

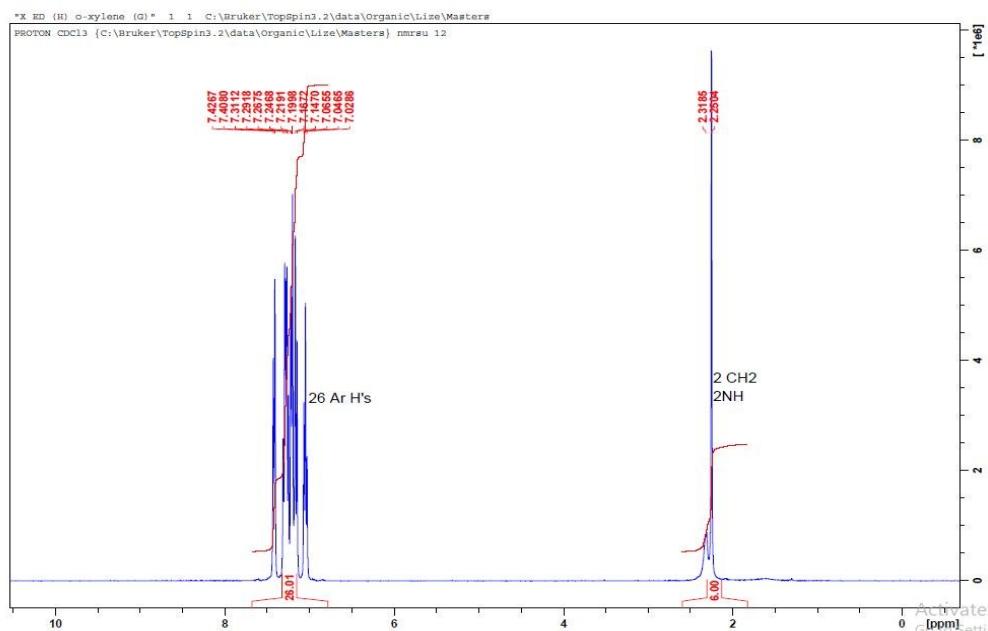
Supplementary Information

Minor modifications afford improved host selectivities in xanthenyl-type host systems

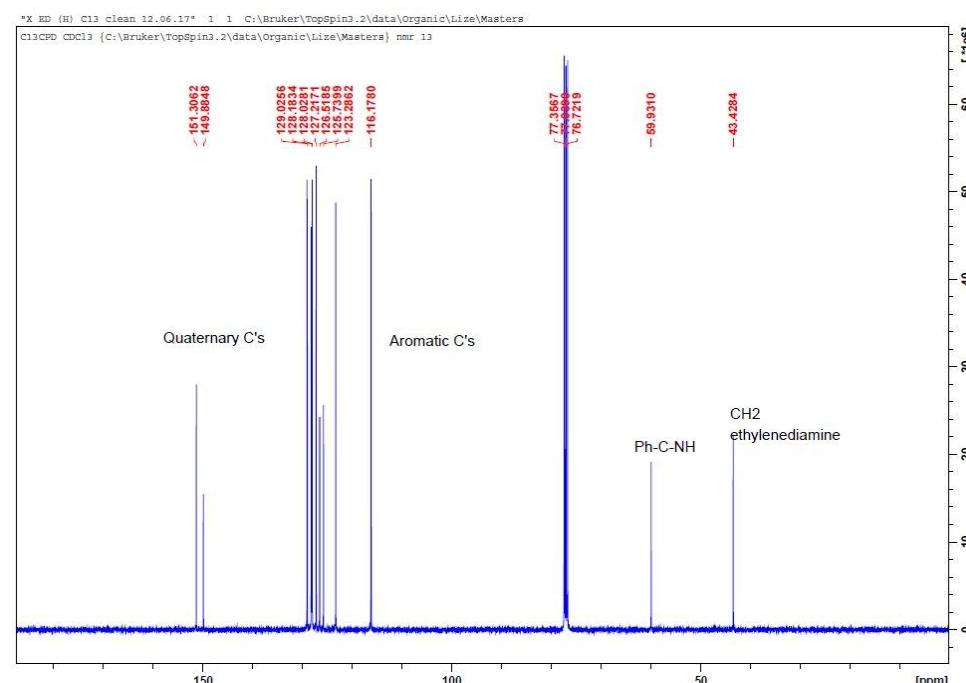
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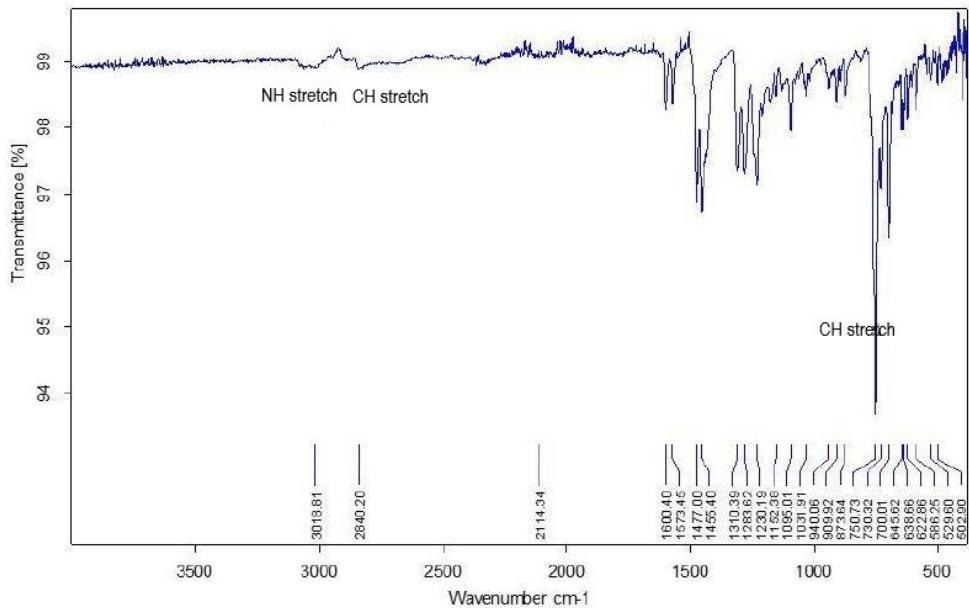
(a)



