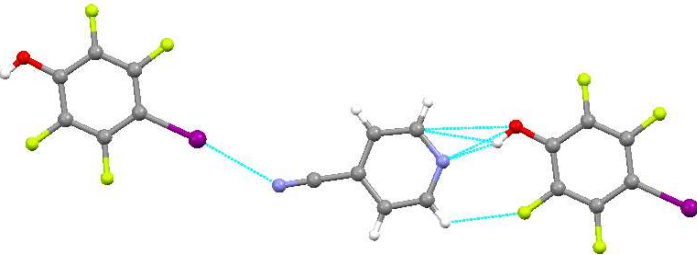
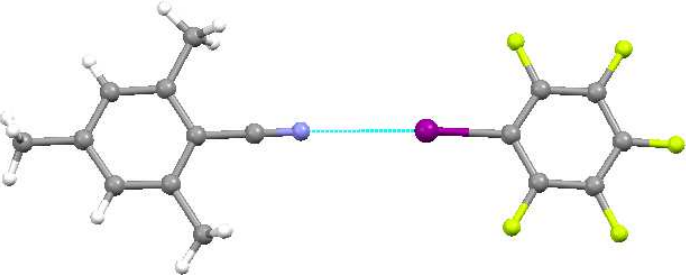
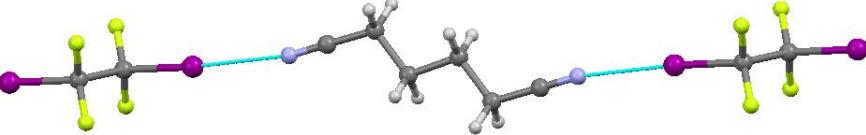


Electronic  
Supplementary  
Information

## The Energy Frameworks of Aufbau Synthron Modules in 4-Cyanopyridine Co-crystals.

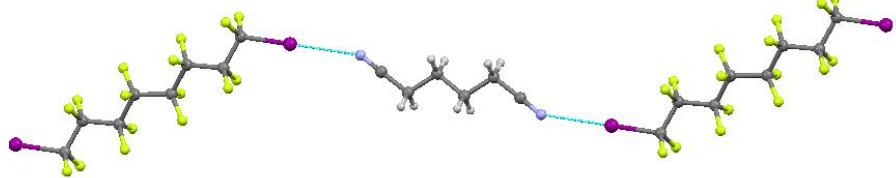
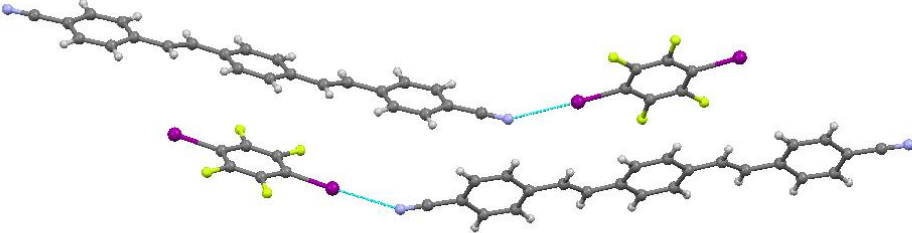
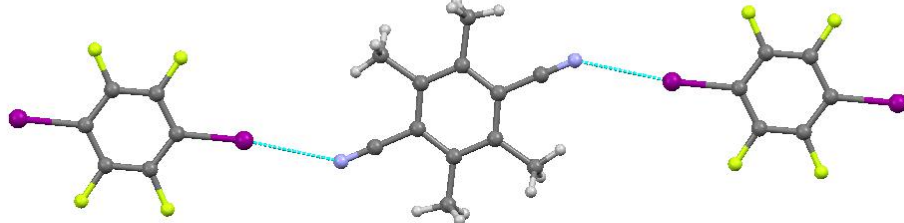
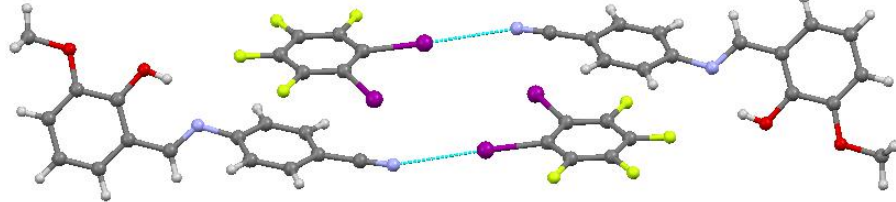
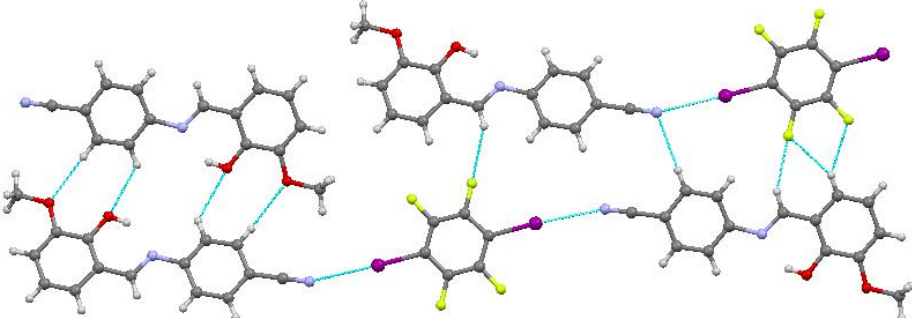
Y. V. Torubaev, I. V. Skabitsky.

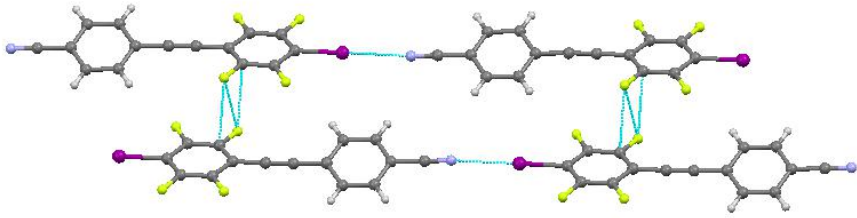
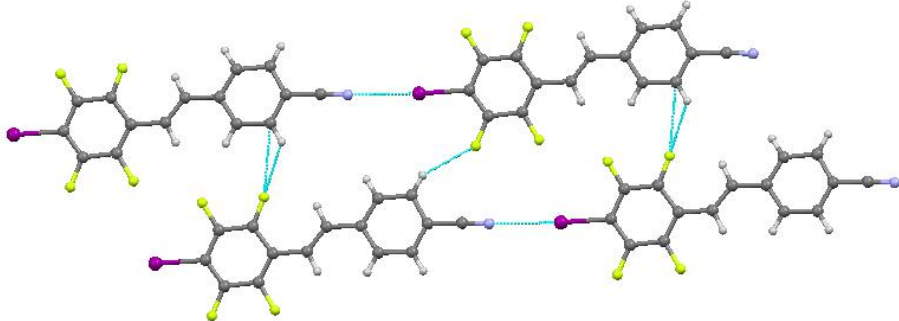
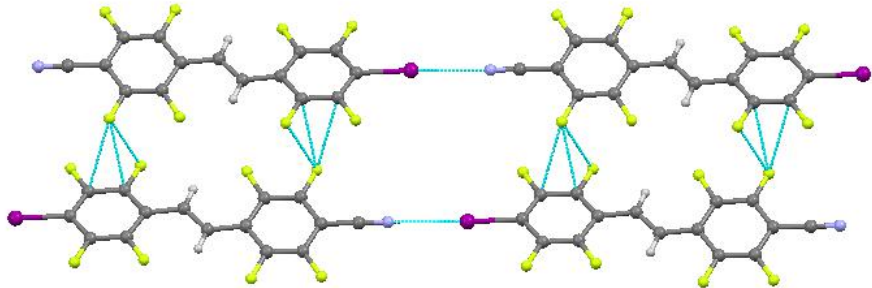
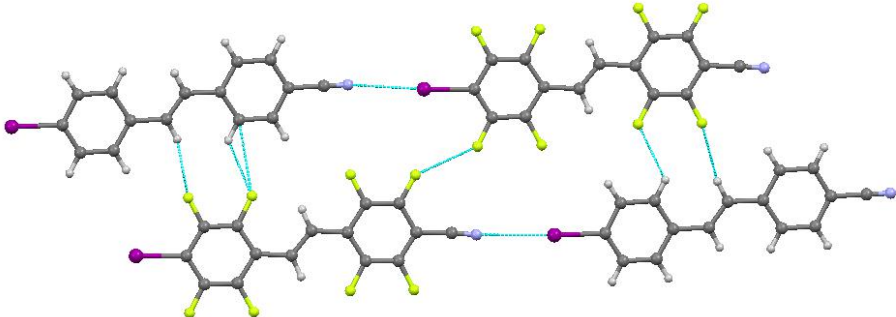
**Table S1.** Selected examples of intermolecular CN---I-R<sub>f</sub><sup>1</sup> bonding in the co-crystals (from the Cambridge Structural Database <sup>2</sup>)

