

## Supporting Information

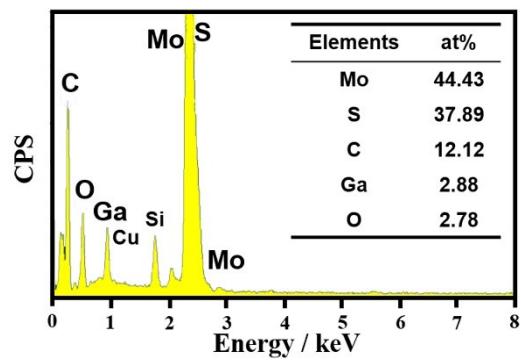
# 2D Organ-like Molybdenum Carbide (MXene) Coupled with MoS<sub>2</sub> Nanoflowers Enhances the Catalytic Activity in the Hydrogen Evolution Reaction

Jie Ren<sup>a</sup>, Hui Zong<sup>a</sup>, Yuyun Sun<sup>a</sup>, Yu Feng<sup>a</sup>, Zhenguo Wang<sup>a</sup>, Le Hu<sup>a</sup>, Shijing Gong<sup>a</sup>,  
Ke Yu<sup>\*a,b</sup> and Ziqiang Zhu<sup>a</sup>

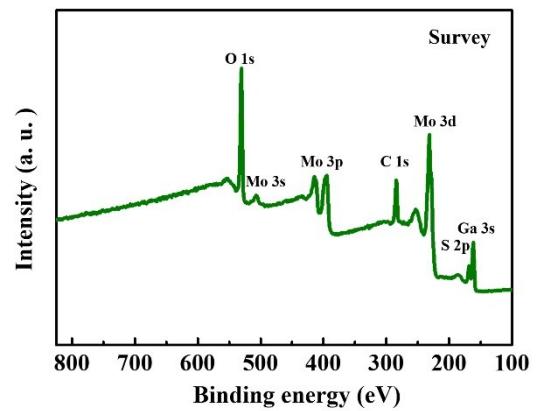
a Key Laboratory of Polar Materials and Devices (MOE), Department of Electronics,  
East China Normal University, Shanghai 200241, China

b Collaborative Innovation Center of Extreme Optics, Shanxi University, Taiyuan,  
Shanxi 030006, People's Republic of China

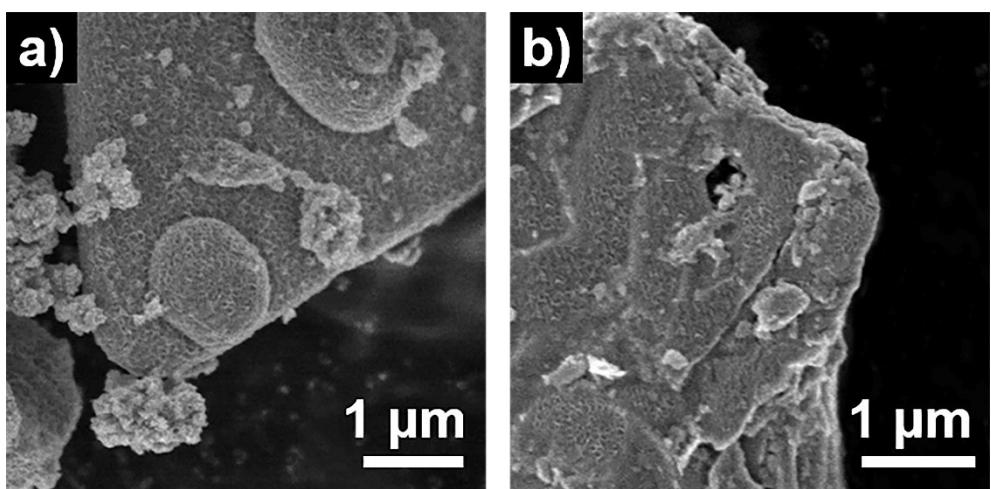
\* Author to whom correspondence should be addressed. E-mail: yk5188@263.net. Tel:  
+86-21-54345198.