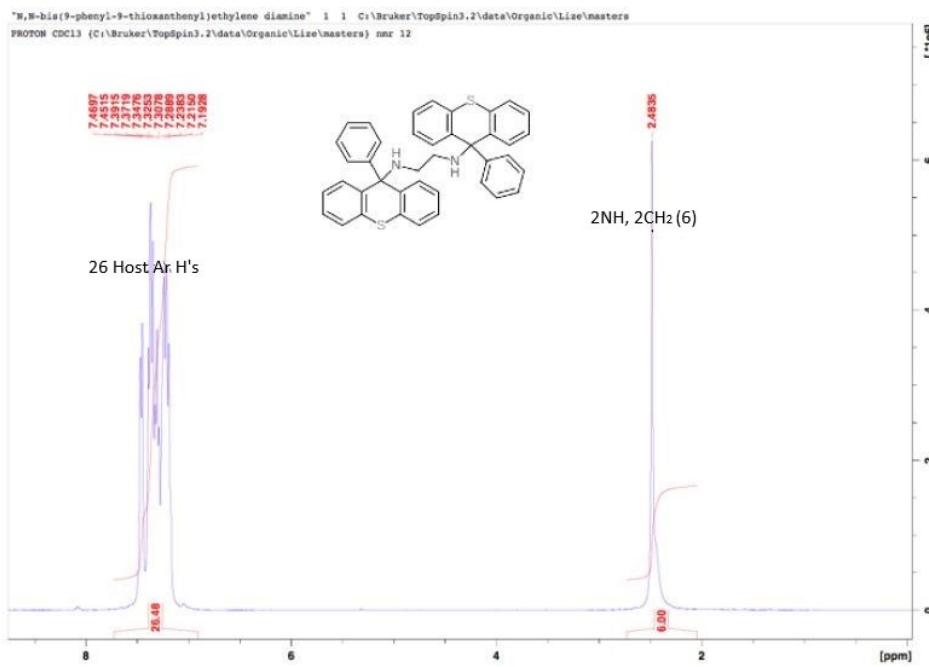
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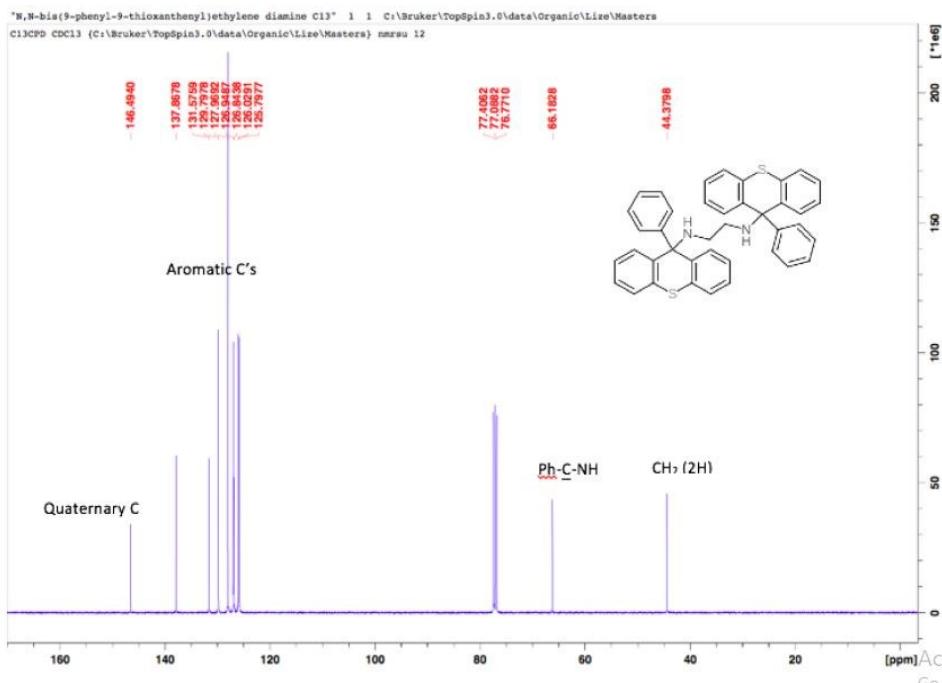
(c)



(d)



(e)



(f)

