Electronic Supplementary Information (ESI)

Improved Ultraviolet Radiation Stability of Mn²⁺ Doped CsPbCl₃ Nanocrystals via B-

site Sn Doping

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Fig. S1 XRD patterns of CsPbCl₃ NCs synthesized with various Sn /Pb ratios. An enlarged view of the peaks marked by the outline is shown in the right panel, in which the vertical lines are drawn to guide the eye. It is found that the diffraction peak near 32 degree shifts to a large angle with increasing Sn content, indicating that increased Sn^{2+} ions are doped in the NCs.

PL decay fitting details

The lifetime data were fitted by a biexponential function:

$$I(t) = A_1 \exp(-t/\tau_1) + A_2 \exp(-t/\tau_2)_{.....(1)}$$

where τ_1 and τ_2 are the time constants, respectively, A_1 and A_2 are the normalized amplitudes of the components, respectively. The average lifetime was given by:

$$\tau_{ave} = (A_1 \tau_1^2 + A_2 \tau_2^2) / (A_1 \tau_1 + A_2 \tau_2)$$
(2)

Table S1. Mn²⁺ PL decay parameters of Mn:CsPbCl₃ NCs synthesized with various Sn/Mn/Pb ratios.

Sn/Mn/Pb ratios		- (
	A_1	$\tau_1(ms)$	A_2	$\tau_2 (ms)$	τ_{avg} (ms)
0/1/1	0.04	0.14	0.96	1.75	1.74
0.5/1/1	0.15	0.16	0.93	1.78	1.76
1/1/1	0.02	0.03	0.98	1.8	1.80
2/1/1	0.05	0.19	0.95	1.78	1.77
3/1/1	0.06	0.23	0.94	1.75	1.73



Fig. S2 XRD patterns of Mn:CsPbCl₃ NCs with Sn/Mn/Pb molar ratios of 0/1/1, 1/1/1, and 2/1/1 under ultraviolet illumination at various times. No significant change was observed in all samples, indicating that the optical degradation of these NCs did not originate directly from its structural degradation.



Fig. S3 PL intensities of band-edge excitons (a) and Mn^{2+} ions (b) and PL lifetimes of Mn^2 ions in the Mn:CsPbCl₃ NCs as a function of illumination time. The PL intensities are normalized for their initial samples without ultraviolet illumination.

Sn/Mn/Pb=0/1/1	Fitting parameters				τ_{avg} (ms)
	\mathbf{A}_{1}	τ ₁ (ms)	A ₂	τ ₂ (ms)	"·g()
Oh	0.49	1.76	0.49	1.76	1.76
1h	0.94	1.76	0.08	1.08	1.73
5h	0.08	0.39	0.91	1.69	1.66
10h	0.82	1.61	0.17	0.41	1.55
15h	0.20	0.39	0.79	1.55	1.48
20h	0.25	0.36	0.75	1.40	1.32
25h	0.32	0.33	0.68	1.12	1.02

Table S2. Mn^{2+} PL decay parameters of Mn:CsPbCl₃ NCs with Sn/Mn/Pb ratio of 0/1/1 under ultraviolet illumination at various times.

Table S3. Mn^{2+} PL decay parameters of Mn:CsPbCl₃ NCs with Sn/Mn/Pb ratio of 1/1/1 under ultraviolet illumination at various times.

Sn/Mn/Pb=1/1/1	Fitting parameters				τ_{avg} (ms)
	A ₁	τ ₁ (ms)	A_2	τ_2 (ms)	
Oh	0.50	1.80	0.50	1.80	1.80
1h	0.50	1.81	0.50	1.81	1.81
5h	0.50	1.83	0.50	1.83	1.83
10h	0.50	1.83	0.50	1.83	1.83
15h	0.50	1.81	0.50	1.81	1.81
20h	0.50	1.76	0.50	1.76	1.76
25h	0.09	0.64	0.91	1.76	1.73

Table S4. Mn^{2+} PL decay parameters of Mn:CsPbCl₃ NCs with Sn/Mn/Pb ratio of 2/1/1 under ultraviolet illumination at various times.

Sn/Mn/Pb=2/1/1	Fitting parameters				τ_{avg} (ms)
	A ₁	τ ₁ (ms)	A ₂	τ ₂ (ms)	
0h	0.95	1.78	0.05	0.19	1.77
1h	0.50	1.77	0.50	1.77	1.77
5h	0.01	0.23	0.99	1.78	1.78
10h	0.01	0.58	0.99	1.78	1.78
15h	0.03	0.08	0.97	1.75	1.75
20h	0.14	0.49	0.86	1.67	1.62
25h	0.24	0.50	0.76	1.50	1.41



Fig. S4 PL spectra and Mn^{2+} emission decay curves of Mn:CsPbCl₃ NCs with Sn/Mn/Pb molar ratios of 0/2/1 (a, d), 1/2/1 (b, e), and 2/2/1 (c, f) under ultraviolet illumination at various times.