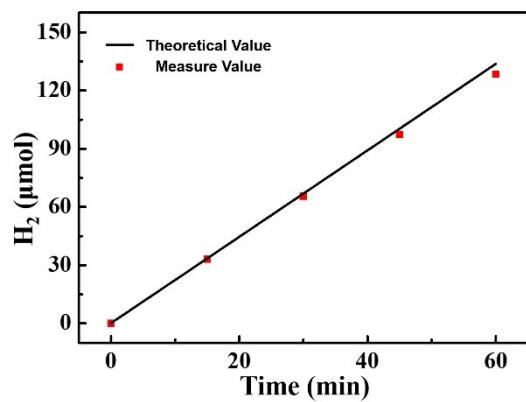
**Fig. S1** EDX spectrum of  $\text{MoS}_2@\text{Mo}_2\text{CT}_x$ .



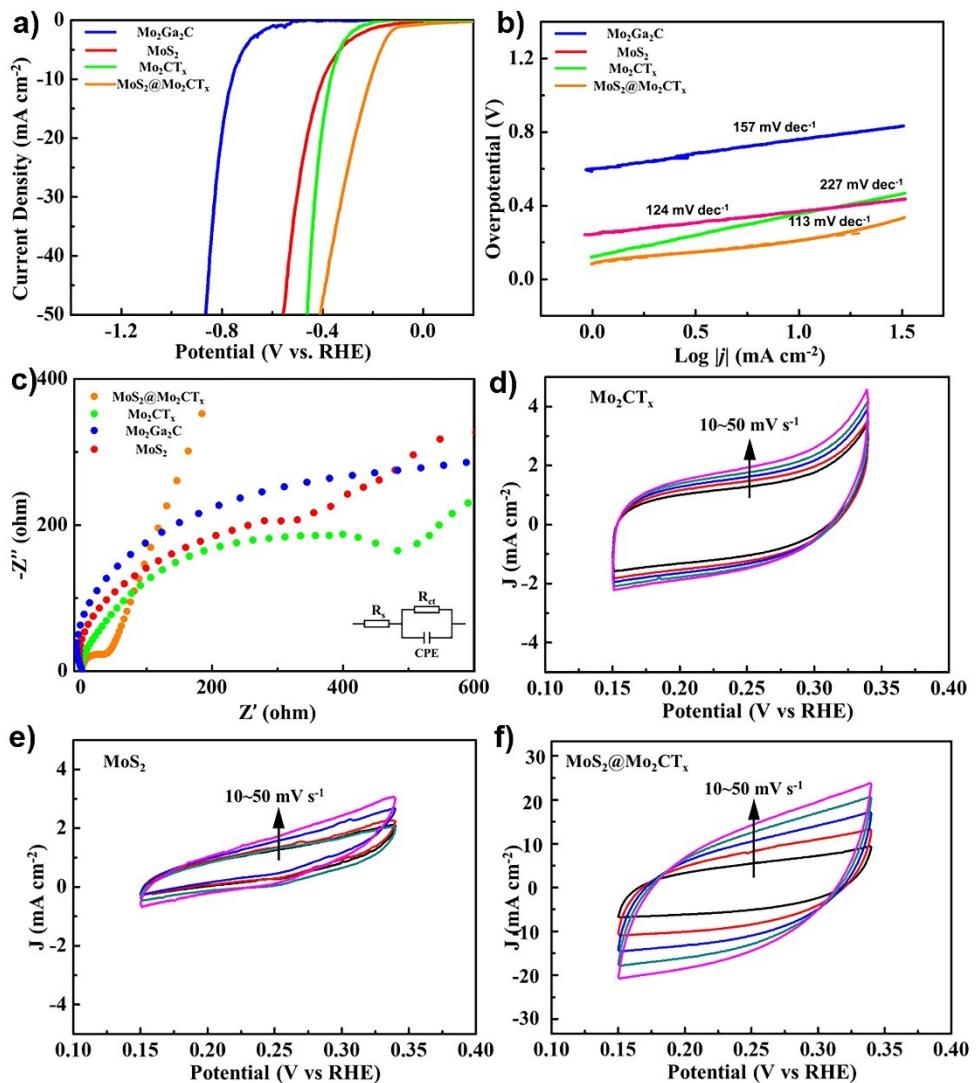
**Fig. S2** XPS survey spectra of  $\text{MoS}_2@\text{Mo}_2\text{CT}_x$  nanohybrids



**Fig. S3** (a) SEM of  $\text{MoS}_2@\text{Mo}_2\text{CT}_x$  at the beginning of HER; (b) SEM of the composite after 8 hours.