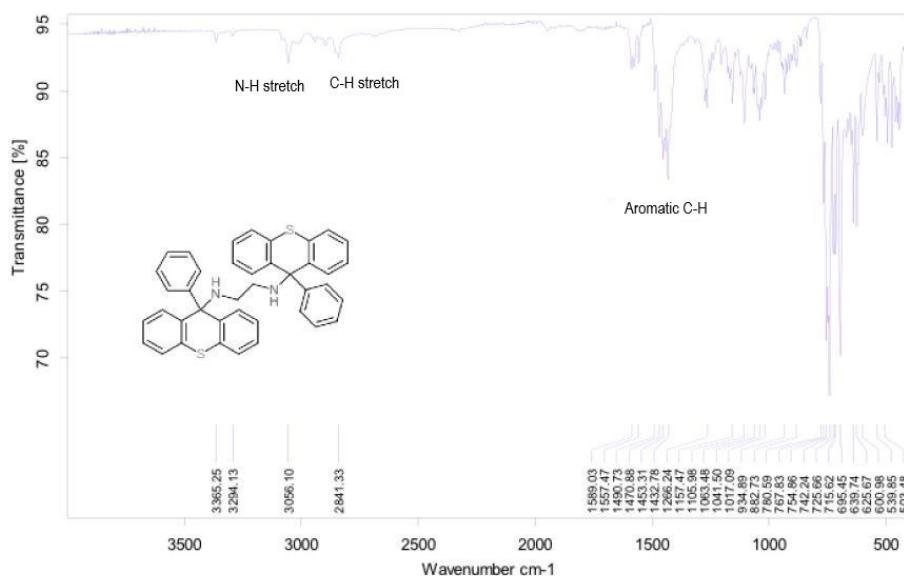
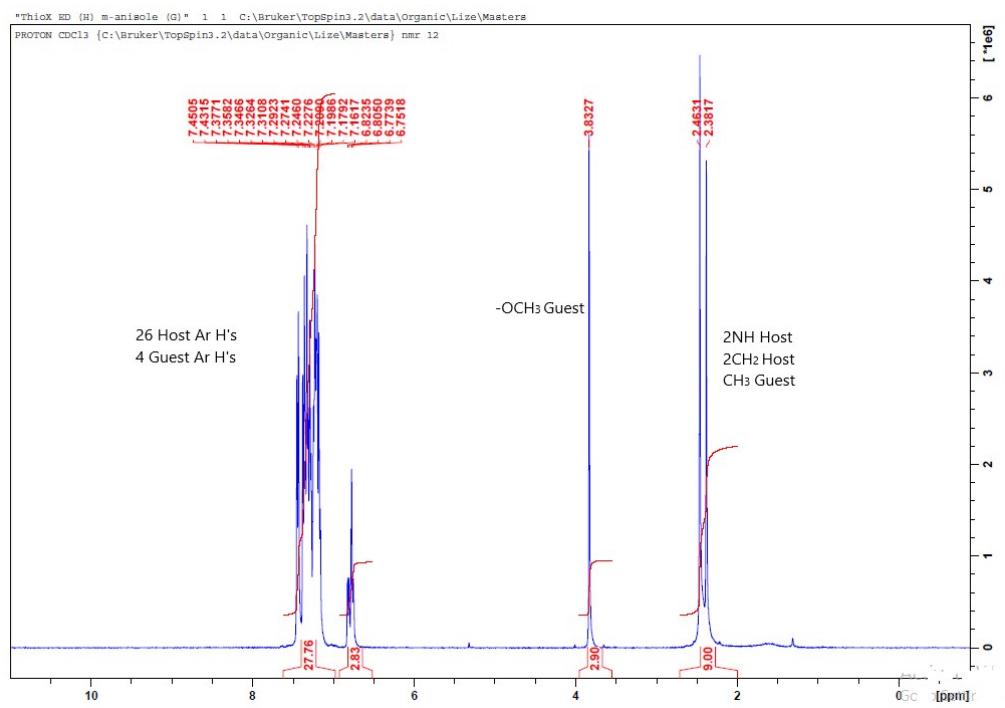
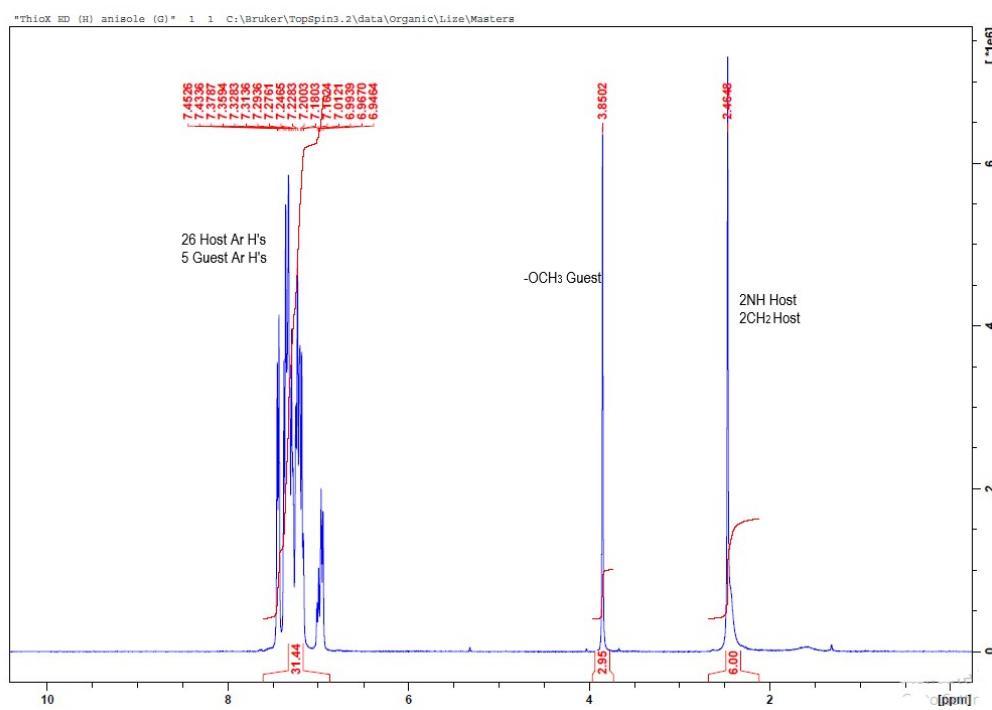
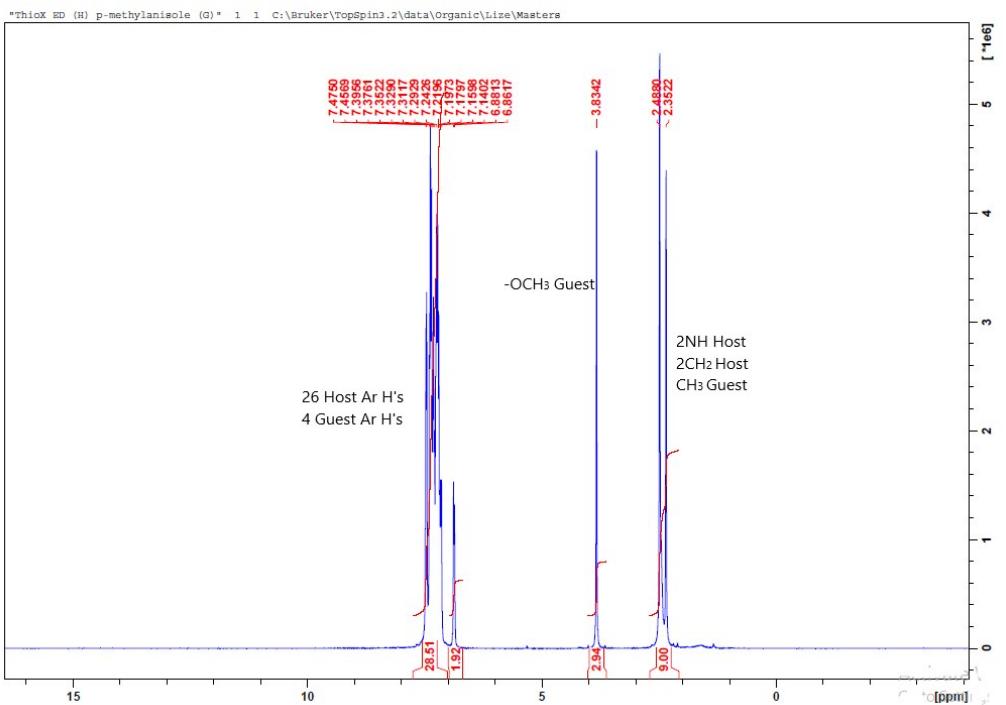