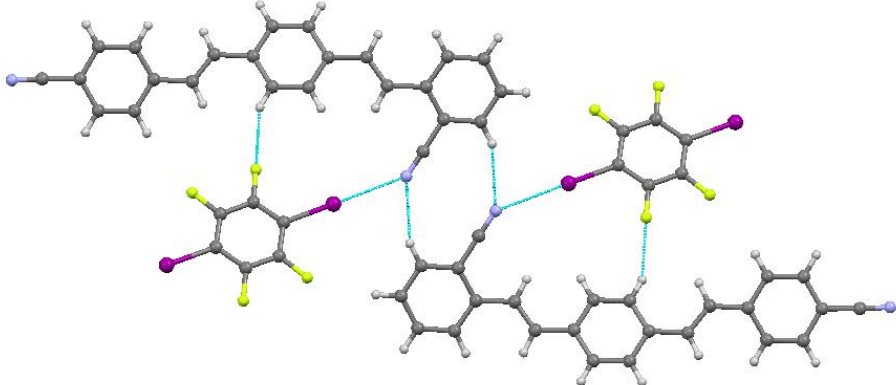
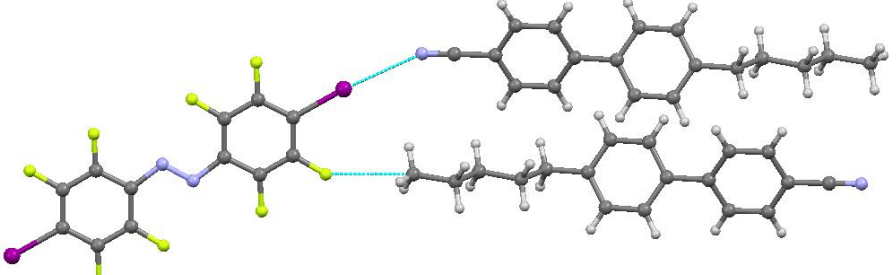
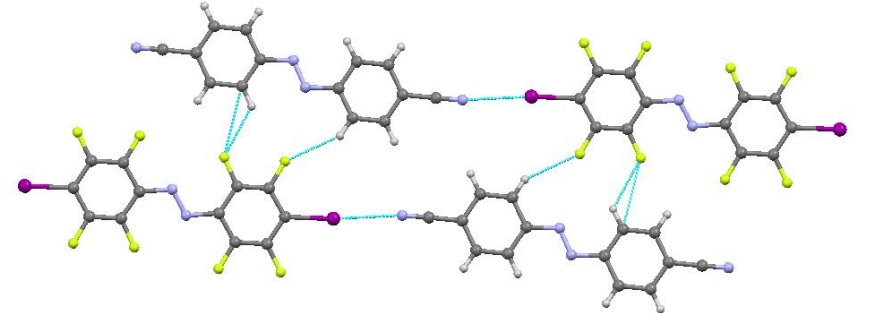
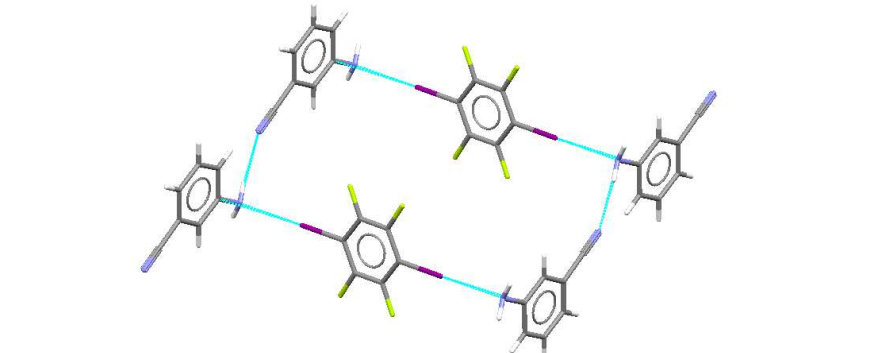
	CSD Identifier	Structure	Rederence
1	BUNJEB		C.B.Aakeroy, C.L.Spartz, S.Dembowski, S.Dwyre, J.Desper, <i>IUCrJ</i> (2015), <b>2</b> , 498
2	EBIHEF		A.Wasilewska, M.Gdaniec, T.Polonski, <i>CSD Communication</i> (2016)
3	EBIIHH		P.Metrangolo, T.Pilati, G.Resnati, A.Stevenazzi, <i>Chem.Commun.</i> (2004), 1492

<sup>1</sup> R<sub>f</sub> – perfluorinated aryl or alkyl group

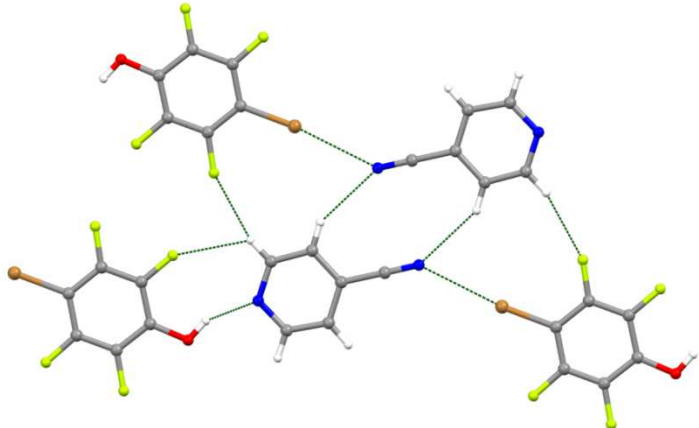
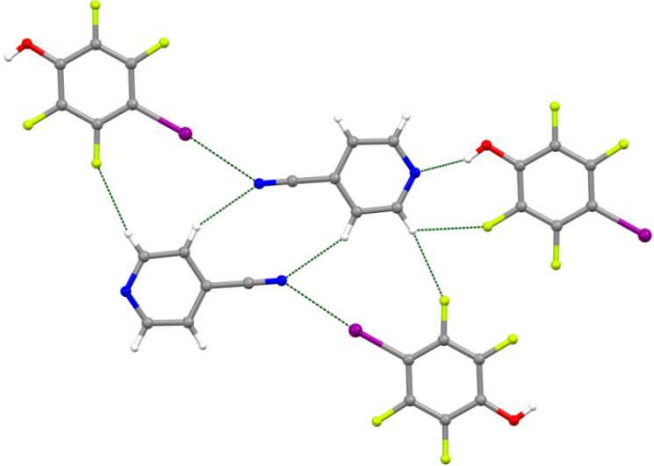
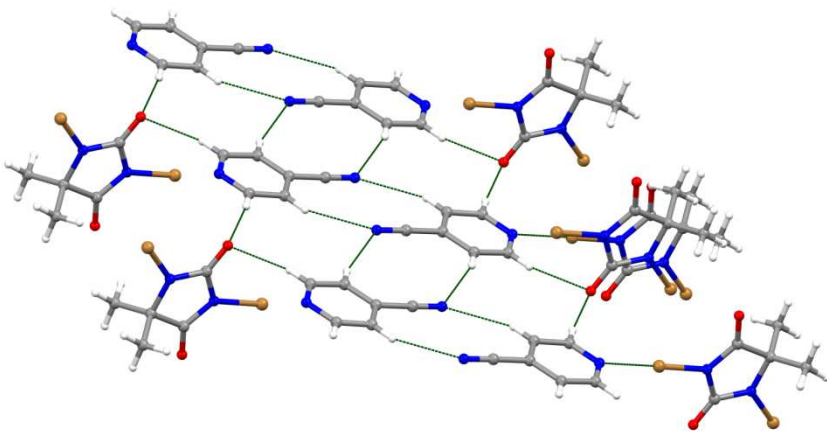
<sup>2</sup>The Cambridge Structural Database C. R. Groom, I. J. Bruno, M. P. Lightfoot and S. C. Ward, *ActaCryst.* (2016). B72, 171-179 ; DOI: 10.1107/S2052520616003954

