

Supporting Information

Sample preparation

Figure S1. TGA curves of **1**.

Table S1. Bond lengths (Å) and bond angles (°) for **1**.

Table S2. Hydrogen bonds (Å, °) for **1**.

Sample preparation

Aqueous solution (30 mL) containing Dabco (0.56 g, 5 mmol), NaBF₄ (0.55 g, 5 mmol) and HBF₄ (3.07 g, 35 mmol) was evaporated to afford colourless block crystals of **1**.

1: yield 80% (based on NaBF₄). Elemental analysis calcd (%) for C₁₂H₂₈B₆F₂₄N₄Na₂ (795.22): C 18.13, H 3.55, N 7.05; found: C 18.22, H 3.50, N 7.04.

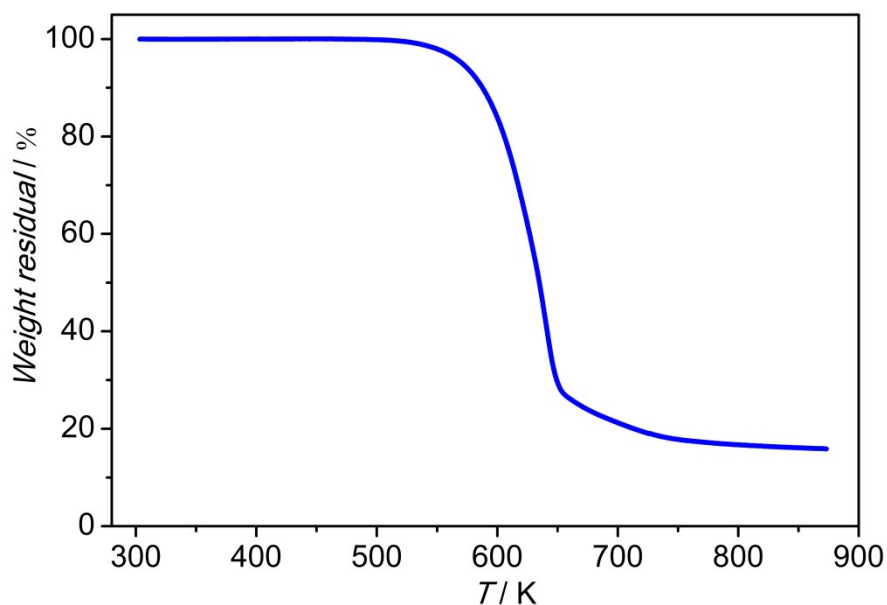


Figure S1. TGA curves of **1**.

Table S1. Bond lengths (Å) and bond angles (°) for **1**.

293 K

Na1–F1	2.847(3)	Na1–F1–B1	107.1(2)
Na1–F1#3	2.847(3)	Na2–F4–B1	126.9(2)
Na1–F1#4	2.847(3)	F1–Na1–F1#3	114.20(6)
Na1–F1#5	2.847(3)	F1–Na1–F1#4	65.80(6)
Na1–F1#6	2.847(3)	F1–Na1–F1#5	180.0
Na1–F1#7	2.847(3)	F1–Na1–F1#6	65.80(6)
Na2–F4	2.496(3)	F1–Na1–F1#7	114.20(6)
Na2–F4#1	2.496(3)	F1#3–Na1–F1#4	180.0
Na2–F4#2	2.496(3)	F1#3–Na1–F1#5	65.80(6)
Na2–F4#8	2.496(3)	F1#3–Na1–F1#6	114.20(6)

Na2–F4#9	2.496(3)	F1#3–Na1–F1#7	65.80(6)
Na2–F4#10	2.496(3)	F1#4–Na1–F1#5	114.20(6)
		F1#4–Na1–F1#6	65.80(6)
		F1#4–Na1–F1#7	114.20(6)
		F1#5–Na1–F1#6	114.20(6)
		F1#5–Na1–F1#7	65.80(6)
		F1#6–Na1–F1#7	180.0
		F4–Na2–F4#1	105.19(6)
		F4–Na2–F4#2	105.19(6)
		F4–Na2–F4#8	180.0
		F4–Na2–F4#9	74.81(6)
		F4–Na2–F4#10	74.81(6)
		F4#1–Na2–F4#2	105.19(6)
		F4#1–Na2–F4#8	74.81(6)
		F4#1–Na2–F4#9	74.81(6)
		F4#1–Na2–F4#10	180.0
		F4#2–Na2–F4#8	74.81(6)
		F4#2–Na2–F4#9	180.0
		F4#2–Na2–F4#10	74.81(6)
		F4#8–Na2–F4#9	105.19(6)
		F4#8–Na2–F4#10	105.19(6)
		F4#9–Na2–F4#10	105.19(6)

Symmetry codes: #1 z, x, y ; #2 y, z, x ; #3 $-y+1/2, z-1/2, x$; #4 $y+1/2, -z+1/2, -x+1$; #5 $-x+1, -y, -z+1$; #6 $-z+1, x-1/2, -y+1/2$; #7 $z, -x+1/2, y+1/2$; #8 $-x+1, -y+1, -z+1$; #9 $-y+1, -z+1, -x+1$; #10 $-z+1, -x+1, -y+1$.

Table S2. Hydrogen bonds (Å, °) for **1** at 293 K.

D–H···A	D–H	D···A	D–H···A
N1–H1C···F2	0.98	2.896(3)	124
N1–H1C···F2 [z, x, y]	0.98	2.896(3)	124
N1–H1C···F2 [y, z, x]	0.98	2.896(3)	124
N2–H2C···F1 [$x-1/2, y, -z+1/2$]	0.98	2.931(3)	127
N2–H2C···F1 [$-z+1/2, x-1/2, y$]	0.98	2.931(3)	127
N2–H2C···F1 [$y, -z+1/2, x-1/2$]	0.98	2.931(3)	127