Figure S1. Characterization spectra for (a) ¹H-NMR spectrum for H₂, (b) ¹³C-NMR spectrum for H₂, (c) IR spectrum for H₂, (d) ¹H-NMR spectrum for H₁, (e) ¹³C-NMR spectrum for H₁ and (f) IR spectrum for H₁.

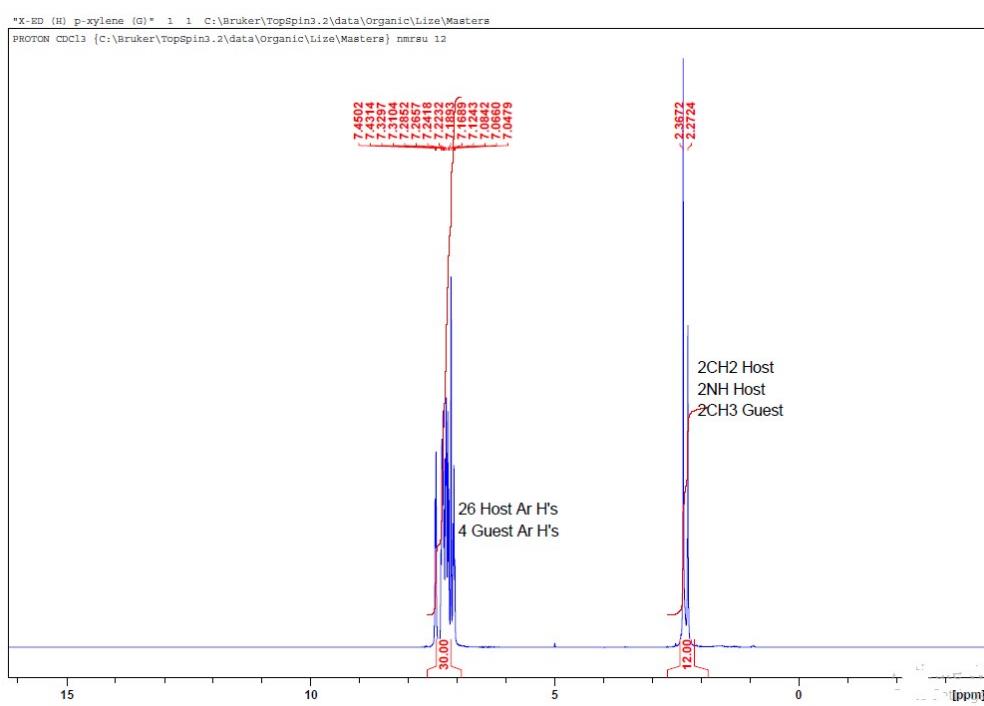
(a)