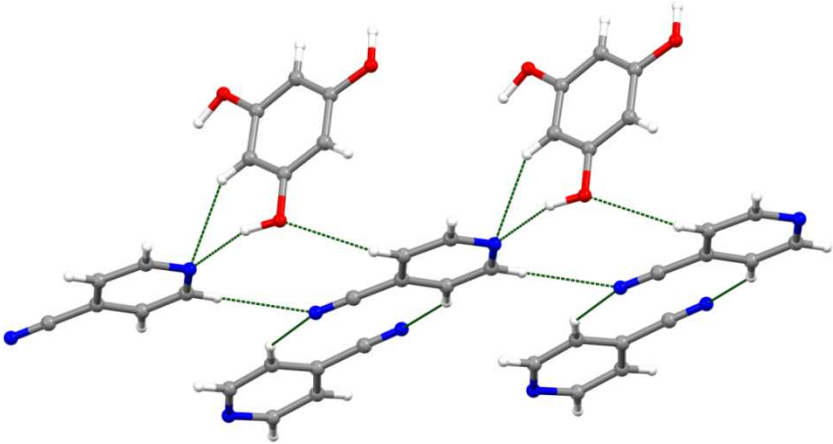
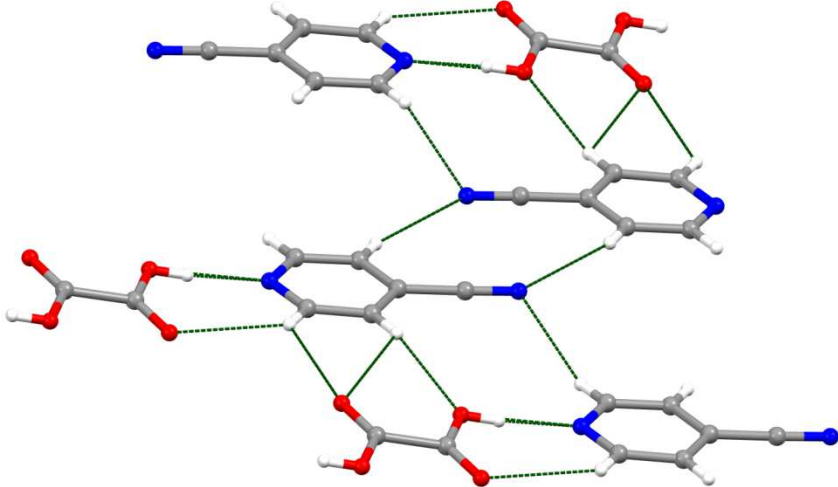
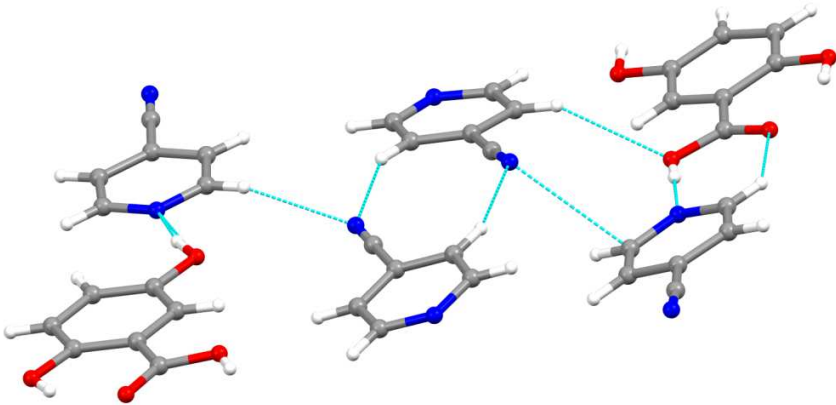
4	EBIHON		<p>P.Metrangolo, T.Pilati, G.Resnati, A.Stevenazzi, <i>Chem. Commun.</i> (2004), 1492</p>
5	FAKFII		<p>Dongpeng Yan, A.Delori, G.O.Lloyd, T.Friscic, G.M.Day, W.Jones, Jun Lu, Min Wei, D.G.Evans, XueDuan, <i>Angew. Chem., Int. E</i> <i>d.</i> (2011), <b>50</b>, 12483</p>
6	HUMLOQ		<p>D.Britton, W.B.Gleason, <i>Acta Crystallogr., Se</i> <i>ct. E: Struct. Rep. Onli</i> <i>ne</i> (2002), <b>58</b>, o1375</p>
7	IWONAL		<p>M.Zbacnik, M.Vitkovic, V.Vulic, I.Nogalo, D.Cincic, <i>Cryst. Growth Des.</i> (2016), <b>16</b>, 6381</p>
8	IWONoz		<p>M.Zbacnik, M.Vitkovic, V.Vulic, I.Nogalo, D.Cincic, <i>Cryst. Growth Des.</i> (2016), <b>16</b>, 6381</p>

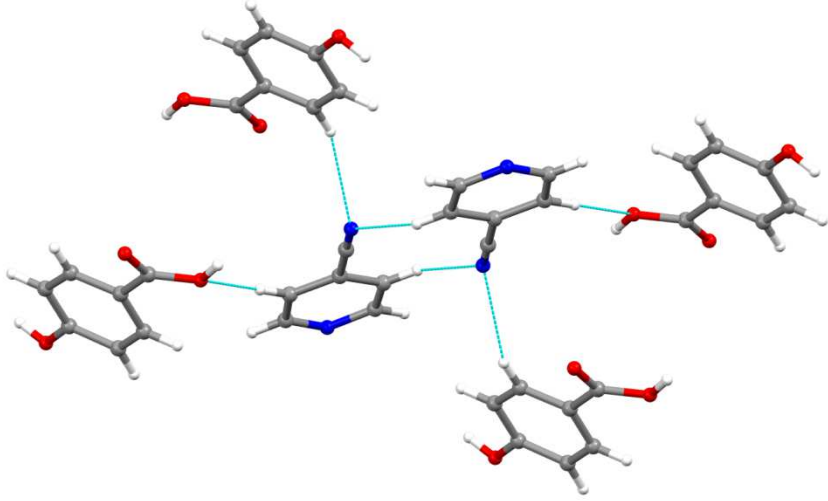
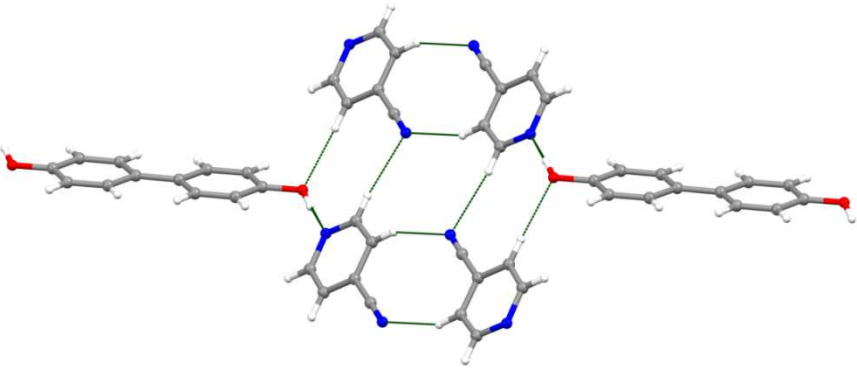
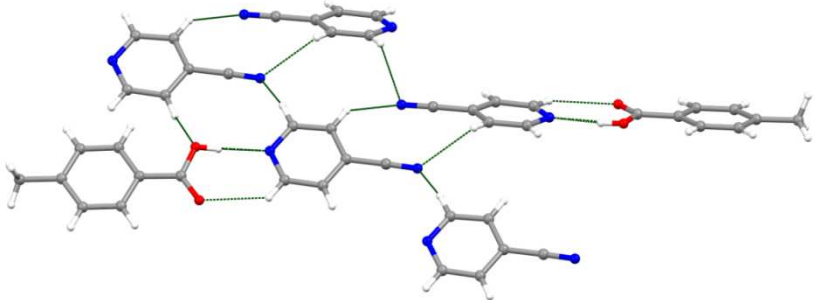
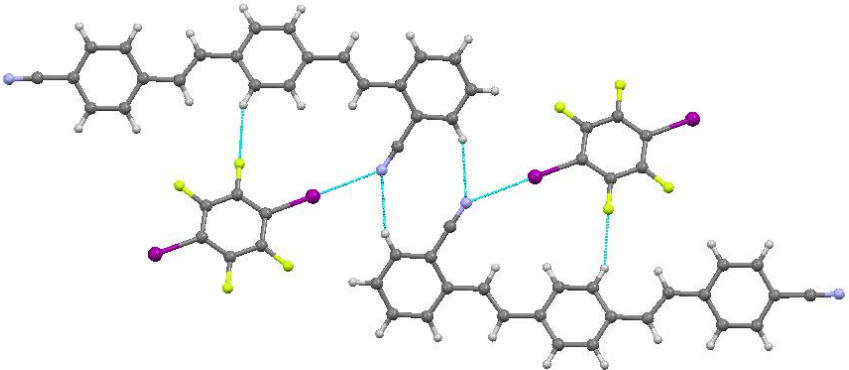
9	NUHDEB		<p>F.Frausto, Z.C.Smith, T.E.Haas, S.W.Thomas III, <i>Chem. Commun.</i> (2015), <b>51</b>, 8825</p>
10	QORZUT		<p>R.Mariaca, G.Labat, N.-R.Behrnd, M.Bonin, F.Helbling, P.Eggli, G.Couderc, A.Neels, H.Stoeckli- Evans, J.Hulliger, <i>J. Fluorine Chem.</i> (2009), <b>130</b>, 175</p>
11	QOSCAD		<p>R.Mariaca, G.Labat, N.-R.Behrnd, M.Bonin, F.Helbling, P.Eggli, G.Couderc, A.Neels, H.Stoeckli- Evans, J.Hulliger, <i>J. Fluorine Chem.</i> (2009), <b>130</b>, 175</p>
12	QOSCOR		<p>R.Mariaca, G.Labat, N.-R.Behrnd, M.Bonin, F.Helbling, P.Eggli, G.Couderc, A.Neels, H.Stoeckli- Evans, J.Hulliger, <i>J. Fluorine Chem.</i> (2009), <b>130</b>, 175</p>

