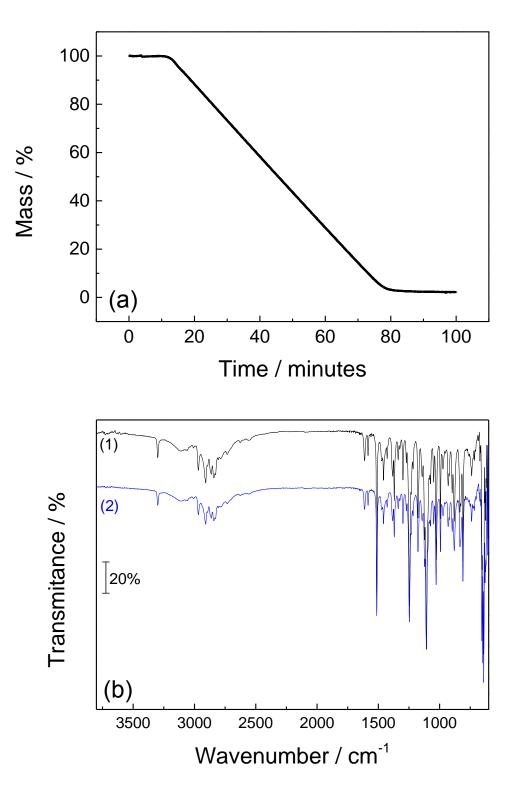
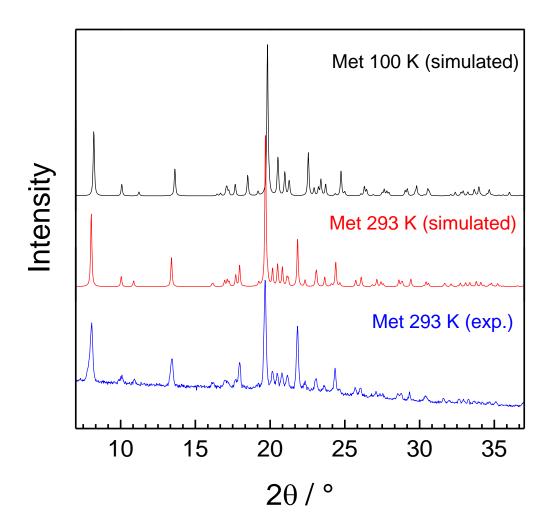
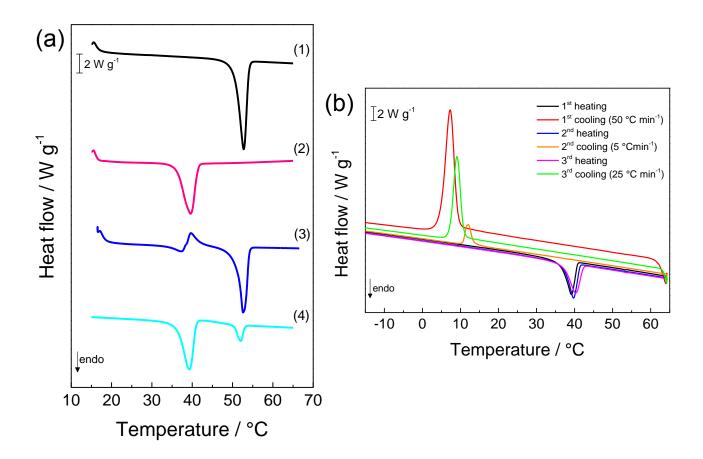
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**Figure S1.** (a) TG curve of metoprolol, sample mass of 0.99 mg, in open  $\alpha$ -alumina crucible, dry air (50 mL min<sup>-1</sup>), heated up to 150 °C at 10 °C min<sup>-1</sup> and kept in isotherm during 100 min; (b) (1) FTIR spectra of metoprolol base and (2) sublimated material collected in a glass tube.



**Figure S2.** X-ray powder diffractogram of metoprolol, form I, simulated at 100 K [CCDC 1882466, Rossi, et. al., 2019] in comparison with form I experimental and simulated powder diffractograms (CCDC 1883843) at room temperature.



**Figure S3.** Examples of metoprolol DSC curves: (a) Heating runs: (1) first heating; (2) after cooling at 5 °C min<sup>-1</sup>; (3) after cooling for 2h30 on the fridge; (4) after another cooling on DSC at 5 °C min<sup>-1</sup>; and (b) other sample with subsequent cycles: at heating rate of 10 °C min<sup>-1</sup> and different cooling hates: 50 °C min<sup>-1</sup> on the 1<sup>st</sup> cooling, 5 °C min<sup>-1</sup> on the 2<sup>nd</sup> cooling and 25 °C min<sup>-1</sup> on the 3<sup>rd</sup> cooling.