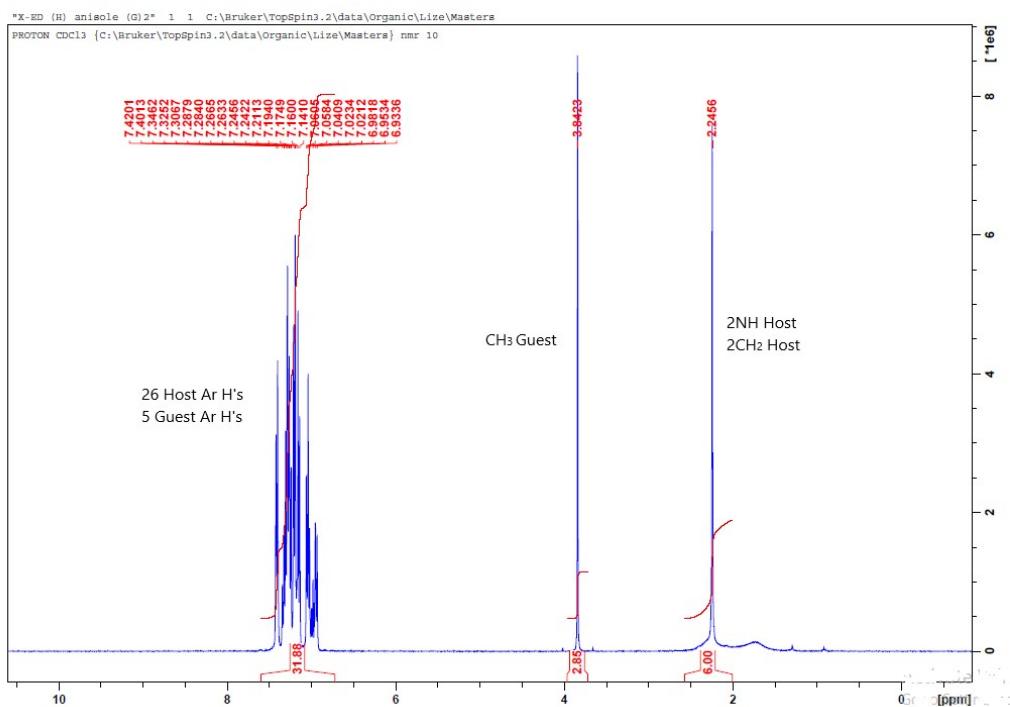
(c)



(d)



(e)



(f)

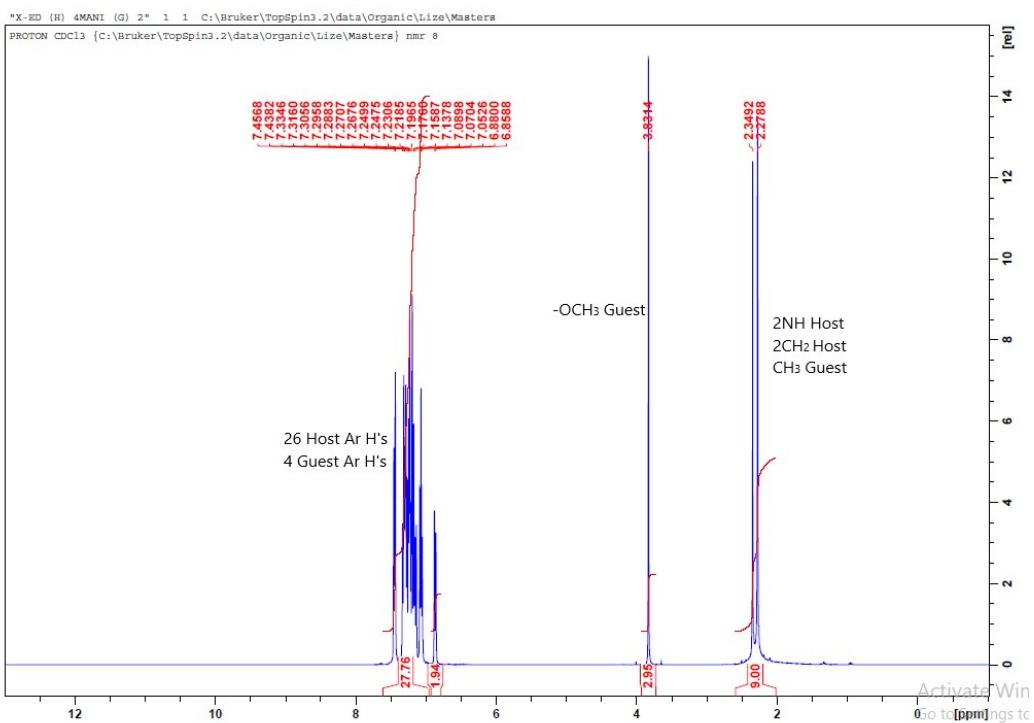


Figure S2. Individual inclusion spectra for (a) ¹H-NMR for ANI with H₁, (b) ¹H-NMR for 3MANI with H₁, (c) ¹H-NMR for 4MANI with H₁, (d) ¹H-NMR for p-Xy with H₂, (e) ¹H-NMR for ANI with H₂ and (f) ¹H-NMR for 4MANI with H₂.

Table S3. (a) Duplicate data for equimolar competition experiments of anisole and methylanisole isomers with \mathbf{H}_1

Combination:	Batch 1	Batch 2	Average	e.s.d.'s
ANI 2MANI	98.29:1.71	94.09:5.91	96.19:3.54	(2.10): (2.10)
ANI 3MANI	54.40: 45.60	57.88:42.12	56.14:43.86	(1.74): (1.74)
ANI 4MANI	35.59:64.42	36.04:63.97	35.82:64.20	(0.23): (0.23)
23 MANI	11.58:88.42	14.91:85.09	13.25:86.76	(1.67): (1.67)
34 MANI	40.14:59.86	40.56:59.44	40.35:59.65	(0.21): (0.21)
24 MANI	3.75:96.25	4.18:95.82	3.79:96.04	(0.22): (0.22)
ANI 23MANI	35.10:8.12:56.78	38.56:7.89:53.55	36.83:8.01:55.17	(1.73): (0.12): (1.62)
ANI 34MANI	27.36:32.73:39.91	29.54:29.97:40.49	28.45:31.38:40.20	(1.09): (1.41): (0.29)
ANI 24MANI	36.31:4.51:59.18	35.69:3.03:61.28	36.00:3.77:60.23	(0.31): (0.74): (1.05)
234 MANI	4.50:40.89:54.60	3.12:43.39:53.46	3.18:42.14:54.03	(0.69): (1.25): (0.57)
ANI 234MANI	31.06:2.88:19.59:46.47	27.02:3.04:24.10:45.85	29.04:2.96:21.85:46.16	(2.02): (0.08): (2.26): (0.31)