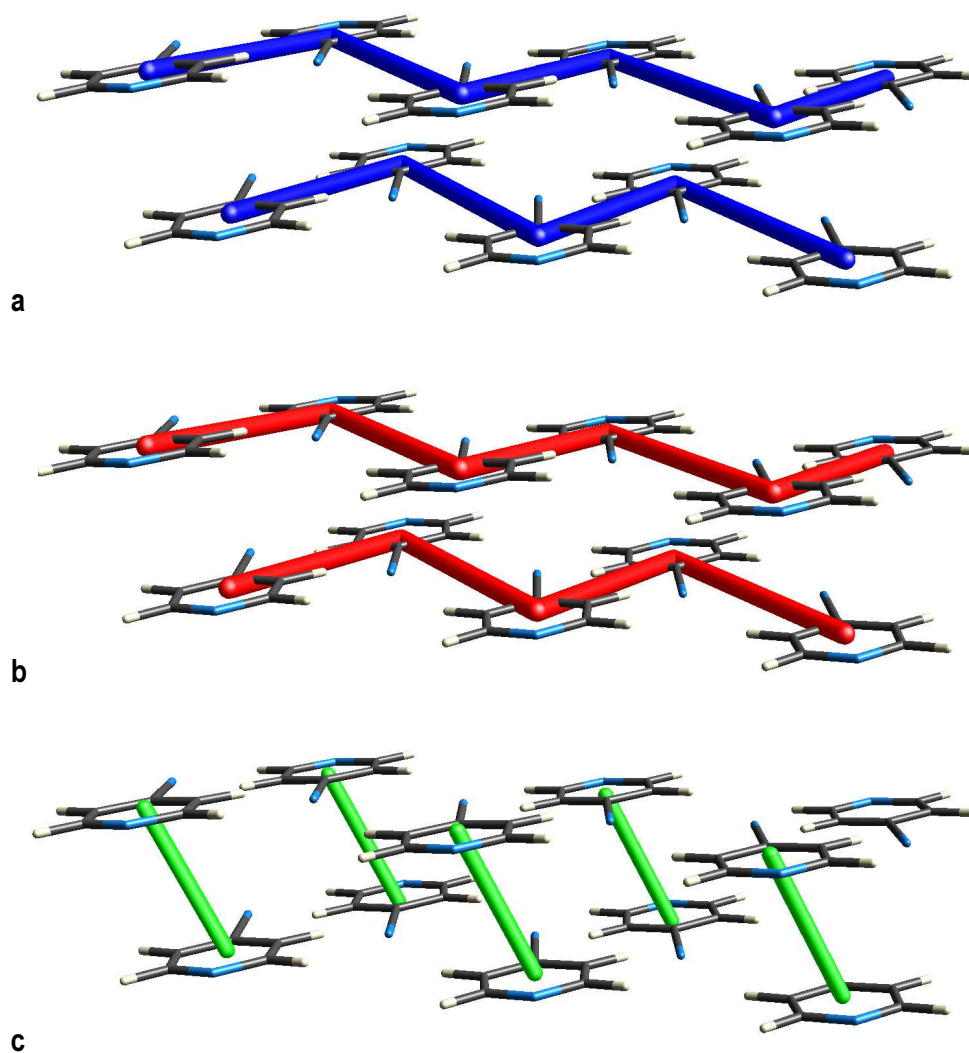
13	SAHJAP		<p>Guoling Fan, Dongpeng Yan, <i>Scientific Reports</i> (2014), <b>4</b>, 4933</p>
14	ZABLOG		<p>O.S.Bushuyev, T.Friscic, C.J.Barrett, <i>Cryst. Growth Des.</i> (2016), <b>16</b>, 541</p>
15	ZABLUM		<p>O.S.Bushuyev, T.Friscic, C.J.Barrett, <i>Cryst. Growth Des.</i> (2016), <b>16</b>, 541</p>
16	JEHZUU		<p>V.Nemec, D.Cincic, <i>CrystEngComm</i> (2016), <b>18</b>, 7425, doi:10.1039/C6CE0 1703G</p>

**Table S2.** Selected examples of intermolecular CN---H- hydrogen bonded 4-cyanopyridine associates in binary co-crystals (from the Cambridge Structural Database [2])

	CSD Identifier		
1	BULVAH		C.B.Aakeroy, C.L.Spartz, S.Dembowski, S.Dwyre, J.Desper, <i>IUCrJ</i> (2015), <b>2</b> , 498
2	BUNJEB		C.B.Aakeroy, C.L.Spartz, S.Dembowski, S.Dwyre, J.Desper, <i>IUCrJ</i> (2015), <b>2</b> , 498
3	EZIBIA		I.Nicolas, O.Jeannin, D.Pichon, M.Fourmigue, <i>CrystEngComm</i> (2016), <b>18</b> , 9325

4	KIHZEH		<p>J.A.Bis, P.Vishweshwar, D.Weyna, M.J.Zaworotko, <i>Mol.Pharmaceutics</i> (2007), <b>4</b>, 401</p>
5	PAVGOK		<p>Wen-Ni Zheng, <i>Acta Crystallogr., Sect. E</i> <i>Struct. Rep. Online</i> 2012), <b>68</b>, o1625</p>
6	PEKTAC		<p>V.Stilinovic, B.Kaitner, <i>Cryst. Growth Des.</i> (2012), <b>12</b>, 5763</p>

7	SOLFEG		<p>A.Mukherjee, G.R.Desiraju, <i>Cryst. Growth Des.</i> (2014), <b>14</b>, 1375</p>
8	TEHNAW		<p>J.A.Bis, P.Vishweshwar, R.A.Middleton, M.J.Zaworotko, <i>Cryst. Growth Des.</i> (2006), <b>6</b>, 1048</p>
9	USOBOU		<p>Xing-Wei Cai, Hong- Fei Lu, <i>Acta Crystallogr., Sect. E: Struct. Rep. Online</i> (2011), <b>67</b>, o1555</p>
10	SAHJAP		<p>Guoling Fan, Dongpeng Yan, <i>Scientific Reports</i> (2014), <b>4</b>, 4933</p>