**Fig. S4** Electrocatalytic efficiency of H<sub>2</sub> production on MoS<sub>2</sub>@Mo<sub>2</sub>CT<sub>x</sub> at a potential of ca. -400 mV, measured for 60 min.



**Fig. S5** (a) Polarization curves of different materials at a scanning rate of 10 mV s<sup>-1</sup> in 0.5 M H<sub>2</sub>SO<sub>4</sub>; (b) Tafel plots of different materials; (c) EIS spectra of different materials over the frequency range from 100 kHz to 10 Hz at  $\eta = 400$ ; cyclic voltammograms of MoS<sub>2</sub>@Mo<sub>2</sub>CT<sub>x</sub>, within different rates ranging from 10 to 50 mV s<sup>-1</sup> in the region from 0.15 to 0.34 V.



**Fig. S6** Digital photograph of the three-electrode system with the ion exchange membrane. The commercial Pt/C electrode and Ag/AgCl electrode were used as the counter electrode and reference electrode respectively.

**Table S1** A comparison of MoS<sub>2</sub>@Mo<sub>2</sub>CT<sub>x</sub> electrocatalyst with recently reported non-noble metal catalysts in HER performance (1 M KOH).

Catalysts	Overpotential at $j=10 \text{ mA cm}^{-2}$ (mV)	Tafel slope (mV dec <sup>-1</sup> )	References	Cites
MoS <sub>2</sub> @Mo <sub>2</sub> CT <sub>x</sub>	176	207	This work	
MoS <sub>2</sub> /Ti <sub>3</sub> C <sub>2</sub> -MXene@C	135	45	<i>Adv. Mater.</i> , 2017, <b>29</b> , 1607017	S1
Mo <sub>2</sub> C@2D-NPC	45	52	<i>ACS Nano</i> , 2017, <b>4</b> , 3933-3942	S2
Mo <sub>2</sub> C	195	67	<i>ACS Nano</i> , 2017, <b>4</b> , 3933-3942	S2
Co <sup>3+</sup> -Cr <sub>2</sub> CT <sub>x</sub>	404	137	<i>J. Am. Chem. Soc.</i> , 2019, <b>141</b> , 9610-9616	S3
Nb-doped Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> MXene	445	154	<i>Adv. Sci.</i> , 2019, <b>11</b> , 1900116.	S4
N,P-doped Mo <sub>2</sub> C@C	156	87	<i>ACS Nano</i> , 2016, <b>9</b> , 8851-8860	S5
Mo <sub>2</sub> CT <sub>x</sub> :Co	180	59	<i>J. Am. Chem. Soc.</i> , 2019, <b>141</b> , 17809-17816	S6
2H MoSe <sub>2</sub> /Ti <sub>3</sub> C <sub>2</sub> MXene	152	211	<i>Electrochim. Acta</i> , 2019, <b>326</b> , 134976	S7

**Table S2** A comparison of MoS<sub>2</sub>@Mo<sub>2</sub>CT<sub>x</sub> electrocatalyst with recently reported non-noble metal catalysts in HER performance (0.5 M H<sub>2</sub>SO<sub>4</sub>).