Table S3. (b) Duplicate data for equimolar competition experiments of anisole and methylanisole isomers with \mathbf{H}_2

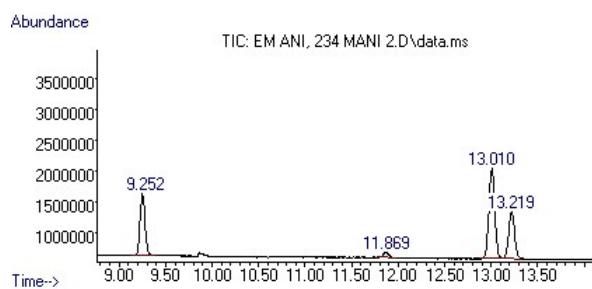
Combination:	Batch 1	Batch 2	Average	e.s.d.'s (%)
ANI 2MANI	97.47:2.54	95.07:4.93	96.27:3.73	(1.20): (1.20)
ANI 3MANI	91.80:8.20	89.91:10.09	90.86:9.14	(0.95): (0.95)
ANI 4MANI	14.54:85.46	14.08:85.92	14.31:85.69	(0.23): (0.23)
23 MANI	^a	^a	-	-
34 MANI	6.55:93.45	6.20:93.80	6.37:93.63	(0.18): (0.18)
24 MANI	3.88:96.12	3.50:96.50	3.69:96.31	(0.19): (0.19)
ANI 23MANI	^a	^a	-	-
ANI 34MANI	11.25:4.77:83.98	9.81:4.17:86.02	10.53:4.47:85.00	(0.72): (0.30): (1.02)
ANI 24MANI	10.11:1.76:88.13	11.73:2.36:85.91	10.92:2.06:87.02	(0.81): (0.30): (1.11)
234 MANI	1.25:3.95:94.80	2.17:5.65:92.18	1.71:4.80:93.49	(0.46): (0.85): (1.31)
ANI 234MANI	11.58:1.15:5.28:81.99	11.09:1.76:3.89:83.26	11.34:1.46:4.59:82.63	(0.25): (0.31): (0.70): (0.64)

^aDid not crystallize**Table S3. (c)** Duplicate data for equimolar competition experiments of ethylbenzene and xylene isomers with \mathbf{H}_2

Combination:	Batch 1	Batch 2	Average	e.s.d.'s
p,o-xylene	96.66:3.34	96.79:3.21	96.73:3.27	(0.07): (0.07)
p,m-xylene	96.09:3.91	96.39:3.61	96.24:3.76	(0.15): (0.15)
m,o-xylene	^a	^a	-	-
p,m,o-xylene	96.85:1.60:1.55	96.12:2.16:1.72	96.49: 1.88: 1.63	(0.37): (0.28): (0.09)
p-xyl and EB	91.94:8.06	92.63:7.37	92.29:7.71	(0.35): (0.35)
o-xyl and EB	^a	^a	-	-
m-xyl and EB	^a	^a	-	-
p,o-xyl and EB	92.77:1.89:5.35	92.75:2.18:5.08	92.76: 2.04:5.22	(0.01): (0.15): (0.14)
p,m-xyl and EB	90.94:2.85:6.22	92.02:1.97:6.01	91.48:2.41:6.12	(0.54): (0.44): (0.11)
o,m-xyl and EB	^a	^a	-	-
p,m,o-xyl and EB	91.42:1.51:1.86:5.21	92.61:1.04:1.11:5.25	92.02:1.28:1.49:5.23	(0.60): (0.24): (0.38): (0.02)

^aNo inclusion occurred

(a)



(b)

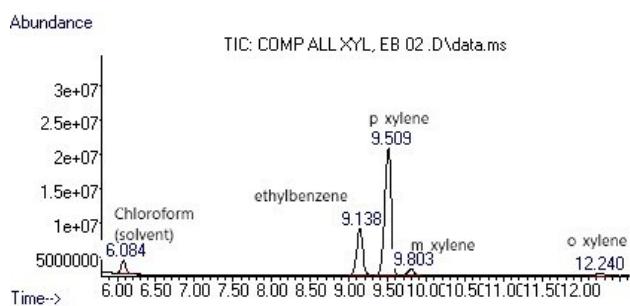


Figure S4 Chromatograph of standards for guests (a) anisoles and methylanisole isomers and (b) ethylbenzene and xylene isomers.