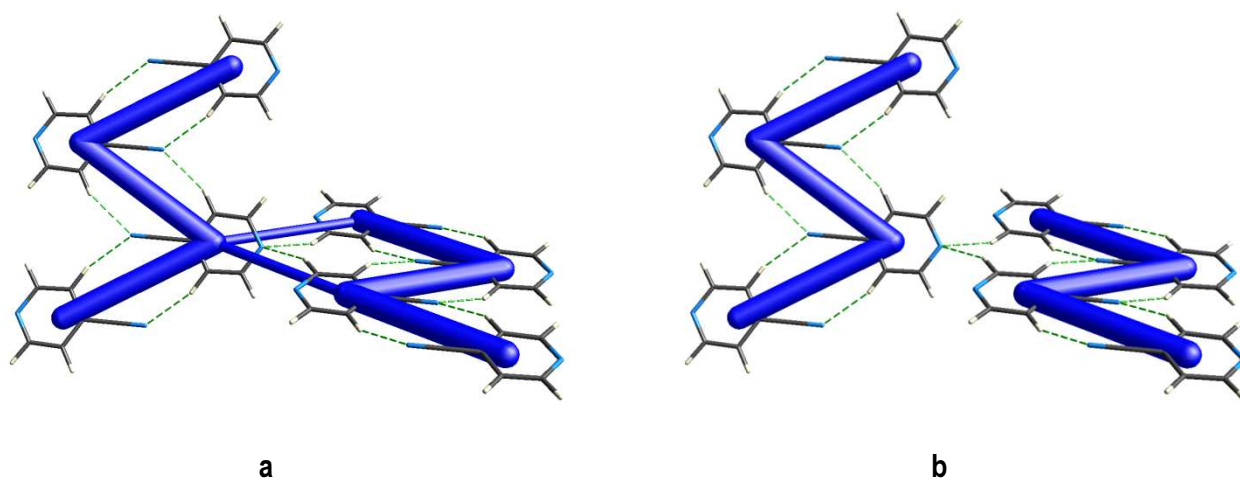


**Figure S1.** Energy frameworks in 4CNpy. a) total energy , b) electrostatic c) dispersion(15 kJ/mol cut-off).

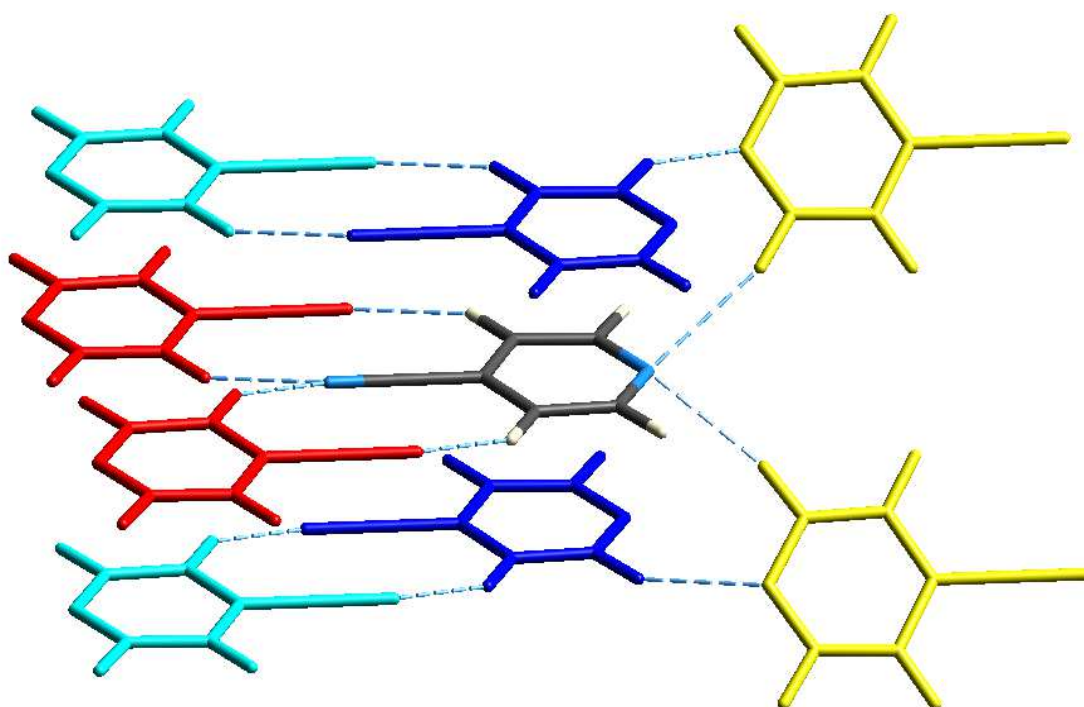
Note that total energy is the sum of the four energy components, **scaled** using the scale factors below:

Energy Model	k_ele	k_pol	k_disp	k_rep
B3LYP/6-31G(d,p) electron densities	1.057	0.740	0.871	0.618





**Figure S2.** Fragment of 4-CNpy crystal packing, showing parallel *zig-zag* energy frameworks at 10 kJ/mol (a) and 12 kJ/mol (b) cut-off.



**Figure S3 (for Table S3).** Intermolecular interactions in 4CNpy.

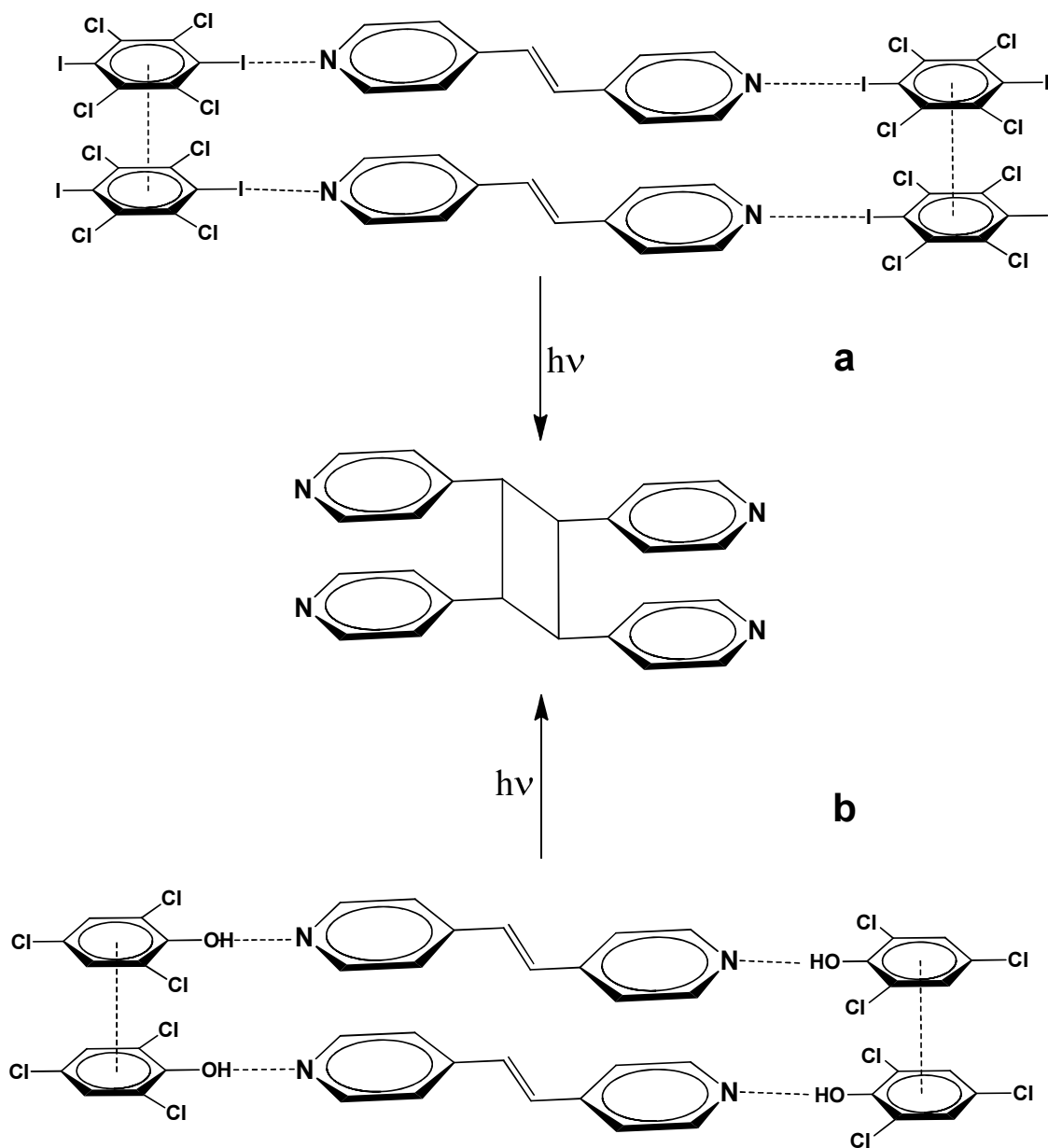
**Table S3.** Interaction Energies (kJ/mol) in 4CNpy.