Catalysts	Overpotential at j=10 mA cm <sup>-2</sup> (mV)	Tafel slope (mV dec <sup>-1</sup> )	References	Cites
MoS <sub>2</sub> @Mo <sub>2</sub> CT <sub>x</sub>	176	113	<b>This work</b>	
Ti <sub>2</sub> C	609	124	<i>ACS Energy Lett.</i> , 2016, 1(3), 589-594	S8
Mo <sub>2</sub> C	283	82	<i>ACS Energy Lett.</i> , 2016, 1(3), 589-594	S8
Mo <sub>2</sub> CT <sub>x</sub>	189	75	<i>ACS Energy Lett.</i> , 2016, 1(3), 589-594	S8
L-Mo <sub>2</sub> C	145	157	<i>ACS Appl. Mater. Interfaces</i> , 2018, <b>47</b> , 40500-40508	S9
Ti <sub>3</sub> C <sub>2</sub> Flakes	390	188	<i>ACS Sustain. Chem. Eng.</i> , 2018, <b>6</b> , 8976-8982	S10
Layered-Ti <sub>2</sub> CT <sub>x</sub> (F-term)	265	138	<i>Nano Energy</i> , 2018, <b>47</b> , 512-518	S11
N-Ti <sub>2</sub> CT <sub>x</sub>	215	67	<i>J. Mater. Chem. A</i> , 2018, <b>6</b> , 20869-20877	S12

## Supplementary References

- S1. X. Wu, Z. Wang, M. Yu, L. Xiu and J. Qiu, *Adv. Mater.*, 2017, **29**, 1607017.
- S2. C. Lu, D. Tranca, J. Zhang, F. N. Rodri Guez Hernandez, Y. Su, X. Zhuang, F. Zhang, G. Seifert and X. Feng, *ACS Nano*, 2017, **11**, 3933-3942.
- S3. S. Y. Pang, Y. T. Wong, S. Yuan, Y. Liu, M. K. Tsang, Z. Yang, H. Huang, W. T. Wong and J. Hao, *J. Am. Chem. Soc.*, 2019, **141**, 9610-9616.
- S4. J. Yu, G. Li, H. Liu, L. Zeng, L. Zhao, J. Jia, M. Zhang, W. Zhou, H. Liu and Y. Hu, *Adv. Sci. (Weinh)*, 2019, **6**, 1901458.
- S5. Y. Y. Chen, Y. Zhang, W. J. Jiang, X. Zhang, Z. Dai, L. J. Wan and J. S. Hu, *ACS Nano*, 2016, **10**, 8851-8860.
- S6. D. A. Kuznetsov, Z. Chen, P. V. Kumar, A. Tsoukalou, A. M. Kierzkowska, P. M. Abdala, O. V. Safonova, A. Fedorov and C. R. Muller, *J. Am. Chem. Soc.*, 2019, **141**, 17809-17816.
- S7. N. Li, Y. Zhang, M. Jia, X. Lv, X. Li, R. Li, X. Ding, Y. Z. Zheng and X. Tao, *Electrochim. Acta*, 2019, **326**, 134976.
- S8. J. Xiong, J. Li, J. Shi, X. Zhang, N. T. Suen, Z. Liu, Y. Huang, G. Xu, W. Cai, X. Lei, L. Feng, Z. Yang, L. Huang and H. Cheng, *ACS Energy Lett.*, 2018, **3**, 341-348.
- S9. W. Yuan, Q. Huang, X. Yang, Z. Cui, S. Zhu, Z. Li, S. Du, N. Qiu and Y. Liang, *ACS Appl. Mater. Interfaces*, 2018, **10**, 40500-40508.
- S10. T. A. Le, Q. V. Bui, N. Q. Tran, Y. Cho, Y. Hong, Y. Kawazoe and H. Lee, *ACS Sustain. Chem. Eng.*, 2018, **6**, 8976-8982.
- S11. S. Li, P. Tuo, J. Xie, X. Zhang, J. Xu, J. Bao, B. Pan and Y. Xie, *Nano Energy*, 2018, **47**, 512-518.
- S12. Y. Yoon, A. P. Tiwari, M. Lee, M. Choi, W. Song, J. Im, T. Zyung, H. K. Jung, S. S. Lee, S. Jeon and K. S. An, *J. Mater. Chem. A*, 2018, **6**, 20869-20877.