Table S5. (a) All interactions for H₁ with anisole and methylanisole isomers

Non-covalent interaction	H ₁ •ANI	H ₁ •3MANI	H ₁ •4MANI	Symmetry
π···π (H···H and H···G)	4.641(1)–5.989(1) Å	4.527(9)–5.9691(9) Å	4.571(1)–5.998(1) Å	
H···G major	[7] 4.909(1)–5.986(1) Å ^a	[7] 5.028(1)–5.757(1) Å	[6] 4.964(2)–5.535(2) Å	
H···G minor			[7] 5.220(2)–5.869(2) Å	
CH···π (host–host and host–guest)				
C _(H) –H _(H) ···Cg _(H)	2.85 Å, 144°			x, y, 1+z
C _(G) –H _(G) ···Cg _(H)		2.78 Å, 170°		1–x, 2–y, 1–z
C _(G) –H _(G) ···Cg _(H)			2.74 Å, 134°	–1+x, y, z
H-bonding (intramolecular)	Non-classical	Non-classical	Non-classical	
C _(H) –H _(H) ···N _(H)	2.62(2) Å, 103°	2.910(2) Å, 102°	2.756(2) Å, 103°	x, y, z
C _(H) –H _(H) ···N _(H)	3.443(2) Å, 158°		3.456(2) Å, 155°	x, y, z
C _(H) –H _(H) ···N _(H)	2.07(2) Å, 102°	2.763(2) Å, 103°	2.914(2) Å, 102°	x, y, z
C _(H) –H _(H) ···N _(H)		3.456(2) Å, 153°		x, y, z
C _(H) –H _(H) ···N _(H)	3.455(2) Å, 151°	3.422(2) Å, 151°	3.437(2) Å, 148°	x, y, z
C _(H) –H _(H) ···N _(H)	2.768(2) Å, 103°	2.772(2) Å, 103°	2.766(2) Å, 103°	x, y, z
C _(H) –H _(H) ···N _(H)	2.898(2) Å, 102°	2.921(2) Å, 101°	2.902(2) Å, 101°	x, y, z
C _(H) –H _(H) ···O _(G)			2.55 Å, 132°	1–x, –y, –z
C _(H) –H _(H) ···O _(G) –C _(G)	2.70 Å, 127°(<)			1–x, 1–y, 1–z
N _(H) –H _(H) ···C _(H) –C _(H)	2.83 Å, 158°(<)			–x, 1–y, 1–z
Short contacts (host/guest and guest/guest) ^b				
C ₆₂ –H ₆₂ ···C ₇₅ –C ₇₄	2.87 Å, 141°(<)			x, y, 1+z
C ₇₂ –H ₇₂ ···C ₅₂ –C ₅₃	2.86 Å, 143°(<)			–x, 1–y, 1–z
C _(H) –H _(H) ···S _(H) –C _(H)		2.95 Å, 146° (<)		1–x, 2–y, 1–z
C _(H) –H _(H) ···H _(H) –C _(H)		2.35 Å, 128° (<)		1–x, 1–y, 1–z
C _(H) –H _(H) ···H _(G) –C _(G)		2.36 Å, 143° (<)		1–x, 2–y, 1–z
C _(H) –H _(H) ···H _(G) –C _(G)		2.94 Å, 130° (<)		x, 1+y, z
C _(H) –H _(H) ···H _(G) –C _(G)		2.37 Å, 158° (<)		2–x, 2–y, 1–z
C _(H) –H _(H) ···H _(H) –C _(H)			2.38 Å, 127° (<)	2–x, 1–y, –z
C _(H) –H _(H) ···H _(G) –C _(G)			2.27 Å, 158° (<)	2–x, –y, –z
C _(G) –H _(G) ···H _(H) –C _(H)			2.26 Å, 141° (<)	1–x, 1–y, 1–z
C _(G) –H _(G) ···H _(H) –C _(H)			2.25 Å, 141° (<)	1–x, –y, 1–z

^aNumber pf H···G interactions are indicated in parentheses.^bDistances denoted by < are contacts that measure less than the sum of the van der Waals radii of the atoms involved while those denoted by << is this sum minus 0.2 Å.

Table S5. (b) All interactions for H₂ with anisole and methylanisole isomers

Non-covalent interaction	H ₂ •ANI	H ₂ •4MANI	Symmetry
π···π (H···H and H···G) H···G interactions	4.080(1) – 5.920(1) Å [5] 4.725(1)–5.920(1) Å	4.0427(7) – 5.756(3) Å [H···H only]	
CH···π (host–host)			
C _(H) –H _(H) ···Cg _(H)	2.94 Å, 76°		x, y, z
C _(H) –H _(H) ···Cg _(H)	2.57 Å, 97°		x, y, z
C _(H) –H _(H) ···Cg _(H)	2.92 Å, 128°		x, 1+y, z
C _(H) –H _(H) ···Cg _(H)	2.63 Å, 162°		2-x, 2-y, 2-z
C _(H) –H _(H) ···Cg _(H)	2.81 Å, 105°		x, y, z
C _(H) –H _(H) ···Cg _(H)	2.92 Å, 128°		1-x, 1-y, 1-z
C _(G) –H _(G) ···Cg _(H)	2.90 Å, 137°		x, y, z
C _(H) –H _(H) ···Cg _(H)	2.61 Å, 99°		x, y, z
C _(H) –H _(H) ···Cg _(H)	2.83 Å, 132°		1+x, y, z
C _(H) –H _(H) ···Cg _(H)	2.90 Å, 133°		1-x, -y, 1-z
C _(H) –H _(H) ···Cg _(H)	2.75 Å, 105°		x, y, z
C _(H) –H _(H) ···Cg _(H)	2.69 Å, 155°		1-x, -y, -z
C _(G) –H _(G) ···Cg _(H)	2.78 Å, 135°		-1+x, -1+y, z
H-bonding (intramolecular)	Non-classical	Non-classical	
C _(H) –H _(H) ···N _(H)	2.805(3) Å, 102°		x, y, z
N _(H) –H _(H) ···C _(G) –C _(G)		2.63 Å, 156° (<<)	1-x, -y, 1-z
N _(H) –H _(H) ···C _(G) –C _(G)		2.37 Å, 160° (<)	1+x, y, z
C _(H) –H _(H) ···O _(H) –C _(H)	2.64 Å, 164° (<)		1+x, 1+y, z
C _(H) –H _(H) ···O _(H) –C _(H)		2.63 Å, 155° (<)	-1+x, -1+y, z
Short contacts (host–guest and guest···guest) ^{a,b}			
C _(H) –H _(H) ···H _(G) –C _(G)		2.30 Å, 127° (<)	1+x, 1+y, z

^aDistances denoted by < are contacts that measure less than the sum of the van der Waals radii of the atoms involved while those denoted by << is this sum minus 0.2 Å.