	N	Symop	R	Electron Density	E_ele	E_pol	E_dis	E_rep	E_tot
--	---	-------	---	------------------	-------	-------	-------	-------	-------

2	-x, -y, -z	6.47	B3LYP/6-31G(d,p)	-24.2	-4.4	-10.5	22.1	-24.3
			B3LYP/DGDZVP	-27.3	-4.5	-10.5	25.6	-25.5
4	x+1/2, -y, -z+1/2	7.21	B3LYP/6-31G(d,p)	-7.9	-1.2	-7.1	8.1	-10.5
			B3LYP/DGDZVP	-9.9	-1.4	-7.1	10.0	-11.6
2	-x, -y, -z	6.47	B3LYP/6-31G(d,p)	-7.1	-1.5	-6.9	2.8	-12.9
			B3LYP/DGDZVP	-8.1	-1.6	-6.9	4.7	-12.8
2	x, y, z	3.82	B3LYP/6-31G(d,p)	5.0	-0.7	-22.4	10.1	-8.5
			B3LYP/DGDZVP	0.6	-0.7	-22.4	18.4	-8.0

**Table S4.** Intermolecular interaction energy (TONTO CE-B3LYP / DGDZVP)

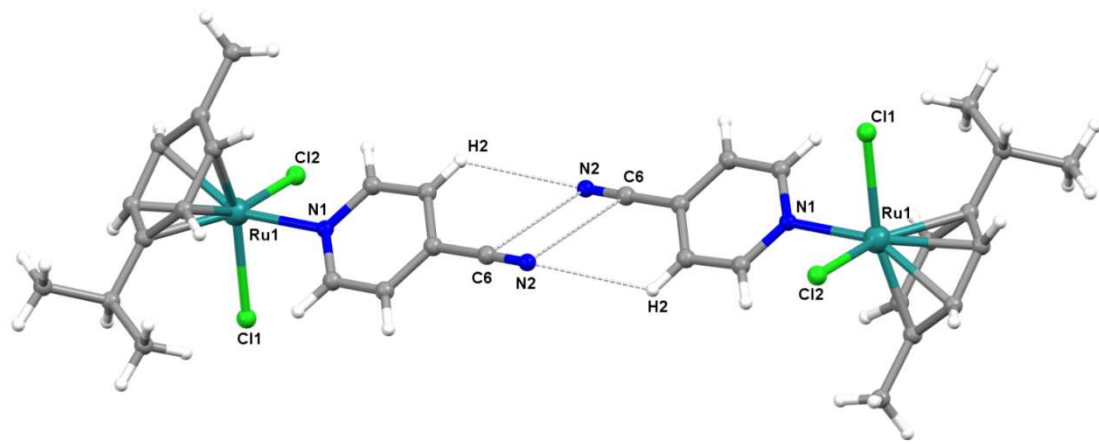
Intermol. interaction energy, -kJ/mol	4CNpy	Co-crystal 1	Co-crystal 2	Co-crystal 3	Co-crystal 4
---N <sub>py</sub> bonded	n/a	19.4	22.4	19.6	19.6
---N <sub>CN</sub> bonded	n/a	n/a	n/a	12.3	11.6
[4CNpy] <sub>2</sub> dimers	25.5	26	n/a	n/a	20
4CNpy stacking	8.0	11	10.0	10.9	9.0
Acceptor-acceptor stacking	n/a	21	10.7	32.2 – 32.7	29.1



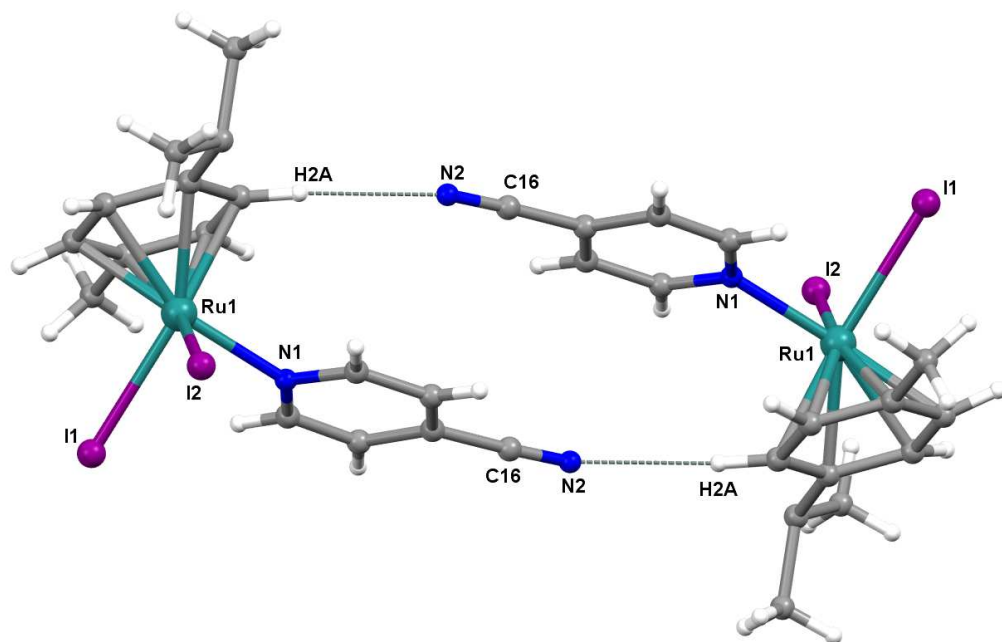
**Scheme S1.** Solid state photochemical [2+2] cycloaddition of trans-1,2-bis(4-pyridyl)ethylene templated by  $\pi$ - $\pi$  stacked columnar modules of XB **(a)** [3] and HB **(b)** [4] donor co-formers.

3. E. Bosch, S. J. Kruse, H. R. Krueger, and R. H. Groeneman, *Crystal Growth & Design*, vol. 19, no. 6, pp. 3092–3096, Apr. 2019. DOI: 10.1021/acs.cgd.9b00275.

4. G. Campillo-Alvarado, C. Li, D. C. Swenson, and L. R. MacGillivray, *Crystal Growth & Design*, vol. 19, no. 5, pp. 2511–2518, Mar. 2019. DOI: 10.1021/acs.cgd.9b00035



a



b

**Figure S4.** Fragment or crystal packing of **CymRuX<sub>2</sub>(4CNpy)** (X=Cl (CIDFIE [5]), I (5, disorder of *i*-Pr group is omitted for clarity)),

Selected distances (Å):

**CymRuCl<sub>2</sub>(4CNpy)** (CIDFIE)

N(2)-H(2) 2.70,

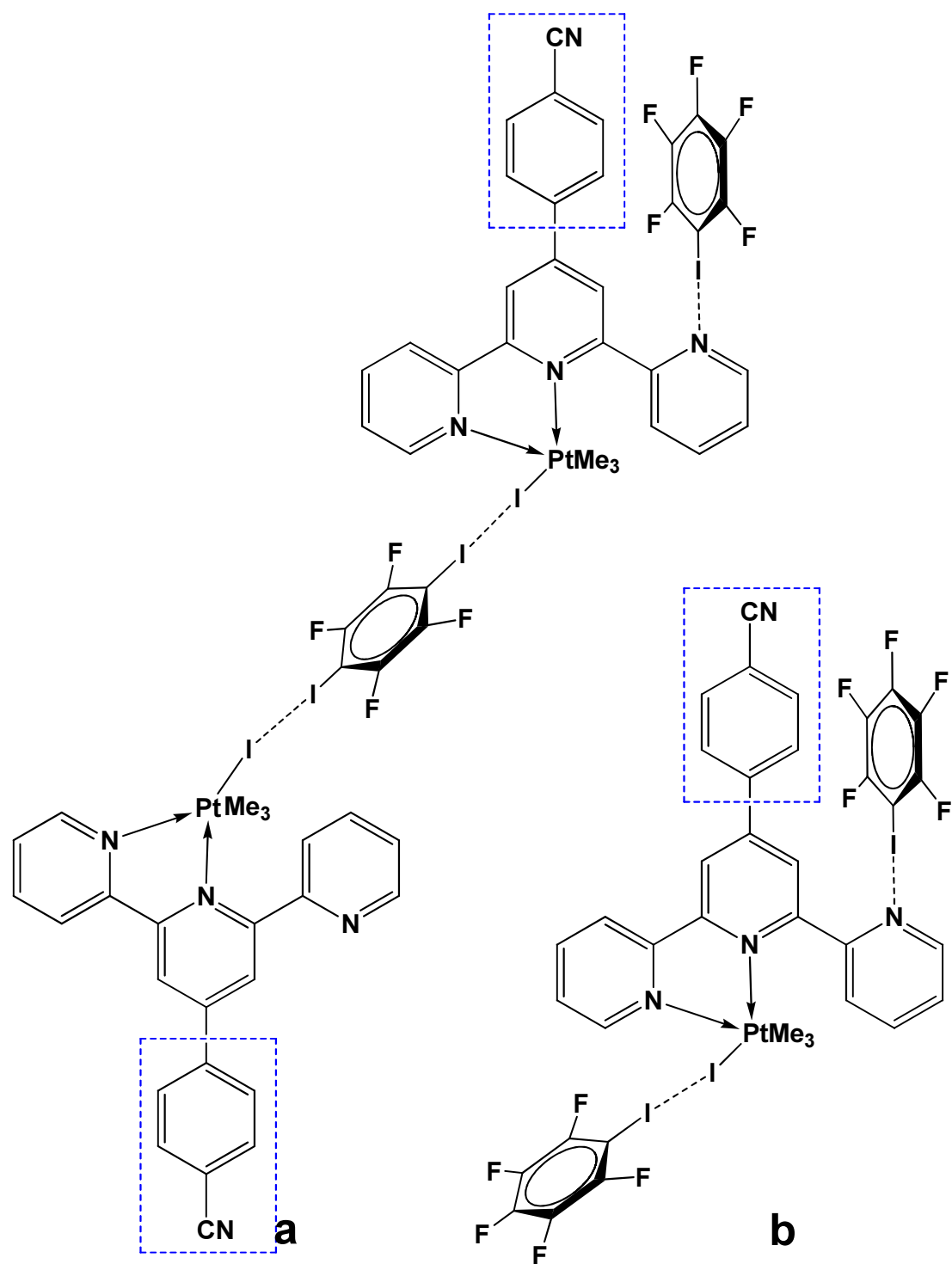
C(6)-N(2) 3.16(1);

