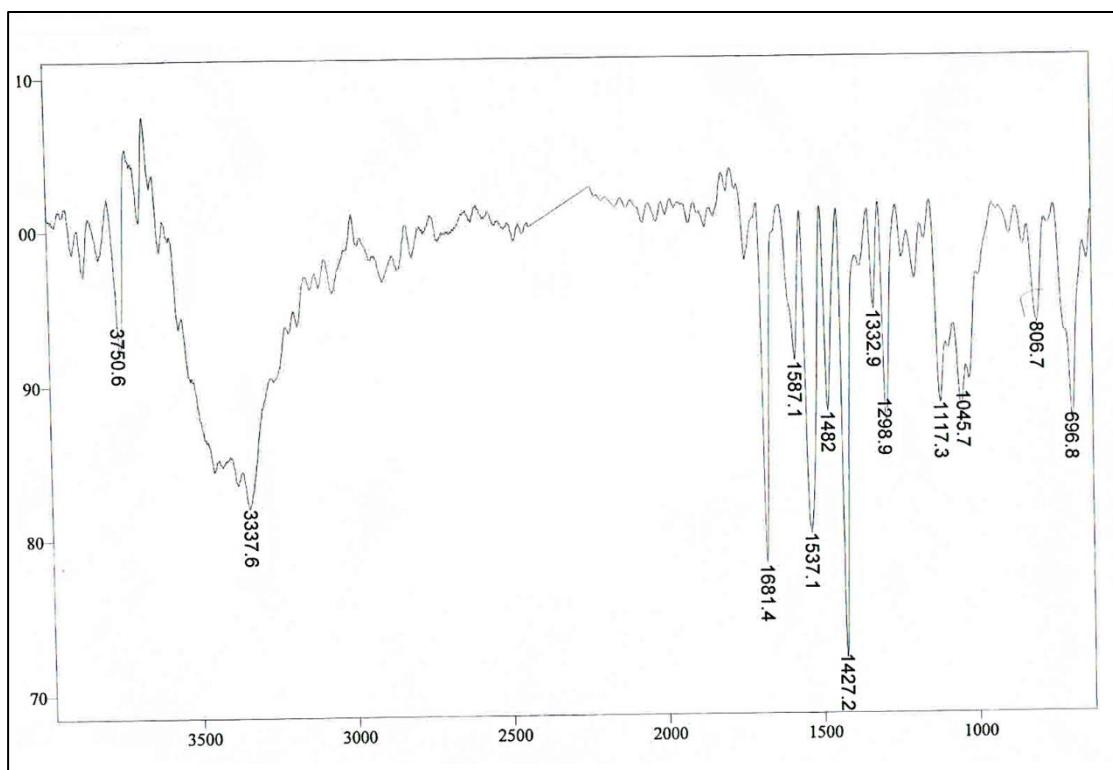
**Fig. S9** IR Spectrum of compound 4 (solution-method)



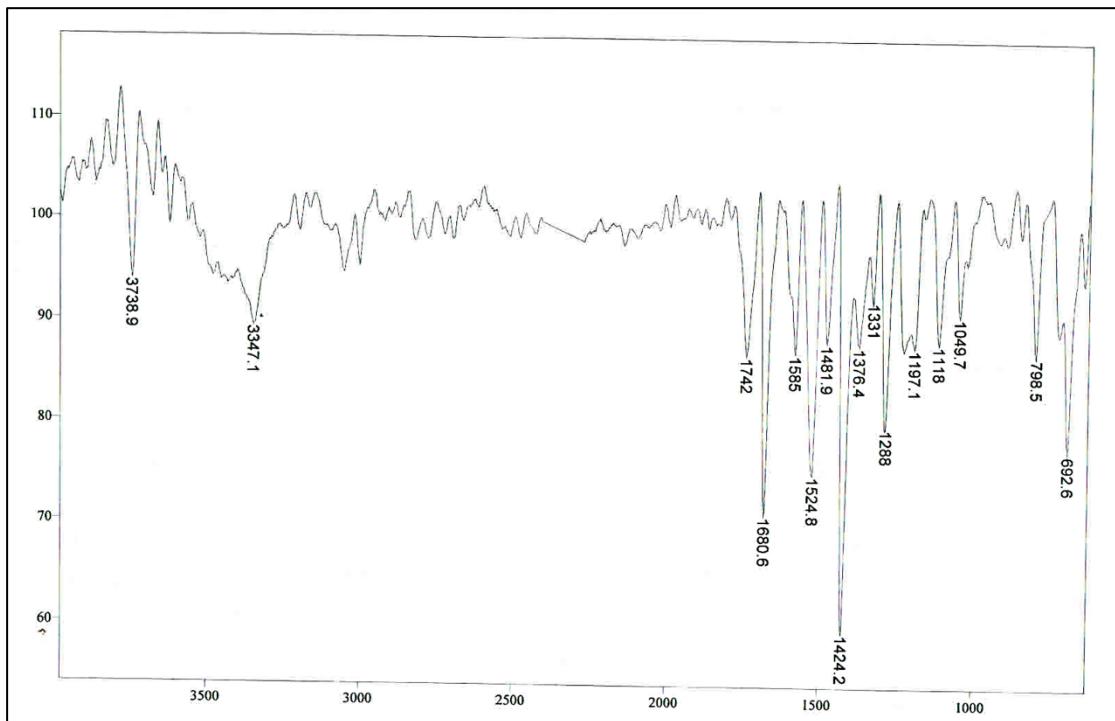
**Fig. S10** IR Spectrum of compound 1 (Grinding method)



**Fig. S11** IR Spectrum of compound 2 (Grinding method)



**Fig. S12** IR Spectrum of compound 1 (Thermal method)



**Fig. S13** IR Spectrum of compound 2 (Thermal method)

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