Table S5. (c) All interactions for H₂ with ethylbenzene and xylene isomers

Non-covalent interaction	H ₂ • <i>p</i> -Xy	Symmetry
π–π (H···H and H···G)	4.074(1) – 5.926(1) Å 5H–G 4.781(1)–5.926(1) Å	
CH···π (host–host)		
C _(H) –H _(H) ···Cg _(H)	3.00 Å, 127°	x, y, z
C _(G) –H _(G) ···Cg _(H)	2.76 Å, 145°	x, 1+y, z
H–bonding		
N _(H) –H _(H) ···C _(G) –C _(G)	2.77 Å, 159° (<)	1–x, 2–y, 1–z
Short contacts (host···guest and guest···guest) ^{a,b}		
C _(H) –H _(H) ···H _(G) –C _(G)	2.33 Å, 130° (<)	x, -1+y, z
C _(G) –H _(G) ···C _(H) –C _(H)	2.87 Å, 139° (<)	x, y, z

^aDistances denoted by < are contacts that measure less than the sum of the van der Waals radii of the atoms involved while those denoted by << is this sum minus 0.2 Å.

Summary of H···H interactions of inclusion compounds

Interaction	H ₂ • <i>p</i> -Xy	
	#	Range
π···π	–	4.047(1)–5.926(1) Å
CH···π	3	2.51–2.80 Å, 102–163°
Non-classical	2	2.811(2)–3.486(2) Å, 102–165°
H–bonding		
Other short contacts	1	2.56 Å, 165°

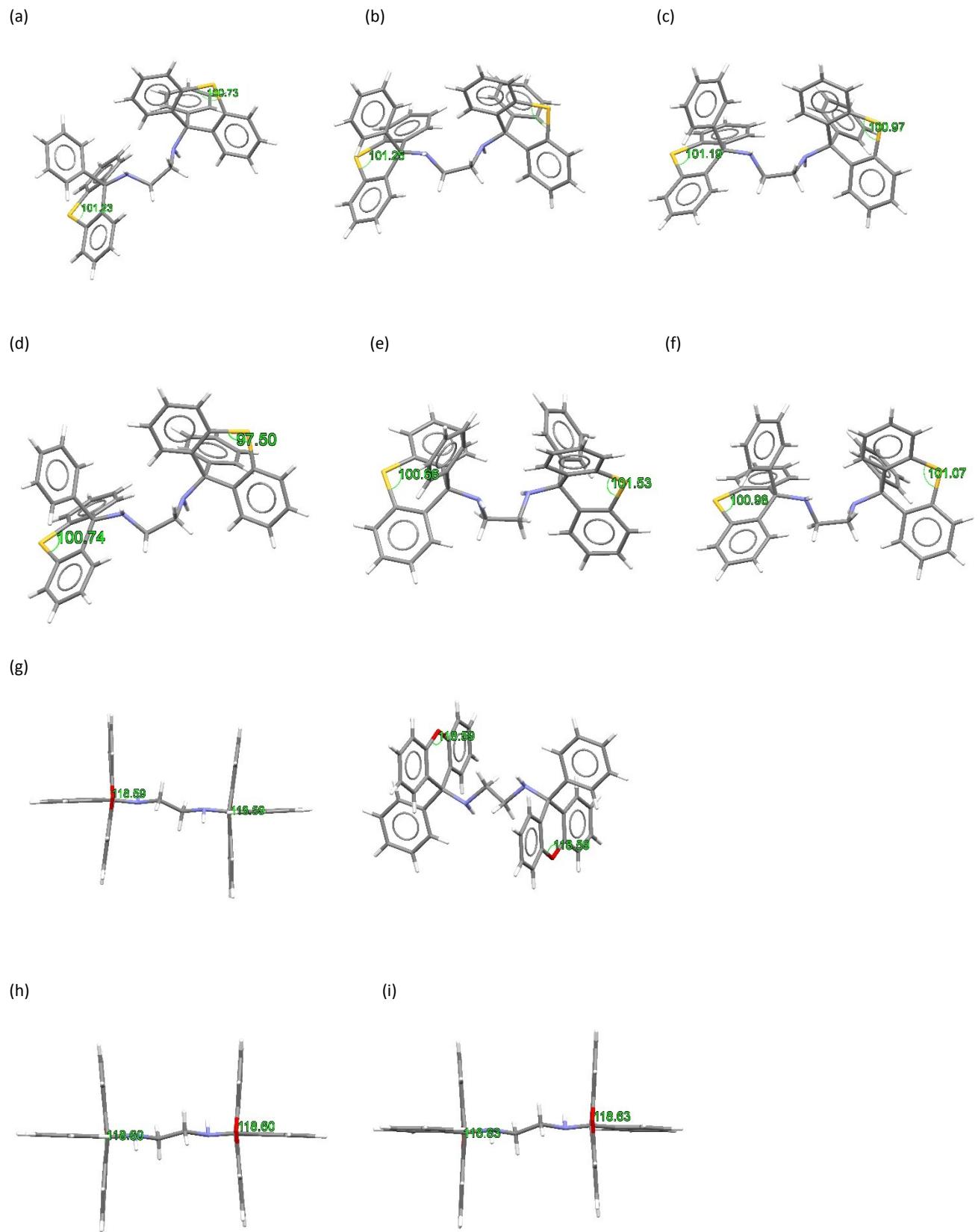


Figure S6. Host geometry within the respective complexes (a) $\mathbf{H}_1 \bullet p\text{-Xy}$, (b) $\mathbf{H}_1 \bullet o\text{-Xy}$, (c) $\mathbf{H}_1 \bullet \text{EB}$, (d) $\mathbf{H}_1 \bullet \text{ANI}$, (e) $\mathbf{H}_1 \bullet 3\text{MANI}$, (f) $\mathbf{H}_1 \bullet 4\text{MANI}$, (g) $\mathbf{H}_2 \bullet p\text{-Xy}$, (h) $\mathbf{H}_2 \bullet \text{ANI}$ and (i) $\mathbf{H}_2 \bullet 4\text{MANI}$.