**CymRuI<sub>2</sub>(4CNpy)**

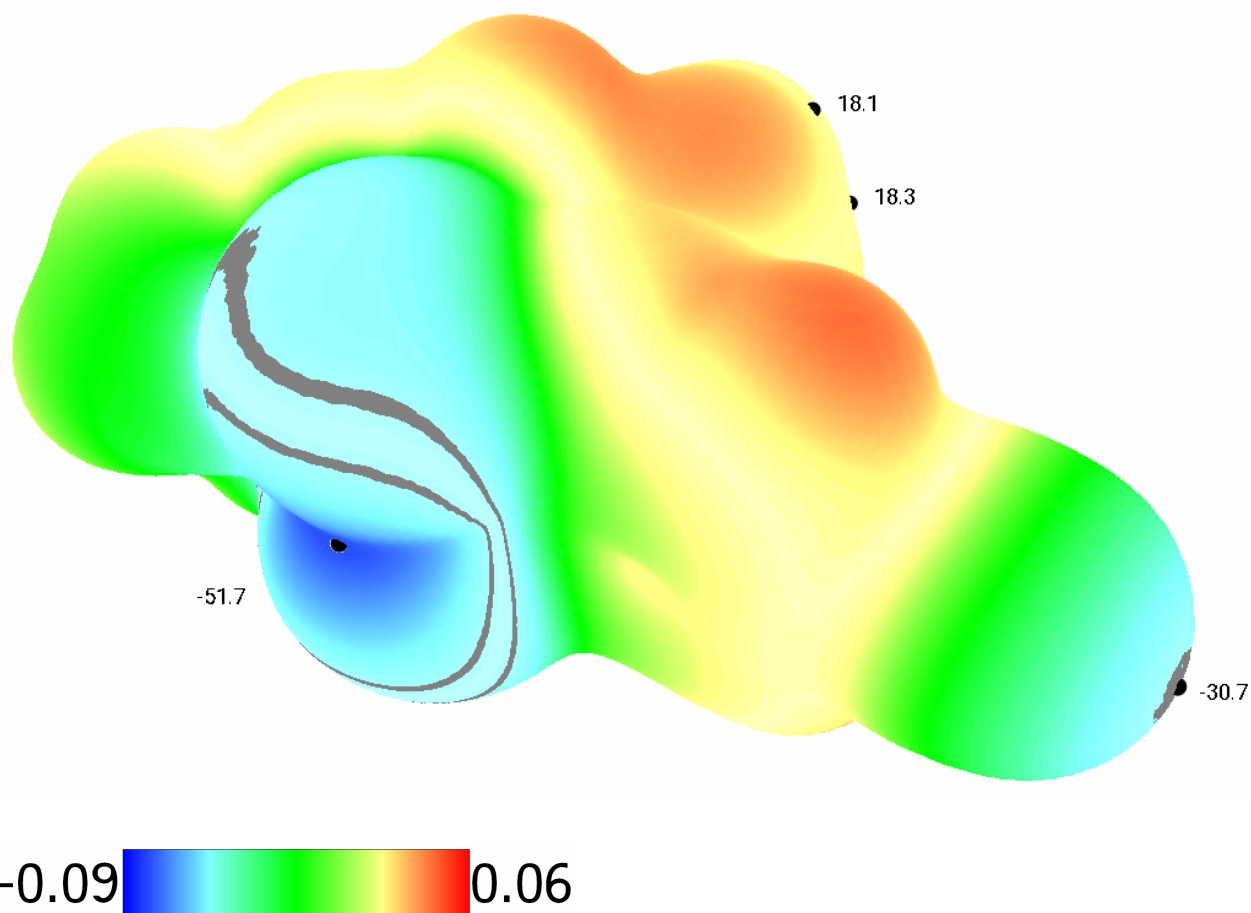
Ru(1)-I(1) 2.741(8)

Ru(1)-I2 2.728(1)

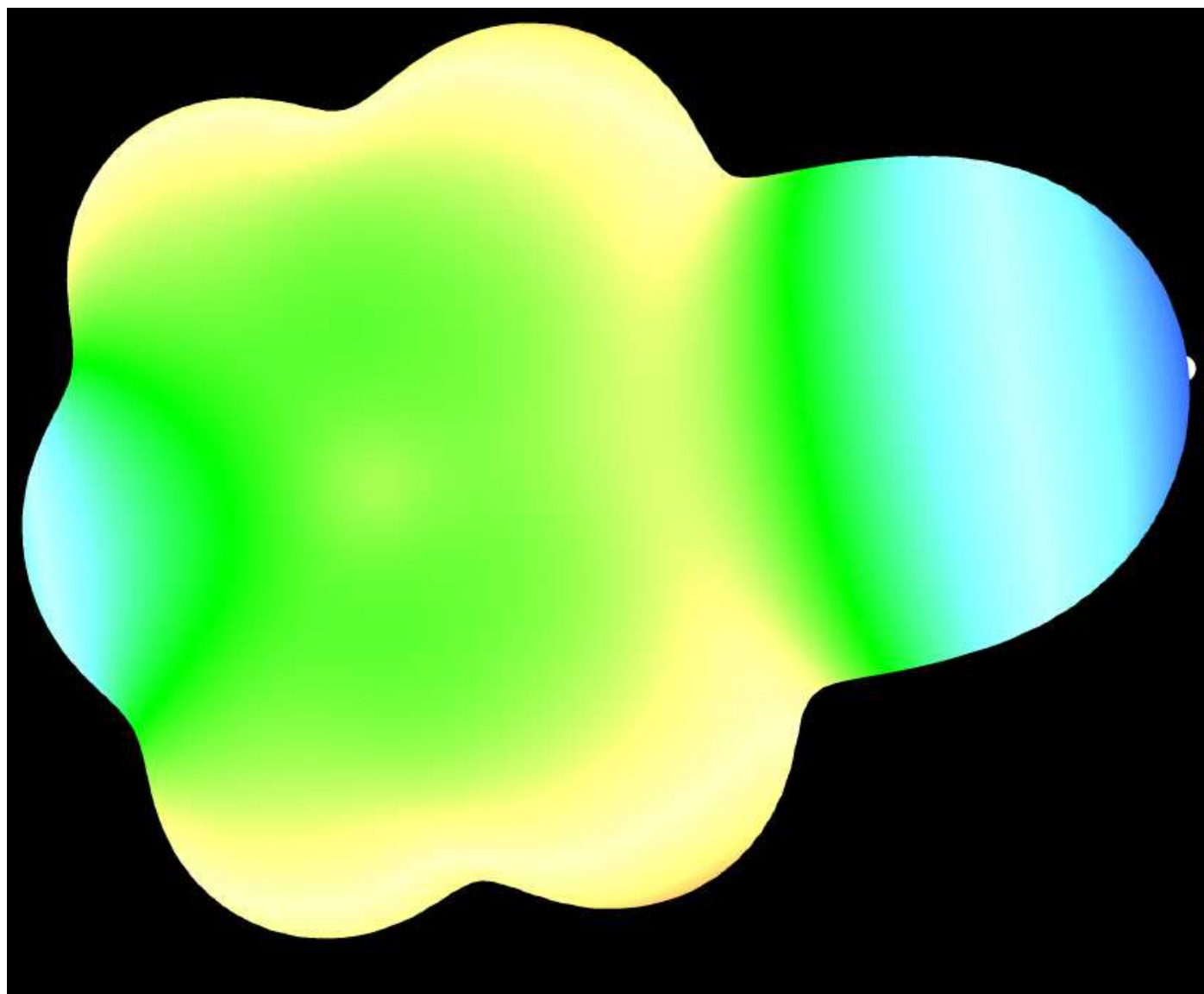
Ru(1)-N(1) 2.135(4)




**Scheme S2** Preference for I---N<sub>py</sub> and I---I-Pt XBs in (4'-(4-cyanophenyl)-2,2':6',2''-terpyridine) trimethyl platinum(IV) iodide / p-C<sub>6</sub>F<sub>4</sub>I<sub>2</sub> (a) and C<sub>6</sub>F<sub>5</sub>I (b) co-crystals [6].