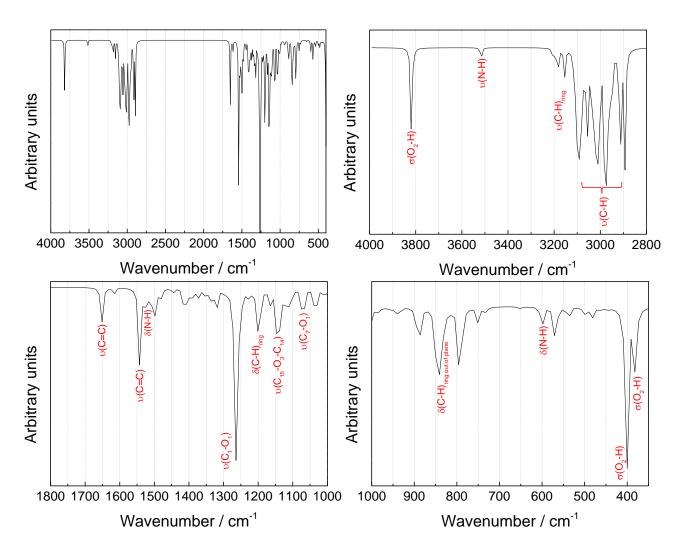
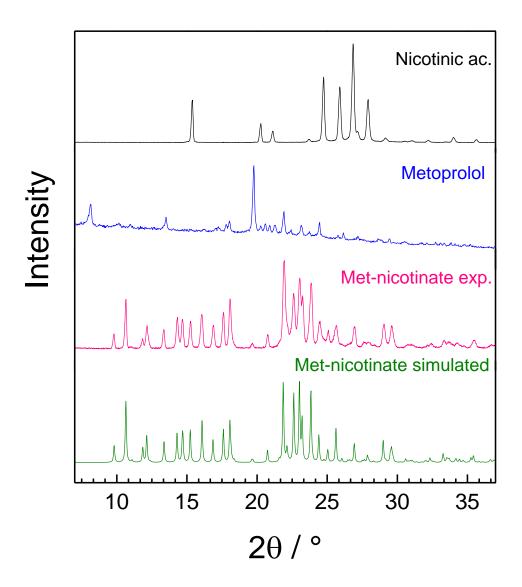
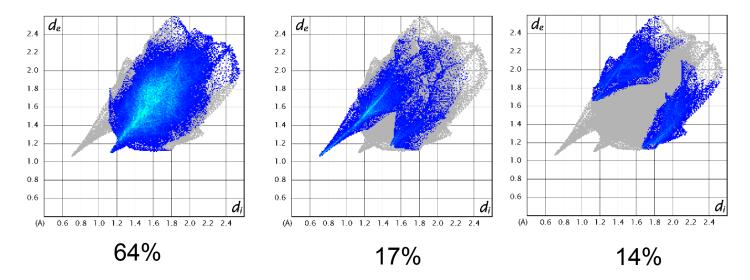


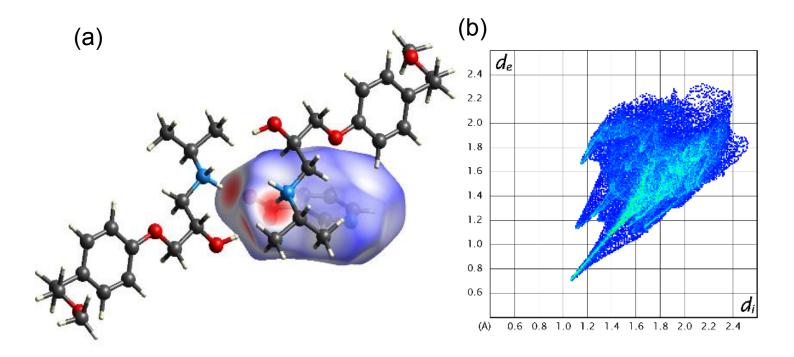
Figure S4. Simulated FTIR spectrum for metoprolol isolated molecule.



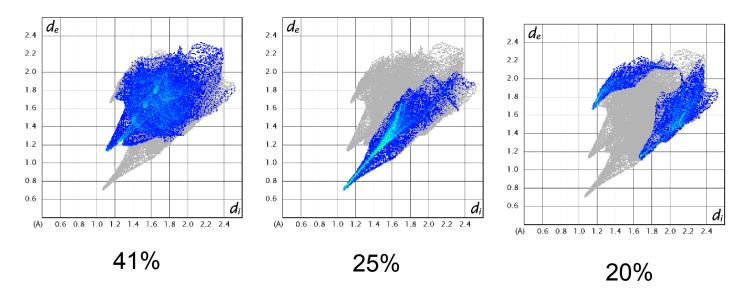
**Figure S5.** X-ray powder diffractograms of nicotinic acid, metoprolol base (Form I), and metoprolol nicotinate experimental and simulated.



**Figure S6.** Fingerprint plots for the metoprolol cation in metoprolol nicotinate broken down into contributions from  $H\cdots H$  (left),  $H\cdots O$  (middle) and  $H\cdots C$  (right) close contacts.



**Figure S7.** (a)  $d_{\text{norm}}$  surfaces of the nicotinate anion in the metoprolol nicotinate crystal lattices. Neighboring counterions associated with close contacts are also shown; (b) Fingerprint for the nicotinate anion in metoprolol nicotinate.



**Figure S8.** Fingerprint plots for the nicotinate anion in metoprolol nicotinate broken down into contributions from  $H\cdots H$  (left),  $H\cdots O$  (middle) and  $H\cdots C$  (right) close contacts.

Table S1. Molecular structure of the coformers tested in this work

| Coformador               | pka | Brand           | Result |
|--------------------------|-----|-----------------|--------|
| NH<br>NH<br>O<br>thymine | 9.7 | Sigma<br>>99.5% | x      |
| NH                       | 9.5 | Sigma<br>>99%   | X      |
| NH<br>NH<br>uracil       | 9.4 | Fluka<br>≥99%   | X      |

| N<br>4,4-bipiridil<br>NH <sub>2</sub>        | 4.3        | Fluka<br>>99% | X    |
|--|------------|---------------|------|
| HO NH <sub>2</sub> L-glutamine               | 9.1        |               | x    |
| OH<br>Nicotinic acid                         | 2.0<br>4.9 | Sigma<br>≥99% | Salt |
| benzamide                                    | 23.3       | Sigma<br>>98% | X    |
| NH <sub>2</sub> NH <sub>2</sub> nicotinamide | 3.4        | Sigma<br>>99% | X    |

| NH <sub>2</sub> 3.7 Sigma >99% x |
|----------------------------------|
|----------------------------------|