Electronic Supporting Information

For paper

Hydrogen bonding vs stacking interaction in the crystals of the simplest coumarin derivatives: study from the energetic viewpoint

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Table S1. Pairwise interaction energies (kcal mol^{-1}) of the basic molecule with neighbouring ones in crystal of (I).

	Symmetry operation		Contribution to	
Dimer		E _{int} , kcal the mol ⁻¹ inte	the total	Interaction type
			interaction	
			energy, %	
I_d1	-x,1-y,1-z	-12.85	19.3	N−H···O (2)
I_d2	1+x,y,z	-10.06	15.1	stacking
I_d3	-1+x,y,z	-10.06	15.1	stacking
I_d4	x,1/2-y,1/2+z	-4.02	6.0	C–H···O1
I_d5	x,1/2-y,-1/2+z	-4.02	6.0	C–H···O1
I_d6	1+x, 1/2-y, -1/2+z	-3.49	5.2	C−H···O3
I_d7	-1+x, 1/2-y, 1/2+z	-3.49	5.2	С−Н…ОЗ
I_d8	1-x,1-y,1-z	-3.13	4.7	Stacking C=O
I_d9	2-x,1/2+y,1/2-z	-2.52	3.8	С-НО
I_d10	2-x,-1/2+y,1/2-z	-2.52	3.8	С-НО
I_d11	1+x, 1/2-y, 1/2+z	-2.48	3.7	non-specific
I_d12	-1+x, 1/2-y, -1/2+z	-2.48	3.7	non-specific
I_d13	2-x,-y,1-z	-2.02	3.0	non-specific
I_d14	1-x,1/2+y,1/2-z	-1.62	2.4	non-specific
I_d15	1-x,-1/2+y,1/2-z	-1.62	2.4	non-specific
I_d16	3-x,-y,1-z	-0.22	0.3	non-specific
	Tota	l energy is -66.	58 kcal mol ⁻¹	

Dimer	Symmetry operation	E _{int} , kcal mol ⁻¹	Contribution to the total interaction energy, %	Interaction type	
II_d1	-x,1-y,1-z	-12.39	19.2	N−H···O (2)	
II_d2	1+x,y,z	-10.19	15.8	stacking	
II_d3	-1+x,y,z	-10.19	15.8	stacking	
II_d4	x,1/2-y,1/2+z	-3.54	5.5	C–H···O1	
II_d5	x,1/2-y,-1/2+z	-3.54	5.5	C–H···O1	
II_d6	1+x, 1/2-y, -1/2+z	-3.53	5.5	С–Н⋯ОЗ	
II_d7	-1+x,1/2-y,1/2+z	-3.53	5.5	С–Н…ОЗ	
II_d8	2-x,1/2+y,1/2-z	-2.54	3.9	C-HN	
II_d9	2-x,-1/2+y,1/2-z	-2.54	3.9	C-HN	
II_d10	1-x,1-y,1-z	-2.40	3.7	Stacking C=O	
II_d11	2-x,-y,1-z	-2.06	3.2	non-specific	
II_d12	1+x, 1/2-y, 1/2+z	-1.92	3.0	non-specific	
II_d13	-1+x,1/2-y,-1/2+z	-1.92	3.0	non-specific	
II_d14	1-x,1/2+y,1/2-z	-1.86	2.9	non-specific	
II_d15	1-x,-1/2+y,1/2-z	-1.86	2.9	non-specific	
II_d16	3-x,-y,1-z	-0.37	0.6	non-specific	
	Total energy is -64.40 kcal mol ⁻¹				

Table S2. Pairwise interaction energies (kcal mol⁻¹) of the basic molecule with neighbouring ones in crystal of (II).

			Contribution to	
Dimer	Symmetry operation	E _{int} , kcal	the total	Internation type
		mol ⁻¹	interaction	interaction type
			energy, %	
III_d1	1-x,1-y,2-z	-10.98	15.7	N–H \cdots S (2)
III_d2	1-x,1-y,1-z	-10.13	14.4	stacking
III_d3	1/2-x,1/2+y,3/2-z	-9.48	13.5	C−H····S
III_d4	1/2-x,-1/2+y,3/2-z	-9.48	13.5	C−H···S
III_d5	1-x,-y,1-z	-6.03	8.6	С–Н…О (2)
III_d6	x,1+y,z	-5.37	7.7	Non-specific
III_d7	x,-1+y,z	-5.37	7.7	Non-specific
III_d8	1/2+x, 1/2-y, 1/2+z	-3.50	5.0	N-H…π
III_d9	-1/2+x,1/2-y,-1/2+z	-3.50	5.0	N-H…π
III_d10	1/2+x, 3/2-y, 1/2+z	-2.50	3.6	non-specific
III_d11	-1/2+x,3/2-y,-1/2+z	-2.50	3.6	non-specific
III_d12	1/2-x,1/2+y,1/2-z	-0.64	0.9	non-specific
III_d13	1/2-x,-1/2+y,1/2-z	-0.64	0.6	non-specific
Total energy is -70.10 kcal mol ⁻¹				

Table S3. Pairwise interaction energies (kcal mol⁻¹) of the basic molecule with neighbouring ones in crystal of (III).

			Contribution to	
Dimor	Symmetry operation	E _{int} , kcal	the total	Interaction type
Dinte		mol ⁻¹	interaction	interaction type
			energy, %	
IV_d1	2-x,1-y,1-z	-15.63	24.5	stacking
IV_d2	1-x,1-y,1-z	-14.62	22.9	stacking
IV_d3	1-x,-y,2-z	-10.42	16.3	N−H···S (2)
IV_d4	2-x,-y,1-z	-6.67	10.4	$C-H\cdots S(2)$
IV_d5	x,1+y,z	-3.18	5.0	non-specific
IV_d6	x,-1+y,z	-3.18	5.0	non-specific
IV_d7	1-x,2-y,1-z	-3.00	4.7	non-specific
IV_d8	1-x,1-y,2-z	-2.15	3.4	non-specific
IV_d9	2-x,1-y,-z	-1.95	3.1	non-specific
IV_d10	1+x,y,-1+z	-1.58	2.5	C-HN
IV_d11	-1+x,y,1+z	-1.58	2.5	C-HN
IV_d12	2-x,2-y,-z	0.08	-0.1	non-specific
	Tota	l energy is -63.	87 kcal mol ⁻¹	

Table S4. Pairwise interaction energies (kcal mol⁻¹) of the basic molecule with neighbouring ones in crystal of (**IV**).



Figure S1. Packing of molecules (top) and energy-vector diagram (bottom) in the crystals (I). The projections along a, b and crystallographic direction are presented. The chains/columns are highlighted by different in blue.



Figure S2. Packing of molecules (top) and energy-vector diagram (bottom) in the crystals (II). The projections along a, b and crystallographic direction are presented. The chains/columns are highlighted by different in blue.



Figure S3. Packing of molecules (top) and energy-vector diagram (bottom) in the crystals (III). The projections along a, b and crystallographic direction are presented.



Figure S4. Packing of molecules (top) and energy-vector diagram (bottom) in the crystals (IV). The projections along a, b and crystallographic direction are presented. The stacked column is highlighted in blue. The layer is highlighted in yellow.