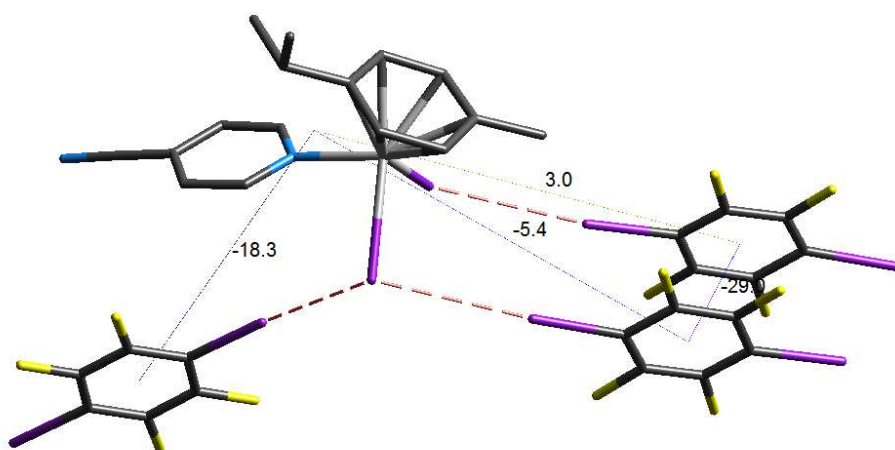


**Figure S5.** Molecular electrostatic potential map for  $\text{CymRuCl}_2(4\text{CNpy})$ , ( $0.001 \text{ e}/\text{\AA}^3$  electron density isosurface), showing the iso-potential areas for Cl and  $V_{\text{max}}$  on CN group (colored in gray).

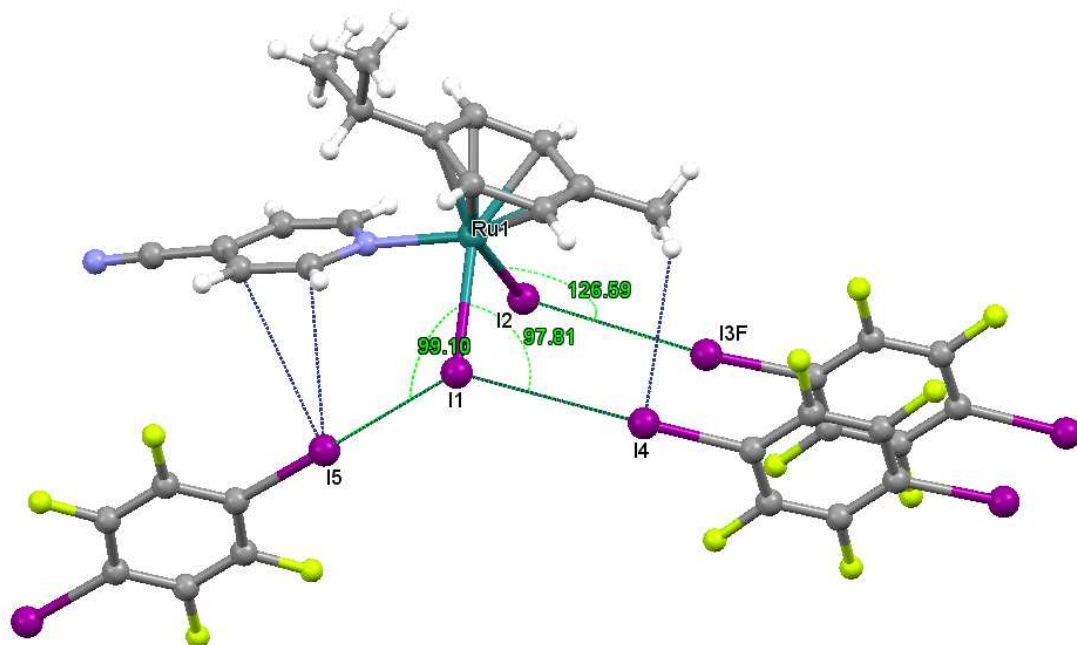


-0.09  0.06

**Figure S6.** Molecular electrostatic potential map for **4CNpy<sub>1</sub>** (0.001 e/A<sup>3</sup> electron density isosurface), showing  $V_{\max}$  on CN group (-33.3 kcal/mol). Compare with N<sub>py</sub> (-26.6 kcal/mol)



a



b

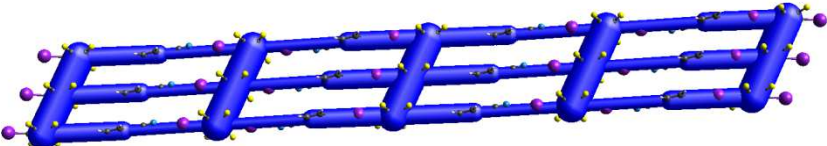
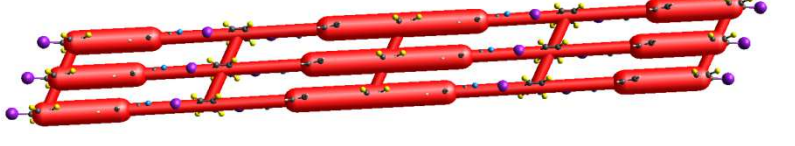
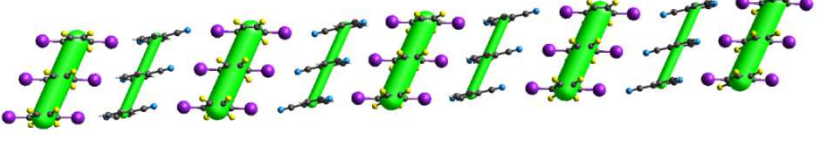
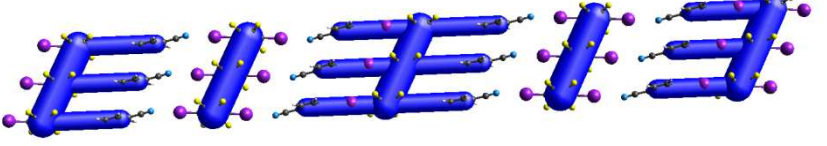
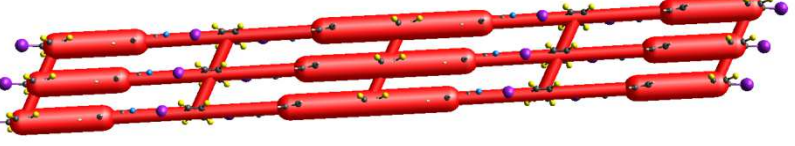
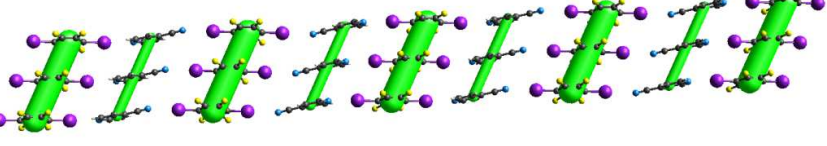
**Figure S7.** Fragment or crystal packing of **CymRu<sub>2</sub>(4CNpy)<sub>p</sub>-DITFB**, showing (a) intermolecular interaction energies (Crystal Explorer 17.5 TONTO B3LYP-DGDZVP , kJ/mol) and (b) I---I halogen bond angles (I---I-Ru).



**Table S5.** Energy frameworks in 2. .a) **total energy** , b) **electrostatic** c) **dispersion** at different cut-off levels.

Note that **total energy** is the sum of the four energy components, **scaled** using the scale factors below:

Energy Model	k_ele	k_pol	k_disp	k_rep
B3LYP/6-31G(d,p) electron densities	1.057	0.740	0.871	0.618

Cutoff (kJ/mol)	Full energy framework	Electrostatic	Dispersion
-12			
-17			
-22	